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5,5'-Diamino-BIPHEP ligands bearing small selector units for non-covalent binding of chiral analytes in solution†

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A dynamic axially chiral BIPHEP-ligand with 3,5-dichlorobenzoyl amide selector units for non-covalent binding of phenylalanine derivatives has been developed. Interaction studies in solution were performed with rhodium(II) complexes under exclusion of the metal being involved in binding. (*R*_{ax}, *S*^{Phe}) and (*S*_{ax}, *S*^{Phe}) adducts were observed as significantly separated species in NMR spectroscopy.

Interactions between chiral substances represent a fundamental principle in chiral amplification, self-organisation and chirality sensing. This requires covalent, coordinative or non-covalent binding of the chiral molecules. Impressive examples of enantioselective self-assembly¹ and binding of chiral compounds² have been reported in supramolecular chemistry for capsules and clusters.

On a molecular level, it has been shown that metal complexes are able to differentiate between enantiomers in a quite effective way. One interesting field of application is the sensing of enantiomeric excesses of small chiral molecules by complexation to transition metals bearing chiral ligands. Canary *et al.*^{3,4} and Anslyn *et al.*^{3,5} reported copper complexes with chiral ligands for direct enantiomeric excess determination *via* CD spectroscopy or colorimetric analysis. Wolf *et al.*⁶ demonstrated palladium complexes with dynamic chiral ligands for this purpose.

The formation of diastereomeric substrate–catalyst adducts that occur in metal catalyzed enantioselective hydrogenation (*re* or *si* coordination) represents a related case of asymmetric recognition.⁷ The *vice versa* process of metal-coordinated and (–)-menthol functionalised substrates, selectively coordinating one additional phosphine ligand enantiomer upon addition of a racemic ligand mixture, has been shown as well.⁸

Impressive results in enantioselective catalysis have been shown by Mikami *et al.* utilizing chiral diamines as aligning ligands for the control of a dynamic chiral phosphine ligand.^{9,10} Previously, Mikami and Noyori reported ruthenium and titanium complexes bearing

racemic ligands in combination with enantiopure diamine or biphenol co-ligands as efficient catalytic systems for asymmetric activation.^{11,12} Faller *et al.* used a similar approach with racemic rhodium complexes and chiral phosphorous ligand additives.^{13,14}

Since free coordination sites are blocked by the aligning ligands, they often have to be cleaved off prior to catalysis.¹⁵ Brown *et al.* and Mikami *et al.* reported a strategy to overcome this limitation by using elaborated enantiopure chiral COD-analogues that are cleaved off under catalysis conditions for alignment of dynamic chiral ligands.^{16,17} Another approach focused on the use of chiral ionic liquids with free amino functionalities for metal coordination.^{18,19}

However, the common ground of all these findings is that a chiral analyte is introduced *via* metal coordination. Not only does this render the metal centre a prerequisite for the concept to work; it also often limits further catalytic applications at the metal centre. Furthermore, all mentioned examples share the requirement of the analyte to bind to the specific metal, *e.g.* *via* free amines or phosphines, which intrinsically excludes analytes without these groups. Our investigations were focused on changing this principle and introducing selector groups for non-covalent analyte binding. We consider this new concept as highly promising, because no metal vacancy is blocked by coordination and thus can be utilised for further purposes (Fig. 1).

We focused on the synthesis of a 5,5'-diamino functionalised BIPHEP ligand enabling selector bonding. Although axially chiral BINAP ligands bearing methylamino²⁰ or amino²¹ groups in the backbone are known in literature, these molecules were not suitable for our purpose for several reasons: methylamino groups were considered to be too flexible, thus direct bonding of the NH₂-moiety was considered to be a more promising approach because the enhanced rigidity should have a beneficial effect on analyte binding. Since future studies will aim at controlling the axial stereochemistry, the tropos nature of BIPHEP-type ligands was also advantageous. Modification of BIPHEP-ligands in the 6-position leads to atropos ligands as it is the case for the widely used MeO-BIPHEP ligand family. In contrast, the 3-position directs the selector to close proximity of the diphenylphosphino groups potentially making substrate coordination difficult due to additional steric repulsion. Little is known about effects of substitution at the

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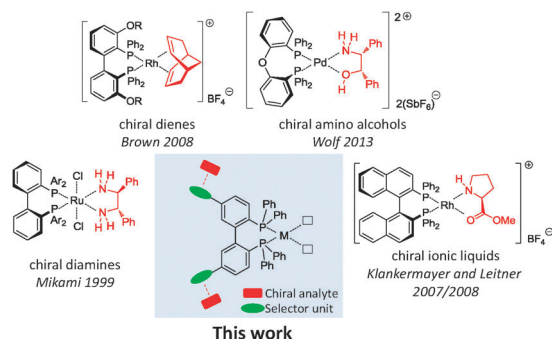


Fig. 1 Previous investigations of complexes bearing chiral or dynamic chiral ligands in binding of chiral analytes. The fundamental concept of our new strategy is depicted centred.

4-position,²² however, it can be assumed that substituents in this position will have less influence on the axial stereochemistry of the BIPHEP-ligand. Therefore, we decided to develop a synthetic strategy for the 5-substituted derivative (Fig. 2) in line with our previously investigated hydroxy functionalised BIPHEPs.²³

The synthetic procedure starts with nitration of commercially available *ortho*-biphenol²⁴ and selective crystallisation of the 5,5'-dinitro derivative. Conversion of the phenol groups into triflates and reduction of the nitro groups yielded **1**. *N*-Boc-protection allowed for subsequent palladium catalysed C–P-coupling with HP(O)Ph₂. Acidic cleavage thus led to **2** which can easily be isolated as hydrogen chloride salt (Fig. 2A).

Several types of selector units can be attached to ligand backbone **2**. We chose state of the art chiral HPLC column selectors, *i.e.* interacting sites of amylose and cellulose CSPs (Chiralpak IC/IE series) as model and focused on the 3,5-dichlorophenyl moiety attached to the ligand core *via* carboxylic acids or isocyanates. The resulting amide or urea NH and CO functionalities as well as the electron deficient aromatic rings are known for efficient binding of analytes in HPLC stationary phase design. Reaction with acyl chlorides readily yielded the corresponding amides exemplified by the preparation of **3**. Amide bond formation can also be applied to the established coupling of diamides with EEDQ as it was shown for

glycine derivative **4**. In even closer analogy to HPLC stationary phases, reaction with isocyanates resulted in urea derivative **5** (Fig. 2B).

Selective reduction of the phosphine oxides was achieved with PhSiH₃. Subsequent coordination to the metal precursor [Rh(COD)₂](BF₄) led to the corresponding rhodium(i) complex (Fig. 3A). X-ray crystallographic investigation²⁵ illustrates the twisted biphenyl core with metal coordinated diphenylphosphino groups on one and the rigid 3,5-dichlorobenzoyl amides on the other side (Fig. 3B). Reduction and coordination was investigated for selector modified BIPHEP oxides **3** and **4**. However, it was found that the reduction product of **4** gives rise to very broad signals in the NMR-spectra in non-polar solvents rendering interaction studies almost impossible. Since very sharp signals were observed in case of complex **6**, we focused on this derivative in the current study.

We subsequently investigated derivatives of the amino acid phenylalanine as interacting small chiral analytes. Derivatives of phenylalanine are of great pharmaceutical importance, *e.g.* in the case of Parkinson's drug L-3,4-dihydroxyphenylalanine (L-DOPA). Besides, they represent the product of enantioselective catalyses, thus samples of varying enantiomeric excess are often analysed. Phenylalanine bears good hydrogen bond donor and acceptor properties and derivatives can easily be obtained by modification of the amino and carboxy functionalities.

Complex **6** was dissolved in CDCl₃ and NMR spectra were recorded with two, five and ten equivalents of (*S*)-phenylalanine (*S*^{Phe}) derivative. We were interested in signal splitting due to dynamic formation of (*R*_{ax}, *S*^{Phe}) and (*S*_{ax}, *S*^{Phe}) adducts. (*S*)-Acetylphenylalanine methyl ester **7** was the first derivative to be investigated. While no interaction was detected with two or five equivalents of substrate, a small splitting (9 ppb) of the complex's NH-groups in ¹H-NMR was observed in the case of ten equivalents amidoester (Table 1).

With this result, we implemented electron rich aromatic substituents to the amino acid in order to potentially improve substrate binding efficiency. We thus studied amide bond formation products

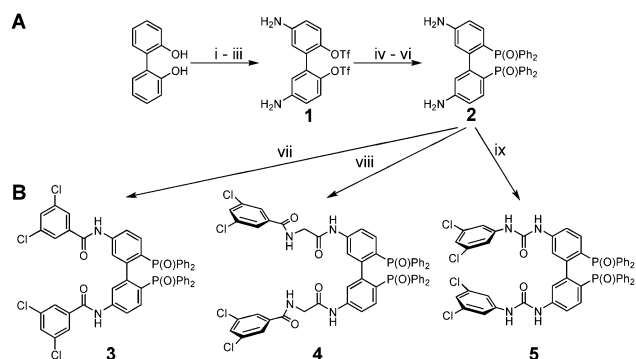


Fig. 2 Synthetic overview of selector modified BIPHEP oxides. (A) Preparation of the ligand backbone: (i) HOAc, HNO₃, 16%, (ii) Tf₂O, pyridine, 85%, (iii) Fe, aq. HCl, 84%, (iv) Boc₂O, 84%, (v) Pd₂(dba)₃, dppb, DIPEA, HP(O)Ph₂, 26%, (vi) HCl in 2-propanol, 93%; (B) modification of **2** with various types of selectors: (vii) 3,5-dichlorobenzoyl chloride, pyridine, 92%, (viii) *N*-3,5-dichlorobenzoyl glycine, EEDQ, 60%, (ix) 3,5-dichlorophenyl isocyanate, 52%.

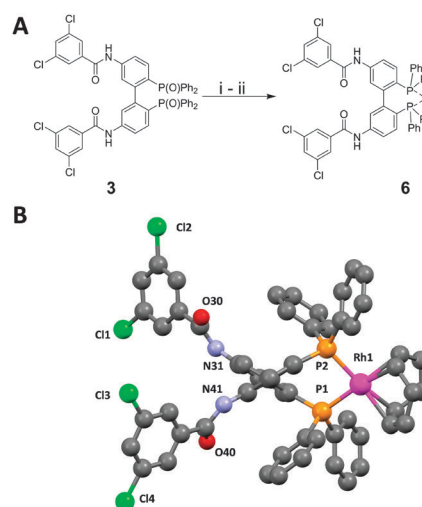
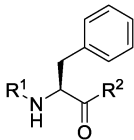

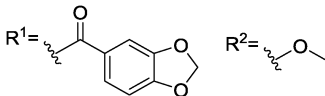
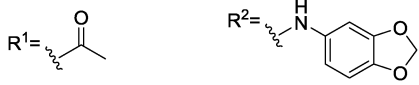
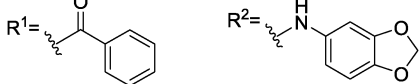
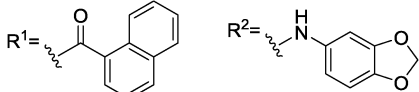
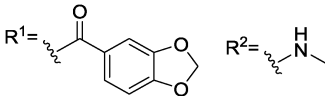



Fig. 3 Synthesis and characterization of rhodium(i)-complex **6**. (A) Reduction of phosphine oxide **3** and subsequent coordination to a rhodium(i)-precursor: (i) PhSiH₃, 70%; (ii) [Rh(COD)₂](BF₄), 88%; (B) X-ray crystallographic investigation of **6**. The depicted structure represents the view along the central biphenyl C–C bond. Hydrogen atoms and the BF₄[−] counter anion are omitted for clarity.

Table 1 Non-covalent interaction of complex **6** with various derivatives of (*S*)-phenylalanine in solution: The $\Delta\delta$ values are shown for the amide NH signal of **6** in ^1H -NMR and the phosphorous resonance (in brackets) in $^{31}\text{P}\{^1\text{H}\}$ -NMR

			2 eq. ^a $\Delta\text{ppb } ^1\text{H (NH), } [^{31}\text{P}]$	5 eq. ^a $\Delta\text{ppb } ^1\text{H (NH), } [^{31}\text{P}]$	10 eq. ^a $\Delta\text{ppb } ^1\text{H (NH), } [^{31}\text{P}]$
7			—, [—]	—, [—]	9, [—]
8			—, [—]	—, [—]	—, [—]
9			92, [—] ^b	146, [—] ^b	171, [—] ^c
10			—, [12]	9, [25] ^b	22, [37] ^c
11			10, [12] ^b	27, [25] ^c	26, [26] ^c
12			57, [27]	107, [49]	117, [51] ^c
13			322, [10]	406, [18]	445, [27] ^b

^a 6.7 mM **6** in dry degassed CDCl_3 . Diamide concentration was varied. ^b Substrate dissolves completely upon warming. ^c Incomplete dissolution of substrate.

from piperonyl chloride (**8**) and 3,4-(dioxymethylene)aniline (**9**). Interestingly, no interaction took place in case of **8** while a splitting of 92 ppb (^1H -NMR, amide NH of **6**) was already observed in the presence of two equivalents of **9**.

Next, we aimed at increasing the steric demand at the amino group of **9** by replacing the acetyl moiety by benzoyl (**10**) and 1-naphthoyl (**11**) groups. These enlarged derivatives should potentially lead to increasing repulsion with the diphenylphosphino groups. Although **10** and **11** were only moderately soluble in CDCl_3 (saturated solutions were obtained with less than 10 equivalents), both showed splitting of the $^1\text{P-Rh}$ doublet in $^{31}\text{P}\{^1\text{H}\}$ -NMR (Table 1) for the first time.

At this point, we concluded that the most crucial part for strong non-covalent binding would be the presence of two amide bonds. It seemed reasonable to include the methylamide analogues of **7** and **8**. Both can easily be obtained by aminolysis of the corresponding methyl esters. *N*-Piperonyl derivative **12** was found to lead to significant splitting in ^1H -NMR and remarkable splitting of 0.05 ppm in $^{31}\text{P}\{^1\text{H}\}$ -NMR. Intriguingly, the small acetyl derivative **13** gives the highest values regarding splitting in ^1H -NMR. The two adducts of ($R_{\text{ax}}, S^{\text{Phe}}$) and ($S_{\text{ax}}, S^{\text{Phe}}$) are separated by 0.45 ppm (NH signal) in ^1H -NMR spectra with 10 equivalents of diamide (Table 1 and Fig. 4A).

In all cases, the splitting effect is not limited to the NH signal in ^1H -NMR and can also be observed, for example, when analysing the

3,5-dichlorobenzoyl H^{Ar} signals (see ESI[†]). This supports our initial idea of amide hydrogen bonding and π -stacking of the 3,5-dichlorobenzoyl fragment being the predominant interaction sites.

It has to be noted that binding of the analyte to the metal centre can be excluded because control experiments with [(BIPHEP)Rh(COD)](BF_4) did not show any splitting even with ten equivalents of **12** or **13** (Fig. 4D and ESI[†]). Besides, adding ten equivalents of **13** to the free ligand results in splitting of signals as well (see ESI[†]).

Next, we verified that the two species observed correspond to the adducts ($R_{\text{ax}}, S^{\text{Phe}}$) and ($S_{\text{ax}}, S^{\text{Phe}}$). Thus, we repeated the interaction experiment using two equivalents of racemic **13**. This led to observation of one single species in the NMR spectrum without any signal splitting. In contrast, signal splitting was observed when adding **13** with 50% ee but the Δppm value was significantly smaller compared to enantiopure **13** (Fig. 4B). It has to be noted, that the two NH-signals do not correspond to R^{Phe} and S^{Phe} , as it would be the case for common chiral shift reagents, thus their relative intensity does not change with different ee values of the analyte. Splitting is the result of the complex sensing the chirality of the analyte through non-covalent coordination to the selector units forming two pairs of adducts: ($R_{\text{ax}}, S^{\text{Phe}}$) and ($S_{\text{ax}}, S^{\text{Phe}}$). These adducts are diastereomeric and thus are observed at different chemical shifts in NMR spectra. Coordination and de-coordination can occur rapidly in solution.

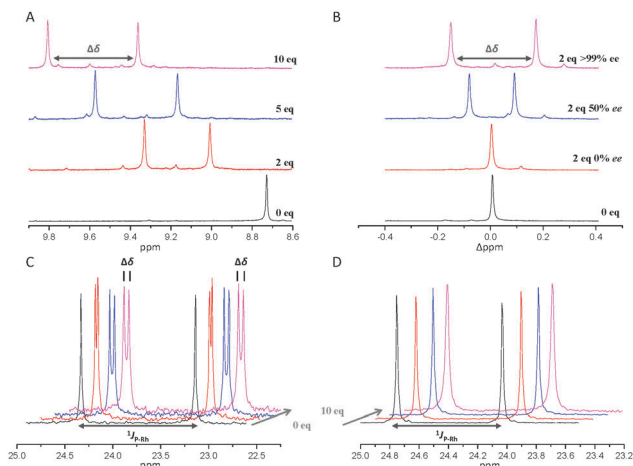


Fig. 4 NMR-spectra with varying amounts of phenylalanine analyte. (A) ^1H -NMR spectra of **6** with 0, 2, 5 and 10 equivalents (*S*)-**13**. The amide NH region of **6** is depicted. (B) ^1H -NMR spectra of **6** with 2 equivalents of **13** of 0, 50 and >99% ee (*S*-enantiomer). The relative deviation of the initial resonance is depicted. (C) $^{31}\text{P}\{^1\text{H}\}$ -NMR of **6** with 0, 2, 5 and 10 equivalents of (*S*)-**12**. (D) $^{31}\text{P}\{^1\text{H}\}$ -NMR of $[(\text{BIPHEP})\text{Rh}(\text{COD})](\text{BF}_4)$ with 0, 2, 5 and 10 equivalents of (*S*)-**12**.

The observed chemical shift of the R_{ax} -enantiomer in case of an analyte of 50% ee is the weighted average of 75% ($R_{\text{ax}}, S^{\text{Phe}}$) and 25% ($R_{\text{ax}}, R^{\text{Phe}}$). This is why coalescence occurs for the signal splitting of racemic **13** because the observed chemical shift for R_{ax} is the weighted average of 50% ($R_{\text{ax}}, S^{\text{Phe}}$) and 50% ($R_{\text{ax}}, R^{\text{Phe}}$) while it is the completely enantiomeric 50% ($S_{\text{ax}}, S^{\text{Phe}}$) and 50% ($S_{\text{ax}}, R^{\text{Phe}}$) for S_{ax} . The splitting correlates linearly with the ee of the added diamide analyte: 0.45 ppm splitting is observed for ten equivalents enantiopure **13**, while a value of 0.23 ppm is observed with the same amount of analyte of 50% ee, which is in good agreement with the theoretically halved value (see ESI† for a plot and measurements with analyte of 25 and 75% ee). This represents a rare example of a racemic complex for the determination of ee values by simply measuring NMR spectra and evaluating signal splitting. The method is related to the pro-chiral solvating-agent (pro-CSA) concept recently reported by Hill and Ariga with achiral porphyrine host molecules.^{26–28}

In conclusion, we described the 5,5'-diamino-BIPHEP ligand as a versatile starting point for the synthesis of selector modified stereodynamic ligands. Our investigations were focused on the corresponding complexes obtained with rhodium(i), yet further applications may arise from immobilization of the ligands or formation of ionic adducts.

The rhodium complex bearing a 3,5-dichlorobenzoyl amide functionalized 5,5'-BIPHEP ligand was studied with regard of non-covalent binding of (*S*)-phenylalanine derivatives in solution. The binding was followed by NMR spectroscopy and observation of signal splitting due to formation of ($R_{\text{ax}}, S^{\text{Phe}}$) and ($S_{\text{ax}}, S^{\text{Phe}}$) adducts. The extent of splitting was found to be dependent on the substitution pattern of the phenylalanine analyte and additionally on its concentration. Diamides were plausibly found to bind more efficiently compared to amidoester derivatives. Introduction of electron rich aromatic substituents seemed to be of subaltern effect. The extent of signal splitting due to binding of the analytes was also

found to be dependent on its enantiomeric excess. This makes ee determination possible.

Our study represents the first example of non-covalent and non-coordinative binding of chiral analytes with a selector modified transition metal complex in solution. These findings open up the perspective of developing self-recognising stereodynamic systems applicable in catalysis and beyond.

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Notes and references

- H. Jdrzejewska, M. Wierzbicki, P. Cmoch, K. Rissanen and A. Szumna, *Angew. Chem., Int. Ed.*, 2014, **53**, 13760.
- D. Fiedler, D. H. Leung, R. G. Bergman and K. N. Raymond, *J. Am. Chem. Soc.*, 2004, **126**, 3674.
- L. A. Joyce, M. S. Maynor, J. M. Dregna, G. M. da Cruz, V. M. Lynch, J. W. Canary and E. V. Anslyn, *J. Am. Chem. Soc.*, 2011, **133**, 13746.
- Z. Dai, X. Xu and J. W. Canary, *Chirality*, 2005, **17**, S227.
- J. F. Folmer-Andersen, V. M. Lynch and E. V. Anslyn, *J. Am. Chem. Soc.*, 2005, **127**, 7986.
- P. Zhang and C. Wolf, *Chem. Commun.*, 2013, **49**, 7010.
- J. M. Brown, *Organometallics*, 2014, **33**, 5912.
- N. W. Alcock, J. M. Brown and P. J. Maddox, *J. Chem. Soc., Chem. Commun.*, 1986, 1532.
- K. Mikami, T. Korenaga, M. Terada, T. Ohkuma, T. Pham and R. Noyori, *Angew. Chem., Int. Ed.*, 1999, **38**, 495.
- K. Mikami, K. Aikawa and T. Korenaga, *Org. Lett.*, 2000, **3**, 243.
- T. Ohkuma, H. Doucet, T. Pham, K. Mikami, T. Korenaga, M. Terada and R. Noyori, *J. Am. Chem. Soc.*, 1998, **120**, 1086.
- K. Mikami and S. Matsukawa, *Nature*, 1997, **385**, 613.
- J. W. Faller, M. R. Mazzieri, J. T. Nguyen, J. Parr and M. Tokunaga, *Pure Appl. Chem.*, 1994, **66**, 1463.
- J. W. Faller and J. Parr, *J. Am. Chem. Soc.*, 1993, **115**, 804.
- K. Mikami, H. Kakuno and K. Aikawa, *Angew. Chem., Int. Ed.*, 2005, **44**, 7257.
- T. Punniyamurthy, M. Mayr, A. S. Dorofeev, C. J. R. Bataille, S. Gosiewska, B. Nguyen, A. R. Cowley and J. M. Brown, *Chem. Commun.*, 2008, 5092.
- K. Aikawa, Y. Takabayashi, S. Kawauchi and K. Mikami, *Chem. Commun.*, 2008, 5095.
- M. Schmitkamp, D. Chen, W. Leitner, J. Klankermayer and G. Francio, *Chem. Commun.*, 2007, 4012.
- D. Chen, M. Schmitkamp, G. Francio, J. Klankermayer and W. Leitner, *Angew. Chem., Int. Ed.*, 2008, **47**, 7339.
- (a) R. ter Halle, B. Colasson, E. Schulz, M. Spagnol and M. Lemaire, *Tetrahedron Lett.*, 2000, **41**, 643; (b) M. Berthod, C. Saluzzo, G. Mignani and M. Lemaire, *Tetrahedron: Asymmetry*, 2004, **15**, 639; (c) M. Berthod, G. Mignani and M. Lemaire, *J. Mol. Catal. A: Chem.*, 2005, **233**, 105.
- (a) T. Okano, H. Kumobayashi, S. Akutagawa, J. Kiji, H. Konishi, K. Fukuyama and Y. Shimano, *US Pat.*, 4 705 895, 1987; (b) Q.-h. Fan, C.-y. Ren, C.-h. Yeung, W.-h. Hu and A. S. C. Chan, *J. Am. Chem. Soc.*, 1999, **121**, 7407; (c) Q.-H. Fan, Y.-M. Chen, X.-M. Chen, D.-Z. Jiang, F. Xi and A. S. C. Chan, *Chem. Commun.*, 2000, 789.
- K. Aikawa, Y. Miyazaki and K. Mikami, *Bull. Chem. Soc. Jpn.*, 2012, **85**, 201.
- (a) F. Maier and O. Trapp, *Angew. Chem., Int. Ed.*, 2014, **53**, 8756; (b) F. Maier and O. Trapp, *Chirality*, 2013, **25**, 126; (c) G. Storch and O. Trapp, *Angew. Chem., Int. Ed.*, 2015, **54**, 3580.
- O. Diels and A. Bibergeil, *Ber. Dtsch. Chem. Ges.*, 1902, **35**, 302.
- CCDC 1415494, compound (6).
- J. Labuta, J. P. Hill, S. Ishihara, L. Hanyková and K. Ariga, *Acc. Chem. Res.*, 2015, **48**, 521.
- A. Shundo, J. Labuta, J. P. Hill, S. Ishihara and K. Ariga, *J. Am. Chem. Soc.*, 2009, **131**, 9494.
- J. Labuta, S. Ishihara, T. Šikorský, Z. Futera, A. Shundo, L. Hanyková, J. V. Burda, K. Ariga and J. P. Hill, *Nat. Commun.*, 2013, **4**, 2188.