Enantioselective Synthesis of Functionalized 1,5-Cyclononadienes by Intramolecular Cycloalkylation under α,α' -Diallyl Coupling**

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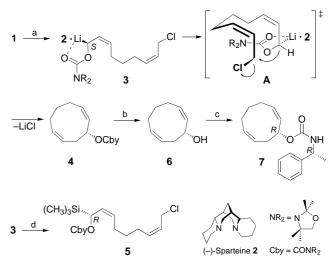
In spite of the biological importance of medium-size carbocycles, only a few ring-closing reactions are known for the synthesis of nine-membered monocarbocyclic rings.^[1] Classic examples are the acyloin condensation of nonanedioic acid dimethyl esters^[2a] and the McMurry reaction^[2b]; however, for reasonable yields, a high dilution of the substrate and a long reaction time are always necessary.^[2c]

When we applied our protocol for an intramolecular lithium—ene reaction^[3] on the (2Z,7Z)-9-chloro-2,7-nonadienyl carbamate **1**, we found a suprisingly efficient synthesis of a nine-membered carbocycle. The reaction proceeds to completion without any by-products within 2 h, even in 0.15 m solution. On the other hand, the (2E,7E)-isomer furnishes the expected cis-1,2-dialkenylcyclopentane **14**, as we recently reported (see Scheme 5).^[3] Compound **1** was synthesized from the known compound cis,cis-nona-2,7-diene-1,9-diol (Scheme 1).^[4]

Scheme 1. Synthesis of the dienyl carbamate 1. a) 0.3 equiv NaH, 0.3 equiv CbyCl, THF, Δ , 85%; b) 1.0 equiv *n*BuLi, 1.3 equiv MsCl, 1.0 equiv LiCl, THF, $-78\,^{\circ}\text{C} \rightarrow \text{RT}$, 73%. [23] Cby = 2,2,4,4-tetramethyl-1,3-oxazolidin-3-ylcarbonyl, Ms = methanesulfonyl.

Treatment of **1** with *n*-butyllithium/(-)-sparteine (**2**)^[5] in toluene at $-88\,^{\circ}$ C provided the (1R,2Z,7Z)-2,7-cyclononadienyl carbamate **4** in 73 % yield and with an *ee* value of 88 % (e.r. = 94:6; Scheme 2).^[6,7] The *S* configuration of the intermediate lithium complex **3** was confirmed by conversion into the silane (R)-**5** (87 % *ee*).^[8,9] Thus, the α,α' -coupling^[10] of the allyl moieties in **3** proceeds under inversion of configuration at the metal-bearing carbon atom and, therefore, presumably passes through the transition state **A**.^[11]

The cleavage^[12] of the carbamate group led to the (R)-cyclononadienol **6** and addition of this to (R)-1-phenylethyl



Scheme 2. Cycloalkylation of the dienyl carbamate **1** and synthesis of the silane **5**. a) 2.0 equiv nBuLi, 2.0 equiv **2**, toluene, $-88\,^{\circ}$ C, $73\,^{\circ}$, e.r. = 94:6 (88 % ee.); b) 1.) 2.0 equiv CH₃SO₃H, CH₃OH, Δ ; 2.) 4.0 equiv KOH, CH₃OH, Δ , 96 %; c) 3.0 equiv (R)-(+)-1-phenylethyl isocyanate, CH₂Cl₂, Δ , 63 %; d) 1.5 equiv nBuLi, 1.5 equiv (-)-sparteine, 2.7 equiv (CH₃)₃SiCl, toluene, $-78\,^{\circ}$ C, 61 %, 87 % $ee^{[23]}$

isocyanate furnished the carbamate 7. The X-ray analysis^[13] of 7 (Figure 1) clearly shows the relative configuration l and hence the R configuration for (+)-6 and (+)-4.

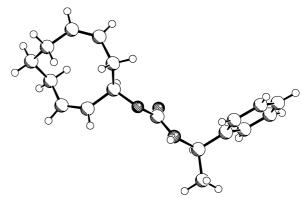


Figure 1. X-ray crystal structure analysis of 7.[13]

The analogous TBDPS-oxydienyl carbamate (R)-8 has been synthesized starting from (S)-epichlorohydrin (Scheme 3). Cyclization of (R)-8 (100% ee) by means of nBuLi/2 furnished the enantiomerically pure cyclononadiendiol derivatives (1R,5S)-9 and (1S,5S)-9, in a diastereomeric ratio of 94:6, which are easily separated by chromatography. The stereogenic center at C-5 affects neither the deprotonation nor the cyclization step; by use of the achiral base nBuLi/TMEDA, (1R,5S)-9 and (1S,5S)-9 were obtained in the ratio 53:47. Corresponding to this, rac-8 furnished, after treatment with nBuLi/2, the epimers (1R,5S)-9 and (1R,5S)-9 in 74% yield with a diastereomeric ratio of 55:45 and each with an ee value of 92% (Scheme 4). After deprotection to the alcohol (1R,5S)-10 and X-ray analysis, the relative configuration I was confirmed (Figure 2).

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^[+] X-ray analysis

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OCD

$$\begin{array}{c}
C & R^1 = H, R^2 = EE \\
C & R^1 = TBDPS, R^2 = EE \\
C & R^1 = TBDPS, R^2 = H
\end{array}$$
OTBDPS

OTBDPS

OTBDPS

OFF

Scheme 3. Synthesis of the enantiomericaly pure dienyl carbamate **8**. a) 1.) 1.0 equiv nBuLi, 0.5 equiv (S)-(+)-epichlorohydrin, 1.0 equiv BF $_3$ · OEt $_2$, THF, $-78\,^{\circ}\text{C} \rightarrow \text{RT}$; 2.) 1.0 equiv NaH, THF, $-10\,^{\circ}\text{C}$, 89%; b) 2.0 equiv HCCCH $_2$ OCb $_1^{[14a]}$ 2.0 equiv nBuLi, 2.0 equiv BF $_3$ · OEt $_2$, THF, $-78\,^{\circ}\text{C} \rightarrow \text{RT}$, 98%; $_1^{[14b]}$ c) 1.2 equiv TBDPSCl, 2.5 equiv imidazole, DMF, 83%; d) 0.07 mass% amberlyst 15, MeOH, 40°C, 95%; e) H $_2$, 5.0 mass% Lindlar catalyst, 0.15 equiv quinoline, MeOH, 96%; f) 1.0 equiv nBuLi, 1.2 equiv MsCl, 1.0 equiv LiCl, THF, $-78\,^{\circ}\text{C} \rightarrow \text{RT}$, 76%. $_1^{[23]}$ Cb = diisopropylaminocarbonyl, EE = ethoxyethyl, TBDPS = t tert-butyldiphenylsilyl.

TBDPSO, TBDPSO, TBDPSO, (R)-8
$$\xrightarrow{A}$$
 + \xrightarrow{B} (OCb (1R,5S)-9 (1S,5S)-9 (1S,5S)-9

Diamine = (-)-sparteine TMEDA 79% (d.r. = 94 : 6, each 100% ee) 70% (d.r. = 53 : 47, each 100% ee)

 rac -8 \xrightarrow{A} (1R,5S)-9 + (1R,5R)-9

Diamine = (-)-sparteine TMEDA 74% (d.r. = 55 : 45, each 92% ee) 71% (d.r. = 54 : 46)

Scheme 4. Cycloalkylation of the dienyl carbamate **8** and deprotection of the product **9**. a) 2.0 equiv nBuLi, 2.0 equiv diamine, toluene, $-88\,^{\circ}$ C, 2 h; b) 3.0 equiv tetrabutylammonium fluoride, THF, $74\,\%.^{[23]}$ TMEDA = N,N,N',N'-tetramethylethylenediamine.

(1R,5R)-10

92% ée

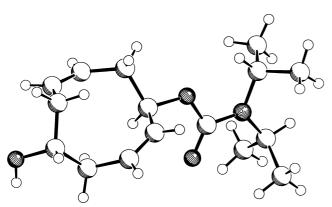


Figure 2. X-ray crystal structure analysis of (1R,5R)-10. [16]

The (Z,Z)-cyclonona-1,5-diene structure is destined for Cope rearrangements which pass through boatlike transition states:^[17] After conversion of **6** into the potassium alcoholate **11** an anionic oxy-Cope rearrangement^[18a] takes place at room temperature. The aldehyde (+)-13, obtained after protonation of the potassium enolate **12**, shows 95 % stereoselectivity for the rearrangement.^[18b] This seems to be the highest grade of chiral transmission in anionic oxy-Cope rearrangements of dienols which only contain a stereogenic center at the oxygenbearing carbon atom.^[19] The absolute configuration of **13** was determined by its synthesis from **14**^[3] through deprotection of the enol.^[20] Together with the *R* configuration of **6**, a reaction pathway via the transition state **B** with an equatorial oxido group^[19c] takes place (Scheme 5).

Scheme 5. Enantioselective anionic oxy-Cope rearrangement. a) 1.3 equiv KN(Si(CH₃)₃)₂, THF, RT, 75%; b) 15 equiv AliBu₂H, toluene, RT, 65%.

The method reported herein provides simple, flexible, and also enantioselective access to highly functionalized nine-membered carbocycles. The application of the homoenolate chemistry^[5a, 21] on the configuratively stable secondary lithium species generated by deprotonation of **4** bears additional possibilities in the synthesis of nine-membered ring compounds.^[22]

Experimental Section

Under argon, compounds **1** (500 mg, 1.52 mmol) and **2** (713 mg, 3.04 mmol) were dissolved in toluene (10 mL). After cooling to $-88\,^{\circ}$ C, a $1.6\,^{\circ}$ M solution of n-butyllithium in hexane (1.90 mL, 3.04 mmol) was slowly injected and the solution was stirred at this temperature for 2 h. Subsequently, CH₃OH (1 mL) and saturated aqueous NH₄Cl (1 mL) were added and the reaction mixture was allowed to warm to room temperature. Standard work up and purification by flash column chromatography over silica gel (Et₂O:pentane = 2:5) furnished **4** (324 mg, 73 %, 88 % ee) as a colorless solid (m.p. 72 °C).

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(1R, 5R) - 9

92% ee

^[1] The synthesis of nine-membered rings is particularly strongly disfavored because of the adverse entropic term, the Baeyer strain, and the Pitzer strain, as well as the transannular interactions: E. L. Eliel, S. H. Wilen in *Stereochemistry of Organic Compounds*, Wiley, New York, 1994, pp. 675–685.

^[2] a) A. T. Bloomquist, Y. C. Meinwald, J. Am. Chem. Soc. 1958, 80, 630-632; b) T. Lecta in Active Metals (Ed.: A. Fürstner), VCH, Weinheim, 1995, pp. 85-132; J. E. McMurry, K. L. Kees, J. Org. Chem. 1977, 42, 2655-2656; c) Other ring-closing reactions leading to monocarbocyclic nine-membered rings; intramolecular 1,4-addition (35% yield): P. Deslongchamps, B. L. Roy, Can. J. Chem. 1986, 64, 2068-2075; Thorpe-Ziegler condensation (3% yield): J. P. Schaefer, J. J. Bloomfield, Org. React. 1967, 15, 3-203.

- [3] A. Deiters, D. Hoppe, Angew. Chem. 1999, 111, 529-532; Angew. Chem. Int. Ed. 1999, 38, 546-548.
- [4] P. Lennon, M. Rosenblum, J. Am. Chem. Soc. 1983, 105, 1233-1241.
- [5] Reviews: a) D. Hoppe, T. Hense, Angew. Chem. 1997, 109, 2376 2410;
 Angew. Chem. Int. Ed. Engl. 1997, 36, 2282 2316; b) P. Beak, A. Basu,
 D. J. Gallagher, Y. S. Park, S. Thayumanavan, Acc. Chem. Res. 1996, 29, 552 560.
- [6] If the reaction is carried out at $-78\,^{\circ}$ C, **4** is obtained in 66% yield and 86% *ee.* By using the achiral base *n*BuLi/TMEDA in Et₂O at $-78\,^{\circ}$ C, *rac*-**4** is yielded in 76%.
- [7] Compound 4: Mp 72 °C; $[\alpha]_{10}^{30} = +14.8 \ (c = 0.59, \text{ CHCl}_3; 88\% \ ee);$ ¹H NMR (400 MHz, CDCl₃): $\delta = 1.34 1.56 \ (\text{m}, 14\text{ H}, \text{ CH}_2, \text{ CH}_3(\text{Cby})), 1.85 2.03 \ (\text{m}, 4\text{ H}, \text{ CH}_2), 2.22 \ \text{and} 2.43 \ (\text{both m}, \text{ each with } 1\text{ H}, \text{ CH}_2), 3.70 \ \text{and} 3.71 \ (\text{s}, 2\text{ H}, \text{ CH}_2(\text{Cby})), 5.10 \ (\text{dd}, 1\text{ H}, \text{ CHO}, {}^3J = 7.8, 7.8 \ \text{Hz}), 5.38 \ (\text{dt}, 1\text{ H}, \text{ CH}, {}^3J = 9.6, 9.6 \ \text{Hz}), 5.47 \ (\text{dt}, 1\text{ H}, \text{ CH}, {}^3J = 5.8, 10.1 \ \text{Hz}), 5.62 \ (\text{m}, 1\text{ H}, \text{ CH}), 5.79 \ (\text{dt}, 1\text{ H}, \text{ CH}, {}^3J = 10.1, 8.5 \ \text{Hz}).$ The enantiomeric ratio of 4 was determined by gas chromatography on a chiral stationary phase (Beta-Dex 120, Supelco, USA).
- [8] Compound **5**: $[a]_D^{50} = +10.3$ (c = 1.27, CHCl₃; 87% ee); ¹H NMR (300 MHz, CDCl₃): $\delta = 0.05$ (s, 9H, Si(CH₃)₃), 1.30–1.51 (m, 14H, CH₂, CH₃(Cby)), 1.96–2.30 (m, 4H, CH₂), 3.69 (s, 2H, CH₂(Cby)), 4.07 (dd, 2H, CH₂Cl, ⁴J = 1.8, ³J = 5.1 Hz), 5.35–5.42 (m, 3H, CH, CHSi), 5.58–5.62 (m, 2H, CH). The enantiomeric ratio of the silane **5** was determined by ¹H NMR shift experiments in the presence of [Eu(hfc)₃] (hfc = 3-(heptafluoropropylhydroxymethylene)-D-camphorate). After hydrogenation of the double bonds and hydrogenolytic cleavage of the chlorine atom, the corresponding 1-trimethylsilylnonyl carbamate was obtained. ^[3] Its enantiomer was synthesized by the sBuLi/(-)-sparteine method. ^[5a] A correlation of the optical rotations assigned the *R* configuration for (+)-**5**.
- [9] Stereoinversion in the silylation of allyllithium-(-)-sparteine complexes: K. Behrens, R. Fröhlich, O. Meyer, D. Hoppe, Eur. J. Org. Chem. 1998, 2397–2403; H. Paulsen, C. Graeve, D. Hoppe, Synthesis 1996, 141–144.
- [10] a) H. Yamamoto et al. reported intermolecular α,α'-allyl couplings by the use of allylbarium reagents.^[10b] However, an intramolecular coupling reaction was unknown: A. Yanagisawa, K. Yasue, H. Yamamoto, *Synlett* 1996, 842–844; A. Yanagisawa, H. Hibino, S. Habaue, Y. Hisada, H. Yamamoto, *J. Org. Chem.* 1992, 57, 6386–6387; b) A. Yanagisawa, H. Yamamoto in *Active Metals* (Ed.: A. Fürstner), VCH, Weinheim, 1995, pp. 61–84.
- [11] Presumably, the topology of the transition state A is influenced by π-π* interactions of the allylic moieties by precoordination and, therefore, favor the entropic term of the cyclization. In principle a γ,γ'-coupling between C-3 and C-7 is also possible and would lead to the cis configured five-membered ring. However, this would cause an increase of transannular and diaxial interactions of the side chains with the ring system; for this reason, the formation of a nine-membered ring takes place.
- [12] F. Hintze, D. Hoppe, Synthesis 1992, 1216-1218.
- [13] X-ray crystal structure analysis of 7: $C_{18}H_{23}NO_2$, $M_r=285.37$, light yellow crystal, $0.20\times0.10\times0.05$ mm, a=23.258(1), c=5.106(1) Å, $\gamma=120^\circ$, V=2392.0(5) Å³, $\rho_{calcd}=1.189$ g cm⁻³, $\mu=0.77$ cm⁻¹, absorption correction with SORTAV (0.985 $\leq T \leq 0.996$), Z=6, trigonal, space group $P3_1$ (No. 144), $\lambda=0.71073$ Å, T=198 K, ω and φ scans, 20 958 reflections collected ($\pm h, \pm k, \pm l$), $[(\sin\theta)/\lambda]=0.65$ Å⁻¹, 7097 independent ($R_{int}=0.056$) and 4457 observed reflections [$I\geq 2\sigma(I)$], 387 refined parameters, R=0.066, $wR^2=0.131$, max./min. residual electron density 0.42/-0.25 eÅ⁻³, Flack parameter -0.1(14), the asymmetric unit contains two independent, nearly identical molecules, hydrogens were calculated and refined as riding atoms. [16b]
- [14] a) For 2-alkynyl carbamates, see: S. Dreller, M. Dyrbusch, D. Hoppe, Synlett 1991, 397-400; b) For a similar reaction sequence, see: S. Hatakeyama, H. Irie, T. Shintani, Y. Noguchi, H. Yamada, M. Nishizawa, Tetrahedron 1994, 50, 13369-13376.
- [15] Compound (1*R*,5*S*)-**9**: $[\alpha]_D^{20} = -37.7$ (c = 0.43, CHCl₃, 100% ee);
 ¹H NMR (300 MHz, CDCl₃): $\delta = 1.07$ (s, 9H, tBu), 1.22 (d, 12H, CH₃(Cb), ${}^3J = 6.9$ Hz), 1.93 2.00 (m, 1H, CH₂), 2.11 (dd, 2H, CH₂, ${}^3J = 6.9$, 3.6 Hz), 2.19 2.38 (m, 2H, CH₂, CH₂), 2.53 (ddd, 1H, CH₂, ${}^3J = 8.7$, 9.0, ${}^2J = 12.9$ Hz), 3.80 4.03 (m, 3H, CH(Cb), CHOSi), 5.03 (dd, 1H, CHOCb, ${}^3J = 6.0$, 6.9 Hz), 5.35 (dt, 1H, CH, ${}^3J = 10.6$,

- 5.7 Hz), 5.67 5.81 (m, 3 H, CH, CH, CH), 7.32 7.70 (m, 10 H, CH(phenyl)). After cleavage of the TBDPS group, the enantiomeric ratio of (1R,5S)-10 was determined as described in reference [7].
- [16] a) X-ray crystal structure analysis of (1R,5R)-10: $C_{16}H_{27}NO_3$, M_r = 281.39, colorless crystal, $0.40 \times 0.25 \times 0.15$ mm, a = 26.159(5), b =10.884(2), c = 12.918(3) Å, $\beta = 113.71(2)^{\circ}$, V = 3367.5(12) Å³, $\rho_{\text{calcd}} =$ $1.110\,\mathrm{g}\,\mathrm{cm}^{-3}$, $\mu = 6.04\,\mathrm{cm}^{-1}$, absorption correction with ψ -scan data $(0.794 \le T \le 0.915)$, Z = 8, monoclinic, space group C2 (No. 5), $\lambda =$ 1.54178 Å, T = 223 K, $\omega/2\theta$ scans, 3697 reflections collected ($\pm h$, +k, $\pm l$), $[(\sin\theta)/\lambda] = 0.62 \text{ Å}^{-1}$, 3611 independent $(R_{\text{int}} = 0.077)$ and 3159 observed reflections $[I \ge 2\sigma(I)]$, 372 refined parameters, R = 0.044, $wR^2 = 0.119$, max./min. residual electron density 0.24/ - 0.17 e Å⁻³, Flack Parameter 0.3(2), the asymmetric unit contains two independent, nearly identical molecules, hydrogens were calculated and refined as riding atoms; b) The data sets were collected with Nonius CAD4 and Nonius KappaCCD diffractometers equipped with a Nonius FR590 sealed tube generator or a Nonius FR591 rotating anode generator. The following programs were used: For data collection, EXPRESS (Nonius B.V., 1994) and COLLECT (Nonius B.V., 1998); for data reduction, MolEN (K. Fair, Enraf-Nonius B.V., 1990) and Denzo-SMN (Z. Otwinowski, W. Minor, Methods Enzymol. 1997, 276, 307-326); for absorption corrections for CCD data, SORTAV (R. H. Blessing, Acta Crystallogr. Sect. A 1995, 51, 33-37; R. H. Blessing, J. Appl. Crystallogr. 1997, 30, 421-426); for structure solution, SHELXS-97 (G. M. Sheldrick, Acta Crystallogr. Sect. A 1990, 46, 467-473); for structure refinement, SHELXL-97 (G. M. Sheldrick, University of Göttingen, 1997); for graphics, SCHAKAL (E. Keller, University of Freiburg, 1997). Crystallographic data (excluding structure factors) for the structures reported in this paper have been deposited with the Cambridge Crystallographic Data Centre as supplementary publication no. CCDC-139019 and -139020. Copies of the data can be obtained free of charge on application to CCDC, 12 Union Road, Cambridge CB21EZ, UK (fax: (+44)1223-336-033; e-mail: deposit@ccdc.cam.ac.uk).
- [17] By heating 4 up to 220°C, a stereospecific Cope rearrangement takes place; [22] Cope rearrangement of (Z,Z)-cyclonona-1,5-diene: E. Vogel, W. Grimme, E. Dinné, Angew. Chem. 1963, 75, 1103.
- [18] a) Review of anionic oxy-Cope rearrangements: L. A. Paquette, Tetrahedron 1997, 53, 13 971 – 14020; b) anionic oxy-Cope rearrangement of rac-6: L. A. Paquette, G. D. Crouse, A. K. Sharma, J. Am. Chem. Soc. 1982, 104, 4411 – 4423. The enantiomeric ratio of 13 was determined as described in reference [7].
- [19] a) S.-Y. Wei, K. Tomooka, T. Nakai, Tetrahedron 1993, 49, 1025 1042;
 b) L. A. Paquette, G. D. Maynard, Angew. Chem. 1991, 103, 1392 1394; Angew. Chem. Int. Ed. Eng. 1991, 30, 1368 1370;
 c) in anionic oxy-Cope rearrangements the oxido group prefers an equatorial position: L. A. Paquette, G. Maynard, J. Am. Chem. Soc. 1992, 114, 5018 5027.
- [20] K. Tomooka, N. Komine, T. Sasaki, H. Shimizu, T. Nakai, *Tetrahedron Lett.* 1998, 39, 9715–9718.
- [21] Review of homoenolate chemistry: D. Hoppe, Angew. Chem. 1984, 96, 930–946; Angew. Chem. Int. Ed. Engl. 1984, 23, 932–948.
- [22] A. Deiters, D. Hoppe, unpublished results.
- [23] All new compounds were analytically pure (error in C,H,N elemental analyses ± 0.4).