Brief Articles

Structure-Activity Study on the in Vitro Antiprotozoal Activity of Glutathione Derivatives

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A series of N-, S-, and COOH-blocked glutathione derivatives were evaluated against the pathogenic parasites Trypanosoma brucei, Trypanosoma cruzi, and Leishmania donovani in vitro, to identify the determinants necessary for activity and for further development into an active lead structure. The results show that N,S-blocked glutathione diesters are the most effective inhibitors of *T. brucei* with structures **14–16** being the most active, **14** having an $IC_{50} \sim 1.9 \ \mu$ M. The toxicity effects observed for glutathione derivatives **12**, **14**, and **16** have been correlated to the K562 antileukemic activity of these compounds and their inhibitory effects on the glyoxalase system of the host. Diester compounds based on S-2,4-dinitrophenylglutathione (17-22) were found to be significantly better inhibitors of *T. brucei* with ED₅₀'s in the range 16–0.19 μ M. Compounds **19** and **20** were the two best inhibitors, with an ED₅₀ of \sim 1.07 and 0.19 μ M, respectively; however **20** displayed toxicity in parasitic assays. Monoesters, monoamides, and diamides tested generally exhibited low in vitro activity. The compounds did not inhibit glutathionylspermidine synthetase and trypanothione reductase enzyme targets in the unique trypanothione pathway of these parasites. Diester compounds per se were considered to be ineffective inhibitors of trypanothione metabolism suggesting that these compounds might act as prodrugs, being hydrolyzed in situ into a variety of glutathione derivatives which include combinations of monoesters, free acids, and amines, some of which are inhibitors of trypanothione metabolism.

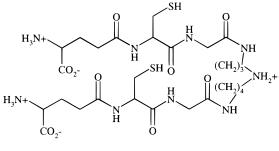
Introduction

The effects of trypanosomiasis and leishmaniasis are a serious problem to the health and economics of developing countries. Current chemotherapy is inadequate. One approach to the development of new drugs is to identify differences in metabolism between the host and pathogen and exploit these differences to design toxic agents. Trypanothione, like glutathione in the host, plays a central role in the metabolism of trypanosomatids acting both as a source of reducing equivalents and as a defense against oxidative stress.¹ Drugs effective against African trypanosomiasis^{1,2} in humans and animals are few and include suramin (in human and camels), pentamidine (in humans), and berenil (in cattle), the arsenicals melarsoprol (in humans) and cymelarsan (in cattle), and eflornithine (D,L- α -difluoromethylornithine (DFMO); in humans only). The ionic nature of suramin and the diamidines however limits their usefulness as they are not therapeutically active in the late CNS stage of this disease as they cannot cross the blood-brain barrier. In late-stage

sleeping sickness the only effective compounds are ones which create oxidative stress by altering trypanothione levels, ^{1,2} such as melarsoprol, which produces fatal side effects in ~5% of treated cases,³ or DFMO, an inhibitor of ornithine decarboxylase, an essential enzyme in spermidine synthesis and a precursor in trypanothione biosynthesis.

A key difference in the metabolism of the trypanosomatids^{1,4} is the replacement of the tripeptide glutathione (GSH) and its dependent enzymes in the host by an analogous family utilizing trypanothione (N^1, N^8) bis(glutathionyl)spermidine, T(SH)₂). Alternative chemotherapeutic strategies to find less toxic drugs include the inhibition of trypanothione reductase with trypanothione mimics⁵ and glutathionylspermidine synthetase (GSS) with glutathione-modified substrate analogues.⁶ Despite the successful inhibition of these enzymes in situ these compounds are inactive against trypanosomes in vitro.^{5,6} The inability of rational drug design studies to produce compounds active in vitro against parasitic cells led us to adopt a lead directed approach to identify potential antiparasitic compounds.⁷ We present here structure-activity results aimed at identifying the

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Trypanothione ((TSH)₂)

necessary determinants required of glutathione analogues to act as toxic agents against trypanosomes in vitro.

Chemistry

Glutathione ester derivatives 1-10 and 12-13 were prepared using known methodology published in the literature.^{9,11,12,21} Compound 11 was synthesized by the (dimethylamino)pyridine (DMAP) catalyzed acylation of 2 with benzyl chloroformate (CbzCl)⁹ in a manner similar to 10. Derivatives 14 and 16 were synthesized by the esterification of 9 and *N*,*S*-dibenzyloxycarbonylglutathione (mp lit.⁸ 105–107 °C) with their respective alcohols according to literature procedures.¹¹ The *N*acetyl derivative 15 was synthesized by DMAP acylation⁹ of *S*-benzyloxycarbonylglutathione dimethyl ester with acetic anhydride in a manner similar to 8.⁹ *S*-Benzyloxycarbonylglutathione dimethyl ester was prepared by cleavage of the Cbz group of 16 with HBr/ AcOH.¹¹

S-2,4-Dinitrophenylglutathione derivatives 17–19, 21 and 22 were synthesized by triethylamine (TEA) catalyzed N-acylation of a mixture of S-2,4-dinitrophenylglutathione mono- and diesters with CbzCl⁹ and chromatographic separation. The mixture of mono- and diesters was prepared by SOCl₂ esterification of S-2,4dinitrophenylglutathione in the appropriate alcohol as described for **3**;¹¹ however in this case monoesters¹¹ were not obtained as the exclusive product but a mixture of mono- and diesters was isolated. The monoesters were prepared for use in another study. Compound 20 was prepared in a manner similar to that described for **10**, 12, 14 and 16 by esterification of N-benzyloxycarbonyl-S-2,4-dinitrophenylglutathione with cyclohexanol according to literature procedures.¹¹ S-Bromobenzylglutathione and S-2,4-dinitrophenylglutathione derivatives were primarily chosen as the basis of these investigations due to their previously reported efficacy against a range of glutathione-dependent enzymes.^{12-15,22}

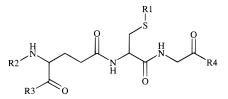
Results and Discussion

Glutathione derivatives 1-16 (see Table 1) were evaluated for antiparasitic activity against bloodstream form *Trypanosoma brucei* trypomastigotes and both *Leishmania donovani* and *Trypanosoma cruzi* amastigotes in mouse peritoneal macrophages. The inhibition of parasitic growth observed was expressed as a percentage of the untreated control at concentrations up to ~30 μ M (see Table 1). One of the 16 compounds evaluated, **2**, showed moderate activity against *T. brucei*, with another 3, the dibutyl esters of *N*-benzyloxycarbonyl-*S*-bromobenzylglutathione **14**, the dimethyl esters of *N*-acetyl-*S*-benzyloxycarbonylglutathione **15**, and *N*,*S*-dibenzyloxycarbonylglutathione **16** showing considerable activity with ED_{50} 's of 1.9, 3.23 and 17 μ M, respectively. Compounds **14** and **16** had activity against *L. donovani* and *T. cruzi*, but in both cases toxicity to host macrophages was also observed. Examination of the data obtained for compounds **1**–**16** (Table 1) with *T. brucei* indicates that diester compounds such as **14**– **16** generally show higher activity relative to diamides **13**, monoamides **11**, and monoesters **10**. The nature of the ester group also appears important if we compare the dimethyl **12** versus the dibutyl esters **14**, with the latter showing greater activity against *T. brucei* and low-grade activity against *L. donovani*.

The presence and nature of the N-blocking group appear important with respect to toxicity against amastigotes especially if we compare 7, 12, and 15. Replacement of the formyl group of 7 by an N-acetyl group (15) increases the activity of this compound against T. brucei while substitution with a benzyloxycarbonyl group (16) produces a lower degree of inhibition relative to 15 and toxicity effects. This result is not surprising in view of the inhibitory activity of the deesterified derivatives of 14-16 against enzymes of the mammalian glyoxalase system (GLI and GLII)^{10,14,15} which are in the order 14 > 16 > 15.^{10,15} Inhibition of the glyoxalase system by S-bromobenzylglutathione diester derivatives has been investigated as a strategy to antileukemic agents.¹⁸ The K562 antileukemic properties of diesters 1-16 (see Table 1) correlate well with the toxicity results observed for compounds 12, 14 and 16 in parasitic assays. This result prevents us from improving the efficacy of the compounds by extrapolating data obtained in the study of glutathione diester antitumor agents,¹⁸ due to their toxicity and general nonselectivity in their mode of action.

The nature of the S-blocking group appears important if we compare **12** with **16** as the latter compound with an S-benzyloxycarbonyl group has higher activity against T. brucei. However both the S-bromobenzyl and Sbenzyloxycarbonyl groups are good inhibitors of the glyoxalase system^{10,12-15,21} and so potentially have the ability to generate toxicity in parasitic assays depending on the nature of groups attached at the N-site. S-2,4-Dinitrophenylglutathione is a weak inhibitor of GLI (K_i) $\sim 0.77 \text{ mM}^{21}$) but a moderate inhibitor of glutathione reductase ($K_i^{50\%} \sim 30~\mu M^{22}$), an enzyme structurally related to trypanothione reductase, and therefore possibly a better starting material for the synthesis of antiparasitic compounds. The antiparasitic activities of diesters 17–21 (see Table 2) are significantly better especially in the cases of 19 and 20, with activities in the submicromolar range against *T. brucei*. The optimum length for the ester group appears to lie between a butyl group and a hexyl group. Precedence for the improvement in activity of butyl esters (19) has been reported for a series of alkyl ester prodrugs based on ibuprofren.²³ Compound **20**, the cyclohexyl diester, is one of the most active compounds as expected with an $ED_{50} \sim 0.21 \ \mu M$ and the most toxic based on the pharmacological responses of glutathione-based anticancer agents.¹⁸ However the cytotoxicity in this case may be associated with the nature of the cyclohexyl group and its metabolism, as the K562 antileukemic

Table 1. Biological Properties of Glutathione Derivatives against T. brucei, L. donovani, and T. cruzi



		compound			ED ₅₀ (µM) T. brucei ^a	% inhibition (µM)		ID ₅₀ (µM)
	R1	R2	R3	R4		L. donovani ^b	T. cruzi ^b	K562
1	BrBz	Н	OH	OH	>30 (40) ^a	na	na	>50
2	BrBz	Н	OH	NH_2	<30 (60) ^a	n.a	na	>50
3	BrBz	Н	OH	OMe	$>30 (20)^a$	na	na	>50
4	BrBz	Н	OMe	OMe	na	8.7	na	>50
5	BrBz	Н	NH_2	NH_2	na	4.0	na	>50
6	BrBz	HCO	OH	OH	na	na	11	>50
7	BrBz	HCO	OMe	OMe	na	na	12	>50
8	BrBz	CH ₃ CO	OH	OH	$>30 (40)^a$	na	na	>50
9	BrBz	Cbz	OH	OH	$>30 (40)^a$	na	na	>50
10	BrBz	Cbz	OH	OMe	na	na	na	>50
11	BrBz	Cbz	OH	NH_2	na	na	na	>50
12	BrBz	Cbz	OMe	OMe	$>30 (40)^a$	T/0	T/0	16.7
13	BrBz	Cbz	NH_2	NH_2	$>30(20)^{a}$	na	na	>50
14	BrBz	Cbz	OBu	OBu	1.9 ± 0.04	68	T/+	16.7
15	Cbz	CH ₃ CO	OMe	OMe	3.2 ± 2.4	na	na	>50
16	Cbz	Cbz	OMe	OMe	17 ± 0.66	T/100	T/100	1.26

^{*a*} The percent (%) inhibition, in parentheses, at 30 μ M is expressed as the percent (%) decrease in viable parasites relative to the control. ^{*b*} T/100, toxic to macrophages/100% inhibition of parasites; T/0, toxic to macrophages/0% inhibition of parasites; T/+, toxic to macrophages/inhibition of parasites; na, not active at ~30 μ M.

Table 2. Biological Properties of S-2,4-Dinitrophenylglutathione Derivatives against T. brucei, L. donovani, and T. cruzi

	compound					ED ₅₀ (µM)		
	R1	R2	R3	R4	T. brucei	L. donovani	T. cruzi	K562
17	2,4-DNP	Cbz	OMe	OMe	16.2 ± 0.64	> 30	na	23.4
18	2,4-DNP	Cbz	OPr	OPr	6 ± 0.16	>30	na	17.6
19	2,4-DNP	Cbz	OBu	OBu	1.07 ± 2.1	>30	na	21.8
20	2,4-DNP	Cbz	OcHx	OcHx	0.19 ± 0.04	T/+a	$3.0^a\pm 0.11$	15.9
21	2,4-DNP	Cbz	OHx	OHx	5.8 ± 0.32	>50	<50	>50
22	2,4-DNP	Cbz	OOct	OOct	na	>50	na	24.7

^{*a*} T/+, toxic to macrophages (~20 μ M)/inhibition of parasites; na, not active at ~30 μ M.

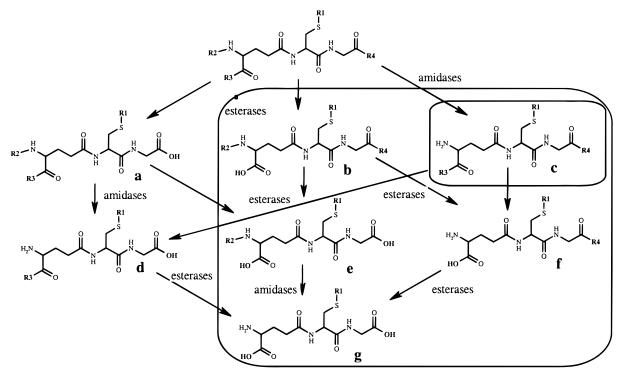
activity of **20** is not significantly different from that of **17–19** and **22**. Comparison of compound **14** with **19** indicates that the presence of the *S*-2,4-dinitrophenyl group enhances the *T. brucei* activity of **19** by \sim 20-fold relative to the *S*-bromobenzyl group.

Based on the current understanding of glutathione chemistry and the determinants required for glutathione¹¹⁻¹⁵ and trypanothione^{2,5,6} to bind to their respective enzymes, we speculate that compounds 14-**21** are ineffective as inhibitors of trypanothione² or glutathione enzymes.11-15 Diesters, per se, lack the minimum requirements for binding, the presence of a free α -Glu-COOH group.^{12,13} To be effective as inhibitors these diester compounds would have to undergo partial or complete de-esterification by nonspecific esterases on entry into the parasite. The N-blocked monoesters or diacids are discounted as inhibitors of the two enzymes responsible for trypanothione biosynthesis, glutathionylspermidine synthetase^{6,24} and trypanothione synthetase,²⁴ due to the requirement of the former for a glutathione derivative with a free α -amino group⁶ and the latter for an intact glutamyl residue. This leaves trypanothione reductase as the final candidate, known to bind derivatives of glutathione lacking a γ -Glu group or a free COOH group as in the case of N-benzyloxycarbonyl-L-cysteinylglycyl-3-methylaminopropylamide

disulfide.¹⁶ Compounds **1–3**, **8**, **9**, **12–16** and **19** were tested at a relatively high concentration of 100 μ M as inhibitors of trypanothione reductase using 5,5'-dithiobis{*N*-[3-(dimethylamino)propyl]-2-nitrobenzamide} as a colorimetric substrate as described by Davioud-Charvet.²⁵ No loss in trypanothione reductase activity was observed suggesting that the target of these compounds was another trypanothione dependent enzyme.

The mechanism of action of these compounds is consistent with the proposed mechanism reported for cancer cells treated with GSH esters where diesters were found to be the most effective compounds for delivery of GSH to cells due to their partial de-esterification to free acid and monoester in situ by nonspecific esterases, within 1 h of addition.¹⁷ These compounds (c; see Scheme 1) were considered prodrugs¹⁷⁻²⁰ functioning primarily by increasing the membrane penetration of glutathione derivatives prior to cleavage into three possible glutathione derivatives (d, f, g; Scheme 1), of which only two were observed¹⁷ (\mathbf{f} , \mathbf{g}), as expected. However in this case seven derivatives $(\mathbf{a}-\mathbf{g};$ Scheme 1) are possible based on the current understanding of detoxification mechanisms, which may account for the diverse range of therapeutic activities found with these compounds.²⁶ Of these potential derivatives only four or five types (b, e, f, g, and possibly c) are expected to

Scheme 1



be observed in reality depending on the nature of the groups R2, R3, and R4, located on the glutathione inhibitor (see Scheme 1). A preliminary study on four of these glutathione derivatives (b, e, f, g) linked to sensors^{27,28} tested against ammonium sulfate fractionated cell extracts of the insect trypanosome Crithidia fasciculata indicates that two of the four inhibitors (f and **e**) were strongly recognized by proteins present in the 40% and 60-100% (NH₄)₂SO₄ extracts, respectively. In both cases, up to 76% of the recognition could be reversed by the addition of inhibitor to the solution. The 40% extract contains proteins of high molecular weight associated with trypanothione biosynthesis (e.g. GSS and TSS²⁴) and the latter low molecular weight proteins (<30 kDa) associated with the trypanosomal peroxidase cycle (e.g. tryparedoxin peroxidase²⁹ and the tryparedoxins^{30,31}), while the 40-60% (NH₄)₂SO₄ extract contains the enzyme trypanothione reductase³² which showed no binding to **e**, consistent with enzymic studies. The recognition by these different classes of proteins for particular inhibitors is consistent with the known binding preferences of enzymes associated with trypanothione biosynthesis and to some extent tryparedoxins in view of the latter's relationship to glutaredoxin³¹ and ribonucleotide reductase which is known to be inhibited by glutathione derivatives³³ such as **e** or **g**. The recognition of **f** by a different type of proteins to **e** may explain the higher inhibitory activity of compound 15 containing an N-acetyl group than a benzyloxycarbonyl group (Cbz). The N-acetyl group is readily deacylated by amidases, and therefore this compound as a monoester could possibly inhibit a different parasitic target (e.g. GSS^{6,24} or TRS²⁴) to that inhibited by Cbz-protected glutathiones (10–14, 16) where de-esterification occurs more readily than deacylation and the resulting compounds resemble **b** or **e** (see Scheme 1).

Based on this reasoning the overall mode for the in vitro action of these compounds can be related to the balance of three factors: a, delivery of the compound into the cell; b, cleavage of the peptide by nonspecific esterases and amidases in the cytosol into several inhibitory forms; c, recognition by the target or targets, inhibition, and cell death.

The activity of compounds **15**, **17–19**, **21** and **22** clearly suggests that N,S-blocked GSH diester derivatives fulfill many of the above requirements for in vitro activity, and optimization of their structure to utilize the prodrug mechanisms proposed to generate the therapeutically most important active structure of those considered should eventually lead to a useful chemotherapeutic agent.

Conclusion

Glutathione derivatives have been shown for the first time to possess significant antiparasitic activity in vitro with only a few compounds showing selectivity against *T. brucei*. A limited structure–activity study indicates that N,S-blocked glutathione diesters are active against *T. brucei* parasites and that *N*-acetyl-*S*-benzyloxycarbonylglutathione dimethyl ester **15** and the *N*,*S*-benzyloxycarbonyl-*S*-2,4-dinitrophenylglutathione diester derivatives **17–19** and **21** represent lead structures possessing minimal toxicity which potentially could be developed further to yield a therapeutically active agent for the treatment of trypanosomiasis or leishmaniasis.

Experimental Section

GSH, DMAP and CbzCl were obtained from the Avocado Chemical Co. Ltd. (U.K.). Silica gel 60 F_{234} and alumina TLC plates were purchased from BDH Chemicals (Poole, Dorset, U.K.). NMR spectra were obtained on a JEOL 270 MHz FT-NMR spectrometer using TMS as an internal standard and FAB mass spectra on a VG Autospec instrument using cesium bombardment at 25kV in a matrix of 3-nitrobenzyl alcohol (NOBA). *S*-(4-Bromobenzyl)glutathione **(1)** was synthesized by the method of Vince et al.²¹ and *S*-2,4-dinitrophenylglutathione

by the method of Sokolovsky et al.⁸ Glutathione derivatives 2-10, 12, and 13 were synthesized by previously reported methods.¹¹

N-Benzyloxycarbonyl-*S*-(4-bromobenzyl)glutathione glycinamide (11) was prepared by N-acylation of 2 (0.5 g; 1.0 mmol) with the CbzCl/DMAP salt (2 equiv) according to literature procedures⁹ and recrystallized from CH₃CN to recover colorless crystals (0.24 g; 40%): mp 178–182 °C; ¹H NMR (D₆MSO) δ 8.34 (t, Gly-NH, J = 4.6), 8.2 (d, Cys-NH, J = 9.3), 7.5 (d, Ar-m-2H, J = 9.3), 7.35 (s, Ar-5H), 7.3 (d, Ar-o-2H, J = 9.3), 7.2 (s, NH), 5.0 (s, O-CH₂–Ar), 4.5 (m, CH), 2.5 (m, CH₂), 2.2 (t, 2H, γ -Glu-CH₂), 1.8–2.0 (m, 2H, γ -Glu-CH₂); FABMS m/z 609 ((M + H)⁺, 40), 633 ((M + Na)⁺, 32); HRFABMS calcd for C₂₅H₃₀N₄O₇SBr⁻¹/₂H₂O: C, 48.6; H, 4.7; N, 9.0. Found: C, 49.14; H, 4.41; N, 8.49.

N-Benzyloxycarbonyl-S-(4-bromobenzyl)glutathione di-n-butyl ester (14) was synthesized by SOCl₂ (0.093 g; 0.72 mmol) esterification of N-benzyloxycarbonyl-S-(4-bromobenzyl)glutathione (0.2 g; 0.33 mmol) (9) in n-butanol (15 mL) using literature procedures as described for 12.11 The solvent was removed by azeotropic vacuum distillation with water to give an oily residue which was titurated with Et₂O to afford colorless crystals (0.13 g; 56% yield): mp 110-115 °C; ¹H NMR (D₆MSO) δ 8.8 (m, glyNH), 8.2 (d, cysNH, J = 9.9), 7.8 (d, GluNH, J = 9.9), 7.6 (d, 2 × Ar-*m*-H, J = 9.1), 7.4 (m, Ar-5H), 7.3 (d, $2 \times \text{Ar-}o-H$, J = 9.1), 5.1 (s, CH₂Ar), 4.65 (m, GluCH), 4.1 (t, $2 \times CH_2$ -OCO, J = 7.2), 4.0 (d, glyCH₂, J = 6.5), 3.7 (s, S-CH₂), 3.6 (m, HN-CH-CH₂), 2.8 (dd, CH_a-S, J = 6.6, J = 0.3), 2.7 (dd, CH_b-S, J = 6.6, J = 0.3), 2.3 (t, CH₂CO, J = 5.7), 2.0 (m, GluCH₂CHCOO), 1.6 (m, $2 \times CH_2$ -CH₂-CH₃), 1.4 (m, $2 \times CH_2$ -CH₃), 1.0 (t, $2 \times$ -CH₂-CH₃, J = 9.6). Anal. Calcd for C33H44N3O8SBr·3/2H2O: C, 52.93; H, 6.28; N, 5.6. Found: C, 52.77; H, 6.47; N, 6.19.

N-Acetyl-*S*-(benzyloxycarbonyl)glutathione dimethyl ester (15) was synthesized by cleavage of the Cbz group from compound 16 (1.5 g; 2.5 mmol) with HBr/AcOH as described for 7,¹¹ followed by acetylation with acetic anhydride/DMAP (4 equiv) as described for 8,⁹ to afford colorless crystals (0.38 g), purified by PTLC (1:10; MeOH/CHCl₃) to afford an analytical sample (0.11 g; 8.7%): mp 193–195 °C; ¹H NMR (D₆MSO) δ 8.46 (t, Gly-NH, *J* = 5.1), 8.26 (d, 2 × NH, *J* = 7.1), 7.43 (s, Ar-5H), 5.28 (s, O-CH₂-Ar), 4.58 (m, CH), 4.26 (m, CH), 3.66 (s, 2 × OCH₃), 2.96 (m, CH₂), 2.26 (t, 2H, γ -Glu-CH₂), 1.88 (m,s, 5H, γ -Glu-CH₂, -COCH₃); FABMS *m*/*z* 512 ((M + H)⁺, 67), 534 ((M + Na)⁺, 100), 1023 ((2M + H)⁺, 3), 1045 ((2M + Na)⁺, 5). Anal. Calcd for C₂₂H₂₉N₃O₉S: C, 51.65; H, 5.72; N, 8.22. Found: C, 51.51; H, 5.69; N, 8.27.

N,S-Dibenzyloxycarbonylglutathione dimethyl ester (16) was synthesized by $SOCl_2$ (1.6 g; 12.5 mmol) esterification of *N,S*-dibenzyloxycarbonylglutathione⁸ (1 g; 1.7 mmol) in MeOH (20 mL) as described for 12^{11} to afford colorless crystals on recrystallization from methanol (0.7 g; 66.7% yield): mp 167–168 °C; ¹H NMR (CDCl₃) δ 7.35 (s, 2 × ArH), 5.25 (s, SCO_2 -CH₂-Ar), 5.1 (s, Ar-CH₂-S), 4.7 (m, CH), 4.0 (d, Ar -CH₂-S), 3.7 (s, 2 × OCH₃), 3.3 (m, CH), 2.1 (m, Gly-CH₂, γ -Glu-CH₂CH₂). Anal. Calcd for C₂₈H₃₃N₃O₁₀S: C, 55.71; H, 5.51; N, 6.96. Found: C, 55.7; H, 5.7; N, 6.9.

N-Benzyloxycarbonyl-*S*-(2,4-dinitrophenyl)glutathione Diesters. Procedure A. A mixture of *S*-(2,4-dinitrophenyl)glutathione mono- and diesters was prepared by SOCl₂ esterification of *S*-(2,4-dinitrophenyl)glutathione with the appropriate alcohol using literature procedures as described for **3**.¹¹ The crude mixture of mono- and diesters on evaporation of solvent was triturated with diethyl ether and the residue N-acylated with CbzCl/TEA (1.1 equiv) in MeOH containing a catalytic amount of DMAP. After stirring overnight at 55 °C, the solution was evaporated in vacuo and the residue triturated with Et₂O to remove the monoester. The residue left was extracted into AcOEt, washed water, dried (MgSO₄) and evaporated in vacuo and then retriturated with Et₂O to give the diester. The Et₂O layer containing the crude monoester was acidified to pH 4 with dilute hydrochloric acid (10%) and extracted into EtOAc, washed H_2O , dried (MgSO₄) and evaporated in vacuo to give the monoester for use elsewhere. An analytical sample of monosubstituted and disubstituted products were prepared by purification using preparative thinlayer chromatography (PTLC).

Procedure B. This was prepared by $SOCl_2$ (2.2 equiv) esterification of *N*-benzyloxycarbonyl-*S*-2,4-dinitrophenyl-glutathione in the appropriate alcohol as described for **12**.¹¹

N-Benzyloxycarbonyl-*S*-(2,4-dinitrophenyl)glutathione dimethyl ester (17) was prepared using procedure A from a mixture (1 g) of *S*-(2,4-dinitrophenyl)glutathione methyl mono- and diester and isolated by PTLC (MeOH/ CHCl₃; 1:3) as pale yellow crystals (0.78 g; 63% yield): mp 142–145 °C; ¹H NMR (D₆MSO) δ 9.0 (d, ArH₃), 8.7 (t, glyNH), 8.5 (d, cysNH), 8.4 (dd, ArH₅), 8.0 (d, ArH₆), 7.7 (d, GluNH), 7.3–7.4 (m, Ar-5H), 5.1 (s, CH₂Ar), 4.65 (td, GluCH), 4.2 (td, HN–CH-CH₂), 3.95 (d, glyCH₂), 3.7 (dd, CH₂–S), 3.6 (s, 2 × CH₃OCO), 2.25 (t, CH₂CO), 2.0 (td, GluCH₂CHCOO); CIMS *m*/*z* 653 ((M + NH₄)⁺, 20), 636 ((MH)⁺, 35), 436 ((M – C₆H₃N₂O₄S), 100), 502 ((M – C₈H₇O₂), 50); HRFABMS calcd for C₂₆H₃₀N₅O₁₂S 63.3802, found 636.3802. Anal. Calcd for C₂₆H₂₉N₅O₁₂S: C, 49.13; H, 4.60; N, 11.02. Found: C, 49.54; H, 4.57; N, 10.56.

N-Benzyloxycarbonyl-S-(2,4-dinitrophenyl)glutathione dipropyl ester (18) was prepared using procedure A from a mixture (1 g) of S-(2,4-dinitrophenyl)glutathione npropyl mono- and diester and isolated by PTLC (MeOH/CHCl₃; 3:17) as pale yellow crystals (0.83 g; 64% yield): mp 110 °C; ¹H NMR (D_6MSO) δ 9.0 (d, ArH₃), 8.7 (t, glyNH), 8.5 (d, cysNH), 8.4 (dd, ArH₅), 8.0 (d, ArH₆), 7.7 (d, GluNH), 7.3-7.4 (m, Ar-5H), 5.1 (s, CH_2Ar), 4.65 (td, GluCH), 4.1 (q, $2 \times CH_3$ -CH₂-CH₂), 4.0 (td, HN-CH-CH₂), 3.95 (d, glyCH₂), 3.7(dd, CH₂-S), 2.25 (t, CH₂CO), 1.6 (q, $2 \times CH_3$ -CH₂-CH₂), 2.0 (td, GluCH₂CHCOO), 1.0 (t, $2 \times CH_3$ -CH₂); CIMS m/z 709 ((M - NH_4)⁺, 30), 692 ((MH)⁺, 50), 493 ((MH)⁺ - C₆H₃N₂O₄S), 80), $664 ((M - C_2H_4), 40), 464 (M - (C_2H_4 + C_6H_3N_2O_4S), 100);$ HRFABMS calcd for $C_{30}H_{38}N_5O_{12}S$ 692.2237, found 692.2234. Anal. Calcd for C₃₀H₃₇N₅O₁₂S (*M*_r 691.2159): C, 52.1; H, 5.35; N, 10.13. Found: C, 52.04, H, 5.33; N, 10.25.

N-Benzyloxycarbonyl-*S*-(2,4-dinitrophenyl)glutathione dibutyl ester (19) was prepared using procedure A from a mixture (0.5 g) of *S*-(2,4-dinitrophenyl)glutathione *n*-butyl mono- and diester and isolated by PTLC (EtOAc/ CHCl₃; 1:1) as pale yellow crystals (0.179 g; 28% yield): mp 115 °C; ¹H NMR (D₆MSO) δ 9.0 (d, ArH₃), 8.7 (t, glyNH), 8.5 (d, cysNH), 8.4 (dd, ArH₅), 8.0 (d, ArH₆), 7.7 (d, GluNH), 7.3– 7.4 (m, Ar-5H), 5.1 (s, CH₂Ar), 4.65 (td, GluCH), 4.1 (t, 2 × CH₂-OCO), 4.0 (td, HN-CH-CH₂), 3.95 (d, glyCH₂), 3.7 (dd, CH₂-S), 2.25 (t, CH₂CO), 2 (td, GluCH₂CHCOO), 1.6 (m, 2 × CH₃-CH₂-CH₂), 1.15 (m, 2 × CH₂-CH₃), 0.95 (t, 2 × CH₃-CH₂); FABMS *m*/*z* 720 ((M + H)⁺, 61), 742 ((M + Na)⁺, 80). Anal. Calcd for C₃₂H₄IN₅O₁₂S·¹/₂H₂O: C, 52.72; H, 5.76; N, 9.61. Found: C, 52.54; H, 5.95; N, 10.25.

N-Benzyloxycarbonyl-S-(2,4-dinitrophenyl)glutathione dicyclohexyl ester (20) was prepared using procedure B from N-benzyloxycarbonyl-S-(2,4-dinitrophenyl)glutathione (0.5 g, 0.82 mmol) in cyclohexanol (25 mL). Azeotropic removal of the solvent with water gave an oily residue which on trituration with Et₂O gave a hygroscopic yellow solid which was purified by PTLC (MeOH/CHCl₃; 1:9) (0.4 g; 63% yield): mp 135-137 °C; ¹H NMR (D₆MSO) δ 8.9 (s, ArH₃), 8.6 (t, glyNH), 8.4 (m, cysNH & ArH₅), 8.0 (d, ArH₆), 7.7 (d, GluNH), 7.3–7.4 (m, Ar-5H), 5.1 (s, CH₂Ar), 4.6 (m, GluCH & 2 \times CH_{cHx}), 4.0 (td, HN-CH-CH₂), 3.95 (d, glyCH₂), 3.7 (dd, CH₂-S), 2.3 (t, CH₂CO), 2.0 (m, GluCH₂CHCOO), 1.4 (m, 2 × (CH₂-CH–CH₂)), 1.2 (m, 2 × (CH₂)_{3 cHx}); FABMS m/z 772 (M + H)⁺, 794 (M + Na)⁺; HRFABMS calcd for $C_{36}H_{46}N_5O_{12}S$ 772.2863, found 772.2827. Anal. Calcd for $C_{36}H_{45}N_5O_{12}S \cdot 2H_2O$: C, 53.52; H, 6.11; N, 8.67. Found: C, 53.23; H, 5.52; N, 8.68.

N-Benzyloxycarbonyl-*S*-(2,4-dinitrophenyl)glutathione dihexyl ester (21) was prepared using procedure A from a mixture (0.5 g, 0.89 mmol) of *S*-(2,4-dinitrophenyl)glutathione *n*-hexyl mono- and diester and isolated by PTLC (MeOH/CHCl₃; 1:9) as pale yellow solid (0.56 g; 94% yield): mp 130–132 °C; ¹H NMR (D₆MSO) δ 9.0 (d, ArH₃), 8.7 (t, glyNH), 8.5 (d, cysNH), 8.4 (dd, ArH₅), 8.0 (d, ArH₆), 7.7 (d, GluNH), 7.3–7.4 (m, Ar), 5.1 (s, CH₂Ar), 4.65 (td, GluCH), 4.1 (t, 2 × CH₂OCO), 4.0 (td, HN–CH–CH₂), 3.95 (d, glyCH₂), 3.7 (dd, CH₂–S), 2.3 (t, CH₂CO), 2.0 (td, GluCH₂CHCOO), 1.6 (t, 2 × CH₂–CH₂–OCO), 1.3 (m, 2 × (CH₂)₃), 0.9 (t, 2 × CH₃–CH₂); FABMS *m*/*z* 776 ((M + H)⁺, 75), 798 ((M + Na)⁺, 100); HRFABMS calcd for C₃₆H₅₀N₅O₁₂S 776.3176, found 776.3664. Anal. Calcd for C₃₆H₄₉N₅O₁₂S: C, 55.72; H, 6.37; N, 9.03. Found: C, 55.75; H, 6.58; N, 9.00.

N-Benzyloxycarbonyl-*S*-(2,4-dinitrophenyl)glutathione dioctyl ester (22) was prepared using procedure A from a mixture (0.5 g, 0.81 mmol) of *S*-(2,4-dinitrophenyl)glutathione *n*-octyl mono- and diester and isolated by PTLC (MeOH/CHCl₃; 1:9) as a pale yellow solid (0.57 g; 82% yield): mp 144 °C; ¹H NMR (D₆MSO) δ 9.0 (d, ArH₃), 8.7 (t, glyNH), 8.5 (d, cysNH), 8.4 (dd, ArH₅), 8.0 (d, ArH₆), 7.7 (d, GluNH), 7.3–7.4 (m, ArH), 5.1 (s, CH₂Ar), 4.65 (td, GluCH), 4.1 (t, 2 × CH₂OCO), 4.0 (td, HN–CH–CH₂), 3.95 (d, glyCH₂), 3.7 (dd, CH₂–S), 2.3 (t, CH₂CO), 2.0 (td, GluCH₂CHCOO), 1.6 (t, 2 × CH₂–CH₂–OCO), 1.3 (m, 2 × (CH₂)₅), 0.9 (t, 2 × CH₃–CH₂); FABMS *m*/*z* 832 ((M + H)⁺, 60), 854 ((M + Na)⁺, 100). Anal. Calcd for C₄₀H₅₇N₅O₁₂S: C, 57.75; H, 6.91; N, 8.42. Found: C, 57.29; H, 7.10; N, 8.27.

Evaluation of the Parasitic Activity of Glutathione Derivatives in Vitro. Parasites: *T. brucei (*strain S427) bloodstream form trypomastigotes were maintained in HMI-18 medium^{34,35} with 20% heat inactivated foetal calf serum (HIFCS) (Harlan Sera-Lab, Crawley, U.K.) at 37 °C in 5% CO_2 -air mixture. *L. donovani* (strain MHOM/ET/67/L82) was maintained routinely in special pathogen free (SPF) female golden hamsters by serial passage every 6–8 weeks. *T. cruzi* (strain MHOM/BR/OO/Y) trypomastigotes were derived from MDCK fibroblasts in Dulbecco's modified Eagle medium (Life Technologies Ltd., Paisley, Scotland) with 10% HICFS at 37 °C in a 5% CO₂-air mixture.

In vitro assays: *T. brucei:* All compounds were tested in triplicate in a 3-fold dilution series from a top concentration of 30 μ M. Parasites were diluted to 2 × 10⁵/mL and added in equal volumes to the test compounds in 96-well, flat bottom Microtest III tissue culture plates (Becton Dickinson and Co., NJ). Appropriate controls with pentamidine isethionate (Rhone-Poulenc-Rorer) as the positive were set up in parallel. Plates were maintained for 3 days at 37 °C in a 5% CO₂—air mixture. Compound activity was determined by the use of a tetrazolium salt colorimetric assay³⁷ on day 3.

L. donovani and T. cruzi: Peritoneal macrophages were harvested from female CD1 mice (Charles Rivers Ltd., Margate, U.K.) by peritoneal lavage 24 h after starch (Merck Ltd., Leics, U.K.) induced recruitment. After washing cells were dispensed into 16-well Lab-tek tissue culture slides (Nunc Inc., IL) at 4 \times 10⁴/well in a volume of 100 μ L of RPMI-1640 medium (Sigma-Aldrich Co. Ltd., Dorset, U.K.) and 10% HIFCS. After 24 h, macrophages were infected at a ratio of 10:1 (4 \times 10⁵/well) with *L. donovani* amastigotes or 5:1 (2 \times 10⁵/well) with *T. cruzi* trypomastigotes. Infected macrophages were then maintained in the presence of drug in a 3-fold dilution series in quadruplicate for 5 days in the case of L. donovani and 3 days with T. cruzi. Drug activity was evaluated from the percentages of macrophages cleared of amastigotes in treated cultures. Sodium stibogluconate (NaSb^v) (Glaxo-Wellcome, Dartford, U.K.) and nifurtimox (Bayer, U.K.) were used as the respective controls.³⁶

Evaluation of the Cytotoxicity of Glutathione Derivatives in Vitro. Cytotoxicity testing on K562 (human, chronic myelogenous leukemia) was undertaken independently on our behalf by the Christie Patterson Institute for Cancer Research, Manchester. The cytotoxicity of drugs were evaluated using the semi automated MMT assay developed by the National Cancer Institute (NCI), Maryland, based on the original work of Mosmann.³⁷ Cell cultures were grown in RPMI medium with 10% calf fetal serum (CFS) at 37 °C in a 5% CO₂-air mixture in a humidified incubator. Plates were incubated with compound for 5 days at 37 °C in a 5% CO₂-humidified air mixture prior to determination of activity by the use of a tetrazolium salt colorimetric $assay^{35,37}$ on day 5.

Trypanothione Reductase: Isolation and Assay. Trypanothione reductase was isolated as previously described³² from Crithidia fasciculata grown on Schneider's insect medium (Sigma-Aldrich Co. Ltd., Dorset, U.K.), containing 0.0002% streptomycin sulfate, FMN (2.4 mg/L) and 10% heat inactivated CFS (Sigma-Aldrich Co. Ltd., Dorset, U.K.). All enzymatic and nonenzymatic reactions were conducted in a flatbottomed 96-well microtiter plates (Sterilin) in a total volume of 100 μ L. The plates on incubation at room temperature were then read using a 410 nm filter in a Dynatech MR5000 multiscan microplate reader connected to a Star Multipart LC-10 printer. Inhibition studies were undertaken on compounds 1-3, 8, 9 and 12-16 using a one-spot enzyme inhibition assay as previously described²⁵ with 5,5'-dithiobis{N-[3-(dimethylamino)propyl]-2-nitrobenzamide}·HCl salt²⁵ as the colorimetric substrate. The reaction was initialized by the addition of 80 μ L of enzyme solution containing 42 \times 10⁻⁴ U of trypanothione reductase and the reaction stopped after 10 min by the addition of 20 µL of acetonitrile. Measurements were undertaken against suitable controls, with and without enzyme, NADPH, or substrate.

5,5'-Dithiobis{N-[3-(dimethylamino)propyl]-2-nitrobenzamide}·HCl salt was prepared in a manner similar to the literature²⁵ and obtained as a pale yellow amorphous powder: ESIMS m/z 565 ([MH]⁺, 15), 283 (100); HRESIMS calcd for C₂₄H₃₃N₆O₆S₂ [MH]⁺ 565.1903, found 565.1905.

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