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COMMUNICATION

Phosphine-catalyzed intramolecular γ -umpolung addition of α -aminoalkylallenic esters: facile synthesis of 3-carbethoxy-2-alkyl-3-pyrrolines[†][‡]

Ian P. Andrews, Brian R. Blank and Ohyun Kwon*

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An array of *N*-tosylated α -aminoalkylallenic esters was prepared and their cyclization under the influence of nucleophilic phosphine catalysts was explored. The α -aminoalkylallenic esters were prepared through aza-Baylis–Hillman reactions or novel DABCO-mediated decarboxylative rearrangements of allenylic carbamates. Conversion of these substrates to 3-carbethoxy-2-alkyl-3-pyrrolines was facilitated through Ph₃P-catalyzed intramolecular γ -umpolung addition.

Nitrogen-containing heterocycles are ubiquitous structural elements in the realm of natural products and pharmaceutically relevant compounds.¹ Amongst these important scaffolds, 3-pyrrolines occupy a uniquely versatile position in that they can be further transformed into their fully saturated pyrrolidines or more highly oxidized pyrrole and pyrrolidinone counterparts.² Of the various approaches toward functionalized 2,5-dihydropyrroles, Lu's phosphine-catalyzed [3+2] annulation of allenes and imines has emerged as one of the premiere methodologies.^{2b,3} Since its first report, this annulation has undergone many developments, including modifications,⁴ applications,⁵ and asymmetric renditions.⁶ Despite the great utility of this phosphine-catalyzed allene-imine [3+2] annulation, it has its limitations—the most evident being the necessity for imines devoid of α -protons. This prerequisite has limited the scope of the reaction to the use of aryl-substituted imines. Overcoming this limitation would expand the potential use of organocatalyzed 3-pyrroline synthesis to applications requiring non-aryl groups substituted onto the heterocycle.7 The successful implementation of alkyl-substituted N-sulfonyl imines was reported only recently: used in conjunction with a Me₃P-catalyzed isomerization of 3-alkynoates or when using diphenylphosphinoyl imines along with a highly nucleophilic peptide-based chiral phosphine.^{8,9} The utility of the resultant 3-pyrrolines was demonstrated in the formal syntheses of (\pm) -allosecurenine and (+)-trachelanthamidine, respectively.^{8,9} The development of new methods for forming these types of heterocycles would be a boon to the synthetic community.



To test the proposed cycloisomerization, we required a method for synthesizing the necessary α -aminoalkylallenoates. For this purpose, we turned to a known tertiary amine-catalyzed rearrangement of allylic N-tosyl carbamates and tested it on the similar, but never before utilized, allenylic N-tosyl carbamates 3.12,13 We were pleased to discover that treatment of the allenvlic carbamates with a nucleophilic catalyst facilitated the desired rearrangement. The reaction was best facilitated by slowly adding the allenvlic carbamate to a solution of one equivalent of DABCO (Table 1).¹³ Interestingly, we isolated the allenoate **1a** in the highest yield (56%) when using dimethyl sulfide as the catalyst in MeCN.¹⁴ Dimethyl sulfide failed to provide any desired rearrangement products from any of our other tested substrates. The DABCO-mediated reaction was tolerant of various alkyl groups, providing the desired products in modest yields. The best result was realized when a methyl substituent was present, providing the desired allenoate 1b in 65% yield (entry 2). Extending the chain decreased the efficiency (entries 3 and 4). The reaction also proceeded smoothly when



Scheme 1 Phosphine-catalyzed intramolecular γ-umpolung addition.

⁶⁰⁷ Charles E. Young Drive East, Los Angeles, USA.

E-mail: ohyun@chem.ucla.edu; Tel: 310-206-9648

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Table 1 Decarboxylative rearrangements of allenylic N-tosyl carbamates

		DABCO (1 equiv)	Ts
	3 OEt	inverse addition over 12 hours 1 OE	t
Entry	R	Product	Yield ^a (%)
1	Н	1 a	56 ^b
2	methyl	1b	65
3	ethyl	1c	56
4	n-pentyl	1d	27
5	i-propyl	1e	52
6	cyclopropyl	1f	45
7	cyclopentyl	1g	45
8	cyclohexyl	1h	51
9	phenyl	1i	0
^{<i>a</i>} Isolated y	vield after column	chromatography. ^b Dimetl	nyl sulfide was

bulkier substituents, namely isopropyl and cycloalkyl groups, were present, albeit in slightly lower yields (entries 5–8). Notably, when an aryl substituent was present, the reaction yielded none of the desired product (entry 9); we could, however, synthesize those allenoates **1** featuring an aryl group as the R unit through aza-Baylis–Hillman reactions between ethyl 2,3-butadienoate and aryl imines.^{13,15}

With the requisite α -aminoalkylallenic esters in hand, we explored their reactivity under nucleophilic catalysis (Table 2). We were pleased to find that addition of 20 mol% of Ph₃P to the allenoate **1i** in CH₂Cl₂ at room temperature provided the dihydropyrrole 2i in 62% yield (entry 1). Screening of various solvents (entries 1-5) revealed that benzene was the most efficient reaction medium in terms of the product yield and reaction time, providing the dihydropyrrole in 93% isolated yield after 18 h (entry 5). Use of the more-nucleophilic n-Bu₃P markedly decreased the reaction time, but led to a dramatic loss in product yield (entry 6). Next, we extended the reaction to β' -alkyl–substituted α -aminoalkylallenic esters in an attempt to access 2-alkyl-substituted dihydropyrroles. In a very gratifying initial trial, we transformed the ethyl-substituted allenic ester into the desired 2,5-dihydropyrrole in 92% yield (entry 7). In salient contrast to the reaction of the phenyl-substituted derivative 1i, the β' -ethyl–substituted allenic ester **1c** required a significantly

Table 2 Optimization of the phosphine-catalyzed cycloisomerization

			HTS Ph ₃ P (20 mo	1%)				
1 OEt Conditions NR 1 OEt Ts 2								
Entry	R	Solvent	Additives	Time	Product	$\operatorname{Yield}^{a}(\%)$		
1	Ph	CH_2Cl_2	none	40 h	2i	62		
2	Ph	PhMe	none	40 h	2i	89		
3	Ph	THF	none	40 h	2i	0^b		
4	Ph	MeCN	none	40 h	2i	0		
5	Ph	PhH	none	18 h	2i	93		
6 ^{<i>c</i>}	Ph	PhH	none	1 h	2i	29		
7	Et	PhH	none	52 h	2c	92		
8	Сур	PhH	none	10 days	1g	44		
9	Et	PhH	AcOH/NaOAc	32 h	2c	97		
10	Сур	PhH	AcOH/NaOAc	4 days	1g	85		

^{*a*} Isolated yield after silica gel chromatography. ^{*b*} Recovered starting material. ^{*c*} Bu₃P was used.

prolonged reaction time (52 h, *cf.* 18 h). Increasing the steric bulk of the alkyl group generally decreased the yield and increased the reaction time. The substrate bearing a cyclopentyl group required a reaction for 10 days at 40 °C to reach completion, resulting in an isolated yield of only 44% for the desired pyrroline (entry 8). We made several efforts to minimize the reaction time and improve the yields of the alkyl-substituted substrates. Based on reports of added Brønsted acid/base pairs increasing reaction efficiencies for various phosphine-catalyzed reactions, we screened several additives for their effects. To our delight, the addition of sodium acetate and acetic acid (0.5 equivalents each) shortened the reaction times and generally improved the yields (entries 9 and 10).^{110,16}

Under the optimized conditions, we converted a number of β' -alkyl– and β' -aryl–substituted α -aminoalkylallenic esters 1 to their corresponding dihydropyrroles 2 (Table 3). The yields for these cycloisomerizations were generally very high. The allenoates 1a and 1j underwent cyclizations in higher yields in the absence of any additives (entries 1 and 10). All of the other alkyl-substituted allenoates benefited from the combination of catalytic Ph₃P and Brønsted acid/base, cyclizing in moderate to excellent yields. The allenoates bearing short straight-chain alkyl units, namely methyl and ethyl groups, cyclized in 99 and 97% yields, respectively (entries 2 and 3). The presence of a longer n-pentyl group resulted in lower reaction efficiency, with the allenoate cyclizing in 69% yield (entry 4). Branched alkyl groups were also tolerated well, with the isopropyl-, cyclopropyl-, cyclopentyl-, and cyclohexyl-substituted allenoates cyclizing in 97, 95, 85, and 93% yields, respectively (entries 5-8). Aryl-substituted a-aminoalkylallenic esters also cyclized in high yields (entries 9-12), with the more-electronegative 4-chlorophenyland 4-cyanophenyl-substituted allenoates resulting in slightly diminished product yields of 88 and 85%, respectively (entries 10 and 12) relative to the more-electron-rich phenyland 4-methoxyphenyl-substituted allenoates, which both cyclized in 99% yields (entries 9 and 11).

Fig. 1 displays our proposed mechanism for the cycloisomerization. Addition of Ph_3P to the α -aminoalkylallenic ester 1 leads to the formation of the phosphonium dienolate 4. Conversion to

		Ph ₃ P (20 mol%)				
	1 OEt HOAc/N	laOAc (50 mol % e benzene	ach) N ts	R 2		
Entry	R	Time (h)	Product	Yield (%)		
1	Н	18	2a	87 ^a		
2	Me	19	2b	99		
3	Et	32	2c	97		
4	n-Pent	48	2d	69		
5	<i>i</i> -Pr	77	2e	97		
6	cyclopropyl	72	2f	95		
7	cyclopentyl	96	2g	85		
8	cyclohexyl	96	2h	93^b		
9	phenyl	12	2i	99		
10	4-chlorophenyl	4.5	2j	88^a		
11	4-methoxyphenyl	22	2k	99		
12	4-cyanophenyl	2	21	85		

 Table 3 Phosphine-catalyzed intramolecular γ-umpolung additions

^{*a*} Yield in the absence of any additives. ^{*b*} Based on recovered starting material (69% isolated yield).



Fig. 1 Mechanism for the intramolecular γ -umpolung addition.

the sulfonamide anion **5** via proton transfer, followed by 5-endo cyclization, yields the phosphorous ylide **6**. The well-established equilibration between **6** and **7**, followed by β -elimination of the phosphine catalyst, furnishes the final product **2**.

In conclusion, we have developed a method for the highyield formation of 3-carbethoxy-2-alkyl-3-pyrrolines by way of phosphine-catalyzed cycloisomerizations of α -aminoalkylallenic esters. The necessary substrates can be prepared in moderate yields: for aryl-substituted allenoic esters, through aza-Baylis–Hillman reactions; for alkyl-substituted allenoic esters, through novel decarboxylative rearrangements of allenylic carbamates catalyzed by a tertiary amine. This paper illustrates the first example of a 5-endo cyclization occurring via intramolecular γ -umpolung addition to an activated allene.

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Notes and references

- Nitrogen-containing heterocycles are present in six of the ten bestselling pharmaceuticals of 2009 (as ranked by worldwide sales): Lipitor[®] (Pfizer), Zyprexa[®] (Lilly), Crestor[®] (AstraZeneca), Seroquel[®] (AstraZeneca), Nexium[®] (AstraZeneca), and Plavix[®] (Bristol-Meyers Squibb). See: D. J. Mack, M. Brichacek, A. Plichta and J. T. Njardarson. Top 200 Pharmaceutical Products by Worldwide Sales in 2009. http://cbc.arizona.edu/njardarson/ group/top-pharmaceuticals-poster (accessed in February 2011).
- C. M. Huwe and S. Blechert, *Tetrahedron Lett.*, 1994, **35**, 7099;
 (b) Z. Xu and X. Lu, *J. Org. Chem.*, 1998, **63**, 5031;
 (c) M. P. Green,
 J. C. Prodger and C. J. Hayes, *Tetrahedron Lett.*, 2002, **43**, 6609;
 (d) W. Sun, X. Ma, L. Hong and R. Wang, *J. Org. Chem.*, 2011, **76**, 7826.
- 3 (a) Z. Xu and X. Lu, *Tetrahedron Lett.*, 1997, **38**, 3461; (b) Z. Xu and X. Lu, *Tetrahedron Lett.*, 1999, **40**, 549.
- 4 (a) X. Zhu, C. E. Henry and O. Kwon, *Tetrahedron*, 2005, 61, 6276;
 (b) I. P. Andrews and O. Kwon, *Org. Syn.*, 2011, 88, 138;
 (c) G. Zhao and M. Shi, *J. Org. Chem.*, 2005, 70, 9975;
 (d) S. Zheng and X. Lu, *Org. Lett.*, 2008, 10, 4481.
- S. Castellano, H. D. G. Fiji, S. S. Kinderman, M. Watanabe, P. De Leon, F. Tamanoi and O. Kwon, J. Am. Chem. Soc., 2007, 129, 5843; (b) M. Watanabe, H. D. G. Fiji, L. Guo, L. Chan, S. S. Kinderman, D. J. Slamon, O. Kwon and F. Tamanoi, J. Biol. Chem., 2008, 283, 9571; (c) J. Lu, L. Chan, H. D. G. Fiji, R. Dahl, O. Kwon and F. Tamanoi, Mol. Cancer Ther., 2009, 8, 1218; (d) Z. Wang, S. Castellano, S. S. Kinderman, C. E. Argueta, A. B. Beshir, G. Fenteany and O. Kwon, Chem. Eur. J., 2011, 17, 649; (e) D. Cruz, Z. Wang, J. Kibbie, R. Modlin and O. Kwon, Proc. Nattl. Acad. Sci. U.S. A., 2011, 108, 6769; (f) Y. Du and

X. Lu, J. Org. Chem., 2003, **68**, 6463; (g) T. Q. Pham, S. G. Pyne, B. W. Skelton and A. H. White, J. Org. Chem., 2005, **70**, 6369.

- (a) L. Jean and A. Marinetti, *Tetrahedron Lett.*, 2006, **47**, 2141;
 (b) A. Scherer and J. A. Gladysz, *Tetrahedron Lett.*, 2006, **47**, 6335;
 (c) N. F. Bregeot, L. Jean, P. Retailleau and A. Marinetti, *Tetrahedron*, 2007, **63**, 11920;
 (d) Y. Fang and E. N. Jacobsen, *J. Am. Chem. Soc.*, 2008, **130**, 5660;
 (e) S. Zheng and X. Lu, *Org. Lett.*, 2008, **10**, 4481;
 (f) N. Pinto, N. F. Bregeot and A. Marinetti, *Eur. J. Org. Chem.*, 2009, **74**, 146.
- 7 J. R. Liddell, Nat. Prod. Rep., 2002, 19, 773.
- 8 M. Sampath, P.-Y. B. Lee and T.-P. Loh, Chem. Sci., 2011, 2, 1988.
- 9 X. Han, F. Zhong, Y. Wang and Y. Lu, Angew. Chem. Int. Ed., 2012, 51, 767.
- 10 Albeit not in a catalytic form, Cristau demonstrated the first γ -umpolung additions to activated allenes; see: (a) H.-J. Cristau, J. Viala and H. Christol, Tetrahedron Lett., 1982, 23, 1569. For reactions catalytic in tertiary phosphine, see: (b) B. Trost and C.-J. Li, J. Am. Chem. Soc., 1994, 116, 10819; (c) X. Lu and C. Zhang, Synlett, 1995, 6, 645; (d) Z. Chen, G. Zhu, Q. Jiang, D. Xiao, P. Cao and X. Zhang, J. Org. Chem., 1998, 63, 5631; (e) C. Alvarez-Iberra, A. Csaky and C. Gomez de la Oliva, J. Org. Chem., 2000, 65, 3544; (f) C. Lu and X. Lu, Org. Lett., 2002, 4, 4677; (g) Z. Pakulski, O. M. Demchuk, J. Frelek, R. Luboradzki and K. M. Pietrusiewicz, Eur. J. Org. Chem., 2004, 69, 3913; (h) D. Virieux, A.-F. Guillouzic and H.-J. Cristau, Tetrahedron, 2006, 62, 3710; (i) Y. K. Chung and G. C. Fu, Angew. Chem., Int. Ed., 2009, 48, 2225; (j) X.-Y. Guan, Y. Wei and M. Shi, Org. Lett., 2010, 12, 5024; (k) Q. Zhang, L. Yang and X. Tong, J. Am. Chem. Soc., 2010, 132, 2550; (1) S. W. Smith and G. C. Fu, J. Am. Chem. Soc., 2009, 131, 14231.
- (a) X.-F. Zhu, J. Lan and O. Kwon, J. Am. Chem. Soc., 2003, 11 125, 4716; (b) X.-F. Zhu, C. E. Henry and O. Kwon, Tetrahedron, 2005, 61, 6276; (c) X.-F. Zhu, C. E. Henry, J. Wang, T. Dudding and O. Kwon, Org. Lett., 2005, 7, 1387; (d) X.-F. Zhu, A. Schaffner, R. C. Li and O. Kwon, Org. Lett., 2005, 7, 2977; (e) Y. S. Tran and O. Kwon, Org. Lett., 2005, 7, 4289; (f) T. Dudding, O. Kwon and E. Mercier, Org. Lett., 2006, 3643; (g) E. Mercier, B. Fonovic, C. Henry, O. Kwon and T. Dudding, Tetrahedron Lett., 2007, 48, 3617; (h) X.-F. Zhu. C. E. Henry and O. Kwon, J. Am. Chem. Soc., 2007, 129, 6722; (i) C. E. Henry and O. Kwon, Org. Lett., 2007, 9, 3069; (j) Y. S. Tran and O. Kwon, J. Am. Chem. Soc., 2007, 129, 12632; (k) V. Sriramurthy, G. A. Barcan and O. Kwon, J. Am. Chem. Soc., 2007, 129, 12928; (1) G. S. Creech and O. Kwon, Org. Lett., 2008, 10, 429; (m) G. S. Creech, X. Zhu, B. Fonovic, D. Travis and O. Kwon, Tetrahedron, 2008, 64, 6935; (n) H. Guo, Q. Xu and O. Kwon, J. Am. Chem. Soc., 2009, 131, 6318; (o) S. Vardhineedi and O. Kwon, Org. Lett., 2010, 12, 1084; (p) S. N. Khong, Y. S. Tran and O. Kwon, Tetrahedron, 2010, 66, 4760; (q) T. J. Martin, Y. S. Tran, V. Vakhshori and O. Kwon, Org. Lett., 2011, 13, 2586; (r) J. Szeto, V. Sriramurthy and O. Kwon, Org. Lett., 2011, 13, 5420; (s) Y. C. Fan and O. Kwon, Phosphine Catalysis, in Science of Synthesis, ed. B. List, Asymmetric Organocatalysis, Lewis Base and Acid Catalysis, Georg Thieme Verlag KG, Stuggart, New York, 2012, vol. 1, pp. 713-782.
- 12 (a) M. Ciclosi, C. Fava, R. Galeazzi, M. Orena and J. Sepulveda-Arques, *Tetrahedron Lett.*, 2002, 43, 2199; (b) S. Kobbelgaard, S. Brandes and K. A. Jørgensen, *Chem. Eur. J.*, 2008, 14, 1464.
- 13 See the Supporting Information[‡] for details regarding reaction optimization. Cursory attempts at employing commercially available cinchona alkaloids for the asymmetric variant of the rearrangement failed to produce the desired products in appreciable yields.
- 14 L.-W. Ye, X.-L. Sun, Q.-G. Wang and Y. Tang, Angew. Chem. Int. Ed., 2007, 46, 5951.
- 15 B. J. Cowen, L. B. Saunders and S. Miller, J. Am. Chem. Soc., 2009, 131, 6105.
- 16 B. M. Trost and G. R. Dake, J. Am. Chem. Soc., 1997, 119, 7595.