Cell culture and in vitro cytotoxicity testing were performed as described.¹⁸

General Procedures. Method A. Coupling Reaction with Dicyclohexylcarbodiimide/1-Hydroxybenzotriazole. To a solution of amino acid ester hydrochloride (13 mmol) in dry pyridine (10 mL) were added N-methylmorpholine (1.64 mL, 15 mmol), Cbz-protected amino acid (13 mmol), HOBT (20 mmol), and slowly a solution of DCC (20 mmol) in dry pyridine (20 mL), at 0 °C. The reaction mixture was stirred at 0 °C for 4 h and then 36 h at room temperature. The urea formed was removed by filtration, and the filtrate was evaporated. The oily residue was dissolved in ethyl acetate (200 mL) and washed successively with 5% NaHCO₃, water, 0.1 M HCl, and water, dried (MgSO₄), and evaporated. The crude product was obtained as a solid. The final purification was achieved by silica gel column chromatography and recrystallization.

Method B. Saponification of Peptide Esters in Aqueous Dioxane. To a solution of peptide esters (10 mmol) in a mixture of dioxane (90 mL) and water (10 mL) was added 1 M NaOH (13 mmol), and the mixture was stirred at room temperature overnight. The aqueous phase was acidified with solid citric acid while being chilled. Solvent was removed under reduced pressure at 45 °C. The residue was dissolved in ethyl acetate (200 mL) and washed with 0.01 M HCl and water. The organic layer was dried (MgSO₄) and evaporated under reduced pressure to yield crude product. In most cases, the products obtained needed no further purification.

L-Leucine Methyl Ester Hydrochloride (4). The title compound was prepared from L-leucine (5 g, 0.03 mol) according to the method used to prepare N° -Cbz-lysine methyl ester hydrochloride:¹⁵ yield 11 g (93%); mp 153-154 °C; R_f (D) 0.45.

N-(Benzyloxycarbonyl)-D-valine (5). D-Valine (6 g, 0.05 mol) was dissolved in a mixture of 2 M NaOH (117 mL) and ether (26 mL) and stirred at 5 °C, and benzyl chloroformate (11.7 mL, 0.08 mol) was added drop by drop. The reaction mixture was stirred for 3 h at 5 °C, and 25 °C for an additional 30 h. The aqueous phase was separated from ether. The pH was then adjusted to 2.0 with 2 M HCl, and extracted with ethyl acetate (3 × 50 mL).

N-[N-(Benzyloxycarbonyl)-D-valylleucyl-N⁶-(benzyl $oxylcarbonyl)lysyl]-1-\beta$ -D-arabinofuranosylcytosine (2a). A solution of 9 (1.3 g, 2.1 mmol), N-hydroxysuccinimide (0.35 g, 3.0 mmol), and EDCI (0.4 g, 3.0 mmol) in dry DMF (50 mL) was stirred for 2 h at room temperature. To the solution was added 1 (0.5 g, 2.1 mmol). The mixture was stirred at 45 °C for 20 h. The solvent was removed in vacuum, and the oily residue was dissolved in ethyl acetate, washed successively with 0.01 M HCl and water, dried $(MgSO_4)$, and concentrated under vacuum. The product was purified by column chromatography on silica gel 60 with solvent B to give 1.44 g (82%) of 2a (after crystallization from ethyl acetate–ether): mp 136–138 °C; $[\alpha]^{20}_{D} = +16.7^{\circ}$ (c = 0.7, MeOH); R_{f} (B) 0.33; ¹H NMR (DMSO- d_{6}) δ 0.85 (q, J = 5 Hz, 12 H, 2 C(CH₃)₂), 1.00-2.05 (m, 10 H, 4 CH₂, 2 CH), 2.97 $(m, 2 H, CH_2), 3.32 (s, 1 H, NH), 3.62 (m, J = 5 Hz, 2 H, 5'-CH_2),$ 3.75-4.00 (m, 3 H, 3 OH), 4.06 (m, 1 H, CH), 4.37 (m, 2 H, 2CH), 4.90-5.15 (m, 5 H, 2 OCH₂, H-4'), 5.50 (m, 2 H, H-2', H-3'), 6.05 (d, J = 4 Hz, 1 H, H-1'), 7.15 (d, J = 8 Hz, 1 H, CH), 7.35 (m, T)11 H, 2 C₆H₅, NH), 7.90 (m, 1 H, NH), 8.05 (d, J = 8 Hz, 1 H, CH), 8.18 (m, 1 H, NH), 10.90 (d, 1 H, NH). Anal. (C₄₂H₅₇N₇O₁₂) C, H, N.

N-(D-Valylleucyllysyl)-1-β-D-arabinofuranosylcytosine (2). Compound 2a (1.27 g, 1.5 mmol) was hydrogenolyzed over 10% Pd/C (0.4 g) in dry methanol (100 mL) in the presence of 2% HCOOH. The catalyst was removed by filtration, the filtrate was evaporated in vacuum, and the residue was recrystallized from methanol-ether to afford the title compound: yield 0.7 g (87%); mp 126-130 °C; $[\alpha]^{20}_{D} = +14.6^{\circ}$ (c = 0.2, MeOH); ¹H NMR (DMSO- d_{6}) δ 0.80 (q, J = 5 Hz, 12 H, 2 C(CH₃)₂), 1.07-2.05 (m, 10 H, 4 CH₂, 2 CH), 2.75 (m, 2 H, CH₂), 3.22 (d, 1 H, CH), 3.62 (d, J = 5 Hz, 2 H, 5'-CH₂), 3.80-4.10 (m, 3 H, 3 OH), 4.40 (m, 2 H, 2 CH), 4.50-6.00 (m, 7 H, 2 NH₂, 3 H, H-3', H-2', H-4'), 6.05 (d, J = 5 Hz, 1 H, H-1'), 7.15 (d, J = 8 Hz, 1 H, CH), 8.10 (d, J = 8 Hz, 1 H, CH), 8.35 (m, 2 H, NH). Anal. (C₂₆H₄₅N₇O₈) C, H, N.

cis-Bis(pyridine)platinum(II) Organoamides with Unexpected Growth Inhibition Properties and Antitumor Activity

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The platinum(II) organoamides [Pt(NRCH₂)₂L₂] (L = pyridine (py), R = p-HC₆F₄, C₆F₅, p-IC₆F₄, p-ClC₆F₄, p-C₆F₅C₆F₄, L = 4-methylpyridine, R = p-HC₆F₄) and [Pt(NRCH₂CH₂NR')(py)₂] (R = p-HC₆F₄, R' = C₆F₅, p-BrC₆F₄, or p-MeC₆F₄) inhibit the growth of murine L1210 leukemia cells in culture with ID₅₀ values for continuous exposure in the range 0.6–2.7 μ M. Representative complexes are also active against L1210 cells in 2-h pulse exposures, as well as against the cisplatin-resistant variant L1210/DDP and human colonic carcinoma cell lines HT 29 and BE. Three complexes [Pt(NRCH₂)₂L₂] (R = p-HC₆F₄, C₆F₅, or p-IC₆F₄) have good activity ($T/C \ge 180\%$) against P388 leukemia in mice, and all other compounds tested are active except when R = p-C₆F₅C₆F₄, L = py. Although the molecular basis of the biological activity of these complexes is not known, the observation of good activity for amineplatinum(II) compounds with no hydrogen substitutents on the nitrogen donor atoms introduces a new factor in the anticancer behavior of platinum(II) complexes.

Introduction

Initial studies of the antitumor effects of cis-[PtCl₂(py)₂] (py = pyridine) revealed moderate activity against Ehrlich ascites carcinoma in vivo¹ but none against the sarcoma 180 tumor model.² More recently, the compound has been shown to be cytostatic against murine L1210 leukemia cells in culture, though with substantially less potency than either cis-[PtCl₂(NH₃)₂] (cisplatin) or, surprisingly,

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trans-[PtCl₂L₂] (L = py or 4-methylpyridine (mepy)), and inactive against P388 leukemia in mice.³ The poor biological activity of cis-[PtCl₂(py)₂] may be attributable to the fact that the complex lacks an important structural characteristic of active amineplatinum(II) complexes,^{4,5} viz. a hydrogen substituent on a N-donor atom. We now report that novel fluorocarbon-substituted organoamidoplatinum(II) complexes $[Pt(NRCH_2)_2L_2]$ and [Pt- $(NRCH_2CH_2NR')(py)_2]$ (R \neq R' = polyfluorophenyl; L = py or mepy) are potent inhibitors of the growth of murine leukemia (L1210) and human colonic carcinoma cells (HT 29 and BE) in culture and that representative examples have high activity against the P388 tumor in mice despite the fact that they also lack hydrogen substituents on the N-donor atoms. In addition, the compounds studied show activity toward cisplatin-resistant L1210 cells in vitro but not against cisplatin-resistant P388 in mice.

Results

Chemistry. The fluorocarbon-stabilized organoamidoplatinum(II) complexes $[Pt(NRCH_2)_2L_2]$ and $[Pt-(NRCH_2CH_2NR')(py)_2]$ ($R \neq R' = polyfluorophenyl)$ have recently been prepared by decarboxylation,⁶⁻⁸ a synthetic method unprecedented in organoamidometallic chemistry,⁹⁻¹² and their structures have been established by X-ray crystallography⁶ and spectroscopic methods.^{7,8} They are stable to atmospheric moisture and can be recovered from pyridine by addition of water, in contrast with the extreme moisture-sensitivity of simple (ethane-1,2-diaminato)platinum(II) complexes, e.g. $[Pt(NRCH_2)_2bpy]$ (R = H or Me; bpy = 2,2'-bipyridine).^{13,14} N,N'-Bis(2,3,5,6-tetra-

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		∠L L R′		
compd	R		L	
la	p-HC ₆ I	F4	ру	
1b	C_6F_5		ру	
1 c	p-ClC ₆	F4	ру	
1 d	$p-IC_6F$	4	ру	
1e	p-HC ₆ I	F4	mepy	
lf	$p-C_6F_5$	C ₆ F ₄	ру	
compd	R	R′	L	
2a	$p-HC_6F_4$	p-BrC ₆ F ₄	ру	
2b	$p-HC_6F_4$	p-MeC ₆ F ₄	ру	
2c	p-HC ₆ F ₄	C ₆ F ₅	ру	

^a Other compounds: cis-[PtCl₂(py)₂] (3), (p-HC₆F₄NHCH₂)₂ (4).

fluorophenyl)ethane-1,2-diamine, needed for comparison of biological properties with those of the platinum organoamides, was prepared by reaction between N-(2,3,5,6tetrafluorophenyl)ethane-1,2-diamine⁸ with pentafluorobenzene: p-HC₆F₄NHCH₂CH₂CH₂NH₂ + C₆F₅H + Et₃N \rightarrow (p-HC₆F₄NHCH₂)₂ + Et₃NHF.

It is surprising that the synthesis requires much more severe conditions (180 °C/86 h) than those (refluxing EtOH/7 h) for the preparation of p-HC₆F₄NHCH₂CH₂NH₂ from ethane-1,2-diamine, since the two-carbon chain would be expected to insulate the NH₂ group from the electron withdrawing p-HC₆F₄ group.

Biological Testing Results. Initially, we investigated the growth inhibitory effects of a wide range of complexes, $[Pt(NRCH_2)_2L_2]$ (L = py, R = p-HC₆F₄, C₆F₅, p-ClC₆F₄, p-BrC₆F₄, p-IC₆F₄, p-MeC₆F₄, p-C₆F₅C₆F₄, c_{4} , p-BrC₆F₄, p-IC₆F₄, p-MeC₆F₄, p-C₆F₅C₆F₄, r 2,3,5-F₃C₆H₂; L = mepy, R = p-HC₆F₄ or C₆F₅) and [Pt(NR-(CH₂)₃NR)(py)₂] (R = p-HC₆F₄ or C₆F₅), against murine L1210 leukemia cells in culture. All were found to be active with ID₅₀ values for continuous exposure in the range 0.6-3.0 μ M, and several (1a-f, Table I) were chosen for more intensive examination together with the unsymmetrical complexes [Pt(NRCH₂CH₂NR')(py)₂] (2a-c) (Table I). The selected examples inhibit the growth of L1210 cells in culture (Table II) with similar potencies to that¹⁵ of cisplatin when assessed by both cell counting and MTT dye assays. It is noteworthy the $[Pt(NRCH_2)_2L_2]$ complexes are significantly more active than cis-[PtCl₂(py)₂]^{3,16} (3), whereas the protonated organoamide ligand (p- $HC_6F_4NHCH_2)_2$ (4) was found to be essentially inactive (Table II). Activity was preserved at much the same levels for the representative complexes 1a, 1c, and 1f in exper-

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Table II.	In	Vitro	Growth	Inhibition	Properties of	of []	Pt(NRCH ₂)	$_{2}L_{2}$	and	[Pt(N	VRCH	CH	NR')L	"] (Comple	axes
								4-4		F (-		<u> </u>		4.		

			in vitro IL	50 values ^a			
	L1210 l	L1210 leukemia		BE human arcinoma	L1210/DI)P leukemia	
compd	CC ^b	MTT	CC ^b	MTT°	CC ⁶	MTT ^c	
1a	$ \begin{array}{r} 1.2 \pm 0.2 \\ (1.6 \pm 0.1) \end{array} $	1.6 ± 0.4 (1.8 ± 0.2)	1.0-1.4	1.1–1.7	1.0 ± 0.2 (1.7)	1.4 ± 0.1 (2.0)	
1 b	0.9 ± 0.1 [0.3] ^d	1.5 ± 0.4	0. 9– 1.0	1.1-1.2	0.9 ± 0.1 [0.4] ^d	1.5 ± 0.2	
1 c	0.8 ± 0.1 (1.5 ± 0.1)	1.1 ± 0.1 (1.7 ± 0.1)	0. 9- 1.1 -	1.1–1.3	0.5 (1.5)	0.8 (1.7)	
1 d	0.6 ± 0.1	0.7 ± 0.2	-	-	0.4 ± 0.1	0.7 ± 0.1	
1 e	2.0 ± 0.8 [0.3] ^d	2.7 ± 0.2	0.8–1.0	1.0-1.1	0.7 [0.3] ^d	1.1	
1 f	1.2 ± 0.1 (2.1 ± 0.1)	1.1 ± 0.1 (2.1 ± 0.1)	-	-	0.5 (2.0)	0.7 (2.0)	
2a	0.7 ± 0.1	0.9 ± 0.1	-	-	0.6 ± 0.1	0.7 ± 0.1	
2Ъ	1.0 ± 0.2	1.6 ± 0.8	-	-	0.7 ± 0.2	0.7 ± 0.1	
2c	1.7 ± 0.1	1.6 ± 0.3	-	-	0.9 ± 0.1	-	
3	5.1 ± 1.5^{e}	5.6 ± 1.4	-	-	$3.0 \pm 1.0^{\prime}$	6.2 ± 2.0	
CisPt ^g	0.6 ± 0.2 (5.0)	1.1 ± 0.6	0.7-3.0	-	6.7 ± 0.2	20.3 ± 4.5	
4	37.0	-	-	-	-	-	

^a ID₅₀, concentration of drug in μ M to inhibit by 50% the growth in culture of murine L1210 leukemia or its cisplatin-resistant variant (L1210/DDP) or human colon carcinoma cells HT-29 and BE. ^bCoulter counter (CC). ^cTetrazolium (MTT) assay. Results are after continuous exposure (48 h) or after 2-h pulse testing (in parentheses) and are single determinations, ranges, or mean \pm range of two to four determinations. ^dTested against P388 (first column) or P388/DDP (fifth coulumn) for 48 h of continuous exposure. ^eReported values: ID₅₀ = 7.3 \pm 0.7 (ref 3a); 4.4 (ref 3b). ^fReported value: 3.3 (ref 3b). ^gCisplatin.

Table III. In Vivo A	Intitumor Activity of	$[Pt(NRCH_2)_2L_2]$ and	1 [Pt(NRCH ₂ CH ₂ NR')	L_2 Complexes
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	L	L1210 ^a		388ª	P388	/DDP ^a	
compd	% T/C ^b	$\mu mol/kg^c$	%T/C	$\mu mol/kg$	% T/C	$\mu mol/kg$	
1a	172	113	206	113	115	113	
1 b	105	538	207	269	91	538	
1 c	102	259	165	97	117	258	
1 d			180	104			
1e			126	136			
1 f			112	385			
2 a			158	93	114	93	
2b			165	69	131	110	
3			96	75	94	150	
CisPt			248	20	122	20	
4			108	140			

^aCell lines tested were murine leukemias L1210, P388 and cisplatin-resistant P388 (P388/DDP). ^bResults are mean survival time of treated animals compared with those of control animals expressed as a percentage (% T/C). Activity is indicated if T/C > 125% for L1210 and P388, and >150% for P388/DDP. ^c The doses represent the optimum dose for active compounds, or the maximum tolerated dose or highest dose used for inactive compounds.

iments where the compounds were incubated with the cells for only 2 h (Table II). By contrast 2-h pulse treatments with cisplatin resulted in an approximate 8-fold loss in potency (Table II). High activity was also observed for representative compounds tested against two human colon cancer cell lines, BE and HT-29, again with ID_{50} values comparable with those of cisplatin. Testing against the cisplatin-resistant L1210 cell line (L1210/DDP) in vitro revealed that complexes 1a-f and 2b were at least as active as they were against the sensitive cells (Table II) and again they were more active than 3. This activity against cisplatin-resistant cells was maintained in 2-h pulse treatments (Table II). The responses of murine P388 and cisplatin-resistant P388 cells to the organoamides 1b and 1e in culture were similar to those of the L1210 and L1210/DDP cell lines.

The organoamidoplatinum(II) complexes except 1f are active (T/C > 125%) against P388 leukemia in the mouse model and three (1a, 1b, and 1d) have good activity $(T/C \ge 180\%)$ (Table III). This markedly contrasts the inactivity of 3 in vivo, (T/C = 109%),³ which we have confirmed (Table III). Optimum activity for 1a-e, 2a, and 2b was lower and was observed at much higher doses than for cisplatin. This lower efficacy and potency may arise from the lack of solubility of the organoamides in aqueous media. Initial in vivo testing against L1210 leukemia has shown that 1a is active (T/C = 172%), whereas 1b, 1c, and 3^3 are not. T/C values in testing against B16 melanoma were 130% (113 μ mol/kg), 130% (269 μ mol/kg), and 134% (259 μ mol/kg) for 1a, 1b, and 1c, respectively. These do not reach the value $(T/C \ge 150\%)$ considered active against this tumor type. By contrast with the cell culture results for behavior against cisplatin-resistant lines L1210/DDP and P388/DDP, 1a-c, 2a, and 2b were inactive against P388/DDP in the mouse model.

Discussion

The most noteworthy feature of the cell culture testing results is the pronounced activity against the cisplatinresistant lines. In both these tests and against normal L1210 cells, the activity of the [N,N'-bis(polyfluorophenyl)ethane-1,2-diaminato(2-)]bis(pyridine)platinum(II)complexes 1a-f and 2a-c is significantly greater than that $of <math>3^{3,16}$ (Table II). In general, the ID₅₀ values for L1210 are similar to or less than those of the anomalously active trans-PtCl₂(py or mepy)₂.³ A further interesting feature is the maintenance of high activity in 2-h pulse testing whereas cisplatin's declines 8-fold. This suggests that sustained exposure to the compounds is not necessary for activity. Variation in the fluorocarbon groups including the presence of two different groups (2a–c) has little effect on ID_{50} values, nor does the change of neutral ligand from pyridine to 4-methylpyridine. Some structure–activity variation may emerge with examination of a wider range of polyfluorophenyl groups, but this requires development of new synthetic methods. The currently used decarboxylation reactions^{6–8} lead to $p-XC_6F_4$ substituents on nitrogen.

By contrast with the in vitro data, the in vivo results show considerable sensitivity to change of ligand. Activity against P388 increases in the sequence: R, R' = p-C₆F₅C₆F₄ < p-HC₆F₄, p-BrC₆F₄ < p-HC₆F₄, p-MeC₆F₄ $\simeq p$ -ClC₆F₄ < p-IC₆F₄ < C₆F₅ $\simeq p$ -HC₆F₄ for L = py, and L = mepy < py for R = p-HC₆F₄ (Table III). These relationships unfortunately do not provide a clear lead in the design of more active drugs of the types 1 or 2. The consistent P388 in vivo activity of these complexes, and the activity of 1a against L1210, highlight the inactivity of cis-PtCl₂(py)₂, the difference being far more marked than that observed in cell culture testing. There are unpublished NIH data¹⁷ for four cis- $[PtX_2L_2]$ (L = pyridine or substituted pyridine) complexes. Two are inactive in the P388 screen, while $cis-[PtCl_2(4-Pr-py)_2]$ and $cis-[Pt(C_2O_4)(py)_2]$ have moderate activity (optimum T/C = 140%), though even the latter is inactive on our testing schedule. Thus, the high activity of 1a, 1b, and 1d is unexpected, as is the consistent activity of the complexes (except 1f). The activity is of particular interest since the organoamides do not have a hydrogen substituent on the nitrogen donor atoms of the amine ligands (or the amide ligands), a characteristic feature of active cis-diamineplatinum(II) complexes.^{4,5} Indeed, it is also unusual for active compounds to have four nitrogen donor atoms.⁵

Although the molecular basis of the biological activity of complexes 1a-f is not known, some possibilities can be excluded. For example, the cytostatic and antitumor behavior does not derive from the possible hydrolysis products, $(RNHCH_2)_2$ (R as in Table I), since (p- $HC_6F_4NHCH_2$ has poor cytostatic behavior against L1210 cells (Table II) and is inactive against P388 in vivo (Table III). Similarly, cytostasis is not due to pyridine, which has $ID_{50} > 100 \ \mu M$. The compounds are unlikely to be prodrugs for cis-[PtCl₂(py)₂], since this complex is less effective than 1a-f in vitro and inactive in vivo. The ¹H and ¹⁹F NMR spectra of 1a in MeOH/D₂O (9:1, v/v) and the ¹⁹F NMR spectrum in EtOH/MEM (9:1, v/v) (MEM = minimum essential medium) are unchanged after 2 h, hence the undissociated complex may be the extracellular active agent in 2-h pulse testing against L1210 leukemia. However, the concentrations needed for NMR measurements are significantly higher than those in biological testing.

It is possible that the organoamide ligands increase the biological effectiveness of the $Pt(py)_2^{2-}$ units; alternatively the $Pt(NRCH_2)_2^{2-}$ group may itself contribute to biological activity. In any case, observation of good activity for amineplatinum(II) compounds with no hydrogen substituents on the nitrogen donor atoms introduces a new factor in the anticancer behavior of platinum(II) complexes. A detailed study of breakdown products of $[Pt(NRCH_2)_2L_2]$ complexes under biologically relevant conditions has begun.

Experimental Section

(a) Reagents and Synthesis. Spectroscopic and analytical

methods are as in earlier papers.^{7,8,18} Cisplatin was obtained from the Institute of Drug Technology, Melbourne, Australia. *cis*dichlorobis(pyridine)platinum(II) was prepared by the reported method,¹⁹ and the identity was confirmed by indexation of the X-ray powder photograph against parameters derived from the X-ray crystal structure.²⁰ Preparations of [*N*,*N*'-bis(polyfluorophenyl)ethane-1,2-diaminato(2–)]platinum(II) complexes were carried out by the reported methods^{7,8} and the compounds had spectroscopic properties identical with those reported.

N,N'-Bis(2,3,5,6-tetrafluorophenyl)ethane-1,2-diamine (4). N-(2,3,5,6-Tetrafluorophenyl)ethane-1,2-diamine⁸ (9.1 mmol), pentafluorobenzene (9.0 mmol), and triethylamine (72 mmol) were heated in a sealed Carius tube at 180 °C for 86 h. The resulting light yellow suspension was evaporated to dryness under vacuum. Extraction with ether, filtration, drying over MgSO₄, and evaporation under vacuum gave the crude title compound, which was purified by column chromatography (silica, Kieselgel 60, 70-230 mesh, Merck) with CH₂Cl₂/petroleum ether (bp 60-65 °C) (2:3 v/v) as eluant. Evaporation to crystallization gave a colorless solid: yield 41%; mp 90-91 °C; ¹⁹F NMR spectrum (acetone-d₆) 142.2 (m, 4 F, F3,5), 160.4 (m, 4 F, F2,6); ¹H NMR spectrum (acetone-d_s) 3.71 (m, 4 H, CH₂), 5.47 (br, 2 H, NH), 6.66 (tt, ${}^{3}J_{H,F} = 10$ Hz, ${}^{4}J_{\rm H,F} = 7$ Hz, 2 H, HC₆F₄); infrared spectrum [intense features] 3448 (s) [ν (NH)], 1648 (vs), 1528 and 1509 (vs), 1466 (s), 1456 (s), 1144 (vs) $[\nu(CF)]$, 919 (s) $[\nu(CF)]$ cm⁻¹; mass spectrum m/z 356 (13, M⁺), 179 (37, HC₆F₄NHCH₃⁺), 178 (100, HC₆F₄NHCH₂⁺). Anal. $(C_{14}H_8F_8N_2)$ C, H, F, N.

(b) Biological Testing. In Vitro. L1210 or cisplatin-resistant L1210 (L1210/DDP) mouse leukemia cells were grown as suspension cultures in MEM (minimum essential medium, Eagle's) plus 1% glutamine and 15% fetal calf serum (Flow Laboratories). Drugs were dissolved in ethanol (1a-e, 2a-c and 3; final maximum solvent concentration 1.0%), ethanol/DMSO (1f; maximum solvent concentration each 0.5%), or DMF (3; maximum solvent concentration 0.5%). Cisplatin was dissolved in saline. Growth inhibition was tested in log-phase cells that were counted using a Coulter counter after 48 h in the presence of drug or solvent or after 2 h of drug exposure followed by 46 h of growth. Growth inhibition was also measured by mitochondrial metabolism of a tetrazolium salt (MTT) using an assay modified from the method of Mosmann.²¹ The human colon lines HT-29 and BE were grown for 12 h in EMEM (MEM with Earle's salts) with 1% glutamine and 15% fetal calf serum. Drugs or solvents were then added as for L1210 experiments and the cells were incubated for a further 2 days. Cells were released from the surface of the culture wells with pronase and counted or used in MTT experiments. The ID_{50} , or dose causing 50% inhibition of cell growth, was determined from the curve of percentage growth versus dose. Control cultures exposed only to the vehicle were included with each test.

In Vivo.²² Tumors for in vivo testing were obtained from the NCI and included P388 and L1210 leukemias and B16 melanoma. The cisplatin-resistant tumor (P388/DDP) was a gift from Dr. S. M. Schmid, Southern Research Institute, Birmingham, AL. The tumors were kept frozen in liquid nitrogen and were reinitiated every 20 passages (10 for the P388/DDP). The leukemias were passaged weekly by intraperitoneal (ip) inoculation of 10⁶ (P388/DDP), 10⁵ (P388), or 10⁴ (L1210) cells in DBA/2 mice. A

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1:10 brei of the B16 melanoma was prepared, and 0.5 mL was inoculated subcutaneously in C57BL/6 mice fortnightly.

For the antitumor testing, 10^6 P388 leukemia cells (or 5×10^6 for P388/DDP, 10^5 for L1210, or 0.5 mL of 1:10 brei for B16) were injected ip into groups of six BDF₁ mice on day 0. On days 1, 5, and 9, Pt compounds in 1% Tween-20/saline (1:9) were administered ip at five concentrations over a 16-fold range incorporating the maximum tolerated dose. Negative control groups were untreated or received only the vehicle, and the positive control group received 5-fluorouracil (50 mg/kg) or cisplatin (6 mg/kg, for B16 and P388/DDP). Animals were observed for 30-60 days. Data are expressed as % T/C, the mean survival time of treated animals vs that of control animals expressed as a per-

centage.

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Synthesis and Antibronchospastic Activity of 8-Alkoxy- and 8-(Alkylamino)imidazo[1,2-a]pyrazines

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Theophylline still occupies a dominant place in asthma therapy. Unfortunatly its adverse central nervous system (CNS) stimulant effects can dramatically limit its use, and adjustments in the dosage are often needed. We have synthesized a new series of imidazo[1,2-a]pyrazine derivatives which are much more potent bronchodilators than theophylline in vivo and do not exhibit the CNS stimulatory profile. In vitro studies on isolated rat uterus and guinea pig trachea confirm the high potentialities of these derivatives. 6-Bromo-8-(methylamino)imidazo[1,2-a]pyrazine-3-carbonitrile (23) is identified as the most potent compound of the series. As in the case of theophylline, phosphodiesterase inhibition appears unlikely to be the unique mechanism of action of this series of heterocycles.

Many imidazo[1,2-a]pyrazine derivatives have been acclaimed for their pharmacological profile, ranging from antidepressant to inotropic or antiulcer activity.¹ In an earlier paper, several 8-hydrogeno- or 8-bromoimidazo-[1,2-a]pyrazines were found to exhibit uterine-relaxing, antibronchospastic, and cardiac-stimulating properties.² In the present study, we report the synthesis of new 8alkoxy- and 8-(alkylamino)imidazo[1,2-a]pyrazines (Table I) and investigate their potential as bronchodilator agents. Their pharmacological profile is compared to that of theophylline. 8-(Methylamino)imidazo[1,2-a]pyrazine derivatives demonstrate high antibronchospastic activity in vitro and in vivo. Interestingly, these compounds do not exhibit the adverse central nervous system (CNS) stimulant effects of theophylline.

Chemistry

The classical condensation of an α -halogeno carbonyl compound with aminopyrazine (1) or 2-amino-3,5-dibromopyrazine (9) led to the imidazo[1,2-*a*]pyrazine series. Imidazo[1,2-*a*]pyrazine (2) yielded 3,5-dibromoimidazo-[1,2-*a*]pyrazine (3) via bromination with bromine in acetic acid.^{3,4} The 3-bromo-8-substituted series was obtained via telesubstitution of the bromine atom in position 5 after a nucleophilic attack at the 8-position (Scheme I). This mechanism, which has already been reported for the formation of 5-bromo-8-methoxyimidazo[1,2-*a*]pyrazine³ (4) and extended here to amino or alkylamino derivatives, was Scheme I. Synthesis of 8-Alkoxy- and 8-(Alkylamino)-3-bromoimidazo[1,2-a]pyrazines by Telesubstitution of 3,5-Dibromoimidazo[1,2-a]pyrazine $N \longrightarrow N^{H_2} \longrightarrow N \longrightarrow N^{H_2} \longrightarrow N^{H_2$



Scheme III



first described in the azoloazine series with a bridgehead nitrogen atom by Bradac et al.⁴ on the triazolo[4,3-a]-

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