

Metal-Free One-Pot Synthesis of 3-Phosphinoylbenzofurans via Phospha-Michael Addition/Cyclization of H-Phosphine Oxides and *in Situ* Generated *ortho*-Quinone Methides

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Supporting Information

ABSTRACT: A novel metal-free one-pot protocol for the effective and efficient synthesis of 3-phosphinoylbenzofurans via a phospha-Michael addition/cyclization of H-phosphine oxides and *in situ* generated *ortho*-quinone methides is described. Based on the expeditious construction of $C(sp^2)$ – P bonds, asymmetric synthesis of optically pure 3-phosphinoylbenzofurans containing chiral P-stereogenic center has also been probed by using chiral R_{p} -(-)-menthyl phenylphosphine oxide.

romatic organophosphorus compounds and their deriva-their biological properties and synthetic value.¹ For example, some structurally related organophosphorus compounds have been used as biologically active molecules in medicinal chemistry,^{1b,d} as phosphorus ligands in asymmetric catalysis,^{1a,f} and as buildings blocks in organic synthesis.^{1c} In recent decades, focusing on the synthesis of phosphorylated heterocycles through $C(sp^2)-P$ bond formation, several methodologies have been established on the basis of classical reactions of halophosphine electrophiles with active carbon nucleophiles such as organometallic reagents,² transition-metal-catalyzed cross-coupling reactions of phosphines with aryl (pseudo) halides,^{3,4} Friedel-Crafts reaction,⁵ and P-centered radicals addition to unsaturated systems.⁶ Despite significant progress in the construction of $C(sp^2)$ -P bonds, only a few approaches (Scheme 1) have been developed for the synthesis of 3phosphinoylbenzofurans.⁷ In approach (a), for example, Tsvetkov and Griffiths have developed the protocols for the synthesis of 3-phosphinoylbenzofurans using bisphosphine oxides and dialkyl benzoylphosphonates through Wittig-Horner and rearrangement reactions, respectively.^{7a,c} In 2012, Swamy and co-workers reported a three-step synthetic procedure using allenylphosphine oxides as key intermediates.^{7b,d} Among the above-mentioned approaches, notably, the C(sp²)-P bonds were constructed through traditional substitution reactions using water-sensitive P^{III} reagents. Recently, Liang and co-workers have reported a novel synthesis of 3-phosphinoylbenzofuran derivatives via a copper(II) catalyzed intermolecular cascade annulation reaction of nucleophilic diphenylphosphine oxide and propargylic alco-



Scheme 1. Previous Methods for $C(sp^2)-P$ Bond Construction and Our Design for the Synthesis of 3-Phosphinoylbenzofurans



hols.^{7e} However, the above-mentioned methodologies generally either require expensive metal catalysts or suffer from multistep preparation of the reaction precursors in some cases.

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Recently, metal-free $C(sp^2)-P$ bond construction reactions have attracted much attention as appealing alternatives to the metal-mediated and environmentally friendly processes.⁸ For example, Miura et al. developed a metal-free approach to phosphorylated heterocycles using Tf₂O through electrophilic phosphination/cyclization of alkynes.^{8h} Despite these elegant improvements, there is still high demand for novel methodologies to efficiently and effectively construct $C(sp^2)-P$ bonds for the synthesis of 3-phosphinoylbenzofuran compounds.

ortho-Quinone methides (o-QMs) are highly reactive as well as ephemeral intermediates which have been employed for decades in the synthesis of natural products and bioactive molecules.⁹ Due to their intrinsic electrophilic reactivity, o-QMs have been generally used in a series of 1,4-conjugate addition, [4 + n]-cycloaddition, intramolecular [5 + 2]-cycloaddition, and oxa- 6π -electrocyclization.¹⁰ Very recently, the Kang group developed a novel organocatalytic phosphonylation through 1,4-conjugate addition of in situ formed o-QMs with trialkylphosphites for the construction of diaryl phosphonates.¹¹ To our knowledge, however, there are few reports on the reaction of o-QMs as nucleophile acceptors with Hphosphine oxides. In light of the importance of 3-phosphinoylbenzofurans compounds and our interest in the development of transformations with o-QMs, herein we report an effective onepot metal-free method for $C(sp^2)$ –P bond formation through intermolecular phospha-Michael addition/intramolecular cyclization reaction (Scheme 1), wherein we envision that both nucleophilic P-center and electrophilic o-QMs would be generated in situ from H-phosphine oxides and chemically stable o-hydroxyl-benzyl alcohols (o-HBAs) under metal-free conditions, respectively.¹

With the above-mentioned consideration, we initiated our screenings for the model reaction of *o*-hydroxyl-benzyl alcohol (*o*-HBA) **1a** and diphenylphosphine oxide **2a** under metal-free reaction conditions (Table 1). Initially, several acids such as TsOH, AcOH, and TFA were chosen for this transformation at ambient temperature in CH₃CN to facilitate the generation of *o*-QM intermediates. Pleasingly, the desired product **3aa** was

Table 1.	Optimization	of the	Reaction	Conditions ^{<i>a</i>}

$\begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} OH \\ H \\ OH \end{array} \end{array} \begin{array}{c} Ph \\ + \end{array} \begin{array}{c} Ph \\ Ph \end{array} \begin{array}{c} O \\ H \\ Ph \end{array} \begin{array}{c} 1) \ solvent, \ temperature \\ additive \ (0.5 \ equiv) \\ \hline 2) \ K_2 CO_3, \ temperature \end{array} \begin{array}{c} \begin{array}{c} Ph \\ Ph \\ OH \end{array} \begin{array}{c} Ph \\ Ph \end{array} \begin{array}{c} Ph \\ Ph \\ Ph \end{array} \end{array} \begin{array}{c} Ph \\ Ph \\ Ph \end{array} $					
	1a	2a		3aa	
entry	additive	solvent	temp (°C)	yield (%) ^b	
1	TsOH	CH ₃ CN	25	14	
2	AcOH	CH ₃ CN	25	29	
3	TFA	CH ₃ CN	25	42	
4	_	CH ₃ CN	80	82	
5	_	$(ClCH_2)_2$	80	92	
6	_	toluene	80	75	
7	-	EtOAc	80	76	
8	-	THF	80	79	
9	-	DMF	80	0	

^{*a*}A Schlenk tube (15 mL) charging with a magnetic stirring bar and **1a** (0.2 mmol, 1.0 equiv), **2a** (0.3 mmol, 1.5 equiv), solvent (2 mL) and additive (0.1 mmol, 0.5 equiv) (if applicable) was sealed without degassing and heated at the indicated temperature until **1a** was disappeared; and then K_2CO_3 as base (0.6 mmol, 3.0 equiv) was added to the resulting mixture and stirred at the indicated temperature. ^{*b*}Yield of isolated product. TFA = trifluoroacetic acid.

obtained, despite low yields (entries 1–3). By considering the acidic instability of *in situ* formed *o*-QM intermediates, this model reaction was examined without an acid additive at increased temperature (80 °C). Surprisingly, the titled reaction proceeded smoothly and gave an improved yield of 82%, clearly demonstrating that the temperature enhancement could positively promote the formation of *o*-QMs (entry 4). In an attempt to further improve the yield, several solvents were examined (entries 5–9). With exception of polar DMF (entry 9), the desired product **3aa** could be obtained in good yields using toluene, EtOAc, (CICH₂)₂, CH₃CN, and THF, in which (CICH₂)₂ as the optimal solvent afforded **3aa** in 92% yield (entry 5).

With the optimal reaction conditions in hand, the scope of *in* situ generated o-QMs in our metal-free, one-pot reaction was first examined. A series of o-HBAs as o-QM precursors were tested in the presence of diphenylphosphine oxide (2a) (Scheme 2). The substrates bearing electron-donating sub-

Scheme 2. Scope of *o*-Hydroxyl Benzyl Alcohols (*o*-HBAs)^{a,b}



^{*a*}Unless otherwise specified, reactions were carried out using 1 (0.2 mmol) and **2a** (0.3 mmol) in $(\text{ClCH}_2)_2$ (2 mL) at 80 °C until 1 disappeared, and then $K_2\text{CO}_3$ (0.6 mmol) was added. ^{*b*}Times listed were for the whole one-pot process.

stituents including methyl (**3ca**, **3ja**), *tert*-butyl (**3da**, **3ia**), and methoxy (**3ba**) groups at ring A afforded the corresponding products in 65–97% yields. The nitro group and halogen (F, Cl, and Br) containing substrates at ring A gave **3ea**–**3ha** in moderate yields under the standard reaction conditions, providing possibilities for late-stage chemical transformations of the products. Besides, aryl and alkyl acetylene based substrates were also screened, affording products (**3ka**–**3ra**) in good to excellent yields. Substrates with different aryl (e.g., 4-MeC₆H₄, 4-ClC₆H₄, 4-*n*-amyl-C₆H₄, **3**-FC₆H₄) substituted acetylene motifs were further investigated, and the desired 3phosphinoylbenzofurans **3ka**–**3na** were afforded in 59–96% yields. Besides, the substrates having alkyl (e.g., *tert*-butyl, cyclopropyl, and *n*-amyl) substituted acetylene units were also tolerable under standard reaction conditions, giving the desired products **3oa** (78%), **3pa** (75%), and **3qa** (88%), respectively. Additionally, one example using the symmetric *o*-HBA substrate **1r** was conducted, and interestingly bis-3phosphinoylbenzofuran product **3ra** was smoothly delivered in 73% yield. To test the synthetic potential of this protocol, we carried out a gram-scale reaction with 5 mmol (1.12 g) of **1a**, and the titled product **3aa** was obtained with analogous reactivity in good yield (91%, 1.85 g).

Next, we explored the scope with respect to the H-phosphine oxides (Scheme 3). Gratifyingly, the reaction of H-phosphine

Scheme 3. Scope of H-Phosphine Oxides^{*a,b*}



^{*a*}Unless otherwise specified, reactions were carried out using 1a (0.2 mmol) and 2 (0.3 mmol) in $(ClCH_2)_2$ (2 mL) at 80 °C until 1a disappeared, and then K_2CO_3 (0.6 mmol) was added. ^{*b*}Times listed were for the whole one-pot process.

oxides bearing the electron-donating groups including methyl (2b, 2c) and methoxy (2d) at the *para* or *ortho* position of the phenyl ring with *o*-HBA (1a) under optimal conditions readily gave products 3ab-3ad in 83%, 92%, and 70% yields, respectively. The substrate 2e with electron-withdrawing chlorine at the phenyl ring could also proceed smoothly, affording the desired product 3ae in 97% yield. While di(benzyl)phosphineoxide (2f) was introduced as the reactant, the product 3af was obtained in good yield (80%). Moreover, the unsymmetrical phosphine oxide (2g) was also effective for this transformation, giving the desired product 3ag in 82% yield. Unexpectedly, diethyl phosphonate (2h) was ineffective in this reaction following the decomposition of the starting materials.

Recently, optically pure R_{p} -(-)-menthyl phenylphosphine oxide 4 was used as an excellent building block for the intramolecular rearrangement reaction and intermolecular addition reaction in our lab, leading to various optically pure tertiary phosphine oxides compounds.¹³ Considering the importance of chiral P-stereogenic phosphorus compounds in organic synthesis, the reactions using the chiral H-phosphine oxide were further explored in the asymmetric synthesis of chiral 3-phosphinoylbenzofurans through phospha-Michael addition/cyclization reactions. As shown in Scheme 4, various *o*-HBAs were examined in this metal-free, one-pot stereoselective reaction. Generally, all the reactions gave the corresponding chiral 3-menthylphenylphosphinoylbenzofurans (**5a**–**5h**) with very high diastereoselectivity in good yields (up to 92% yield). The absolute configuration of **5a** was Scheme 4. Stereoselective Synthesis of Chiral 3-Menthylphenylphosphinoylbenzofurans a,b,c



^{*a*}Unless otherwise specified, reactions were carried out using 1 (0.2 mmol) and 4 (0.3 mmol) in $(ClCH_2)_2$ (2 mL) at 80 °C until 1 disappeared, and then K₂CO₃ (0.6 mmol) was added. ^{*b*}Times listed were for the whole one-pot process. ^{*c*}The diastereomeric ratio were detected by ³¹P NMR spectroscopy.

unambiguously established by X-ray crystallographic analysis (CCDC 1568383).

Based on the above results and previous reports, a plausible mechanism was proposed as shown in Scheme 5. The thermal

Scheme 5. Proposed Mechanism



dehydration of *o*-HBA **1a** first resulted in the generation of *o*-QM intermediate, and then an intermolecular phospha-Michael addition of nucleophilic phosphinous acid **2a**' tautomerized from **2a** produced the alkynylphosphine oxide intermediate **3aa**' (CCDC 1568384). Following the isomerization of **3aa**', *in situ* formed allenylphosphine oxide intermediate **A** quickly underwent an intramolecular 1,4-addition/isomerization to afford the final product 3-phosphinoylbenzofuran **3aa**.

In summary, we have developed a novel metal-free one-pot protocol featuring phospha-Michael addition/cyclization of Hphosphine oxides with *in situ* generated *o*-QMs, leading to an effective method for the construction of $C(sp^2)$ -P bonds in various structurally interesting 3-phosphinoylbenzofurans. Importantly, asymmetric synthesis of 3-phosphinoylbenzofurans containing chiral P-stereogenic centers has been explored by using optically pure $R_{\rm P}$ -(-)-menthylphenylphosphine oxide. Further investigations on the development of phosphorus-directed methodologies initiated by such type of metal-free phospha-Michael addition are currently underway in our group.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.or-glett.7b03863.

Experimental procedure and spectra data (PDF)

Accession Codes

CCDC 1568383–1568384 contain the supplementary crystallographic data for this paper. These data can be obtained free of charge via www.ccdc.cam.ac.uk/data_request/cif, or by emailing data_request@ccdc.cam.ac.uk, or by contacting The Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax: +44 1223 336033.

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Notes

The authors declare no competing financial interest.

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