

Highly Efficient Approach to Orthogonally Protected (2*S*,4*R*)- and (2*S*,4*S*)-4-Hydroxyornithine

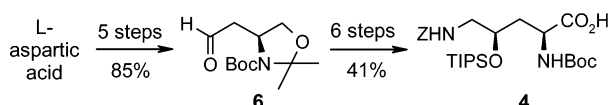
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ABSTRACT



A concise stereoselective approach to both orthogonally protected (2*S*,4*R*)- and (2*S*,4*S*)-4-hydroxyornithine, key constituents of the biphenomycin- and clavalanine-type antibiotics, respectively, has been developed. The approach is based on bis(oxazoline) copper(II)-complex-catalyzed diastereoselective Henry reactions of nitromethane with the homoserine-derived aldehyde **6**. The synthesis of this versatile chiral building block has been markedly improved.

4-Hydroxyornithine is a rare amino acid found in lentils¹ (e.g., *Lens culinaris* Medik.) and some members of the genus *Vicia*² (e.g., *V. unijuga* A. Br.). Moreover, it is a key constituent of the β -lactam antibiotic clavalanine³ and the cyclopeptide antibiotics biphenomycin A and B **1**.⁴ The latter have received considerable attention in recent years due to their high in vitro and in vivo antibacterial activity against multiresistant gram-positive pathogens.⁵

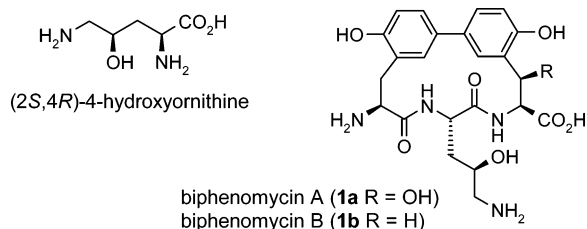


Figure 1.

As part of our program directed toward a convergent total synthesis of these antibiotics⁶ and analogues thereof with a modified biaryl moiety, we sought an efficient access to an

appropriate N^{α} , N^{δ} , O^{γ} -protected (2*S*,4*R*)-4-hydroxyornithine building block. While a number of synthetic pathways, including stereoselective approaches, have been developed for the synthesis of 4-hydroxyornithine,⁷ only two approaches dealt with the synthesis of a derivative bearing N^{α} , N^{δ} , O^{γ} -protection suitable for peptide synthesis. Schmidt et al. reported a 13 step synthesis starting from (*R*)-isopropylidene glyceraldehyde to form the *N,O*-acetal **2**, albeit in low overall yield.⁸ More recently, Rudolph et al. described a very concise

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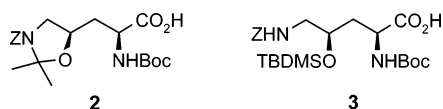
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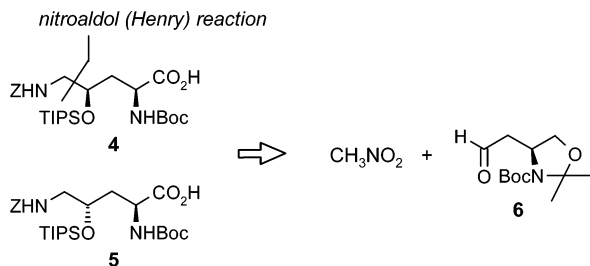
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access to the TBDMS-protected 4-hydroxyornithine **3** starting from (*S*)-*N*-Boc aspartic acid *tert*-butyl ester. This approach, which is based on an initial homologization of the acid side chain to form an α -nitroketone and its subsequent diastereoselective reduction to the corresponding β -nitro alcohol, however, also suffers from a low overall yield.^{5,9}



In this paper, we disclose a short and efficient stereoselective approach to both orthogonally $N^{\alpha}, N^{\delta}, O^{\gamma}$ -protected (2*S*,4*R*)- and (2*S*,4*S*)-4-hydroxyornithine based on an asymmetric nitroaldol (Henry) reaction of nitromethane with the homoserine-derived aldehyde **6**¹⁰ (Scheme 1).¹¹

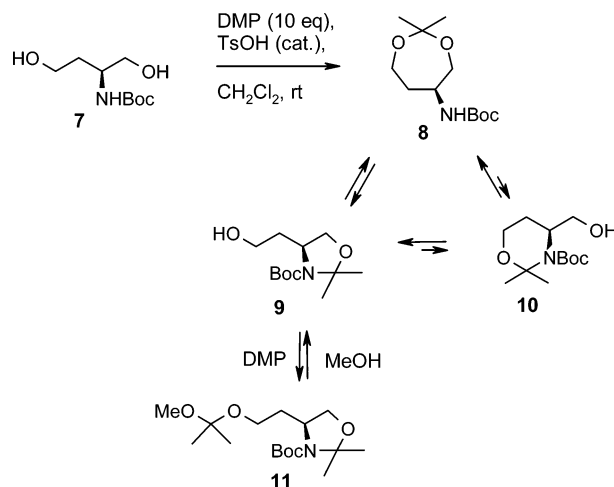
Scheme 1. Retrosynthetic Analysis



A simple two-step preparation of building block **6** starting from readily available *N*-Boc-protected (*S*)-2-amino-1,4-butanediol **7**¹² was previously reported by Ksander and co-workers.^{13,14} This commonly used approach, however, suffers from low regioselectivity in the formation of the five-

membered cyclic *N,O*-acetal **9** (Scheme 2). Thus, this crucial step was reported to proceed in only 42–48% yield by reacting **7** with excess 2,2-dimethoxypropane (DMP) (10 equiv, CH_2Cl_2 , rt) in the presence of a catalytic amount of TsOH due to concurrent formation of the corresponding six-membered cyclic *N,O*-acetal **10**.^{13,15}

Scheme 2. Acid-Catalyzed *N,O*-Acetal Formation of *N*-Boc-Protected (*S*)-2-Amino-1,4-butanediol **7** with DMP



In the course of our studies, however, the structure of this byproduct was revised on the basis of ^1H – ^{13}C HMBC and ^1H – ^{15}N HMQC NMR experiments to be the corresponding seven-membered cyclic *O,O*-acetal **8**. Indeed, this isomer was shown to be the product of kinetic control, which slowly equilibrates with **9** under the reported reaction conditions. No evidence has been found for the occurrence of the six-membered cyclic *N,O*-acetal **10**.¹⁶ Furthermore *N,O*-acetal **9** was shown to exist also in an equilibrium with its 1-methyl-1-methoxyethyl (MIP) ether derivative **11**, a fact not mentioned in the previous papers.¹⁷ In the end, trapping **9** to form **11** allows the overall equilibrium to shift to the desired five-membered ring system. Accordingly, slight modification of the reported reaction conditions, i.e., using 2,2-dimethoxypropane as the solvent and addition of 2-methoxypropene (3.0 equiv) to trap liberated methanol, led to **9** in high overall yield (92%) after mild hydrolysis (wet silica gel, CH_2Cl_2 , rt) of the intermediate MIP ether **11**. Finally, Swern oxidation provided the desired aldehyde **6** in 97% yield (Scheme 3).¹⁴

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(10) For recent use of this versatile chiral building block, see: (a) Dondoni, A.; Massi, A.; Minghini, E.; Sabbatini, S.; Bertolasi, V. *J. Org. Chem.* **2003**, *68*, 6172–6183. (b) Catalano, J. G.; Deaton, D. N.; Furfine, E. S.; Hassell, A. M.; McFayden, R. B.; Miller, A. B.; Miller, L. R.; Shewchuk, L. M.; Willard, D. H.; Whright, L. L. *Bioorg. Med. Chem. Lett.* **2004**, *14*, 275–278. (c) Dondoni, A.; Catozzi, N.; Marra, A. *J. Org. Chem.* **2004**, *69*, 5023–5036. (d) Dondoni, A.; Giovanni, P. P.; Massi, A. *Org. Lett.* **2004**, *6*, 2929–2932.

(11) An analogous Henry reaction-based strategy for the synthesis of 4-hydroxyornithine was previously reported by Rudolph et al. (ref 9). This approach, however, suffers both from an unfavorable stereoselectivity and from a very low yield in the key nitroaldol reaction step and therefore was not pursued further.

(12) Although *N*-Boc-protected (*S*)-2-amino-1,4-butanediol **7** is commercially available (Aldrich), its relatively high price leads us to recommend its preparation on a multigram scale from inexpensive L-aspartic acid (see Supporting Information).

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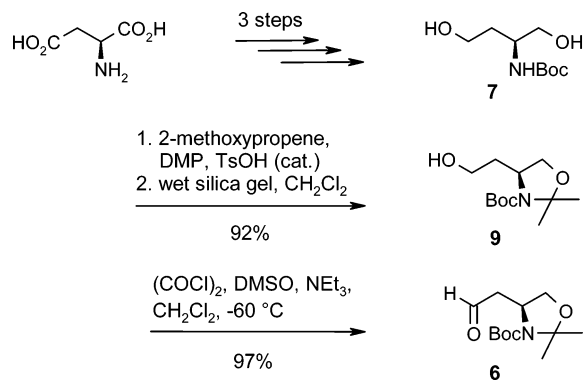
(14) For an alternative approach to **6** starting from *N*-*tert*-butoxycarbonyl-L-aspartic acid γ -benzyl ester, see: Ouerfelli, O.; Ishida, M.; Shinozaki, H.; Nakanishi, K.; Ohfun, Y. *Synlett* **1993**, 409–410.

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(16) Ab initio MO calculations (DFT/B3LYP/6-311G+) showed an energy difference of about 7.8 kcal/mol between **9** and the six-membered cyclic *N,O*-acetal **10**, indicating that only traces of this regioisomer are to be expected.

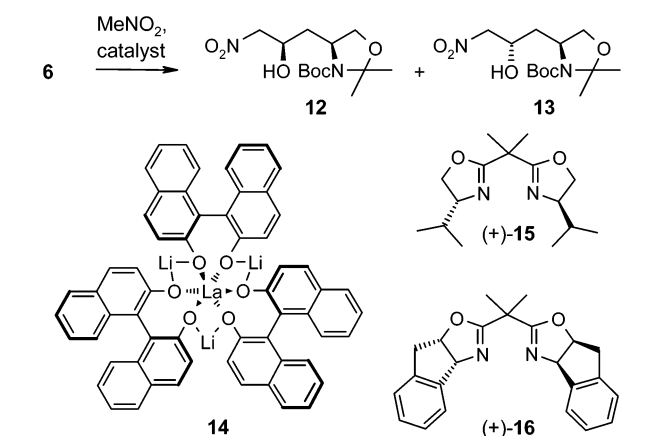
(17) Products **8**, **9**, and **11** were obtained in a ratio of 30:45:25 (as determined by ^1H NMR analysis of the crude reaction product) by treatment of **7** with DMP (10 equiv) and TsOH (0.1 equiv, CH_2Cl_2 , rt, 36 h) according to ref 15.

Scheme 3. Improved Synthesis of Aldehyde 6



With key building block **6** in hand, its nitroaldol (Henry) reaction with nitromethane was examined (Table 1). LiAlH_4 ,¹⁸ TBAF,¹⁹ as well as *t*-BuOK-catalyzed²⁰ Henry reactions led to nitro alcohols **12** and **13** with low diastereoselectivity, reflecting that the existing stereogenic center is too far away from the newly created one to exert appreciable asymmetric induction (Table 1, entries 1–3).²¹ An obvious way of resolving this problem was the introduction of additional chiral information, i.e., application of a chiral catalyst. In fact, double stereodifferentiation using Shibasaki's well-established heterobimetallic (*S*)-BINOL

Table 1. Diastereoselective Henry Reaction of Aldehyde **6** with Nitromethane



entry	catalyst	conditions	yield (%) ^a	ratio ^b 12:13
1	LiAlH_4	THF, rt	53	56:44
2	TBAF	THF, rt	33	43:57
3	<i>t</i> -BuOK	<i>t</i> -BuOH/THF, 0 °C	72	23:77
4	14	THF, -40 °C	45	98:2
5	$\{\text{Cu}[(+)\text{-15}]\}(\text{OAc})_2$	EtOH, rt	87	92:8
6	$\{\text{Cu}[(+)\text{-15}]\}(\text{OAc})_2$	EtOH, rt	85	9:91
7	$\{\text{Cu}[(+)\text{-16}]\}(\text{OAc})_2$	EtOH, rt	94	97:3
8	$\{\text{Cu}[(+)\text{-16}]\}(\text{OAc})_2$	EtOH, rt	91	8:92

^a Isolated yield. ^b Determined by HPLC analysis of crude reaction mixtures.

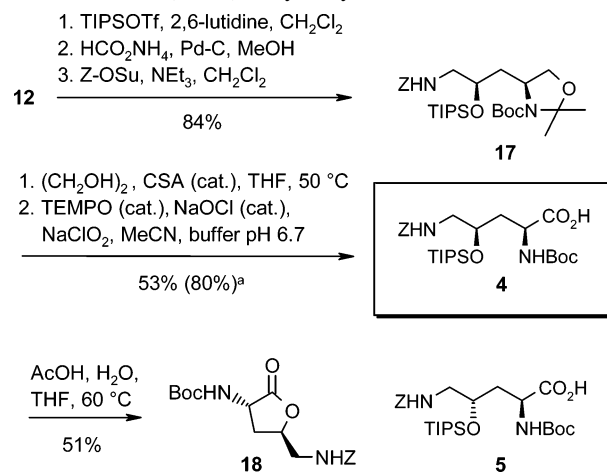
catalyst **14**²² (5 mol %, THF, -40 °C, 3 days) led to **12** with high diastereoselectivity, albeit in low yield (Table 1, entry 4).

Recently, other highly efficient chiral catalysts for asymmetric Henry reactions have been developed. Thus, Corey²³ and Maruoka²⁴ have utilized chiral quaternary ammonium fluorides as catalysts, while Trost²⁵ has presented a dinuclear zinc catalyst. Salen-cobalt(II) complexes have been used by Yamada,²⁶ whereas Jørgensen²⁷ and Evans²⁸ have introduced bis(oxazoline)-copper(II) complexes. The latter seemed to be the catalysts of choice, at least for aliphatic aldehydes, with respect to attainable yields and degree of stereoselectivity.

Indeed, application of Evans' bis(oxazoline) copper(II) acetate-based catalysts $\{\text{Cu}[(+)\text{-15}]\}(\text{OAc})_2$ and in particular $\{\text{Cu}[(+)\text{-16}]\}(\text{OAc})_2$ (5 mol %, EtOH, rt, 5 days) gave the desired nitro alcohol **12** both with high diastereoselectivity and in high yield (Table 1, entries 5 and 7). Finally, to selectively obtain diastereomer **13**, aldehyde **6** was reacted with nitromethane in the presence of the enantiomeric catalysts $\{\text{Cu}[(+)\text{-15}]\}(\text{OAc})_2$ and $\{\text{Cu}[(+)\text{-16}]\}(\text{OAc})_2$, respectively. In these cases, slightly lower stereoselectivities and yields were observed, reflecting a mismatched pairing (Table 1, entries 6 and 8).

We next turned our attention to the transformation of β -nitro alcohol **12** into the desired amino acid building block **4** (Scheme 4).²⁹ Protection of the hydroxyl group as a

Scheme 4. Synthesis of Orthogonally Protected (2*S*,4*R*)-4-Hydroxyornithine **4**



^a Based on recovered starting material.

TIPS ether proceeded smoothly under standard conditions (TIPSOTf/2,6-lutidine). Reduction of the nitro group was accomplished using ammonium formate as a hydrogen source and palladium on carbon as the catalyst to afford the

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(21) Low stereoselectivity in nitroaldol reactions with aldehydes bearing a stereogenic center at the β -position has been observed previously. See ref 9 and references therein.

corresponding amine, which was transformed (Z)-OSu/NEt₃ to the N^δ-Z-protected 4-hydroxyornithine derivative **17** in 84% overall yield (three steps). Selective hydrolysis of the N,O-acetal using several methods (e.g., pyridinium tosylate, MeOH, 60 °C; I₂, MeOH, rt or CeCl₃·7H₂O/oxalic acid, rt) proved to be difficult due to concomitant partial cleavage of the TIPS ether. Fortunately, we were able to effect this transformation cleanly and in good yield (64%, 96% based on recovered starting material) using ethylene glycol/CSA (THF, 50 °C, 2 days).³⁰ The final oxidation of the amino alcohol was best accomplished with TEMPO/NaOCl/NaClO₂³¹ to give the desired carboxylic acid **4** {mp 45–47 °C, [α]_D²² +67.6 (c 1.64, CH₂Cl₂)} in 83% yield without

epimerization. The absolute configuration of product **4** was established to be 2*S*,4*R* by subsequent transformation into the known γ-lactone **18**³² {mp 148–149 °C, lit. mp 143–145 °C; [α]_D²³ –25.2 (c 0.85, CHCl₃), lit. [α]_D²³ –22.4 (c 0.5, CHCl₃)}.

According to this reaction sequence, **13** was converted to **5** {mp 53–55 °C, [α]_D²⁰ –28.5 (c 1.00, CH₂Cl₂)} in 57% overall yield (66% based on recovered starting materials) in five steps.

In conclusion, we have developed a short and highly efficient approach to orthogonally protected (2*S*,4*R*)- and (2*S*,4*S*)-4-hydroxyornithine building blocks **4** and **5**, respectively, based on bis(oxazoline) copper(II)-complex-catalyzed diastereoselective Henry reactions of nitromethane with aldehyde **6**. In addition, a greatly improved procedure for a multigram synthesis of this valuable chiral building block and its precursor **9** has been developed.

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Supporting Information Available: Experimental procedures and characterization for all new compounds. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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