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Research paper

Part III: Novel checkpoint kinase 2 (Chk2) inhibitors; design, synthesis and biological evaluation of pyrimidine-benzimidazole conjugates



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ABSTRACT

Recently a dramatic development of the cancer drug discovery has been shown in the field of targeted cancer therapy. Checkpoint kinase 2 (Chk2) inhibitors offer a promising approach to enhance the effectiveness of cancer chemotherapy. Accordingly, in this study many pyrimidine-benzimidazole conjugates were designed and twelve feasible derivatives were selected to be synthesized to investigate their activity against Chk2 and subjected to study their antitumor activity alone and in combination with the genotoxic anticancer drugs cisplatin and doxorubicin on breast carcinoma, (ER+) cell line (MCF-7). The results indicated that the studied compounds inhibited Chk2 activity with high potency (IC₅₀ = 5.6 nM - 46.20 nM). The studied candidates exhibited remarkable antitumor activity against MCF-7 (IG₅₀ = 6.6 μ M - 24.9 μ M). Compounds **10a-c**, **14** and **15** significantly potentiated the activity of the studied genotoxic drugs, whereas, compounds **9b** and **20–23** antagonized their activity. Moreover, the combination of compound **10b** with cisplatin revealed the best apoptotic effect as well as combination of cells are in the S phase with no cells at G2/M. Structure-activity relationship was discussed on the basis of molecular modeling study using Molecular modeling Environment program (MOE).

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1. Introduction

Cancer is one of the main causes of death around the world that results from high rate of proliferation for abnormal cells [1,2]. Many factors, as infectious agents, tobacco, chronic inflammation and UV light, can lead to genomic instability which can stimulate cancerous injuries [1,2]. Radiotherapy and chemotherapy are often used in addition to surgery to decrease the probabilities cancer recurrence. The generic mode of chemotherapies, genomic instability results from targeting DNA in cancer treatment, the rapid emergence of

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https://doi.org/10.1016/j.ejmech.2018.01.072 0223-5234/© 2018 Elsevier Masson SAS. All rights reserved. drug resistance via development of chemo-resistance by cancer cells, have emphasized the urgent need to solve their shortcomings and find new strategy for cancer treatment [3,4]. The key components of the DNA damage response pathway are two kinases named as ataxia-telangiectasia mutated (ATM) and Rad3-related (ATR). They subsequently phosphorylate several substrates essential for DNA repair, cell cycle arrest, transcription, and apoptosis [5–9]. Cell cycle checkpoints represent the restriction points between each phase of the cell cycle whereby the entire process can be delayed/ stopped its progress to enable the proper sequential of the process ensuring that each stage occurs in an imperative system or to allow time for DNA repair [10–12]. Downstream phosphorylation targets of ATM and ATR are the effectors serine/threonine checkpoint kinases 1 and 2 (Chk1& Chk2) [13], which in turn phosphorylate

partially overlapping residues in other target proteins to induce cell cycle arrest and facilitate DNA repair. Chk1 is activated by ATR phosphorylation on Ser-317 and Ser-345 whereas Chk2 is activated by ATM phosphorylation on Thr-68 [14]. Downstream targets of Chk2 include Cdc25A and Cdc25C, which, on phosphorylation, go through degradation and cytoplasmic relocalization, respectively, and induce cell arrest at G1, S, and G2-M phases [15]. Another important target of Chk2 is p53, the phosphorylation of which on Ser20 regulates p53 transcriptional activation. Moreover, Chk2 phosphorylation of Hdmx Ser367, a negative regulator of p53, enhances its degradation [16,17] and endorses the accumulation of p53 and transcriptional induction of p53 responsive genes. Chk2 also phosphorylates the transcription factor E2F-1 on Ser364, thereby enhancing its stability and encourages apoptosis [18,19].

According to the current understanding of Chk2 function in tumor cells, in both biological and genetic context, Chk2 plays an important role in maintaining genomic integrity by acting as a signal transducer of DNA damage. Chk2 is endogenously activated in precancerous lesions with genomic instability and in cancer cells grown in culture. Studies have shown that the activated Chk2 acts as a survival factor for cancer cells. There are several rationales for the development of Chk2 inhibitors. First, wide range of anticancer drugs and ionizing radiation caused activation of Chk2 in tumor cells which seriously limited their effect in these cells [20-26]. These shortcomings of current chemotherapeutic approaches can be addressed by selective inhibition of Chk2. Second, inhibition of Chk2 in normal cells leads to protection normal tissues during chemotherapy or radiation therapy that increase the therapeutic indices of DNA-targeted and ionizing radiation(IR) agents in these cells. Therefore, Chk2 inhibitors would protect healthy tissues as well as sensitize the tumor to chemotherapy. Therefore, inhibition of Chk2 can address limitations of current cancer therapies and rationalize the basis of using Chk2 inhibitors as chemotherapeutic agents [27-30].

2-Arylbenzimidazole derivatives as 2-(4-(4-hydroxyphenylthio) phenyl)-1H-benzo[d]imidazole-5-carboxamide (1) (Fig. 1A) are defined as potent and selective inhibitors of Chk2 [31-33] and 4fluoro-2-(4-{[(3S,4R)-4-(2-hydroxypropan-2-yl)pyrrolidin-3-yl] amino}-6,7-dimethoxyquinazolin-2-yl)phenol (XBJ) (Fig. 1B) were described as a new series of potent and selective ATP-competitive inhibitors of Chk2 that was generated on the basis of structurebased design [34]. A series of pyrazole-benzimidazole conjugates were developed by Galal et al. on the basis of structure based design such as N-isopropyl-2-(4-(3-methyl-4-nitro-1-phenyl-1H-pyrazol-5-yloxy)phenyl)-1H-benzo[d]imidazole-5-carboxamide (2)(Fig. 1C) [35] and 2-(4-(4-((2-carbamoylhydrazono)methyl)-3methyl-1-phenyl-1H-pyrazol-5-yloxy)phenyl)-5-nitro-1H-benzo [d]imidazole (3) [36] exhibiting high potency as Chk2 inhibitors (Fig. 1D).

On the other hand, there are great numbers of antitumor active agents possessing the pyrimidine nucleus [37–41] such as pyrimidinylmethyl benzimidazole derivatives (I) (Fig. 2) which exhibited potent antitumor effect against panel of cancer cell lines [38] and AZD6738 (Fig. 1E) is orally active anti-tumor single agent acting as selective ATR inhibitor in phase I clinical trial and used in cancer treatment as a monotherapy or in combination with chemo/ radiation therapy in patients with solid tumors [40,41].

The present study aimed to synthesize new candidates of 2arylbenzimidazoles, on the basis of molecular hybridization of 2arylbenzomidazoles and pyrimidines, guided by molecular docking studies, while assessing their biological activities not only as Chk2 inhibitors and antitumor agents, but also in combination with cisplatin or doxorubicin.

2. Results and discussion

2.1. Rational design

Although, 2-arylbenzimidazoles represent one of the most potent selective class of Chk2 inhibitors and showed effective radioprotection of human T-cells subjected to ionizing radiation during radiotherapy [31–33], their antitumor effect is not reported and so they cannot be used either as single agent in monotherapy treatment or in combined chemotherapies where they do not potentiate the antitumor effect of the genotoxic anticancer drugs [42]. To address their shortage and following the same strategy used in the formation of pyrazole-benzimidazole conjugates [35,36] which succeeded in the synthesis of new candidates of 2-arylbenzimidazoles with effective anticancer properties and potentiated antitumor effect of genotoxic drugs [35,36] that formed on the basis of hybridization two different bioactive molecules with complementary pharmacophoric functions or with different mechanisms of action often exhibited enhanced effects [43–45].

A small library 212 of pyrimidine-benzimidazole conjugates were designed by replacing the lateral aryl group of 2-arylbenzimidazoles (II) with pyrimidine moieties in formula I (Fig. 2). Most of the designed pyrimidine-benzimidazole conjugates of formula III (Fig. 2) revealed high binding affinities towards Chk2 receptor (PDB code: 2XBJ) [34] according to MOE program (supplementary material: Table 2S). Twelve synthetically feasible compounds were chosen among the best fifty candidates that exhibiting the highest binding affinities into Chk2 for chemical synthesis. Variation of the substituents permitted the deduction of structure-activity relationship. Finally, their possible interactions with Chk2 were investigated by docking using the MOE program into the crystal structure of Chk2 in complex with the potent and selective 2-(quinazolin-2-yl)phenol inhibitor (PDB code: 2XBJ) [34].

2.2. Chemistry

The target pyrimidine-benzimidazole conjugates were synthesized via straight forward synthetic routes as shown in Schemes 1–3. They were classified into three series according to the linker (X) between 2-phenylbenzimidazoles and pyrimidines moieties (Fig. 2). The desired pyrimidines derivatives 7a-g have been achieved synthesis of 6-aryl-4-oxo-2-thioxo-1,2,3,4bv tetrahydropyrimidine-5-carbonitriles (4a-f) that were prepared following a previously described protocol [38,39]. The intermediate compounds, ethyl 2-(5-cyano-4-(4-florophenyl)-6-oxo-1,6dihydropyrimidin-2-ylthio)acetate (5) and ethyl 2-(5-cyano-4-(4chlorophenyl)-6-oxo-1,6-dihydropyrimidin-2-ylthio)acetate (6). were obtained respectively by alkylation of compounds 4b and 4c with ethyl bromoacetate in benzene and potassium carbonate. Chloropyrimidine derivatives 7a-g were obtained through treatment of hydroxypyrimidine derivatives 4a, 5, 4c, 4d, 4e, 4f and 6, respectively, with phosphorus oxychloride (Scheme 1).

The first series of target compounds **9a-c** and **10a-c** with ether linker (X = O, Fig. 2) was obtained by reaction of **7a-c** with 4hydroxybenzaldehyde in DMF, potassium hydroxide and copper (I) iodide that yielded phenyl-pyrimidinyl ether derivatives **8a-c** in good yield. Subsequently, reaction of 3,4-diaminobenzoic with compounds **8a-c** in DMF and sodium thiosulphate afforded acid derivatives of 2-arylbenzimidazoles (**9a-c**). Finally, coupling of 2arylbenzimidazoles **9a-c** with ammonium carbonate in DMF and CDI yielded amides of analogues of 2-arylbenzimidazoles **10a-c** (Scheme 2).

The second series of pyrimidine-benzimidazole conjugates







E:

AZD6738

Fig. 1. A-E: Reported potent Chk2 inhibitors (A–D) and AZD6738, potent antitumor active agent with selective inhibition against ATR (E).



Fig. 2. Rational design of the target pyrimidine-benzimidazole conjugates via hybridization of effective antitumor active agents (I) and selective Chk2 inhibitors (II).



Conditions: i) K₂CO₃, ethanol, reflux, 6h, ii) ethyl bromoacetate, K₂CO₃, benzene, reflux, 3h, iii) POCl₃, stirring, 50 °C, 3h,

Scheme 1. Synthetic routes of chloropyrimidine derivatives 7a-g.

12–15 (Scheme 3) with amino linker (X = NH; Fig. 2) were obtained by reaction of chloropyrimidines **7d,e** with 4-aminobenzoic acid to form pyrimidinyl aminobenzoic acid derivatives **11a,b** in the presence of DMF, potassium hydroxide and catalytic amount of copper (I) iodide. Then, 3,4-diaminobenzoic acid was reacted with compounds **11a,b** in polyphosphoric acid to form acid derivatives of 2arylbenzimidazoles **12** and **13**, respectively. Finally, coupling of **12** and **13** with ammonium carbonate in the presence of CDI and DMF yielded the target amides of pyrimidine-benzimidazole conjugates **14** and **15**, respectively (Scheme 3).

The third series of pyrimidine-benzimidazole conjugates **16–23** with thio-ether linker (Scheme 3) obtained via similar steps that started by reaction of chloropyrimidines **7c**, **7e**, **7f** and **7g** with 4-

mercaptobenzoic acid to form phenyl pyrimidine thio-ether derivatives **11e**, **11c**, **11d**, and **11f**, respectively, in the presence of potassium hydroxide, catalytic amount of copper (I) iodide and DMF. Thio-ether derivatives **11c-f** were reacted with 3,4diaminobenzoic to afford pyrimidine-benzimidazole conjugates of thioether linker **16–19** which followed by formation of the target amides analogues **20–23** via coupling with ammonium carbonates (Scheme 3).

The synthesized compounds were characterized by physicochemical and spectral means. The MS, ¹HNMR, ¹³CNMR spectral data and elemental analyses were found in accord with the assigned molecular structures.



Conditions: i) 4-hydroxybenxalehyde, KOH, DMF, CuI, 70 °C, 2h, ii) 3,4diaminobenzoic acid, Na₂S₂O₅, DMF, reflux, 24 h, iii) (NH₄)₂CO₃, CDI, DMF, stirring, r.t., 24 h.

Scheme 2. Synthetic routes of pyrimidine-benzimidazole conjugates 9a-c and 10a-c.

2.3. Evaluation of the efficacy of target compounds as Chk2 inhibitors

The checkpoint Kinase Assay was used to demonstrate the chemical inhibition of Chk2 function by compounds 9a-c, 10a-c, 14, 15 and 20–23. The IC₅₀ values of synthesized compounds as Chk2 inhibitors in addition to 2-(4-(4-hydroxyphenylthio)phenyl)-1Hbenzo[d]imidazole-5-carboxamide (1) as lead compound have been evaluated (Table 1). The activity of synthesized compounds has not been examined against other kinases. The determination of IC₅₀ of studied compounds against Chk2 function profile, the newly synthesized 2-arylbenzimidazoles are effective potent inhibitors against Chk2 (Table 1) with IC₅₀ in a range from 5.56 nM to 46.20 nM. Six candidates (9b, 23, 9a, 14, 9c and 20) out of twelve synthesized 2-phenylbenzimidazoles were found to be more potent than the lead compound 1 (IC₅₀ = 32.40 nM). Compounds **9b** and **23** are the most potent derivatives (IC₅₀ \approx 5.55 nM). Compounds 10a, 10b, 21, 15 and 10c exhibited comparable inhibitory effect with the lead compounds. In the first series with ether linker, acid conjugates **9a-c** are more potent than the amides analogues 10a-c. In the second series with amino linker, compound 14 is more effective than compound 15. In the third series with thioether linker, compound 22 (IC₅₀ = 46.20 ± 3.44 nM) showed the least potency among the studied compounds. The order of activity is as follows: 23 = 9b, 9a, 14, 9c, 20, 10a, 10b, 21, 15, 10c and finally 22 (Table 1).

2.4. Cytotoxicity of the synthesized Chk2 inhibitors and their combinations with doxorubicin and cisplatin

The cytotoxicity of the studied compounds 9b, 10a-c, 14, 15 and

20-23 in addition, to two reference compounds (as a positive control) cisplatin and doxorubicin were tested on human breast cancer cell line (MCF7) using sulphorhodamine-B (SRB) assay [46] (Table 2). Compounds 9b, 10a-c, 14, 15 and 20–23 showed potential cytotoxicity on MCF7. The compounds **14** ($GI_{50} = 6.6 \pm 0.047 \mu M$), **15** (GI₅₀ = 7.1 \pm 0.01 μ M) and **23** (GI₅₀ = 6.2 \pm 0.07 μ M) showed higher cytotoxic effect than doxorubicin ($GI_{50} = 8.4 \pm 0.04$) while, a comparable toxicity to cisplatin was observed ($GI_{50} = 6.8 \pm 0.04$). On the other hand, compounds **9b**, **10a-c** and **20–22** exhibited less potential cytotoxity than the two reference drugs, doxorubicin and cisplatin with GI₅₀ in range from $(11.7 \pm 0.07 \,\mu\text{M} \text{ to } 24.9 \pm 0.1 \,\mu\text{M})$. Preliminary structure activity relationships can be evaluated from the obtained results as follow: compound **9b** ($GI_{50} = 24.9 \pm 0.10$) is less cytotoxic than its amide analogue **10b** ($GI_{50} = 14.6 \pm 0.06$). For compounds **10a** ($GI_{50} = 20.2 \pm 0.02$) and **10c** ($GI_{50} = 11.7 \pm 0.07$), the presence of Cl in 10a negatively affected on the activity. For amides of series 2, compounds 14 and 15 exhibited the highest activity among the studied compounds. For compounds 20-23 of series 3, compound **23** ($GI_{50} = 6.24 \pm 0.07$) is the most active compounds with obvious higher activity than its analogue 22 $(GI_{50} = 14.6 \pm 0.03)$ that revealed the importance of presence of ethyl acetate group on thiol of pyrimidine moiety. For analogues 20 $(GI_{50} = 11.8 \pm 0.05)$, **21** $(GI_{50} = 15.0 \pm 0.03)$ and **22** exhibited comparable cytotoxic effect against MCF-7. Studying of SAR indicated that acid conjugate 9b is less effective than its amide analogue 10b. Ethyl thioacetate group is highly effective on the activity as observed in compound 23. Compounds with amino linker as in compounds 14 and 15 are more effective as cytotoxic agents than that with ether or thio-ether linkers.

Resistance of the tumor to the used chemotherapy is a common phenomenon that can result in rapid disease progression during or



Conditions: i) KOH, DMF, CuI, 70 °C, 2h, ii) 3,4-diaminobenzoic acid, polyphosphoric acid, 150°C, 4h, iii) (NH₄)₂CO₃, CDI, DMF, stirring, r.t., 24 h.

Scheme 3. Synthetic routes of pyrimidine-benzimidazole conjugates 12–23.

shortly after completion of treatment. One of the treatment modalities to overcome the impact of drug resistance is the concurrent use of two or more chemotherapy agents with different mechanisms of action and different modes of drug resistance. Therefore, one of the main goals of the present study was to investigate the combined effect of synthesized compounds **9b**, **10a-c**, **14**, **15**, and **20–23** with genotoxic drugs as cisplatin and doxorubicin. In this study, combination effect was expressed as a potentiation index (PI), which is the ratio of GI₅₀ for the Chk2 inhibitor alone and GI₅₀ for the Chk2 inhibitor in combination with anticancer drug (Table 2). The studied compounds **10a-c**, and **14**, **15** potentiated the cytotoxic effect of both doxorubicin and cisplatin (Table 2) with dramatic decrease in their IG₅₀. Compound **10b** exhibited the best potentiation effect with the highest PI whereas acid conjugate **9b** and compounds of series 3 with thio-ether linker **20–23** antagonize the cytotoxic effect of both drugs.

The foregoing results indicated that the synthesized compounds are effective as Chk2 inhibitors as well as effective antitumor

Table 1Investigation of the activity of pyrimidine-benzimidazole conjugates 9a-9c, 10a-10c,14, 15 and 20-23 as Chk2 inhibitors.

Compound#	IC ₅₀ (nM)	Compound#	IC ₅₀ (nM)
9a	15.35 ± 4.36	14	25.80 ± 2.99
9b	5.56 ± 6.80	15	33.40 ± 3.27
9c	27.88 ± 6.66	20	28.00 ± 7.12
10a	31.00 ± 4.22	21	35.90 ± 9.41
10b	32.40 ± 2.07	22	46.20 ± 3.44
10c	33.80 ± 8.08	23	5.55 ± 4.09
1	32.40 ± 6.17		

1 = Lead compound; 2-(4-(4-hydroxyphenylthio)phenyl)-1*H*-benzo[*d*]imidazole-5-carboxamide.

 IC_{50} : The concentration of synthesized compound that inhibits Chk2 by 50%. IC_{50} values expressed in nM \pm SD, all values are the mean of at four replicate experiments.

Table 2

Cytotoxicity (GI₅₀) of studied compounds **9b**, **10a-c**, **14**, **15** and **20–23** alone and their combination with genotoxic drugs as Doxorubicin and Cisplatin against MCF7 cells.

Compound	$GI_{50}\left(\mu M\right)$	combination $GI_{50}\left(\mu M ight)$		PI	
		Doxorubicin	Cisplatin	Doxorubicin	Cisplatin
9b	24.90 ± 0.10	>100	>100	<1	<1
10a	20.25 ± 0.02	0.1 ± 0.02	0.06 ± 0.004	202.00	336.60
10b	14.60 ± 0.06	0.04 ± 0.02	0.04 ± 0.01	365.00	365.00
10c	11.70 ± 0.07	0.08 ± 0.08	0.1 ± 0.06	146.25	117.00
14	6.60 ± 0.04	0.05 ± 0.041	ND	132.00	ND
15	7.13 ± 0.01	0.09 ± 0.01	0.05 ± 0.05	78.80	142.00
20	11.80 ± 0.05	>100	>100	<1	<1
21	15.00 ± 0.03	>100	>100	<1	<1
22	14.61 ± 0.03	>100	>100	<1	<1
23	06.24 ± 0.07	>100	>100	<1	<1
DOX	8.40 ± 0.04				
CIS	6.82 ± 0.04				

Each value is the mean \pm SD of 5 independent experiments performed in triplicates; the surviving fraction = 0.D. (treated cells)/0.D. (control cells). [48].

 IG_{50} : the concentration that inhibited growth by 50% relative to untreated cells. PI: the ratio of the mean GI_{50} for the ChK2 inhibitor alone and the mean GI_{50} for the ChK2 inhibitor in combination with a cytotoxic agent.

The statistical significance of the results was analyzed using one-way ANOVA followed by Tukey multiple comparison test.

activity. The checkpoint kinase 2 (Chk2) is a critical enzyme involved in the DNA damage-response pathway, which is activated by phosphorylation prompting cellular response such as DNA repair, cell-cycle regulation or apoptosis. Moreover, inhibition of Chk2 is thought to sensitize p53-mutated or p53-deficient cancerous cells leading to cell death [47]. Due to their multiple pharmacological effects, the IC_{50} (Table 1) that measure the degree of kinase inhibition of the synthesized compounds is different from IG₅₀ that is associated with cell death which may not confined by only one pathway (Table 2). Their activities as Chk2 inhibitors result from their binding to Chk2 receptor via interaction with its active sites as reported in the literature data bases of 2arylbenzimidazoles class. So, we can deduce that there are different modes of actions controlling their inhibitory effect against Chk2 and their cytotoxicity on MCF-7 cells, which reflect the success of the hybridization of pyrimidine and 2-arylbenzimidazole in the production of molecules with complex modes of action that impact our multiple targets simultaneously that are better in controlling complex disease systems, less disposed to drug resistance and being the standard of care in cancer treatment [43,45].

2.5. Cellular inhibitory effect of synthesized compounds against Chk2 alone and in combination with genotoxic anticancer drug

Several kinases in the DNA-repair pathway are activated after cells are exposed to cisplatin. Among the two kinases involved in checkpoint activation, namely ATM and ATR, cisplatin preferentially activates ATR kinase [47,48], which phosphorylates p53 at serine-15 to initiate activation of the p53 protein [49]. Also, cisplatin activates Chk2, but the effect of cisplatin on Chk2 appears to be independent of ATM [48,50,51].

The chemical evaluation of the synthesized compounds proved their potency as Chk2 inhibitors. It is important to determine their cellular inhibitory effects against Chk2 in cancer cells alone and in the combination with cisplatin. To address our target MCF-7 cells were exposed to single agent treatment with IG_{50} of cisplatin, **9b**, **10b**, or their combination (IG₅₀ of cisplatin $+0.1\mu$ M of studied compound, 9b or 10b) (Fig. 3). Treated and untreated MCF-7 cells (negative control) were harvested, and proteins were extracted, separated on SDS-PAGE gel, transferred onto NC membrane and then probed with phosphorylated anti-Chk2 antibody [52,53]. The western blot reaction showed the immunogenic band recognized phosphorylation (activation) of Chk2 at 64.8 kDa which was observed in the untreated cancer cells (lane 4) and the cells treated with cisplatin alone (lane 3) and in combination of compound 9b (lane 2) (Fig. 3). A complete disappearance of the immunogenic band of Chk2 phosphorylation reaction (activation process) in the cases of treating cells with IG₅₀ of compounds **9b** (lane 1), **10b** (lane 5) and the combination of **10b** with cisplatin (lane 6), was observed. These results suggest that the expression of Chk2 protein is induced by cisplatin [49,50] and its combination with 9b whereas it was completely inhibited in combination of cisplatin and compound 10b (Fig. 3). The specific blockade of Chk2 activity by compounds 9b and 10b as well as combination of 10b with cisplatin in breast cancer cells proved that both compounds both 9b and 10b alone exhibited chemical inhibition of Chk2 activity but also the combined treatment of **10b** with cisplatin potentiates the kinase inhibitory effect of compound **10b** that suggest this combination may be important therapy for resistant types of tumors. Interestingly, combination of 9b and cisplatin induced expression of Chk2 which indicated that the combined treatment of 9b and cisplatin antagonized the inhibitory effect of 9b which in accord with the observed antagonism effect for antitumor activity on MCF-7 cells in the potentiation assay.



Fig. 3. Western Blot analysis showing reactivity of different samples on MCF-7 cells. Lane M: Pre-stained molecular weight marker, Lane 1: IG_{50} of compound **9b** (C9) Lane 2: IG_{50} of cisplatin +0.1 μ M of compound **9b** (Cis + C9b), Lane 3: IG_{50} of cisplatin (Cis), Lane 4: control, Lane 5: IG_{50} of compound **10b** (C10b), Lane 6: IG_{50} of cisplatin +0.1 μ M of compound **10b** (Cis + C10b).

2.6. Effect of synthesized Chk2 inhibitors on apoptosis

The apoptotic effect was studied to further investigate antitumor activity of synthesized compounds. The experiments included untreated cells (control cells), treated MCF-7 cells with GI_{50} of cisplatin, compound **9b** or **10b** alone or combination of GI_{50} of cisplatin with 0.1 μ M of synthesized conjugates **9b** or **10b** (Fig. 4A–G). As shown in Fig. 4F, combination of compound **10b** with cisplatin had the best apoptotic effect on the MCF-7 cells while cells treated with compound **9b** and cisplatin (Fig. 4D) showed the worst anti-apoptotic effect compared to the untreated cells (Fig. 4A). Regarding the MCF-7 cells treated with IG₅₀ of cisplatin, compounds **9b** and **10b**, the number of the living cells were 207 ± 8.54 (Fig. 4B), 284 ± 6.91 (Figs. 4C) and 232 ± 18.69 cells (Fig. 4E), respectively. These results can be summarized in Fig. 4G to reflect the antitumor activity of compounds **9b** and **10b** as well as



G: Apoptotic effect of compounds 9b and 10b and their combination with cisplatin MCF-7



Fig. 4. A-G: Effect of synthesized Chk2 inhibitors on apoptosis of MCF7 cells. **A**, untreated cells (control). **B**, Gl₅₀ of cisplatin alone. **C**, Gl₅₀ of compound **9b**. **D**, Gl₅₀ of cisplatin in combination with 0.1 μM of compound **9b**. **E**, Gl₅₀ of compound **10b**. **F**, Gl₅₀ of cisplatin in combination with 0.1 μM of compound **10b**. **G**, Gl₅₀ of cisplatin in combination with 0.1 μM of compound **10b**. **G**, Gl₅₀ of cisplatin in combination with 0.1 μM of compound **10b**. **G**, antification of the apoptotic effect by calculating mean number of living cells in 5 point of dimension 1700 μm for each of stained with propidium iodide by using confocal microscope (Zeiss LSM 710), the objective lens is 20X and magnification 200X.



Fig. 5. A-E: Effects of compound **10b** on drug-induced cell cycle arrest in MCF-7 breast cancer cells. **A**, effects of cells treated with DMSO (control cells). **B**, effect of cells treated with GI₅₀ of doxorubicin alone. **C**, effects of cells treated with GI₅₀ of compound **10b**. **D**, effects of cells treated with GI₅₀ of doxorubicin in combination with 0.1 µM of compound **10b**. Histograms, cell cycle distribution were assessed by propidium iodide DNA staining as shown in Materials and Methods. **E**, quantification of the effects for total cells with studied concentrations of compound **10b** alone and in combination with doxorubicin-induced cell cycle arrest in MCF-7 cells at 48 h.

the potentiation effect of compound **10b** to the cytotoxicity of cisplatin in addition to the antagonism effect of compound **9b** to cisplatin.

2.7. Cell cycle analysis

Flow cytometry was used to check the activity of the synthesized Chk2 inhibitors on the cell cycle phases. The experiments were carried out on MCF-7 breast carcinoma cell line, including untreated cells (control cells), treated cells with GI₅₀ of compound 10b, treated MCF7 cells with GI₅₀ of doxorubicin alone or GI₅₀ of doxorubicin in combination with $0.1 \,\mu M$ of compound **10b** (Fig. 5A–E). The analysis of control cells revealed that untreated cells showed the expected pattern for continuously growing cells and the existences of 75.09% of cells were in G1 phase, 17.24% in Sphase and 7.67% in G2/M phase (Fig. 5A). While treating cells with GI₅₀ of doxorubicin, the number of the arrested cells in S-phase increased (21.3%) and non-significant changes in the number of cells in phases of G1 and G2/M by comparison to control cells (Fig. 5B). Comparing the effect of compound 10b with untreated control, compound **10b** alone arrested MCF7 cells at G2/M by 11.19% and almost no changes at G0/G1 phase (72.08%) or S phase(16.73%) of the cell cycle (Fig. 5C). These results suggest that the G₂-M phase accumulation was caused by an increase in the duration of the G2-M phase. The combination of doxorubicin with compound 10b caused obvious change on the cell cycle distribution by comparison to the effect of both doxorubicin or compound **10b** with significant arrest in S phase (43.8%) and a subsequent decrease in G1 to 56.6% and complete absence of cells in G2-M (Fig. 5D). This result suggests not only S phase accumulation but almost complete arrest of many cells in the S phase. A summary of the cell cycle distribution found in MCF-7 carcinoma cells treated with compounds 10b and combination with doxorubicin with the studied concentrations compared to control cells at 48 h (Fig. 5E).

2.8. Molecular modeling

The molecular modeling of the target compounds **9a-c**, **10a-c**, **14**, **15** and **20–23** using MOE (molecular modeling environment) program was studied to get deeper insight into the molecular bases of the inhibitory potency and to find out the interaction between the new synthesized pyrimidine-benzimidazole conjugates and Chk2 receptor.

Molecular modeling calculations and local docking were done using MOE [54] to evaluate the binding free energies of these inhibitors into the target to Chk2 kinase receptor (PDB: (2XBJ).

2.8.1. Validation of the docking performance and accuracy

To validate the docking accuracy of the program used, docking of the native co-crystallized XBJ (4-fluoro-2-(4-{[(3S,4R)-4-(2-hydroxypropan-2-yl)pyrrolidin-3-yl]amino}-6,7-dimethoxyquinazolin-2-yl)phenol) ligand of Chk2 kinase (PDB: (2XBJ) into the binding sites. The docked pose was docked super-imposed on the native co-crystalized ligand on its binding sites with a root mean square deviation (RMSD) of 1.8 Å, S-score of -19.32 and bound by five hydrogen bonds (Fig. 6A–B).

2.8.2. The binding affinities of the synthesized compounds into ChK2 receptor

All synthesized compounds were docked into same groove of the binding site of the native co-crystalized XBJ (4-fluoro-2-(4-{[(3S,4R)-4-(2-hydroxypropan-2-yl)pyrrolidin-3-yl]amino}-6,7dimethoxyquinazolin-2-yl)phenol) ligand of Chk2 kinase into the binding sites and for docking calculations, the protein structure (PDB: (2XBJ) was first separated from the inhibitor molecule and refined using energy minimization with added hydrogens. Docking calculations were carried out using standard default variables for the MOE program [49].

The binding affinities were evaluated on the basis of the binding free energy S-score and hydrogen bonds with their distance between the designed compounds and the amino acids in the receptor (Table 3).

All the synthesized compounds **9a-c, 10a-c, 14, 15** and **20–23** gave higher score than the Lead compound; 2-(4-(4-hydroxyphenylthio)phenyl)-1*H*-benzo[*d*]imidazole-5-

carboxamide (1) which docked into Chk2 with S-score fitness in a range from (-15.95 kcal/mol) to (-20.01 kcal/mol) (Table 3). Compounds **9b**, **9a**, and **23** exhibited the highest score fitness of -20.01, -18.73 and -18.07, respectively (Table 3).

The active-site of amino acids residues in Lead compound formed one H-bond with GLU 308 and in the native ligand formed three hydrogen bonds with amino acids residue GLU 308, GLU 351 and MET 304 beside two H-bonds with NO₃ 1516. The synthesized compounds **9a-c**, **10a-c**, **14**, **15** and **20–23** were bound into the binding site through up to two hydrogen bonds, to a higher extent with LYS 349, LYS 294, MET 304, NO₃ 1516 and GLY 232, and to a lower extent with ASN 269 (C=O), GLU 273, GLU 305 and ASP 368 amino acids (Table 3).

Compound **9b** showed the highest binding free energies (–20.0135 kcal/mol) among the studied compounds (Table 3). It also bound into ChK2 kinase by two hydrogen bonds into the binding site through its carboxylic group at position 5 with NH of Met 304 and its terminal cyano group with NH of LYS 249 (Fig. 7). The docking results for compound **9b** are highly correlated to its inhibitory activity against ChK2 kinase with highest activity among the synthesized compounds (IC₅₀ = 5.56 ± 6.80 nM).

Compound **10b** exhibited S-score fitness into Chk2 kinase receptor (-16.01 kcal/mol) comparable to lead compound **1** as well as they exhibited similar inhibitory activity against ChK2 $(IC_{50} = 32.40)$ (Table 1). Compound **10b** also bound into the binding site of the receptor by two hydrogen bonds with MET 304 and LYS 349 (Fig. 8A–B) (Table 3). Compound **10b** and lead compound **1** revealed lower inhibitory activity than compound **9b** (Table 1).

Compound **9a** showed binding free energy (-18.73 kcal/mol) higher than lead compound (-15.56 kcal/mol). Also, **9a** showed better interaction than lead compound **1** by forming 2 H-bonds with the receptor binding site moieties MET 304 and LYS 249 and carboxylate at position 5 and terminal cyano of **9a**, respectively (Fig. 9) compared to only 1 H-bond between the lead compound and GLU308. The docking results for compound **9a** is correlated to its inhibitory activity against ChK2 kinase (IC₅₀ = $15.35 \pm 4.36 \text{ nM}$) which is also higher than lead compound. The plausible binding modes and the docking orientations for compounds **9a**, **9b** and **10b** superimposed in the binding site of Chk2 receptor that are surrounded by a Gaussian contact surface are represented in Fig. 10.

2.8.3. Structure-activity relationship

Structure-activity relationship was deduced from the forgoing results on the basis of molecular modeling by getting deeper insight into the molecular bases of the inhibitory potency (Table 1) and by correlating the interaction of pyrimidine-benzimidazole conjugates **9a-c**, **10a-c**, **14**, **15**, **20**–**23** and Chk2 receptor by the change in the binding free energy according to change of substituents (Table 3). Binding free energy score is a combination of different scoring functions includes H-bonds, ionic interaction, hydrophobic interaction, metal ligation and the interaction between hydrophobic and polar atoms. In the first series (X = O), the binding free energies of the acid derivatives, **9a-c**, are higher than the amide analogues **10a-c** (Table 3) which in accord with the observed higher inhibitory effect of the acid derivatives, **9a-c**, than the amide analogues **10a-c**



Fig. 6. A-B: Proposed binding mode for the interaction of the native co-crystallized ligand with Chk2 receptor (PDB: 2XBJ), five hydrogen bonds are formed with GLU 308, GLU 351, NO₃ 1516, MET 304, NO₃ 1516 (violet dotted lines), (A: 2D and B: 3D). (For interpretation of the references to color in this figure legend, the reader is referred to the Web version of this article.)

Table 3

Molecular Operating Environment (MOE) [54] flexible docking results for compounds **9a-c**, **10a-c**, **14**, **15** and **20–23** docked by (MOE) 2008.10 into Chk2 kinase (PDB: (2XBJ) in comparison to the co-crystallized XBJ ligand.

Compound	S- score ^a (kcal/mol)	Hydrogen bonds between atoms of compounds and amino acids of receptor				
		Compounds	receptor		Туре	Distance (A)
		Atoms	Atoms	Residues		
XBJ ^b	-19.32	H 4502	OE 1322	GLU 308	H-don	2.03
		H 4512	0 2049	GLU 351	H-don	2.23
		N 4511	02 4562	NO3 1516	H-don	2.86
		O 4539	N 1262	MET 304	H-acc	2.79
		N 4511	02 4562	NO3 1516	H-acc	2.86
1 ^c	-15.57	H 4964	OE 1322	GLU 308	H-don	1.19
9a	-18.73	O 5152	N 1262	MET 304	H-acc	1.49
		N 5153	NZ 2026	LYS 349	H-acc	2.79
9b	-20.01	O 5152	N 1262	MET 304	H-acc	1.51
		N 5211	NZ 2026	LYS 249	H-acc	3.00
9c	-16.43	O 5152	01 4489	NO3 1516	H-acc	1.92
		O 5152	01 4489	NO3 1516	H-don	1.92
10a	-16.95	N 2403	02 2214	NO3 1516	H-acc	2.55
10b	-16.01	H 5117	0 1267	MET 304	H-don	1.59
		O 5309	NZ 2026	LYS 349	H-acc	2.88
10c	-16.22	H 2418	01 2213	NO3 1516	H-don	2.57
14	-17.26	N 2387	N 150	GLY 232	H-acc	2.85
		0 2415	ND 338	ASN 269	H-acc	3.13
15	-16.42	H 2422	OE 365	GLU 273	H-don	1.90
		N 2383	N 150	GLY 232	H-acc	2.52
20	-17.50	H 2422	0 620	GLU 305	H-don	1.55
		O 2409	NZ 987	LYS 349	H-acc	2.79
21	-16.52	O 2418	N 1122	ASP 368	H-acc	2.88
22	-15.95	N 2386	N 150	GLY 232	H-acc	2.99
		N 2403	02 2214	NO ₃ 1516	H-acc	2.60
23	S	H 5013	O 298	ALA 230	H-don	2.25
		N 4998	N 1262	MET 304	H-acc	2.98

^a Binding free energy.

^b 4-Fluoro-2-(4-{[(35,45)-4-(1-hydroxy-1-methylethyl)pyrrolidin-3-yl]amino}-6,7-dimethoxy-quinazolin-2-yl) phenol.

^c Lead compound: 2-(4-(4-hydroxyphenylthio)phenyl)-1*H*-benzo[*d*]imidazole-5-carboxamide (1).

(Table 1). For the acid candidates **9a-c**: compound **9b** is one of the most potent compound with the lowest $IC_{50} = 5.56$ nM and the highest binding free energy (S-score = -20.01 kcal/mol, Table 3). Decreasing of binding free energy by substitution with chlorine in **9c** (S-score = -16.43 kcal/mol) was observed and negatively affected the inhibitory effect ($IC_{50} = 27.88 \text{ nM}$) compared to the unsubstituted **9a** (S-score = -18.73 kcal/mol, IC₅₀ = 15.35 nM). For the amide analogues 10a-c, comparable activities were detected $(IC_{50} \approx 32 \text{ nM})$ as well as comparable free energies (Sscores ≈ -16 kcal/mol) were found. For amides of the second series (compounds **14** and **15**, X = S), comparing the effect of the nitro group in compound 14 to that of the methoxy group in compound 15, it can be deduced that the presence of the nitro group was more favorable for activity ($IC_{50} = 25.80 \text{ nM}$) and led to enhancement of both, the binding free energy (S-score = -17.26 kcal/mol) and Hbonding interaction with the receptor, $(IC_{50} = 33.40 \text{ nM}, \text{ S-})$ score = -16.42 kcal/mol). For the third series (**20–23**, X = NH), compound **23** exhibited the highest activity with a remarkable decrease in the IC_{50} (5.55 nM) (Table 1) and an increase in the binding free energy (S-score = -18.07 kcal/mol, Table 3) with respect to its compound 22, which exhibited the least potency among the studied compounds $(IC_{50} = 46.20 \text{ nM}, \text{ Table 1})$ and lowest S-score that reflecting the importance of ethyl thioacetate as substituent. This can be attributed to strong hydrophobic interaction with the receptor and H-bonding interaction with its carbonyl group which monopolizes the activity, no matters the compound was an acid with the ether linker as **9b** or an amide with amino linker as 23, or the substituent on the phenyl ring was floro or chloro. For analogues 20, 21 and 22, the OCH₃ substitution was favored as compound **20** was more active than **21** having CH₃ substitution, or Cl in 22, which have negative effect on activity and

causes a great decrease in the activity of **22**, which can be correlated with the increase of binding free energy in case of substitution with the methoxy group with respect to both methyl and chloro as substituents in compounds **21** and **22**, respectively.

3. Conclusion

The studied compounds 9b, 10a-c, 14, 15 and 20-23 exhibited potent inhibition effect against enzymatic activity of Chk2. The inhibitory effect of compounds 9b and 10b was also evaluated by western blot using anti-Chk2 which indicated that compounds 9b, **10b** and combination of **10b** with cisplatin blocked the enzymatic activity of Chk2 which was proved by complete disappearance of characteristic band of Chk2 at 64.8 KDa. Moreover, the studied compounds 9b, 10a-c, 14, 15 and 20-23 exhibited remarkable cytotoxicity against MCF-7. Compounds 10a-c, 14 and 15 potentiated the activity of both cisplatin and doxorubicin whereas compounds 9b and 20-23 antagonized their cytotoxicity. The antitumor effect of compounds **9b** and **10b** was proved also by studying their apoptotic effect. Flow cytometry indicated that combination of compound **10b** with doxorubicin blocked the cell cycle to almost complete arrest in the S phase. The foregoing results necessitated further elucidation of pyrimidine-benzimidazole conjugates.

4. Material and methods

Microanalyses, spectral data of the compounds were performed in the Microanalytical National Research Centre and Pharmaceutical faculty, Cairo University, Egypt. The IR spectra (4000-400 cm⁻¹) were recorded using KBr pellets in a Jasco FT/IR 300E



699



B:



Fig. 7. A-B: Compound **9b** exhibited good fitting with ChK2 receptor (pdb: 2XBJ) via hydrogen bonds and hydrophobic interaction. Two direct H-bonds with Met 304 and LYS 249 are formed. Green line represents the distance of the hydrogen bonds and violet dots represent the score percentage of the H-bonds interaction with the receptor (**Fig. 7A**). Docking orientation of Compound 9b into ChK2 receptor (pdb: 2XBJ) receptor with a gaussian contact surface surrounding the van der Waals surface is drawn around the binding site (**Fig. 7B**). (For interpretation of the references to color in this figure legend, the reader is referred to the Web version of this article.)

Fourier transform infrared spectrophotometer on a Perkin Elmer FT-IR 1650 spectrophotometer. Starting materials, reagents, and solvents for reactions were reagent grade and used as purchased. Chromatography solvents were HPLC grade and were used without further purification. Thin layer chromatography (TLC) analysis was performed using Merck silica gel 60 F-254 thin layer plates. Flash column chromatography was carried out using columns prepacked with $40-63\,\mu\text{m}$ silica or using standard chromatography techniques when performed on a large scale. NMR spectra were recorded on a Bruker Advance 300 MHz spectrometer. The samples were referenced to the appropriate internal non-deuterated solvent

peak. The data is given as follows: chemical shift (δ) in ppm, multiplicity (where applicable), coupling constants (J) in Hz (where applicable), and integration (where applicable). The mass spectra were recorded using a Finnigan mat SSQ 7000 (Thermo. Inst. Sys. Inc., USA) spectrometer at 70 eV. The used petroleum ether had a boiling temperature in the 60–80 °C range.

4.1. Molecular docking

The crystal structures of the Checkpoint kinase (PDB: 2XBJ) [34] in complex with its co-crystalized native ligand: (XBJ): (4-fluoro-2-



Fig. 8. A-B: The interaction of compound 10b with Chk2 receptor (pdb: 2XBJ), 2 H-bonds are formed with MET 304 and LYS 349. Green line represents the distance of the hydrogen bonds and violet dots represent the score percentage of the H-bonds interaction with the receptor (Fig. 8A). Docking orientation of Compound 10b into ChK2 receptor (pdb: 2XBJ) receptor with a gaussian contact surface surrounding the van der Waals surface is drawn around the binding site (Fig. 8B). (For interpretation of the references to color in this figure legend, the reader is referred to the Web version of this article.)

(4-{[(3S,4R)-4-(2-hydroxypropan-2-yl)pyrrolidin-3-yl]amino}-6,7dimethoxyquinazolin-2-yl)phenol), as a potent and selective Inhibitor of Checkpoint Kinase 2, was retrieved from the Protein Data Bank, http://www.rcsb.org/pdb/home/home.do. The key amino acids of the active site were identified using data in PDB sum, http://www.ebi.ac.uk/pdbsum.

- The constructed 3D structures of small library designed compounds about 200 derivatives were energetically minimized through Force field MMFF94x Optimization using gradient of 0.0001 kcal/mol.
- Validation of the MOE software by re-docking of the native cocrystalized on its binding site was done. The S-score is: -19.32 kcal/mol and RMSD is 1.8 Å



B:



Fig. 9. A-B: The interaction of compound **9a** with Chk2 receptor (pdb: 2XBJ), 2 H-bonds are formed with MET 304 and LYS 349. Green line represents the distance of the hydrogen bonds and violet dots represent the score percentage of the H-bonds interaction with the receptor (Fig. 7A). Docking orientation of Compound **9a** into ChK2 receptor (pdb: 2XBJ) receptor with a gaussian contact surface surrounding the van der Waals surface is drawn around the binding site (Fig. 7B). (For interpretation of the references to color in this figure legend, the reader is referred to the Web version of this article.)

- Docking calculations were carried out using standard default variables for the MOE program and the tested compounds were docked in the binding site of the native co-crystalized ligand.
- The Dock scoring in MOE software was done using London dG scoring function, rotatable bonds were allowed; the best 10 poses were retained and analysed.
- The binding site of the human crystal structure of Chk2 is composed of the following amino acids: Leu226, Gly227, Val234, Ala247, Ile286, Leu301, Glu302, Leu303, Met304, Glu305, Gly307, Glu308, Glu351, Asn352, and Leu354.
- The binding free energy was used to rank the binding affinity of the synthesize compounds to Chk2 receptor. In addition, in case

of compounds having simalr S-scores, hydrogen bonds between the ligand and amino acids in Chk2 receptors were used in the ranking of the compounds. Evaluation of the hydrogen bonds was done by measuring the hydrogen bond length which doesn't exceed 3 Å.

- Twelve compounds out of the best fifty were selected for chemical synthesis in this study and biological evaluations were performed as shown in the forgoing results.
- The mode of interaction of the native ligand 2XBJ receptors within their crystal structures was used as a standard docked model as well as for RMSD calculation.



Fig. 10. Docking orientations and superimposition for Compound 9a "colored in blue", compound 9b "colored in red" along with ompound 10b "colored in yellow" into ChK2 receptor (pdb: 2XBJ) receptor. A gaussian contact surface surrounding the van der Waals surface is drown around the binding site. (For interpretation of the references to color in this figure legend, the reader is referred to the Web version of this article.)

4.2. Chemistry

4.2.1. Synthesis of compounds 5 and 6

A solution of 6-(4-fluorophenyl)-4-oxo-2-thioxo-1,2,3,4-tetrahydropyrimidine-5-carbonitrile **(4b)** or 6-(4-chlorophenyl)-4-oxo-2-thioxo-1,2,3,4-tetrahydropyrimidine-5-carbonitrile **(4c)** [39] (10 mmol) in benzene (50 mL) and K₂CO₃ (1.38g, 10 mmol) was treated with ethyl bromoacetate (1.66g, 10 mmol) added dropwise. The reaction mixtures were stirred for 1 h at room temperature then were refluxed for 3 h. The solvent was evaporated under reduced pressure. The solids obtained were filtered off and were crystalized from benzene as yellow crystals.

4.2.1.1. Ethyl 2-(5-cyano-4-(4-florophenyl)-6-oxo-1,6dihydropyrimidin-2-ylthio)acetate **(5)**. Yield, (84%), m.p. 159–162 °C. ¹H NMR (300 MHz, DMSO) δ 1.03–1.12 (t, 3H, CH₃), 3.99–4.19 (m, 4H, 2CH₂), 7.39–7.49(m, 2H), 7.96–8.07(m, 2H, 2H-Ar). ¹³C NMR (75 MHz, DMSO) δ 13.81, 32.93, 61.25, 93.00, 115.45, 115.57, 115.74, 115.81, 131.28, 131.34, 131.39, 130.77, 162.41, 165.12, 165.96, 165.74, 167.91. MS, *m*/*z* (%): 333(M+, 28%), Anal. Calcd for C₁₅H₁₂FN₃O₂S (FW: 333): C, 54.05; H, 3.63; F, 5.70; N, 12.61; S, 9.62 Found: C, 54.21; H, 3.44; N, 12.54; S, 9.77.

4.2.1.2. Ethyl 2-(4-(4-chlorophenyl)-5-cyano-6-oxo-1,6dihydropyrimidin-2-ylthio)acetate **(6)**. Yield, (78%), m.p. 178–180 °C. ¹H NMR (300 MHz, DMSO) δ 1.05–1.13 (t, 3H, CH₃), 4.09–4.20 (m, 4H, 2CH₂), 7.34–7.37 (d, *J* = 8.22 Hz 2H), 7.48–7.51 (d, *J* = 8.22 Hz, 2H). MS, *m/z* (%): 351 (15%), 349 (44%) for M⁺. Anal. Calcd for C₁₅H₁₂ClN₃O₃S (FW: 349): C, 51.51; H, 3.46; Cl, 10.14; N, 12.01; S, 9.17. Found: C, 51.74; H, 3.55; Cl, 10.02; N, 12.22; S, 9.40.

4.2.2. Synthesis of compounds 7a-g

Phosphorus oxychloride (20 mL) was added dropwise to compounds **4a**, **5**, **4c**, **4d**, **4e**, **4f** or **6** (10 mmol) with stirring at r.t for 1 h and then, stirred for 3 h at 50 °C. The reaction mixtures were left to cool to room temperature and then were added to ice bath with stirring. The yellow solids obtained were filtered off, washed with petroleum ether and crystalized from dichloromethane as yellow crystals of compounds 7a, 7b, 7c-7f [38] and 7g.

4.2.2.1. Ethyl 2-(4-chloro-5-cyano-6-(4-fluorophenyl)pyrimidin-2-ylthio)acetate (7b). Yield (50%), m.p. 143–145 °C. ¹H NMR (300 MHz, DMSO) δ 1.10–1.24 (t, 3H, CH₃), 4.07–4.20 (m, 4H, 2CH₂), 7.37–7.40 (m, 2H), 8.10–8.13 (m, 2H, 2H-Ar). MS, *m/z* (%):351 (58%), 353 (19%) M⁺. Anal. Calcd for C₁₅H₁₁CIFN₃O₂S (FW: 351): C, 51.21; H, 3.15; Cl, 10.08; F, 5.40; N, 11.94; S, 9.11. Found: C, 51.09; H, 3.17; Cl, 10.22; N, 12.07; S, 9.21.

4.2.2.2. Ethyl 2-(4-chloro-6-(4-chlorophenyl)-5-cyanopyrimidin-2-ylthio)acetate (7g). Yield (64%), m.p. 166–169 °C. δ 1.05–1.13 (t, 3H, CH₃), 4.09–4.20 (m, 4H, 2CH₂), 7.54–7.57(d, J = 8.22 Hz 2H), 7.98–9.01(d, J = 8.22 Hz, 2H). MS, m/z (%): 368(19%), 366(60%) for M+. Anal. Calcd for C₁₅H₁₁Cl₂N₃O₂S (FW: 366): C, 48.93; H, 3.01; Cl, 19.26; N, 11.41; S, 8.71. Found: C, 48.77; H, 3.24; Cl, 19.00; N, 11.39; S, 8.82.

4.2.3. Synthesis of compounds 8a-c

4.2.3.1. Method A. A solution of 4-hydroxybenzaldehyde (1.22 g, 10 mmol), potassium hydroxide (2.24 g, 40 mmol), triphenylphosphine (0.524 g, 2 mmol) and cupper (I) iodide (0.19 g, 1 mmol) in DMF (20 mL) was treated with compound **7a**, **7b** or **7c** (10 mmol). The reaction mixture was heated to 70 °C and stirred under Argon for 2 h. The reaction mixture was quenched with water and extracted with ethyl acetate. The organic extracts were dried (Na₂SO4), concentrated, and purified by column chromatography (0–25% EtOAc/n-hexane) to obtain yellow powders.

4.2.3.2. 4 - (4 - for mylphenoxy) - 6 - phenyl - 2 - thioxo - 1, 2 - dihydropyrimidine-5-carbonitrile (8a). Yield (64%). m.p. 214–217 °C. ¹H NMR (300 MHz, DMSO) 7.17–7.20(d, <math>J = 8.70 Hz, 2H), 7.33–7.37(m, 1H), 7.46–7.49(m, 2H), 7.69–7.71(m, 2H), 7.88–7.91(d, J = 8.70 Hz, 2H), 10.03(s, 1H, CHO), 12.33(br., 1H, D₂O exchangeable). MS, m/z (%): 333(M+, 24%). Anal. Calcd for C₁₈H₁₁N₃O₂S (FW: 333): C, 64.85; H, 3.33; N, 12.60; S, 9.62. Found: C, 64.69; H, 3.50; N, 12.44; S, 9.75.

4.2.3.3. *Ethyl* 2-(5-cyano-4-(4-fluorophenyl)-6-(4-formylphenoxy) pyrimidin-2-ylthio) acetate (8b). Yield (64%), m.p. 153–155 °C. ¹H NMR (300 MHz, DMSO) δ 1.02–1.07 (t, 3H), 3.92–3.99(m, 4H), 7.47–7.53(m, 2H), 7.57–7.60(d, J=8.70 Hz, 2H), 8.05–8.11(m, 4H),10.07(s, 1H, CHO) . ¹³C NMR (75 MHz, DMSO) δ 13.81, 33.32, 61.14, 114.22,115.87, 116.17, 122.46, 130.96, 131.32, 131.53, 131.64, 134.31, 155.66, 167.63, 167.90, 168.77, 192.00. MS, *m/z* (%): 437(M+, 24%). Anal. Calcd for C₂₂H₁₆FN₃O₄S (FW: 437): C, 60.40; H, 3.69; F, 4.34; N, 9.61; S, 7.33. Found: C, 60.29; H, 3.65; N, 9.77; S, 7.42.

4.2.3.4. 6-(4-Chlorophenyl)-4-(4-formylphenoxy)-2-thioxo-1,2dihydropyrimidine-5-carbonitrile (8c). Yield (64%), m.p. 244–247 °C. ¹H NMR (300 MHz, DMSO) 7.33–7.37(d, J = 8.52 Hz, 2H), 7.44–7.47(d, J = 8.70 Hz, 2H), 7.89–7.91(d, J = 8.52 Hz, 2H), 8.02–8.05(d, J = 8.70 Hz, 2H), 10.07(s, 1H, CHO), 11.25(br., 1H, D2O exchangeable). MS, m/z (%): 367(M+, 28%). Anal. Calcd for C₁₈H₁₀ClN₃O₂S (FW: 367): C, 58.78; H, 2.74; Cl, 9.64; N, 11.42; S, 8.72. Found: C, 58.55; H, 2.51; Cl, 9.44; N, 11.61; S, 8.97.

4.2.4. Synthesis of compounds 9-a-c

4.2.4.1. Method B. Compounds **8a-c** (10 mmol) were treated with a solution of 3,4-diaminobenzoic acid (1.53g, 10 mmol) in DMF (30 mL), followed by Na₂S₂O₅ (2.85g, 15 mmol). The mixture was heated to 90 °C for 24 h. The reaction mixture was cooled to room temperature then was quenched with water and the solid obtained was filtered and purified by column chromatography (0–100% EtOAc/n-hexane) to obtain yellow powders.

4.2.4.2. 2-(4-(5-cyano-6-phenyl-2-thioxo-1,2-dihydropyrimidin-4-yloxy)phenyl)-1H-benzo[d]imidazole-5-carboxylic acid (**9a**). Yield (64%), m.p. 262–265 °C. ¹H NMR (300 MHz, DMSO) δ 7.27–7.35(m, 3H), 7.50–7.53(d, J = 8.70 Hz, 2H), 7.65–7.68(m, 2H), 7.70–7.73(d, J = 8.25 Hz, 1H), 7.88–7.91(d, J = 7.2 Hz, 1H), 8.23 (s, 1H), 8.32–8.35 (d, J = 8.70 Hz, 2H), 11.56 (br., 1H, D₂O exchangeable), 12.69 (br., 1H, D₂O exchangeable), 13.07 (br., 1H, D₂O exchangeable). MS, m/z (%):465 (M+, 37%). Anal. Calcd for C₂₅H₁₅N₅O₃S (FW: 465): C, 64.51; H, 3.25; N, 15.05; S, 6.89. Found: C, 64.32; H, 3.51; N, 15.21; S, 6.65.

4.2.4.3. 2-(4-(5-cyano-2-(2-ethoxy-2-oxoethylthio))-6-(4-fluorophenyl)pyrimidin-4-yloxy) phenyl)-1H-benzo[d]imidazole-5-carboxylic acid (**9b** $). Yield (67%), m.p. <math>271-273 \,^{\circ}C.$ ¹H NMR (300 MHz, DMSO) δ 0.97–1.08 (t, 3H, CH₃), 3.92–4.08 (m, 4H, 2CH₂), 7.37–7.48 (m, 2H), 7.50–7.53 (d, $J = 8.70 \,$ Hz, 2H), 7.70–7.73 (d, $J = 8.25 \,$ Hz, 1H), 7.88–7.91 (d, $J = 8.25 \,$ Hz, 1H), 8.05–8.08 (m, 2H), 8.23 (s, 1H), 8.32–8.35 (d, $J = 8.70 \,$ Hz, 2H), 10.56 (br., 1H, D₂O exchangeable), 11.57 (br., 1H, D₂O exchangeable). ¹³C NMR (75 MHz, DMSO) δ 13.86, 33.35, 61.28, 110.29, 114.87, 115.87, 115.22, 116.29, 120.87, 122.33, 128.11, 129.32, 130.96, 131.32, 131.52, 131.64, 134.31, 147.52, 148.22, 149.22, 150.22, 152.47, 156.66, 167.03, 167.89, 168.44. MS, m/z (%): 569(M+, 38%). Anal. Calcd for C₂₉H₂₀FN₅O₅S (FW: 569): C, 61.15; H, 3.54; F, 3.34; N, 12.30; S, 5.63. Found: C, 61.01; H, 3.39; N, 12.56; S, 5.77.

4.2.4.4. 2-(4-(6-(4-chlorophenyl)-5-cyano-2-thioxo-1,2-dihydropyrimidin-4-yloxy)phenyl)-1H-benzo[d]imidazole-5-carboxylic acid (**9c**). Yield (64%), m.p. 241–244 °C. ¹H NMR (300 MHz, DMSO) 7.27–7.30(d, <math>J=8.52 Hz, 2H), 7.40–7.43(d, J=8.70 Hz, 2H), 7.55–7.58(d, J=8.52 Hz, 2H), 7.70–7.73(d, J=8.25 Hz, 1H), 7.88–7.91(d, J=8.25 Hz, 1H), 8.06 (s, 1H), 8.12–8.13(d, J=8.70 Hz, 2H), 11.33 (br., 1H, D₂O exchangeable), 13.07(br., 1H, D2O exchangeable). ¹³C NMR (75 MHz, DMSO) δ 113.77, 115.44, 115.22, 116.98, 120.87, 124.21, 125.61, 129.11, 130.66, 131.96, 131.52, 131.64, 138.31, 146.31, 148.22, 149.22, 150.22, 153.00, 156.66, 164.03, 166.40, 167.89, 168.44, 179.44. MS, *m/z* (%):499 (M+,

21%). Anal. Calcd for $C_{25}H_{14}ClN_5O_3S$ (FW: 499): C, 60.06; H, 2.82; Cl, 7.09; N, 14.01; S, 6.41. Found: C, 60.31; H, 2.75; Cl, 6.85; N, 14.31; S, 6.22.

4.2.5. Synthesis of compounds 10a-c

4.2.5.1. Method C. Compounds **9a-c** (10 mmol) were dissolved in DMF (20 mL), and 1,1-carbonyldiimidazole (0.16 g, 10 mmol) was added. The mixture was stirred for 30 min at room temperature and then cooled to 0 °C. Ammonium carbonate (0.19 g, 20 mmol) was added, and the reaction mixture was stirred for 24 h at room temperature. The solvent was evaporated under reduced pressure, and the residue quenched with water (30 mL) and extracted with ethyl acetate washed with water (3 × 50 mL), and dried (MgSO4). The organic extracts were then collected and concentrated. The residue was purified by column chromatography (0–100% EtOAc/n-hexane) and obtained as yellow powder.

4.2.5.2. 2-(4-(5-cyano-6-phenyl-2-thioxo-1,2-dihydropyrimidin-4-yloxy)phenyl)-1H-benzo[d]imidazole-5-carboxamide (10a). Yield (64%), m.p. 263–265 °C. ¹H NMR (300 MHz, DMSO) δ 7.30–7.57(m, 5H), 7.66–7.69(m, 2H), 7.74–7.77(d, *J* = 8.25 Hz, 1H), 7.96–7.99(d, *J* = 8.25 Hz, 1H), 8.15 (s, 1H), 8.25–8.28 (d, *J* = 8.70 Hz, 2H), 8.54 (br., 2H, D₂O exchangeable), 11.42, (br., 1H, D₂O exchangeable). 13.01(br., 1H, D₂O exchangeable). MS, *m*/*z* (%):465 (M+, 21%). Anal. Calcd for C₂₅H₁₆N₆O₂S (FW: 464): C, 64.64; H, 3.47; N, 18.09; S, 6.90. Found: C, 64.68; H, 3.55; N, 17.94; S, 6.71.

4.2.5.3. $2-(4-(5-cyano-2-(2-ethoxy-2-oxoethylthio)-6-(4-fluorophenyl)pyrimidin-4-yloxy) phenyl)-1H-benzo[d]imidazole-5-carboxamide (10b). Yield (57%). m.p. 281–283 °C. ¹H NMR (300 MHz, DMSO) <math>\delta$ 0.98–1.22 (t, 3H, CH₃), 3.99–4.21(m, 4H, 2CH₂), 7.38–7.47(m, 2H), 7.50–7.53(d, J = 8.70 Hz, 2H), 7.69–7.72(d, J = 8.25 Hz, 1H), 7.88–7.91(d, J = 7.2 Hz, 1H), 8.01–8.12(m, 2H), 8.19 (s, 1H), 8.32–8.35 (d, J = 8.70 Hz, 2H), 8.69(br., 2H, D₂O exchangeable), 11.31(br., 1H, NH, D₂O exchangeable). MS, m/z (%): 568(M+, 22%). Anal. Calcd for C₂₉H₂₁FN₆O₄S (FW: 568): C, 61.26; H, 3.72; F, 3.34; N, 14.78; S, 5.64. Found: C, 61.12; H, 3.57; N, 14.88; S, 5.80.

4.2.5.4. 2-(4-(6-(4-chlorophenyl)-5-cyano-2-thioxo-1,2-dihydropyrimidin-4-yloxy)phenyl)-1H-benzo[d]imidazole-5-carboxamide (10c). Yield (61%). m.p. 278–280 °C. ¹H NMR (300 MHz, DMSO) 7.21–7.24(d, <math>J = 8.70 Hz, 2H), 7.43–7.46 (d, J = 8.52 Hz, 2H), 7.63–7.66 (d, J = 8.52 Hz, 2H), 7.78–7.81 (d, J = 8.20 Hz, 1H), 8.08–8.11 (d, J = 8.20 Hz, 1H), 8.26 (s, 1H), 8.32–8.35 (d, J = 8.70 Hz, 2H), 8.64 (br., 2H, D₂O exchangeable), 11.68 (br., 1H, D₂O exchangeable), 13.09 (br., 1H, D₂O exchangeable). MS, m/z (%):498 (M+, 27%), 500(M+, 8%). Anal. Calcd for C₂₅H₁₅ClN₆O₂S (FW: 498): C, 60.18; H, 3.03; Cl, 7.11; N, 16.84; S, 6.43. Found: C, 60.02; H, 3.22; Cl, 7.04; N, 16.77; S, 6.56.

4.2.6. Synthesis of compounds 11a-f

4.2.6.1. 4 - (5 - Cyano - 6 - (4 - nitrophenyl) - 2 - thioxo - 1, 2 - dihydropyrimidin-4-ylamino)benzoic acid (**11a**). Prepared according to method (A) by reaction of 4-chloro-6-(4-nitrophenyl)-2-thioxo-1,2-dihydropyrimidine-5-carbonitrile (**7d**) with 4-aminobenzoic acid. Yield (57%). m.p. 259–262 °C. ¹H NMR (300 MHz, DMSO) 4.52 (br., 1H, D₂O exchangeable), 7.24–7.27(d, <math>J = 8.70 Hz, 2H), 7.85–7.89(d, J = 8.52 Hz, 2H), 8.09–8.11(d, J = 8.52 Hz, 2H), 8.32–8.35(d, J = 8.70 Hz, 2H), 12.74 (br., 1H, D₂O exchangeable) 13.25 (br., 1H, D₂O exchangeable). MS, m/z (%): 393(M+, 28%). Anal. Calcd for C₁₈H₁₁N₅O₄S (FW: 393): C, 54.96; H, 2.82; N, 17.80; S, 8.15. Found: C, 54.78; H, 2.97; N, 17.69; S, 8.27.

4.2.6.2. 4-(5-Cyano-6-(4-methoxyphenyl)-2-thioxo-1,2dihydropyrimidin-4-ylamino)benzoic acid (11b). Prepared according to method (A) by reaction of 4-chloro-6-(4-methoxyphenyl)-2-thioxo-1,2-dihydropyrimidine-5-carbonitrile(**7e**) with 4-aminobenzoic acid. Yield (42%). m.p. $271-274 \,^{\circ}$ C. ¹H NMR (300 MHz, DMSO) 3.83 (s, 3H), 5.22 (br., 1H, D₂O exchangeable), 7.06–7.09 (d, $J = 8.64 \,\text{Hz}$, 2H), 7.37–7.40(d, $J = 8.52 \,\text{Hz}$, 2H), 7.59–7.61(d, $J = 8.52 \,\text{Hz}$, 2H), 7.82–7.85(d, $J = 8.64 \,\text{Hz}$, 2H), 12.44 (br., 1H, D₂O exchangeable) 12.94 (br., 1H, D₂O exchangeable). MS, m/z (%): 378(M+, 28%). Anal. Calcd for C₁₉H₁₄N₄O₃S (FW: 378): C, 60.31; H, 3.73; N, 14.81; S, 8.47. Found: C, 60.48; H, 3.66; N, 14.94; S, 8.61.

4.2.6.3. 4-(5-Cyano-6-(4-methoxyphenyl)-2-thioxo-1,2dihydropyrimidin-4-ylthio)benzoic acid (**11c**). Prepared according to method (A) by reaction of 4-chloro-6-(4-methoxyphenyl)-2thioxo-1,2-dihydropyrimidine-5-carbonitrile (**7e**) with 4mercaptobenzoic acid. Yield (42%). m.p. 283–285 °C. ¹H NMR (300 MHz, DMSO) δ 3.82 (s, 3H), 7.16–7.19 (d, J = 8.64 Hz, 2H), 7.57–7.60 (d, J = 8.52 Hz, 2H), 7.68–7.71 (d, J = 8.52 Hz, 2H), 8.02–8.05 (d, J = 8.64 Hz, 2H), 12.74 (br., 1H, D₂O exchangeable) 13.52 (br., 1H, D₂O exchangeable). MS, m/z (%): 395(M+, 42%). Anal. Calcd for C₁₉H₁₃N₃O₃S₂ (FW: 395): C, 57.71; H, 3.31; N, 10.63; S, 16.22. Found: C, 57.65; H, 3.22; N, 10.71; S, 16.09.

4.2.6.4. 4-(5-*Cyano*-2-*thioxo*-6-*p*-*tolyl*-1,2-*dihydropyrimidin*-4*ylthio)benzoic acid* (**11d**). Prepared according to method (A) by reaction of 4-chloro-6-*p*-tolyl-2-thioxo-1,2-dihydropyrimidine-5carbonitrile (**7f**) with 4-mercaptobenzoic acid. Yield (51%), m.p. 252–254 °C. ¹H NMR (300 MHz, DMSO) δ 2.31 (s, CH₃), 7.16–7.19 (d, *J* = 8.64 Hz, 2H), 7.27–7.30(d, *J* = 8.33 Hz, 2H), 7.66–7.69 (d, *J* = 8.52 Hz, 2H), 8.08–8.11 (d, *J* = 8.52 Hz, 2H), 12.79 (br., 1H, D₂O exchangeable) 13.10 (br., 1H, D₂O exchangeable). MS, *m/z* (%): 379(M+, 40%). Anal. Calcd for C₁₉H₁₃N₃O₂S₂ (FW: 379): C, 60.14; H, 3.45; N, 11.07; S, 16.90. Found: C, 60.02; H, 3.22; N, 11.31; S, 16.74.

4.2.6.5. 4-(6-(4-Chlorophenyl)-5-cyano-2-thioxo-1, 2-dihydropyrimidin-4-ylthio)benzoic acid (**11e**). Prepared according to method (A) by reaction of 4-chloro-6-(4-Chlorophenyl)-2-thioxo-1,2-dihydropyrimidine-5-carbonitrile (**7c** $) with 4-mercaptobenzoic acid. Yield (62%), m.p. 274–277 °C. ¹H NMR (300 MHz, DMSO) <math display="inline">\delta$ 7.35–7.38 (d, *J* = 8.50 Hz, 2H), 7.48–7.51(d, *J* = 8.50 Hz, 2H), 7.65–7.68(d, *J* = 8.48 Hz, 2H), 8.08–8.11 (d, *J* = 8.48 Hz, 2H), 12.55 (br., 1H, D₂O exchangeable) 13.37 (br., 1H, D₂O exchangeable). MS, *m/z* (%):399 (M+, 28%), 401(M+, 9%) 28%). Anal. Calcd for C₁₈H₁₀ClN₃O₂S₂ (FW: 398.99): C, 54.07; H, 2.52; Cl, 8.87; N, 10.51; S, 16.04. Found: C, 53.89; H, 2.34; Cl, 8.66; N, 10.64; S, 16.12.

4.2.6.6. 4-(6-(4-chlorophenyl)-5-cyano-2-(2-ethoxy-2-oxoethylthio) pyrimidin-4-ylthio)benzoic acid **(11f)**. Prepared according to method (A) by reaction of 4-chloro-6-(4-Chlorophenyl)-2-(2-ethoxy-2-oxoethylthio)-1,2-dihydropyrimidine-5-carbonitrile **(7g)** with 4-mercaptobenzoic acid. Yield (54%), m.p. 264–267 °C. ¹H NMR (300 MHz, DMSO) δ 1.08–1.22 (t, 3H, CH₃), 4.03–4.21(m, 4H, 2CH₂), 2H), 7.55–7.759(m, 4H), 7.82–7.85(d, *J* = 8.24 Hz, 2H), 8.01–8.04(d, *J* = 8.52 Hz, 2H), 12.50 (br., 1H, D₂O exchangeable). MS, *m/z* (%): 485 (M+, 31%), 487 (M+, 11%). Anal. Calcd for C₂₂H₁₆ClN₃O₄S₂ (FW: 485): C, 54.37; H, 3.32; Cl, 7.30; N, 8.65; S, 13.20. Found: C, 54.54; H, 3.49; Cl, 7.14; N, 8.57; S, 13.41.

4.2.7. Synthesis of compounds 12, 13 and 16–19

4.2.7.1. Method D. Compounds **11a-f** (10 mmol) were treated with a solution of 3,4-diaminobenzoic acid (1.53g, 10 mmol) in polyphosphoric acid (20 mL). The mixture was heated to $150-180 \degree$ C for 4 h. The reaction mixture was cooled to room temperature then was quenched with water and neutralized with NH₄OH the solid was filtered and was purified by column chromatography (0–100%

EtOAc/n-hexane).

4.2.7.2. 2-(4-(5-cyano-6-(4-nitrophenyl)-2-thioxo-1,2-dihydropyrimidin-4-ylamino)phenyl)-1H-benzo[d]imidazole-5-carboxylic acid (**12**). Prepared by reaction of compound**11a**with 3,4-diaminobenzoic acid according to method**D** $. Yield (28%), m.p. 305–307 °C. ¹H NMR (300 MHz, DMSO) <math>\delta$ 5.65 (s., 1H, D₂O exchangeable), 7.17–7.19 (d, J=8.36 Hz, 2H), 7.64–7.66 (d, J=8.12 Hz, 1H), 7.72–7.74(d, J=8.36 Hz, 2H), 7.82–7.84 (d, J=8.12 Hz, 1H), 7.92–7.94 (d, J=8.36 Hz, 2H), 8.20 (s, 1H), 8.38–8.40(d, J=8.36 Hz, 2H), 12.04 (s., 1H, D₂O exchangeable), 12.15 (s., 1H, D₂O exchangeable), 12.75(br, 1H, D₂O exchangeable), 12.55 (s., 1H, D₂O exchangeable), 12.75(br, 1H, D₂O exchangeable). MS, m/z (%):509 (M+, 45%). Anal. Calcd for C₂₅H₁₅N₇O₄S (FW: 509): C, 58.93; H, 2.97; N, 19.24; S, 6.29. Found: C, 58.71; H, 2.68; N, 19.55; S, 6.42.

4.2.7.3. 2-(4-(5-cyano-6-(4-methoxyphenyl)-2-thioxo-1,2-dihydropyrimidin-4-ylamino)phenyl)-1H-benzo[d]imidazole-5-carboxylic acid(**13**). Prepared by reaction of compound**11b**with 3,4-diaminobenzoic acid according to method**D** $. Yield (25%), m.p. 287–289 °C. ¹H NMR (300 MHz, DMSO) <math>\delta$ 3.81 (s, 3H), 5.24 (br., 1H, D₂O exchangeable), 7.05–7.08(d, J = 8.35 Hz, 2H), 7.22–7.25(d, J = 8.25 Hz, 2H), 7.66–7.69(d, J = 8.35 Hz, 2H), 7.22–7.25(d, J = 8.25 Hz, 2H), 7.66–7.69(d, J = 8.35 Hz, 2H), 7.88–7.91(m, 3H), 8.15–8.18(d, J = 8.35 Hz, 1H), 8.23 (s, 1H), 11.43 (br., 1H, D₂O exchangeable), 12.55 (br., 1H, D₂O exchangeable), 13.33(br., 1H, D₂O exchangeable), 13.57 (br., 1H, D₂O exchangeable), 13.33(br., 1H, D₂O exchangeable). ¹³C NMR (75 MHz, DMSO) δ 58.32, 115.17, 116.12, 116.54, 119.84, 120.87, 124.21, 125.61, 129.33, 129.98, 131.96, 131.46, 133.20, 139.01, 146.31, 147.27, 148.31, 150.75, 152.52, 159.24, 163.99, 165.24, 166.87, 181.07. MS, m/z (%):494 (M+, 45%). Anal. Calcd for C₂₆H₁₈N₆O₃S (FW: 494): C, 63.15; H, 3.67; N, 16.99; S, 6.48. Found: C, 63.25; H, 3.70; N, 16.81; S, 6.66.

4.2.7.4. 2-(4-(5-cyano-6-(4-methoxyphenyl)-2-thioxo-1,2-dihydropyrimidin-4-ylthio)phenyl)-1H-benzo[d]imidazole-5-carboxylic acid (**16**). Prepared by reaction of compound**11c**with 3,4-diaminobenzoic acid according to method**D** $. Yield (27%), m.p. 301–303 °C. ¹H NMR (300 MHz, DMSO) <math>\delta$ 3.81 (s, 3H), 7.01–7.04(d, J = 8.25 Hz, 2H), 7.46–7.49(m, 4H), 7.73–7.77(d, J = 8.25 Hz, 2H), 7.85–7.88(d, J = 8.35 Hz, 1H), 8.09–8.12(d, J = 8.35 Hz, 1H), 8.23 (s, 1H), 11.61 (br., 1H, D₂O exchangeable), 12.89 (br., 1H, D₂O exchangeable), 13.03(br., 1H, D₂O exchangeable). MS, m/z (%): 511 (M+, 31%). Anal. Calcd forC₂₆H₁₇N₅O₃S₂ (FW: 511): C, 61.04; H, 3.35; N, 13.69; S, 12.54. Found: C, 61.22; H, 3.41; N, 13.48 S, 12.33.

4.2.7.5. 2-(4-(5-cyano-2-thioxo-6-p-tolyl-1,2-dihydropyrimidin-4ylthio)phenyl)-1H-benzo[d]imidazole-5-carboxylic acid (17). Prepared by reaction of compound **11d** with 3,4-diaminobenzoic acid according to method **D**. Yield (27%), m.p. 265–267 °C. ¹H NMR (300 MHz, DMSO) δ 2.33 (s, 3H), 7.12–7.15(d, *J* = 8.26 Hz, 2H), 7.46–7.49(d, *J* = 8.25 Hz, 2H), 7.68–7.71(d, *J* = 8.36 Hz, 2H), 7.80–7.83(d, *J* = 8.36 Hz, 2H), 7.91–7.94(d, *J* = 8.35 Hz, 1H), 8.05–8.08(d, *J* = 8.35 Hz, 1H), 8.21 (s, 1H), 11.61 (br., 1H, D₂O exchangeable), 12.78 (br., 1H, D₂O exchangeable), 13.22 (br., 1H, D₂O exchangeable). MS, *m/z* (%): 495 (M+, 45%). Anal. Calcd for-C₂₆H₁₇N₅O₂S₂ (FW: 495): C, 63.01; H, 3.46; N, 14.13; S, 12.94. Found: C, 63.22; H, 3.59; N, 14.21; S, 13.09.

4.2.7.6. 2-(4-(6-(4-chlorophenyl)-5-cyano-2-thioxo-1,2-dihydropyrimidin-4-ylthio)phenyl)-1H-benzo[d]imidazole-5-carboxylic acid (**18**). Prepared by reaction of compound**11e**with 3,4-diaminobenzoic acid according to method**D** $. Yield (25%), m.p. 322–324 °C. ¹H NMR (400 MHz, DMSO) <math>\delta$ 7.27–7.29(d, J = 8.36 Hz, 2H), 7.55–7.57 (d, J = 8.32 Hz, 2H), 7.63–7.65(d, J = 8.36 Hz, 1H), 7. 74-7. 76 (d, J = 8.36 Hz, 2H), 7.91–7.93(d, J = 8.36 Hz, 1H), 8.01–8.03(d, J = 8.36 Hz, 1H), 8.22 (s, 1H), 12.04 (s., 1H, D₂O

exchangeable), 12.16 (s., 1H, D₂O exchangeable), 12.75 (br., 1H, D₂O exchangeable). ¹³C NMR (75 MHz, DMSO) δ 114.69, 114.99, 116.30, 117.50, 123.38, 124.22, 124.59, 125.16, 125.46, 129.26, 133.30, 135.43, 137.70, 140.45, 144.21, 154.95, 155.28, 161.73, 166.58, 167.77, 180.05. MS, *m/z* (%): 515 (M+, 39%), 517 (M+, 14%). Anal. Calcd for C₂₅H₁₄ClN₅O₂S₂ (FW: 515): C, 58.19; H, 2.73; Cl, 6.87; N, 13.57; S, 12.43. Found: C, 58.31; H, 2.97; Cl, 6.56; N, 13.32; S, 12.54.

4.2.7.7. 2-(4-(6-(4-chlorophenyl)-5-cyano-2-(2-ethoxy-2-oxoethylthio) pyrimidin-4-ylthio)phenyl)-1H-benzo[d]imidazole-5-carboxylic acid (**19**). Prepared by reaction of compound **11f** with 3,4-diaminobenzoic acid according to method D.

¹H NMR (300 MHz, DMSO) δ 1.18–1.29 (t, 3H, CH₃), 4.02–4.18(m, 4H, 2CH₂), 7.48–7.51(d, J = 8.25 Hz, 2H), 7.62–7.65 (d, J = 8.35 Hz, 2H), 7.72–7.75(d, J = 8.35 Hz, 2H), 7.83–7.86 (d, J = 8.25 Hz, 1H), 7.96–7.99 (d, J = 8.35 Hz, 2H), 8.15–8.18 (d, J = 8.22 Hz, 1H), 8.25 (s, 1H), 11.56 (br., 1H, D₂O exchangeable), 12.75 (br., 1H, D₂O exchangeable), 12.75 (br., 1H, D₂O exchangeable), MS, m/z (%): 601(M+, 25%), 603(M+, 8%). Anal. Calcd forC₂₉H₂₀ClN₅O₄S₂ (FW: 601): C, 57.85; H, 3.35; Cl, 5.89; N, 11.63; S, 10.65. Found: C, 57.67; H, 3.29; Cl, 5.77; N, 11.42; S, 10.72.

4.2.8. Synthesis of compounds 14, 15 and 20-23

4.2.8.1. 2 - (4 - (5 - cyano - 6 - (4 - nitrophenyl) - 2 - thioxo - 1, 2 - dihydropyrimidin - 4 - ylamino)phenyl) - 1H - benzo[d]imidazole - 5-carboxamide (14). Prepared by the reaction of compound 12 with ammonium carbonate according to method**C.** $Yield (68%), m.p. 291–293 °C. ¹H NMR (400 MHz, DMSO) <math>\delta$ 5.53–5.58 (br., 1H, D₂O exchangeable), 7.24–7.26(d, J = 8.36 Hz, 2H), 7.76–7.78(d, J = 8.36 Hz, 1H), 7.94–7.96(d, J = 8.44 Hz, 2H), 7.97–7.99(m, 3H), 8.14 (s, 1H), 8.21–8.23(d, J = 8.44 Hz, 2H), 8.51–8.55 (br., 2H, D₂O exchangeable), 12.17 (br., 1H, D₂O exchangeable), 12.37 (br., 1H, D₂O exchangeable), 12.37 (br., 1H, D₂O exchangeable), 12.37 (br., 1H, D₂O exchangeable). MS, m/z (%):508 (M+, 24%). Anal. Calcd for C₂₅H₁₆N₈O₃S (FW: 508): C, 59.05; H, 3.17; N, 22.04; S, 6.31. Found: C, 59.31; H, 3.25; N, 22.13; S, 6.19.

4.2.8.2. 2-(4-(5-cyano-6-(4-methoxyphenyl)-2-thioxo-1,2-dihydropyrimidin-4-ylamino)phenyl)-1H-benzo[d]imidazole-5-carboxamide (15). Prepared by reaction of compound 13 with ammonium carbonate according to method**C.** $Yield (67%), m.p. 318–320 °C. ¹H NMR (400 MHz, DMSO) <math>\delta$ 3.85 (s, 3H), 5.09 (br., 1H, D₂O exchangeable), 6.94–6.96 (d, J = 8.24 Hz, 2H), 7.26–7.28(d, J = 8.36 Hz, 2H), 7.65–7.67(d, J = 8.24 Hz, 2H), 7.72–7.74(d, J = 8.32 Hz, 1H), 7.88–7.90(d, J = 8.36 Hz, 2H), 7.95–7.97 (d, J = 8.32 Hz, 1H), 8.13 (s, 1H), 8.61–8.65 (br., 2H, D₂O exchangeable), 12.17 (br., 1H, D₂O exchangeable), 12.37 (br., 1H, D₂O exchangeable). MS, *m*/*z* (%):493 (M+, 34%). Anal. Calcd forC₂₆H₁₉N₇O₂S (FW: 493): C, 63.27; H, 3.88; N, 19.87; S, 6.50. Found: C, 63.50; H, 3.77; N, 19.67; S, 6.39.

4.2.8.3. 2-(4-(5-cyano-6-(4-methoxyphenyl)-2-thioxo-1,2-dihydropyrimidin-4-ylthio)phenyl)-1H-benzo[d]imidazole-5-carboxamide (**20**). Prepared by the reaction of compound**16**with ammonium carbonate according to method**C.** $Yield (60%), m.p. <math>325-327 \,^{\circ}$ C. ¹H NMR (300 MHz, DMSO) δ 3.81 (s, 3H), 7.03–7.06 (d, J = 8.35 Hz, 2H), 7.43–7.46(d, J = 8.70 Hz, 2H), 7.60–7.63(d, J = 8.20 Hz, 1H), 8.04–8.07(d, J = 8.20 Hz, 1H), 8.23 (s, 1H), 8.69 (br., 2H, D₂O exchangeable), 12.47 (br., 1H, D₂O exchangeable), 13.21(br., 1H, D₂O exchangeable). ¹³C NMR (75 MHz, DMSO) δ 55.99. 115.03, 115.92, 116.44, 121.17, 122.47, 123.66, 127.42, 127.89, 129.41, 130.11, 132.42, 134.55, 139.65, 144.89, 152.67, 160.99. 162.69, 163.41, 168.09, 180.13. MS, m/z (%): 510 (M⁺, 16%). Anal. Calcd for C₂₆H₁₈N₆O₂S₂ (FW: 510): C, 61.16; H, 3.55; N, 16.46; S, 12.56. Found: C, 61.01; H, 3.42; N, 16.33; S, 12.72.

4.2.8.4. 2-(4-(5-cyano-2-thioxo-6-p-tolyl-1,2-dihydropyrimidin-4vlthio)phenyl)-1H-benzo[d]imidazole-5-carboxamide (21). Prepared by reaction of compound 17 with ammonium carbonate according to method C. Yield (27%), m.p. 299-302 °C. ¹H NMR (300 MHz, DMSO) δ2.33 (s, 3H), 7.12-7.13(d, *J* = 8.25 Hz, 2H), 7.24-7.27(d, J = 8.25 Hz, 2H), 7.46-7.49(d, J = 8.70 Hz, 2H),7.68–7.71(d, J = 8.70 Hz, 2H), 7.91–7.94(d, J = 8.25 Hz, 1H), 8.05-8.08(d, J=8.25 Hz, 1H), 8.21 (s, 1H), 8.61 (br., 2H, D₂O exchangeable), 12.78 (br., 1H, D₂O exchangeable), 13.22 (br., 1H, D₂O exchangeable). ¹³C NMR (75 MHz, DMSO) δ 21.59, 115.45, 116.3117.21, 122.34, 124.98, 127.49, 128.71, 128.99, 129.44, 130.10, 131.11, 134.58, 137.66, 140.21, 145.34, 152.13, 163.33, 164.03, 168.00, 180.27. MS, *m*/*z* (%): 494 (M⁺, 45%). Anal. Calcd forC₂₆H₁₈N₆OS₂ (FW: 494): C, 63.14; H, 3.67; N, 16.99; S, 12.97. Found: C, 63.14; H, 3.67; N. 16.99; S. 12.97.

4.2.8.5. 2 - (4 - (6 - (4 - chlorophenyl) - 5 - cyano - 2 - thioxo - 1, 2 - dihydropyrimidin-4 - ylthio)phenyl) - 1H - benzo[d]imidazole-5-carboxamide (22). Prepared by reaction of compound 18 with ammonium carbonate according to method**C.** $Yield (25%), m.p. 330–332 °C. ¹H NMR (300 MHz, DMSO) <math>\delta$ 7.22–7.25(d, J = 8.70 Hz, 2H), 7.33–7.51(m, 4H), 7.65–7.68(d, J = 8.70 Hz, 2H), 7.91–7.94(d, J = 8.25 Hz, 1H), 8.08–8.11(d, J = 8.25 Hz, 1H), 8.21 (s, 1H), 8.65 (br., 2H, D₂O exchangeable), 12.49 (br., 1H, D₂O exchangeable), 13.08 (br., 1H, D₂O exchangeable). ¹³C NMR (75 MHz, DMSO) δ 115.22, 115.55, 116.54, 120.74, 122.20, 127.41, 127.87, 128.47, 129.82, 130.10, 131.87, 135.32, 135.55, 139.24, 145.90, 152.74, 163.04, 163.49, 168.11, 180.55. MS, m/z (%): 514 (M+, 24% for Cl^{35.5}), 516 (M+, 10% for Cl^{37.5}). Anal. Calcd for C₂₅H₁₅ClN₆OS₂ (FW: 514): C, 58.30; H, 2.94; Cl, 6.88; N, 16.32; S, 12.45. Found: C, 58.48; H, 2.77; Cl, 6.64; N, 16.45; S, 12.61.

4.2.8.6. 2-(4-(6-(4-chlorophenyl)-5-cyano-2-(2-ethoxy-2oxoethylthio) pyrimidin-4-ylthio) phenyl)-1H-benzo[d]imidazole-5carboxamide (23). Prepared by the reaction of compound 19 with ammonium carbonate according to method C. Yield (28%), m.p. 317–320 °C. ¹H NMR (300 MHz, DMSO) δ 1.21–1.29 (t, 3H, CH₃), 4.04-4.18(m, 4H, 2CH₂), 7.40-7.43(d, J = 8.25 Hz, 2H), 7.52-7.55(d, J = 8.70 Hz, 2H), 7.67–7.70(d, J = 8.25 Hz, 2H), 7.87–7.90(d, J = 8.20 Hz, 1H), 7.97-8.00(d, J = 8.70 Hz, 2H), 8.12-8.15(d, J = 8.20 Hz, 1H), 8.23 (s, 1H), 8.55 (br., 2H, D₂O exchangeable), 12.71(br., 1H, D₂O exchangeable). ¹³C NMR (75 MHz, DMSO) δ14.12, 35.46, 60.54, 100.44, 115.20, 116.87, 117.31, 122.34, 127.44, 127.65, 127.72, 128.78, 129.83, 129.99, 131.20, 131.68, 133.49, 134.09, 139.54, 145.63, 152.44, 164.47, 168.21, 169.87, 176.55, 192.21. MS, m/z (%): $600 (M^+,\ 39\%$ for $Cl^{35.5}),\ 602 (M^+,\ 16\%$ for $Cl^{37.5}).$ Anal. Calcd. for-C29H21ClN6O3S2 (FW: 600): C, 57.95; H, 3.52; Cl, 5.90; N, 13.98; S, 10.67. Found: C, 57.79 H, 3.33; Cl, 5.87; N, 14.08; S, 10.81.

4.3. In vitro checkpoint Kinase Assay

The CycLex[®] Checkpoint Kinase Assay/Inhibitor Screening Kit was used to measure the activities of checkpoint kinases by using a phospho-Cdc25C (Ser216) monoclonal antibody to provide a specific and sensitive method. Chk2 positive control (CY-E1162-2) was used. Assay conditions were based on published protocols with minor modifications. The optical density (O.D.) of each well was measured using a spectrophotometric plate reader at dual wavelengths of 450/492 nm using ELISA microplate reader. Then, assays were repeated by using mixture of optimum concentration of positive control that exhibited maximum activity of checkpoint kinase 2 (0.01 unit/well) and specific concentration of each com- $B = 1.5 \,\mu mol/L$, pound $A = 2.5 \,\mu mol/l$, $C = 0.1 \,\mu mol/L$ $D = 0.050 \,\mu mol/L$ and $E = 0.025 \,\mu mol/L$. The IC₅₀ (concentration that produces 50% of enzyme inhibition) by synthesized compounds as Chk2 inhibitors has been calculated as the mean value of at four replicate experiments. %inhibition= (1-(OD of the compound at specific conc. – OD of negative control)÷ (OD of pos. control - OD of negative control)) X100, (All data are represented in supplementary material).

4.4. Western blot

MCF-7 breast carcinoma including untreated cells (control cells), treated cells with GI₅₀ of compound 8, 9, GI₅₀ of cisplatin alone or GI₅₀ of cisplatin in combination with 0.1 µM of compound 8, 9. Cells were grown to ~70% confluency and reagents were added as indicated concentrations. Cells were lysed in lysis buffer containing Tris-HCl (20 mM), NaCl (0.5 M), Triton X-100(0.25%), EDTA (1 mM), EGTA, (1 mM), glycophosphate (10 mM), NaF (10 mM), Na₃VO₄ (300 IM), benzamidine (1 mM), PMSF (2 IM) and DTT (1 mM). The protein concentration was determined by the BCA protein assay (Thermo Scientific, Rockford, IL, USA). Proteins were separated on SDS-PAGE, transferred on to nitrocellulose membrane (NC). NC sheets were cut into 0.5 cm strips [52]. Followed by blocking in 5% dry skimmed milk in TBS-T for 2 h on rocker platform and then, following washing by TBS-T. The strips were probed with phosphorylated anti-Chk2 antibody (Sigma-Aldrich) in 1:200 concentrations. Anti-mouse IgM peroxidase (Sigma-Aldrich) diluted at 1: 1000 in TBS was added to NC strips for 1 h on a rocker platform, The o-phenylene diamine tablet (Sigma-Aldrich) dissolved in 10 mL methanol and completed to 50 mL with TBS was added to NC stripes for 30 min [53].

Immunoreactivity was detected using the ECL detection system (Amersham Biosciences).

4.5. Cytotoxicity assay

4.5.1. Cell lines

Breast carcinoma, estrogen receptor positive (ER+) cell line (MCF-7) was used in this study. It was obtained frozen in liquid nitrogen (-180 °C) from the American Type Culture Collection (ATTC, USA). The tumor cell lines were maintained by serial sub culturing at the National Cancer Institute, Cairo, Egypt. They were cultured in RPMI-1640 medium supplemented with 10% fetal bovine serum, 100 U/ml penicillin, 100 mg/mL streptomycin, and 3 mM glutamine and incubated in humidified incubator at 37 °C and 5% CO₂. The cells were sub-cultured every three days. Cisplatin and doxorubicin were purchased from Sigma Chemical Co. (St. Louis, MO, USA). The cytotoxicity of the synthesized compounds in addition to two reference compounds (positive control) cisplatin and doxorubicin were tested on MCF7 cancer cell lines, using sulphorhodamine-B (SRB) method [46]. In brief; cells were seeded at a density of 3×10^3 cells/well in 96-well microliter plates. They were left to attach for 24 h before incubation with studied compounds or drugs. Then, cells were treated with different concentrations of the Chk2 inhibitors (0.05, 0.1, 0.15 and 0.2 µmol). For each concentration, three wells were used and incubation was continued for 48 h. A negative control well-containing solvent as DMSO its concentration does not exceed (1% v/v) was included. At the end of incubation, cells were fixed with 20% trichloroacetic acid (TCA), stained with 0.4% SRB dye. The optical density (O.D.) of each well was measured spectrophotometrically at 570 nm using ELISA microplate reader (TECAN sunrise[™], Germany). The mean survival fraction at each drug concentration was calculated as follows: O.D. of the treated cells/O.D. of the control cells. The GI₅₀ (concentration that produces 50% of cell growth inhibition) value of each drug was calculated using sigmoidal dose response curve fitting models (Graph Pad Prizm software, version 5). Each value is the mean \pm SD of 5 independent experiments performed in triplicates; the surviving fraction = O.D. (treated cells)/O.D. (control cells).

The statistical significance of the results was analyzed using one-way ANOVA followed by Tukey multiple comparison tests.

4.5.2. Combined effect of different concentrations of the tested compounds and IG_{50} of cisplatin or doxorubicin on MCF7

The combination regimens were designed using GI₅₀ of doxorubicin and cisplatin with different concentrations of Chk2 inhibitors on MCF-7 cells. Different concentrations of the studied compounds were chosen to test their combination with doxorubicin and cisplatin for further experiments. MCF7 was plated in 96 well plates at concentration 3×10^3 cells/well and were left to attach for 24 h before incubation with drugs. MCF7 cells were treated with either GI₅₀ of doxorubicin or cisplatin alone and with different concentrations of the Chk2 inhibitors (0.05, 0.1, 0.15 and 0.2 µmol) and incubated for 48 h. At the end of incubation, cells were fixed with 20% trichloroacetic acid (TCA) and stained with 0.4% SRB dye as previously described [46].

4.6. Propidium iodide staining

Cells were plated in 60-mm dishes at 2×10^5 cells/mL RPMI. On the following day, the cells were treated with GI₅₀ of cisplatin, compound **8** or **9** alone or combination of GI₅₀ of cisplatin with 0.1 μ M of synthesized conjugates **8** or **9** for 48 h, fixed with 2% paraformaldehyde solution, and incubated with propidium iodide buffer (0.1% sodium citrate, 0.1% Triton X-100, and 0.1 mg/mL propidium iodide). Subsequently, nuclei of the cells were observed with confocal microscope (Zeiss LSM 710), the objective lens is 20X, the cells were counted in 5 points of dimension 1700 μ m² magnification 200X.

4.7. Cell cycle analysis

Briefly cell cycle phase modifications were examined by flow on propidium iodide–stained cells using Cycle TESTTM PLUS DNA Reagent (BECTON DICKINSON, San Jose, California, U.S.A.) that was used following the manufacturer's protocol, including that for staining with propidium iodide for cell-cycle analysis on MCF-7 cells that treated with DMSO or with either GI₅₀ of doxorubicin alone, and with 0.1 μ M or with GI₅₀ of studied compounds as previously described [46]. Cells were analyzed by flow cytometry to quantify the numbers of apoptotic cells and the distribution of cells throughout the cell cycle.

In conclusion, our synthesized compounds demonstrated strong antitumor activity alone and in combination with other genotoxic drugs as cisplatin and doxorubicin via inhibition of Chk2, induction of apoptosis and cell cycle arrest.

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Appendix A. Supplementary data

Supplementary data related to this article can be found at https://doi.org/10.1016/j.ejmech.2018.01.072.

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