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First synthesis of P-chirogenic prophosphatranes

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ABSTRACT

Syntheses are reported for the novel P-chirogenic bicyclic prophosphatranes $P(RNCH_2CH_2)_2N(OCH_2CH_2)_-$ (**8**) in which the two R groups [i.e., 1-methylenenaphthyl and 1,2-methoxybenzyl (**8**)] and in its corresponding phosphine oxide (**9**) are different. Also synthesized was the transannulated protonated phosphatrane cation in the salt [HP(RNCH_2CH_2)_3N]Cl in which the three R groups [i.e., 1-methylenenaphthyl, 1,2-methoxybenzyl, and (*S*)-1-phenethyl (**14**)] are different, and also in its corresponding deprotonated prophosphatrane form P(RNCH_2CH_2)_3N (**15**). Good analytical resolution of racemic **9** is reported, whereas only partial resolution was achieved for diastereomeric **14**.

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Nucleophilic organophosphorus catalysts enjoy wide use in organic reactions,¹ and many phosphine ligands are exceedingly effective in metal-assisted cross couplings.² Such transformations are of increasing interest in the synthesis of pharmaceuticals,² fine chemicals,³ and compounds of biological interest.⁴ Bulky electronrich phosphines can be particularly useful for aryl chloride⁵ and also for aromatic and heteroaromatic silanoate⁶ substrates in cross coupling reactions. Prochiral olefins have been hydrogenated enantioselectively using chiral phosphites in the presence of a rhodium catalyst.⁷ Allylation catalyzed by palladium/chiral phosphine ligand systems appears to be of considerable utility⁸ and such a ligand has also been reported to efficiently facilitate an asymmetric Heck^{8e,9} as well as other Heck-type reactions.^{9b}

Chiral organophosphorus compounds catalyze a wide variety of useful enantiomeric transformations as organocatalysts.¹⁰ Chiral phosphines catalyze the enantioselective acylation of alcohols to effect their kinetic resolution^{10a,11} as well as desymmetrization of *meso* diols via acylation.^{10a} Chiral phosphines also catalyze reactions of prochiral substrates as in MBH^{10a,12} and aza-MBH reactions,^{10a} formation of asymmetric quaternary carbon centers in the addition of nucleophiles to the γ -position of ethyl butynoate and ethyl 2,3-butadineate,^{10a} and also in Steglich rearrangements.^{10a} Cycloadditions of the [4 + 2] type are catalyzed by chiral phosphines as well as such reactions of the relatively rare [3 + 2] variety.^{10a}

Non-chiral prophosphatranes of type **A** (with two or three identical R groups) in Figure 1 have been reported to catalyze a wide variety of important organic reactions¹³ and to function as excellent ligands for palladium-catalyzed transformations.¹⁴ Because a significant number of such reactions are carried out with prochiral substrates, it would be of interest to investigate the efficacy of P-chirogenic prophosphatranes in such reactions. P-chirogenic compounds¹⁵ form a sizeable sub-class of chiral phosphines, but no examples of P-chirogenic prophosphatranes have been reported until now.

Prophosphatrane structures of types **A** and **B**, for example, feature a P-chiral 'pocket' for substrate activation by the nucleophilic phosphorus. Such a pocket in **A** and/or **B** may be anticipated to favor high product ee values since the R groups are held quite rigidly in place by the bicyclic framework and by the planarity of all four nitrogens.^{1,2} This rigidity would be reasonably maintained by any changes in the N \rightarrow P transannular distance that could potentially occur during the catalytic cycle.

Here we report initial results for the synthesis of chiral **8** and **9** in Scheme 1, and chiral **14** and **15** in Scheme 2.

The synthesis of racemic **8** was achieved by monoprotection of an NH_2 group in **2** with BOC to provide intermediate **3**, which was



Figure 1. Prophosphatranes.



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Scheme 1. Synthesis of prophosphatrane 8 and its oxide 9.

then followed by reductive amination of the second NH_2 group with α -naphthaldehyde, BOC deprotection, and reductive amination with *ortho*-benzaldehyde to afford compound **6** bearing different R groups attached to the two terminal NH groups. Under similar conditions for the synthesis of non-chiral prophosphatranes of type **A**,¹⁶ compound **6** was converted into the oxa-prophosphatrane salt **7**, which was then deprotonated to afford oxa-prophosphatrane **8**. The overall yield of **8** through seven steps from **1** is 28%. This is the first report of the synthesis of a prophosphatrane featuring two different heteroatoms on the phosphorus. For purification and optical resolution of an air- and moisturestable derivative of **8**, this type **B** product was converted into its corresponding oxide **9** in 80% yield via CH₃SiOOSiCH₃ oxidation.

Our synthesis of a P-chirogenic prophosphatrane of type **A** with three different R groups on the P–N nitrogens relied on the successful synthesis of the intermediate tetraamine **13** in Scheme 2. Starting with **6**, which was synthesized from **1** as shown in Scheme 1, the two terminal NH groups were protected by BOC, followed by protection of the OH group as the mesylate **11**, which was allowed to react with (s)-1-methyl benzylamine to provide **12**. Following

BOC deprotection of **12**, **13** was afforded in very good overall yield (66% from **6** through four steps) and in 21% overall yield from **1** through nine steps. This represents the first synthesis of a tetraamine of type **13** with three different R groups on the terminal NH groups. Our methodology is amenable to scale-up and easy purification of the products of each step via column chromatography. Analogously to the synthesis of non-chiral prophosphatranes of type of **A**,¹⁶ diastereomeric **15** was obtained by reaction of **13** with ClP((NMe₂)₂, which was prepared in situ via the reaction of HMPT {P(NMe₂)₃} with PCl₃, to form the salt **14** which was then deprotonated with *t*BuOK in THF. The ³¹P NMR spectrum of **15** showed two peaks [δ 127.89 (s), 127.83 (s)] consistent with the presence of two diastereomers.

Results of chiral HPLC separation experiments with **9** and **14** are shown in Figures 2 and 3, respectively. Racemic prophosphatrane oxide **9** was successfully resolved into a pair of enantiomers on a Chiralpak IA column with a mobile phase of methyl *t*-butyl ether/methanol/acetic acid/triethylamine (85:15:0.3:0.2). The diastereomeric phosphatrane salt **14** was partially separated to a pair of diastereomers on a Chiralpak IC column with a mobile phase of



Scheme 2. Synthesis of diastereomeric prophosphatrane 15.



Figure 2. Enantiomeric separation of 9 on a Chiralpak IA column (15 cm).



Figure 3. Partial diastereomeric separation of 14 on a Chiralpak IC column (25 cm).

methyl *t*-butyl ether/methanol/acetic acid/triethylamine (85:15: 0.3:0.2).

In conclusion, we have successfully synthesized the first examples of P-chirogenic prophosphatranes. Thus $P(RNCH_2CH_2)_2-N(OCH_2CH_2)-(\mathbf{8})$ bearing two different R groups and its corresponding phosphine oxide (**9**) are reported, as well as $P(RNCH_2-CH_2)_3N$ (**15**) featuring three different R groups. Good analytical resolution of racemic **9** and partial analytical resolution of diastereomeric prophosphatrane salt **14** was achieved. Our work

sets the stage for further enantiomeric resolution and subsequent evaluation of compounds of the title type in inducing enantioselectivity in organic reactions.

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Supplementary data

Supplementary data (experimental details and characterizational data [NMR spectral (¹H, ¹³C, ³¹P) HRM spectral and HPLC conditions]) associated with this article can be found, in the online version, at doi:10.1016/j.tetlet.2011.01.041.

References and notes

- (a) Methot, J. L.; Roush, W. R. Adv. Synth. Catal. 2004, 346, 1035; (b) Adrio, L. A.; Kuok, K. J. Organomet. Chem. 2009, 35, 62.
- Metal-Catalyzed Cross-Coupling Reactions; de Meijere, A., Diederich, F., Eds.; Wiley-VCH: Weinheim, 2004.
- 3. Zapf, A. Angew. Chem., Int. Ed. 2003, 42, 5394.
- 4. Nicolaou, K. C.; Bulger, P. G.; Sarlah, D. Angew. Chem., Int. Ed. 2005, 44, 4442.
- (a) Garabatos-Perere, J. R.; Butenschoen, H. J. Organomet. Chem. 2008, 693, 357;
 (b) Barrios-Landeros, F.; Carrow, B. P.; Hartwig, J. F. J. Am. Chem. Soc. 2009, 131, 8141;
 (c) Mauger, C. C.; Mignani, G. S. A. Aldrichim. Acta 2006, 39, 17024;
 (d) Tewari, A.; Hein, M.; Zapf, A.; Beller, M. Tetrahedron 2005, 61, 9705.
- Denmark, S. E.; Smith, R. C.; Chang, W.-T. T.; Muhuhi, J. M. J. Am. Chem. Soc. 2009, 131, 3104–3118.
- Lyubimov, S. E.; Kalinin, V. N.; Tyutyunov, A. A.; Olshevskaya, V. A.; Dutikova, Y. V.; Cheong, C. S.; Petrovskii, P. V.; Safronov, A. S.; Davankov, V. Chirality 2009, 21, 2–5.
- (a) Howell, G.; Minnaard, A. J.; Feringa, B. L. Org. Biomol. Chem. 2006, 4, 1278;
 (b) Trost, B. M.; Frederiksen, M. U. Angew. Chem., Int. Ed. 2005, 44, 308–310; (c)

Boaz, N. W.; Ponasik, J. A.; Large, S. E.; Debenham, S. D. *Tetrahedron: Asymmetry* **2004**, *15*, 2151; (d) Acemoglu, L.; Williams, J. M. J. In *Handbook of Organopalladium Chemistry for Organic Synthesis*; Negishi, E., Ed.; John Wiley & Sons: Hoboken, NJ, 2002; Vol. 2, pp 1945–1979; (e) Gilbertson, S. R.; Genov, D. G.; Rheingold, A. R. Org. *Lett.* **2000**, *2*, 2885; (f) Nemoto, T.; Kanematsu, M.; Tamura, S.; Hamada, Y. Adv. Synth. Catal. **2009**, *351*, 1778.

- (a) Mata, Y.; Dieguez, M.; Pamies, O.; Claver, C. Org. Lett. 2005, 7, 5597–5599; (b) Tietze, L. E.; Lotz, F. In Asymmetric Synthesis; Christman, B., Ed., 2nd ed.; Wiley-VCH: Weinheim, 2008; pp 155–160.
- (a) Gröger, H.; Burda, E. In Phosphorous Ligands in Asymmetric Catalysis; Börner, A., Ed.; Wiley-VCH: Weinheim, 2008; Vol. 3, pp 1175–1195; (b) Andrushko, V.; Börner, A. In Phosphorus Ligands in Asymmetric Catalysis; Börner, A., Ed.; Wiley-VCH: Weinheim, 2008; Vol. 2, pp 715–748; (c) Glueck, D. S. Chem. Eur. J. 2008, 14, 7108; (d) Erre, G.; Enthaler, S.; Junge, K.; Gladiali, S.; Beller, M. Coord. Chem. Rev. 2008, 252, 471; (e) Benincori, T.; Marchesi, A.; Mussini, P. R.; Pilati, T.; Ponti, A.; Rizzo, S.; Sannicolo, F. Chem. Eur. J. 2009, 15, 86.
- (a) Vedejs, E.; Daugulis, O.; Diver, S. T. J. Org. Chem. **1996**, *61*, 430; (b) Vedejs, E.; Daugulis, O. J. Am. Chem. Soc. **1999**, 121, 5813; (c) Duffey, T. A.; MacKay, J. A.; Vedejs, E. J. Org. Chem. **2010**, 75, 4674–4685.
- (a) Wei, Y. Acc. Chem. Res. 2010, 43, 1005–1018; (b) Lei, Z.-Y.; Liu, X.-G.; Shi, M.; Zhao, M. Tetrahedron: Asymmetry 2008, 19, 2058–2062.

- (a) Chintareddy, V. R.; Wadhwa, K.; Verkade, J. G. J. Org. Chem. 2009, 74, 8118;
 (b) Wadhwa, K.; Chintareddy, V. R.; Verkade, J. G. J. Org. Chem. 2009, 74, 6681;
 (c) Wadhwa, K.; Verkade, J. G. J. Org. Chem. 2009, 74, 5683;
 (d) Wadwha, K.; Verkade, J. G. J. Org. Chem. 2009, 74, 5683;
 (d) Wadwha, K.; Verkade, J. G. J. Org. Chem. 2009, 74, 4368;
 (e) Wadhwa, K.; Verkade, J. G. J. Org. Chem. 2009, 74, 4368;
 (f) Wadhwa, K.; Verkade, J. G. J. Org. Chem. 2009, 74, 4368;
 (e) Wadhwa, K.; Verkade, J. G. J. Org. Chem. 2009, 50, 4307;
 (f) Chintareddy, V. R.; Verkade, J. G. J. Org. Chem. 2010, 75, 7166–7174.;
 (h) Raders, S. M.; Verkade, J. G. J. Org. Chem. 2010, 75, 5308.
- (a) Venkat Reddy, C. V.; Kingston, J. V.; Verkade, J. G. J. Org. Chem. 2008, 73, 3047; (b) Raders, S.; Kingston, J. V.; Verkade, J. G. J. Org. Chem. 2010, 75, 1744; (c) Kingston, J. V.; Verkade, J. G. J. Org. Chem. 2007, 72, 2816; (d) Venkat Reddy, C. V., Kingston, J. V.; Verkade, J. G., in preparation.; (e) Zhou, Y.; Verkade, J. G. Adv. Synth. Catal. 2010, 352, 616.
- (a) Taylor, A. M.; Altman, R. A.; Buchwald, S. L. J. Am. Chem. Soc. 2009, 13, 9900;
 (b) Benetskii, E. B.; Davankov, V. A.; Petrovskii, P. V.; Rastorguev, E. A.; Grishina, T. B.; Gavrilov, K. N.; Rosset, S.; Bailat, G.; Alexakis, A. Russ. J. Org. Chem. 2008, 44, 1846; (c) Chen, Y.-L.; Froehlich, R.; Hoppe, D. Tetrahedron: Asymmetry 2009, 20, 1144; (d) Chan, V. S. H.; Chiu, M.; Bergman, R. G.; Toste, D. F. J. Am. Chem. Soc. 2009, 131, 6021; (e) Barta, K.; Eggenstein, M.; Hoelscher, M.; Francio, G.; Leitner, W. Eur. J. Org. Chem. 2009, 6198; (f) Imamoto, T. In Phosphorus Ligands in Asymmetric Catalysis; Börner, A., Ed.; Wiley-VCH: Weinheim, 2008; Vol. 3, pp 1201–1210, 1267.
- 16. Kisanga, P. B.; Verkade, J. G. Tetrahedron 2001, 57, 467.