

Synthesis of Spiro Diphosphines and Their Application in Asymmetric Hydrogenation of Ketones

Jian-Hua Xie, Li-Xin Wang, Yu Fu, Shuo-Fei Zhu, Bao-Min Fan, Hai-Feng Duan, and Qi-Lin Zhou*

State Key Laboratory and Institute of Elemento-organic Chemistry, Nankai University, Tianjin 300071, China

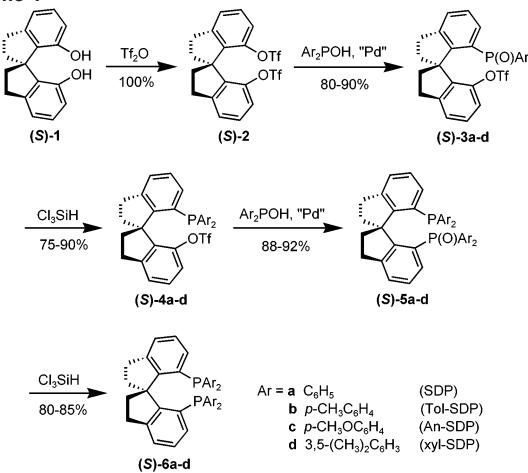
Received December 24, 2002; E-mail: qlzhou@public.tpt.tj.cn

The design of new chiral ligands is the key in the development of transition metal catalyzed asymmetric synthesis.¹ Many chiral diphosphine ligands have been prepared and applied in asymmetric catalytic reactions with excellent enantioselectivities.^{1,2} Among the chiral diphosphine ligands that have been reported, the atropisomeric *C₂*-symmetric phosphines with a biaryl scaffold initiated by Noyori and co-workers³ with BINAP were found to have the widest application in the transition metal catalyzed reactions.⁴ Planar chiral diphosphines based on ferrocene or paracyclophane backbones have also been applied to a number of reactions with a remarkable degree of success.⁵ However, the spiro diphosphine compounds, another type of axially chiral ligands, have not been synthesized until now.⁶ Recently, we designed chiral phosphoramidite ligands (SIPHOS)⁷ containing a 1,1'-spirobiindane backbone and demonstrated that these ligands can be highly efficient for the Rh-catalyzed asymmetric hydrogenation of functionalized olefins. Especially, in the case of asymmetric hydrogenation of α -arylethenylamines, the spiro monophosphoramidite ligands provided a significantly higher level of enantiocontrol compared to that of the monophosphoramidite ligands derived from BINOL.^{7b} We now describe the synthesis of spiro diphosphines **6** (SDP) containing 1,1'-spirobiindane as a new chiral scaffold and their application in the ruthenium-catalyzed asymmetric hydrogenation of simple ketones with high activity (*S/C* up to 100 000) and excellent enantioselectivity (ee up to 99.5%).

Chiral spiro diphosphines **6** were easily prepared from enantiomerically pure (*S*)-1,1-spirobiindane-7,7-diol (**1**)⁸ (Scheme 1). The diol (**1**) was converted into triflate (**2**) in quantitative yield. Monophosphinylation of triflate (**2**) with diarylphosphine oxide in the presence of Pd catalyst, followed by reduction with trichlorosilane, generated (*S*)-7-(diarylphosphino)-7'-trifluoromethanesulfonyloxy-1,1'-spirobiindanes ((*S*)-**4**).⁹ Phosphinylation and reduction of compounds (*S*)-**4** provided desired diphosphines (*S*)-**6** in high yields. Using the same procedure, the diphosphines (*R*)-**6** were also synthesized from (*R*)-1,1-spirobiindane-7,7-diol.¹⁰

The catalytic asymmetric hydrogenation of prochiral ketones appears to be the most facile route to produce enantiomerically enriched secondary alcohols. A number of efficient catalysts have been developed for the asymmetric hydrogenation of functionalized ketones.^{4j,11} In contrast, only a few catalysts have been reported in the asymmetric hydrogenation of simple ketones.¹² Recently, a significant breakthrough was achieved by Noyori and co-workers by using diphosphine–ruthenium–diamine complexes as catalysts in the hydrogenation of ketones.¹³ The most effective catalyst was *trans*-[((*S*)-Xyl-BINAP)Ru((*S*)-DAIPEN)Cl₂]¹⁴ which has extremely high activity and enantioselectivity in the hydrogenation of a wide range of ketones.^{13e,f} To date, only two other chiral diphosphine ligands, PhanePhos¹⁵ and P-Phos,¹⁶ have been reported to approach the utility of Noyori's Xyl-BINAP in this important reaction.¹⁷ We are delighted to find that the ruthenium complexes of spiro diphosphine ligands **6** serve as excellent catalysts for the asymmetric

Scheme 1

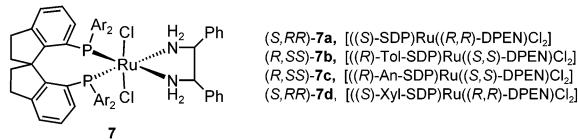


hydrogenation of aromatic, heteroaromatic, and α,β -unsaturated ketones.

The catalysts **7** (Figure 1) were prepared by reacting ligands **6** with $[(C_6H_6)RuCl_2]_2$ in DMF at 100 °C, followed by the treatment of the resulting reddish brown solution with 1 equiv of DPEN¹⁴ at room temperature. The complexes, thus obtained, were used directly in the catalytic reactions. Initial tests with catalyst $[(S\text{-}SDP)Ru((R,R)\text{-DPEN})Cl_2]$ (*S,RR*)-**7a**) in the asymmetric hydrogenation of acetophenone in 2-propanol in the presence of *t*-BuOK (*S/B* = 70) at room temperature provided (*S*)-1-phenylethanol in quantitative yield and 90% ee over 1.5 h at *S/C* = 5000 (Table 1, entry 1). This result is slightly better than that obtained with $[(R\text{-BINAP})Ru((R,R)\text{-DPEN})Cl_2]$ (87% ee).¹³ A systematical investigation on the effect of substituents in the ligands **6** indicated that the introduction of 3,5-dimethyl groups to *P*-phenyl rings, (*S,RR*)-**7d**, dramatically increased the enantioselectivity to 99% ee (entry 4).¹⁸ The enantioselectivity remained to be 98% ee even when the ratio of substrate to catalyst (*S/C*) was increased to 100 000 (entry 5).

A variety of aromatic, heteroaromatic, and α,β -unsaturated ketones can be hydrogenated by catalyst (*S,RR*)-**7d** with excellent enantioselectivities. The results summarized in Table 1 are better than or comparable to those achieved with Xyl-BINAP–Ru–DAIPEN,^{13a} Xyl-PhanePhos–Ru–DPEN,¹⁵ and Xyl-P-Phos–Ru–DPEN¹⁶ systems. It deserves commendation that the hydrogenation of acetylferrocene with (*S,RR*)-**7d** produced (*S*)-1-ferrocenylethanol in 98% ee at *S/C* = 5000.¹⁹ The enantiomerically enriched 1-ferrocenylethanol is a crucial starting material in the synthesis of many chiral ferrocene compounds such as ferrocenylethylamines and ferrocenylphosphines.²⁰ Our study provides a practical method to the synthesis of ferrocenylethanol and related compounds.

In conclusion, we have developed novel chiral diphosphine ligands with spiro biindane as a new chiral scaffold, which are highly effective for the asymmetric hydrogenation of ketones. The extremely high activity and enantioselectivity of their ruthenium

**Figure 1.****Table 1.** Asymmetric Hydrogenation of Ketones^a

entry	cat.	ketone		time (h)	convn ^b (%)	ee ^c (%)
		Ar	R			
1	7a	C ₆ H ₅	CH ₃	1.5	100	90 (S)
2	7b	C ₆ H ₅	CH ₃	3	99	89 (S)
3	7c	C ₆ H ₅	CH ₃	2.5	100	92 (S)
4	7d	C ₆ H ₅	CH ₃	1.5	100	99 (S)
5 ^d	7d	C ₆ H ₅	CH ₃	72	98	98 (S)
6	7d	o-ClC ₆ H ₄	CH ₃	3.5	99	98 (S)
7	7d	o-BrC ₆ H ₄	CH ₃	6.5	100	99.2 (S)
8	7d	m-BrC ₆ H ₄	CH ₃	3	99	99.2 (S)
9	7d	m-CF ₃ C ₆ H ₄	CH ₃	2	99	99 (S)
10	7d	p-CH ₃ C ₆ H ₄	CH ₃	1.5	100	99.2 (S)
11	7d	p-OCH ₃ C ₆ H ₄	CH ₃	4.5	100	98 (S)
12	7d	p-ClC ₆ H ₄	CH ₃	1.5	100	99 (S)
13	7d	p-BrC ₆ H ₄	CH ₃	3	100	99 (S)
14	7d	C ₆ H ₅	C ₂ H ₅	3.5	99	99.5 (S)
15	7d	C ₆ H ₅	PhCH ₂	46	100	98 (S)
16	7d	2-naphthyl	CH ₃	4	98	99.2 (S)
17 ^e	7d	ferrocenyl	CH ₃	5	100	98 (S)
18	7d	2-furyl	CH ₃	5	99	98 (S)
19	7d	2-thienyl	CH ₃	5	98	98 (S)
20 ^f	7d	trans-PhCH=CH	CH ₃	3	100	96 (S)

^a Reactions were conducted at 20–25 °C under 50 atm of H₂ pressure using a 2.0–2.5 M solution in 2-propanol containing **7d** (*S/C* = 5000) and *t*-BuOK (*S/B* = 70). ^b Determined by GC or ¹H NMR. ^c The ee were determined by chiral GC or HPLC. The absolute configuration was determined by comparison of the sign of optical rotation or retention time with literature data. ^d *S/C* = 100 000, at 40 °C. ^e Using a 1.0 M solution in 2-propanol, *S/B* = 50. ^f *S/B* = 50.

complexes for the hydrogenation of a variety of prochiral ketones indicated a good potential for wide application of these spiro diphosphine ligands. Studies of these spiro ligands in other transition metal catalyzed asymmetric reactions are in progress.

Acknowledgment. We thank the National Natural Science Foundation of China, the Major Basic Research Development Program (Grant G2000077506), and the Ministry of Education of China for financial support.

Supporting Information Available: Preparations and properties of compounds **2–6** and **7**, procedures for asymmetric hydrogenation of ketones, GC behavior of chiral alcohols (PDF). This material is available free of charge via the Internet at <http://pubs.acs.org>.

References

- For reviews, see: (a) *Comprehensive Asymmetric Catalysis*; Jacobsen, E. N., Pfaltz, A., Yamamoto, H., Eds.; Springer: Berlin, 1999; Vol. I–III. (b) *Catalytic Asymmetric Synthesis*, 2nd ed.; Ojima, I., Ed.; Wiley: New York, 2000. (c) Lin, G.-Q.; Li, Y.-M.; Chan, A. S. C. *Principles and Applications of Asymmetric Synthesis*; Wiley: New York, 2001.
- Handbook of Enantioselective Catalysis*; Brunner, H., Zettlmeier, W., Eds.; VCH: New York, 1993; Vol. 2.
- Miyashita, A.; Yasuda, A.; Takaya, H.; Toriumi, K.; Ito, T.; Souchi, T.; Noyori, R. *J. Am. Chem. Soc.* **1980**, *102*, 7933.
- (a) Schmid, R.; Foricher, J.; Cereghetti, M.; Schonholzer, P. *Helv. Chim. Acta* **1991**, *74*, 370. (b) Zhang, X.; Uemura, T.; Matsumura, K.; Sayo, N.; Kumobayashi, H.; Takaya, H. *Synlett* **1994**, *501*. (c) Uemura, T.; Zhang, X.; Matsumura, K.; Sayo, N.; Kumobayashi, H.; Ohta, T.; Nozaki, K.; Takaya, H. *J. Org. Chem.* **1996**, *61*, 5510. (d) Gelpke, A. E. S.; Kooijman, H.; Spek, A. L.; Hiemstra, H. *Chem. Eur. J.* **1999**, *5*, 2472. (e) Benincori, T.; Cesari, E.; Piccolo, O.; Sannicolo, F. *J. Org. Chem.* **2000**, *65*, 2043. (f) Benincori, T.; Piccolo, O.; Rizzo, S.; Sannicolo, F. *J. Org. Chem.* **2000**, *65*, 8340. (g) Pai, C.-C.; Lin, C.-W.; Chen, C.-C.; Chan, A. S. C.; Wong, W. T. *J. Am. Chem. Soc.* **2000**, *122*, 11513. (h) Saito, T.; Yokoawa, T.; Ishizaki, T.; Moroi, T.; Sayo, N.; Miura, T.; Kumobayashi, H. *Adv. Synth. Catal.* **2001**, *343*, 264.
- (a) Togni, A.; Breutel, U.; Schnyder, A.; Spindler, F.; Landert, H.; Tijani, A. *J. Am. Chem. Soc.* **1994**, *116*, 4062. (b) Pye, P. J.; Rossen, K.; Reamer, R. A.; Tsou, N. N.; Volante, R. P.; Reider, P. J. *J. Am. Chem. Soc.* **1997**, *119*, 6207. (c) Pye, P. J.; Rossen, K.; Reamer, R. A.; Volante, R. P.; Reider, P. J. *Tetrahedron Lett.* **1998**, *39*, 4441.
- For an example of spiro phosphinite ligands, see: Chan, A. C. S.; Hu, W.-H.; Pai, C.-C.; Lau, C.-P.; Jiang, Y.-Z.; Mi, A.-Q.; Yan, M.; Sun, J.; Lou, R.-L.; Deng, J.-G. *J. Am. Chem. Soc.* **1997**, *119*, 9570.
- SIPHOS = *N*-diethyl(1,1'-spirobiindane-7,7'-diyl)phosphoramidite. (a) Fu, Y.; Xie, J.-H.; Hu, A.-G.; Zhou, H.; Wang, L.-X.; Zhou, Q.-L. *Chem. Commun.* **2002**, *480*. (b) Hu, A.-G.; Fu, Y.; Xie, J.-H.; Zhou, H.; Wang, L.-X.; Zhou, Q.-L. *Angew. Chem., Int. Ed.* **2002**, *41*, 2348.
- (a) Birman, V. B.; Rheingold, A. L.; Lam, K.-C. *Tetrahedron: Asymmetry* **1999**, *10*, 125. (b) Zhang, J.-H.; Liao, J.; Cui, X.; Yu, K.-B.; Deng, J.-G.; Zhu, S.-F.; Wang, L.-X.; Zhou, Q.-L.; Chung, L.-W.; Ye, T. *Tetrahedron: Asymmetry* **2002**, *13*, 1363.
- Uozumi, Y.; Tanahashi, A.; Lee, S.-Y.; Hayashi, T. *J. Org. Chem.* **1993**, *58*, 1945.
- For experimental details, see the Supporting Information.
- (a) Noyori, R.; Takaya, H. *Acc. Chem. Res.* **1990**, *23*, 345. (b) Burk, M. J.; Gross, M. F.; Harper, G. P.; Kalberg, C. S.; Lee, J. R.; Martinez, J. P. *Pure Appl. Chem.* **1996**, *68*, 37. (c) Naota, T.; Takaya, H.; Murahashi, S. *Chem. Rev.* **1998**, *98*, 2599. (d) Ireland, T.; Tappe, K.; Grossheimann, G.; Knochel, P. *Chem. Eur. J.* **2002**, *8*, 843.
- (a) Bakos, J.; Tóth, I.; Heil, B.; Markó, L. *J. Organomet. Chem.* **1985**, *279*, 23. (b) Zhang, X.; Taketomi, T.; Yoshizumi, T.; Kumobayashi, H.; Akutagawa, S.; Mashima, K.; Takaya, H. *J. Am. Chem. Soc.* **1993**, *115*, 3318. (c) Jiang, Q.; Jiang, Y.; Xiao, D.; Cao, P.; Zhang, X. *Angew. Chem., Int. Ed.* **1998**, *37*, 1100.
- (a) Ohkuma, T.; Ooka, H.; Hashiguchi, S.; Ikariya, T.; Noyori, R. *J. Am. Chem. Soc.* **1995**, *117*, 2675. (b) Ohkuma, T.; Ooka, H.; Yamakawa, M.; Ikariya, T.; Noyori, R. *J. Am. Chem. Soc.* **1995**, *117*, 10417. (c) Ohkuma, T.; Ooka, H.; Yamakawa, M.; Ikariya, T.; Noyori, R. *J. Org. Chem.* **1996**, *61*, 4872. (d) Doucet, H.; Ohkuma, T.; Murata, K.; Yokozawa, T.; Kozawa, M.; Katayama, E.; England, A. F.; Ikariya, T.; Noyori, R. *Angew. Chem., Int. Ed.* **1998**, *37*, 1703. (e) Ohkuma, T.; Koizumi, M.; Doucet, H.; Pham, T.; Kozawa, M.; Murata, K.; Katayama, E.; Yokozawa, T.; Ikariya, T.; Noyori, R. *J. Am. Chem. Soc.* **1998**, *120*, 13529. (f) Noyori, R.; Ohkuma, T. *Angew. Chem., Int. Ed.* **2001**, *40*, 40.
- Xyl-BINAP = 2,2'-bis(di-3,5-xylylphosphino)-1,1'-binaphthyl. Tol-BINAP = 2,2'-bis(di-4-tolylphosphino)-1,1'-binaphthyl. DAIPEN = 1,1-dianisyl-2-isopropyl-1,2-ethylenediamine. DPEN = 1,2-diphenylethylenediamine.
- PhanePhos = 4,12-bis(diphenylphosphino)-[2,2]paracyclophane. Xyl-PhanePhos = 4,12-bis(di-3,5-xylylphosphino)-[2,2]paracyclophane. See: Burk, M. J.; Hems, W.; Herzberg, D.; Malan, C.; Zanotti-Gerosa, A. *Org. Lett.* **2000**, *2*, 4173.
- P-Phos = 2,2',6,6'-tetramethoxy-4,4'-bis(diphenylphosphino)-3,3'-bipyridine. See: Wu, J.; Chen, H.; Kwok, W.-H.; Guo, R.-W.; Zhou, Z.-Y.; Yeung, C.-H.; Chan, A. S. C. *J. Org. Chem.* **2002**, *67*, 7908.
- (17) For review, see: Togni, A. *Angew. Chem., Int. Ed. Engl.* **1996**, *35*, 1475. For recent articles, see: (a) Dübner, F.; Knochel, P. *Angew. Chem., Int. Ed.* **1999**, *38*, 379. (b) Schwink, L.; Knochel, P. *Chem. Eur. J.* **1998**, *5*, 950.
- With the catalyst **(S,SS)-7d**, which represents a mismatch in chirality between SDP and DPEN, (*S*)-1-phenylethanol was produced in 28% ee.
- [*((S)*-Tol-BINAP)*Ru*(*((S)*-DAIPEN)*Cl*₂] (87% ee) and [*((S)*-Xyl-PhanePhos)*Ru*(*((R,R)*-DPEN)*Cl*₂] (92% ee) were also used in the hydrogenation of acetylferrocene, see refs 13f and 15.
- For review, see: Togni, A. *Angew. Chem., Int. Ed. Engl.* **1996**, *35*, 1475. For recent articles, see: (a) Dübner, F.; Knochel, P. *Angew. Chem., Int. Ed.* **1999**, *38*, 379. (b) Schwink, L.; Knochel, P. *Chem. Eur. J.* **1998**, *5*, 950.

JA029907I