Phosphine-Catalyzed Reaction of Cyanohydrins with Activated Alkynes

Acétou Siby, Olivier Loreau,* Frédéric Taran*

CEA, IBITECS, Service de Chimie Bioorganique et de Marquage, Gif sur Yvette 91191, France Fax +33(1)69087991; E-mail: frederic.taran@cea.fr *Received 29 March 2009; revised 3 April 2009*

Abstract: A new method for preparing α -cyanoacrylates and α -cyanoacnylates is described. The procedure uses a phosphine-catalyzed α -C addition of cyanide ion, generated in situ from cyanohydrins, to activated alkynes. An unexpected tandem reaction producing benzylidenecyclopentanones is also described.

Key words: cyanoacrylates, alkynes, catalysis, phosphines

 α -Cyanoacrylates and α -cyanoenones are common building blocks for the synthesis of numerous heterocycles; their use as monomers in polymer synthesis is also well established. These products have been mostly synthesized by the Knoevenagel condensation using active methylene compounds such as cyanoacetates and α -cyanoketones.¹ However, α -cyanoenones may be reactive or unstable to base, so that classical Knoevenagel conditions are not always suitable for the preparation of these compounds.² Only few alternative synthetic approaches to α -cyanoenones such as isoxazoline oxidation,³ ring opening of isoxazoles by bases,⁴ and α -cyanoacylation of aldehydes⁵ have been described in literature.

During the course of our work on the preparation of ¹⁴Clabeled cyanoacrylates and their polymer derivatives,⁶ we have been interested in developing a mild procedure based on previous work described by our group using labeled KCN as starting material for the synthesis of α -cyanoenones. We have described a series of phosphine-catalyzed α -addition reactions of pronucleophiles on activated alkynes that allows the formation of polyfunctionalized activated alkenes.⁷ We contemplated using this type of chemistry for the formation of α -cyanoacrylates and α -cyanoenones through α -C-addition of cyanide ion to activated alkynes **1** (Scheme 1). We designed cyanohydrins, easily prepared from reaction of KCN with aldehydes or ketones, as cyanide sources.⁸

Cyanohydrines may indeed react as an indirect pronucleophile upon reaction with the zwitterion intermediate generated by Michael addition of the phosphine catalyst on the activated alkyne. H-abstraction of the cyanohydrin by this intermediate should release cyanide ion that may then add onto the vinyl phosphonium intermediate. Elimination of the phosphine after H⁺ transfer should then afford the desired acrylonitrile **2**. To test the feasibility of this reaction, various cyanohydrins were reacted with ethyl phenylpropiolate (**1a**) in the presence of several phosphines (Table 1).

To our delight, the reaction was found to work through the desired pathway and afforded ethyl *trans*- α -cyanocinnamate (**2a**) at room temperature in acceptable yield (Table 1, entry 1). Higher yields and shorter reaction times are obtained upon heating and by using MePh₂P,



Scheme 1 Proposed route to labeled acrylonitrile 2

SYNTHESIS 2009, No. 14, pp 2365–2370 Advanced online publication: 25.05.2009 DOI: 10.1055/s-0029-1216827; Art ID: Z05209SS © Georg Thieme Verlag Stuttgart · New York

Table 1Optimization of the Reaction of Cyanohydrins with Alkyne1a

		$= \frac{\begin{array}{c} R \\ HO \\ CN \end{array}}{\begin{array}{c} R_{3}P \\ Ca \end{array}} $	t.)	N
	1a		2a C	O ₂ Et
Entry	R R' HO CN	R ₃ P (mol%)	Conditions	Yield of 2a (%) ^a
1	HOCN	Bu ₃ P (20%)	toluene, r.t., 20 h	66
2		Bu ₃ P (20%)	toluene, 110 °C, 1 h	63
3		MePh ₂ P (20%)	toluene, 110 °C, 1 h	72
4		Ph ₃ P (20%)	toluene, 110 °C, 1 h	30
5		MePh ₂ P (5%)	toluene, 110 °C, 6 h	72
6		MePh ₂ P (20%)	<i>i</i> -PrOH, 90 °C, 7 h	40
7		MePh ₂ P (20%)	CH ₂ Cl ₂ , 40 °C, 24 h	72
8	HOCN	MePh ₂ P (20%)	toluene, 110 °C, 7 h	73
9		MePh ₂ P (20%)	toluene, 110 °C, 5 h	61
10	HO' 'CN	MePh ₂ P (20%)	toluene, 110 °C, 4 h	74 ^b

^a Isolated yields.

^b A by-product in 17% yield resulting from the α -O-addition of the cyanohydrin was isolated in this case.

which was found the most active catalyst (compare entry 3 with 2 and 4, Table 1). Among the tested cyanohydrins, acetone cyanohydrin was found the most efficient (compare entries 8, 9, and 10 with 3, Table 1). The other cyanohydrins required longer reaction times; in each cases the formation of the corresponding ketone or aldehyde by-products were observed in the crude reaction mixture.

The reaction proceeded smoothly on a panel of activated alkynes affording the desired α -cyanoacrylates **2a–c** and α -cyanoenones **2d–g** products in moderate to good yields. In the case of alkynes activated by a ketone function, the reaction requires room temperature conditions due to the thermal instability of the resulting products.

Compounds **2** were fully characterized and their structure established by NMR and MS experiments as well as by comparison with published analytical data. By-product **3d** (26% isolated yield) has also been observed when enolizable yne-one starting material **1d** was used. The structure of this by-product (Figure 1) was determined by extensive spectral analysis. The *E*-form of the double bound was assigned by comparison with the published NMR data.⁹



Figure 1 Structure of by-product 3d

Control experiments proved that this cyclopentanone product resulted from a phosphine-catalyzed condensation of alkyne **1d** on cyanoenone **2d** (Scheme 2).

One possible reaction pathway yielding product **3d** from **1d** and **2d** is shown in Scheme 3.

The zwitterionic intermediate **4** produced by conjugate addition of the phosphine catalyst undergoes an intramolecular proton migration (1,3-proton shift) from the methyl group. A nucleophilic attack of the resulting enolate **5** to cyanoacrylate **2d** produces the carbanionic intermediate **6**, which then undergoes intramolecular α -C-addition on the α , β -unsaturated ketone moiety generating ylide **7**. Proton transfer followed by phosphine elimination then produces **3d**.



Scheme 2 Control experiment yielding product 3d

Synthesis 2009, No. 14, 2365–2370 © Thieme Stuttgart · New York



 Table 2
 Phosphine-Catalyzed Reaction of Acetone Cyanohydrin with Activated Alkynes^a

^a Isolated yields.

We therefore decided to take advantage of this new reaction for the straightforward synthesis of poly-functionalized benzylidenecyclopentanones. After several attempts, we developed a two-step, one-pot process that worked in reasonable yields (Scheme 4). The process is based on sequential addition of 1) cyanohydrin to generate α -cyanoacrylonitrile derivative, and 2) the alkyne-one to produce cyclopentatones **3** as a mixture of diastereoisomers.



Scheme 3 Proposed route to cyclopentanone 3d



Scheme 4 Phosphine-catalyzed reaction yielding benzylidenecyclopentanones 3



Scheme 5 Synthesis of ¹³C-labeled cyanoacrylate 2h

In conclusion, we have developed a simple and practical method for the preparation of α -cyanoacrylates and α -cyanoenones starting from activated alkynes and cyanohydrins. Besides the practical advantages of the method, this reaction is interesting for the preparation of labeled acrylonitriles. Preliminary investigations conducted in our laboratory proved that this reaction is highly suitable for the synthesis of labeled α -cyanoacrylates: reaction of eth-yl phenylpropiolate (**1a**) with ¹³C-labeled cyclohexanone cyanohydrin, obtained as previously described,¹⁰ afforded product **2h** in high yield and isotopic enrichment (IE) (Scheme 5).

During the course of this work a new construction of benzylidenecyclopentanone from a phosphine-catalyzed cycloaddition of α -cyanoacrylates and α -cyanoenones with 4-arylbut-3-yn-2-ones has also been highlighted. This allowed us to optimize a tandem phosphine-catalyzed onepot process that afforded polyfunctionalized cyclopentanones in moderate yields.

All reagents were used directly as obtained commercially, unless otherwise noted. Melting points were determined on a Büchi 535 capillary melting point apparatus and are uncorrected. Flash chromatography was carried out on Merck silica gel (40–63 μ m). IR spectra were obtained on a PerkinElmer system 2000 FT-IR spectrophotometer. ¹H NMR (400 MHz) and ¹³C NMR (100 MHz) were measured on a Bruker Avance 400 MHz spectrometer. Electrospray mass spectra were obtained using an ESI/TOF Mariner Mass Spectrometer.

MePh₂P-Catalyzed Synthesis of α -Cyanoacrylates and α -Cyanoenones; General Procedure

To a solution of ethyl arylpriopiolate **1** (1 mmol) and acetone cyanohydrin (100 μ L, 1.1 mmol) in anhyd toluene (5 mL) was added diphenylmethylphosphine (37.3 μ L, 0.2 mmol) under argon. The mixture was heated at reflux and then evaporated under reduced pressure. The residue was purified by flash chromatography on a silica gel column to give α -cyanoacrylates as the final product (Table 2). Identical procedure, but at r.t., was applied to 4-arylbut-3-yn-2-one for the preparation of α -cyanoenones (Table 2).

(E)-2-Cyano-3-phenylacrylic Acid Ethyl Ester (2a)

Purified by column chromatography (CH₂Cl₂-hexane, 50:50); beige solid; mp 50–51 $^{\circ}$ C.

IR (KBr): 3030, 2981, 2223, 1727, 1607, 1572, 1496, 1465, 1444, 1386, 1367, 1301, 1258, 1202, 1089, 1010, 970, 887, 849, 768, 683, 579, 524 $\rm cm^{-1}.$

¹H NMR (400 MHz, CDCl₃): δ = 8.25 (s, 1 H), 8 (d, *J* = 7.2 Hz, 2 H), 7.55 (m, 3 H), 4.4 (q, *J* = 7.1 Hz, 2 H), 1.4 (t, *J* = 7.1 Hz, 3 H).

¹³C NMR (100 MHz, CDCl₃): δ = 162.4, 154.9, 133.2, 131.4, 131.0, 129.2, 115.4, 102.9, 62.6, 14.1.

MS (ESI): $m/z = 202 [M + H]^+$.

(*E*)-2-Cyano-3-(4-methoxyphenyl)acrylic Acid Ethyl Ester (2b) Purified by column chromatography (CH_2Cl_2 -hexane, 70:30); yellow solid; mp 81–82 °C.

IR (KBr): 3415, 3084, 3025, 2993, 2943, 2844, 2577, 2215, 1968, 1912, 1748, 1586, 1562, 1513, 1461, 1431, 1364, 1320, 1263, 1211, 1182, 1127, 1089, 1018, 984, 894, 837, 817, 763, 737, 723, 625, 586 cm⁻¹.

¹H NMR (400 MHz, CDCl₃): $\delta = 8.2$ (s, 1 H), 8 (d, J = 8.9 Hz, 2 H), 7 (d, J = 8.9 Hz, 2 H), 4.4 (q, J = 7.1 Hz, 2 H), 3.9 (s, 3 H), 1.4 (t, J = 7.1 Hz, 3 H).

¹³C NMR (100 MHz, CDCl₃): δ = 163.7, 163.0, 154.3, 133.6, 124.3, 116.1, 55.5, 14.1.

MS (EI, 70 eV): m/z (%) = 231 (100, [M⁺]), 216 (7), 203 (37), 186 (53), 158 (22), 143 (9), 115 (11), 77 (8), 29 (9).

(E)-3-(3,4-Dichlorophenyl)-2-cyanoacrylic Acid Ethyl Ester (2c)

Purified by column chromatography (CH₂Cl₂-hexane, 70:30); beige solid; mp 118–120 °C.

IR (KBr): 3424, 3087, 3065, 3037, 2985, 2938, 2905, 2225, 1921, 1804, 1720, 1611, 1585, 1548, 1473, 1445, 1396, 1366, 1354, 1281, 1261, 1209, 1136, 1094, 1029, 1014, 973, 904, 826 $\rm cm^{-1}.$

¹H NMR (400 MHz, CDCl₃): δ = 8.15 (s, 1 H), 8 (d, *J* = 2.0 Hz, 1 H), 7.9 (dd, *J* = 2.0, 8.4 Hz, 1 H), 7.7 (d, *J* = 8.4 Hz, 1 H), 4.4 (q, *J* = 7.1 Hz, 2 H), 1.4 (t, *J* = 7.1 Hz, 3 H).

¹³C NMR (100 MHz, CDCl₃): δ = 162.7, 151.8, 137.5, 133.7, 132.6, 131.3, 131.1, 129.2, 114.7, 104.8, 62.9, 14.0.

MS (EI, 70 eV): m/z (%) = 273 (12, [M (2 ³⁷Cl)]⁺), 271 (68, [M (³⁷Cl, ³⁵Cl)]⁺), 269 (100, [M (2 ³⁵Cl)]⁺), 245 (9), 243 (50) 241 (75), 228 (7), 226 (41), 224 (63), 196 (39), 161 (63), 124 (20), 75 (18), 50 (9), 29 (56).

HRMS: m/z calcd for $C_{12}H_9^{35}Cl_2O_2$ + Na: 291.9908; found: 291.9901; m/z calcd for $C_{12}H_9^{35}Cl^{37}ClO_2$ + Na: 293.9879; found: 293.9885.

3-Oxo-2-[1-phenylmeth-(*E*)-ylidene]butyronitrile (2d)

Purified by column chromatography (CH₂Cl₂-hexane, 70:30); white solid; mp 81-82 °C.

IR (KBr): 3380, 3030, 3008, 2216, 1697, 1586, 1567, 1494, 1450, 1419, 1363, 1317, 1294, 1243, 1200, 1186, 1081, 1026, 980, 960, 761, 687, 637, 561, 547, 522 cm⁻¹.

¹H NMR (400 MHz, CDCl₃): δ = 8.25 (s, 1 H), 8 (d, J = 7.3 Hz, 2 H), 7.55 (m, 3 H), 2.66 (s, 3 H).

¹³C NMR (100 MHz, CDCl₃): δ = 191.11, 153.1, 133.48, 131.39, 131.26, 129.28, 117.18, 109.64, 27.81.

MS (ESI): $m/z = 172 [M + H]^+$

4,4-Dimethyl-3-oxo-2-[1-phenylmeth-(*E*)-ylidene]pentanenitrile (2e)

Purified by column chromatography (CH₂Cl₂-hexane, 50:50); yellow solid; mp 69–72 °C.

IR (KBr): 3360, 3064, 3020, 2973, 2935, 2876, 2210, 1968, 1897, 1689, 1583, 1565, 1497, 1478, 1446, 1396, 1367, 1343, 1321, 1294, 1214, 1134, 1075, 1035, 1035, 1002, 984, 956, 939, 834, 795, 761, 733, 685, 625, 600, 570, 523 cm⁻¹.

¹H NMR (400 MHz, CDCl₃): δ = 8.23 (s, 1 H), 8.0 (d, J = 7.2 Hz, 2 H), 7.5 (m, 3 H), 1.4 (s, 9 H).

¹³C NMR (100 MHz, CDCl₃): δ = 198.9, 156.1, 132.9, 131.9, 131.1, 129.1, 113.2, 107.31, 44.54, 26.4.

MS (EI, 70 eV): m/z (%) = 213 (30, [M⁺]), 185 (34), 170 (12), 157 (63), 129 (98), 102 (41), 77 (21), 57 (100), 41 (47), 29 (23).

3-Oxo-2-[1-thiophen-2-ylmeth-(E)-ylidene]butyronitrile (2f)

Purified by column chromatography (CH₂Cl₂-heptane, 80:20); yellow solid; mp 79–81 °C.

IR (KBr): 3443, 3082, 3046, 2364, 2330, 2212, 1833, 1717, 1695, 1676, 1579, 1504, 1413, 1360, 1268, 1243, 1201, 1054, 976, 948, 858, 786, 742, 635, 600, 567, 518 cm⁻¹.

¹H NMR (400 MHz, CDCl₃): δ = 8.3 (s, 1 H), 7.8 (dd, *J* = 3.7, 15.8 Hz, 2 H), 7.3 (q, *J* = 4.1 Hz, 1 H), 2.6 (s, 3 H).

¹³C NMR (100 MHz, CDCl₃): δ = 190.9, 144.8, 137.9, 136.2, 135.7, 128.8, 117.5, 106.1, 27.9.

MS (EI, 70 eV): m/z (%) = 177 (100, [M]⁺), 162 (65), 135 (42), 108 (18), 90 (13), 63 (14), 43 (99).

HRMS: *m*/*z* calcd for C₉H₇NOS + Na: 200.0146; found: 200.0150.

(*E*)-3-(4-Methoxyphenyl)-2-(3-methylbenzofuran-2-carbonyl)acrylonitrile (2g)

Purified by column chromatography (CH₂Cl₂-heptane, 80:20); orange solid; mp 102.7–104.6 °C.

IR (KBr): 2958, 2838, 2219, 1726, 1650, 1567, 1488, 1465, 1434, 1385, 1363, 1339, 1292, 1260, 1221, 1196, 1176, 1150, 1098, 1048, 977, 940, 902, 872, 787, 746, 725, 683 cm⁻¹.

¹H NMR (400 MHz, CDCl₃): δ = 8.3 (s, 1 H), 7.7 (m, 3 H), 7.5 (m, 3 H), 7.4 (t, *J* = 8.1 Hz, 1 H), 7.3 (t, *J* = 7.1 Hz, 1 H), 7.1 (dd, *J* = 1.98, 8.2 Hz, 1 H), 3.9 (s, 3 H), 2.6 (s, 3 H).

¹³C NMR (100 MHz, CDCl₃): δ = 177.6, 159.9, 155, 154.3, 146.6, 133.2, 130.1, 129.1, 120.0, 116.2, 114.5, 112.5, 110.2, 55.4, 9.9.

MS (EI, 70 eV): *m*/*z* (%) = 317 (100, [M⁺]), 302 (19), 286 (69), 274 (10), 251 (11), 210 (18), 159 (57), 103 (29), 77 (37).

HRMS: *m/z* calcd for C₂₀H₁₅NO₃ + Na: 340.0950; found: 340.0943.

(*E*)-2-¹³C-Cyano-3-phenylacrylic Acid Ethyl Ester (2h)

¹H NMR (400 MHz, CDCl₃): δ = 8.3 (d, *J* = 13.9 Hz, 1 H), 8 (d, *J* = 7.4 Hz, 2 H), 7.5 (m, 3 H), 4.4 (q, *J* = 7.1 Hz, 2 H), 1.5 (t, *J* = 7.1 Hz, 3 H).

¹³C NMR (100 MHz, CDCl₃): δ = 162.4, 155, 133.2, 131.4, 131, 129.2, 115.4, 102.9 (d, J = 84 Hz), 62.7, 14.1.

MS (ESI): $m/z = 203 [M + H]^+$.

MePh₂P-Catalyzed Construction of Polyfunctionalized Cyclopentanones; General Procedure

To a solution of the activated alkyne (1 mmol) and acetone cyanohydrin (100 μ L, 1.1 mmol) in anhyd toluene (5 mL) was added diphenylmethylphosphine (37.3 μ L, 0.2 mmol) under argon. The mixture was stirred at r.t. for 12 h, then 4-arylbut-3-yn-2-one (1 mmol, 1 equiv) dissolved in anhyd toluene (5 mL) was added dropwise over 5 h (syringe pump, 0.2 equiv/h). After the complete addition, the mixture was further stirred overnight at r.t. and then concentrated under reduced pressure. The residue was purified by flash chromatography on a silica gel column to give the final product.

1-Acetyl-2-benzylidene-3-oxo-5-phenylcyclopentanecarbonitrile (3d)

Purified by column chromatography (CH_2Cl_2 -hexane, 70:30); colorless oil.

IR (film): 3428, 3062, 3032, 2925, 2855, 2240, 1963, 1720, 1619, 1575, 1494, 1450, 1410, 1360, 1312, 1287, 1234, 1180, 1146, 1108, 1080, 1029, 1001, 966, 927, 827, 759, 736, 698, 662, 619, 606 cm⁻¹.

¹H NMR (400 MHz, CDCl₃): δ = 7.9 (s, 1 H), 7.4 (m, 9 H), 7.3 (dd, *J* = 1.8, 5.2 Hz, 2 H), 3.7 (dd, *J* = 7.2, 12.9 Hz, 1 H), 3.1 (dd, *J* = 12.9, 17.6 Hz, 1 H), 2.9 (dd, *J* = 7.3, 17.6 Hz, 1 H), 2.05 (s, 3 H).

¹³C NMR (100 MHz, CDCl₃): δ = 199, 198.5, 140.12, 134, 132.99, 131.0, 129.9, 129.03, 129.15, 128.07, 107, 62.04, 50.11.

MS (EI, 70 eV): m/z (%) = 315 (13, [M]⁺), 273 (100), 244 (22), 230 (15), 196 (13), 167 (6), 140 (9), 103 (8), 77 (10), 43 (73).

HRMS: m/z calcd for C₂₁H₁₇NO₂ + Na: 338.1157; found: 338.1158.

2-Benzylidene-1-cyano-3-oxo-5-phenylcyclopentanecarboxylic Acid Ethyl Ester (3e)

Purified by column chromatography (EtOAc-heptane, 75:25), yellow oil.

IR (film): 3446, 3082, 3003, 2935, 1744, 1729, 1625, 1575, 1493, 1469, 1452, 1404, 1367, 1316, 1264, 1230, 1177, 1108, 933, 909, 853, 779, 762, 694, 649, 623, 604, 546, 524 cm⁻¹.

¹H NMR (400 MHz, CDCl₃): δ = 7.9 (s, 1 H), 7.2–7.6 (m, 10 H), 3.9 (m, 3 H), 3.1 (dd, *J* = 13.6, 17.6 Hz, 1 H), 2.9 (dd, *J* = 7, 17.6 Hz, 1 H), 1 (t, *J* = 7.15 Hz, 3 H).

 ^{13}C NMR (100 MHz, CDCl₃): δ = 199.2, 166.05, 139.9, 143.3, 132.84, 132.6, 130.83, 130.15, 129.0, 128.9, 128.7, 127.93, 115.7, 63.2, 56.4, 50.4, 40.2, 13.6.

MS (EI, 70 eV): m/z (%) = 345 (36, [M]⁺), 299 (12), 272 (35), 241 (91), 213 (100), 195 (35), 168 (24), 141 (46), 114 (16), 103 (17), 91 (9), 77 (20), 29 (35).

HRMS: *m/z* calcd for C₂₂H₁₉NO₃ +Na: 368.1263; found: 368.1259.

2-(4-Bromobenzylidene)-1-cyano-3-oxo-5-phenylcyclopentanecarboxylic Acid Ethyl Ester (3f)

Purified by column chromatography (CH₂Cl₂-heptane, 80:20), yellow oil.

IR (film): 3417, 3091, 3033, 2985, 2930, 2363, 2242, 1888, 1736, 1719, 1618, 1582, 1562, 1543, 1487, 1455, 1402, 1363, 1305, 1264, 1226, 1178, 1152, 1108, 1072, 1050, 1025, 1007, 946, 922, 852, 820, 769, 727, 697, 672, 631, 606, 562, 521 cm⁻¹.

¹H NMR (400 MHz, CDCl₃): δ = 7.8 (s, 1 H), 7.6 (d, *J* = 8.4 Hz, 2 H), 7.4 (m, 5 H), 7.3 (m, 2 H), 3.9 (m, 3 H), 3.1 (dd, *J* = 13.4, 17.2 Hz, 1 H), 2.9 (dd, *J* = 7.2, 17.2 Hz, 1 H), 1.1 (t, *J* = 7.2 Hz, 3 H).

¹³C NMR (100 MHz, CDCl₃): δ = 199, 166, 134.5, 134.2, 133.3, 132, 131.6, 131.5, 129.1, 128.9, 127.9, 125.5, 115.4, 63.4, 56.3, 50.4, 13.6.

MS (EI, 70 eV): m/z (%) = 425 (33, [M (⁸¹Br)]⁺), 423 (34, [M (⁷⁹Br)]⁺), 379 (17), 377 (16), 352 (35), 350 (35), 321 (77), 319 (77), 293 (60), 291 (60), 270 (13), 243 (11), 229 (41), 212 (100), 202 (16), 167 (13), 139 (24), 104 (37), 78 (27), 29 (55).

1-Cyano-2-(4-methoxybenzylidene)-3-oxo-5-phenylcyclopentanecarboxylic Acid Ethyl Ester (3g)

Purified by column chromatography (CH₂Cl₂-heptane, 80:20); yellow oil.

IR (film): 3548, 3059, 2961, 2934, 2872, 2840, 2242, 2099, 1889, 1728, 1664, 1621, 1577, 1516, 1494, 1463, 1448, 1391, 1367, 1293,

¹H NMR (400 MHz, CDCl₃): δ = 7.9 (s, 1 H), 7.6 (m, 2 H), 7.5 (m, 3 H), 7.3 (d, *J* = 8.7 Hz, 2 H), 6.9 (d, *J* = 8.7 Hz, 2 H), 3.9 (m, 6 H), 3.1 (dd, *J* = 3.7, 17.5 Hz, 1 H), 2.9 (dd, *J* = 7.1, 17.5 Hz, 1 H), 1 (t, *J* = 7.2 Hz, 3 H).

¹³C NMR (100 MHz, CDCl₃): δ = 199.3, 166.1, 160, 139.8, 132.9, 132.7, 130.8, 130.1, 129.1, 128.7, 126.2, 115.8, 114.2, 63.1, 56.6, 55.2, 50, 40.4, 13.6.

MS (EI, 70 eV): m/z (%) = 375 (61, [M]⁺), 329 (36), 302 (21), 241 (19), 213 (48), 195 (23), 134 (100), 119 (13), 91 (15).

Acknowledgment

We warmly thank Dr. E. Zekri, E. Leonce, and D. Buisson for experimental assistance with NMR, IR, and MS measurements.

References

- (a) Clark, J. H. *Chem. Rev.* **1980**, *80*, 429. (b) Gugelchuk,
 M. M.; Hart, D. J.; Tsai, Y. M. *J. Org. Chem.* **1981**, *46*,
 3671. (c) Brunskill, J. S. K.; De, A.; Ewing, D. F. *J. Chem. Soc., Perkin Trans. 1* **1978**, 629.
- (2) (a) Yoshimatsu, M.; Yamaguchi, S.; Matsubara, Y. J. Chem. Soc., Perkin Trans. 1 2001, 2560. (b) Brillon, D.; Sauvé, G. J. Org. Chem. 1992, 57, 1838.
- (3) Padwa, A.; Kline, D. N.; Perumattam, J. *Tetrahedron Lett.* 1987, 28, 913.
- (4) Ciller, J. A.; Martin, N.; Seoane, C.; Soto, J. L. J. Chem. Soc., Perkin Trans. 1 1985, 2581.
- (5) Yoshimatsu, M.; Timura, Y. J. Org. Chem. 2002, 67, 5678.
- (6) (a) Garcia-Garcia, E.; Gil, S.; Andrieux, K.; Desmaële, D.; Nicolas, V.; Taran, F.; Georgin, D.; Andreux, J. P.; Roux, F.; Couvreur, P. *Cell. Mol. Life Sci.* 2005, *62*, 1.
 (b) Ryoung Kim, H.; Gil, S.; Andrieux, K.; Nicolas, V.; Appel, M.; Chacun, H.; Desmaële, D.; Taran, F.; Georgin, D.; Couvreur, P. *Cell. Mol. Life Sci.* 2007, *64*, 356. (c) Kim, H. R.; Andrieux, K.; Gil, S.; Taverna, M.; Chacun, H.; Desmaële, D.; Taran, F.; Georgin, D.; Couvreur, P. *Biomacromolecules* 2007, *8*, 793. (d) Kim, H. R.; Andrieux, K.; Chacun, H.; Appel, M.; Desmaële, D.; Taran, F.; Georgin, D.; Couvreur, P.; Taverna, M. *Electrophoresis* 2007, *28*, 2252.
- (7) (a) Lecerclé, D.; Sawicki, M.; Taran, F. Org. Lett. 2006, 8, 4283. (b) Gabillet, S.; Lecerclé, D.; Loreau, O.; Dézard, S.; Gomis, J.-M.; Taran, F. Synthesis 2007, 515. (c) Gabillet, S.; Lecerclé, D.; Loreau, O.; Carboni, M.; Dézard, S.; Gomis, J.-M.; Taran, F. Org. Lett. 2007, 9, 3925. (d) Hanedanian, M.; Loreau, O.; Taran, F.; Mioskowski, C. Tetrahedron Lett. 2004, 45, 7035. (e) Hanedanian, M.; Loreau, O.; Sawicki, M.; Taran, F. Tetrah edron 2005, 61, 2287.
- (8) Cyanohydrins have been successfully used as a cyanide source in palladium-catalyzed cyanation reactions, for example: Sundermeier, M.; Zapf, A.; Beller, M. Angew. Chem. Int. Ed. 2003, 42, 1661.
- (9) (a) Zhu, J.-L.; Ko, Y.-C.; Kuo, C.-W.; Shia, K.-S. *Synlett* 2007, 1274. (b) Kreher, U. P.; Rosamilia, A. E.; Raston, C. L.; Scott, J. L.; Strauss, C. R. *Org. Lett.* 2003, *5*, 3107.
- (10) Carr, G.; Whittaker, D. J. Chem. Soc., Perkin Trans. 2 1989, 359.