## ChemComm

## COMMUNICATION

ROYAL SOCIETY OF CHEMISTRY

View Article Online View Journal | View Issue

Cite this: Chem. Commun., 2014, 50, 8912

Received 8th May 2014, Accepted 17th June 2014

DOI: 10.1039/c4cc03479a

www.rsc.org/chemcomm

Phosphine-catalyzed asymmetric [4+1] annulation of activated  $\alpha$ , $\beta$ -unsaturated ketones with Morita– Baylis–Hillman carbonates: enantioselective synthesis of spirooxindoles containing two adjacent quaternary stereocenters<sup>†</sup>

Fang-Le Hu, Yin Wei and Min Shi\*

The asymmetric [4+1] annulation of activated  $\alpha$ , $\beta$ -unsaturated ketones with MBH carbonates catalyzed by bifunctional thiourea-phosphine catalysts derived from an axially chiral binaphthyl scaffold has been developed, giving spirooxindoles with two adjacent quaternary stereocenters in good yields with high enantioselectivities and moderate diastereoselectivities under mild conditions.

Nucleophilic phosphines are widely used as Lewis base catalysts in organic reactions. On the basis of phosphine catalysis, a variety of unique carbo- and heterocyclic frameworks, some of which are commonly present in bioactive natural products and medicinally important compounds, can be easily acquired.<sup>1</sup> It has grown significantly over the past decade and now it is an active area of organocatalysis.<sup>2</sup> Recently, Morita-Baylis-Hillman (MBH) adducts have aroused organic chemists' interest after Lu's pioneering work on the intra- and intermolecular [3+n] annulations (n = 2, 3, 4, and 6) using MBH carbonates as 1,3-dipoles with various activated alkenes catalyzed by a trivalent phosphine,<sup>3</sup> affording the corresponding cycloadducts in good yields and high regioselectivities under mild conditions.<sup>4</sup> Furthermore, recently some examples of asymmetric [3+2] annulation of MBH carbonates with activated alkenes have also been disclosed.<sup>5</sup> More recently, Zhang, Huang, He and Shi as well as their co-workers have also developed [4+1] annulations utilizing MBH carbonates as 1,1-dipoles to obtain the heterocyclic products in high yields, respectively.<sup>6</sup> In this aspect, we have further developed a series of bifunctional thiourea-phosphine catalysts derived from natural amino acids and applied them in asymmetric [3+2] annulations of MBH carbonates with electrondeficient alkenes, furnishing the corresponding highly functionalized cyclopentenes in good yields and ee values.<sup>7</sup> We have also reported the first highly enantioselective [4+1] annulations between

dicyano-2-methylenebut-3-enoates and MBH carbonates catalyzed by chiral bifunctional phosphine, providing an efficient synthesis of highly functionalized cyclopentenes bearing one all-carbon quaternary stereogenic center.<sup>8a</sup> To the best of our knowledge, no asymmetric [4+1] annulation of activated alkenes with MBH carbonates has been further disclosed after our previous report.<sup>8b</sup>

The spirooxindole backbones are commonly present in a number of biologically active compounds as well as natural products.<sup>9</sup> They have recently become one of the most attractive synthetic targets and many kinds of elegant synthetic approaches have been developed for their syntheses.<sup>10</sup> In continuation of our efforts to seek efficient approaches toward the construction of the spirooxindole motif,<sup>11</sup> we wish to report chiral phosphine catalyzed asymmetric [4+1] annulation of MBH carbonates with activated  $\alpha$ , $\beta$ -unsaturated ketones, effectively providing 2*H*-spiro[furan-3,3'-indolin]-2'-one derivatives with two adjacent quaternary stereocenters.

Initially, we examined the reaction of activated  $\alpha$ ,  $\beta$ -unsaturated ketone 1a (0.1 mmol) with MBH carbonate 2a (0.11 mmol, 1.1 equiv.) catalyzed by a bifunctional chiral phosphine catalyst TP1 derived from a natural amino acid in toluene (2 mL) at room temperature. The desired [4+1] cycloadduct 3a was obtained in 92% total yield along with 1:2 dr within 12 h, the ee value of the major product is 11%, however, the minor product's ee value is 86% (Table 1, entry 1). In order to improve the ee values, multifunctional thiourea-phosphines with more sterically hindered substituents TP2, TP3 and TP4 were further tested in this reaction, providing the spirooxindole 3a containing two adjacent quaternary stereocenters in good yields along with moderate stereochemical outcomes (up to 3:1 dr and up to 69% ee) (Table 1, entries 2-4). We next utilized bifunctional chiral phosphines derived from an axially chiral binaphthyl scaffold TP5-TP7 as catalysts for this reaction. TP6 is the best catalyst in terms of enantioselectivity, affording 3a in 97% total yield along with 3:1 dr, and the major product's ee value is 96% (Table 1, entries 5-7). During the investigation of the nucleophilic phosphine catalysis, we have noticed that substituents in the phosphorus center of chiral phosphine catalysts could remarkably influence the diastereoselectivities,<sup>11j,12</sup> TP8-TP10 were therefore synthesized and subjected to

State Key Laboratory of Organometallic Chemistry, Shanghai Institute of Organic Chemistry, Chinese Academy of Sciences, 354 Fenglin Road, Shanghai 200032, P. R. China. E-mail: mshi@mail.sioc.ac.cn

<sup>†</sup> Electronic supplementary information (ESI) available: Experimental procedures, characterization data for new compounds. CCDC 991742 and 990849. For ESI and crystallographic data in CIF or other electronic format see DOI: 10.1039/c4cc03479a

Table 1 Optimization of reaction conditions for the asymmetric  $\left[ 4\!+\!1 \right]$  annulation



<sup>*a*</sup> The reaction carried out using **1a** (0.1 mmol), **2a** (0.2 mmol) and catalyst (0.02 mmol) in solvent (2.0 mL). <sup>*b*</sup> Isolated yield. <sup>*c*</sup> Determined by chiral HPLC analysis.

this reaction. It was found that the bifunctional chiral phosphine catalysts **TP8** and **TP9**, having the methyl groups on the 3,5positions of the benzene ring, showed poor performance in terms of enantioselectivity (Table 1, entries 8 and 9). Increasing the steric hindrance of the thiourea moiety did not further increase the enantioselectivity, affording **3a** in total 96% yield along with 6:1 dr, 75% and 95% ee, respectively (Table 1, entry 10). Finally, the solvent effects have also been examined using **TP6** as the catalyst. In dichloromethane (DCM), similar results were obtained, but in fluorobenzene, benzotrifluoride or *p*-xylene, the ee values of **3a** (major diastereomer) decreased remarkably (Table 1, entries 11–14). Therefore, the best reaction conditions have been determined to be **TP6** (20 mol%) in toluene at room temperature within 12 hours.

With the optimized reaction conditions in hand, we next turned our attention to the scope and limitations of this reaction using a variety of activated  $\alpha$ , $\beta$ -unsaturated ketones **1** with MBH carbonates **2**. It was found that a variety of *N*-protected  $\alpha$ , $\beta$ -unsaturated ketones were suitable for this reaction, producing the desired products **3b**, **3c**, **3d** and **3e** in high yields along with

 Table 2
 Substrate scope of the asymmetric [4+1] annulation of 1 with 2

$R^{1} \xrightarrow{\text{NC}} R^{2} \xrightarrow{\text{Ar}^{1}} CO_{2}Et \xrightarrow{\text{TP6} (20 \text{ mol}\%)} R_{1} \xrightarrow{\text{Ar}^{1}} CO_{2}Et \xrightarrow{\text{NC}} R^{3}$							
Entry <sup>a</sup>	R <sup>1</sup>	$\mathbb{R}^2$	Ar <sup>1</sup>	R <sup>3</sup>	Yield <sup>b</sup> (%)	ee <sup>c</sup> (%)	dr <sup>c</sup> (%)
1	1 <b>b</b> , H	Me	$C_6H_5$	<b>2a</b> , 4-NO <sub>2</sub>	<b>3b</b> , 97	98	2/1
2	1c, H	MOM	$C_6H_5$	<b>2a</b> , 4-NO <sub>2</sub>	3c, 92	98	2/1
3	1d, H	Allyl	$C_6H_5$	$2a, 4-NO_2$	3d, 99	98	3/1
4	1e, H	Ph	$C_6H_5$	$2a, 4-NO_2$	3e, 90	97	2/1
5	<b>1f</b> , H <sup>f</sup>	Н	$C_6H_5$	$2a, 4-NO_2$	3f, 65	95	$3/1^{d}$
6	<b>1g</b> , 5-Me <sup>f</sup>	Н	$C_6H_5$	$2a, 4-NO_2$	3g, 61	95	$3/1^{d}$
7	1h, 5-F <sup>f</sup>	н	C <sub>6</sub> H <sub>5</sub>	$2a, 4-NO_2^2$	3h, 58	90	$3/1^{d}$
8	<b>1i</b> , 7-Br <sup>f</sup>	н	$C_6H_5$	$2a, 4-NO_2$	3i, 62	96	$3/1^{d}$
9	1j, H <sup>f</sup>	н	3-MeC <sub>6</sub> H <sub>4</sub>	$2a, 4-NO_2$	3i, 75	96	$4/1^{d}$
10	$1\mathbf{k}, \mathbf{H}^{f}$	н	$4-ClC_6H_4$	$2a, 4-NO_2$	3k, 51	90	$3/1^{d}$
11	1a. H <sup>f</sup>	Bn	C <sub>6</sub> H <sub>5</sub>	<b>2b.</b> 4-Cl <sub>2</sub>	<b>31</b> . 90	97	3/1
12	1a, H <sup>f</sup>	Bn	C <sub>6</sub> H <sub>5</sub>	2b, 4-CN	<b>3m</b> , 98	98	3/1
13	$1a, H^e$	Bn	C <sub>6</sub> H <sub>5</sub>	2d, 3,4-Cl <sub>2</sub>	<b>3n</b> , 85	81	1/1
14	<b>1f</b> , H <sup>f</sup>	Bn	C <sub>6</sub> H <sub>5</sub>	2e, 3,5-Cl <sub>2</sub>	30, 89	96	2/1
15	<b>1f</b> , H	Bn	$C_6H_5$	2 <b>f</b> , 4-Me	NR		

 $^{a}$  **1** (0.01 mmol), **1** (0.11 mmol) and **TP6** (0.02 mmol) were stirred in 2 mL of toluene within 12 h.  $^{b}$  Isolated yield.  $^{c}$  Determined by chiral HPLC.  $^{d}$  Determined by <sup>1</sup>H NMR analysis.  $^{e}$  The reaction mixtures were stirred for 4 days.  $^{f}$  The reaction mixtures were stirred for 48 h.

moderate de and high ee values (up to 3:1 dr, 97-98% ee) (Table 2, entries 1-4). To our delight, as for the N-unprotected  $\alpha,\beta$ -unsaturated ketone **1f**, the corresponding [4+1] annulation product 3f was obtained in 65% yield along with 95% ee and 3:1 dr, although the reactivity of 3f was reduced and the reaction was conducted for 48 h (Table 2, entry 5). When other N-unprotected  $\alpha,\beta$ -unsaturated ketones were used in this reaction, the corresponding products were acquired in moderate yields (up to 62% yield) along with excellent enantioselectivities (90-96% ee) and moderate dr values (up to 3:1 dr), respectively (Table 2, entries 6–8). When substituents were introduced on the Ar<sup>1</sup> group, the reaction could also proceed smoothly to give the desired products in up to 75% yield and up to 4:1 dr (96% ee, respectively) (Table 2, entries 9 and 10). Finally, substrates with electron-withdrawing substituents on the aromatic ring of MBH carbonates were also tested in this reaction, giving the desired products in good yields along with moderate de values and high ee values (Table 2, entries 11-14). Only in the case of MBH carbonate 2e, the corresponding adduct 30 was obtained in a relatively low ee value (81% ee), presumably due to the influence of steric hindrance (Table 2, entry 13). Unfortunately, when the methyl group was introduced into the aromatic ring of MBH carbonate, the reactivity of 2f decreased remarkably with no desired product being formed (Table 2, entry 15).

It should be pointed out that the two diastereoisomers 3f-3k can be isolated by column chromatography and the structures of minor diastereomers 3f'-3k' have also been determined by <sup>1</sup>H and <sup>13</sup>C NMR spectroscopy, MS and HRMS analyses. The structure of **1c** has been determined by X-ray diffraction to be *Z* configuration and the absolute configuration of **3i** has also been unambiguously determined by X-ray diffraction. Their CIF data are presented in the ESI.<sup>†</sup>

The plausible mechanism of these [4+1] annulations is proposed in Scheme S1 in the ESI† on the basis of previous literature,<sup>3a,d,6,8,13</sup> and the plausible transition state has also been presented in Fig. S1 in the ESI.†

In conclusion, the asymmetric [4+1] annulation of activated  $\alpha$ , $\beta$ -unsaturated ketones derived from isatin with MBH carbonates catalyzed by chiral bifunctional phosphine has been developed for the first time, which provides highly efficient and enantioselective synthesis of spirooxindoles containing two adjacent quaternary stereocenters. Further efforts are in progress to develop the use of this reaction in organic synthesis.

We thank the Shanghai Municipal Committee of Science and Technology (11JC1402600), the National Basic Research Program of China (973)-2010CB833302, and the National Natural Science Foundation of China (20472096, 21372241, 21361140350, 20672127, 21102166, 21121062, 21302203 and 20732008).

## Notes and references

- For selected reviews, see: (a) S. E. Denmark and G. L. Beutner, *Angew. Chem., Int. Ed.*, 2008, 47, 1560; (b) X. Lu, C. Zhang and Z. Xu, *Acc. Chem. Res.*, 2001, 34, 535; (c) J. L. Methot and W. R. Roush, *Adv. Synth. Catal.*, 2004, 346, 1035; (d) A. Marinetti and A. Voituriez, *Synlett*, 2010, 174; (e) S. Xu and Z. He, *Sci. Sin.: Chim.*, 2010, 40, 856; (f) Z.-M. Wang, X.-Z. Xu and O. Kwon, *Chem. Soc. Rev.*, 2014, 43, 2927.
- 2 For selected reviews, see: (a) L.-W. Ye, J. Zhou and Y. Tang, Chem. Soc. Rev., 2008, 37, 1140; (b) B. J. Cowen and S. J. Miller, Chem. Soc. Rev., 2009, 38, 3102; (c) Y. C. Fan and O. Kwon, Chem. Commun., 2013, 49, 11588 and references therein.
- 3 (a) Y. Du, X. Lu and C. Zhang, Angew. Chem., Int. Ed., 2003, 42, 1035;
  (b) Y. Du, J. Feng and X. Lu, Org. Lett., 2005, 7, 1987; (c) J. Feng, X. Lu, A. Kong and X. Han, Tetrahedron, 2007, 63, 6035; (d) S. Zheng and X. Lu, Org. Lett., 2008, 10, 4481; (e) S. Zheng and X. Lu, Tetrahedron Lett., 2009, 50, 4532; (f) S. Zheng and X. Lu, Org. Lett., 2009, 11, 3978.
- 4 (a) C.-W. Cho, J.-R. Kong and M. J. Krische, Org. Lett., 2004, 6, 1337; (b) C.-W. Cho and M. J. Krische, Angew. Chem., Int. Ed., 2004, 43, 6689; (c) B. M. Trost, R. M. Machacek and H. C. Tsui, J. Am. Chem. Soc., 2005, 127, 7014; (d) Y.-S. Du, X.-L. Han and X.-Y. Lu, Tetrahedron Lett., 2004, 45, 4967; (e) D. J. V. C. van Steenis, T. Marcelli, M. Lutz, A. L. Spek, J. H. van Maarseveen and H. Hiemstra, Adv. Synth. Catal., 2007, 349, 281; (f) Y.-Q. Jiang, Y.-L. Shi and M. Shi, J. Am. Chem. Soc., 2008, 130, 7202; (g) H.-L. Cui, X. Feng, J. Peng, J. Lei, K. Jiang and Y.-C. Chen, Angew. Chem., Int. Ed., 2009, 48, 5737; (h) J. Peng, X. Huang, H.-L. Cui and Y.-C. Chen, Org. Lett., 2010, 12, 4260; (i) T. Furukawa, T. Nishimine, E. Tokunaga, K. Hasegawa, M. Shiro and N. Shibata, Org. Lett., 2011, 13, 3972; (j) T. Furukawa, J. Kawazoe, W. Zhang, T. Nishimine, E. Tokunaga, T. Matsumoto, M. Shiro and N. Shibata, Angew. Chem., Int. Ed., 2011, 50, 9684; (k) F. Zhong, J. Luo, G. Y. Chen, X. Dou and Y. Lu, J. Am. Chem. Soc., 2012, 134, 10222; (l) G.-Y. Chen, F. Zhong and Y. Lu, Org. Lett., 2012,

14, 3955; (*m*) B. Wang, X. Companyó, J. Li, A. Moyano and R. Rios, *Tetrahedron Lett.*, 2012, 53, 4124; (*n*) R. Zhou, J.-F. Wang, H.-B. Song and Z.-J. He, *Org. Lett.*, 2011, 13, 580.

- 5 (a) B. Tan, N. R. Candeias and C. F. Barbas III, J. Am. Chem. Soc., 2011, 133, 4672; (b) F.-R. Zhong, X.-Y. Han, Y.-Q. Wang and Y.-X. Lu, Angew. Chem., Int. Ed., 2011, 50, 7837; (c) Q.-G. Wang, S.-F. Zhu, L.-W. Ye, C.-Y. Zhou, X.-L. Sun, Y. Tang and Q.-L. Zhou, Adv. Synth. Catal., 2010, 352, 1914; (d) Y. Wang, L. Liu, T. Zhang, N.-J. Zhong, D. Wang and Y.-J. Chen, J. Org. Chem., 2012, 77, 4143; (e) F.-R. Zhong, G.-Y. Chen, X.-Y. Han, W.-J. Yao and Y.-X. Lu, Org. Lett., 2012, 14, 3764; (f) J. Peng, X. Huang, L. Jiang, H.-L. Cui and Y.-C. Chen, Org. Lett., 2011, 13, 4584; (g) N.-J. Zhong, F. Wei, Q.-Q. Xuan, L. Liu, D. Wang and Y.-J. Chen, Chem. Commun., 2013, 49, 11071.
- 6 (a) Z.-L. Chen and J.-L. Zhang, Chem. Asian J., 2010, 5, 1542;
  (b) P.-Z. Xie, Y. Huang and R.-Y. Chen, Org. Lett., 2010, 12, 3768;
  (c) J.-J. Tian, R. Zhou, H.-B. Song and Z.-J. He, J. Org. Chem., 2011, 76, 2374;
  (d) P.-Z. Xie, E.-Q. Li, J. Zheng, X. Li, Y. Huang and R.-Y. Chen, Adv. Synth. Catal., 2013, 355, 161-169;
  (e) W. Yuan, H.-F. Zheng, Z.-H. Yu, Z.-L. Tang and D.-Q. Shi, Eur. J. Org. Chem., 2014, 583-591.
- 7 (a) H.-P. Deng, Y. Wei and M. Shi, Org. Lett., 2011, 13, 3348;
  (b) H.-P. Deng, Y. Wei and M. Shi, Adv. Synth. Catal., 2012, 354, 783;
  (c) H.-P. Deng, D. Wang, Y. Wei and M. Shi, Beilstein J. Org. Chem., 2012, 8, 1098;
  (d) F.-L. Hu, Y. Wei and M. Shi, Tetrahedron, 2012, 68, 7911.
- 8 (a) X.-N. Zhang, H.-P. Deng, L. Huang, Y. Wei and M. Shi, *Chem. Commun.*, 2012, **48**, 8664; (b) X.-Y. Han, W.-J. Yao, T.-L. Wang, Y.-R. Tan, Z.-Y. Yan, J. Kwiatkowski and Y.-X. Lu, *Angew. Chem., Int. Ed.*, 2014, **53**, 5643–5647.
- 9 (a) H. Lin and S. J. Danishefsky, Angew. Chem., Int. Ed., 2003, 42, 36;
  (b) M. M.-C. Lo, C. S. Neumann, S. Nagayama, E. O. Perlstein and S. L. Schreiber, J. Am. Chem. Soc., 2004, 126, 16077; (c) G. C. Bignan, K. Battista, P. J. Connolly, M. J. Orsini, J. C. Liu, S. A. Middleton and A. B. Reitz, Bioorg. Med. Chem. Lett., 2005, 15, 5022; (d) C. V. Galliford and K. A. Scheidt, Angew. Chem., Int. Ed., 2007, 46, 8748; (e) S. Kotha, A. C. Deb, K. Lahiri and E. Manivannan, Synthesis, 2009, 165.
- 10 For selected reviews, see: (a) A. B. Dounay and L. E. Overman, Chem. Rev., 2003, 103, 2945; (b) C. V. Galliford and K. A. Scheidt, Angew. Chem., Int. Ed., 2007, 46, 8748; (c) G. S. Singh, M. Dhooghe and N. De Kimpe, Chem. Rev., 2007, 107, 2080; (d) F. Zhou, Y.-L. Liu and J. Zhou, Adv. Synth. Catal., 2010, 352, 1381; (e) K. Shen, X. Liu, L. Lin and X.-M. Feng, Chem. Sci., 2012, 3, 327; (f) J. E. M. N. Klein and R. J. K. Taylor, Eur. J. Org. Chem., 2011, 6821; (g) G. S. Singh and Z. Y. Desta, Chem. Rev., 2012, 112, 6104; (h) N. R. Ball-Jones, J. J. Badillo and A. K. Franz, Org. Biomol. Chem., 2012, 10, 5165.
- (a) H.-B. Yang and M. Shi, Org. Biomol. Chem., 2012, 10, 8236;
  (b) Z. Lian and M. Shi, Org. Biomol. Chem., 2012, 10, 8048;
  (c) Z. Lian and M. Shi, Eur. J. Org. Chem., 2012, 581; (d) Z. Lian and M. Shi, Tetrahedron, 2012, 68, 2401; (e) H.-B. Yang, X.-Y. Guan, Y. Wei and M. Shi, Eur. J. Org. Chem., 2012, 2792; (f) D. Du, Y. Jiang, Q. Xu and M. Shi, Adv. Synth. Catal., 2013, 355, 2249; (g) Q. Wang, Z. Lian, Q. Xu and M. Shi, Adv. Synth. Catal., 2013, 355, 3344;
  (h) L.-Y. Mei, Y. Wei, Q. Xu and M. Shi, Organometallics, 2012, 31, 7591; (i) X.-N. Zhang, G.-Q. Chen, X. Dong, Y. Wei and M. Shi, Adv. Synth. Catal., 2013, 355, 3351; (j) F.-L. Hu, Y. Wei and M. Shi, Adv. Synth. Catal., 2014, 356, 736.
- 12 Y.-L. Yang, C.-K. Pei and M. Shi, Org. Biomol. Chem., 2011, 9, 3349.
- 13 R. Zhou, C. Wang, H. Song and Z. He, Org. Lett., 2010, 12, 976.