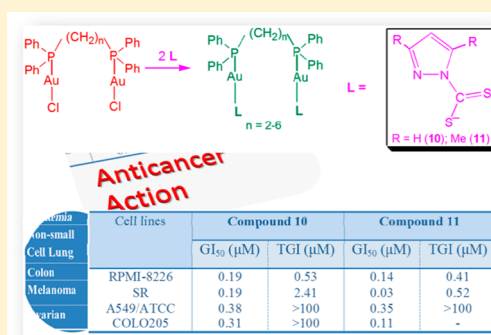


Phosphinogold(I) Dithiocarbamate Complexes: Effect of the Nature of Phosphine Ligand on Anticancer Properties

Frankline K. Keter,^{†,‡} Ilia A. Guzei,^{†,‡} Margo Nell,[§] Werner E. van Zyl,^{†,¶} and James Darkwa^{*,†}[†]Department of Chemistry, University of Johannesburg, P.O. Box 524, Auckland Park 2006, South Africa[‡]Department of Chemistry, University of Wisconsin-Madison, 1101 University Avenue, Madison, Wisconsin 53706, United States[§]Department of Pharmacology, University of Pretoria, Pretoria 0002, South Africa[¶]Advanced Materials Division, Mintek, Private Bag X3015, Randburg 2125, South Africa

S Supporting Information

ABSTRACT: The reactions of potassium salts of the dithiocarbamates **L** {where **L** = pyrazolyldithiocarbamate (**L1**), 3,5-dimethylpyrazolyldithiocarbamate (**L2**), or indazolyldithiocarbamate (**L3**)} with the gold precursors [AuCl(PPh₃)], [Au₂Cl₂(dppe)], [Au₂Cl₂(dppp)], or [Au₂Cl₂(dpph)] lead to the new gold(I) complexes [AuL(PPh₃)] (**1–3**), [Au₂L₂(dppe)] (**4–6**), [(Au₂L₂)(dppp)] (**7–9**), and [Au₂(L)₂(dpph)] (**10–12**) [where dppe = 1,2-bis(diphenylphosphino)ethane, dppp = 1,3-bis(diphenylphosphino)propane, and dpph = 1,6-bis(diphenylphosphino)hexane]. These gold compounds were characterized by a combination of NMR and infrared spectroscopy, microanalysis, and mass spectrometry; and in selected cases by single-crystal X-ray crystallography. Compounds **4–6**, which have dppe ligands, are unstable in solution for prolonged periods, with **4** readily transforming to the Au₁₈ cluster [Au₁₈S₈(dppe)₆]Cl₂ (**4a**) in dichloromethane. Compounds **1–3** and **7–12** are all active against human cervical epithelioid carcinoma (HeLa) cells, but the most active compounds are **10** and **11**, with IC₅₀ values of 0.51 μM and 0.14 μM, respectively. Compounds **10** and **11** are more selective toward HeLa cells than they are toward normal cells, with selectivities of 25.0 and 70.5, respectively. Further tests, utilizing the 60-cell-line Developmental Therapeutics Program at the National Cancer Institute (U.S.A.), showed **10** and **11** to be active against nine other types of cancers.



INTRODUCTION

Gold compounds in the oxidation states of +1 and +3 continue to be of interest in medicinal chemistry because their efficacies toward certain diseases can be fine-tuned.^{1,2} For instance, a number of gold(III) dithiocarbamates have been reported as anticancer agents by Fregona and co-workers.³ This includes the classics [Au(DMDT)X₂] and [Au(ESDT)X₂] (where X = Cl, Br; DMDT = *N,N*-dimethyldithiocarbamate; ESDT = ethylsarcosinedithiocarbamate), which were shown to be 1–4 times more active than *cis*-platin and particularly potent against the lymphoproliferative-type HL-60 cell line.^{3,4} Various modes of action have been proposed for these compounds, from weak DNA binding to proteasome inhibition.^{5–7} A second generation of gold(III) dithiocarbamate (dtc) anticancer compounds includes the gold(III)-dipeptidedithiocarbamate derivatives AuX₂(dtc-Sar-AA-O(*t*-Bu))] (X = Cl, Br; Sar = Sarcosine; AA = Gly, α -aminoisobutyric acid, and L-Phe)⁸ and several [AuX₂(pdtc)] (X = Cl, Br; pdtc = oligopeptidedithiocarbamate) compounds, which are active against four human tumor cell lines at lower IC₅₀ values than is *cis*-platin.⁹

Gold(I) thiolate compounds have also been of interest as anticancer agents since the first reports on the antiarthritic agent auranofin,^{1,10,11} and other gold(I) thiolates^{12–14} were

shown to possess anticancer properties. These include the report by Tiekinck and co-workers that triorganophosphinogold(I) dithiocarbamates of general formula [(R₃P)Au(S₂CNR₂)] were active against seven human cancer cell lines.¹³ Of these, [(Et₃P)Au(S₂CNEt₂)] (Et = ethyl) was the most active against the IGROV ovarian cell line (ID₅₀ = 12 ng/mL).¹³

An early report by Mirabelli et al.,¹⁴ supported by others,^{12,13} on various gold(I) compounds of structural formula R₃P–Au–X (R = ethyl; X = thioglucose (SR)) has demonstrated that the presence of a P–Au–S motif in these compounds enhances their anticancer activity. Notably, phosphinogold(I) thiolate complexes of type [Au(PR₃)(SR)] are more active than gold(I) thiolates of type [Au(SR)], suggesting that the presence of phosphine ligands increases the lipophilicity and membrane permeability of the phosphinogold(I) complexes that make them active.¹²

It has been established that gold compounds, such as auranofin, act against cancer cells via the mitochondria by inhibiting thioredoxin reductase.^{11–18} In doing so, gold(I) binds the C-terminal of the redox-active selenocysteine, leading

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to cytotoxic effects.^{18–20} For these reasons, several phosphinogold(I) thiolate complexes have been prepared and investigated for their anticancer activities. Examples include the hydrophilic tetrakis((tris(hydroxymethyl)phosphino)gold(I) complex, which was reported to be cytotoxic toward HCT-15 tumor cells.²¹ Other examples are [Au(PPh₃)(sppa)] (sppa = sulfanylpropenoate),²² [Au(PEt₃)(K₃TSC)] (K₃TSC = vitamin K₃ derivative),²³ and the gold(I) 7-azacoumarin complex,²⁴ which exhibit activity against HeLa and A2780cis cells, [(AuOmS)₂(Ph₂P(CH₂)₂PPh₂)] (OmS = lupinylsulfide), exhibiting activity against ovarian carcinomas,²⁵ and [Et₃PAu-(S₂CNEt₂)] and [Ph₃PAu(S₂CNC₂C₄H₈)], which exhibit activity against breast, ovarian, and colon cancer.^{12,13a} This strongly suggests that one strategy to make gold(I) anticancer drugs is to prepare molecules that have a phosphine and a sulfur-containing ligand bound to gold(I) to have the P–Au–S motif.

For phosphinogold(I) complexes, the nature of the phosphine ligand appears to be important in regulating their anticancer behavior. For example a report on the anticancer properties of dinuclear di(phosphino)alkane gold(I) chloride by Mirabeli et al.²⁶ suggested two key factors determine the activity of these gold(I) compounds. The first is to have phenyl groups on the phosphorus, and the second is the effect of CH₂ linkers, with the compound with two CH₂ linkers being more active than compounds with three, four, five, or six CH₂ linkers. However, a recent report by Raubenheimer and co-workers indicates that dinuclear di(phosphino)alkane gold(I) azole complexes with longer CH₂ linkers are highly active for a number of cancer cell lines, and at very low dosage.²⁷ Clearly, from these two reports, the nature of the di(phosphino)-alkanegold(I) and the ancillary ligand bound to the gold(I) is important in determining the anticancer activity of these compounds.

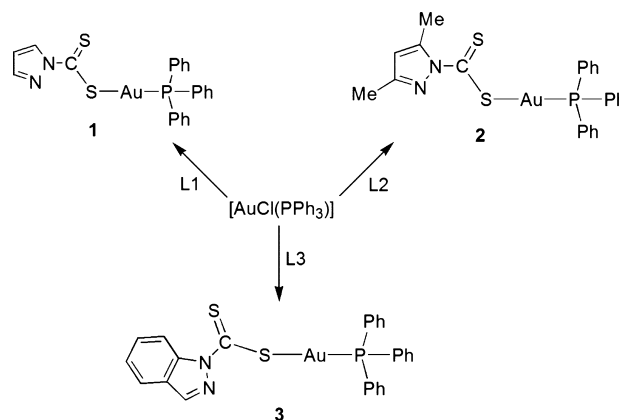
We report here new phosphinogold(I) dithiocarbamate complexes, using triphenylphosphine, 1,2-bis-(diphenylphosphino)ethane (dppe), 1,3-bis-(diphenylphosphino)-propane (dppp), 1,6-bis-(diphenylphosphino)hexane (dpph), and the pyrazol-1-yl- and indazol-1-yl-based dithiocarbamate ligands (dtcs) to test the hypothesis that combining alkane chain length in the diphosphino ligand and using dithiocarbamate as the thiolate can lead to new gold anticancer compounds. Our choice of pyrazoles and indazole as the backbone of the dithiocarbamate is based on their medicinal properties.^{28,29} This approach to prepare phosphinogold(I) dithiocarbamate compounds has resulted in compounds that have excellent anticancer activities against several cancer cell lines. Several reports on the chemistry of gold(I) dithiocarbamate compounds,^{30–39} gold(I) thiosemicarbazones, and gold(I) thiosemicarbazone complexes^{40,41} have appeared in the literature. Some of these compounds have anticancer activity that has been attributed to the synergistic effect of the metals and the thiolate ligands, but none of these compounds has phosphine ligands.

We have further shown for the first time that dithiocarbamate can act as a sulfur transfer reagent to form the high-nuclearity gold cluster [Au₁₈S₈(dppe)₆]²⁺ from [Au(L1)(dppe)], a previously known cluster that was erroneously reported as neutral [Au₁₈S₈(dppe)₆].H₂O.⁴² The heavier congener of this cluster, [Au₁₈Se₈(dppe)₆]²⁺, is known.⁴³

RESULTS AND DISCUSSION

Triphenylphosphinogold(I) Dithiocarbamate Complexes (1–3). The gold(I) complexes 1–3 were prepared from the reaction of [AuCl(PPh₃)] and the dtc ligands L1–L3 (Scheme 1) and were isolated as orange-yellow solids. The

Scheme 1. Preparations of (Monophosphino)gold(I) Dithiocarbamate Complexes 1–3



complexes were characterized by a combination of spectroscopy, mass spectrometry, and microanalysis; in the case of 3, single-crystal X-ray crystallography was also used. Complex 1 displays a typical ¹H NMR spectrum for these compounds, which shows dtc protons for 4-pz and 5-pz proton peaks at 6.27 and 7.63 ppm, respectively. The ³¹P{¹H} NMR spectra of complexes 1–3 displayed broad signals between 29.9 and 36.3 ppm.

Crystal data, together with the data collection and refinement parameters, are presented in Table 1. Compound 3 exhibits the expected near-linear geometry about the Au(I) center with the P1–Au1–S1 angle spanning 175.36(2)°. A molecular drawing of 3 is shown in Figure 1. The Au–P distance of 2.2533(6) Å is in excellent agreement with the average value of 2.26(2) Å obtained by averaging 628 Au–PPh₃ distances from 501 complexes reported to the Cambridge Structural Database (CSD).⁴⁴ The Au–S distance in 3 (2.3272(6) Å) is slightly longer than the Au–P distance, but is somewhat shorter than the average distance of 2.38(16) Å computed based on 89 Au–S distances for 58 dithiocarbamate complexes reported to the CSD. However, the difference with the latter is not statistically significant as Au–S bond lengths fall in a broad range. In closely related compounds these distances are as follows: [Au(S₂CNEt₂)(PPh₃)], Au–S (2.338(4) Å) and Au–P (2.251(3) Å)³³; [Au(S₂CNC₁₀H₂₀O₄)(PPh₃)] (where C₁₀H₂₀O₄ = aza-15-crown-5), Au–S (2.339(3) Å) and Au–P (2.253(3) Å).³¹ The C–S distances are different, with the C1–S1 distance to the coordinated S atom being longer (single bond, 1.732(3) Å) than the C1–S2 bond that clearly shows multiple character (1.663(3) Å).

(Diphosphino)ethanegold(I) Dithiocarbamate Complexes (4–6). Complexes 4–6 (Scheme 2) were prepared from the reaction of [Au₂Cl₂(dppe)] with 2 equiv of L1–L3 and were isolated as flaky yellow solids in moderate yields. In the ¹H NMR spectra of 4–6, the CH₂ protons of the di(phosphino)alkane ligand appeared upfield (1.8–2.9 ppm) as second-order multiplets, and the ¹³C{¹H} NMR spectra showed peaks at ca. 215.0 and ca. 129.2–133.5 ppm, which were assigned to C(=S) and the phenyl carbons,

Table 1. Crystallographic Data for Compounds 3, 4a, and 12

	3	4a	12
empirical formula fw	C ₂₆ H ₂₀ AuN ₂ PS ₂ 652.50	C ₁₅₆ H ₁₄₄ Au ₁₈ Cl ₂ P ₁₂ S ₈ 6263.13	C ₄₆ H ₄₂ Au ₂ N ₄ P ₂ S ₄ 1234.95
temp (K)	100(2)	105(2)	150(2)
wavelength (Å)	0.710 73	0.710 73	0.774 90 ^a
cryst syst	monoclinic	monoclinic	monoclinic
space group	<i>P</i> 2 ₁ / <i>n</i>	<i>P</i> 2 ₁ / <i>n</i>	<i>P</i> 2 ₁ / <i>n</i>
<i>a</i> (Å)	9.1382(12)	17.6854(7)	13.499(11)
<i>b</i> (Å)	22.487(3)	34.4361(14)	11.560(19)
<i>c</i> (Å)	11.3760(14)	29.1894(12)	15.396(15)
β (deg)	91.541(2)	92.4570(10)	108.60(6)
volume (Å ³)	2336.8(5)	17760.5(12)	2277(5)
<i>Z</i>	4	4	2
density (calc) (Mg/m ³)	1.855	2.342	1.801
abs. coeff. (mm ⁻¹)	6.560	15.077	8.334
<i>F</i> (000)	1264	11376	1196
final <i>R</i> indices (<i>R</i> 1)	0.0587	0.0473	0.0404
reflections collected	36 929	188 093	31 033
completeness to θ	99.3%	99.8%	99.2%
goodness-of-fit on <i>F</i> ²	1.19	1.02	1.02
final <i>R</i> indices [<i>I</i> > 2 σ (<i>I</i>)]	<i>R</i> 1 = 0.0210, <i>wR</i> 2 = 0.0542	<i>R</i> 1 = 0.0473, <i>wR</i> 2 = 0.1258	<i>R</i> 1 = 0.0404, <i>wR</i> 2 = 0.0984
<i>R</i> indices (all data)	<i>R</i> 1 = 0.0216, <i>wR</i> 2 = 0.0544	<i>R</i> 1 = 0.0677, <i>wR</i> 2 = 0.1347	<i>R</i> 1 = 0.0566, <i>wR</i> 2 = 0.1071
larg. diff. peak hole (e.Å ⁻³)	1.15 and -0.88	2.14 and -1.44	1.45 and -0.86

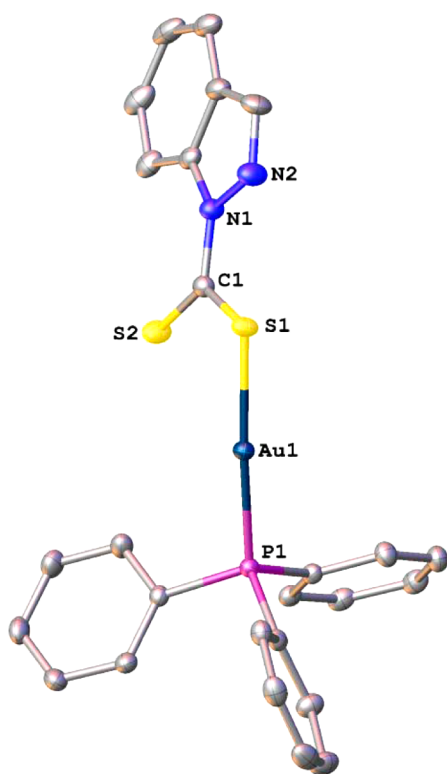
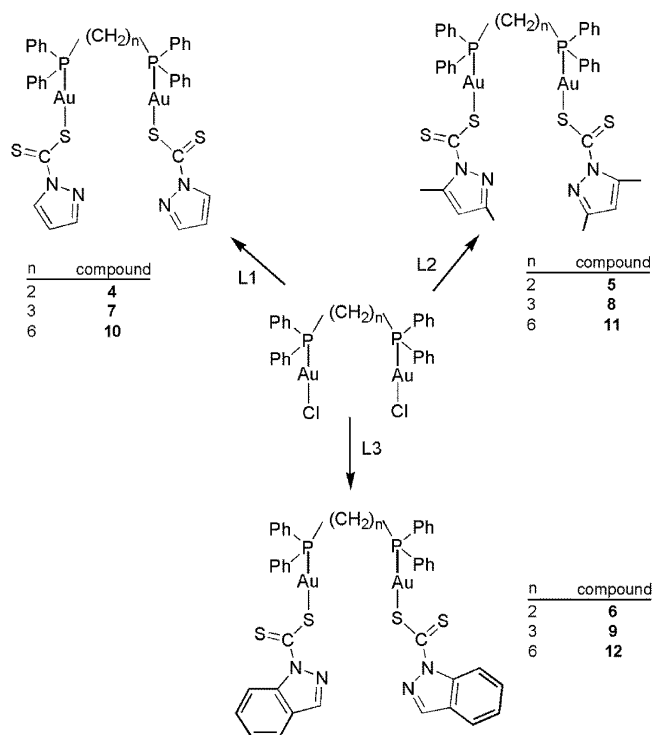
^aUsing synchrotron radiation tuned to $\lambda = 0.7749$ Å.

Figure 1. Molecular structure of 3 drawn with 30% probability ellipsoids. H atoms are omitted for clarity. Selected bond lengths [Å] and angles [deg]: Au1–P1, 2.2533(6); Au1–S1, 2.3272(6); P1–C1, 1.811(2); N1–C1, 1.399(3); S2–C1, 1.663(3); P1–Au1–S1, 175.36(2); C1–S1–Au1, 100.04(9); C9–P1–Au1, 118.20(8); S2–C1–S1, 124.85(15).

respectively; the ³¹P{¹H} NMR spectra showed broad singlets between 29.0 and 33.0 ppm. The spectroscopic data for 4–6 are similar to the data for the bis(diphenylphosphino)digold(I) cations with azotate neutral heterocycles [μ -(dppe)Au₂-

Scheme 2. Preparations of (Diphosphino)alkylgold(I) Dithiocarbamate Complexes 4–12



(pzH)₂ClO₄ and [μ -dppmAu₂(pzH)₂]₂ClO₄ (pzH = pyrazole).⁴⁵

Attempts to grow crystals of compound 4 were unsuccessful. On prolonged standing, a solution of 4 in a mixture of dichloromethane and diethylether at room temperature led to the isolation of the chloride salt of the cationic gold(I) cluster [Au₁₈S₈(dppe)₆]²⁺ (4a), whose molecular structure was established by single crystal X-ray crystallography (Table 1,

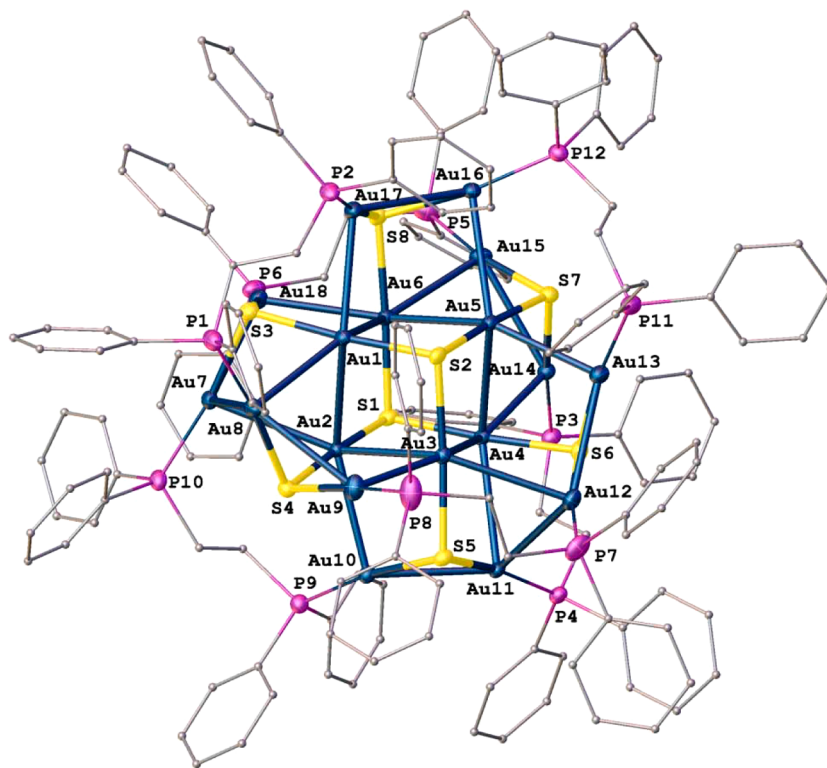


Figure 2. Molecular structure of **4a**. H atoms are omitted for clarity. Selected bond lengths [Å] and angles [deg]: Au1–P1, 2.2533(6); Au1–S2, 2.343(3); Au1–S3, 2.9263(7); Au1–Au2, 2.9452(7); Au1–Au7, 3.1395(7); Au7–P10, 2.268(4); S2–Au1–S3, 175.80(12); Au10–Au2–Au3, 77.409(18); S2–Au1–Au2, 92.05(8); S1–Au2–Au1, 109.71(8); P8–Au9–S4, 109.71(8).

Figure 2). The molecular structure of **4a** contains 18 gold atoms, 8 sulfur atoms, and 6 dppe units, with 2 chloride counterions and some unidentified solvents of crystallization (see Supporting Information for more details). It is isomorphous with the Se analogue $[\text{Au}_{18}\text{Se}_8(\text{dppe})_6]^{2+}$.⁴³ The geometry of **4a** is similar to that of the selenium analogue $[\text{Au}_{18}\text{Se}_8(\text{dppe})_6]^{2+}$.⁴³ At the core of the cluster is a Au_6S_2 cubane composed of atoms Au1–Au6, S1, and S2 in such a fashion that each face contains three Au and one S atoms; the S atoms are positioned across the body diagonal of the cubane. The three Au atoms of each face are capped with a Au_2S triangle. Each S atom is μ_3 -bridging: atoms S1 and S2 coordinate to three Au atoms of the cubane, whereas the other S atoms ligate one Au atom from the cubane and two Au atoms from a capping Au_2S triangle. Each of the Au atoms in the capping triangles is ligated by one P atom of the six bidentate dppe ligands. The Au–Au distances vary between 2.9147(7) and 3.2838(8) Å; the Au–S bond length falls within a 2.328(4)–2.374(3) Å range, and the 12 Au–P distances average 2.260(7) Å, a value similar to that in compound **3**. Other high nuclearity gold(I) sulfide complexes with bridging diphosphines have been reported.^{46–48} As for the neutral hydrated cluster $[\text{Au}_{18}\text{S}_8(\text{dppe})_6]\cdot\text{H}_2\text{O}$,³² its formulation implies a mixed-valent $\text{Au}^0\text{–Au}^{\text{I}}$ compound, contrary to our findings and unlike the related cluster $[\text{Au}_{18}\text{Se}_8(\text{dppe})_6]^{2+}$.^{43,48} The formation of **4a** is unique because dithiocarbamate as a sulfur source is clearly unusual in forming high-nuclearity clusters. The formation of **4a** from **4** is likely to occur via a C–S bond cleavage, triggered by the heterocycle attached to the CS_2 in the dithiocarbamate. This is supported by $[\text{Au}(\text{dithiocarbamate})(\text{dppe})]$ complexes that are stable in solution and have solid-state structures where the dithiocarbamate

ligands remain intact, for example, $[\text{Au}_2(\text{S}_2\text{CNET}_2)_2(\text{dppe})]$,⁴⁹ $[\text{Au}_2(\text{S}_2\text{CNET}_2)_2\{\mu-(\text{PPh}_2)_2\text{C}=\text{CH}_2\}]$,⁴⁹ and $[\text{Au}(\text{S}_2\text{CNC}_{10}\text{H}_{20}\text{O}_4)(\text{dppe})]$ (where $\text{C}_{10}\text{H}_{20}\text{O}_4$ = aza-15-crown-5).³¹ The dppe backbone and aurophilicity of the two Au atoms in **4** must have aided the cluster formation since increase in the alkyl chain length in the diphosphine gave stable diphosphinogold(I) dithiocarbamate complexes (vide infra).

(Diphosphino) Propyl and Hexyl Gold(II) Dithiocarbamate Complexes (7–12). To avoid the cluster formation observed during crystallization of **4**, the $-\text{CH}_2-$ linker count within the diphosphines was increased to three in dppp and six in dpph. Their respective diphosphinogold(I) dithiocarbamates (**7–12**) were synthesized from the reaction between $[\text{Au}_2\text{Cl}_2(\text{dppp})]$ or $[\text{Au}_2\text{Cl}_2(\text{dpph})]$ and 2 equiv of **L1**, **L2**, or **L3** (Scheme 2). The ^1H and $^{31}\text{P}\{^1\text{H}\}$ NMR spectra of **7–12** showed similar patterns as those of **4–6**, with the CH_2 protons in the backbone of the diphosphine ligands appearing upfield in the region of 1.49–2.93 ppm and singlets for phosphorus between 29.0 and 33.0 ppm. These complexes were stable in solution for several weeks, with no signs of decomposition. ESI-MS of **9** (m/z = 1190.88, Figure S1, Supporting Information) showed molecular ions that correspond to a binuclear complex with a molecular ion peak at m/z = 1190.88, and the crystal structure of **12** (Figure 3) further indicates that the length of the alkyl linker in the diphosphine ligand is important in preventing compounds **7–12** from transforming into clusters.

Crystal data, together with the data collection and refinement parameters, are presented in Table 1. The molecular drawing of **12** is shown in Figure 3. The dinuclear complex resides on a crystallographic inversion center, and only one-half of it is symmetry independent; the most important distances are discussed. The symmetry-independent half of complex **12** can

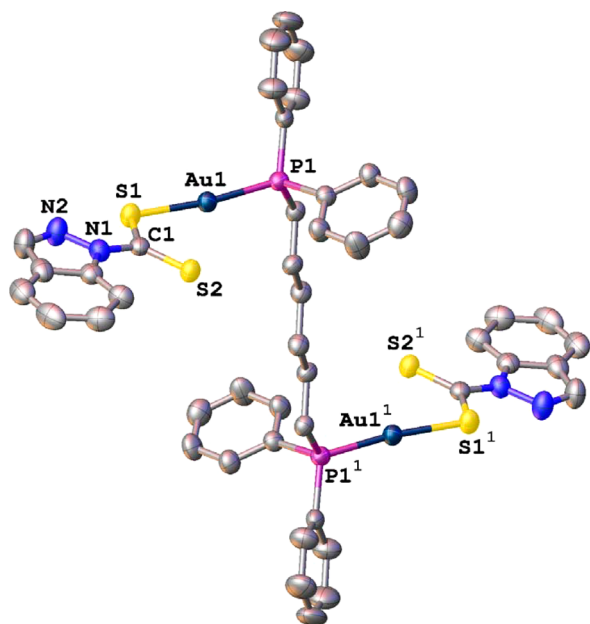


Figure 3. Molecular structure of **12** drawn with 50% probability ellipsoids. H atoms are omitted for clarity. Selected bond length [Å] and angles [deg]: Au1–P1, 2.255(2); Au1–S1, 2.312(3); P1–C9, 1.818(6); N1–C8, 1.390(8); S1–C1, 1.720(7); S2–C1, 1.656(6); P1–Au1–S1, 173.41(6); C1–S1–Au1, 100.9(2); C15–P1–Au1, 112.6(8); S2–C1–S1, 124.1(4). Selected symmetry-related atoms are labeled with a superscript.

be considered a congener of complex **3**. In complex **12** the coordination environment about the Au atoms is nearly linear, with the P1–Au1–S1 angle spanning 173.41(6)°; the Au1–S1 (2.312(3) Å) and Au1–P1 (2.255(2) Å) distances are typical and in good agreement with the ones observed in complex **3**. The single C1–S1 distance to the ligating atom S1 (1.720(7) Å) is longer than the formally double C1=S2 bond of 1.656(6) Å, but both values closely match the corresponding values in complex **3**. In the lattice, pairs of clusters are packed with an inversion center at the center of the Au–S⋯Au–S parallelepiped. Within each parallelepiped, the nonbonding Au⋯Au distance measures 3.986(3) Å, and the Au⋯S distance is 3.598(4) Å. Both distances exceed the sum of the van der Waals radii of the involved elements; thus, there are no aurophilic interactions in the lattice. The structural findings for complex **12** are similar to those reported by Uson et al.⁵⁰ and Cookson et al.⁵¹ for the gold(I) complexes [Au(PPh₃)(S₂C-aza-15-crown-5)] and [Au(C₅H₅NS)₂], respectively. The other

bonding distances and angles are in the same range and correspond with the expected values. Crystal data, together with the data collection and refinement parameters, are presented in Table 1.

Biological Activity. Fifteen compounds (**L1**–**L3**, **1**–**12**) were initially screened for their ability to inhibit cell growth on human cervix epitheloid carcinoma (HeLa) cells in vitro. All data were acquired in triplicate, and the final values were recorded as averages. The dose values that caused 50% inhibition of cell growth (IC₅₀) are listed in Table 2. To establish the activities of phosphine gold(I) complexes it was important to first establish the activities of the ligands (**L1**–**L3**) to determine whether activities of the metal complexes could be due to the presence of the dithiocarbamate ligands. All the free ligands were inactive against HeLa cells.

The gold(I) complexes were grouped into three sets according to ligand type, PPh₃ (**1**–**3**), dppp (**7**–**9**), and dppe (**10**–**12**), for testing. Complexes **4**–**6** (the dppe set) were not tested because of the instability of **4** in solution discussed earlier. Although complexes **1**–**3**, **7**–**9**, and **12** had quite good IC₅₀ values (2.2–7.0 μM), their tumor specificity (TS) values were very low (Table 2), and they were less active than *cis*-platin (0.476 μM). The low TS factors suggest that these compounds were toxic to both tumor and normal cells and were thus not further investigated. Compounds **10** and **11** had IC₅₀ values of 0.51 μM and 0.14 μM, respectively, compared to 0.45 μM for *cis*-platin. These two gold compounds (**10** and **11**) displayed very good TS values of 25.0 and 70.5, respectively (Table 2). Our findings are comparable to and in some cases much better than activities reported for phosphine gold(I) thiolate complexes,^{12,13,23} which further buttresses the conception that a P–Au–S motif enhances the therapeutic effect of phosphine gold(I) thiolate compounds as anticancer agents. Complexes **10** and **11** were further investigated against a panel of 60 cancer cell lines of the Developmental Therapeutics Program (DTP) at the National Cancer Institute (U.S.A.), where the concentration that inhibits growth of cells by 50% (GI₅₀), the concentration that causes total tumor growth inhibition (TGI), and the least concentration required to kill 50% of tumor cells (LC₅₀) were determined. The 60 cell lines were organized into subpanels, representing various histologies, for example, nonsmall cell lung-, colon-, breast-, ovarian-, renal-, prostate-, and CNS cancers, leukemia, and melanoma. Because of the extensive data, we have highlighted only the most important findings in Tables 3 and 4, but more data are provided as Supporting Information (Figures S2 and S3; Tables S1 and S2). Complexes **10** and **11** were, in general,

Table 2. Growth Inhibition Data of Complexes **1**–**3**, **7**–**9**, and **10**–**12** against HeLa Cells and Corresponding Tumor Specificities

	drug	HeLa IC ₅₀ (μM)	lymph (resting) IC ₅₀ (μM)	lymph (stimulated) IC ₅₀ (μM)	tumor specificity (TS)
PPh ₃ derivatives	1	2.56 ± 0.12	9.05 ± 3.50	8.55 ± 3.37	3.4
	2	2.63 ± 0.10	9.53 ± 3.18	8.63 ± 3.77	3.5
	3	2.56 ± 0.17	13.98 ± 10.50	5.08 ± 2.48	3.7
dppp derivatives	7	3.07 ± 0.48	7.05 ± 2.02	5.02 ± 1.72	1.9
	8	6.96 ± 0.51	6.13 ± 2.92	2.94 ± 0.91	0.7
	9	3.38 ± 0.30	18.48 ± 3.88	8.16 ± 4.10	3.9
dppe derivatives	10	0.51 ± 0.08	18.89 ± 5.66	6.68 ± 3.40	25.0
	11	0.14 ± 0.01	18.23 ± 15.90	1.38 ± 0.39	70.5
	12	4.04 ± 0.49	37.74 ± 12.60	30.25 ± 11.06	8.4
	<i>cis</i> -platin	0.45 ± 0.09			

Table 3. Selected DTP Antitumor Inhibition Results of Complex 10

type of cancer	panel/cell line	GI ₅₀ (μM)	TGI (μM)	LC ₅₀ (μM)
leukemia	RPMI-8226	0.19	0.53	>100
	SR	0.19	2.41	>100
nonsmall cell lung	A549/ATCC	0.38	>100	>100
	COLO205	0.31	>100	>100
	KM12	0.43	1.73	>100
melanoma	UACC-62	0.38	>100	>100
ovarian	OVCAR-3	0.70	14.20	>100
renal	RXF393	0.45		>100

Table 4. Selected DTP Antitumor Inhibition Results of Complex 11

type of cancer	panel/cell line	GI ₅₀ (μM)	TGI (μM)	LC ₅₀ (μM)
leukemia	RPMI-8226	0.14	0.41	>100
	SR	0.03	0.52	90
nonsmall cell lung	A549/ATCC	0.35	>100	>100
	COLO205	0.11		>100
	HCC-2998	0.25	0.80	2.99
melanoma	UACC-62	0.25		>100
ovarian	OVCAR-8	0.60	>100	>100
renal	A498	0.38		>100

active against all of the 60 cell lines in the subpanels mentioned above, with GI₅₀ values in the range of 0.03–100 μM.

Compound **10** showed good activity against all cell lines tested but was particularly highly active against the leukemia RPMI-8226 and SR cell lines, with a GI₅₀ value of 0.19 μM for each cell line and TGI values of 0.53 and 2.41 μM for these cell lines, respectively (Table 3). This indicates the inhibitory property of **10** at very low molar concentrations. However, what was interesting is that the LC₅₀ for **10** was greater than 100 μM for most cell lines, suggesting minimal cytotoxic properties of **10**. It, therefore, implies that this complex is cytostatic. Similar results have been reported in literature, where a compound is cytostatic/antiproliferative and not necessarily cytotoxic.⁵²

We observed a similar activity pattern for complex **11** against all the cell lines, but **11** was even more potent than **10**. For instance, the GI₅₀ values of **11** against the leukemia cell lines RPMI-8226 and SR were 0.14 μM and 0.03 μM, respectively. In fact, the best activity registered for all the compounds tested was that of **11** against the SR cell line (0.03 μM) (Table 4). The TGI concentrations recorded were similarly quite low, RPMI-8226 (0.41 μM) and SR (0.52 μM), which corroborates the cytostatic properties of **11**. Of interest is that compound **11**, unlike **10**, showed very good cytotoxic activity against the HCC-2998 cell line (LC₅₀ = 2.99 μM). Compound **11** also showed activity against the colon cancer COLO205 and HCC-2998 cell lines, with GI₅₀ of 0.11 μM and 0.25 μM, respectively. In general, complexes **10** and **11** recorded high cytostatic median values (GI₅₀ = −6.25; TGI = −4.51) and cytotoxic median value (LC₅₀ = −4.05) against the 60 cell lines (Table 5,

Figures S5 and S7, Supporting Information), which in the DTP program indicates high activity across most of the panel of cell lines.

Following on the in vitro results, compound **11** was further tested in vivo in hollow fiber studies, but the activity data suggested that the compound could not be delivered to the targets. We are therefore investigating the use of various drug delivery vehicles, including the use of β- and γ-cyclodextrin, to deliver this drug in vivo.

CONCLUSIONS

We have prepared the first examples of phosphinogold(I) dithiocarbamates derived from heterocycles. The stability of these phosphinogold(I) dithiocarbamate complexes depends on the nature of the phosphine ligand used. Triphenylphosphino and diphenylphosphinoalkyl ligands with alkyl chains longer than ethyl produce stable gold dithiocarbamates in solution, but the diphenylphosphinoethanogold(I) dithiocarbamates are unstable and were found to transform to a Au₁₈ cluster. All the phosphinogold(I) dithiocarbamates that are stable in solution are active against HeLa cancer cells, suggesting the importance of the P–Au–S moiety in conferring activity to the compounds. Compounds with hexyl chain were found to be the most active and extremely selective; in particular compounds **10** and **11** were 25.0 and 70.5 times more selective for HeLa cells than normal cells. Bis-(diphosphines) alkanes with longer CH₂ linkers appear to hold better promise as anticancer agents than their shorter CH₂-linker counterparts, similar to the observation by Mirabeli et al.²⁷ We also found compounds **10** and **11** to have excellent activities for nine other cancer cell lines in vitro. Of the nine cancer cell lines tested, the best activity against RPMI-8226 was found for **10** (GI₅₀ = 0.19 μM), while the best activity against SR cells (GI₅₀ = 0.03 μM) was for **11**. Although activities for **10** and **11** in vivo were not so great, we believe finding drug delivery vehicles to transport these two compounds would improve their activities in vivo. We are therefore investigating the use of various delivery vehicles for these compounds.

EXPERIMENTAL SECTION

Materials and Instrumentation. All manipulations were performed under a dry, deoxygenated nitrogen atmosphere using Schlenk techniques. All commercially available chemicals were used as received. Pyrazol-1-ylthiocarbamate (**L1**), 3,5-dimethylpyrazol-1-ylthiocarbamate (**L2**), and indazol-1-ylthiocarbamate (**L3**) were synthesized according to literature methods.^{53,54} Gold starting materials [Au₂Cl₂(dppe)], [Au₂Cl₂(dppp)], and [Au₂Cl₂(dpph)] {dppe = 1,2-bis(diphenylphosphino)ethane; dppp = 1,3-bis(diphenylphosphino)propane; and dpph = 1,6-bis(diphenylphosphino)hexane} were synthesized according to the literature procedures.⁵⁵ Infrared (IR) spectra were recorded as KBr pellets on a Bruker Tensor27 spectrophotometer. ¹H, ¹³C{¹H}, and ³¹P{¹H} NMR spectra were recorded on a Varian 2000 spectrometer (¹H, 300 MHz; ¹³C, 75.4 MHz; and ³¹P, 121.5 MHz) in CDCl₃ or D₂O at room temperature. Elemental analysis was performed on a Fisons elemental analyzer at the University of Cape Town, South Africa. ESI-MS spectra

Table 5. Cytostatic and Cytotoxic Median Data of 10 and 11

drug	GI ₅₀			TGI			LC ₅₀		
	MG-MID	Δ	range	MG-MID	Δ	range	MG-MID	Δ	range
10	−5.93	0.8	2.73	−4.36	1.92	2.28	−4.02	0.69	0.71
11	−6.25	1.22	1.96	−4.51	1.87	2.38	−4.05	1.47	1.52

were recorded on a Waters API Quattro Micro spectrometer at the University of Stellenbosch, South Africa. The mass spectra were collected using 3.0 s cyclical scans and applying the sample cone voltage of 15 V at the source block temperature of 100 °C. Desolvation temperature was 350 °C at desolvation cone gas flow rate of 350 L/h.

Synthesis of Triphenylphosphinegold(I) Complexes. *Pyrazolyl-1-dithiocarbamate-triphenylphosphinogold(I) (1).* Complex **1** was prepared by dissolving **L1** (0.05 g, 0.3 mmol) in deionized water (10 mL), after which a solution of $[\text{AuCl}(\text{PPh}_3)]$ (0.15 g, 0.3 mmol) in dichloromethane (10 mL) was added. The resultant biphasic mixture was vigorously stirred at room temperature for 30 min, during which time the color of the organic layer changed from colorless to red. The aqueous and organic layers were separated, and the organic layer was dried over anhydrous MgSO_4 . The solvent was removed from the organic extract in vacuo to afford an orange-yellow solid. Yield = 0.17 g (93%). ^1H NMR (CDCl_3): δ 7.63 (s, 1H, 5-pz); 7.57 (s, 1H, 3-pz); 7.49 (m, 9H, PPh_3); 7.35 (m, 6H, PPh_3); 6.27 (s, 1H, 4-pz). $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3): δ 213.4 (C(C=S)); 144.0 (C(5-pz)); 134.2–129.1 (phenyl region); 140.8 (C(3-pz)); 106.1 (C(4-pz)). IR (KBr, cm^{-1}): $\nu_{\text{C=N}} = 1620$, $\nu_{\text{C-S}} = 1121$, $\nu_{\text{C=S}} = 884$. $^{31}\text{P}\{^1\text{H}\}$ NMR (CDCl_3): δ 35.2 (PPh_3). Anal. Calc. for $\text{C}_{24}\text{H}_{22}\text{AuN}_2\text{PS}_2$: C 43.86, H 3.01, N 4.65. Found: C 43.53, H 3.05, N 4.62%.

Compounds **2** and **3** were prepared using the procedure described for **1** above, using the reagents indicated for each compound.

3,5-dimethylpyrazolyl-1-dithiocarbamate-triphenylphosphinogold(I) (2). **L2** (0.1 g, 0.47 mmol), $[\text{AuCl}(\text{PPh}_3)]$ (0.24 g, 0.47 mmol). Yield = 0.17 g (57%). ^1H NMR (CDCl_3): δ 7.44, 7.29 (m, 15H, PPh_3); 6.12, 5.82 (s, 1H, 4-pz); 2.29 (m, 6H, 3, 5-pz). $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3): δ 215.5 (C(C=S)); 149.7 (C(5-pz)); 134.1–128.7 (phenyl region); 142.1 (C(3-pz)); 112.4, 104.1 (C(5-pz)); 13.9 (C(CH_3 , 5-pz)); 12.31 (C(CH_3 , 3-pz)). IR (KBr, cm^{-1}): $\nu_{\text{C=N}} = 1580$, $\nu_{\text{C-S}} = 1154$, $\nu_{\text{C=S}} = 854$. $^{31}\text{P}\{^1\text{H}\}$ NMR (CDCl_3): δ 33.8 (PPh_3). Anal. Calc. for $\text{C}_{24}\text{H}_{22}\text{AuN}_2\text{PS}_2$: C 45.72, H 3.52, N 4.44. Found: C 45.89, H 3.89, N 3.29%.

Indazolyl-1-dithiocarbamate-triphenylphosphinogold(I) (3). **L3** (0.1 g, 0.4 mmol), $[\text{AuCl}(\text{PPh}_3)]$ (0.2 g, 0.4 mmol). Yield = 0.15 g (58%). ^1H NMR (CDCl_3): δ 9.21 (d, 1H, $^4J_{\text{HH}} = 9.0$ Hz, Ha); 8.17 (s, 1H, Hb); 7.73 (d, 1H, $^3J_{\text{HH}} = 8.4$ Hz, He); 7.61, 7.50 (m, 15H, PPh_3); 7.58 (1H, Hd); 7.35 (t, 1H, $^4J_{\text{HH}} = 15$ Hz, Hc). $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3): δ 214.2 (C(C=S)); 139.5 (C(5C-pz)); 134.7–129.5 (phenyl region); 129.5 (C(3C-pz)); 126.9 (C(7C-Ph)); 124.6 (C(4C-Ph)); 121.4 (C(8C, 9C-Ph)); 118.7 (C(6C-Ph)). IR (KBr, cm^{-1}): $\nu_{\text{C=N}} = 1601$, $\nu_{\text{C-S}} = 1148$, $\nu_{\text{C=S}} = 905$. $^{31}\text{P}\{^1\text{H}\}$ NMR (CDCl_3): δ 36.8 (PPh_3). Anal. Calc. for $\text{C}_{26}\text{H}_{20}\text{AuN}_2\text{PS}_2$: C 47.86, H 3.09, N 4.29. Found: C 47.41, H 3.04, N 3.85%.

Synthesis of Dinuclear Di(phosphino)alkylgold(I) Complexes. *Bis-(pyrazolyl-1-dithiocarbamate)-bis-(diphenylphosphino)ethane Dinuclear Gold(I) (4) and $[\text{Au}_{18}\text{S}_8(\text{dppe})_6]\text{Cl}_2$ (4a).* To a solution of $[\text{Au}_2\text{Cl}_2(\text{dppe})]$ (0.08 g, 0.09 mmol) in CH_2Cl_2 (10 mL) was added **L1** (0.03 g, 0.19 mmol) previously dissolved in water (10 mL). The resultant mixture was stirred at room temperature for 20 min during which time the CH_2Cl_2 layer changed color from a white suspension to a clear orange solution, while the yellowish water layer became clear. The two layers were separated and the organic fraction dried over anhydrous MgSO_4 . The organic fraction was evaporated in vacuo to obtain **4** as an orange solid. Yield = 0.07 g (73%). ^1H NMR (CDCl_3): δ 8.78 (d, 2H, $^3J_{\text{HH}} = 2.7$ Hz, 5-pz); 7.80 (s, 2H, 3-pz); 7.78 (m, 8H, $(\text{Ph})_2\text{P}(\text{CH}_2)_2\text{P}(\text{Ph})_2$); 7.49 (m, 12H, $(\text{Ph})_2\text{P}(\text{CH}_2)_2\text{P}(\text{Ph})_2$); 6.43 (s, 2H, 4-pz); 2.87 (bs, 4H, CH_2 , $(\text{Ph})_2\text{P}(\text{CH}_2)_2\text{P}(\text{Ph})_2$). $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3): δ 214.2 (C(C=S)); 149.7 (C(5-pz)); 142.0 (C(3-pz)); 134.1–128.7 (phenyl region); 111.4 (C(4-pz)); 24.6 (C($\text{Ph}_2\text{P}(\text{CH}_2)_2\text{PPh}_2$)). IR (KBr, cm^{-1}): $\nu_{\text{C=N}} = 1590$, $\nu_{\text{C-S}} = 1190$, $\nu_{\text{C=S}} = 889$. $^{31}\text{P}\{^1\text{H}\}$ NMR (CDCl_3): δ 36.1 ($\text{Ph}_2\text{P}(\text{CH}_2)_2\text{PPh}_2$). Anal. Calc. for $\text{C}_{34}\text{H}_{30}\text{Au}_2\text{N}_4\text{P}_2\text{S}_4$: C 37.85, H 2.80, N 5.14, S 11.89. Found: C 38.18, H 3.01, N 4.57, S 11.73%.

Attempts to crystallize **4a** in a solution of dichloromethane/ether led to the formation of the gold cluster, $[\text{Au}_{18}\text{S}_8(\text{dppe})_6]\text{Cl}_2$, (**4a**).

Anal. Calc. for $\text{C}_{26}\text{H}_{20}\text{AuCl}_2\text{N}_2\text{PS}_2$: C 29.71, H 2.32, N 4.10. Found: C 29.80, H 2.00, N 3.80%.

Compounds **5**–**12** were prepared using the procedure described for **4** above, using the appropriate starting materials.

Bis-(3,5-dimethylpyrazolyl-1-dithiocarbamate)-bis-(diphenylphosphino)ethane Dinuclear Gold(I) (5). **L2** (0.04 g, 0.19 mmol), $[\text{Au}_2\text{Cl}_2(\text{dppe})]$ (0.1 g, 0.09 mmol). Yield = 0.09 g (88%). ^1H NMR (CDCl_3): δ 7.64 (m, 8H, $\text{Ph}_2\text{P}(\text{CH}_2)_2\text{PPh}_2$); 7.35 (m, 12H, $\text{Ph}_2\text{P}(\text{CH}_2)_2\text{PPh}_2$); 6.09 (s, 2H, 4-pz); 2.69 (bs, 4H, $\text{Ph}_2\text{P}(\text{CH}_2)_2\text{PPh}_2$); 2.26 (d, 12H, CH_3 , 3,5-pz). $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3): δ 216.5 (C(C=S)); 152.7 (C(5-pz)); 143.1 (C(3-pz)); 135.1–128.6 (phenyl region); 112.6 (C(4-pz)); 14.0 (C(5-pz)); 12.4 (C(3-pz)); 25.7 (C($\text{Ph}_2\text{P}(\text{CH}_2)_2\text{PPh}_2$)). IR (KBr, cm^{-1}): $\nu_{\text{C=N}} = 1603$, $\nu_{\text{C-S}} = 1263$, $\nu_{\text{C=S}} = 989$. $^{31}\text{P}\{^1\text{H}\}$ NMR (CDCl_3): δ 35.7 ($\text{Ph}_2\text{P}(\text{CH}_2)_2\text{PPh}_2$). Anal. Calc. for $\text{C}_{38}\text{H}_{38}\text{Au}_2\text{N}_4\text{P}_2\text{S}_4$: C 40.22, H 3.37, N 4.94, S 11.30. Found: C 39.89, H 3.59, N 4.67, S 11.15%.

Bis-(indazolyl-1-dithiocarbamate)-bis-(diphenylphosphino)ethane Dinuclear Gold(I) (6). **L3** (0.04 g, 0.19 mmol), $[\text{Au}_2\text{Cl}_2(\text{dppe})]$ (0.08 g, 0.09 mmol). Yield = 0.07 g (63%). ^1H NMR (CDCl_3): δ 9.16 (s, 1H, Ha); 8.06 (s, 1H, Hb); 7.78 (m, 8H, $(\text{Ph})_2\text{P}(\text{CH}_2)_2\text{P}(\text{Ph})_2$); 7.49 (m, 12H, $(\text{Ph})_2\text{P}(\text{CH}_2)_2\text{P}(\text{Ph})_2$); 7.67 (d, 1H, $^3J_{\text{HH}} = 8.4$ Hz, He); 7.71 (1H, Hd); 7.43 (t, 1H, $^4J_{\text{HH}} = 15$ Hz, Hc); 2.87 (bs, 4H, CH_2 , $(\text{Ph})_2\text{P}(\text{CH}_2)_2\text{P}(\text{Ph})_2$). $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3): δ 213.9 (C(C=S)); 142.1 (C(5-Pz)); 139.0 (C(3-Pz)); 134.1–129.7 (phenyl region); 126.1 (C(7C-Ph)); 123.1 (C(4C-Ph)); 122.2 (C(8C-Ph, 9C-Ph)); 117.5 (C(6C-Ph)); 25.2 (C($\text{Ph}_2\text{P}(\text{CH}_2)_2\text{PPh}_2$)). IR (KBr, cm^{-1}): $\nu_{\text{C=N}} = 1612$, $\nu_{\text{C-S}} = 1137$, $\nu_{\text{C=S}} = 915$. $^{31}\text{P}\{^1\text{H}\}$ NMR (CDCl_3): δ 34.9 ($\text{Ph}_2\text{P}(\text{CH}_2)_2\text{PPh}_2$). Anal. Calc. for $\text{C}_{42}\text{H}_{34}\text{Au}_2\text{N}_4\text{P}_2\text{S}_4$: C 42.79, H 2.91, N 4.75, S 10.88. Found: C 43.24, H 2.82, N 4.65, S 10.57%.

Bis(pyrazolyl-1-yl)dithiocarbamate-bis-(diphenylphosphino)propane Dinuclear Gold(I) (7). **L1** (0.08 g, 0.46 mmol), $[\text{Au}_2\text{Cl}_2(\text{dppp})]$ (0.2 g, 0.23 mmol). Yield = 0.15 g (59%). ^1H NMR (CDCl_3): δ 8.73 (s, 2H, $^3J_{\text{HH}} = 2.7$ Hz, 5-pz); 7.73 (m, 8H, $\text{Ph}_2\text{P}(\text{CH}_2)_3\text{PPh}_2$); 7.51 (s, 4H, $^3J_{\text{HH}} = 1.8$ Hz, 3-pz); 7.41 (m, 12H, $\text{Ph}_2\text{P}(\text{CH}_2)_3\text{PPh}_2$); 6.41 (s, 2H, $^3J_{\text{HH}} = 2.7$ Hz, 4-pz); 2.93 (m, 2H, $\text{Ph}_2\text{PCH}_2\text{CH}_2\text{CH}_2\text{PPh}_2$); 1.99 (m, 4H, $\text{Ph}_2\text{PCH}_2\text{CH}_2\text{CH}_2\text{PPh}_2$). $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3): δ 218.9 (C(C=S)); 147.5 (C(5-pz)); 140.8 (C(3-pz)); 133.2–129.2 (phenyl region); 106.2 (C(4-pz)); 28.5 (2C, $^1J_{\text{C-P}} = 62.7$ Hz, P– CH_2 , $\text{Ph}_2\text{PCH}_2\text{CH}_2\text{CH}_2\text{PPh}_2$); 21.4 (C, $\text{Ph}_2\text{PCH}_2\text{CH}_2\text{CH}_2\text{PPh}_2$). IR (KBr, cm^{-1}): $\nu_{\text{C=N}} = 1628$, $\nu_{\text{C-S}} = 1088$, $\nu_{\text{C=S}} = 891$. $^{31}\text{P}\{^1\text{H}\}$ NMR (CDCl_3): δ 32.8. Anal. Calc. for $\text{C}_{35}\text{H}_{32}\text{Au}_2\text{N}_4\text{P}_2\text{S}_4$: C 38.47, H 2.95, N 5.13, S 11.74. Found: C 38.99, H 3.08, N 5.09, S 11.45%.

Bis(3,5-dimethylpyrazolyl-1-yl)dithiocarbamate-bis-(diphenylphosphino)propane Dinuclear Gold(I) (8). **L2** (0.1 g, 0.46 mmol), $[\text{Au}_2\text{Cl}_2(\text{dppp})]$ (0.2 g, 0.23 mmol). Yield = 0.13 g (50%). ^1H NMR (CDCl_3): δ 7.66 (m, 8H, $\text{Ph}_2\text{P}(\text{CH}_2)_3\text{PPh}_2$); 7.34 (m, 12H, $\text{Ph}_2\text{P}(\text{CH}_2)_3\text{PPh}_2$); 5.85 (s, 2H, 4-pz); 2.91 (m, 2H, $\text{Ph}_2\text{PCH}_2\text{CH}_2\text{CH}_2\text{PPh}_2$); 1.88 (m, 4H, $\text{Ph}_2\text{PCH}_2\text{CH}_2\text{CH}_2\text{PPh}_2$); 2.24, 2.26 (d, 12H, CH_3 , 3,5-pz). $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3): δ 215.9 (C(C=S)); 153.3 (C(5-pz)); 144.2 (C(3-pz)); 133.4–129.2 (phenyl region); 112.5 (C(4-pz)); 27.9 (2C, $^1J_{\text{C-P}} = 58.4$ Hz, P– CH_2 , $\text{Ph}_2\text{PCH}_2\text{CH}_2\text{CH}_2\text{PPh}_2$); 20.2 (C, $\text{Ph}_2\text{PCH}_2\text{CH}_2\text{CH}_2\text{PPh}_2$); 13.8 (C(CH_3 , 5-pz)); 13.2 (C(CH_3 , 3-pz)). IR (KBr, cm^{-1}): $\nu_{\text{C=N}} = 1653$, $\nu_{\text{C-S}} = 1099$, $\nu_{\text{C=S}} = 957$. $^{31}\text{P}\{^1\text{H}\}$ NMR (CDCl_3): δ 29.2. Anal. Calc. for $\text{C}_{39}\text{H}_{40}\text{Au}_2\text{N}_4\text{P}_2\text{S}_4$: C 40.77, H 3.51, N 4.88, S 11.16. Found: C 40.99, H 3.74, N 4.79, S 11.10%.

Bis(indazolyl-1-yl)dithiocarbamate-bis-(diphenylphosphino)propane Dinuclear Gold(I) (9). **L3** (0.04 g, 0.22 mmol), $[\text{Au}_2\text{Cl}_2(\text{dppp})]$ (0.1 g, 0.11 mmol). Yield = 0.07 g (54%). ^1H NMR (CDCl_3): δ 9.13 (s, 1H, Ha); 8.16 (s, 1H, Hb); 7.73 (1H, Hd); 7.69 (d, 1H, $^3J_{\text{HH}} = 8.0$ Hz, He); 7.66 (m, 8H, $\text{Ph}_2\text{P}(\text{CH}_2)_3\text{PPh}_2$); 7.45 (t, 1H, $^4J_{\text{HH}} = 16.8$ Hz, Hc); 7.42 (m, 12H, $\text{Ph}_2\text{P}(\text{CH}_2)_3\text{PPh}_2$); 2.89 (m, 2H, $\text{Ph}_2\text{PCH}_2\text{CH}_2\text{CH}_2\text{PPh}_2$); 1.92 (m, 4H, $\text{Ph}_2\text{PCH}_2\text{CH}_2\text{CH}_2\text{PPh}_2$). $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3): δ 213.8 (C(C=S)); 140.3 (C(5C-pz)); 139.1 (C(3C-pz)); 133.5–129.2 (phenyl region); 127.9 (C(7C-Ph)); 124.2 (C(4C-Ph)); 121.1 (C(8C-Ph, 9C-Ph)); 118.3 (C(6C-Ph)); 28.6 (2C, $^1J_{\text{C-P}} = 64.2$ Hz, P– CH_2 , $\text{Ph}_2\text{PCH}_2\text{CH}_2\text{CH}_2\text{PPh}_2$); 19.8 ($\text{Ph}_2\text{PCH}_2\text{CH}_2\text{CH}_2\text{PPh}_2$). IR (KBr, cm^{-1}):

$\nu_{\text{C=N}} = 1606$, $\nu_{\text{C-S}} = 1150$, $\nu_{\text{C=S}} = 993$. $^{31}\text{P}\{^1\text{H}\}$ NMR (CDCl_3): δ 29.2. ESI-MS: m/z 1190.9 $[\text{Au}_2(\text{L3})_2(\text{dppp})]^+$ (5%). Anal. Calc. for $\text{C}_{43}\text{H}_{36}\text{Au}_2\text{N}_4\text{P}_2\text{S}_4$: C 43.29, H 3.04, N 4.70, S 10.75. Found: C 43.23, H 3.05, N 4.62, S 10.56%.

Synthesis of Dinuclear Bis(pyrazol-1-ylthiocarbamate)-bis-(diphenylphosphino)hexane Gold(I) (10). L1 (0.02 g, 0.11 mmol), $[\text{Au}_2\text{Cl}_2(\text{dpph})]$ (0.05 g, 0.05 mmol). Yield = 0.04 g (70%). ^1H NMR (CDCl_3): δ 8.78 (d, 2H, $^3J_{\text{HH}} = 2.7$ Hz, 5-pz); 7.80 (s, 2H, 3-pz); 7.78 (m, 8H, $(\text{Ph})_2\text{P}(\text{CH}_2)_6\text{P}(\text{Ph})_2$); 7.49 (m, 12H, $(\text{Ph})_2\text{P}(\text{CH}_2)_6\text{P}(\text{Ph})_2$); 6.43 (s, 2H, 4-pz); 2.87 (bs, 4H, $\text{Ph}_2\text{P}(\text{CH}_2)_2(\text{CH}_2)_2(\text{CH}_2)_2\text{PPh}_2$); 1.49 (bs, 8H, $\text{Ph}_2\text{P}(\text{CH}_2)_2(\text{CH}_2)_2(\text{CH}_2)_2\text{PPh}_2$). $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3): δ 213.8 (C(C=S)); 140.3 (C(5C-pz)); 139.1 (C(3C-pz)); 133.5–129.2 (phenyl region); 127.9 (C(7C-Ph)); 124.2 (C(4C-Ph)); 121.1 (C(8C-Ph, 9C-Ph)); 118.3 (C(6C-Ph)); 30.1 (4C, $\text{PPh}_2(\text{CH}_2)_2(\text{CH}_2)_2(\text{CH}_2)_2\text{PPh}_2$); 25.9 (2C, $\text{Ph}_2\text{P}(\text{CH}_2)_2(\text{CH}_2)_2(\text{CH}_2)_2\text{PPh}_2$). IR (KBr, cm^{-1}): $\nu_{\text{C=N}} = 1620$, $\nu_{\text{C-S}} = 1108$, $\nu_{\text{C=S}} = 960$. $^{31}\text{P}\{^1\text{H}\}$ NMR (CDCl_3): δ 32.3. Anal. Calc. for $\text{C}_{38}\text{H}_{38}\text{Au}_2\text{N}_4\text{P}_2\text{S}_4$: C 40.22, H 3.37, N 4.94, S 11.30. Found: C 40.17, H 3.45, N 4.36, S 10.28%.

Synthesis of Dinuclear Bis(3,5-dimethylpyrazol-1-ylthiocarbamate)-bis-(diphenylphosphino)hexane Gold(I) (11). L2 (0.09 g, 0.42 mmol), $[\text{Au}_2\text{Cl}_2(\text{dpph})]$ (0.2 g, 0.2 mmol). Yield = 0.12 g (50%). ^1H NMR (CDCl_3): δ 7.67 (m, 8H, $\text{Ph}_2\text{P}(\text{CH}_2)_6\text{PPh}_2$); 7.40 (m, 12H, $\text{Ph}_2\text{P}(\text{CH}_2)_6\text{PPh}_2$); 6.00 (s, 2H, 4-pz); 2.76 (bs, 4H, $\text{Ph}_2\text{P}(\text{CH}_2)_2(\text{CH}_2)_2(\text{CH}_2)_2\text{PPh}_2$); 2.18, 2.27 (d, 12H, CH₃, 3,5-pz); 1.60, 1.52 (bs, 8H, $\text{Ph}_2\text{P}(\text{CH}_2)_2(\text{CH}_2)_2(\text{CH}_2)_2\text{PPh}_2$). $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3): δ 215.9 (C(C=S)); 153.3 (C(5-pz)); 144.2 (C(3-pz)); 133.4–129.2 (phenyl region); 112.5 (C(4-pz)); 31.2 (4C, $\text{PPh}_2(\text{CH}_2)_2(\text{CH}_2)_2(\text{CH}_2)_2\text{PPh}_2$); 26.7 (2C, $\text{Ph}_2\text{P}(\text{CH}_2)_2(\text{CH}_2)_2(\text{CH}_2)_2\text{PPh}_2$); 13.8 (C(5-pz)); 13.2 (C(3-pz)). IR (KBr, cm^{-1}): $\nu_{\text{C=N}} = 1653$, $\nu_{\text{C-S}} = 1099$, $\nu_{\text{C=S}} = 957$. $^{31}\text{P}\{^1\text{H}\}$ NMR (CDCl_3): δ 32.7. Anal. Calc. for $\text{C}_{42}\text{H}_{46}\text{Au}_2\text{N}_4\text{P}_2\text{S}_4$: C 42.36, H 3.89, N 4.70, S 10.77. Found: C 42.12, H 3.84, N 4.62, S 10.56%.

Synthesis of Dinuclear Bis(indazol-1-ylthiocarbamate)-bis-(diphenylphosphino)hexane Gold(I) (12). L3 (0.1 g, 0.44 mmol), $[\text{Au}_2\text{Cl}_2(\text{dpph})]$ (0.2 g, 0.22 mmol). Yield = 0.14 g (52%). ^1H NMR (CDCl_3): δ 9.18 (d, 1H, $^4J_{\text{HH}} = 8.4$ Hz, (Ha)); 8.10 (s, 1H, (Hb)); 7.49 (d, 1H, $^3J_{\text{HH}} = 8.7$ Hz, (He)); 7.24 (t, 1H, $^4J_{\text{HH}} = 15.0$ Hz, (Hd)); 7.32 (t, 1H, $^4J_{\text{HH}} = 15.0$ Hz, (Hc)); 7.66 (m, 8H, $\text{Ph}_2\text{P}(\text{CH}_2)_6\text{PPh}_2$); 7.42 (m, 12H, $\text{Ph}_2\text{P}(\text{CH}_2)_6\text{PPh}_2$); 2.39 (m, 4H, $\text{Ph}_2\text{P}(\text{CH}_2)_2(\text{CH}_2)_2(\text{CH}_2)_2\text{PPh}_2$); 1.57 (m, 8H, $\text{Ph}_2\text{P}(\text{CH}_2)_2(\text{CH}_2)_2(\text{CH}_2)_2\text{PPh}_2$). $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3): δ 213.8 (C(C=S)); 140.3 (C(5C-pz)); 139.1 (C(3C-pz)); 133.5–129.2 (phenyl region); 127.9 (C(7C-Ph)); 124.2 (C(4C-Ph)); 121.1 (C(8C-Ph, 9C-Ph)); 118.3 (C(6C-Ph)); 28.9 (4C, $\text{PPh}_2(\text{CH}_2)_2(\text{CH}_2)_2(\text{CH}_2)_2\text{PPh}_2$); 25.6 (2C, $\text{Ph}_2\text{P}(\text{CH}_2)_2(\text{CH}_2)_2(\text{CH}_2)_2\text{PPh}_2$). IR (KBr, cm^{-1}): $\nu_{\text{C=N}} = 1657$, $\nu_{\text{C-S}} = 1099$, $\nu_{\text{C=S}} = 994$. $^{31}\text{P}\{^1\text{H}\}$ NMR (CDCl_3): δ 31.8. Anal. Calc. for $\text{C}_{46}\text{H}_{42}\text{Au}_2\text{N}_4\text{P}_2\text{S}_4$: C 44.74, H 3.43, N 4.54, S 10.39. Found: C 44.37, H 3.46, N 4.50, S 10.32%.

Crystal Structure Determination. Crystals of **3** and **4a** were mounted in oil on a glass fiber, and data collection was performed on a Bruker CCD-1000 diffractometer with Mo K_α ($\lambda = 0.71073$ Å) radiation at a diffractometer-to-crystal distance of 4.9 cm. The reflections were successfully indexed by an automated indexing routine built in the SMART program. The data were corrected for Lorentz and polarization effects. Intensity data for **12** were collected at 150 K on a D8 goniostat equipped with a Bruker APEXII CCD detector at beamline 11.3.1 at the Advanced Light Source (Lawrence Berkeley National Laboratory) using synchrotron radiation tuned to $\lambda = 0.7749$ Å. The data frames were collected using the program APEX2 and processed using the program SAINT routine within APEX2. Option SQUEEZE of program PLATON⁵⁶ was used to correct the diffraction data for diffuse scattering effects and to identify the solvate molecule.

The absorption correction for the all three complexes was based on fitting a function to the empirical transmission surface as sampled by multiple equivalent measurements. A successful solution by the direct methods provided all non-hydrogen atoms from the E-map. All non-hydrogen atoms were refined with anisotropic displacement coefficients. All hydrogen atoms were included in the structure factor

calculation at idealized positions and were allowed to ride on the neighboring atoms with relative isotropic displacement coefficients.⁵⁷

■ TESTING OF COMPOUNDS FOR ANTICANCER ACTIVITY

Biological Reagents and Instrumentation. All commercial reagents were used as received. Phosphate-buffered saline (PBS), Eagle's RPMI-1640 medium, 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide (MTT) kit, phytohemagglutinin-protein form (PHA-P) and the 96-well flat-bottomed culture plates were all purchased from BD Biosciences Ltd. Cisplatin was purchased from Sigma Aldrich. Eagle's medium with 0.1 mM nonessential amino acids was prepared by adding 2 mM L-glutamine, 1.0 mM sodium pyruvate, and 5% bovine fetal calf serum to the pure Eagle's RPMI-1640 medium. A total of fifteen compounds (**L1–L3**, **1–3**, and **7–12**) were screened for their anticancer activities. Human cervix epithelial carcinoma (HeLa) cells and human lymphocytes (PBMCs), from preservative-free heparinised peripheral blood, were obtained from the Department of Pharmacology and Pretoria Medical Hospital, University of Pretoria, South Africa. The absorbance values were recorded on a Whittaker Microplate Reader 2001 spectrophotometer at 570 nm and the reference wavelength of 630 nm.

Two complexes (**10** and **11**) were further tested against the 60-cell-line panel at the DTP using their internal procedures.^{58,59} From this study, three important (cytostatic and cytotoxic) properties are evident, namely, cytostatic values that include molar concentrations of drug required for 50% growth inhibition (GI_{50}), cytostatic values that include molar concentrations of drug required for total growth inhibition (TGI), and cytostatic values that include molar concentrations of drug required to kill 50% of the cell population (LC_{50}).

Cell Culture and Drug Treatment. HeLa cells were cultured in Eagle's medium with 0.1 mM nonessential amino acids, 2 mM L-glutamine, 1.0 mM sodium pyruvate, and 5% bovine fetal calf serum at 37 °C in an atmosphere of 5% CO_2 . Cells were placed in 96-well sterile plates, at a density of 1×10^4 cells/well in 100 μL of medium, and incubated for 1 h. Subsequently, the ligands or gold compounds were added, with concentrations ranging from 0 to 100 μM . Cytotoxicity was determined by using MTT to stain treated HeLa cells after 7 d, according to literature methods.⁶⁰ MTT dye is reduced by living cells to yield a soluble formazan product that can be assayed colorimetrically.⁶⁰ A 20 μL volume of freshly prepared MTT (5 mg/mL) was added to each well, and the cells were incubated for another 4 h. Cell survival was evaluated by measuring absorbance at 570 nm, using a Whittaker Microplate Reader 2001. All experiments were performed in triplicate.

The inhibition of the growth of normal cells by the complexes tested was also measured by employing human lymphocytes (PBMC) cells. The same procedure described above was used, except that the treated PBMC cells were incubated for 3 d as opposed to 7 d for HeLa cells. The aim of testing these compounds on normal cells was to determine whether the compounds could target the cancerous (HeLa) cells specifically and not the normal cells. Lymphocytes were divided into two groups, namely, (i) normal cells that were stimulated using PHA-P so as to increase their proliferation rate (stimulated lymphocytes) and (ii) unstimulated normal cells (resting lymphocytes).

■ ASSOCIATED CONTENT

■ Supporting Information

Crystallographic information data (CIF); ESI-MS of **9**; biological data for compounds **11** and **12**. This material is available free of charge via the Internet at <http://pubs.acs.org>. The deposition numbers CCDC 680273, 949434, and 920397 contain the supplementary crystallographic data for compounds **3**, **4a**, and **12**, respectively. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.

■ AUTHOR INFORMATION

Corresponding Author

*Email: jdarkwa@uj.ac.za.

Present Address

[†]School of Chemistry and Physics, University of KwaZulu-Natal, Westville Campus, Private Bag X54001, Durban, 4000, South Africa.

Notes

The authors declare no competing financial interest.

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