# Asymmetric Construction of Highly Functionalized Spirobarbiturate-Cyclopentenes through Chiral Phosphine-Catalyzed [3+2] Annulation of Morita–Baylis–Hillman Carbonates with Barbiturate-Derived Alkenes

Yang Liu,<sup>a</sup> Wenjun Yang,<sup>a</sup> Yang Wu,<sup>a</sup> Biming Mao,<sup>a</sup> Xing Gao,<sup>a</sup> Honglei Liu,<sup>a</sup> Zhanhu Sun,<sup>a</sup> Yumei Xiao,<sup>a</sup> and Hongchao Guo<sup>a,\*</sup>

<sup>a</sup> Department of Applied Chemistry, China Agricultural University, 2 West Yuanmingyuan Road, Beijing 100193, People's Republic of China Fax: (+86)-10-6273-0784; e-mail: hchguo@cau.edu.cn

Received: April 27, 2016; Revised: June 9, 2016; Published online:

Supporting information for this article is available on the WWW under http://dx.doi.org/10.1002/adsc.201600450.

**Abstract:** A multifunctional chiral phosphine-catalyzed enantioselective [3+2] annulation of Morita-Baylis–Hillman carbonates with barbiturate-derived alkenes has been achieved under mild conditions, providing a variety of chiral spirobarbiturate-cyclopentenes in moderate to excellent yields with moderate to excellent diastereo- and enantioselectivities.

**Keywords:** alkenes; allenoates; annulation; phosphines; spirobarbiturates

Barbituric acid derivatives are important pharmacological compounds, displaying various activities such as sedative, anesthetic, anxiolytic, anticonvulsant, analeptic, anticancer, anti-AIDS and immunomodulating activities.<sup>[1]</sup> As a type of barbiturate, spirobarbiturates also showed a range of pharmacological and physiological activities and have attracted much attention.<sup>[2,3]</sup> Considering functionalized five-membered carbocycles are a common structural motif in medicinally important compounds and natural products,<sup>[4]</sup> spirobarbiturate-cyclopentenes with the combination of barbiturate and cyclopentene moieties are very appealing in the search for medicinally active compounds. Recently, a synthetic approach to spirobarbiturate-cyclopentenes has been achieved through a ring-closing metathesis (RCM) reaction (Scheme 1).<sup>[5]</sup> However, this method is not feasible for multi-substituted spirobarbiturate-cyclopentenes. Especially, to the best of our knowledge, an asymmetric construction of spirobarbiturate-cyclopentenes has not been reported. In this context, we considered developing an enantioselective tool to synthesize chiral spirobarbiturate-cyclopentenes through phosphine-catalyzed asymmetric [3+2] annulation of barbiturate-derived alkenes with Morita–Baylis–Hillman (MBH) carbonates.

Phosphine-catalyzed annulation reactions are very powerful methods in the syntheses of carbocyclic and heterocyclic compounds and serve as the key step in the total synthesis of some natural products.<sup>[6]</sup> Particularly, these annulations have also successfully been applied in the synthesis of complex spirocyclic compounds.<sup>[7]</sup> Among various annulation reactions, the phosphine-catalyzed [3+2] annulation reaction of MBH carbonates with olefins has emerged as an important tool to synthesize cyclopentenes since Lu first reported a phosphine-catalyzed [3+2] annulation of MBH carbonates as a three-carbon synthon with electron-deficient alkenes.<sup>[6j,8]</sup> Employing this reaction, a few spirocyclic molecules fused with a cyclopentene moiety have been constructed. Lu, Shi and Barbas independently developed chiral phosphine-catalyzed asymmetric [3+2] annulation reactions of methyleneindolinones with MBH carbonates to prepare



Scheme 1. Synthesis of spirobarbiturate-cyclopentenes.

© 2016 Wiley-VCH Verlag GmbH & Co. KGaA, Weinheim

These are not the final page numbers! **77** 

Wiley Online Library

1

Adv. Synth. Catal. 0000, 000, 0-0



chiral spirocyclopentene oxindoles.<sup>[9]</sup> Barbas achieved the asymmetric [3+2] annulation of methylenebenzofuranone with MBH carbonates to give spirocyclopentene benzofuranones.<sup>[10]</sup> Shi described the asymmetric [3+2] annulation of MBH carbonates with 2-arylideneindane-1,3-diones to provide spirocyclopenteneindenediones.<sup>[11]</sup> Chen and Liu used MBH carbonates of isatin for the asymmetric [3+2] annulation with alkenes to furnish cyclopentene-fused spirocyclic heterocycles.<sup>[12]</sup> As part of our continuing efforts on phosphine-catalyzed annulations,<sup>[13]</sup> herein, we present the multifunctional chiral phosphine-catalyzed asymmetric [3+2] annulations of MBH carbonates with barbiturate-derived alkenes to synthesize enantioenriched spirobarbiturate-cyclopentenes (Scheme 1).

In our initial attempts, we screened different kinds of chiral phosphines in the asymmetric reaction between barbiturate-derived alkene (1a) and MBH carbonate (2a) in toluene at 80 °C. As shown in Table 1, Kwon phosphines P1 and P2<sup>[14]</sup> displayed opposing results. With the use of chiral phosphine P1 as the catalyst, the desired spirobarbiturate-cyclopentene 3aa was obtained in excellent yield but with moderate

Table 1. Catalyst screen and optimization studies.<sup>[a]</sup>

enantioselectivity (entry 1). In contrast, chiral phosphine P2 displayed excellent enantioselectivity but moderate catalytic activity under otherwise identical conditions (entry 2). Spirocyclic chiral phosphine **P3**<sup>[15]</sup> promoted the reaction to give the product **3aa** in 83% yield albeit with 65% ee (entry 3). Satisfactorily, two multifunctional phosphines P4 and P5<sup>[16]</sup> demonstrated good catalytic capability (entries 4 and 5). Although **P5** delivered better enantioselectivity than P4 did, P4 is a preferred catalyst for this reaction in terms of both yield and enantioselectivity (entry 4 vs. 5). With P4 as the catalyst, we next performed solvent screening to improve the enantioselectivity. These experiments revealed that trifluorotoluene was an optimal solvent (entries 6 and 7). In trifluorotoluene, the ee value was increased to 93% (entry 7). Using 4 Å MS as an additive, the yield was slightly increased to 80% (entry 8). Since the base has been demonstrated to be favorable to phosphine-catalyzed annulations in the previous report,<sup>[17]</sup> a catalytic amount of K<sub>2</sub>CO<sub>3</sub> was also examined as an additive. To our delight, the yield was indeed marginally enhanced to 85% (entry 9). The base might help in formation of a phosphonium enolate zwitterion from



| Entry            | Px        | Solvent           | Temperature [°C] | Time [h] | Yield [%] <sup>[b]</sup> | ee [%] <sup>[c]</sup> |
|------------------|-----------|-------------------|------------------|----------|--------------------------|-----------------------|
| 1                | P1        | toluene           | 80               | 18       | 98                       | 37                    |
| 2                | P2        | toluene           | 80               | 18       | 45                       | 91                    |
| 3                | <b>P3</b> | toluene           | 80               | 16       | 83                       | 65                    |
| 4                | P4        | toluene           | 80               | 16       | 87                       | 85                    |
| 5                | P5        | toluene           | 80               | 16       | 67                       | 91                    |
| 6                | <b>P4</b> | DCE               | 80               | 18       | 76                       | 90                    |
| 7                | <b>P4</b> | PhCF <sub>3</sub> | 80               | 16       | 76                       | 93                    |
| 8 <sup>[d]</sup> | <b>P4</b> | PhCF <sub>3</sub> | 80               | 16       | 80                       | 93                    |
| 9 <sup>[e]</sup> | P4        | PhCF <sub>3</sub> | 80               | 16       | 85                       | 96                    |

<sup>[a]</sup> Unless otherwise stated, reactions of **1a** (0.1 mmol) and **2a** (0.12 mmol) were performed in the presence of phosphine (0.02 mmol) in 2 mL of solvent.

<sup>[b]</sup> Isolated yield. The dr is >20:1, determined by <sup>1</sup>H NMR analysis.

<sup>[c]</sup> Determined by chiral HPLC analysis.

 $^{[d]}$  50 mg of 4 Å MS were used as the additive.

<sup>[e]</sup> 50 mg of 4 Å MS and 20 mol% of  $K_2CO_3$  were used.

*Adv. Synth. Catal.* **0000**, 000, 0-0

## These are not the final page numbers! **77**

#### © 2016 Wiley-VCH Verlag GmbH & Co. KGaA, Weinheim



Table 2. Scope of MBH carbonates 2.<sup>[a]</sup>



P4 (20 mol%) PhCF<sub>3</sub>, 4 Å MS K<sub>2</sub>CO<sub>3</sub>, 80 °C



| Entry            | R   | Time [h] | 3          | Yield [%] <sup>[b]</sup> | <i>ee</i> [%] <sup>[c]</sup> |
|------------------|---|----------|------------|--------------------------|------------------------------|
| 1 <sup>[d]</sup> | $2-MeC_{6}H_{4}$ ( <b>2b</b> )              | 18       | 3ab        | 88                       | 91                           |
| 2                | $3-\text{MeC}_{6}\text{H}_{4}(2\mathbf{c})$ | 18       | 3ac        | 99                       | 93                           |
| 3                | $4 - \text{MeC}_{6}H_{4}$ (2d)              | 16       | 3ad        | 80                       | 91                           |
| 4                | $2-\text{MeOC}_6H_4$ (2e)                   | 16       | 3ae        | 98                       | 91                           |
| 5                | $4-\text{MeOC}_6H_4$ (2f)                   | 17       | 3af        | 98                       | 93                           |
| 6                | $2-FC_6H_4$ (2g)                            | 18       | 3ag        | 90                       | 90                           |
| 7                | $3-FC_{6}H_{4}(2h)$                         | 16       | 3ah        | 80                       | 91                           |
| 8                | $4 - FC_6 H_4$ (2i)                         | 18       | 3ai        | 81                       | 91                           |
| 9                | $2-\text{ClC}_6\text{H}_4(2\mathbf{j})$     | 16       | 3aj        | 97                       | 87                           |
| 10               | $3-ClC_{4}H_{4}(2\mathbf{k})$               | 18       | 3ak        | 64                       | 93                           |
| 11               | $4-ClC_6H_4$ (21)                           | 16       | 3al        | 86                       | 93                           |
| 12               | $2-BrC_6H_4$ ( <b>2m</b> )                  | 20       | 3am        | 96                       | 85                           |
| 13               | $3-\operatorname{BrC}_6H_4(2\mathbf{n})$    | 18       | 3an        | 85                       | 87                           |
| 14               | 2-naphthyl ( $20$ )                         | 16       | <b>3ao</b> | 87                       | 93                           |
| 15               | 2-thienvl (2p)                              | 18       | 3ap        | 90                       | 90                           |
| 16               | i-Pr (2q)                                   | 24       | 3ag        | NR <sup>[e]</sup>        | _                            |
| 17               | cyclohexyl (2r)                             | 24       | 3ar        | NR                       | _                            |

[a] Reactions of 1a (0.1 mmol) and 2 (0.12 mmol) were performed in the presence of P4 (0.02 mmol), K<sub>2</sub>CO<sub>3</sub> (0.02 mmol) and 50 mg of 4Å MS at 80°C in 2 mL of PhCF<sub>3</sub>.

<sup>[b]</sup> Isolated yield. Unless otherwise stated, the dr is >20:1, determined by <sup>1</sup>H NMR analysis.

<sup>[c]</sup> Determined by chiral HPLC analysis.

 $^{[d]} dr = 10:3.$ 

<sup>[e]</sup> No reaction.

MBH carbonate, thus slightly increasing the yield. On the basis of the above-mentioned results, the optimized conditions are as follows: reaction of **1a** and **2a** in trifluorotoluene at 80 °C using 20 mol% of **P4** in the presence of 4 Å MS and  $K_2CO_3$ .

Under the optimal conditions, the scope of the MBH carbonates 2 in this [3+2] annulation was investigated (Table 2). Generally, whether using electronrich or electron-deficient aryl-substituted MBH carbonates, the spirocyclic products were obtained in good to excellent yields (64-99%) with good to excellent diastereo- and enantioselectivities (85-93% ee) (entries 1–15). The high diastereoselectivities might be attributed to steric hindrance from the phosphine and the special structure of barbiturate ring. The position of the substituent on the benzene ring seemed to have no remarkable influence on the activities and stereoselectivities (entries 1-15). Notably, 2-naphthyland 2-thienyl-substituted substrates also reacted efficiently with barbiturate-derived alkene, affording the corresponding spirocyclic products with high yields and excellent ee values (entries 14 and 15). Unfortunately, two alkyl-substituted MBH carbonates did not work (entries 16 and 17). The absolute configuration of the spirocyclic products was assigned by an X-ray crystallographic analysis of the product **3al** (entry 11).<sup>[18]</sup>

Application of the standard conditions to various barbiturate-derived alkenes 1 is summarized in Table 3. Variations of the substituents on the benzene moiety were tolerated. A variety of barbiturate-derived alkenes, regardless of the substitution pattern and electronic nature, worked well to produce the corresponding cycloadducts in high yields (60-99%) with excellent enantioselectivities (91-99% ee) (entries 1-13). All products, except 3ca and 3ka (entries 3 and 11), were generated with >20:1 diastereometric ratios. A 2-naphthyl-substituted alkene (1m) led to excellent yield, diastereoselectivity and enantioselectivity (86% yield, >20:1 dr, 93% ee) (entry 14). The 2-furanylalkene 1n was also compatible with the current system, affording the product 3na in 99% ee, albeit in only 45% yield (entry 15). For phosphinecatalyzed annulation reactions, alkenes bearing aliphatic substituents are generally challenging substrates. In this [3+2] annulation, a cyclohexyl-substi-

*Adv. Synth. Catal.* **0000**, 000, 0–0



Table 3. Scope of barbiturate-derived alkenes 1.<sup>[a]</sup>



| Entry             | R   | Time [h] | 3           | Yield [%] <sup>[b]</sup> | <i>ee</i> [%] <sup>[c]</sup> |
|-------------------|---|----------|-------------|--------------------------|------------------------------|
| 1                 | Ph ( <b>1</b> a)                                  | 16       | <b>3</b> aa | 85                       | 96                           |
| 2                 | $2 - MeC_6H_4$ (1b)                               | 16       | 3ba         | 98                       | 91                           |
| 3 <sup>[d]</sup>  | $3-\text{MeC}_{6}\text{H}_{4}$ (1c)               | 18       | 3ca         | 97                       | 99                           |
| 4                 | $4 - \text{MeC}_{6}H_{4}$ (1d)                    | 16       | 3da         | 98                       | 93                           |
| 5                 | $2-\text{MeOC}_6H_4$ (1e)                         | 18       | 3ea         | 98                       | 97                           |
| 6                 | 4-MeOC <sub>6</sub> H <sub>4</sub> (1f)           | 18       | 3fa         | 92                       | 91                           |
| 7                 | $2-FC_6H_4$ (1g)                                  | 18       | 3ga         | 87                       | 99                           |
| 8                 | $4 - FC_6 H_4$ ( <b>1h</b> )                      | 18       | 3ha         | 90                       | 91                           |
| 9                 | $2-ClC_{6}H_{4}(1i)$                              | 15       | 3ia         | 60                       | 97                           |
| 10                | $4-ClC_{4}H_{4}(1i)$                              | 15       | 3ia         | 99                       | 98                           |
| 11 <sup>[e]</sup> | $2-BrC_6H_4$ (1k)                                 | 20       | 3ka         | 70                       | 99                           |
| 12                | $4-\operatorname{BrC}_{6}\operatorname{H}_{4}(1)$ | 20       | 3la         | 89                       | 98                           |
| 14                | 2-naphthyl ( <b>1m</b> )                          | 16       | 3ma         | 86                       | 93                           |
| 15                | 2-thienvl ( <b>1n</b> )                           | 16       | 3na         | 45                       | 99                           |
| 16                | cyclohexyl (10)                                   | 24       | <b>3</b> 0a | 30                       | 81                           |

<sup>[a]</sup> Unless otherwise stated, reactions of **1** (0.1 mmol) and **2a** (0.12 mmol) were carried out in the presence of **P4** (0.02 mmol), K<sub>2</sub>CO<sub>3</sub> (0.02 mmol) and 50 mg of 4 Å MS at 80 °C in 2 mL of PhCF<sub>3</sub>.

<sup>[b]</sup> Isolated yield. Unless otherwise stated, the dr is >20:1, determined by <sup>1</sup>H NMR analysis.

<sup>[c]</sup> Determined by chiral HPLC analysis.

 $^{[d]} dr = 4:1.$ 

[e] dr = 4:5.

tuted alkene displayed moderate activity, leading to the spirocyclic product **30a** in 30% yield, but gratifyingly, an 81% *ee* was obtained (entry 16).

The products could further be transformed into other useful compounds (Scheme 2). The spirobarbiturate-cyclopentene **3an** was treated with DIBAL-H



Scheme 2. Synthetic transformations of the product.

*Adv. Synth. Catal.* **0000**, 000, 0-0

## These are not the final page numbers! **77**

in THF at 0 °C for 4 h, giving the reduction product **4** in 70% yield. In the presence of  $Pd(Ph_3P)_4$  and butyldi(1-adamantyl)phosphine, the coupling of **3an** with 4chlorophenylboronic acid proceeded smoothly in 1,1dimethoxyethane (DME) at 80 °C to afford the corresponding product **5** in 95% yield.

A reasonable mechanism is proposed in Scheme 3. The allylic phosphonium ylide **A** formed from nucleophilic addition of phosphine to MBH carbonate **2** un-



Scheme 3. A proposed mechanism.



dergoes a nucleophilic attack from the bottom of the carbon-carbon double bond of the barbiturate-derived alkene to give the phosphonium ylide **B**, which subsequently carries out intramolecular conjugate addition to furnish the [3+2] annulation and afford the phosphonium ylide **C**. Consequent formation of a carbon-carbon double bond and simultaneous ejection of phosphine leads to the annulation product **3**.

In summary, we have developed an efficient method for the asymmetric construction of biologically significant chiral spirobarbiturate-cyclopentenes through chiral phosphine-catalyzed asymmetric [3+2] annulation of MBH carbonates with barbiturate-derived alkenes in moderate to excellent yields with moderate to excellent diastereo- and enantioselectivities. The method will be a useful tool in the synthesis of biologically active compounds for the discovery of novel therapeutic agents.

### **Experimental Section**

#### **General Procedure**

Under a nitrogen atmosphere, to a stirred mixture of alkene **1** (0.1 mmol), MBH carbonate **2** (0.12 mmol),  $K_2CO_3$ (0.02 mmol) and 4Å MS (50 mg) in 2 mL of trifluorotoluene was added chiral phosphine **P4** (11 mg, 0.02 mmol) and the resulting mixture was stirred at 80 °C. Upon completion of the reaction as monitored by TLC, the mixture was concentrated under vacuum. The residue was purified through flash column chromatography (ethyl acetate/petroleum ether) to afford the corresponding annulation product.

### Acknowledgements

This work is supported by the NSFC (21372256 and 21572264), the National S&T Pillar Program of China (2015BAK45B01), Research Fund for the Doctoral Program of Higher Education of China (20120008110038) and Chinese Universities Scientific Fund (2016QC090).

#### References

 a) M. W. Johns, Drugs 1975, 9, 448; b) M. C. Smith, B. J. Riskin, Drugs 1991, 42, 365; c) H. Bruner, K. P. Ittner, D. Lunz, S. Schmatloch, T. Schmidt, M. Zabel, Eur. J. Org. Chem. 2003, 855; d) E. Maquoi, N. E. Sounni, L. Devy, F. Oliver, F. Frankenne, H. W. Krell, F. Grams, J. M. Foidart, A. Noel, Clin. Cancer Res. 2004, 10, 4038; e) L. L. Brunton, J. S. Lazo, L. P. Keith, Goodman & Gilman's the Pharmacological Basis of Therapeutics, 11<sup>th</sup> edn., McGraw-Hill, Inc., New York, 2006; f) K. E. Lyons, R. Pahwa, CNS Drugs 2008, 22, 1037; g) C. Uhlmann, W. Froscher, CNS Neurosci. Ther. 2009, 15, 24; h) D. J. Abraham, D. P. Rotella, Burger's Medicinal Chemistry, Drug Discovery and Develop*ment*, Wiley, Hoboken, **2010**; i) J. Wang, C. Medina, M. W. Radomski, J. F. Gilmer, *Bioorg. Med. Chem.* **2011**, *19*, 4985.

- [2] For bioactivities of spirobarbiturate derivatives, see:
  a) W. C. Lee, Jpn. J. Pharmacol. 1953, 2, 123; b) R. J. Fessenden, J. G. Larsen, M. D. Coon, J. S. Fessenden, J. Med. Chem. 1964, 7, 695; c) W. Fraser, C. J. Suckling, H. C. S. Wood, J. Chem. Soc. Perkin Trans. 1 1990, 3137; d) S. B. King, E. Stratford, C. Craig, E. K. Fifer, Pharm. Res. 1995, 12, 1240; e) E. M. Galati, M. T. Monforte, N. Miceli, E. Raneri, Farmaco 2001, 56, 459; f) A. Renard, J. Lhomme, M. Kotera, J. Org. Chem. 2002, 67, 1302; g) D. B. Ramachary, M. Kishor, Y. V. Reddy, Eur. J. Org. Chem. 2008, 975; h) L. Lomlin, J. Einsiedel, F. W. Heinemann, K. Meyer, P. Gmeiner, J. Org. Chem. 2008, 73, 3608.
- [3] For recent examples about synthesis of spirobarbiturates, see: a) M. N. Elinson, A. N. Vereshchagin, N. O. Stepanov, P.A. Belyakov, G.I. Nikishin, Tetrahedron Lett. 2010, 51, 6598; b) M. N. Elinson, A. N. Vereshchagin, N. O. Stepanov, T. A. Zaimovskaya, V. M. Merkulova, G. I. Nikishin, Tetrahedron Lett. 2010, 51, 428; c) K. Mori, S. Sueoka, T. Akiyama, J. Am. Chem. Soc. 2011, 133, 2424; d) A. P. Dieskau, M. S. Holzwarth, B. Plietker, J. Am. Chem. Soc. 2012, 134, 5048; e) E. O. Dorofeeva, M. N. Elinson, A. N. Vereshchagin, N. O. Stepanov, I. S. Bushmarinov, P. A. Belyakov, O. O. Sokolova, G. I. Nikishin, RSC Adv. 2012, 2, 4444; f) D. Bhuyan, R. Sarmaand, D. Prajapati, Tetrahedron Lett. 2012, 53, 6460; g) P. Borah, P. J. Bhuyan, Tetrahedron Lett. 2013, 54, 6949; h) A. N. Vereshchagin, M. N. Elinson, E. O. Dorofeeva, N. O. Stepanov, T. A. Zaimovskaya, G. I. Nikishin, Tetrahedron 2013, 69, 1945; i) E. Soleimani, H. Yazdani, P. Saei, Tetrahedron Lett. 2015, 56, 1635; j) H.-W. Zhao, T. Tian, B. Li, Z. Yang, H.-L. Pang, W. Meng, X.-Q. Song, X.-Q. Chen, J. Org. Chem. 2015, 80, 10380; k) L. De Crescentini, O. A. Attanasi, L. A. Campisi, G. Favi, S. Lillini, F. Ursini, F. Mantellini, Tetrahedron 2015, 71, 7282; l) A. Palasz, D. Ciez, B. Musielak, J. Kalinowska-Tluscik, Tetrahedron 2015, 71, 8911.
- [4] a) R. C. Hartley, S. T. Caldwell, J. Chem. Soc. Perkin Trans. 1 2000, 477; b) Prostaglandins: Prostaglandins, Leukotrienes and Essential Fatty Acids, (Eds.: D. F. Horrobin, M. S. Manku, P. Sirois, P. Borgeat), Churchill Livingston, Edinburgh, 2002; c) carbasugars: G. Rassu, L. Auzzas, L. Pinna, L. Battistini, C. Curti, Stud. Nat. Prod. Chem. 2003, 29, 449; d) H. Wu, H. Zhang, G. Zhao, Tetrahedron 2007, 63, 6454.
- [5] S. Kotha, A. Chandra Deb, R. V. Kumar, *Bioorg. Med. Chem. Lett.* 2005, 15, 1039.
- [6] For selected reviews on phosphine-promoted annulations, see: a) X. Lu, C. Zhang, Z. Xu, Acc. Chem. Res. 2001, 34, 535; b) L.-W. Ye, J. Zhou, Y. Tang, Chem. Soc. Rev. 2008, 37, 1140; c) B. J. Cowen, S. J. Miller, Chem. Soc. Rev. 2009, 38, 3102; d) A. Marinetti, A. Voituriez, Synlett 2010, 174; e) S.-X. Wang, X. Y. Han, F. R. Zhong, Y. Q. Wang, Y. X. Lu, Synlett 2011, 2766; f) Q.-Y. Zhao, Z. Lian, Y. Wei, M. Shi, Chem. Commun. 2012, 48, 1724; g) Y. C. Fan, O. Kwon, Chem. Commun. 2013, 49, 11588; h) Z. Wang, X. Xu, O. Kwon, Chem. Soc. Rev. 2014, 43, 2927; i) Y. Xiao, Z.

Adv. Synth. Catal. 0000, 000, 0-0

These are not the final page numbers! **77** 

© 2016 Wiley-VCH Verlag GmbH & Co. KGaA, Weinheim

5



Sun, H. Guo, O. Kwon, Beilstein J. Org. Chem. 2014, 10, 2089; j) P. Xie, Y. Huang, Org. Biomol. Chem. 2015, 13, 8578; k) Y. Xiao, H. Guo, O. Kwon, Aldrichimica Acta 2016, 49, 3. For applications of phosphine catalysis in the total synthesis of natural products, see: 1) J. C. Wang, M. J. Krische, Angew. Chem. 2003, 115, 6035; Angew. Chem. Int. Ed. 2003, 42, 5855; m) K. Agapiou, M. J. Krische, Org. Lett. 2003, 5, 1737; n) Y. S. Tran, O. Kwon, Org. Lett. 2005, 7, 4289; o) R. A. Jones, M. J. Krische, Org. Lett. 2009, 11, 1849; p) M. Sampath, P.-Y. B. Lee, T. P. Loh, Chem. Sci. 2011, 2, 1988; q) I. P. Andrews, O. Kwon, Chem. Sci. 2012, 3, 2510; r) R. A. Villa, Q. H. Xu, O. Kwon, Org. Lett. 2012, 14, 4634; s) G. A. Barcan, A. Patel, K. N. Houk, O. Kwon, Org. Lett. 2012, 14, 5388; t) L. Cai, K. Zhang, O. Kwon, J. Am. Chem. Soc. 2016, 138, 3298.

- [7] A. Voituriez, A. Marinetti, M. Gicquel, Synlett 2015, 26, 142.
- [8] Y. Du, X. Lu, C. Zhang, Angew. Chem. 2003, 115, 1065; Angew. Chem. Int. Ed. 2003, 42, 1035.
- [9] a) F. Zhong, X. Han, Y. Wang, Y. Lu, Angew. Chem. 2011, 123, 7983; Angew. Chem. Int. Ed. 2011, 50, 7837; b) H.-P. Deng, Y. Wei, M. Shi, Org. Lett. 2011, 13, 3348; c) B. Tan, N. R. Candeias, C. F. Barbas III, J. Am. Chem. Soc. 2011, 133, 4672; d) F. Zhong, G.-Y. Chen, X. Han, W. Yao, Y. Lu, Org. Lett. 2012, 14, 3764.
- [10] K. Albertshofer, B. Tan, C. F. Barbas III, Org. Lett. 2013, 15, 2958.
- [11] F. Hu, Y. Wei, M. Shi, Tetrahedron 2012, 68, 7911.
- [12] a) Y. Wang, L. Liu, T. Zhang, N.-J. Zhong, D. Wang, Y.-J. Chen, J. Org. Chem. 2012, 77, 4143; b) G. Zhan, M.-L. Shi, Q. He, W.-J. Lin, Q. Ouyang, W. Du, Y.-C. Chen, Angew. Chem. Int. Ed. 2016, 55, 2147.
- [13] a) R. Na, C. Jing, Q. Xu, H. Jiang, X. Wu, J. Shi, J. Zhong, M. Wang, D. Benitez, E. Tkatchouk, W. A. Goddard III, H. Guo, O. Kwon, J. Am. Chem. Soc. 2011, 133, 13337; b) C. Jing, R. Na, B. Wang, H. Liu, L. Zhang, J. Liu, M. Wang, J. Zhong, O. Kwon, H. Guo, Adv. Synth. Catal. 2012, 354, 1023; c) Z. Li, H. Yu, H. Liu, L. Zhang, H. Jiang, B. Wang, H. Guo, Chem. Eur. J. 2014, 20, 1731; d) H. Liu, Y. Liu, C. Yuan, G.-P.

Wang, S.-F. Zhu, Y. Wu, B. Wang, Z. Sun, Y. Xiao, Q.-L. Zhou, H. Guo, *Org. Lett.* **2016**, *18*, 1302.

- [14] C. E. Henry, Q. Xu, Y. C. Fan, T. J. Martin, L. Belding, T. Dudding, O. Kwon, J. Am. Chem. Soc. 2014, 136, 11890.
- [15] a) S.-F. Zhu, Y. Yang, L.-X. Wang, B. Liu, Q.-L. Zhou, Org. Lett. 2005, 7, 2333; b) B. Liu, S. F. Zhu, L.-X. Wang, Q.-L. Zhou, Tetrahedron: Asymmetry 2006, 17, 634; c) W. Zhang, S.-F. Zhu, X.-C. Qiao, Q.-L. Zhou, Chem. Asian J. 2008, 3, 2105. For the reviews on the development of chiral spirocyclic ligands and catalysts, see: d) J.-H. Xie, Q.-L. Zhou, Acc. Chem. Res. 2008, 41, 581; e) S.-F. Zhu, Q.-L. Zhou, in: Privileged Chiral Ligands and Catalysts, (Ed.: Q.-L. Zhou), Wiley-VCH: Weinheim, 2011, Chapter 4, pp 137–170.
- [16] a) H. Xiao, Z. Chai, C. W. Zheng, Y. Q. Yang, W. Liu, J. K. Zhang, G. Zhao, Angew. Chem. 2010, 122, 4569; Angew. Chem. Int. Ed. 2010, 49, 4467; b) F. Zhong, X. Han, Y. Wang, Y. Lu, Angew. Chem. 2011, 123, 7983; Angew. Chem. Int. Ed. 2011, 50, 7837; c) X. Han, Y. Wang, F. Zhong, Y. Lu, J. Am. Chem. Soc. 2011, 133, 1726; d) X. Han, F. Zhong, Y. Wang, Y. Lu, Angew. Chem. 2012, 124, 791; Angew. Chem. Int. Ed. 2012, 51, 767; e) F. Zhong, X. Dou, X. Han, W. Yao, Q. Zhu, Y. Meng, Y. Lu, Angew. Chem. 2013, 125, 977; Angew. Chem. Int. Ed. 2013, 52, 943; f) X. Han, W. Yao, T. Wang, Y. R. Tan, Z. Yan, J. Kwiatkowski, Y. Lu, Angew. Chem. 2014, 126, 5749; Angew. Chem. Int. Ed. 2014, 53, 5643; g) W. J. Yao, X. W. Dou, Y. Lu, J. Am. Chem. Soc. 2015, 137, 54; h) T. Wang, Z. Yu, D. L. Hoon, C. Y. Phee, Y. Lan, Y. Lu, J. Am. Chem. Soc. 2016, 138, 265; i) X. Han, W.-L. Chan, W. Yao, Y. Wang, Y. Lu, Angew. Chem. 2016, 128, 6602; Angew. Chem. Int. Ed. 2016, 55, 6492.
- [17] L. Zhang, H. Liu, G. Qiao, Z. Hou, Y. Liu, Y. Xiao, H. Guo, J. Am. Chem. Soc. 2015, 137, 4316.
- [18] CCDC 1470357 (3al) contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data\_ request/cif.

Adv. Synth. Catal. 0000, 000, 0-0

### COMMUNICATIONS

Asymmetric Construction of Highly Functionalized Spirobarbiturate-Cyclopentenes through Chiral Phosphine-Catalyzed [3+2] Annulation of Morita–Baylis–Hillman Carbonates with Barbiturate-Derived Alkenes

Adv. Synth. Catal. 2016, 358, 1-7

Yang Liu, Wenjun Yang, Yang Wu, Biming Mao, Xing Gao, Honglei Liu, Zhanhu Sun, Yumei Xiao, Hongchao Guo\*



7