

Available online at www.sciencedirect.com



Tetrahedron: Asymmetry 15 (2004) 17-19

Tetrahedron: Asymmetry

A concise synthesis of 2-diarylphosphino-2'-methoxy-1,1'binaphthalenes (MOPs) by simple resolution

Yunfei Luo, Feng Wang, Gangguo Zhu and Zhaoguo Zhang*

State Key Laboratory of Organometallic Chemistry, Shanghai Institute of Organic Chemistry, Chinese Academy of Sciences, 354 Fenglin Road, Shanghai 200032, China

Received 27 September 2003; accepted 21 October 2003

Abstract—A practical synthesis of 2-diarylphosphino-2'-methoxy-1,1'-binaphthalene (MOPs) was provided by employing the etherification of 1,1'-bi-2,2'-naphthol on zeolite; the ring opening of the ether with lithium; the reaction with chlorodiaryl phosphine and the resolution with (1S)-(+)-10-camphorsulfonyl chloride as the key steps.

© 2003 Elsevier Ltd. All rights reserved.

To develop new chiral ligands that might realize high enantioselectivity in asymmetric catalytic reactions is an important area in current research. Based on the chiral structure of 1,1'-binaphthalenyl-2,2'-diol, a chiral monophosphine ligand 2'-diphenylphosphanyl-1,1'binaphthalenyl-2-ol was reported by Morgans and co-workers.1 Later, Hayashi and co-workers2 reported an improved synthesis enroute to substituted chiral monophosphine ligands (MOPs) that are efficient for several types of palladium-catalyzed asymmetric reactions including asymmetric hydrosilylation of olefins,³ asymmetric 1,4-hydroboration of 1,3-enynes to form allenyl boranes,⁴ and asymmetric reduction of allylic esters with formic acid.⁵ In these reactions, the use of MOP-type ligands is crucial for obtaining high catalytic reactivity and regioselectivity as well as high enantioselectivity. Moreover, once the parent structure of chiral MOP-type ligands was established, it was expanded into other useful ligands, such as PHEST,⁶ BINAPHOS,⁷ BIPPHOS,⁷ BIPNITE⁷ etc., which now have been utilized in Pd-catalyzed asymmetric allylic substitution,^{6a} Rh-catalyzed highly enantioselective hydroformylation of olefins.⁷ Besides that the hydroxy group can also be converted into the thiol to produce another bidentate ligand 2'-diphenylphosphanyl-1,1'-binaphthalenyl-2thiol.8

Another method provided by Miyano et al.⁹ to prepare 2'-diphenylphosphinoyl-1,1'-binaphthalenyl-2-ol involves a nucleophilic aromatic substitution reaction of 1-methoxy-2-(diphenylphosphino)naphthalene with Grignard reagent. However, the monophosphine ligands made by this method were racemic.

Heinicke and co-workers¹⁰ reported a facile route to synthesize biaryl racemic monophosphine ligands from either dibenzofuran or dinaphthofuran by lithiummediated ring opening and a subsequent reaction with chlorodiphenylphosphine. However, attempts to resolve the racemic 2'-diphenylphosphinoyl-1,1'-binaphthalenyl-2-ol with camphorsulfonyl chloride failed, with only oxidized product being obtained.

It could be envisioned that the oxidized product could undergo further reaction with a chiral acid to form an ester, and the pair of diastereoisomers would have significantly different physical properties that would let them be separated from each other.

According to the literature procedure,¹⁰ we prepared the dinaphthofuran in high yield by employing HY zeolite with a SiO₂/Al₂O₃ ratio of about 14. The ring opening of the dinaphthofuran with lithium at rt in Et₂O afforded the intermediate dilithium salt **3**. The dilithium salt **3** was then quenched with chlorodiphenyl phosphine oxide instead of chlorodiphenyl phosphine to give the racemic hydroxy phosphine oxide **4**, which served as the precursor to be resolved (Scheme 1).

Treatment of 2'-diphenylphosphinoyl-1,1'-binaphthalenyl-2-ol **4** with (1*S*)-(+)-10-camphorsulfonyl chloride in the presence of triethylamine as a base and dichloromethane

^{*} Corresponding author. Tel.: +86-21-64163300x3435; fax: +86-21-64163300; e-mail: zhaoguo@mail.sioc.ac.cn



Scheme 1. Preparation of racemic 2'-diphenylphosphinoyl-1,1'binaphthalenyl-2-ol.

as solvent at room temperature, formed the racemic ester **5** in high yield (Scheme 2).



Scheme 2. Resolution of 2'-diphenylphosphinoyl-1,1'-binaphthalenyl-2-ol.

After extensive screening of the common solvents, we found that the diastereoisomers of **5** could be separated by chromatography on silica gel with diethyl ether and diisopropyl ether (5:1) as an eluent. Thus, **5a** and **5b** were obtained in decent yields with good diastereomeric purity (de: $\ge 95\%$). Both **5a** and **5b** were easily hydrolyzed to (*R*)- or (*S*)-2'-diphenylphosphinoyl-1,1'-binaphthalenyl-2-ol, respectively, in quantitative yield

with the same conditions reported.² Therefore, chiral MOPs oxides could easily be derived from the common starting materials **5a** and **5b**, and the MOPs oxides obtained thus conveniently upgraded to >99% ee by simple recrystallization. The absolute configuration of the two diasteroisomers can simply be determined by converting one of them into an identified ligand, 2-diphenylphosphinoyl-2'-methoxyl-1,1'-binaphthalenes according to literature.² For example, **5b** was converted into 2-diphenylphosphinoyl-2'-methoxyl-1,1'-binaphthalenes with 99% yield upon hydrolysis and methyl protection. The stereochemistry was determined to be (*R*)- for the naphthalene moiety after comparing the sign of specific rotation and spectroscopic data with a reported sample.²

We also made 2'-diphenylphosphanyl-1,1'-binaphthalenyl-2-ol and investigated the direct esterification reaction of it with (1S)-(+)-10-camphorsulfonyl chloride in the presence of NaH in THF. It was found that no esterification occurred and that only the oxidized product **4** could be detected apart from the starting material. This result is the same as the literature.¹⁰

However, when the same strategy was applied to resolve 2'-di(*p*-tolyl)phosphinoyl-1,1'-binaphthalenyl-2-ol, no esterification product could be detected and the starting material was recovered. The reaction also failed to give product when NaH was used as base and DMF as solvent.

However, when 2'-di(p-tolyl)phosphanyl-1,1'-binaphthalenyl-2-ol 6, which was prepared by reaction of chlorodi(*p*-tolyl)phosphine with dilithium salt 3, and (1S)-(+)-10-camphorsulfonyl chloride were employed as the starting materials (Scheme 3), the reaction proceeded in THF with NaH as the base to give the ester 7 in 50% yield, accompanied with the oxidized starting material, 2'-di(p-tolyl)phosphinoyl-1,1'-binaphthalenyl-2-ol and a little of unreacted 6. It should be mentioned that 7 was not the direct esterification product; it was the oxidized product of the direct esterification product. No direct esterification product was isolated even if less than 1 mole equivalent of (1S)-(+)-10-camphorsulfonyl chloride was used, and the yield could not be improved upon, when more (1S)-(+)-10-camphorsulfonyl chloride was employed.



Scheme 3. Resolution of 2'-di(p-tolyl)phosphanyl-1,1'-binaphthalenyl-2-ol.

Ester 7 was also separated into 7a and 7b by chromatography on silica gel with Et_2O as eluent. Both 7a and 7b serve as key intermediates in monophosphine synthesis.¹¹

In summary, we have provided a novel method for preparing chiral MOPs. This method involves the simple resolution of hydroxy monophosphines or their oxides bearing a naphthalene moiety. The enantiomerically pure form of **4** and **6** can allow the rapid construction of a series of chiral monophosphine ligands.

Acknowledgements

We thank the National Natural Science Foundation of China and the Chinese Academy of Sciences for financial support.

References and Notes

- Kurz, L.; Lee, G.; Morgans, D.; Waldyke, J. M. J.; Ward, D. Tetrahedron Lett. 1990, 31, 6321–6324.
- Uozumi, Y.; Tanahashi, A.; Lee, S.-Y.; Hayashi, T. J. Org. Chem. 1993, 58, 1945–1948.
- (a) Uozumi, Y.; Hayashi, T. J. Am. Chem. Soc. 1991, 113, 9887–9888; (b) Hayashi, T.; Uozumi, Y. Pure Appl. Chem. 1992, 60, 1911–1916; (c) Uozumi, Y.; Lee, S.-Y.; Hayashi, T. Tetrahedron Lett. 1992, 33, 7185–7188; (d) Uozumi, Y.; Hayashi, T. Tetrahedron Lett. 1993, 34, 2335–2338; (e) Uozumi, Y.; Kitayama, K.; Hayashi, T. Tetrahedron: Asymmetry 1993, 4, 2419–2422.
- 4. Matsumoto, Y.; Naito, M.; Uozumi, Y.; Hayashi, T. J. Chem. Soc., Chem. Commun. 1993, 1468–1469.
- Hayashi, T.; Iwamura, H.; Naito, M.; Matsumoto, Y.; Uozumi, Y.; Miki, M.; Yanagi, K. J. Am. Chem. Soc. 1994, 116, 775–776.
- (a) Kodama, H.; Taiji, T.; Ohta, T.; Furukawa, I. *Tetrahedron: Asymmetry* 2000, 11, 4009–4015; (b) Uozumi, Y.; Suzuki, N.; Ogiwara, A.; Hayashi, T. *Tetrahedron* 1994, 50, 4293–4302.
- (a) Sakai, N.; Mano, S.; Nozaki, K.; Takaya, H. J. Am. Chem. Soc. 1993, 115, 7033–7034; (b) Nozaki, K.; Li, W.; Horiuchi, T.; Takaya, H. J. Org. Chem. 1996, 61, 7658– 7659; (c) Horiachi, T.; Ohata, T.; Shirakawa, E.; Nozaki, K.; Takaya, H. J. Org. Chem. 1997, 62, 4285–4292; (d) Nozaki, K.; Sakai, N.; Nanno, T.; Higashijima, T.; Mano, S.; Horiachi, T.; Takaya, H. J. Am. Chem. Soc. 1997, 119, 4413–4423.
- (a) Gladiali, S.; Dore, A.; Fabbri, D. *Tetrahedron: Asymmetry* **1994**, *5*, 1143–1146; (b) Gladiali, S.; Medici, S.; Pirri, G.; Pulacchini, S.; Fabbri, D. *Can. J. Chem.* **2001**, *79*, 670–678.
- 9. Hattori, I.; Sakamoto, J.; Hayshizaka, N.; Miyano, S. *Synthesis* **1994**, 199–202.
- Kadyrov, R.; Heinicke, J.; Kinderman, M. K.; Heller, D.; Fischer, C.; Selke, R.; Fischer, A. K.; Jones, P. G. *Chem. Ber.* **1997**, *130*, 1663–1670.
- 11. Analytical data for **5a**, **5b**, **7a**, **7b**. **5a**: ¹H NMR (δ , ppm; $CDCl_3$): 8.18 (dd, 1H, J = 8.4, 2.1 Hz), 7.92 (d, 1H, J = 8.1 Hz, 7.83 (dd, 1H, J = 11.4, 8.4 Hz), 7.77–7.72 (m, 2H), 7.56–7.11 (m, 16H), 6.96 (d, 1H, J = 8.7 Hz), 2.86 (d, 1H, J = 15.0 Hz), 2.46 (d, 1H, J = 15.0 Hz), 2.28–2.19 (m, 1H), 2.04–1.94 (m, 2H), 1.24 (d, 2H, J = 9.0 Hz), 0.80 (s, 3H), 0.57 (s, 3H); ¹³C NMR (δ , ppm; CDCl₃): camphor: 213.2, 57.6, 48.9, 47.4, 42.6, 42.1, 26.6, 24.9, 19.4, 19.3, aryl: 146.0–119.9; ³¹P NMR (δ, ppm; CDCl₃): 29.4; IR (KBr; wavenumber, cm⁻¹): 3056 (m), 2959 (s), 1746 (vs), 1591 (m), 1509 (m), 1437 (s), 1371 (vs), 1198 (vs), 1168 (vs), 1116 (s), 964 (s), 941 (s), 819 (vs), 749 (s), 699 (vs), 539 (s), 524 (s); MS (ESI): m/z 685.2 [M+1], 686.2 [M+2], 707.1 [M+1]Na, 708.1 [M+2]Na; HRMS m/z calcd for C₄₂H₃₈O₅SP [M+1]⁺: 685.2174, found: 685.2172; De: 98%. **5b**: ¹H NMR (δ , ppm; CDCl₃): 8.16 (dd, 1H, J = 8.4, 2.1 Hz), 7.92 (d, 1H, J = 8.1 Hz), 7.84 (dd, 1H, J = 11.4, 8.4 Hz), 7.77-7.72 (m, 2H), 7.56-7.13 (m, 16H), 6.96 (d, 1H, J = 8.4 Hz), 3.1 (d, 1H, J = 15.0 Hz), 2.25–2.16 (m, 1H), 2.08 (d, 1H, J = 15.0 Hz), 1.97–1.88 (m, 2H), 1.82– 1.75 (m, 2H), 1.37-1.20 (m, 2H), 0.68 (s, 3H), 0.41 (s, 3H); ¹³C NMR (δ, ppm; CDCl₃): camphor: 213.4, 57.4, 48.8, 47.4, 42.4, 42.1, 26.6, 24.3, 19.2, 19.0, aryl: 145.7-120.2; ³¹P NMR (δ, ppm; CDCl₃): 29.4; IR (KBr; wavenumber, cm⁻¹): 3055 (m), 2959 (s), 1747 (vs), 1591 (m), 1509 (m), 1437 (s), 1362 (vs), 1199 (s), 1168 (vs), 1116 (s), 965 (s), 941 (s), 817 (vs), 749 (s), 699 (vs), 539 (s), 524 (s); MS (ESI): *m*/*z* 685.2 [M+1], 686.2 [M+2], 707.1 [M+1]Na, 708.1 [M+2]Na; HRMS m/z calcd for $C_{42}H_{38}O_5$ SP $[M+1]^+$: 685.2174, found: 685.2172; De: 95%. 7a: ¹H NMR (δ, ppm; CDCl₃): 8.02 (dd, 1H, J = 9.0, 2.1 Hz), 7.94–7.87 (m, 2H), 7.74 (dd, 2H, J = 11.1, 8.7 Hz), 7.52 (dd, 2H, J = 6.9, 8.7 Hz), 7.38-7.25 (m, 7H), 7.17-7.11 (m, 2H), 7.01-6.94 (m, 2H), 6.89 (dd, 2H, J = 7.8, 2.4 Hz), 2.82 (d, 1H, J = 15.0 Hz), 2.4 (d, 1H, J = 15.0 Hz); 2.87 (s, 3H), 2.25 (s, 3.5H), 2.22–2.19 (m, 0.5H), 2.04–1.76 (m, 4H), 1.26– 1.23 (m, 2H), 0.80 (s, 3H), 0.56 (s, 3H); 13 C NMR (δ , ppm; CDCl₃): 213.2, 57.6, 48.9, 47.4, 42.6, 42.2, 26.6, 25.0, 21.41, 21.40, 21.37, 21.36, 19.5, and 19.3, aryl: 146.0-120.0; ³¹P NMR (δ, ppm; CDCl₃): 29.8. IR (KBr; wavenumber, cm⁻¹): 2958 (m), 1748 (vs), 1601 (m), 1369 (vs), 1216 (s), 1197 (s), 1168 (vs), 1114 (s), 965 (s), 941 (s), 810 (vs), 749 (s), 681 (s), 659 (s), 526 (vs); MS (ESI): m/z 713.2 [M+1]; 714.2 [M+2]; 735.2 [M+1]Na. HRMS calcd for C₄₄H₄₁O₅SPNa [M+1]Na⁺: 735.2341, found: 735.2304. De: 97%. **7b**: ¹H NMR (δ , ppm; CDCl₃): 8.00 (dd, 1H, J = 8.7, 2.1 Hz), 7.95–7.90 (m, 2H), 7.73 (dd, 2H, J = 11.1, 8.4 Hz, 7.52–7.45 (m, 2H), 7.38–7.24 (m, 6H), 7.15–7.12 (m, 2H), 7.03 (dd, 2H, J = 8.1, 2.4 Hz), 6.96 (d, 1H, J = 13.8 Hz), 6.90 (dd, 2H, J = 7.8, 2.4 Hz), 3.08 (d, 1H, J = 15.0 Hz), 2.3 (s, 3H), 2.26 (s, 3.5H), 2.19–2.18 (m, 0.5H), 2.06 (d, 1H, J = 15.0 Hz), 2.00–1.76 (m, 4H), 1.34– 1.25 (m, 2H), 0.67 (s, 3H), 0.40 (s, 3H); ¹³C NMR (δ, ppm; CDCl₃): Camphor: 213.3, 57.4, 48.8, 47.4, 42.36, 42.1, 26.6, 24.3, 21.43, 21.41, 21.38, 21.36, 19.2, 19.0, aryl: 141.5–120.3; ³¹P NMR (δ, ppm; CDCl₃): 29.8; IR (KBr; wave number, cm⁻¹): 2958 (m), 1748 (vs), 1601 (m), 1362 (vs), 1216 (s), 1197 (s), 1167 (vs), 1114 (s), 965 (s), 941 (s), 811 (vs), 774 (s), 749 (s), 681 (s), 660 (s), 526 (vs); MS (ESI) m/z 713.2 [M+1], 714.2 [M+2], 715.2 [M+3], 735.2 [M+1]Na; HRMS m/z calcd for $C_{44}H_{41}O_5SPNa$ [M+1]Na⁺: 735.2328, found: 735.2304; De: 99%.