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# Multicomponent synthesis and anticancer activity studies of novel 6-(Trifluoromethyl)-1, 2, 3, 4-tetrahydropyrimidine-5-carboxylate derivatives

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## ABSTRACT

A series of novel 6-(trifluoromethyl)-1,2,3,4-tetrahydropyrimidine-5-carboxylate derivatives were obtained in good yields from ethyl 4,4,4-trifluoro-3-oxobutanoate, urea, and aryl aldehyde via Biginelli multicomponent reaction. All the corresponding products **4a–4p** were examined against four human cancer cell lines (A549, HepG2, COLO205 and DU145) and compounds **4e**, **4i**, and **4m** which showed promising anticancer activity have been identified.

## GRAPHICAL ABSTRACT



## ARTICLE HISTORY

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## KEYWORDS

6-(trifluoromethyl)-1,2,3,4-tetrahydropyrimidine-5-carboxylate; anticancer activity; Biginelli condensationIntroduction

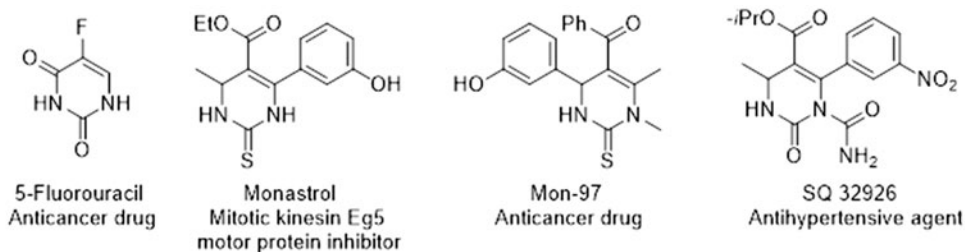
Multicomponent reactions (MCRs)<sup>[1]</sup> are proved effective protocols to build complex molecules in a single step. Dihydropyrimidinones (DHPMs) are key building blocks in N-containing heterocyclic compounds and play an important role in synthetic and medicinal chemistry wherein exhibit wide range of applications such as antiviral,<sup>[2]</sup> anticancer,<sup>[3]</sup> antibacterial and antifungal,<sup>[4]</sup> and calcium channel modulation.<sup>[5]</sup> 5-Fluoro Uracil which has pyrimidine dione shows anticancer activity. Monastrol(I), with the pyrimidine-2-thione motif, specifically inhibits the motility of the mitotic kinesin Eg5,<sup>[6]</sup> meanwhile SQ 32926 (II),<sup>[7]</sup> and Mon-97 (III)<sup>[8]</sup> display antihypertensive and anticancer activities (Fig. 1).

Biginelli<sup>[9–14]</sup> condensation is a simple, well-known, atom-economical, and direct method for the construction of DHPMs by the three-component reaction of aldehyde, 1,3-dicarbonyl compound, and urea under acidic conditions. The available

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**Figure 1.** Biological activity compounds based on Dihydropyrimidinones (DHPMs).

methodologies for the Biginelli reaction in the literature employs catalysts such as  $\text{Fe}_3\text{O}_4$  SBA-15 (mesoporous nanocatalyst),<sup>[15]</sup>  $[\text{Gmim}]\text{Cl}-\text{Cu}(\text{II})$  complex,<sup>[16]</sup>  $\text{Mg}/\text{Ca}-\text{Al}-\text{CO}_3$  hydrotalcite,<sup>[17]</sup> phosphoric acid,<sup>[18]</sup> poly(1-vinyl-3-(3-sulfopropyl) imidazolium hydrogen sulfate) (poly(SIL)),<sup>[19]</sup> chiral organocatalyst,<sup>[20]</sup> ionic liquid ( $[\text{Hmim}]\text{HSO}_4/\text{TMSCl}$ ),<sup>[21]</sup> perchloric acid doped silica ( $\text{HClO}_4/\text{SiO}_2$ ),<sup>[22]</sup> bioglycerol-based sulfonic acid functionalized carbon catalyst,<sup>[23]</sup> and  $t\text{-BuOK}$ ,<sup>[24]</sup> *etc.*<sup>[25,26]</sup> Considering that some of these methods suffer from low yields and high cost of the catalysts, the development of new reaction conditions capable of figuring out a solution to the formidable problems remains highly desirable. Fluorine or trifluoro methyl group at appropriate position of an organic molecule dramatically alters the properties in terms of lipophilicity, lipid solubility, oxidative thermal stability, permeability, oral bio availability thereby enhancement of transport mechanism. In this work, we established a highly effective hydrofluoric acid promoted Biginelli reaction for the preparation of structurally diverse pyrimidinone derivatives bearing trifluoromethyl groups. The anticancer activities of pyrimidinone derivatives were examined against four cancer cell lines such as A549, HepG2, COLO205, DU145 wherein compounds **4e**, **4i**, and **4m** showed promising results.

## Results and discussion

The initial investigation commenced with aryl aldehyde, trifluoromethyl 1,3-dicarbonyl compound, and urea using commercially available aqueous HF (40%) as solvent medium to prepare the desired 6-(trifluoromethyl)-1,2,3,4-tetrahydropyrimidine-5-carboxylate derivatives **4a–4p** (Scheme 1). A mixture of aryl aldehyde, urea, and trifluoromethyl 1,3-dicarbonyl compound in aq HF (40%) were charged in polypropylene bottle and stirred at ambient temperature to afford the desired products in high yields. Different electron withdrawing or donating groups on the aryl aldehyde were examined and the outcome showed satisfactory functional group tolerance (Table 1).

## Cytotoxicity assay

The cytotoxicity of the compounds was determined on the basis of measurement of in vitro growth inhibition of tumor cell lines in 96-well plates by cell-mediated reduction of tetrazolium salt to water insoluble formazan crystals using 5-Fluorouracil as a standard. The cytotoxicity of compounds **4a–4p** was assessed against a panel of four different human tumor cell lines: A549 derived from human alveolar adenocarcinoma epithelial cells (ATCC No. CCL-185), HepG2 derived from human hepatic carcinoma cells

**Table 1.** Preparation of 6-(trifluoromethyl)-1,2,3,4-tetrahydropyrimidine-5-carboxylate derivatives **4a-4p**.

$  \begin{array}{c}  \text{RCHO} + \text{F}_3\text{C}-\text{C}(=\text{O})-\text{CH}_2-\text{C}(=\text{O})-\text{OEt} + \text{H}_2\text{N}-\text{C}(=\text{O})-\text{NH}_2 \xrightarrow[\text{RT, 20-60 min}]{40\% \text{ aq. HF}} \text{Product} \\  \text{1} \qquad \qquad \qquad \text{2} \qquad \qquad \qquad \text{3} \qquad \qquad \qquad \text{4a-4p}  \end{array}  $				
No.	Product	R	Time(min)	Yield (%)
1	<b>4a</b>	C <sub>6</sub> H <sub>5</sub>	30	90
2	<b>4b</b>	4-CH <sub>3</sub> C <sub>6</sub> H <sub>4</sub>	40	87
3	<b>4c</b>	4-ClC <sub>6</sub> H <sub>4</sub>	20	91
4	<b>4d</b>	4-BrC <sub>6</sub> H <sub>4</sub>	25	92
5	<b>4e</b>	4-OCH <sub>3</sub> C <sub>6</sub> H <sub>4</sub>	20	86
6	<b>4f</b>	2-CH <sub>3</sub> C <sub>6</sub> H <sub>4</sub>	20	86
7	<b>4g</b>	4-FC <sub>6</sub> H <sub>4</sub>	25	89
8	<b>4h</b>	4-NO <sub>2</sub> C <sub>6</sub> H <sub>4</sub>	30	85
9	<b>4i</b>	3-CF <sub>3</sub> C <sub>6</sub> H <sub>4</sub>	20	82
10	<b>4j</b>	4-CNC <sub>6</sub> H <sub>4</sub>	20	82
11	<b>4k</b>	2-Furyl	40	85
12	<b>4l</b>	2-Thienyl	30	80
13	<b>4m</b>	2,4-diFC <sub>6</sub> H <sub>3</sub>	40	88
14	<b>4n</b>	3-NO <sub>2</sub> C <sub>6</sub> H <sub>4</sub>	35	79
15	<b>4o</b>	Styryl	40	81
16	<b>4p</b>	Naphthalene	60	80

**Table 2.** Anticancer activities of compounds **4a-4p**.

Test Compounds	IC <sub>50</sub> values in (μg/mL)			
	A549	HepG2	COLO205	DU145
<b>4a</b>	23.2 ± 0.23	31.2 ± 0.25	16.6 ± 0.12	17.4 ± 0.28
<b>4b</b>	15.2 ± 0.11	12.8 ± 0.15	19.2 ± 0.24	12.0 ± 0.12
<b>4c</b>	16.2 ± 0.12	—	18.9 ± 0.33	25.6 ± 0.31
<b>4d</b>	25.8 ± 0.22	12.4 ± 0.33	31.1 ± 0.21	—
<b>4e</b>	9.6 ± 0.15	4.8 ± 0.12	5.1 ± 0.14	4.2 ± 0.19
<b>4f</b>	7.2 ± 0.14	11.4 ± 0.10	10.9 ± 0.20	13.8 ± 0.12
<b>4g</b>	12.9 ± 0.11	17.2 ± 0.27	21.2 ± 0.20	16.5 ± 0.10
<b>4h</b>	11.8 ± 0.19	10.8 ± 0.25	12.4 ± 0.19	9.9 ± 0.29
<b>4i</b>	2.2 ± 0.12	3.1 ± 0.20	2.7 ± 0.22	2.1 ± 0.12
<b>4j</b>	19.5 ± 0.25	17.5 ± 0.10	—	21.5 ± 0.12
<b>4k</b>	32.2 ± 0.22	27.1 ± 0.15	15.7 ± 0.25	22.8 ± 0.12
<b>4l</b>	18.2 ± 0.12	17.1 ± 0.20	18.7 ± 0.28	25.3 ± 0.22
<b>4m</b>	6.1 ± 0.12	4.4 ± 0.24	5.7 ± 0.16	7.8 ± 0.22
<b>4n</b>	11.3 ± 0.13	14.1 ± 0.27	15.8 ± 0.26	17.3 ± 0.32
<b>4o</b>	—	—	—	116.8 ± 0.38
<b>4p</b>	18.4 ± 0.13	17.7 ± 0.30	28.6 ± 0.18	—
<b>5-Fluoro uracil</b> (Standard control)	1.8 ± 0.09	1.9 ± 0.11	1.7 ± 0.08	1.7 ± 0.07

—indicates IC<sub>50</sub> value >116.8 μg/mL; A549: human alveolar adenocarcinoma epithelial cells (ATCC No. CCL-185); HepG2: derived from human hepatic carcinoma cells (ATCC No. HB-8065); COLO205: derived from human colon cancer cells (ATCC No. CCL-222) and DU145: derived from human prostate cancer cells (ATCC No. HTB-81).

(ATCC No. HB-8065), COLO205 derived from human colon cancer cells (ATCC No. CCL-222), and DU145 derived from human prostate cancer cells (ATCC No. HTB-81) using the MTT assay.<sup>[27]</sup> The IC<sub>50</sub> values (50% inhibitory concentration) were calculated from the plotted absorbance data for the dose-response curves. IC<sub>50</sub> values (in μM) are expressed as the average of three independent experiments. Among all the compounds

examined, compounds **4e**, **4i**, and **4m** showed promising activity ( $IC_{50}$  concentration  $<7.8 \mu\text{g/mL}$  on all cancer cell lines). The presence of  $\text{CF}_3$  group at appropriate position of 6-(trifluoromethyl)-1,2,3,4-tetrahydropyrimidine-5-carboxylates, can enhance the properties of lipid solubility, oxidative thermal stability and thereby leading to an enhancement of transport mechanism. Meanwhile compounds **4b**, **4f**, **4g**, **4h** and **4n** showed moderate activity with  $IC_{50}$  concentration  $<21.2 \mu\text{g/mL}$  on all cancer cell lines as shown in Table 2.

## Conclusion

In conclusion, we have demonstrated a simple and efficient method for the preparation of structurally diverse 6-(trifluoromethyl)-1,2,3,4-tetrahydropyrimidine-5-carboxylate derivatives under aq. HF (40%) medium. The anticancer activities of all the pyrimidine derivatives were examined with promising bioactivities.

## Experimental section

### General procedure

A mixture of aryl aldehyde (0.1 mol), urea (0.1 mol) and trifluoromethyl 1,3-dicarbonyl compound (0.1 mol) in aq HF (40%) were charged in polypropylene bottle and stirred at ambient temperature for a period of 60 min. The reaction progress was monitored by thin-layer chromatography (TLC). After completion of the reaction, the reaction mixture was poured into crushed ice, and the resulted solid was filtered. The solid was washed with water and then sodium bicarbonate solution and dried to afford the product. Please see supporting information for further details (yield and data characterization).

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