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Graphical Abstract

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Synthesis of chiral cyclic β -Amino ketones by Ru-catalyzed asymmetric hydrogenation

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ARTICLE INFO

ABSTRACT

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Keywords: Asymmetric hydrogenation Ruthenium Chiral phosphine ligands Amino ketones Synthesis of chiral cyclic β -Amino ketones has been first reported via Ru-catalyzed asymmetric hydrogenation. High enantioselectivities were achieved by using (*S*)-C₃-TunePhos chiral ligand (up to 94% ee).

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Chiral amines and their derivatives are important synthetic targets and powerful pharmacophores for defining new pharmaceutical drugs.¹ Over the past few decades, enantioselective hydrogenation has proven to be an efficient method for chiral amine synthesis.² Our group has recently developed a Rh-catalyzed enantioselective hydrogenation of β -keto enamides as an efficient way to prepare optically pure β -amino ketones and their derivatives (Scheme 1).³ We later envisioned that chiral cyclic β -Amino ketones are also interesting structural motifs. For example, these amino ketones can serve as key synthetic precursors for some antitumor agents (Scheme 2).^{4,5} To the best of our knowledge, there is no report on synthesis of optically pure cyclic β -amino ketones through catalytic asymmetric hydrogenation.

Previous work



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Herein, we present a Ru-catalyzed highly enantioselective hydrogenation of cyclic β -keto enamides to afford chiral cyclic β -amino ketones.



Scheme 2. Examples of commercial available antitumor agents.

Preparation of cyclic β -keto enamide substrates were tried though the direct condensation of commercial available cyclic 1,3diketones with acetamide.⁶ However, all attempts afforded low yields (< 30%). Thus, a two-step procedure was applied. Cyclic 1,3 -diketones were first converted to the corresponding cyclic β enaminones,⁷ followed by acylation with acyl chloride to give the enamide substrates in moderate to good yields (up to 87%, scheme 3). We have to mention that aromatic cyclic 1,3diketones such as 1,3-indanediones didn't give any desired products following this procedure, which dramatically narrowed the substrate scope.

Tetrahedron



Scheme 3. Preparation of cyclic β -keto enamide substrates.

We began our study by investigating the asymmetric hydrogenation of **1a** as the model substrate. Rh/TangPhos and Rh/DuanPhos catalysts were initially tested since they have been previously proved efficient for enantioselective hydrogenation of acyclic β -ketoenamides.³ However, neither of them gave satisfactory enantioselectivities under various reaction conditions (Table 1, entries 1-2). Other commercial available Rh complexes such as Rh/DuPhos and Rh/BINAPINE were also tested to give lower than 60% ee's (Table 1, entries 3-4). After catalyst screening, to our surprise, we found that commercial available Ru(OAc)₂/Binap catalyst gave the highest 92% ee. Later on, our results showed that Ru(OAc)₂/C₃-Tunephos give better enantioselectivities as well as good conversions (94% ee, Table 1, entries 6-7).



Picture 1. Selected structure of screened phosphine ligands.

Table 1.	Selected	results	from	initial	screening	of ca	talysts	for
hydroger	nation of	1a ^a			-			



6	C ₃ -TunePhos- Ru(OAc) ₂	53	94	
7 ^d	C ₃ -TunePhos- Ru(OAc) ₂	86	94	

^a Unless otherwise mentioned, reaction conditions: Catalyst/substrate =1:100, at room temperature, under 1 atm of hydrogen for 24 h.

^b Conversions were determined by ¹H NMR of the crude product.

^c Determined by GC on a chiral phase.

^d Under 5 atm of hydrogen.

Thus, different Ru precursors were systematically investigated. ⁸ As showed in Table 2, $Ru(OAc)_2/C_3$ -TunePhos complex still gave the highest enantioselectivities (Table 2, entries 1-4). Further solvent screening indicated that MeOH served as the best solvent (Table 2, entries 5-11). Increasing the hydrogen pressure or catalyst loading would cause more formation of the corresponding amino-alcohol **3a** as side product.

Table 2. Selected results from initial screening of Rucatalysts for hydrogenation of $1a^a$



Entry	{Ru}	Solvent	Yield of	ee of
			2a (%) ^b	2a (%) ^c
1	$[NH_2Me_2][{RuCl(L)}_2$	MeOH	22	42
	$(\mu$ -Cl) ₃]			
2	[RuCl(p-pymene)L]Cl	MeOH	36	51
3	[RuCl(benzene)L]Cl	MeOH	31	55
4	RuCl ₂ L (DMF) _n	MeOH	<5	n.d.
5	Ru(OAc) ₂ L	THF	31	82
6	$Ru(OAc)_2L$	1,4-dioxane	34	83
7	$Ru(OAc)_2L$	toluene	29	42
8	$Ru(OAc)_2L$	ethyl acetate	47	75
9	Ru(OAc) ₂ L	EtOH	69	91
10	$Ru(OAc)_2L$	ⁱ PrOH	63	85
11	Ru(OAc) ₂ L	МеОН	73	94

^a Unless otherwise mentioned, reaction conditions: Catalyst/substrate =1:100, at room temperature, under 5 atm of hydrogen for 24 h. L= (S)-C₃-TunePhos. Ru complexes were prepared according to reported procedure.
 ^b Conversions were determined by ¹H NMR of the crude product.
 ^c Determined by GC on a chiral phase.

Table 3. Substrate scope and limitations^a

Entry	Substrate	Product	Yield ^b	ee ^c
1	0 NHAc	o * NHAc 2a	73	94
2	O NHAC 1b	O NHAc 2b	68	88
3	0 NHAc	O * NHAc 2c	36	67
4	O NHAc 1d	NHAc 2d	45	61

^a Unless otherwise mentioned, reaction conditions: Ru(OAc)₂/C₃-TunePhos as catalyst, Catalyst/substrate =1:100, Methanol as solvent, at room temperature, under 5 atm of hydrogen for 24 h.

^b Conversions were determined by ¹H NMR of the crude product.

^c Determined by GC on a chiral phase.

With the optimized reaction conditions, 5,5-dimethyl substituted substrate **1b** was tested to give moderate hydrogenation results (Table 3, entry 2). However, hydrogenation of tetra-substituted olefin substrates **1c** showed dramatic loss of reactivities and enantioselectivities (Table 3, entry 3). Low reactivity and enantioselectivity were also observed when cyclopentyl substrate **1d** was tested (Table 3, entry 4).

In conclusion, we reported a Ru-catalyzed asymmetric hydrogenation of cyclic β -keto enamides to afford chiral cyclic β -amino ketones, which afforded a new approach for the synthesis of optically pure cyclic β -amino ketones. Further studies focusing on expanding the substrate scope and improving the reactivities and enantioselectivities are ongoing.

Experimental section

General Procedure for Asymmetric Hydrogenation

In a glovebox filled with nitrogen, Ru catalyst (0.02 mmol) was dissolved in MeOH (5 mL). 0.5 mL of this solution (0.002 mmol) was added into a solution of 0.2 mmmol substrate in 3 mL of degassed MeOH. The resulting solution was then transferred into an autoclave and charged with 5 atm of hydrogen. The hydrogenation was performed at room temperature for 24 h. After carefully releasing the hydrogen, the reaction mixture was passed through a short silica gel column to remove the catalyst. The ee values of all compounds were determined by GC on a chiral stationary phase.

Acknowledgments

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Supplementary data

Supplementary data associated with this article can be found, in the online version

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Support information

Synthesis of chiral cyclic β -amino ketones by Ru-catalyzed asymmetric hydrogenation (

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E-mail: xumu@rci.rutgers.edu
1. General Information
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2. Synthesis of substrate
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S19

General Methods. Starting materials, reagents and solvents were purchased from commercial sources and were used as received. All reactions and manipulations that were

NC'

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sensitive to moisture or air were performed in a nitrogen-filled glovebox or using standard Schlenk techniques, unless otherwise noted. Solvents were dried with standard procedures and degassed with N₂. Column chromatography was performed using 200-400 mesh silica gel supplied by Natland International Corp. Thin-layer chromatography (TLC) was performed on EM reagents 0.25 mm silica 60-F plates. ¹H and ¹³C NMR spectra were recorded on Bruker Avance 400 MHz spectrometers and a Varian VNMRS-500 MHz instrument. GC analysis was carried out on Hewlett-Packard 7890 gas chromatography using chiral capillary columns.

General Procedure for substrate synthesis

10 mmol cyclic 1,3-diketone was added to a mixture of 10 mmol ammonium acetate in dry toluene (20 mL). The mixture was heated for 3 h under reflux using a Dean–Stark water separator. The resulting oily product was separated and recrystallized with ethyl acetate to give the corresponding enaminone as crystals. The enaminone product was dissolved in 50ml THF and 2 equiv. of pyridine were added. To the mixed solution, acetyl chloride (2 equiv.) in 5ml THF was added dropwise at 0 °C. The reaction mixture was warmed up to ambient temperature and stirred additional 5 h at 60°C. The solvent was removed under vacuum and the crude product purified by flash chromatography.

✓ NHAC

N-(3-oxocyclohex-1-en-1-yl)acetamide (1a) ¹H NMR (400 MHz, CDCl3) δ 7.61 (1H, br), 6.49 (1H, s), 2.51 (2H, t, *J* = 6.0), 2.30 (2H, t, *J* = 6.3), 2.07 (3H, s), 1.96 (2H, q, *J* = 6.4); ¹³C NMR (100 MHz, CDCl3) δ 21.4, 24.7, 28.2, 36.6, 111.2, 158.9, 170.1, 201.1

N-(**5**,**5**-dimethyl-3-oxocyclohex-1-en-1-yl)acetamide (**1b**) ¹H NMR (400 MHz, CDCl3) δ 8.67 (1H, br), 6.74 (1H, s), 2.51 (2H, t, *J* = 6.0), 2.37 (2H, s), 2.19 (2H, s), 2.11 (3H, s), 1.05 (6H, s); ¹³C NMR (100 MHz, CDCl3) δ 24.7, 28.1, 32.7, 42.1, 50.5, 110.1, 154.7, 170.1, 201.0

NHAC

N-(2-methyl-3-oxocyclohex-1-en-1-yl)acetamide (1c) ¹H NMR (400 MHz, d₆-DMSO) δ 9.24 (1H, br), 2.77 (2H, t, *J* = 6.0), 2.33 (2H, t, *J* = 6.2), 2.10 (3H, s), 1.88 (2H, q, *J* = 6.4), 1.67 (3H, s); ¹³C NMR (100 MHz, CDCl3) δ 7.95, 23.72, 23.75, 26.9, 36.04, 117.1, 152.0, 167.9, 197.8



N-(2-methyl-3-oxocyclopent-1-en-1-yl)acetamide (1d) ¹H NMR (400 MHz, CDCl₃) δ 8.41 (1H, br), 3.13 (2H, t, *J* = 2.4), 2.34 (2H, t, *J* = 2.5), 2.16 (3H, s), 1.59 (3H, s); ¹³C NMR (100 MHz, CDCl₃) δ 6.69, 24.5, 27.7, 33.6, 118.3, 165.1, 168.9, 206.7

NHAc H NMR (400 MHz, CDCl3)































General Procedure for Asymmetric Hydrogenation

In a glovebox filled with nitrogen, Ru catalyst (0.02 mmol) was dissolved in MeOH (5 mL). 0.5 mL of this solution (0.002 mmol) was added into a solution of 0.2 mmmol substrate in 3 mL of degassed MeOH. The resulting solution was then transferred into an autoclave and charged with 5 atm of hydrogen. The hydrogenation was performed at room temperature for 24 h. After carefully releasing the hydrogen, the reaction mixture was passed through a short silica gel column to remove the catalyst. The ee values of all compounds were determined by GC on Supelco beta Dex^{TM} 390 column.



Supelco beta Dex^{TM} 390 column (30 m ×0.25 mm × 0.25 µm), He 1.0 mL/min, column temperature: 180 °C; t_R = 13.4 min (minor), t_R = 13.7 min (major). (2a)



