

Copper-Catalyzed P–H Insertions of α -Imino Carbenes for the Preparation of 3-Phosphinoylindoles

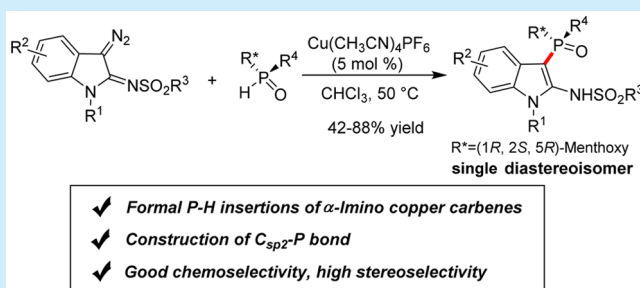
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Supporting Information

ABSTRACT: A highly efficient P–H insertion of α -imino copper carbene into H-phosphine oxides, leading to 3-phosphinoylindoles with a broad substrate diversity and good chemoselectivity, is established. This methodology provides a rapid and efficient approach for the C(sp²)–P bond formation via the P–H insertion strategy. The stereoselective P–H insertion of α -imino copper carbene represents a unique example of asymmetric P–H insertion, affording P-stereogenic 3-phosphinoylindoles with high stereoselectivities.



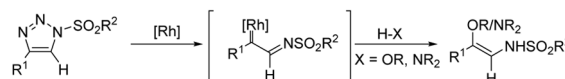
The highly efficient construction of carbon–heteroatom bonds is one of the most pursued goals in modern organic synthesis. Transition-metal-catalyzed insertion of carbenes and carbenoids generated in situ from α -diazo carbonyl compounds into heteroatom–hydrogen bonds (X–H, where X = O, N, S, P, etc.) is an efficient strategy for the construction of C(sp³)–X bonds.¹ Recently, α -imino rhodium carbene, generated in situ from *N*-sulfonyl 1,2,3-triazoles, has emerged as an increasingly valuable intermediate possessing novel reactivity and synthetic potential.² Significant progress has been made by the research groups of Fokin,³ Gevorgyan,⁴ Davies,⁵ Sarpong,⁶ Shi,⁷ and others. In particular, α -imino rhodium carbenes insertions have led to a novel synthetic method toward the formation of C–X bonds, including C(sp²)–X bonds. For example, high regio- and stereoselectivities have been achieved with the insertion of α -imino rhodium carbene into O–H and N–H bonds, affording enamide products (Scheme 1a).^{3a}

Indole and its derivatives are important bioactive heterocycles in natural products and pharmaceuticals.⁸ More recently, indole-embedded α -imino metal carbene in situ generated from 3-diazoindolin-2-imine has been identified as a reactive intermediate in carbene insertions to furnish a variety of indole derivatives.⁹ The transformation of rhodium-catalyzed N–H insertion was reported to build 3-substituted indole derivatives with the C(sp²)–N bond formation (Scheme 1b).^{9c} Despite progress in α -imino carbene insertion transformations, the insertion reactions of indole-embedded α -imino metal carbenes for the generation of C(sp²)–X bonds is much less developed.

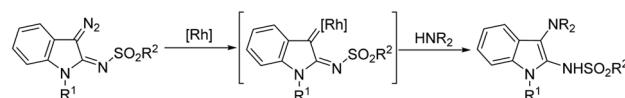
Aryl and heteroaryl phosphorus compounds have been of particular interest due to their biological significance and synthetic value.¹⁰ 3-Phosphoindoles represent novel second-generation NNRTIs (non-nucleoside reverse transcriptase inhibitors) which show remarkable effect on inhibiting the replication of HIV-1 virus (IDX899).¹¹ In consideration of the

Scheme 1. Insertion Reactions of α -Imino Metal Carbenes for the Construction of C(sp²)–Heteroatom Bonds

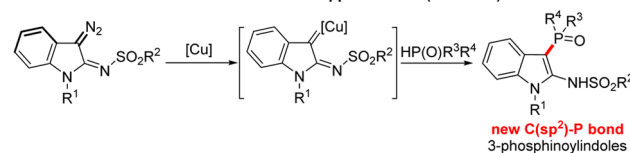
a. O–H and N–H insertions of α -imino rhodium carbene



b. N–H insertion of indole-embedded α -imino rhodium carbene



c. P–H insertion of indole-embedded α -imino copper carbene (this work)

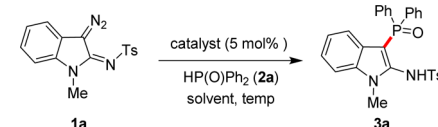


importance of 3-phosphoindoles in pharmaceuticals and inspired by these previous findings shown in Scheme 1, herein we describe our efforts toward the development of the copper-catalyzed P–H insertion of novel α -imino copper carbene to access a variety of 3-phosphoindole derivatives (Scheme 1c). This method will not only expand the repertoire of transformations of α -imino metal carbenes but also provide a straightforward route to C(sp²)–P bond formation and creation of new phosphoindoles. To the best of our knowledge, the P–H insertion reactions of α -imino metal carbene for the construction of C–P bond has never been reported.

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We began to explore the P–H insertion of 3-diazoindolin-2-imine (**1a**) and diphenylphosphine oxide (**2a**) with the common catalysts used for the decomposition of triazoles and diazo compounds, including $\text{Rh}_2(\text{oct})_4$, $[\text{Ru}(p\text{-cymene})\text{Cl}_2]_2$, $(\eta^3\text{-C}_3\text{H}_5)_2\text{Pd}_2\text{Cl}_2$, and $\text{Ni}(\text{cod})_2$ (Table 1, entries 1–4).

Table 1. Optimization of the Reaction Conditions^a



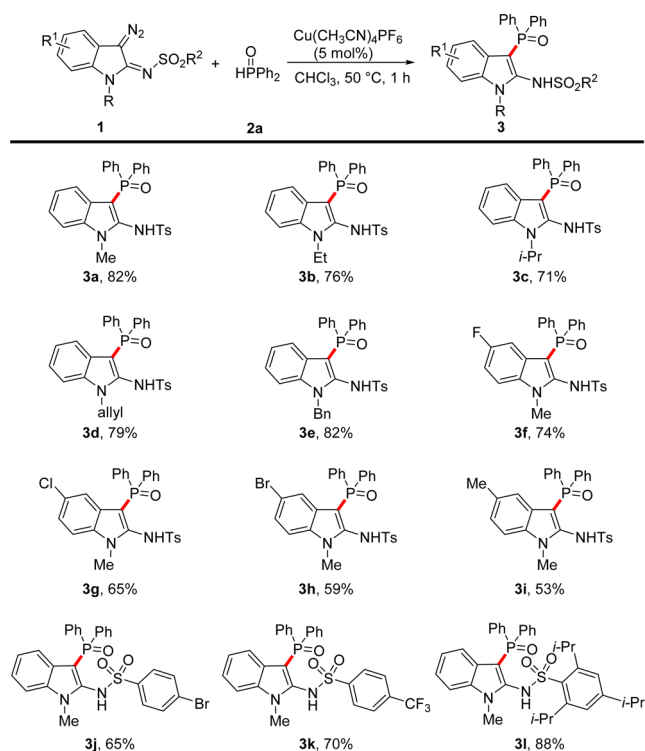
entry	catalyst	solvent	temp (°C)	time (h)	yield ^b (%)
1	$\text{Rh}_2(\text{oct})_4$	DCE	35	24	0
2	$[\text{Ru}(p\text{-cymene})\text{Cl}_2]_2$	DCE	35	24	0
3	$(\eta^3\text{-C}_3\text{H}_5)_2\text{Pd}_2\text{Cl}_2$	DCE	35	24	0
4	$\text{Ni}(\text{cod})_2$	DCE	35	24	0
5	$\text{Cu}(\text{OTf})_2$	DCE	35	24	20
6	$\text{Cu}(\text{OAc})_2$	DCE	35	24	15
7	$\text{Cu}(\text{acac})_2$	DCE	35	24	15
8	Cu_2O	DCE	35	24	23
9	$\text{Cu}(\text{CH}_3\text{CN})_4\text{PF}_6$	DCE	35	24	42
10	$\text{Cu}(\text{CH}_3\text{CN})_4\text{PF}_6$	DCE	35	1	42
11	$\text{Cu}(\text{CH}_3\text{CN})_4\text{PF}_6$	DCE	50	1	56
12	$\text{Cu}(\text{CH}_3\text{CN})_4\text{PF}_6$	DCE	15	1	trace
13		DCE	50	1	0
14	$\text{Cu}(\text{CH}_3\text{CN})_4\text{PF}_6$	DCM	50	1	26
15	$\text{Cu}(\text{CH}_3\text{CN})_4\text{PF}_6$	CHCl_3	50	1	82
16	$\text{Cu}(\text{CH}_3\text{CN})_4\text{PF}_6$	1, 4-dioxane	50	1	64
17	$\text{Cu}(\text{CH}_3\text{CN})_4\text{PF}_6$	toluene	50	1	52

^aReaction condition: **1a** (0.2 mmol), **2a** (0.2 mmol), catalyst (0.01 mmol), solvent (0.8 mL). ^bIsolated yield.

Unfortunately, no desired product but a significant amount of starting material was observed. To our delight, the reaction gave **3a** in 20% yield in the presence of 5 mol % of $\text{Cu}(\text{OTf})_2$ in DCE at 35 °C (Table 1, entry 5). Encouraged by this initial result, we then screened other copper catalysts, of which $\text{Cu}(\text{CH}_3\text{CN})_4\text{PF}_6$ proved superior (Table 1, entries 5–9). By screening the reaction temperature and reaction time, the optimal reaction temperature and reaction time were found to be 50 °C and 1 h, respectively. A trace amount of the desired **3a** was exactly detected when the reaction was carried out at 15 °C (entry 12). A control reaction in the absence of metal catalyst afforded no desired product, indicating the critical role of the copper catalyst in generating the α -imino metal carbene intermediate (entry 11 vs 13). The reaction was subjected to notable solvent effects. Chloroform (CHCl_3) was a better solvent over $\text{ClCH}_2\text{CH}_2\text{Cl}$ (DCE), CH_2Cl_2 (DCM), 1, 4-dioxane, and toluene, providing **3a** in 82% yield (entries 15–17).

With the optimal reaction conditions, we then explored the substrate scope with respect to the 3-diazoindolin-2-imines. As illustrated in Scheme 2, the alkyl group on the 1-position of indole could be varied from methyl, ethyl, isopropyl, allyl, and benzyl to produce **3a**, **3b**, **3c**, **3d**, and **3e** in 82%, 76%, 71%, 79%, and 58% yields, respectively. The structure of **3e** was unambiguously determined by single-crystal X-ray analysis (as shown in the Supporting Information). No insertion product was observed by using N-unprotected 3-diazoindolin-2-imines. Various 3-diazoindolin-2-imines with substituents on the 5-

Scheme 2. Scope of 3-Diazoindolin-2-imines^{a,b}

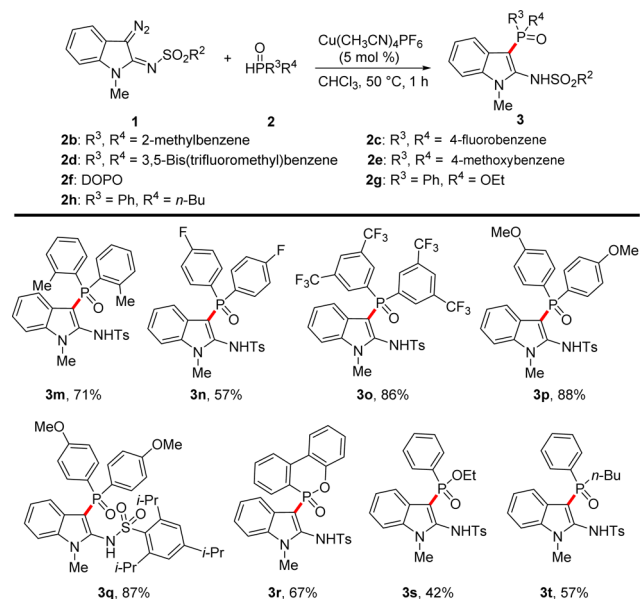


^aReaction conditions: $\text{Cu}(\text{CH}_3\text{CN})_4\text{PF}_6$ (0.01 mmol), **1** (0.20 mmol) and $\text{HP}(\text{O})\text{Ph}_2$ (0.20 mmol) in CHCl_3 (0.8 mL) at 50 °C for 1 h. ^bIsolated yield.

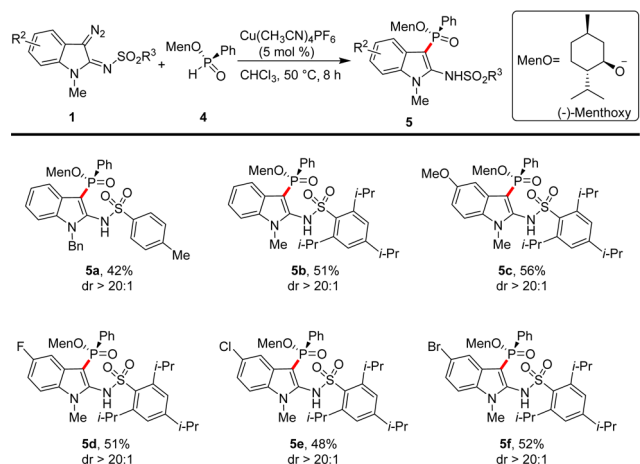
position of indole were feasible substrates in this reaction and delivered **3f–i** in 53–74% yields. The process was tolerant to 3-diazoindolin-2-imines bearing different sulfonyl groups ($\text{R}^2 = 4\text{-BrC}_6\text{H}_4$, $4\text{-CF}_3\text{C}_6\text{H}_4$, $2,4,6\text{-i-Pr}_3\text{C}_6\text{H}_2$), thus giving the corresponding products **3j–l** with high chemoselectivities (65–88% yield).

We next investigated the scope with respect to the H-phosphine oxides (Scheme 3). H-Phosphine oxides **2** under the optimal reaction conditions and reacted with **1** generally proceeded in good yields. Both electron-poor and electron-rich substituted groups in H-phosphine oxides (**2d,e**) could be tolerated and afforded **3o** and **3p** in 86% and 88% yield, respectively. Notably, heterocyclic H-phosphine oxides such as 9,10-dihydro-9-oxa-10-phosphaphenanthrene 10-oxide (DOPO, **2f**) could be introduced to the insertion product. Moreover, ethyl phenylphosphinate (**2g**) was also viable for this transformation, giving the expected product **3s** in 42% yield. In the case where butyl(phenyl)phosphine oxide (**2h**) was used as a substrate, the desired product was isolated in 57% yield. The tosyl group can be removed with concentrated sulfuric acid to afford stable 2-free amino indole **6** in 91% yield (Supporting Information).

Inspired by the success in the P–H insertions of α -imino copper carbene, we then turned our attention to the asymmetric synthesis in an attempt to obtain a chiral 3-phosphindoles. (1*R*,2*S*,5*R*)-(–)-Menthoxypheylphosphinate (**4**) was employed as the chiral reagent for the chirality induction (Scheme 4). The P–H insertion of α -imino copper carbene proceeded smoothly to give the chiral 3-phosphindoles (**5a–f**) in moderate yield with very high diastereoselectivity. The absolute configuration of **5f** was established by X-ray single-crystal analysis (Figure 1).

Scheme 3. Scope of H-Phosphine Oxides^{a,b}

^aReaction conditions: $\text{Cu}(\text{CH}_3\text{CN})_4\text{PF}_6$ (0.01 mmol), **1a** (0.20 mmol) and **2** (0.20 mmol) in CHCl_3 (0.8 mL) at 50 °C for 1 h.
^bIsolated yield.

Scheme 4. Stereoselective Synthesis of Chiral 3-Methoxyarylphosphindoles^{a,b}

^aReaction conditions: $\text{Cu}(\text{CH}_3\text{CN})_4\text{PF}_6$ (0.01 mmol), **1a** (0.20 mmol), and $\text{HP}(\text{O})(\text{OMe})\text{Ph}$ (0.20 mmol) in CHCl_3 (0.8 mL) at 50 °C for 8 h. ^bIsolated yield and the diastereomeric ratio were detected by ^1H NMR (600 MHz) spectroscopy.

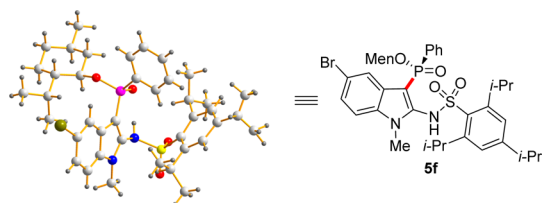


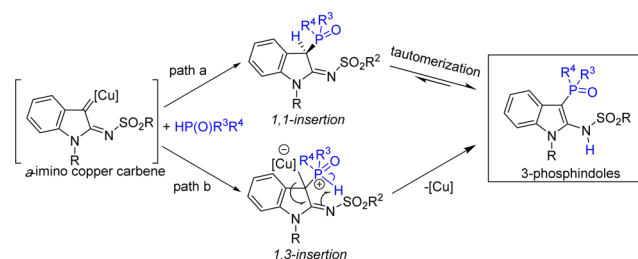
Figure 1. X-ray structure of **5f**.

P-Stereogenic phosphorus compounds have attracted considerable attention due to their wide applications as chiral ligands for asymmetric synthesis.¹² The stereoselective synthesis of *P*-stereogenic compounds is one of the most

important and challenging subjects in synthetic organic chemistry.^{13,14} In this asymmetric *P*–H insertion, the *H*–*P* bond in (1*R*,2*S*,5*R*)-(–)-menthoxyphenylphosphinate (**4**) was converted into a $\text{C}(\text{sp}^2)$ –*P* bond to give chiral 3-phosphindoles with high stereoselectivities. Although there are several examples of *P*–H insertions reported in the formation of the $\text{C}(\text{sp}^3)$ –*P* bond,¹⁵ no asymmetric *P*–H insertion reaction has been achieved to date. The stereoselective *P*–H insertion of α -imino copper carbene represents a unique example of asymmetric *P*–H insertion, affording chiral 3-phosphindoles with a *P*-stereogenic center.

On the basis of the preliminary results of control reactions (as shown in Table 1, entry 11 vs 13) and previous work,^{3a,16} we envisioned that the insertion of α -imino copper carbene into *P*–H bonds may proceed in two plausible pathways (Scheme 5).

Scheme 5. Proposed Mechanism



diazindolin-2-imines may undergo direct 1,1-insertion into the *P*–H bond, followed by tautomerization leading to product of 3-phosphindoles (path a). Alternatively, the insertion may proceed by the formation of zwitterionic *P*-ylide intermediate. Subsequent dissociation of copper results in a proton transfer from the phosphorus to imine moiety, finally furnishing the 1,3-insertion product of 3-phosphindoles (path b).

In summary, we have developed *P*–H insertions of α -imino copper carbenes for the construction of $\text{C}(\text{sp}^2)$ –*P* bonds. The protocol provides an efficient approach to 3-phosphindoles from 3-diazoindolin-2-imines and *H*-phosphine oxides with good chemoselectivity. In addition, the stereoselective *P*–H insertion was also successfully developed with high diastereoselectivity. The investigations on the biological activities of the new 3-phosphindoles are in progress.

■ ASSOCIATED CONTENT

§ Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.orglett.6b03752.

Experimental procedures, characterization data, and NMR spectra of all new compounds (PDF)
 X-ray crystallography of **3e** (CIF)
 X-ray crystallography of **5f** (CIF)

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Notes

The authors declare no competing financial interest.

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