COMMUNICATION

Novel insertion of ethynylbenzene derivatives bearing an electron-withdrawing substituent at the 4-position into a P–C bond and their transannular insertion between a metal atom and an *ipso*-carbon atom of the phosphine ligand †

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Received 21st June 2000, Accepted 28th July 2000 Published on the Web 14th August 2000

Reactions of Cp\*MCl(P–O) (M = Rh, Ir) bearing P–O coordination with ethynylbenzene derivatives having an electron-withdrawing group in the presence of a  $PF_6$  anion result in either insertion into a P–C bond or a transannular addition between a metal and an *ipso*-carbon of the phosphine.

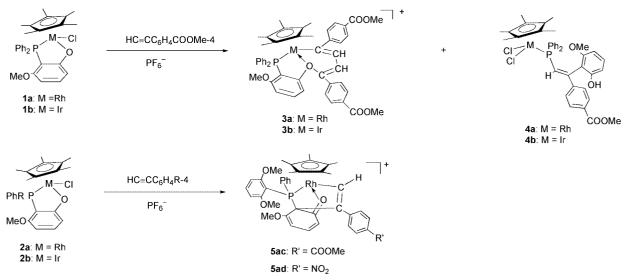
When transition metal halide complexes are allowed to react with alkynes in the presence of anions such as  $PF_6^-$ ,  $BF_4^-$ , etc., it is well-known that vinylidene or carbene complexes may be formed, depending on the solvent.<sup>1</sup> However, we have recently reported that the unprecedented insertion of alkyne into a metal-oxygen  $\sigma$ -bond can occur without the formation of vinylidene complexes.<sup>2</sup> On treatment of the rhodium complex  $[Cp*RhCl(MDMPP-\kappa^2 P, O)] \quad 1a \quad (MDMPP-\kappa^2 P, O = \kappa O-1-O$ κP-2-PPh<sub>2</sub>-3-MeOC<sub>6</sub>H<sub>3</sub>) with mono- and di-substituted alkynes (HC=CPh, HC=CCOOR or ROOCC=CCOOR; R = Me, Et) in the presence of the  $PF_6^-$  anion, single and double insertion of alkynes into the Rh-O bond occurred to form various types of complex, depending on the alkyne, as follows; (1) a complex bearing five- and six-membered rings arising from a double insertion, (2) a seven-membered metallacycle bearing a CO ligand arising from a single insertion and an extraction of CO from an ester and (3) a neutral complex bearing a sevenmembered ring arising from a single insertion.<sup>2</sup>

When ethynylbenzene derivatives bearing an electronwithdrawing substituent at the 4-position of the phenyl group

† Electronic supplementary information (ESI) available: spectroscopic data for complexes 3–5. See http://www.rsc.org/suppdata/dt/b0/ b004984k/ were treated with [Cp\*MCl(MDMPP- $\kappa^2 P, O$ )] (1a: M = Rh;<sup>3</sup> 1b: M = Ir<sup>4</sup>) or [Cp\*RhCl(BDMPP- $\kappa^2 P, O$ )] 2a<sup>5</sup> (BDMPP- $\kappa^2 P, O$ ] 2a<sup>5</sup> (BDMPP- $\kappa^2 P, O$ ]  $\kappa^2 P, O = \kappa O$ -1-O- $\kappa P$ -2-PRPh-3-MeOC<sub>6</sub>H<sub>3</sub>) bearing a P, O bidentate ligand derived from (2,6-dimethoxyphenyl)diphenyl-phosphine (MDMPP) and bis(2,6-dimethoxyphenyl)diphenyl-phosphine (BDMPP) in the presence of KPF<sub>6</sub>, we found that novel types of reaction: an insertion into a P–C bond or a transannular addition between a Rh atom and an *ipso*-carbon atom of the phosphine ligand, occurred readily. We report here the reactions with ethynylbenzene derivatives bearing an electron-withdrawing substituent such as the COOMe or NO<sub>2</sub> group.

Reaction of 1a with  $HC = CC_6H_4COOMe-4$  at room temperature in the presence of KPF<sub>6</sub> in acetone-CH<sub>2</sub>Cl<sub>2</sub> gave two compounds by crystallization from CH<sub>2</sub>Cl<sub>2</sub>-diethyl ether: 3a ± and 4a ‡ as confirmed from elemental analyses and FAB mass spectroscopy (Scheme 1). X-Ray analysis revealed that 3a contained a (P,O,C) tridentate ligand resulting from the head-tohead double-insertion of alkyne into the Rh-O bond.§ Two carbon atoms bearing the phenyl group were connected to the Rh and O atoms. The infrared spectrum showed bands due to methoxycarbonyl and PF<sub>6</sub> groups at 1717 and 839 cm<sup>-1</sup>, respectively. The <sup>1</sup>H NMR spectrum in CD<sub>2</sub>Cl<sub>2</sub> showed one doublet at  $\delta$  1.36 due to the Cp\* protons and three singlets at  $\delta$  3.35, 3.88 and 3.95; the former is due to the methoxy group and the others are due to the methoxycarbonyl groups. The <sup>31</sup>P{<sup>1</sup>H} NMR spectrum showed a doublet at  $\delta$  35.84  $(J_{\rm RhP} = 157.8 \text{ Hz}).$ 

The infrared spectrum of 4a showed three characteristic bands at 3285, 1717 and 1605 cm<sup>-1</sup> due to the OH, carbonyl groups and C–C double bond, respectively. In the <sup>1</sup>H NMR spectrum a broad resonance due to a hydroxyl proton, which

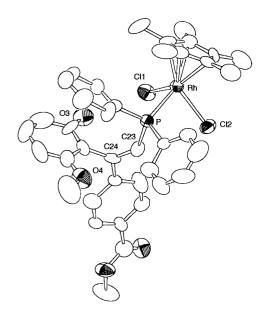


Scheme 1 Reactions of 1 and 2 with  $HC \equiv CC_6H_4R' - 4$  (R = 2,6-(MeO)<sub>2</sub>C<sub>6</sub>H<sub>3</sub>, R' = COOMe or NO<sub>2</sub>).

**2896** J. Chem. Soc., Dalton Trans., 2000, 2896–2897

DOI: 10.1039/b004984k

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**Fig. 1** Molecular structure of **3a**. Selected bond lengths (Å) and angles (°): Rh(1)–P(1) 2.341(5), Rh(1)–Cl(1) 2.428(5), Rh(1)–Cl(2) 2.411(5); P(1)–C(23) 1.83(2), C(23)–C(24) 1.32(2); P(1)–Rh(1)–Cl(1) 87.1(2), P(1)–Rh(1)–Cl(2) 85.0(2), Cl(1)–Rh(1)–Cl(2) 93.4(2), P(1)–C(23)–C(24) 135(1), C(23)–C(24)–C(25) 117(1).

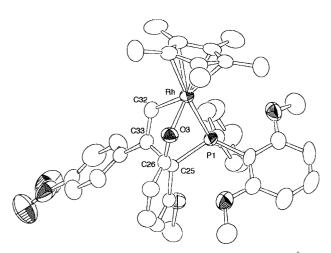


Fig. 2 Molecular structure of 5ad. Selected bond lengths (Å) and angles (°): Rh(1)-P(1) 2.304(2), Rh(1)-O(3) 2.106(4), Rh(1)-C(32) 2.013(6), O(3)-C(26) 1.263(7), C(25)-C(26) 1.505(8), P(1)-C(25) 1.943(6), C(25)-C(33) 1.567(8), C(32)-C(33) 1.340(8); P(1)-Rh(1)-O(3) 81.1(1), P(1)-Rh(1)-C(32) 76.3(2), O(3)-Rh(1)-C(32) 85.2(2), Rh(1)-P(1)-C(25) 90.2(2), P(1)-C(25)-C(26) 105.1(4), C(25)-C(26)-O(3) 118.7(5), Rh(1)-O(3)-C(26) 117.9(4), Rh(1)-C(32)-C(33) 120.4(4), C(32)-C(33)-C(25) 114.2(5), P(1)-C(25)-C(33) 96.4(4).

disappeared on treatment with D<sub>2</sub>O, appeared at  $\delta$  6.74 in addition to three characteristic methyl resonances at  $\delta$  1.41(d), 3.44(s) and 3.83(s), due to 1,2,3,4,5-pentamethylcyclopentadienyl, methoxy and methoxycarbonyl groups, respectively. There were no signs of a PF<sub>6</sub> group in the infrared and <sup>31</sup>P{<sup>1</sup>H} NMR spectra, suggesting that the complex was neutral. The structure was confirmed by X-ray analysis (Fig. 1).¶ As expected, the molecule is neutral and the rhodium atom is surrounded by two chlorine atoms and one phosphorus atom. *Cis*-insertion of alkyne into the P–C bond has occurred, accompanied by cleavage of a Rh–O bond. The Rh–P bond length of 2.341(5) Å is longer than that of **1a** by 0.04 Å, due to the absence of chelation and is 0.02 Å shorter than that (2.366(1) Å) of Cp\*RhCl<sub>2</sub>(MDMPP), probably due to lower steric demand than in the MDMPP ligand.<sup>3</sup> A similar reaction occurred on treatment with 1b in acetone– $CH_2Cl_2$  to give 3b and 4b (Scheme 1).<sup>‡</sup> The fact that a similar reaction in MeOH gave exclusively 3a as the only isolated complex confirmed that the chlorine atom in 4 originated from dichloromethane.

Complex 2a containing the bulky phosphine BDMPP was treated with HC=CC<sub>6</sub>H<sub>4</sub>R-4, (R = COOMe or NO<sub>2</sub>) in acetone–CH<sub>2</sub>Cl<sub>2</sub> in the presence of KPF<sub>6</sub> at room temperature (Scheme 1) to yield complexes 5ac (yellow, R = COOMe) and 5ad (brown, R = NO<sub>2</sub>) as confirmed from elemental analyses and FAB mass spectroscopy.<sup>‡</sup> X-Ray analysis of 5ad revealed that the structure consists of five- and six-membered rings derived from a transannular insertion of alkyne between the Rh atom and the *ipso*-carbon atom of the phosphine ligand, accompanying the subsequent transformation of the Rh–O  $\sigma$ -bond to a Rh–O coordination (Fig. 2).|| The change of this bonding mode caused an elongation of *ca*. 0.06 Å in the Rh–O bond length.

The IR spectrum showed bands due to methoxycarbonyl and ketone groups at 1715 and 1630 cm<sup>-1</sup> for **5ac** and due to a ketone group at 1628 cm<sup>-1</sup> for **5ad**, respectively. The <sup>1</sup>H NMR spectrum showed three methoxy groups at  $\delta$  2.97 (s), 3.06 (bs) and 3.49 (bs) and one methoxycarbonyl group at  $\delta$  3.84 (s) for **5ac** and three methoxy groups at  $\delta$  2.99 (s), 3.08 (bs) and 3.52 (bs) for **5ad**, respectively. A remarkable feature is that the <sup>31</sup>P{<sup>1</sup>H} NMR doublets show large downfield shifts (*ca.*  $\delta$  140).

Further mechanistic studies are now in progress.

## Acknowledgements

We thank Professor Shigetoshi Takahashi and Dr Fumie Takei for measurements of FAB mass spectroscopy. This work was partially supported by a Grant-in-Aid for Scientific Research from the Ministry of Education of Japan.

## Notes and references

‡ Elemental analyses of all new complexes prepared here are in good agreement with the calculated values. Elementary analyses and spectroscopic data are enclosed in the electronic supplementary information.<sup>†</sup> § Crystal data for **3a**: C<sub>49</sub>H<sub>47</sub>O<sub>6</sub>P<sub>2</sub>F<sub>6</sub>Rh, *M* = 1010.8, monoclinic, space group *P*2<sub>1</sub>/*a* (no. 14), *a* = 15.534(3) Å, *b* = 19.986(3) Å, *c* = 15.757(3) Å,  $\beta$  = 103.70(1)°, *V* = 4752(1) Å<sup>3</sup>, *Z* = 4, *D* = 1.413 g cm<sup>-3</sup>,  $\mu$  = 4.96 cm<sup>-1</sup> (MoK*a*), *F*(000) = 2072, *T* = 298 K. Data were collected on a Rigaku AFC5S diffractometer. The structure was solved by Patterson methods, and non-hydrogen atoms were refined anisotropically using full-matrix least-squares based on *F*<sup>2</sup> to give *R*1 = 0.064 for 3298 reflections and *R*<sub>w</sub> = 0.198 for 8328 reflections.

¶ Crystal data for **4a**:  $C_{39}H_{42}O_6PCl_2Rh$ , M = 811.5, orthorhombic, space group  $Pca2_1$  (no. 29), a = 34.835(7) Å, b = 9.135(8) Å, c = 12.305(5) Å, V = 3915(5) Å<sup>3</sup>, Z = 4, D = 1.377 g cm<sup>-3</sup>,  $\mu = 6.55$  cm<sup>-1</sup> (MoKa), F(000) = 1672, T = 300 K. The structure was solved by Patterson methods, and non-hydrogen atoms were refined anisotropically using full-matrix least-squares based on  $F^2$  to give R1 = 0.068 for 2216 reflections and  $R_w = 0.203$  for 3607 reflections.

|| Crystal data for **5ad**: C<sub>39</sub>H<sub>46</sub>NO<sub>9</sub>P<sub>2</sub>F<sub>6</sub>Rh, M = 951.6, monoclinic, space group  $P2_1/a$  (no. 14), a = 15.299(7) Å, b = 14.826(7) Å, c = 19.505(9) Å,  $\beta = 106.78(3)^\circ$ , V = 4235(3) Å<sup>3</sup>, Z = 4, D = 1.492 g cm<sup>-3</sup>,  $\mu = 5.57$  cm<sup>-1</sup> (MoKa), F(000) = 1952, T = 298 K. The structure was solved by Patterson methods, and non-hydrogen atoms were refined anisotropically using full-matrix least-squares based on  $F^2$  to give R1 = 0.065 for 4458 reflections and  $R_w = 0.180$  for 7708 reflections. CCDC reference number 186/2117. See http://www.rsc.org/suppdata/ dt/b0/b004984k/ for crystallographic files in .cif format.

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