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Synthesis of 2-Substituted-Pyrrolidinethiourea Derivatives and Their Antagonist Effect on Vanilloid Receptor

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Abstract—Four pyrrolidine derivatives were prepared by the formation of a 5-membered ring based on capsazepine. Among them, the two carbon extended derivatives, **4a** ($IC_{50}=55 \mu M$) and **4b** ($IC_{50}=3 \mu M$), both showed different levels of antagonist activity against the vanilloid receptor in a ⁴⁵Ca²⁺-influx assay. © 2002 Elsevier Science Ltd. All rights reserved.

Capsaicin (1), the pungent component of chili pepper, activates the vanilloid receptor (VR1) which is a polymodal nociceptor.¹ The activation opens a novel cation selective ion channel in the plasma membrane of peripheral sensory neurons.² Capsaicin induces pain upon topical application in the early stage, which is followed by a period of desensitization.³ Since this agent became a target for the design of a novel class of analgesics, its structure and activity relationships have been studied by modifications of capsaicin.^{4,5} However, the potent synthetic agonists (SDZ-249-482,^{4e} KR-25018^{5a}) could not be developed as orally active analgesics, because of their initial irritancy.⁶ Since it has been impossible to remove the initial excitatory side effect in the agonist, a competitive antagonist has been pursued as a novel pharmacological agent for analgesics, rather than an agonist.

In 1994, the first competitive antagonist, capsazepine (2), was reported by introducing a saturated 7-membered rigid ring system.⁷ The rationale for the major difference between the agonist and the antagonist is in the structural relationship between the catechol or vaniloid aromatic ring (A region) and the amide bond (B region). In the antagonist structure, the A and B

regions are virtually orthogonal, in contrast to the agonist structure, in which they are approximately coplanar.⁷ This conformational difference was supported by an X-ray crystallographic analysis as well as a molecular modeling technique. Recently, tetrahydrobenzazepine and tetrahydroisoquinoline thiourea derivatives have been prepared as antagonists by the replacement of the *p*-chlorophenethyl group with 3-acetoxy-2-benzylpropyl groups.^{5d} As part of our program to find a new scaffold for a competitive antagonist against the capsaicin receptor, we modified capsazepine by introducing a 5-membered pyrrolidine ring (3, 4), which has a similar conformation based on molecular modeling. In this communication, we report the synthesis of 2-substituted-pyrrolidinethiourea derivatives and their biological activities. The molecular modeling studies proposed the pyrrolidine derivatives (3) as target molecules with an orthogonal conformation between the A and B regions. The energy-minimized conformation of 3a was very similar to that of capsazepine, as shown in Figure 1. Both structures showed the *E*,*E*-thiourea- π - π -stacking form (3a: 5.91 kcal/mol; 2: 4.24 kcal/mol) rather than the E,E-thiourea- π - π nonstacking form (3a: 8.86 kcal/mol; 2: 6.68 kcal/mol) because of the stabilizing effect induced by the π - π stacking interaction of the two corresponding benzene moieties.8

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Figure 1. Stereopair modeling representation of the preferred threedimensional conformation. (a) Energy-minimized conformation of 2; (b) energy-minimized conformation of 3a.

Compounds 3a and 3b were prepared in 5 and 6 steps, respectively, from vanillin, as shown in Scheme 1. Reductive amination of vanillin with 3-chloropropylamine, using 10% Pd/C under atmospheric H₂, followed by sequential *N*-butyloxycarbonyl (Boc) protection and *O*-methoxymethyl (MOM) protection, gave 7. The pyrrolidine ring system (8) was constructed by intramolecular cyclization of 7 under basic conditions.⁹ The deprotection of the Boc and MOM groups with trifluoroacetic acid gave 9, which was coupled with phenethylisothiocyanate to provide 3b. Compound 3a also could be prepared by demethylation of 9, using c-HBr and the same coupling reaction as above.

The biological activities of **3a** and **3b** were evaluated as both agonists and antagonists in the ${}^{45}Ca^{2+}$ -influx assay by using the neonantal rat cultured spinal sensory neurons.¹⁰ Unexpectedly, none of derivatives showed both agonist and antagonist activity, in spite of their similar conformations to capsazepine based on the molecular modeling (Table 1). A possible explanation may come from the unfavorable rotation of C(1')–C(2) of **3**, which may affect the binding interaction of the catechol or



Scheme 1. Reaction and conditions: (a) 3-chloropropylamine, 10% Pd/C, H₂, MeOH, rt; Boc₂O, CH₂Cl₂, TEA, rt, 95%, (b) MOMCl, K₂CO₃, DMF, 50 °C, 93%, (c) *n*-BuLi, THF, -78–0 °C, 84%, (d) TFA, CH₂Cl₂, rt, 95%, (e) phenethylisothiocyanate, CH₂Cl₂, rt, 85%, (f) c-HBr, 100 °C, 57%.



Scheme 2. Reaction and conditions: (a) 4'-benzyoxy-3'-methoxybenzyltriphenylphosphoniumbromide, THF, *n*-BuLi, 0 °C, 85%, (b) 10% Pd/C, H₂, CH₃OH, rt, 95%, (c) TFA, CH₂Cl₂, rt, 95%, (d) phenethylisothiocynate, CH₂Cl₂, rt, 82%, (e) c-HBr, 100 °C, 67%.

Table 1. ⁴⁵Ca²⁺-influx activity of the pyrrolidine derivatives



 $^{a}\text{EC}_{50}$ (the concentration of derivatives necessary to produce 50% of the maximal response) and IC_{50} values (the concentration of derivatives necessary to reduce the response to 0.5 μ M capsaicin by 50%) were estimated with at least three replicates at each concentration. Each compound was tested in two independent experiments. Antagonist data were fitted with a sigmodial function.

vanilloid moiety of 3 with the receptor. We extended two carbons between C(1')–C(2) of 3, which could give sufficient flexibility. The preparation of 4a and 4b is shown in Scheme 2. Wittig coupling with *N*-Boc-2-formypyrrolidine 10 and 3'-methoxy-4'-benzyloxybenzyltriphenylphosphoniumbromide, followed by hydrogenation, gave 12. The deprotection of the Boc group with trifluoroacetic acid, followed by coupling with phenethylisothiocyanate, provided 4b. Demethylation of 13 with c-HBr followed by treatment with phenethylisothiocyanate gave 4a. As shown in Table 1, the ⁴⁵Ca²⁺-influx assay revealed that both 4a and 4b show antagonist activity and the vanilloid derivative, 4b (IC₅₀=3 μ M), is more active than the catechol derivative, 4a (IC₅₀=55 μ M). Interestingly, the difference in potency between the vanilloid derivative (4b) and the catechol derivative (4a) is congruent with the reported general tendency in agonists, but opposite to that of capsazepine and its vanilloid derivative.¹¹ These cumulative findings may suggest that the binding site of the vanilloid or catechol moiety in 4 may same as that of the general agonists, such as capsaicin.

In conclusion, four pyrrolidine derivatives¹² were prepared by the formation of a 5-membered ring based on capsazepine. Even though the energy minimized molecular conformations of 3 are very similar to that of capsazepine, they show no biological activities. However, the two carbons extended derivatives, **4a** and **4b**, both show antagonist effects at different levels. This result suggests that the orthogonal conformation may not so critical for the antagonist activity but the variation in the length of the ligand could play more important role for the design of new scaffold as potent antagonist. The studies on the vanilloid type acyclic antagonist are currently being investigated.

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8. The calculations were done using the program SYBYL 6.5 from Tripos Software Inc., St. Louis, MO, USA.

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10. The uptake and accumulation of ${}^{45}Ca^{2+}$ by the pyrrolidine derivatives was studied in neonatal rat cultured spinal sensory neurons by the method described in detailed by the reference, **3b**.

11. See the reference 7 and 5d *N*-(3,4-Dihydroxybenzyl)-*N*'-[(4-chlorophenyl)ethyl]thiourea (EC₅₀, 0.10 μ M), *N*-(4-Hydroxy-3-methoxybenzyl)-*N*'-[(4-chlorophenyl)ethyl]thiourea (EC₅₀, 0.06 μ M); Capsazepine (IC₅₀=0.56 μ M), the vanilloid derivative of Capsazepine (IC₅₀=0.97 μ M).

12. All compounds gave satisfactory spectroscopic data consistent with the proposed structures. Data for 3a: mp 197 °C. ¹H NMR (CDCl₃, 300 MHz), δ 6.98 (m, 8H), 4.60 (m, 1H), 3,99 (m, 2H), 3.85 (m, 2H), 2.72 (m, 2H), 2.40 (m, 1H), 1.87 (m, 3H). IR (KBr) 3384, 2924, 1644, 1618 cm⁻¹. MS (EI) *m/e*, 342 [M⁺]. Anal. calcd for C₁₉H₂₂N₂O₂S: C, 66.64; H, 6.48; N, 8.18. Found: C, 66.59; H, 6.42; N, 8.09. 3b: mp 148 °C. ¹H NMR(CDCl₃, 300 MHz), δ 7.06 (m, 5H), 6.65 (m, 3H), 4.61 (s, 1H), 4.01 (d, 2H, J = 7.14 Hz, 3.81 (m, 2H), 3.82 (s, 3H), 2.72 (m, 2H), 2.38 (m, 1H), 1.90 (m, 3H). IR (KBr) 3390, 2938, 1602, 1516 cm⁻¹. MS (EI) m/e, 356 [M⁺]. Anal. calcd for C₂₀H₂₄N₂O₂S: C, 67.38; H, 6.79; N, 7.86. Found: C, 66.75; H, 6.70; N, 7.65. 4a: mp 158 °C. ¹H NMR (CDCl₃, 300 MHz) δ 7.12 (m, 5H), 6.56 (m, 2H), 6.39 (d, 1H, J=8.04 Hz), 3.67 (m, 3H), 3.58 (m, 1H), 3.21 (t, 1H, J=1.58 Hz), 2.76 (m, 2H), 2.36 (t, 2H, J=7.32 Hz), 1.87 (m, 6H), 1.50 (m, 1H). IR (KBr) 3399, 2920, 1604, 1531, 1455 cm⁻¹. MS (EI) *m/e*, 370 [M⁺]. Anal. calcd for C₂₁H₂₆N₂O₂S: C, 68.08; H, 7.07; N, 7.56. Found: C, 67.89; H, 7.17; N, 7.05. 4b: mp 121 °C. ¹H NMR (CDCl₃, 300 MHz) 7.18 (m, 5H), 6.75 (d, 1H, *J*=8.07 Hz), 6.64 (s, 1H), 6.50 (d, 1H, J = 7.86 Hz), 3.97 (m, 7H), 3.52, (m, 1H), 2.81 (d, 2H, J=3.84 Hz), 2.48 (m, 2H), 1.93 (m, 6H). IR (KBr) 3397, 2920, 1516, 1456, 1384 cm⁻¹. MS (EI) m/e, 384. Anal. calcd for C₂₂H₂₈N₂O₂S: C, 68.72; H, 7.34; N, 7.29. Found: C, 68.48; H, 7.40; N, 7.05.