

# New Applications of Camphor-Derived P,N-Ligands for Asymmetric Pd- and Ir-Catalyzed Reactions

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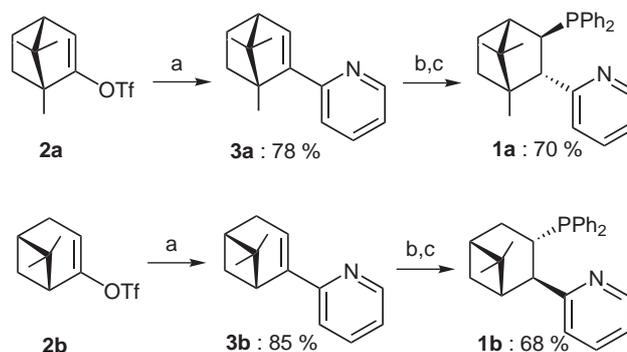
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**Abstract:** Ir-catalyzed asymmetric hydroboration of bicyclic hydrazine **4** in the presence of chiral ligand **1a**, leads to bicyclic alcohol **5** after oxidative workup in good yield (76%) and moderate enantioselectivity (71% ee).

**Key words:** *meso*-bicyclic hydrazine, Ir-catalyzed hydroboration, *t*-BuOK-catalyzed addition, P,N-ligands, asymmetric catalysis

The preparation of new chiral ligands for performing of metal-catalyzed reactions is an important research field.<sup>1</sup> We have recently reported a new synthesis of P,N-ligands **1a,b** which proved to be excellent ligands to perform various iridium-catalyzed hydrogenation reactions.<sup>2</sup> Especially interesting were the hydrogenation of unfunctionalized stilbenes, which proceeded with up to 96% ee and the first asymmetric iridium-catalyzed asymmetric hydrogenation of dehydroamino acids with up to 96.5% ee.<sup>2</sup> Herein, we wish to report another application of ligands **1a,b** for the asymmetric hydroboration of *meso*-bicyclic hydrazines<sup>3</sup> as well as the use of ligand **1a,b** for asymmetric Pd-catalyzed allylic substitution.<sup>4</sup> Both ligands **1a,b** were readily available from the corresponding triflates **2a,b** which were obtained from D-(+)-camphor and (*R*)-(+)-nopinone.<sup>5</sup> The Pd-catalyzed Negishi cross-coupling<sup>6</sup> of **2a,b** with 2-pyridylzinc bromide furnished the unsaturated pyridines **3a,b** in 78–85% yield. The addition of diphenylphosphine oxide in NMP or DMSO to **3a,b** proceeded readily at 70 °C in the presence of *t*-BuOK (20 mol%).<sup>7</sup>

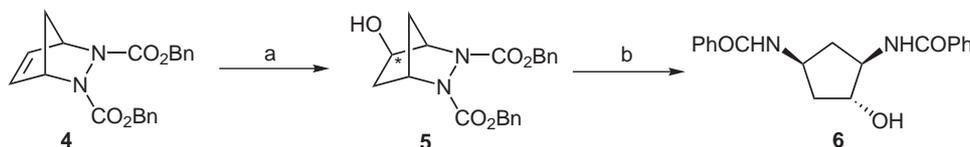
After the reduction of the intermediate phosphine oxides with HSiCl<sub>3</sub> and Et<sub>3</sub>N in toluene (120 °C, 12 h),<sup>8</sup> the desired ligands **1a,b** were obtained in 70% and 68% yields,



**Scheme 1** Reagents and conditions: a) Pd(dba)<sub>2</sub> (2 mol%), dppf (2 mol%), 2-pyridylzinc bromide, reflux, 12 h; b) *t*-BuOK (20 mol%), DMSO, diphenylphosphine oxide, 70 °C, 16 h; c) HSiCl<sub>3</sub> (10 equiv), Et<sub>3</sub>N (20 equiv), toluene, 120 °C, 16 h.

respectively (Scheme 1). We have first investigated the asymmetric hydroboration<sup>9</sup> of the *meso*-bicyclic hydrazine **4** with catecholborane (CatBH) in the presence of [Ir(cod)Cl]<sub>2</sub>. After H<sub>2</sub>O<sub>2</sub> oxidation the alcohol **5** was obtained with variable enantioselectivities (Table 1). Bicyclic alcohol **5** can be readily converted to interesting diaminocyclopentanols of type **6** by a reductive cleavage of the N–N bond and subsequent benzoylation (Scheme 2).<sup>3b</sup>

We found that **1a** and **1b** are effective ligands for the hydroboration of *meso*-hydrazine **4** providing high conversion in short reaction times (4–6 h). Ligand **1a** furnished alcohol **5** with a higher enantioselectivity (58% ee) whereas **1b** provided alcohol **5** with 44% ee at 25 °C (Table 1, entries 1 and 2). Performing the hydroboration at 0 °C led



**Scheme 2** Reagents and conditions: a) i) [Ir(cod)Cl]<sub>2</sub> (1 mol%), ligands **1a** or **1b** (2.1 mol%), catecholborane (2 equiv), THF, 0 °C, ii) 30% H<sub>2</sub>O<sub>2</sub>, NaOH, EtOH; b) i) H<sub>2</sub>, AcOH, Pt (cat), ii) PhCOCl, pyridine.

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**Table 1** Enantioselective Ir-Catalyzed Hydroboration of the *meso*-Hydrazine **4** in the Presence of Ligands **1a,b**

Entry	L*	Molarity	Solvent	Temp (°C)	Yield (%) <sup>a</sup>	%ee <sup>b,c</sup>
1	<b>1a</b>	0.25	THF	25	57	58
2	<b>1b</b>	0.25	THF	25	67	44
3	<b>1a</b>	0.25	THF	25	61 (54) <sup>c</sup>	71 (52) <sup>d</sup>
4	<b>1a</b>	0.25	THF	-20	–	–
5	<b>1a</b>	0.25	toluene	25	30	65
6	<b>1a</b>	0.25	DME	0	63	67
7	<b>1a</b>	0.1	THF	0	56	62
8	<b>1a</b>	0.6	THF	0	76	71
9	( <i>R,S</i> )-Josiphos	0.6	THF	0	61	64 <sup>e</sup>

<sup>a</sup> Yield of analytically pure products.

<sup>b</sup> The enantioselectivity was determined by chiral HPLC (Chiracel AD: *i*-PrOH:*n*-hexane, 20:80, flow: 0.8 mL/min).

<sup>c</sup> The absolute configuration of the major enantiomer with **1a** has been established to be (1*S*,4*R*,5*R*).

<sup>d</sup> Results in parentheses were obtained using 1 equivalent of CatBH.

<sup>e</sup> The absolute configuration of the major enantiomer has been established to be (1*S*,4*R*,5*R*).<sup>3a</sup>

to a further improvement of the enantioselectivity providing alcohol **5** with 71% ee and 61% yield (entry 3). On lowering the reaction temperature further (-20 °C), no significant reaction occurred (entry 4). Using toluene as solvent led to a lower conversion (30% yield) and 65% ee (entry 5) whereas change to DME provided alcohol **5** in 63% yield and 67% ee (entry 6). In order to improve the reaction yield we increased the concentration (compare entries 3, 7 and 8) and obtained our best result with a 0.6 M solution of bicycle **4** (76% yield, 71% ee, entry 8). Lowering the amount of CatBH from 2 equivalents to 1 equivalent in order to avoid an eventual uncatalyzed racemic hydroboration did not succeed and resulted in 54% yield and 52% ee (entry 3). Compared to previous studies using (*R,S*)-Josiphos,<sup>3a</sup> the use of the new ligand **1a** represents an improvement of yield and a slight improvement in enantioselectivity (from 68% ee to 71% ee).

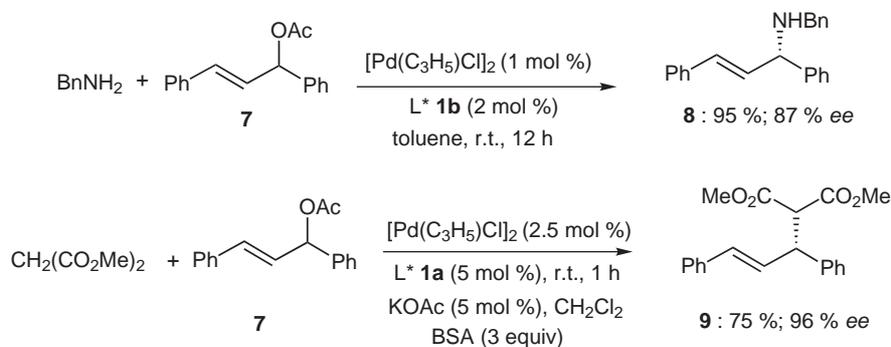
Next, we have used ligands **1a,b** to perform palladium-catalyzed allylation reactions. We examined the Pd(0)-catalyzed amination of 1,3-diphenylallyl acetate **7** with benzylamine as well as the reaction of **7** with dimethyl

malonate in the presence of [Pd(C<sub>3</sub>H<sub>5</sub>)Cl]<sub>2</sub> (1–2.5 mol%) and the new P,N-ligands **1a,b** under standard conditions. We found that both ligands gave good enantioselectivities. Thus, the reaction with benzylamine under standard conditions provided the desired allylic amine **8** in 95% yield and 87% ee using the ligand **1b** (2 mol%). Similarly, substitution with dimethylmalonate provided the expected product **9** in the presence of ligand **1a** (5 mol%) in 75% yield and 96% ee (Scheme 3).

In summary, we have reported the use of the new modular ligands **1a,b** for Ir-catalyzed asymmetric hydroboration and Pd(0)-catalyzed allylation. Further applications of this new family of ligands are currently underway in our laboratories.<sup>11</sup>

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**Scheme 3**

## References

- (1) (a) Noyori, R. *Asymmetric Catalysis in Organic Synthesis*; Wiley: New York, **1994**. (b) Ohkuma, T.; Kitamura, M.; Noyori, R. In *Catalytic Asymmetric Synthesis*; Ojima, I., Ed.; Wiley-VCH: Weinheim, **2000**, 1. (c) Brown, J. M. In *Comprehensive Asymmetric Catalysis*; Jacobsen, E. N.; Pfaltz, A.; Yamamoto, H., Eds.; Springer: Berlin, **1999**, 121. (d) Togni, A.; Halterman, R. L. *Metallocenes*, Vol. 2; Wiley-VCH: Weinheim, **1998**, 685. (e) Beller, M.; Eckert, M. *Angew. Chem. Int. Ed.* **2000**, *39*, 1010; *Angew. Chem.* **2000**, *112*, 1027.
- (2) Bunlaksanusorn, T.; Polborn, K.; Knochel, P. *Angew. Chem. Int. Ed.* **2003**, *42*, *Angew. Chem.* 3941; **2003**, *115*, 4071.
- (3) (a) Pérez Luna, A.; Bonin, M.; Micouin, L.; Husson, H.-P. *J. Am. Chem. Soc.* **2002**, *124*, 12098. (b) Pérez Luna, A.; Ceschi, M.-A.; Bonin, M.; Micouin, L.; Husson, H.-P.; Gougeon, S.; Estenne-Bouhtou, G.; Marabout, B.; Sevrin, M.; George, P. *J. Org. Chem.* **2002**, *67*, 3522.
- (4) (a) Helmchen, G.; Pfaltz, P. *Acc. Chem. Res.* **2000**, *33*, 336. (b) Togni, A.; Bieler, N.; Burckhardt, C.; Köllner, C.; Pioda, G.; Schneider, R.; Schnyder, A. *Pure Appl. Chem.* **1999**, *71*, 1531. (c) Loiseleur, O.; Hayashi, M.; Keenan, M.; Schmees, N.; Pfaltz, A. *J. Organomet. Chem.* **1999**, *576*, 16. (d) Roesky, P. W. *Heteroatom Chem.* **2002**, *13*, 514. (e) Von Matt, P.; Loiseleur, O.; Koch, G.; Pfaltz, A.; Lefebvre, C.; Feucht, T.; Helmchen, G. *Tetrahedron: Asymmetry* **1994**, *5*, 573. (f) Dawson, G. J.; Williams, J. M. J.; Coote, S. J. *Tetrahedron: Asymmetry* **1995**, *6*, 2535. (g) Hou, D.-R.; Reibenspies, J.; Colacot, T. J.; Burgess, K. *Chem.-Eur. J.* **2001**, *7*, 5391. (h) Lightfoot, A.; Schnider, P.; Pfaltz, A. *Angew. Chem.* **1998**, *110*, 3047; *Angew. Chem. Int. Ed.* **1998**, *37*, 2897.
- (5) McMurry, J. E.; Scott, W. J. *Tetrahedron Lett.* **1983**, *24*, 979.
- (6) (a) Negishi, E.-I. *Acc. Chem. Res.* **1982**, *15*, 340. (b) Negishi, E.-I. In *Metal-Catalyzed Cross Coupling Reactions*; Diederich, F.; Stang, P. J., Eds.; Wiley-VCH: Weinheim, **1998**, Chap. 1. (c) Erdik, E. *Tetrahedron* **1992**, *48*, 9577.
- (7) (a) Rodriguez, A. L.; Bunlaksanusorn, T.; Knochel, P. *Org. Lett.* **2000**, *2*, 3285. (b) Bunlaksanusorn, T.; Rodriguez, A. L.; Knochel, P. *Chem. Commun.* **2001**, 745. (c) Bunlaksanusorn, T.; Knochel, P. *Tetrahedron Lett.* **2002**, *43*, 5817.
- (8) Uozumi, Y.; Hayashi, T. *J. Am. Chem. Soc.* **1991**, *113*, 9887.
- (9) Männig, D.; Nöth, H. *Angew. Chem. Int. Ed.* **1985**, *24*, 878; *Angew. Chem.* **1985**, *97*, 854.
- (10) Schnyder, A.; Togni, A.; Wiesli, U. *Organometallics* **1997**, *16*, 255.
- (11) [Ir(COD)Cl]<sub>2</sub> (3.4 mg, 5 mol, 1 mol%), ligand **1a** (4.2 mg, 11 mol, 2.1 mol%) and **4** (182 mg, 0.5 mmol) were placed under Ar in a flame-dried Schlenk tube. THF (0.85 mL) was degassed at -50 °C and added to the mixture at this temperature. The reaction was stirred at r.t. for 30 min and cooled to 0 °C. Catecholborane (0.11 mL, 1 mmol) was added at 0 °C and stirred for 4 h. EtOH (0.5 mL), 3 M NaOH (0.85 mL) and 30% H<sub>2</sub>O<sub>2</sub> (0.5 mL) were added and stirred at 25 °C for 16 h. The reaction mixture was extracted with EtOAc (3 × 10 mL). The organic phase was washed with 1 M NaOH (5 × 10 mL), brine, dried over MgSO<sub>4</sub> and concentrated in vacuo. The crude product was purified by flash chromatography (50% EtOAc in cyclohexane), affording (1*S*,4*R*,5*R*)-**5** (145 mg, 76%, 71% ee) as a colorless liquid. The ee value was determined by HPLC [Chiralcel AD, *n*-hexane-*i*-PrOH, 80:20, 0.8 mL/min, 220 nm]: t<sub>r</sub> (min) = 14.6 (1*S*,4*R*,5*R*), 16.4 (1*R*, 4*S*,5*S*).