Review Article

Highly Stereoselective Hydrogenations—As Key-Steps in the Total Synthesis of Statins

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ABSTRACT Statins are inhibitors of 3-hydroxy-3-methyl-glutaryl coenzyme A reductase (HMG-CoA reductase) and became the standard of care for treatment of hypercholesterolemia because of their efficacy, safety, and long-term benefits. They are administered as diastereo- and enantiomerically pure compounds. We summarize here two new approaches for the total synthesis of the most important representatives, atorvastatin, and rosuvastatin, based on highly stereoselective hydrogenations as keysteps. *Chirality 22:534–541, 2010.* © 2009 Wiley-Liss, Inc.

KEY WORDS: drug synthesis; atorvastatin; rosuvastatin; asymmetric hydrogenation; enantioselectivity; stereoselectivity; statins; asymmetric catalysis

INTRODUCTION

Statins^{1–3} are effective inhibitors of the enzyme 3hydroxy-3-methyl-glutaryl coenzyme reductase (HMG-CoA reductase)⁴ and are a class of hypolipidemic drugs used to lower level of cholesterol⁵ in the blood by reducing the production of cholesterol by the liver in people with or at risk of cardiovascular disease.^{6–8} Recently, statins have become the standard of care for treatment of hypercholesterolemia and hyperlipidaemia^{9–12} due to their efficacy, safety, and long-term benefits.¹³ Among the statin class of drugs, atorvastatin^{14–21} and rosuvastatin^{22–30} are the most powerful lipid-lowering agents (Fig. 1).

Atorvastatin calcium¹³ was the first totally synthetic HMG-CoA-reductase inhibitor developed and marketed as a single enantiomer.^{17–21} Currently, it ranks at the top of the drugs best sold in the world (Atorvastatin calcium tablets are currently marketed by Pfizer under the trade name Lipitor. According to a recent article in Forbes-The World's Ten Best-Selling Drugs: "Pfizer's cholesterol pill Lipitor remains the best-selling drug in the world for the fifth year in a row. Its annual sales were \$12.9 billion, more than twice as much as its closest competitors."). 31 Rosuvastatin calcium is the newest among statins and it was approved as the cholesterol-lowering drug Crestor in August 2003. Rosuvastatin has been called a super-statin, because it appears to reduce low-density lipoprotein (LDL) cholesterol to a greater degree than rivals in its class without additional adverse effects. A number of new methods for the synthesis of rosuvastatin and its precursors has been published recently.26-30

SYNTHESIS OF ATORVASTATIN

We recently reported a new approach for the total synthesis of atorvastatin³² via highly stereoselective hydro-© 2009 Wiley-Liss, Inc. genation of the statin precursor ethyl (5S)-5,6-isopropylidenedioxy-3-oxohexanoate³³ as a key-step. The retrosynthetic pathway to the enantiopure side chain of atorvastatin is depicted on Scheme 1. This involves two-carbon elongation of methyl (S)-2-(2,2-dimethyl-1,3-dioxolan-4-yl)acetate (1) to afford alkyl (S)-4-(2,2-dimethyl-1,3-dioxolan-4-yl)-3oxobutanoate (2) followed by catalytic hydrogenation of the keto group to afford the corresponding hydroxy ester **D**. The protection of the HO-group in compound **D** and the subsequent cleavage of acetal-moiety gives lactone **C**, which may then be used for the preparation of atorvastatin precursor **B**.

As methyl (*S*)-2-(2,2-dimethyl-1,3-dioxolan-4-yl)acetate (1) is commercially available at a rather high price, it was prepared in our lab starting from (*S*)-malic acid by a known sequence (Scheme 2). (*S*)-Malic acid was esterified with methanol in the presence of dimethyl orthoformate to obtain dimethyl (*S*)-malate (3). This material was regioselectively reduced using Moriwake's procedure (sBH₃×SMe₂, THF, cat. NaBH₄).^{34,35} The resulting crude methyl (*S*)-3,4-dihydroxybutanoate was directly transformed without purification into the acetonide (*S*)-1³³ with a yield of 74% after distillation. Saponification of ester (*S*)-1 with aqueous NaOH and subsequent neutralization gave the corresponding acid (*S*)-4 in a chemical yield of 71–

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Fig. 1. Structures of atorvastatin calcium and rosuvastatin calcium.

78%. Acid (*S*)-**4** was elongated by application of a general protocol developed for the synthesis of β -keto esters³⁶ via intermediate formation of imidazolide (*S*)-**5**. After work-up keto ester (*S*)-**2** was obtained after distillation in good yield.³³

To define optimal conditions for the stereoselective hydrogenation of β -keto ester (*S*)-**2** into the desired protected polyol (3*R*,5*S*)-**6** (Scheme 3), we studied the performance of several homogeneous chiral Rh and Ru catalysts.

The hydrogenations were performed under mild conditions (50 bar initial H₂ pressure, room temperature). The results are listed in Table 1. The syn/anti ratios were determined on the basis of NMR data. Hydrogenation of (*S*)-**2** catalyzed with chiral Rh(I) catalysts showed only moderate diastereoselectivities (entries 1–9). Structures of chiral ligands are illustrated on Fig 2. Higher activities were observed in aprotic solvents (compare, entry 1 and 2). The Rh(I) catalyst with (*S*,*S*)-Et-DUPHOS or (*S*,*S*)-Me-DUPHOS as ancillary ligands induced reasonable syn-selectivity. At the same time, with (*R*,*R*)-Et-DUPHOS and (*R*,*R*)-Me-DUPHOS as ligands, completely opposite results were observed.

Ru(II) catalysts based on diphosphines BINAP and Tol-BINAP gave markedly varying results (entries 10–22). Particularly, it was found that the application of commercial or in-house prepared³⁷ Ru[(*R*)-BINAP]Cl₂ afforded a complex mixture of products although the conversion of β-keto ester **2** was complete (entry 10).

In comparison with the related Rh(I) complex (entries 1 and 2), the Ru(II) complex did not work in AcOEt (entry

11). Besides the desired (3R,5S)-6 (syn) alcohol, a side product was formed in MeOH. On the basis of the NMR data,³³ we assumed that a thermodynamic equilibrium is established between (3R,5S)-6 (syn) and the rearrangement product (S,R)-7 during the hydrogenation reaction (Scheme 4). This assumption was confirmed by O-silylation of the mixture of alcohols (3R,5S)-6 and (3R,5S)-7 followed by chromatographic separation of products 8 and 9. Structures of both isomers 8 and 9 were confirmed by elemental analysis, mass-spectra, and NMR data. It is remarkable that 1,3-dioxolane 8 as well as 1,3-dioxane 9 showed only one set of signals in the NMR spectrum. This gives clear evidence that the hydrogenation protocol used produced alcohols (3R,5S)-6 and (3R,5S)-7 with excellent diastereomeric purity. The same rearrangement was also observed with the Ru[(R)-Tol-BINAP]Cl₂ catalyst.

It became obvious that these catalysts have acidic properties and it is thus necessary to neutralize them before use. For this reason, we tested several nitrogenous bases (Et₃N, BnNH₂, imidazole, piperazine). These bases had an uniform influence on the activity of the catalysts. In Table 1 only examples with Et₃N are listed (entries 12–15). The addition of Et₃N completely inhibited the formation of the transacetalization product (*S*,*R*)-**7**. It was shown also that increasing the concentration of Et₃N affected the activity of the catalyst and at a ratio of Ru/Et₃N = 1/2, there was complete inhibition of hydrogenation. It is interesting to note that imidazole and piperazine, which possess two basic centers, behaved similarly to Et₃N. The (*R*)-BINAP complex induced exclusively (within analytical error) the syn-configuration in the product **6**. No significant influ-



Scheme 1. Retrosynthetic pathway to the enantiopure side chain of atorvastatin.



Scheme 2. Synthesis of ethyl (S)-4-(2,2-dimethyl-1,3-dioxolan-4-yl)-3oxobutanoate (2) [(a) MeOH, HC(OMe)₃, cat. HCl, RT, (93%); (b) BH₃·SMe₂, THF, cat. NaBH₄ (1 mol %), RT; (c) Me₂C(OMe)₂, Me₂CO, cat. *p*-TsOH, RT, (74%); (d) 2N NaOH, RT, and neutralization with 2N NaHSO₄, (78%); (e) CDI (*N*,*N*'-carbonyldiimidazole), THF, RT; (f) (EtO₂CCH₂CO₂)₂Mg, THF, RT, 12h, (90%).]

ence of the chiral backbone of the substrate was observed. Partial neutralization of Ru[(*R*)-BINAP]Cl₂ (dimer) could also be achieved by prior treatment with an excess of Et₃N in CH₂Cl₂ followed by evaporation and drying in vacuum. The resulted solid, designated as Ru[(*R*)-BINAP]Cl₂× Et₃N, was tested in the hydrogenation of β -keto ester (*S*)-**2**. Particularly, it was shown that the activity of this complex is comparable to the activity of the catalytic system arising after neutralization in solution (compare with entries 12 and 13).

The activity of the complex is strongly dependent upon the polarity of the alcohol and follows the order: MeOH >EtOH >> i-PrOH (compare entries 16–18).

We noticed that all previously studied catalysts were poorly soluble in the reaction mixture. To overcome this problem we studied the influence of AcONa on the catalytic performance of Ru[(R)-BINAP]Cl₂ (runs 19–20). It is interesting to note that AcONa acted in the same way as Et₃N, for example, lowering the activity with an increase



Scheme 3. Hydrogenation of β -keto ester (*S*)-2 under homogeneously catalyzed conditions.

in the AcONa/Ru ratio and complete inhibition of the hydrogenation at a ratio of 2 (runs 19–20). The parent Ru[(R)-Tol-BINAP]Cl₂×Et₃N complex exhibited similar activity and syn-selectivity (entry 21, compare with entry 16). The solubility of this precatalyst in MeOH was better than the analogous BINAP complex.

Finally, we found that application of Ru[(R)-Tol-BINAP]Cl₂× AcONa (entry 22) showed results similar to Ru[(R)-BINAP]Cl₂ (dimer) with addition of 1 equiv. of AcONa (compare with entry 19).

To optimize the substrate to catalyst ratio, several Ru(II) precatalysts, additives, solvents, and conditions (like, temperature and initial H₂-pressure) have been tested. Finally, the best results were achieved with Ru[(*R*)-Tol-BINAP]Cl₂× AcONa in EtOH at 50°C and 100 bar initial H₂-pressure. Under these conditions it was possible to run the hydrogenation of (*S*)-**2** at a substrate to catalyst ratio of 1000:1 with complete conversion within 4 h. The product (3*R*,5*S*)-**6** (syn) was obtained with at least 99% *de* and ca. 98% chemical purity.

The subsequent transformation of (3R,5S)-6 (syn) into the atorvastatin precursor **15** was accomplished as shown in Scheme 5.³² Protection of the HO-group of (3R,5S)-6 followed by reduction of the ester-moiety gave aldehyde (3R,5S)-**10** in good yield. Upon subsequent cleavage of the acetal, the lactol **11** was obtained. Esterification of the primary hydroxyl group with 4-chlorotoluenesulfonyl chloride, followed by treatment of the sulfonate obtained with NaCN

Entry	Precatalyst	Solvent	Time (min)	Conversion (%)	de (%)
1	${Rh[(S)-BINAP](COD)}BF_4$	MeOH	1000	> 99	54 (syn)
2	$\{Rh[(S)-BINAP](COD)\}BF_4$	AcOEt	120	> 99	52 (syn)
3	$\{Rh[(R)-BINAP](COD)\}BF_4$	AcOEt	120	> 99	50 (anti)
4	$\{Rh[(S,S)-Et-DUPHOS](COD)\}BF_4$	AcOEt	70	> 99	36 (syn)
5	$\{Rh[(R,R)-Et-DUPHOS](COD)\}BF_4$	AcOEt	70	> 99	18 (anti)
6	$\{Rh[(S,S)-Me-DUPHOS](COD)\}BF_4$	AcOEt	120	> 99	40 (syn)
7	$\{Rh[(R,R)-Me-DUPHOS](COD)\}BF_4$	AcOEt	150	> 99	12 (anti)
8	$Rh[(R)-Tol-BINAP](COD)]BF_4$	AcOEt	80	> 99	50 (anti)
9	$\{Rh[(R,S)-JOSIPHOS](COD)\}BF_4$	AcOEt	50	> 99	40 (syn)
10	$Ru[(R)-BINAP]Cl_2$ (dimer)	MeOH	80	> 99	(S,R)-6 (syn) and 9
11	$Ru[(R)-BINAP]Cl_2$ (dimer)	AcOEt	940	< 10	n.d.
12	$Ru[(R)-BINAP]Cl_2$ (dimer) + 0.5 Et_3N	MeOH	100	> 99	> 99 (syn)
13	$Ru[(R)-BINAP]Cl_2$ (dimer) + 1.0 Et_3N	MeOH	200	> 99	98.2 (syn)
14	$Ru[(R)-BINAP]Cl_2$ (dimer) + 2.0 Et_3N	MeOH	1310	0	-
15	$Ru[(S)-BINAP]Cl_2$ (dimer) + 1.0 Et_3N	MeOH	380	> 99	2.3 (syn)
16	$Ru[(R)-BINAP]Cl_2 \times Et_3N$	MeOH	120	> 99	> 99 (syn)
17	$Ru[(R)-BINAP]Cl_2 \times Et_3N$	EtOH	300	> 99	> 99 (syn)
18	$Ru[(R)-BINAP]Cl_2 \times Et_3N$	<i>i</i> -PrOH	340	0	-
19	$Ru[(R)-BINAP]Cl_2$ (dimer) + 1.0AcONa	MeOH	140	> 99	> 99 (syn)
20	$Ru[(R)$ -BINAP] Cl_2 (dimer) + 2.0AcONa	MeOH	1000	0	-
21	$Ru[(R)-Tol-BINAP]Cl_2 \times Et_3N$	MeOH	100	> 99	> 99 (syn)
22	Ru[(R)-Tol-BINAP]Cl ₂ ×AcONa	MeOH	130	> 99	> 99 (syn)

TABLE 1. Stereoselective hydrogenation of β -keto ester (S)-2 with chiral Rh(I) and Ru(II) catalysts³³

HIGHLY STEREOSELECTIVE HYDROGENATIONS



Fig. 2. Chiral diphosphine ligands applied for stereoselective Rh(I)- and Ru(II)-catalyzed hydrogenation of β -keto ester (S)-2.

gave the nitrile **12**. Raney nickel catalyzed hydrogenation of the latter afforded the desired amine **13**, which in turn was submitted to the Paal–Knorr condensation with the appropriate functionalized diketone **14** to give atorvastatin lactol **15** in 71% yield. Oxidation of the acetal and desilylation in order to obtain atorvastatin free acid can be conveniently performed as recently suggested.³⁸

SYNTHESIS OF ROSUVASTATIN

Similarly, as shown earlier for the synthesis of atorvastatins we developed a new pathway for the production of rosuvastatin comprising two highly stereoselective hydrogenations as key-steps.^{39,40} The first part covers the asymmetric hydrogenation of ethyl 5,5-dimethoxy-3-oxopentanoate (**17**) to ethyl 3-hydroxy-5,5-dimethoxypentanoate (**18**) (Scheme 6).⁴¹ From the retrosynthetic analysis depicted in Scheme 6 it follows that the chiral side chain of rosuvastatin acid **A** can be created by diastereoselective and chemoselective reduction of the keto group in structure **B**. The protected hydroxy ylide (*R*)-**19** was postulated as a possible building block for the preparation of **B**, as long as the Wittig coupling reaction with the corresponding aldehyde could be performed and the HO-protecting



Scheme 4. Transacetalization reaction in acidic medium.

group subsequently removed. Ylide (*R*)-**19** could be prepared³⁹ starting from methyl 3,3-dimethoxypropanoate (**16**) by two-carbon elongation followed by the enantioselective hydrogenation of the prochiral keto group in β -keto esters **17** to give finally β -hydroxy ester (*R*)-**18**. Protection of the HO-group and transformation of the (MeO)₂CH-moiety should accomplish the synthesis of ylide (*R*)-**19**.

β-Keto ester **17**, serving as a prochiral substrate for the enantioselective hydrogenation, was assembled by a twocarbon chain elongation of 3,3-dimethoxypropionic acid **20** (Scheme 7). The latter was easily available from ester **16** by hydrolysis. A modified two-step Masamune procedure,³⁶ which involved first the activation of acid **20** with CDI (*N*,*N*-carbonyldiimidazole), followed by subsequent treatment with Mg(O₂CCH₂CO₂Et)₂ prepared in situ, afforded ethyl 5,5-dimethoxy-3-oxopentanoate (**17**) in 91% yield.

For the asymmetric hydrogenation of functionalized β -keto ester **17** (Scheme 8), a number of Rh(I) and Ru(II) catalysts were tested. Structures of applied chiral ligands are illustrated on Figs. 2 and 3. Selected results are summarized in Table 2.

Rh(I) complexes displayed good activities but only poor enantioselectivities (entries 1–8). Ethyl acetate was found to be the solvent of choice, since reactions in methanol and other solvents ran significantly more slowly.

In strong contrast, hydrogenations catalyzed by commercial Ru[(*R*)-BINAP]Cl₂ proceeded smoothly at room temperature and confirmed the previously reported results with related bromide based catalyst.⁴² A route to the required chiral methyl 3-hydroxy-5,5-dimethoxypentanoate (**18**) had been already reported by Genêt et al.⁴² who hydrogenated the corresponding β -keto ester **17** in the presence of Ru(BINAP)Br₂ (2 mol % catalyst, room temperature, 1 atm H₂ pressure) and obtained **18** in 86% yield and with enantioselectivity >95% *ee.* According to our recently published results,³⁹ the hydrogenation of β -keto ester **17** proceed in MeOH with an initial H₂-pressure of 50 bar to give β -hydroxy ester (*R*)-**18** with an enantiose-*Chirality* DOI 10.1002/chir



Scheme 5. Synthesis of atorvastatin by the lactol pathway [(a) *t*-BuPh₂SiCl, imidazole, DMF, 10°C, (94%); (b) DIBALH, Et₂O, -78°C, (77%); (c) HC(OMe)₃, MeOH, cat. *p*-TsOH×Py, 70°C, (64%); (d) *p*-ClC₆H₄SO₂Cl, Py, RT, (96%); (e) NaCN, DMSO, RT, (79%); (f) H₂, Ra–Ni, NH₃, MeOH, RT, (>99%); g) THF/toluene, reflux, 30h, (71%).]



Scheme 6. Retrosynthetic pathway to the enantiopure side chain of statins.



Scheme 7. Synthesis of the prochiral β-keto ester 17 [(a) 2N NaOH, RT, and then 2N NaHSO₄; (b) CDI (*N*,*N*-carbonyl-diimidazole), THF, RT; (c) (EtO₂CCH₂CO₂)₂Mg, THF, RT, 12 h, (91%).]



Scheme 8. Enantioselective hydrogenation of functionalized β-keto ester 17.

lectivity of 98.7% (entry 11). The reaction was complete within 3 h at a ratio of S/C = 500 (entry 11) and within 10 h at a ratio of S/C = 1000 (entry 12). An increase in the temperature to 70°C made it possible to complete the reaction in 1h with a S/C ratio as high as 10,000 (entry 16), thus achieving a turnover frequency (TOF) of 10,000 h⁻¹. Turnover numbers (TONs) of up to 20,000 could be reached at this temperature, and the reaction was complete within 8 h (entry 17). Enantioselectivities at increased temperatures were slightly lower, but still above 95% in all trials. After hydrogenation of **17** at 100°C and 100 bar of initial H₂-pressure in the presence of Ru[(*R*)-BINAP]Cl₂ (dimer) and a substrate/catalyst ratio = 20,000, *Chirality* DOI 10.1002/chir



Fig. 3. Chiral diphosphine ligands used for the Rh(I)- and Ru(II)-catalyzed enantioselective hydrogenation of β -keto ester 17 [for structures of other chiral diphosphines, like (*R*,*S*)-JOSIPHOS, (*S*,*S*)-Me-DUPHOS, and (*R*)-BINAP, see Figure 2].

Entry	Precatalyst	Solvent	S/C	T (°C)	Time (h)	Conversion (%)	ee (%)
1	$\{Rh[(R,S)-JOSIPHOS](COD)\}BF_4$	AcOEt	500	r. t.	2	> 99	15.2 (R)
2	$\{Rh[(R,S)-JOSIPHOS](COD)\}BF_4$	AcOEt	1000	r. t.	20	88	n.d.
3	$\{Rh[(R,R)-DIOP](COD)\}BF_4$	AcOEt	500	r. t.	4	> 99	7.6 (S)
4	$\{Rh[(R)-BINAP](COD)\}BF_4$	AcOEt	500	r. t.	20	> 99	58.6 (S)
5	$\{Rh[(S,S)-CHIRAPHOS](COD)\}BF_4$	AcOEt	500	r. t.	20	> 99	29.4 (S)
6	$\{Rh[(S,S)-DEGUPHOS](COD)\}BF_4$	AcOEt	500	r. t.	18	< 5	n.d.
7	$\{Rh[(S,S)-Me-DUPHOS](COD)\}BF_4$	AcOEt	200	45	6	78	n.d.
8	$\{Rh[(S,S)-Me-catASiumM](COD)\}BF_4$	AcOEt	500	r. t.	10	> 99	6.4(S)
9	$Ru[(R)-BINAP]Cl_2$ (dimer)	CH_2Cl_2	200	r. t.	20	46	n. d.
10	Ru[(R)-BINAP]Cl ₂ (dimer)	EtOH	500	r. t.	20	68	n. d.
11	$Ru[(R)-BINAP]Cl_2$ (dimer)	MeOH	500	r. t.	3	> 99	98.7 (R)
12	Ru[(R)-BINAP]Cl ₂ (dimer)	MeOH	1000	r. t.	10	> 99	98.7 (R)
13	$Ru[(R)-BINAP]Cl_2$ (dimer)	MeOH	2000	r. t.	24	35	n.d.
14	$Ru[(R)-BINAP]Cl_2$ (dimer)	MeOH	2000	50	1	> 99	97.8 (R)
15	Ru[(R)-BINAP]Cl ₂ (dimer; 10 bar)	MeOH	2000	50	4	> 99	98.4 (R)
16	Ru[(R)-BINAP]Cl ₂ (dimer)	MeOH	10,000	70	1	> 99	97.0 (R)
17	Ru[(R)-BINAP]Cl ₂ (dimer)	MeOH	20,000	70	8	> 99	95.8 (R)
18	$Ru[(R)$ -BINAP] Cl_2 (dimer; 100 bar)	MeOH	20,000	100	0.3	73	90.0 (R)
19	Ru[(R)-Tol-BINAP]Cl ₂ (dimer)	MeOH	500	r. t.	3	> 99	98.2 (R)

TABLE 2. Enantioselective hydrogenation of β-keto ester 17 using chiral Rh(I) and Ru(II) catalysts³⁹ at 50 bar of initial H₂-pressure (if not otherwise noted)



Scheme 9. Determination of the enantiomeric excess of alcohol 18 with the assistance of enantiopure phosphite (R)-21.

incomplete conversion and diminished enantioselectivity (90%) was noted (entry 18).

Methanol was found to be the solvent of choice, since the reaction rates dropped dramatically when the reaction was performed in CH_2Cl_2 (entry 9) or ethanol (entry 10).

In contrast, at low hydrogen pressure (10 bar), β -keto ester **17** could be successfully reduced without any loss of enantioselectivity, but the reaction was significantly slower than under a pressure of 50 bar (compare entries 14 and 15). The use of the related Ru((*R*)-Tol-BINAP)Cl₂ complex improved neither the activity nor the stereoselectivity of the hydrogenation (entry 19). The results were identical to those obtained with Ru(BINAP)Cl₂.

Enantiomeric compositions for β -hydroxy ester **18** were determined with help of the chlorophosphite chiral derivatizing agent (*R*)-**21** (Scheme 9),³⁹ prepared from PCl₃ and (*R*)-BINOL. Chlorophosphite (*R*)-**21** reacted quantitatively and rapidly with alcohol **18** to give phosphites **22a** and **22b**. The ³¹P NMR spectrum of the two diastereomeric compounds **22** is characterized by signals separated to the base line (δ_P 151.4 and 154.6 ppm), so it is possible to determine the enantiomeric composition of the catalytic hydrogenation product precisely.

Alcohol (*R*)-**18** was transformed into ylide (*R*)-**19** by the route depicted in Scheme 10. Protection of the HOgroup was performed by treatment with *t*-BuPh₂SiCl in DMF in the presence of imidazole as a base. Selective cleavage of the acetal-moiety in (*R*)-**23** and oxidation was performed in one-pot by treatment with Jones' oxidation reagent at 0°C, proceeding without racemization at the C-3 carbon atom.



Scheme 10. Multistep synthesis of ylide (*R*)-19 [(a) *t*BuPh₂SiCl, imidazole, DMF, 0°C to RT, overnight, (90%); (b) Jones' oxidation: CrO₃, H₂SO₄, acetone/water, 0°C to RT (68%); (c) AlkOC(O)Cl, NEt₃, toluene, 0°C to RT, 1 h; (d) Ph₃P⁺-CH₃Br⁻, *n*BuLi, THF, -78°C to 0°C, then 15, -78°C to 0°C, overnight, (45%).]



Scheme 11. Synthesis of rosuvastatin ethyl ester [(a) MeCN, reflux, 14 h, (70%); (b) HF (aq.), MeCN; (c) Et₂BOMe, NaBH₄, THF/MeOH, -78°C, (85%, over two steps).]

Mixed anhydrides (*R*)-**25a,b** [R = Me (a) or R = Et (b)], prepared by treatment of (*R*)-**24** with 1.5 equiv. of corresponding methyl or ethyl chloroformate in toluene in the presence of Et₃N, were submitted for the synthesis of ylide (*R*)-**19** without additional purification. Treatment of (*R*)-**25a** or (*R*)-**25b** with methyltriphenylphosphonium bromide and *n*BuLi furnished the targeted Wittig reagent (*R*)-**19** in 45% yield.

Wittig coupling of aldehyde **26** and ylide (*R*)-**19** was performed under reflux in CH₃CN over 14 h and gave the rosuvastatin precursor **27** in a yield of 70% (Scheme 11).⁴¹ Removal of the *t*BuPh₂Si protective group by treatment with aqueous HF in CH₃CN and final exclusive diastereoselective reduction of the keto group with Et₂B(OMe) and NaBH₄ in THF/MeOH at -78° C, afforded rosuvastatin ethyl ester in 85% yield.

We have presented here two new approaches for the total syntheses of atorvastatin and rosuvastatin, based on highly stereoselective hydrogenations of intermediate β -keto esters. Optimized syntheses of substrates and subsequent multistep transformations of the chiral hydrogenation products in the desired drug constituents show clearly the great value of stereoselective homogeneous catalysis for the establishment of new and patent-free routes to drugs.

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