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Regioselective *5-endo-dig* Electrophilic lodocyclization of Enediynes: A Convenient Route to lodo-substituted Indenes and Cyclopenta Fused Arenes

Rakesh K. Saunthwal, Abhinandan K. Danodia, Monika Patel, Sushil Kumar and Akhilesh K. Verma*

Abstract: An efficient iodine-mediated regioselective tandem approach for the synthesis of symmetric 4a-g and asymmetric 5a-f indenes cyclopenta iodo-substituted and stereoselective [b]pyridine/thiophene 6a-h from easily accessible enediynes 3a-u via in situ formation of iodonium intermediate followed by regioselective 5-endo-dig cyclization has been described. The intramolecular electrophilic iodocyclization was selectively triggered by the distribution of electronic density along the alkyne bond. Subsequently, the iodo-substituted indenes have been diversified by employing palladium-catalyzed cross-coupling reactions and the coupled products were further confirmed by the X-ray crystallographic studies.

The carbo- and heterocyclic compounds,^[1] constitutes an important class due to their presence in a large number of drug molecules which acquire a remarkable pharmaceutical activity.^[2,3] For instance, indene skeleton with an *exo*-alkylidene moiety is embedded in Sulindac (1) which is a non-steroidal anti-inflammatory drug,^[4] Dimethindene (2) an oral anti-histamine agent and Donepezil (3) ^[5] an anti-Alzheimer drug (Figure 1).^[6,7] In addition to these commercial medicines, indenes are the structural constituent of group IV metallocene bound catalyst which are used for olefin polymerization.^[8]



Figure 1 Biological significance of indenes

Due to immense biological activity, a number of synthetic methodologies have been developed for the construction of indene moiety such as the ring enlargement of substituted cyclopropenes,^[9] the Lewis acid mediated ring closure of substituted 1,3-butadienes ^[10] and the reduction or dehydration of indanones.^[11] During the past decade, a rapid interest has been developed in the area of electrophilic iodocyclization and it emerge as an extremely active and original field for heterocyclic synthesis.^[12-13] The halide functionality introduced through

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electrophilic iodocylization provide a useful avenue for structure elaboration to generate complex molecules.^[14] The electrophilic iodocyclization of alkynes and allene bound substrate are an alternative powerful tool for the construction of diversely functionalized carbocycles and heterocycles having a mono-iodo substituent. However, a number of methods are available in the literature for the synthesis of monoiodonated indenes by electrophilic cyclization.^[15] On introducing an electron-rich alkyne moiety into the domino transformations via cross-coupling chemistry leads to the rapid production of polycyclic frameworks.



Scheme 1. Approach for the synthesis of indene derivatives.

In last few years, Takai,^[16a] Larock,^[16b,c] Kondo ^[1] and Tian ^[16d] groups have demonstrated the transition metal C-H functionalization of alkynes followed by electrophilic annulation to generate indene. Alabugin and co-workers ^[17a] have developed a two-step strategy for the synthesis of benzofulvene via Bu₃SnH mediated 5-*exo-dig* radical cyclization of diaryl enediyne. Subsequently, Lee et. al ^[17b] have explored the synthesis of benzofulvenes via Bergman cyclization using Pd/Cu-catalyst. Farinola group^[17c] demonstrated the application

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of symmetric benzofulvene intermediate for the 5-exo-dig cyclization of macro conjugated molecules with limited substrate scope (Scheme 1a). In 2009, Khan^[18a] and fellow workers reported the iodine-mediated electrophilic cyclization of substituted ethenylmalonates for the construction of 3-iodo-1Hindene derivatives by 5-endo-dig carbocyclization (Scheme 1b). The synthesis of iodinated indenes via electrophilic carbocyclizations of propargylic alcohol in nitromethane have been described by Liang group ^[18b] in 2011 (Scheme 1c). Later, Wang and co-workers^[18c] synthesized substituted 2,3diiodoindenes in dichloromethane (Scheme 1c). They extended the chemistry for the synthesis of 13H-indeno[1,2-Iphenanthrenes from synthesized 2,3-diiodoindene through Suzuki and Scholl oxidative coupling reaction. In 2012, Hashmi and co-workers,^[19a] described gold-catalyzed synthesis of benzofluvene, via dual activation of substrate (Scheme 1d). During our manuscript preparation, an iodonium-induced cyclization of arylallenes has been demonstrated by Grandclaudon et al. for the synthesis of 2-iodoindenes (Scheme 1e).^[19b]

Owing to the broad biological importance of indenes and limitations in the previous reported strategies, such as use of expensive metal catalysts, limited scope or availability of starting materials and requirement of multistep procedures. Due to the above drawbacks and presence of indenes in many natural products, this motivated us for designing an efficient route to synthesize iodo-substituted indenes and heterocvcles. Encouraged by our laboratory results for the synthesis of pyrano[4,3-b]quinoline via iodine-mediated electrophilic cyclization,^[20] herein we report the electrophilic iodocyclization chemistry for the synthesis of diiodoindenes from dialkynyl aromatic/heteroaromatic through sequential Sonogashira/ Sonogashira coupling reaction followed by 5-endo-dia cyclization.

Our approach towards the synthesis of diiodoindene preparation involves of dialkynyl two steps: (i) aromatic/heteroaromatic (3) from 1-bromo-2-iodo-benzene (1) by sequential Sonogashira/Sonogashira coupling reaction and (ii) electrophilic iodocyclization of dialkynyl aromatic/heteroaromatic. The coupling products 3 were obtained in 76-85% yield in the presence of 2.0 + 3.0 mol % of Pd(PPh₃)₂Cl₂ by coupling of **1** with terminal alkyne **2**.^[21] The coupling reaction accommodates a wide variety of functional groups (Scheme 2).



Scheme 2. Sonogashira Reaction of Hindered Aromatic/Heteroaromatic Dihalide with Alkynes

To identify the optimal reaction conditions for the reaction, we examined the reaction of 1,2-bis(phenylethynyl)benzene (**3a**) with 1.2 equiv of I₂, and 2.0 equiv of K₂CO₃, in 2.0 mL of CH₂Cl₂ at 25 °C for 0.5 h, the corresponding cyclized product was obtained in 37% yield in a mixture of E and Z (*E:Z*:75:25) isomers (Table 1, entry 1). Increasing the amount of I₂ from 1.2 equiv to 1.5 equiv and reaction time from 0.5 h to 1 h, the

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desired product 4a/4a' was observed in 47% yield (entry 2). Further increasing of the amount of molecular I₂ from 1.5 equiv to 2.0 and then to 2.4 equiv afforded the mixture of isomers 4a/4a' in 74 and 88% yields respectively (entries 3 and 4). When the reaction was monitored at short interval of time with 2.4 equiv of molecular iodine, the yield of the corresponding iodocyclized product 4a/4a' remain unchanged (entry 5). Inferior results were obtained with K₃PO₄, NaOH, and CsCO₃ (entries 6-9), while no iodocyclized product was detected with Et₃ N (entry 10). Although CH₂Cl₂ is an inadequate solvent for K₂CO₃ however; K₂CO₃ provides a mild basic condition which is essential for a clean and high yielding reaction. K₂CO₃ is presumed to avoid the by-product of the reaction which can degrade the desired product.²² In the absence of base, no progress in the reaction was observed (entry 11). The electrophilic cyclization of 3a using CICH2CH2CI solvent at 25 °C and 60°C successfully provided the desired product 4a/4a' in 72 and 85% vields respectively with 60:40 stereoisomeric ratio (entries 12 and 13). However, other solvents like THF. DMF. MeCN failed to provide the iodocyclized product (entries 14-16).

Table 1. Optimization of iodocyclization of dialkynylbenzene^a

Entry	Solvent	Base	l ₂ reagent	7(°C)	Time(h)	Yield(%) ⁻ 4a/4a'
			(equiv)			$(E:Z)^c$
1	CH ₂ Cl ₂	K ₂ CO ₃	1.2	25	0.5	37(75:25)
2	CH_2CI_2	K ₂ CO ₃	1.5	25	1	47(70:30)
3	CH ₂ Cl ₂	K ₂ CO ₃	2.0	25	1	74(70:30)
4	CH ₂ Cl ₂	K ₂ CO ₃	2.4	25	1	88(70:30)
5	CH ₂ Cl ₂	K ₂ CO ₃	2.4	25	0.5	88(70:30)
6	CH ₂ Cl ₂	K ₃ PO ₄	2.4	25	0.5	66(68:32)
7	CH ₂ Cl ₂	K_3PO_4	2.4	25	1	78(68:32)
8	CH ₂ Cl ₂	NaOH	2.4	25	1	40(70:30)
9	CH ₂ Cl ₂	CsCO ₃	2.4	25	1	61(70:30)
10	CH ₂ Cl ₂	Et ₃ N	2.4	25	1	N. R^{d}
11	CH ₂ Cl ₂	-	2.4	25	1	N. R^d
12	CICH ₂ CH ₂ CI	K ₂ CO ₃	2.4	25	1	72(60:40)
13	CICH ₂ CH ₂ CI	K ₂ CO ₃	2.4	60	1	85(60:40)
14	THF	K ₂ CO ₃	2.4	60	1	N. R^{d}
15	DMF	K_2CO_3	2.4	60	1	N. R^d
16	MeCN	K_2CO_3	2.4	60	1	N. R ^d

[a] Reaction were performed using 1,2-bis(phenylethynyl)benzene **3a** (0.5 mmol), I₂, and base (2.0 equiv) and 1.5 mL of solvent unless otherwise noted.[b]Total isolated yield of two isomers. [c] Stereoisomeric ratio. [d] no reaction.

After establishing the optimal reaction conditions, we explored the substrate scope of electrophilic iodocyclization chemistry by employing various symmetrical **3a–g** and unsymmetrical dialkyne **3h–m** in the presence of molecular iodine and afforded the corresponding mixture of *E* and *Z* isomers in 67–88% yields. Substrate **3a** bearing phenyl substituent at R¹ and R² provided a mixture of *E* and *Z* (70:30) isomer in 88% yield. Alkynes **3b–g** bearing electron-rich substituents at R¹ and R² on reaction with iodine afforded the *E:Z* mixture of corresponding products in 77–88% yields. Substrates **3c**, bearing methyl substitution at the *meta* position of the phenyl ring afforded the desired product **4c** along with **4c'** in comparatively lower yield. Reaction of 1,2-bis((4-ethylphenyl) ethynyl)benzene **3d** provided the mixture of stereoisomers *E:Z* in equal ratio. Substrates **3e–f** bearing a *tert*-

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butyl, and *n*-butyl were found suitable for the reaction and provided the mixture of desired products (4+4') e-f in 87 and 84% yield respectively (entries 5 and 6). The structure of 4e (*E*-isomer) was confirmed by the X-ray crystallographic studies. The electron-rich thiophene alkyne 3g afforded the iodocyclized product in 77% yield (Scheme 3).



Scheme 3. Scope of symmetrical dialkyne [a] The reactions were performed using dialkynylbenzene 3 (0.50 mmol), 2.4 equiv of I₂ and 2.0 equiv of K₂CO₃ in 2.0 mL of CH₂Cl₂ at 25 °C for 0.5 h. [b] Isolated yields. Data CCDC number of **4e** is 845040.

The reaction of unsymmetrical alkyne, 1,2-bis((4-(alkyl)phenyl) ethynyl)benzene containing different substituent at the end of triple bond have also been investigated. Under similar condition the various substrates **3h**-**m** bearing different alkyl coupling partner were found successful for the reaction and provided the mixture of *E* and *Z* isomers **5a**-**f** in 67–80% yields. The structure of **5c** (*Z*-isomer) was confirmed by the X-ray crystallographic studies. Interestingly, when substrates **3j** were used for iodocyclization, the formation of *Z*-isomer was predominated; the possible reason could be due to the spatial arrangement of intermediate during the course of reaction. It is noteworthy that alkynes **3m** bearing a electron-withdrawing trifluoromethyl group at 4-position of the phenyl ring was found successful for the reaction and provided the desired product (**5+5')f** in *E:Z* ::74:26 stereoisomeric ratio in 73% yield (Scheme 4).



Scheme 4. Scope of unsymmetrical dialkyne. [a] The reactions were

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performed using dialkynylbenzene ${\bf 3}$ (0.50 mmol), 2.4 equiv of l_2 and 2.0 equiv of K_2CO_3 in 2.0 mL of CH_2Cl_2 at 25 °C for 0.5 h. [b] Isolated yields. CCDC number of ${\bf 5c}$ is 978613 .



Shielding and deshielding effect

Figure 2. Anisotropic effect

Encourage by the result from X-ray crystallography, NOESY and NMR spectroscopy, we ascertained the mixture of *E* and *Z* isomer. In *Z*-isomer, possibly the π orbital of phenyl ring **A** and **B** are in same plane. When an external magnetic field is applied, then **Ha** (δ 5.85-6.00 ppm) and **Hb** proton comes in same region, therefore **Ha** proton comes perpendicular to applied magnetic field and showed the shielding of **Ha** (δ 5.85-6.00 ppm) proton (Figure 2). In *E*-isomer no anisotropic effect occurred between phenyl ring **A** and **B** (as shown in Figure 2). We presumed that the deshielding of **Ha** (δ 8.85-9.01 ppm) proton might be due to induced dipole interaction. This anisotropic effect also supports the formation of mixture of stereoisomers in case of 1,2-bis((4-(alkyl)phenyl)ethynyl)benzenes (**3**).



Scheme 5. Synthesis of cyclopenta fused arenes via electrophilic iodocyclization

We further extended the scope of the developed chemistry for

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the synthesis of an important class of heterocyclic scaffold. In contrast to the phenyl dialkyne, introduction of heteroatom in the phenyl ring helped in controlling the stereoselectivity of product (Scheme 5). When we performed the reaction using electrondeficient pyridyl substrates containing symmetrical alkynes bearing -Ph (3n),-p-tolyl (3o) and 4-methoxyphenyl (3p) groups with iodine, we observed exclusive Z isomer of 6a, 6b and 6c in 70%, 72% and 75% yields, respectively. Unsymmetrical alkyne(3q) were also successful in providing the iodocyclized product 6d in 73% yield. Interestingly, 2,3-bis (phenylethynyl) pyrazine (3r) and 2,3-bis(phenylethynyl)quinoxaline (3s) were also capable in providing the intriguing iodocyclized products 6e-f in 68-72% yields with excellent stereoselectivity. The synthesis of (E)-6-iodo-4-(iodo(4-methoxyphenyl)methylene)-5-(4-methoxy phenyl) -4H-cyclopenta[b]thiophene (6g) and (E)-4-(6-iodo-4-(iodo(4-methoxyphenyl) methylene)-4H-cyclopenta[b] thiophen-5-yl)-N.N-dimethylaniline (6h) were carried out in good vields with high stereoselectivity using dialkynylthiophene 3t and 3u using optimized reaction conditions. It is noteworthy, that the reaction of heterocyclic dialkyne provided a single stereoisomer probably due to the electronic effect of the heterocyclic moiety.



Scheme 6. Plausible mechanism

On the basis of the above observations, a plausible reaction mechanism for 5-*endo-dig* iodocylization has been described in Scheme 6. Presumably the electrophile (I_2) and the enediynes **3** would generate the cyclic iodonium intermediate **P** and **Q** upon electrophilic attack of the I_2 on the electron deficient carbon of alkyne. Subsequent attack of nucleophile provides the mixture of *E* and *Z* isomers via 5-endo-dig cyclization. The formation of *E* and *Z* isomer depends upon the attack of the iodine nucleophile on to alkyne carbon via route a or b. The present strategy depicts the dual role of I_2 as an electrophile as well as nucleophile.

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Scheme 7. Synthetic elaboration using Pd-catalyzed cross-coupling reactions

We further investigated the structural elaboration of iodosubstituted indene via palladium-catalyzed cross-coupling reactions. To this end, mixture of (4+4')e was functionalized by applying palladium-catalyzed Suzuki and Sonogashira coupling reactions and the corresponding coupling products 8 and 9 were obtained in 78% and 80% yields, respectively (Scheme 7). The structure of Sonogashira coupled product was further confirmed by X-ray crystallographic data (Figure 3).



Figure 3. X-ray crystallographic ORTEP drawing of compound 9 drawn at the 50% probability level . CCDC number of 9 is 978612

In summary, we have described the iodine-mediated 5endo-dig cyclization for the regio- and stereoselective synthesis of diiodo-substituted indenes, cyclopenta[b]pyridine/thiophene from easily accessible enediynes via in situ formation of iodonium intermediate. It is noteworthy, that the reaction of heterocyclic arenes provided the single stereoisomers in good yields. Iodo-cyclization method leads to the synthesis of halogen containing indene derivatives which thereafter elaborated using palladium catalyzed coupling reactions, such as Suzuki-Miyaura and Sonogashira coupling reactions. The Xray crystallographic studies confirmed the stereochemistry of the reaction products. Due to the immense diversity in the substitution pattern, this developed chemistry can be used for the synthesis of various heterocyclic systems.

Experimental Section

General Information Method

Nuclear magnetic resonance spectra were recorded in CDCI₃, ¹H NMR (400 MHz) and ¹³C NMR (100 MHz), at ambient temperature. Chemical shifts (δ) for all protons are reported in parts per million (ppm) and were measured relative to the residual CHCI₃ resonance as an internal reference in the deuterated solvent. Chemical shifts were reported as parts per million (δ in ppm) using tetramethylsilane (TMS) as internal standard or by reference to proton resonances resulting from incomplete deuteration of the NMR solvent. The following abbreviations were used to

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describe the multiplicities: when appropriate (s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet, dd = doublet of doublet, td = triplet of doublet). Reactions were monitored using thin-layer chromatography on commercially prepared silica gel plates and visualized by either UV irradiation or by staining with I_2 . Chemical yields are referred to the pure isolated substances. Chromatography using 100– 200 mesh size silica gels.

General Procedure for the Synthesis of Diiodoindene 4+4'

To a vial dialkynyl aromatic/heteroaromatic **3** (0.50 mmol) and 2.4 equiv of I₂ and 2.0 equiv of K₂CO₃ was added in CH₂Cl₂. The reaction vial was then sealed and stirred at room temperature until TLC revealed complete conversion of the starting material. The solution was washed with saturated solution of Na₂S₂O₃ and then extracted with EtOAc(3X10 mL).The combined organic layers were dried over Na₂SO₄, concentrated under vacuum, and purified by column chromatography using 100– 200 mesh size silica gels (hexane) to afford the corresponding product **3**

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