



Benzyllic-acetoxylation of alkylbenzenes with PhI(OAc)₂ in the presence of catalytic amounts of TsNH₂ and I₂

Haruka Baba, Katsuhiko Moriyama, Hideo Togo *

Graduate School of Science, Chiba University, Yayoi-cho 1-33, Inage-ku, Chiba 263-8522, Japan

ARTICLE INFO

Article history:

Received 16 May 2011

Revised 7 June 2011

Accepted 8 June 2011

Available online 16 June 2011

ABSTRACT

Treatment of alkylbenzenes with (diacetoxyiodo)benzene in the presence of catalytic amounts of *p*-toluenesulfonamide or *p*-nitrobenzenesulfonamide, and molecular iodine in 1,2-dichloroethane at 60 °C gave the corresponding (α -acetoxy)alkylbenzenes in good to moderate yields. The present reaction is a simple method for the introduction of an acetoxy group to the benzylic position of alkylbenzenes.

© 2011 Elsevier Ltd. All rights reserved.

Keywords:

(Diacetoxyiodo)benzene

(Dibenzoyloxyiodo)benzene

Alkylbenzene

p-Toluenesulfonamide

p-Nitrobenzenesulfonamide

(α -Acetoxy)alkylbenzene

Molecular iodine

(Diacetoxyiodo)benzene (DIB) is one of the most useful hyper-valent iodine reagents for organic synthesis because it can serve as an alternative to toxic heavy-metal reagents.¹ The advantages of DIB are as follows: it is a non-metal oxidant and it can be used for not only polar reactions, but also radical reactions to generate oxygen-centered radicals, nitrogen-centered radicals, and carbon-centered radicals.² For the radical reactions, a DIB-molecular iodine (I₂) system is used and the initial formation of acetyl hypoiodite (CH₃CO₂I) from the reaction of DIB and I₂ is the key step.³ Synthetic studies of alkoxy radicals derived from substrates, such as steroid-alcohols and sugars, with DIB and I₂ have been well carried out by Suarez et al.² We have also examined the synthetic utility of nitrogen-centered radicals for the construction of tetrahydroquinolines, benzosultams, and saccharins from sulfonamides with DIB and I₂ under irradiation with a tungsten lamp,⁴ and oxygen-centered radicals for the construction of chromans from 3-arylpropanols with DIB and I₂ under irradiation with a tungsten lamp.⁵ The formation of tetrahydroquinolines, benzosultams, and chromans proceeds through the intramolecular cyclization of the formed sulfonamidyl radicals and alkoxy radicals onto their aromatic rings. Recently, the α -sulfonylamidation of alkylbenzenes with DIB, I₂, and *p*-toluenesulfonamide at 50 °C without solvent was reported to provide α -(*p*-toluenesulfonylamido)alkylbenzenes in moderate to good yields.⁶ In those reactions, the reaction of

ethylbenzene with DIB, I₂, and *p*-toluenesulfonamide in 1,2-dichloroethane gave a mixture of α -(*p*-toluenesulfonylamido)ethylbenzene and α -(acetoxy)ethylbenzene in 5% and 4% yields, respectively.⁶ We have also reported the preparation of α -(*p*-toluenesulfonylamido)alkylbenzenes from alkylbenzenes with *p*-toluenesulfonamide and 1,3-diiodo-5,5-dimethylhydantoin (DIH), which works as synthon of a DIB and I₂ system.⁷ Here, as part of our synthetic study of DIB for organic synthesis,^{4,5} we would like to report the benzylic-acetoxylation of alkylbenzenes with DIB in the presence of catalytic amounts of *p*-toluenesulfonamide and I₂. To the best of our knowledge, the following methods for the practical preparation of α -(acetoxy)alkylbenzenes from alkylbenzenes are known, such as K₂S₂O₈ with transition metal salts in AcOH at 115 °C,^{8a} K₂S₂O₈/LiBr (cat.)/AcONa in AcOH at 115 °C,^{8b} CAN (cerium ammonium nitrate, cat.)/KBrO₃ in AcOH at 90–110 °C,^{8c} NHPI (N-hydroxyphthalimide, cat.)/Co(OAc)₂ (cat.)/HNO₃ (cat.) under air in AcOH at 80 °C,^{8d} DDQ (2,3-dichloro-5,6-dicyano-1,4-benzoquinone) in AcOH under ultrasonic irradiation,^{8e} NaO₄/LiBr(cat.) in AcOH at 90–110 °C,^{8f} and related reactions.^{8g,h} The Pd(OAc)₂-catalyzed α -acetoxylation of alkylbenzenes with DIB in AcOH at 100 °C, and the Pd(OAc)₂/CuI-catalyzed α -acetoxylation of alkylbenzenes with oxygen in AcOH were also reported.⁹ However, there are several drawbacks such as toxicity and/or the high cost of some transition metals.

The present reaction was carried out as follows: ethylbenzene **1a**, DIB, I₂, and *p*-toluenesulfonamide were added to 1,2-dichloroethane under an argon atmosphere. The mixture was warmed at 60 °C for 2 h. After the reaction, the mixture was poured into satd aq sodium

* Corresponding author. Tel./fax: +81 43 290 2792.

E-mail address: togo@faculty.chiba-u.jp (H. Togo).

sulfite solution and extracted with diethyl ether. After removal of the ether solvent, the residue was subjected to preparative TLC on silica gel to give α -(acetoxy)ethylbenzene **2a** in 63% yield as shown in Table 1 (entry 5).¹⁰ In the absence of either I₂ or *p*-toluenesulfonamide under the same conditions, α -(acetoxy)ethylbenzene was not formed at all (entries 8 and 11). This suggests that I₂ and *p*-toluenesulfonamide, respectively, work as catalysts in the present reaction. Instead of using DIB as the oxidant, [bis(trifluoroacetoxy)iodo]benzene was employed. However, [bis(trifluoroacetoxy)iodo]benzene did not work at all, whereas *p*-chloro[(diacetoxymethoxy)iodo]benzene and *p*-methyl[(diacetoxymethoxy)iodo]benzene worked to generate α -(acetoxy)ethylbenzene **2a** in moderate yields (entries 13–15). Oxone®, ¹BuOCl, and DIH (1,3-diiodo-5,5-dimethylhydantoin) also did not work at all as an oxidant, respectively (entries 16~18). When acetonitrile and carbon tetrachloride instead of 1,2-dichloroethane as solvent were used, the yield of α -(acetoxy)ethylbenzene was decreased to ~30% yield (entries 19 and 20). The same treatment of ethylbenzene with DIB, I₂, and *p*-toluenesulfonamide in 1,2-dichloroethane under irradiation conditions with a tungsten lamp provided α -(acetoxy)ethylbenzene in almost the same yield as that with warming conditions at 60 °C (entries 5 and 21). When [bis(benzoyloxy)iodo]benzene was used instead of DIB, α -(benzoyloxy)ethylbenzene **3a** was obtained in good yield (entry 22). These results indicate that the acetoxy group of α -(acetoxy)ethylbenzene **2a** comes from the acetoxy group of DIB. Based on these results, other

Table 1
 α -Acetoxylation of ethylbenzene with a DIB-I₂-TsNH₂ system

Entry	DIB (equiv)	I ₂ (equiv)	TsNH ₂ (equiv)	AcOH (mL)	Yield (%)		
					2a	2	3
1 ^a	2.0	0.2	0.2	1	25		
2 ^b	2.0	0.2	0.2	1	50		
3 ^b	1.5	0.2	0.2	1	42		
4	2.0	0.2	0.2	0	57		
5	2.5	0.2	0.2	0	63		
6	3.0	0.2	0.2	0	59		
7	2.5	0.1	0.2	0	47		
8	2.5	0	0.2	0	0		
9	2.5	0.2	0.3	0	61		
10	2.5	0.2	0.1	0	43		
11	2.5	0.2	0	0	0		
12	2.5	0.5	0.5	0	24		
13	2.5 ^c	0.2	0.2	1	0		
14	2.5 ^d	0.2	0.2	0	56		
15	2.5 ^e	0.2	0.2	0	50		
16	2.0 ^f	0.2	0.2	1	0		
17	2.0 ^g	0.2	0.2	1	0		
18	2.0 ^h	0.2	0.2	1	3		
19 ⁱ	2.5	0.2	0.2	0	33		
20 ^j	2.5	0.2	0.2	0	29		
21 ^k	2.5	0.2	0.2	0	61		
22	2.5 ^l	0.2	0.2	0	67 ^m		

^a The reaction time was 10 min.

^b The reaction time was 1 h.

^c Instead of DIB, [bis(trifluoroacetoxy)iodo]benzene was used.

^d Instead of DIB, *p*-chloro[(diacetoxymethoxy)iodo]benzene was used.

^e Instead of DIB, *p*-methyl[(diacetoxymethoxy)iodo]benzene was used.

^f Instead of DIB, oxone was used.

^g Instead of DIB, ¹BuOCl was used.

^h Instead of DIB, DIH was used.

ⁱ CH₃CN was used as solvent.

^j CCl₄ was used as solvent.

^k Irradiated with a tungsten lamp.

^l Instead of DIB, [bis(benzoyloxy)iodo]benzene was used.

^m Yield of **3a**.

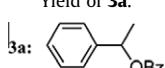
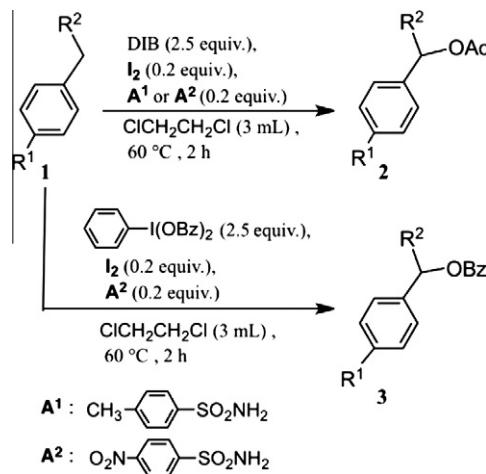
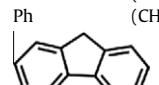
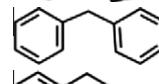
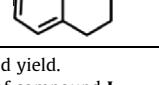


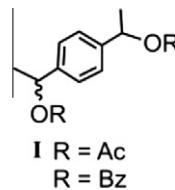
Table 2
 α -Acetoxylation of alkylbenzenes with a DIB-I₂-ArSO₂NH₂ system



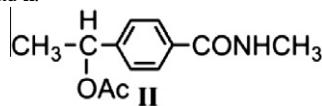
Entry	R ¹	R ²	Yield ^a (%)		
			2	3	
			A ¹	A ²	A ¹
1	H	CH ₃	63	73	67
2	Br	CH ₃	66	72	61
3	C(CH ₃) ₃	CH ₃	66	68	58
4	C ₂ H ₅	CH ₃	38 (44) ^b	29 (51) ^b	43 (27) ^b
5	CO ₂ CH ₃	CH ₃	48	59	37
6	NO ₂	CH ₃	27	35	29
7	CON(CH ₃) ₂	CH ₃	52 (25) ^c	28	35
8	H	CH ₂ CH ₃	42	54	39
9	H	(CH ₂) ₂ CH ₃	48	60	57
10	H	(CH ₂) ₂ OAc	27	31	45
11	H	(CH ₂) ₂ Br	36	42	30
12	H	(CH ₂) ₆ CH ₃	52	57	48
13	Ph	(CH ₂) ₃ CH ₃	64	81	61
14			61 (9) ^d	48 (13) ^d	51 (11) ^e
15			61	37	51
16			55 (1:0.7) ^f	63 (1:0.7) ^f	58 (1:0.8) ^g

^a Isolated yield.

^b Yield of compound **I**.



^c Yield of compound **II**.



^d Yield of compound **III**.

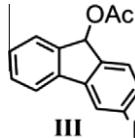
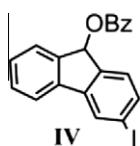
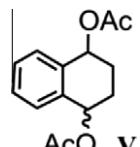
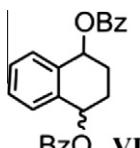


Table 2 (continued)^e Yield of compound IV.^f Yield of compound V.^g Yield of compound VI.

alkylbenzenes, such as 4-bromo-1-ethylbenzene, 4-*t*-butyl-1-ethylbenzene, methyl 4-ethylbenzoate, 4-ethyl-*N,N*-dimethylbenzamide, *n*-propylbenzene, *n*-butylbenzene, *n*-octylbenzene, and 4-pentyl-1-biphenyl were treated with DIB, I₂, and *p*-toluenesulfonamide at 60 °C to provide the corresponding α -(acetoxy)alkylbenzenes in good to moderate yields as shown in Table 2 (entries 2, 3, 5, 7–9, 12, and 13). When 1,4-diethylbenzene was used 1- α -(acetoxyethyl)-4-ethylbenzene was obtained in 38% yield together with 1,4-di(α -acetoxy)ethylbenzene in 44% yield (entry 4). The yields of α -acetoxy compounds were low when *p*-nitroethylbenzene was used (entry 6). Treatment of 4-ethyl-*N,N*-dimethylbenzamide with

DIB in the presence of I₂ and *p*-toluenesulfonamide generated 4-(α -acetoxy)ethyl-*N,N*-dimethylbenzamide in 52% yield together with 4-(α -acetoxy)ethyl-*N*-methylbenzamide in 25% yield, which was an N-demethylated compound (entry 7). The same treatment of fluorene and diphenylmethane with DIB, I₂, and *p*-toluenesulfonamide provided the corresponding α -acetoxy compounds in good to moderate yields (entries 14 and 15). When 1,2,3,4-tetrahydronaphthalene was used, a mixture of *cis*- and *trans*-1,4-diacetoxy-1,2,3,4-tetrahydronaphthalenes was obtained in 55% yield. Moreover, when *p*-nitrobenzenesulfonamide (**A**²), which may have stronger N–H bond energy than that of *p*-toluenesulfonamide due to the presence of an electron-withdrawing group, was used instead of *p*-toluenesulfonamide (**A**¹), the yield of α -acetoxylated compounds was slightly increased (from column **A**¹ to column **A**²). The same treatment of alkylbenzenes with [bis(benzoyloxy)iodo]benzene instead of DIB, generated the corresponding α -benzoyloxy compounds **3**. However, there is not so much difference in yield between DIB and [bis(benzoyloxy)iodo]benzene.

A plausible reaction mechanism is shown in Scheme 1. The initial formation of *N*-iodo-*p*-toluenesulfonamide from the reaction of *p*-toluenesulfonamide with DIB in the presence of I₂ occurs, and this is followed by homolytic N–I bond cleavage to generate a sulfonamidyl radical.

The sulfonamidyl radical abstracts a benzylic hydrogen atom from alkylbenzene to provide a benzyl radical that may be oxidized to benzylic cation with DIB. Once a benzylic cation is formed, it smoothly reacts with acetic acid to give α -(acetoxy)alkylbenzene. Practically, the addition of galvinoxyl free radical completely retarded the present reaction and ethylbenzene was obtained with high recovery under the same conditions.

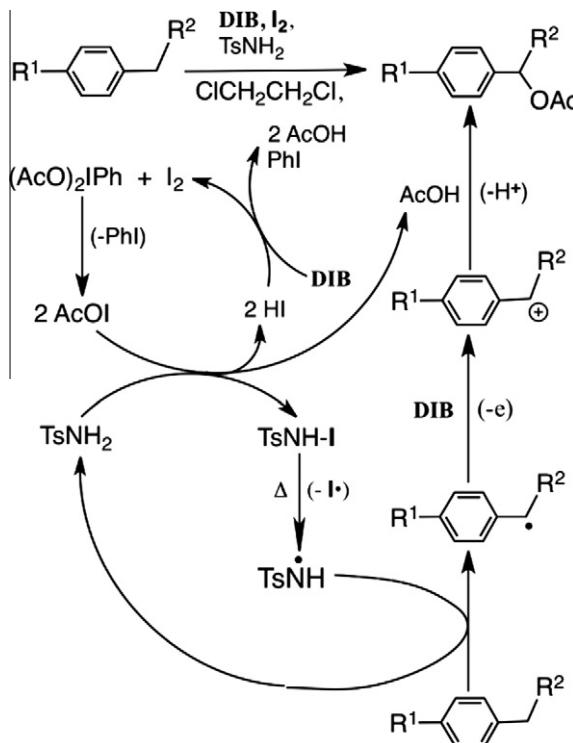
In conclusion, treatment of alkylbenzenes with DIB in the presence of catalytic amounts of I₂ and *p*-toluenesulfonamide at 60 °C gave the corresponding (α -acetoxy)alkylbenzenes in good to moderate yields. The present reaction is a simple method for the α -acetoxylation at the benzylic position of alkylbenzenes. Further synthetic studies using the present reaction are under way in this laboratory.

Acknowledgments

Financial support in the form of a Grant-in-Aid for Scientific Research (No. 20550033) from the Ministry of Education, Culture, Sports, Science, and Technology in Japan, Iodine Research Project in Chiba University, and Futaba Electronics Memorial Foundation is gratefully acknowledged.

References and notes

- Reviews: (a) Moriarty, R. M.; Vaid, R. K. *Synthesis* **1990**, 431; (b) Stang, P. J. *Angew. Chem., Int. Ed. Engl.* **1992**, 31, 274; (c) Prakash, O.; Saini, N.; Sharma, P. K. *Synlett* **1994**, 221; (d) Kitamura, T. *Yuki Gosei Kagaku Kyokaishi* **1995**, 53, 893; (e) Stang, P. J.; Zhdankin, V. V. *Chem. Rev.* **1996**, 96, 1123; (f) Umemoto, T. *Chem. Rev.* **1996**, 96, 1757; (g) Kita, Y.; Takada, T.; Tomha, H. *Pure Appl. Chem.* **1996**, 68, 627; (h) Togo, H.; Hoshina, Y.; Nogami, G.; Yokoyama, M. *Yuki Gosei Kagaku Kyokaishi* **1997**, 55, 90; (i) Varvoglou, A. *Tetrahedron* **1997**, 53, 1179; (j) Zhdankin, V. V. *Rev. Heteroatom Chem.* **1997**, 17, 133; (k) Muraki, T.; Togo, H.; Yokoyama, M. *Rev. Heteroatom Chem.* **1997**, 17, 213; (l) Kitamura, T.; Fujiwara, Y. *Org. Prep. Proced. Int.* **1997**, 29, 409; (m) Varvoglou, A.; Spyroudis, S. *Synlett* **1998**, 221; (n) Zhdankin, V. V.; Stang, P. J. *Tetrahedron* **1998**, 54, 10927; (o) Moriarty, R. M.; Prakash, O. *Adv. Heterocycl. Chem.* **1998**, 69, 1; (p) Togo, H.; Katohgi, M. *Synlett* **2001**, 565; (q) Zhdankin, V. V.; Stang, P. J. *Chem. Rev.* **2002**, 102, 2523; (r) Richardson, R. D.; Wirth, T. *Angew. Chem., Int. Ed.* **2006**, 45, 4402; (s) Ladziata, U.; Zhdankin, V. *Synlett* **2007**, 527; (t) Zhdankin, V. V.; Stang, P. J. *Chem. Rev.* **2008**, 108, 5299; (u) Merritt, E. A.; Olofsson, B. *Synthesis* **2011**, 517.
- Review: (a) Togo, H.; Katohgi, M. *Synlett* **2001**, 565; Papers: (b) de Armas, P.; Carrau, R.; Concepcion, J. I.; Francisco, C. G.; Hernandez, R.; Suarez, E. *Tetrahedron Lett.* **1985**, 26, 2493; (c) Freire, R.; Marrero, J. J.; Rodriguez, M. S.; Suarez, E. *Tetrahedron Lett.* **1986**, 27, 383; (d) Freire, R.; Hernandez, R.; Rodriguez, M. S.; Suarez, E.; Perales, P. *Tetrahedron Lett.* **1987**, 28, 981; (e) Francisco, C. G.; Freire, R.; Rodriguez, M. S.; Suarez, E. *Tetrahedron Lett.* **1987**, 28, 3397; (f) Carrau, R.; Hernandez, R.; Suarez, E.; Betancor, C. *J. Chem. Soc., Perkin*

Scheme 1. Plausible reaction mechanism.

- Trans.* **1** **1987**, 937; (g) Hernandez, R.; Medina, M. C.; Salazar, J. A.; Suarez, E. *Tetrahedron Lett.* **1987**, 28, 2533; (h) de Armas, P.; Francisco, C. G.; Hernandez, R.; Salazar, J. A.; Suarez, E. *J. Chem. Soc., Perkin Trans. 1* **1988**, 3255; (i) Hernandez, R.; Marrero, J. J.; Suarez, E. *Tetrahedron Lett.* **1988**, 29, 5979; (j) Hernandez, R.; Marrero, J. J.; Melian, D.; Suarez, E.; Melian, D. *Tetrahedron Lett.* **1988**, 29, 6661; (k) Hernandez, R.; Marrero, J. J.; Suarez, E. *Tetrahedron Lett.* **1989**, 30, 5501; (l) Dorta, R. L.; Francisco, C. G.; Suarez, E. *Chem. Commun.* **1989**, 1168; (m) Arencibia, M. T.; Freire, R.; Perales, A.; Rodriguez, M. S.; Suarez, E. *J. Chem. Soc., Perkin Trans. 1* **1991**, 3349; (n) Boto, A.; Betancor, C.; Prange, T.; Suarez, E. *Tetrahedron Lett.* **1992**, 33, 6687; (o) de Armas, P.; Francisco, C. G.; Suarez, E. *Angew. Chem., Int. Ed. Engl.* **1992**, 31, 772.
3. (a) Ogata, Y.; Aoki, K. *J. Am. Chem. Soc.* **1968**, 90, 6187; (b) Merkushev, E. B.; Smakhina, N. D.; Kovshenikova, G. M. *Synthesis* **1980**, 486.
4. (a) Togo, H.; Katohgi, M.; Yokoyama, M. *Synlett* **1998**, 131; (b) Togo, H.; Hoshina, Y.; Muraki, T.; Nakayama, H.; Yokoyama, M. *J. Org. Chem.* **1998**, 63, 5193; (c) Katohgi, M.; Togo, H.; Yamaguchi, K.; Yokoyama, M. *Tetrahedron* **1999**, 55, 14885; (d) Togo, H.; Nabana, T.; Yokoyama, M. *J. Org. Chem.* **2000**, 65, 8391; (e) Togo, H.; Harada, Y.; Yokoyama, M. *J. Org. Chem.* **2000**, 65, 926.
5. (a) Muraki, T.; Togo, H.; Yokoyama, M. *Tetrahedron Lett.* **1996**, 37, 2441; (b) Togo, H.; Muraki, T.; Hoshina, Y.; Yamaguchi, K.; Yokoyama, M. *J. Chem. Soc., Perkin Trans. 1* **1997**, 787.
6. Fan, R.; Li, W.; Pu, D.; Zhang, L. *Org. Lett.* **2009**, 11, 1425.
7. Baba, H.; Togo, H. *Tetrahedron Lett.* **2010**, 51, 2063.
8. (a) Belli, A.; Giordano, C.; Citterio, A. *Synthesis* **1980**, 447; (b) Citterio, A.; Santi, R.; Pagani, A. *J. Org. Chem.* **1987**, 52, 4925; (c) Ganin, E.; Amer, I. *J. Mol. Catal. A: Chem.* **1997**, 116, 323; (d) Minisci, F.; Recupero, F.; Gambarotti, C.; Punta, C.; Paganelli, R. *Tetrahedron Lett.* **2003**, 44, 6919; (e) Kumar, V.; Sharma, A.; Sharma, M.; Sharma, U. K.; Sinha, A. K. *Tetrahedron* **2007**, 63, 9718; (f) Shaikh, T. M.; Sudalai, A. *Tetrahedron Lett.* **2005**, 46, 5587; (g) Lee, J. M.; Park, E. J.; Cho, S. H.; Chang, S. J. *J. Am. Chem. Soc.* **2008**, 130, 7824; (h) Other methods with metals: *Comprehensive Organic Transformation*; Larock, R. C., Ed.; VCH Publishers, 1989; p 823.
9. (a) Dick, A. R.; Hull, K. L.; Sanford, M. S. *J. Am. Chem. Soc.* **2004**, 126, 2300; (b) Jiang, H.; Chen, H.; Wang, A.; Lin, X. *Chem. Commun.* **2010**, 46, 7259; (c) Zhang, S.; Luo, F.; Wang, W.; Jia, X.; Hu, M.; Cheng, J. *Tetrahedron Lett.* **2010**, 51, 3317.
10. Typical procedure for preparation of α -acetoxymethylbenzene with ethylbenzene, *DIB*, *I₂*, and *p*-toluenesulfonamide: (Diacetoxyiodo)benzene (5 mmol, 1.61 g), *I₂* (0.4 mmol, 102 mg), *p*-toluenesulfonamide (0.4 mmol, 68.4 mg), and ethylbenzene (2 mmol, 184 mg) were added to dichloroethane (3 mL). The mixture was warmed at 60 °C for 2 h under an argon atmosphere. Then, the mixture was poured into saturated aqueous sodium sulfite solution and extracted with diethyl ether (3 × 20 mL). The organic layer was dried over Na₂SO₄. After filtration and removal of the solvent under reduced pressure, the residue was subjected to preparative TLC on silica gel using a mixture of hexane and ethyl acetate (5:1) as an eluent to give α -acetoxymethylbenzene in 63% yield.
- (α -Acetoxymethylbenzene (*C₁₀H₁₂O₂*): IR (Neat): 1065, 1241, 1743 cm⁻¹; ¹H NMR (CDCl₃, TMS) δ = 1.53 (d, *J* = 6.6 Hz, 3H), 2.07 (s, 3H), 5.88 (q, *J* = 6.6 Hz, 1H), 7.26–7.36 (m, 5H); ¹³C NMR (CDCl₃, TMS) δ = 21.3, 22.1, 72.2, 126.0, 128.0, 128.4, 141.6, 170.2; HRMS (ESI) calcd for C₁₀H₁₂O₂Na [M+Na]⁺: *m/z* 187.0730, found *m/z* 187.0728.
- (α -Benzoyloxyethylbenzene (*C₁₅H₁₄O₂*): IR (Neat): 1109, 1267, 1713 cm⁻¹; ¹H NMR (CDCl₃, TMS) δ = 1.67 (d, *J* = 6.6 Hz, 3H), 6.13 (q, *J* = 6.6 Hz, 1H), 7.29 (t, *J* = 7.3 Hz, 1H), 7.36 (t, *J* = 7.5 Hz, 2H), 7.41–7.44 (m, 4H), 7.55 (t, *J* = 7.5 Hz, 1H), 8.07 (d, *J* = 7.5 Hz, 2H); ¹³C NMR (CDCl₃, TMS) δ = 22.4, 72.9, 126.0, 127.8, 128.3, 128.5, 129.6, 130.5, 132.9, 141.7, 165.7; HRMS (ESI) calcd for C₁₅H₁₄O₂Na [M+Na]⁺: *m/z* 249.0886, found *m/z* 249.0882.
- 4-(α -Acetoxymethyl-1-bromobenzene (*C₁₀H₁₁O₂Br*): IR (Neat): 599, 1071, 1241, 1735 cm⁻¹; ¹H NMR (CDCl₃, TMS) δ = 1.51 (d, *J* = 6.7 Hz, 3H), 2.06 (s, 3H), 5.82 (q, *J* = 6.7 Hz, 1H), 7.22 (d, *J* = 8.3 Hz, 2H), 7.47 (d, *J* = 8.3 Hz, 2H); ¹³C NMR (CDCl₃, TMS) δ = 21.2, 22.1, 71.6, 121.7, 127.8, 131.6, 140.7, 170.1; HRMS (APPI) calcd for C₁₀H₁₀O₂Br [M-H]⁺: *m/z* 240.9859, found *m/z* 240.9854.
- 4-(α -Benzoyloxyethyl-1-bromobenzene (*C₁₅H₁₃O₂Br*): IR (Neat): 711, 1109, 1270, 1718 cm⁻¹; ¹H NMR (CDCl₃, TMS) δ = 1.65 (d, *J* = 6.6 Hz, 3H), 6.07 (q, *J* = 6.6 Hz, 1H), 7.32 (d, *J* = 8.5 Hz, 2H), 7.44 (t, *J* = 7.5 Hz, 2H), 7.48 (d, *J* = 8.5 Hz, 2H), 7.56 (t, *J* = 7.5 Hz, 1H), 8.06 (d, *J* = 7.5 Hz, 2H); ¹³C NMR (CDCl₃, TMS) δ = 22.2, 72.2, 121.7, 127.7, 128.3, 129.6, 130.2, 131.6, 133.0, 140.8, 165.6; HRMS (APPI) calcd for C₁₅H₁₂O₂Br [M-H]⁺: *m/z* 303.0015, found *m/z* 303.0010.
- 1-(α -Acetoxymethyl-4-t-butylbenzene (*C₁₄H₂₀O₂*): IR (Neat): 1064, 1242, 1743 cm⁻¹; ¹H NMR (CDCl₃, TMS) δ = 1.33 (s, 9H), 1.53 (d, *J* = 6.6 Hz, 3H), 2.07 (s, 3H), 5.88 (q, *J* = 6.6 Hz, 1H), 7.28 (d, *J* = 8.6 Hz, 2H), 7.37 (d, *J* = 8.6 Hz, 2H); ¹³C NMR (CDCl₃, TMS) δ = 21.3, 21.9, 31.3, 34.4, 72.1, 125.3, 125.8, 138.5, 150.7, 170.3; HRMS (ESI) calcd for C₁₄H₂₀O₂Na [M+Na]⁺: *m/z* 243.1356, found *m/z* 243.1353.
- 1-(α -Benzoyloxyethyl-4-t-butylbenzene (*C₁₉H₂₂O₂*): IR (Neat): 1110, 1270, 1719 cm⁻¹; ¹H NMR (CDCl₃, TMS) δ = 1.31 (s, 9H), 1.66 (d, *J* = 6.6 Hz, 3H), 6.13 (q, *J* = 6.6 Hz, 1H), 7.36–7.39 (m, 4H), 7.42 (t, *J* = 7.4 Hz, 2H), 7.54 (t, *J* = 7.4 Hz, 1H), 8.07 (d, *J* = 7.4 Hz, 2H); ¹³C NMR (CDCl₃, TMS) δ = 22.2, 31.3, 34.5, 72.7, 125.4, 125.8, 128.3, 129.6, 130.6, 132.8, 138.6, 150.8, 165.8; HRMS (ESI) calcd for C₁₉H₂₂O₂Na [M+Na]⁺: *m/z* 305.1512, found *m/z* 305.1509.
- 1-(α -Acetoxymethyl-4-ethylbenzene (*C₁₂H₁₆O₂*): IR (Neat): 1064, 1240, 1738 cm⁻¹; ¹H NMR (CDCl₃, TMS) δ = 1.23 (t, *J* = 7.8 Hz, 3H), 1.53 (d, *J* = 6.6 Hz, 3H), 2.06 (s, 3H), 2.64 (q, *J* = 7.8 Hz, 2H), 5.86 (q, *J* = 6.6 Hz, 1H), 7.18 (d, *J* = 7.9 Hz, 2H), 7.27 (d, *J* = 7.9 Hz, 2H); ¹³C NMR (CDCl₃, TMS) δ = 15.4, 21.3, 22.0, 28.5, 72.1, 126.1, 127.9, 138.8, 143.9, 170.3; HRMS (ESI) calcd for C₁₂H₁₆O₂Na [M+Na]⁺: *m/z* 215.1043, found *m/z* 215.1040.
- 1-(α -Benzoyloxyethyl-4-ethylbenzene (*C₁₇H₁₈O₂*): IR (Neat): 1113, 1270, 1715 cm⁻¹; ¹H NMR (CDCl₃, TMS) δ = 1.24 (t, *J* = 7.7 Hz, 3H), 1.66 (d, *J* = 6.6 Hz, 3H), 2.64 (q, *J* = 7.7 Hz, 2H), 6.11 (q, *J* = 6.6 Hz, 1H), 7.20 (d, *J* = 8.1 Hz, 2H), 7.37 (d, *J* = 8.1 Hz, 2H), 7.42 (t, *J* = 7.5 Hz, 2H), 7.57 (t, *J* = 7.5 Hz, 1H), 8.07 (d, *J* = 7.5 Hz, 2H); ¹³C NMR (CDCl₃, TMS) δ = 15.4, 22.2, 28.5, 72.8, 126.1, 127.9, 128.2, 129.6, 130.5, 132.8, 138.9, 143.9, 165.7; HRMS (ESI) calcd for C₁₇H₁₈O₂Na [M+Na]⁺: *m/z* 277.1199, found *m/z* 277.1196.
- 1,4-Bis(α -acetoxyethyl)benzene (*C₁₄H₁₈O₄*): IR (Neat): 1065, 1240, 1739 cm⁻¹; ¹H NMR (CDCl₃, TMS) δ = 1.52 (d, *J* = 6.3 Hz, 6H), 2.07 (s, 6H), 5.87 (q, *J* = 6.3 Hz, 2H), 7.33 (s, 4H); ¹³C NMR (CDCl₃, TMS) δ = 21.3, 22.1, 72.0, 126.3, 141.3, 170.3; HRMS (ESI) calcd for C₁₄H₁₈O₄Na [M+Na]⁺: *m/z* 273.1097, found *m/z* 273.1093.
- 1,4-Bis(α -benzoyloxyethyl)benzene (*C₂₄H₂₂O₄*): IR (Neat): 1065, 1268, 1713 cm⁻¹; ¹H NMR (CDCl₃, TMS) δ = 1.66 (d, *J* = 6.6 Hz, 6H), 6.13 (q, *J* = 6.6 Hz, 2H), 7.43 (t, *J* = 7.3 Hz, 4H), 7.45 (s, 4H), 7.55 (t, *J* = 7.3 Hz, 2H), 8.07 (d, *J* = 7.3 Hz, 4H); ¹³C NMR (CDCl₃, TMS) δ = 22.3, 72.6, 126.3, 128.3, 129.6, 130.4, 132.9, 141.4, 165.8; HRMS (ESI) calcd for C₂₄H₂₂O₄Na [M+Na]⁺: *m/z* 397.1410, found *m/z* 397.1404.
- Methyl 4-(α -acetoxyethyl)benzoate (*C₁₂H₁₄O₄*): IR (Neat): 1067, 1236, 1719 cm⁻¹; ¹H NMR (CDCl₃, TMS) δ = 1.53 (d, *J* = 6.6 Hz, 3H), 2.09 (s, 3H), 3.91 (s, 3H), 5.90 (q, *J* = 6.6 Hz, 1H), 7.41 (d, *J* = 8.2 Hz, 2H), 8.02 (d, *J* = 8.2 Hz, 2H); ¹³C NMR (CDCl₃, TMS) δ = 21.1, 22.1, 52.0, 71.7, 125.8, 129.5, 129.8, 146.7, 166.6, 170.1; HRMS (ESI) calcd for C₁₂H₁₄O₄Na [M+Na]⁺: *m/z* 245.0784, found *m/z* 245.0781.
- Methyl 4-(α -benzoyloxyethyl)benzoate (*C₁₇H₁₆O₄*): IR (Neat): 1110, 1263, 1713 cm⁻¹; ¹H NMR (CDCl₃, TMS) δ = 1.68 (d, *J* = 6.5 Hz, 3H), 3.91 (s, 3H), 6.15 (q, *J* = 6.5 Hz, 1H), 7.45 (t, *J* = 7.5 Hz, 2H), 7.50 (d, *J* = 8.4 Hz, 2H), 7.57 (t, *J* = 7.5 Hz, 1H), 8.04 (d, *J* = 8.4 Hz, 2H), 8.08 (d, *J* = 7.5 Hz, 2H); ¹³C NMR (CDCl₃, TMS) δ = 22.3, 52.0, 72.3, 125.8, 128.3, 129.6, 130.0, 130.1, 133.0, 146.8, 165.6, 166.6; HRMS (ESI) calcd for C₁₇H₁₆O₄Na [M+Na]⁺: *m/z* 307.0941, found *m/z* 307.0934.
- 4-(α -Acetoxyethyl)-1-nitrobenzene (*C₁₀H₁₀O₄N*): IR (Neat): 1069, 1238, 1348, 1524, 1739 cm⁻¹; ¹H NMR (CDCl₃, TMS) δ = 1.55 (d, *J* = 6.7 Hz, 3H), 2.11 (s, 3H), 5.92 (q, *J* = 6.7 Hz, 1H), 7.51 (d, *J* = 8.7 Hz, 2H), 8.22 (d, *J* = 8.7 Hz, 2H); ¹³C NMR (CDCl₃, TMS) δ = 21.1, 22.2, 71.2, 123.8, 126.7, 147.4, 149.0, 170.0; HRMS (APPI) calcd for C₁₀H₁₀O₄N [M-H]⁺: *m/z* 208.0604, found *m/z* 208.0604.
- 4-(α -Benzoyloxyethyl)-1-nitrobenzene (*C₁₅H₁₂O₄N*): IR (Neat): 1069, 1263, 1713 cm⁻¹; ¹H NMR (CDCl₃, TMS) δ = 1.68 (d, *J* = 6.5 Hz, 3H), 3.91 (s, 3H), 6.15 (q, *J* = 6.5 Hz, 1H), 7.45 (t, *J* = 7.5 Hz, 2H), 7.50 (d, *J* = 8.4 Hz, 2H), 7.57 (t, *J* = 7.5 Hz, 1H), 8.04 (d, *J* = 8.4 Hz, 2H), 8.08 (d, *J* = 7.5 Hz, 2H); ¹³C NMR (CDCl₃, TMS) δ = 22.3, 52.0, 72.3, 125.8, 128.3, 129.6, 130.0, 130.1, 133.0, 146.8, 165.6, 166.6; HRMS (ESI) calcd for C₁₇H₁₆O₄Na [M+Na]⁺: *m/z* 307.0941, found *m/z* 307.0934.
- 4-(α -Acetoxyethyl)-1-nitrobenzene (*C₁₅H₁₂O₄N*): IR (Neat): 1069, 1238, 1348, 1522, 1719 cm⁻¹; ¹H NMR (CDCl₃, TMS) δ = 1.55 (d, *J* = 6.7 Hz, 3H), 2.11 (s, 3H), 5.92 (q, *J* = 6.7 Hz, 1H), 7.51 (d, *J* = 8.7 Hz, 2H), 8.22 (d, *J* = 8.7 Hz, 2H); ¹³C NMR (CDCl₃, TMS) δ = 21.1, 22.2, 71.2, 123.8, 126.7, 147.4, 149.0, 170.0; HRMS (APPI) calcd for C₁₅H₁₂O₄N [M-H]⁺: *m/z* 270.0761, found *m/z* 270.0754.
- 4-(α -Acetoxyethyl)-N,N-dimethylbenzamide (*C₁₃H₁₇O₃N₂*): IR (Neat): 1065, 1241, 1633, 1733, 3354, 3460 cm⁻¹; ¹H NMR (CDCl₃, TMS) δ = 1.53 (d, *J* = 6.6 Hz, 3H), 2.09 (s, 3H), 3.05 (br, 6H), 5.89 (q, *J* = 6.6 Hz, 1H), 7.39 (q, *J* = 8.5 Hz, 4H); ¹³C NMR (CDCl₃, TMS) δ = 21.3, 22.2, 35.3, 39.6, 72.4, 126.0, 127.3, 128.4, 129.6, 130.3, 133.0, 135.8, 143.2, 165.7, 171.2; HRMS (ESI) calcd for C₁₃H₁₇O₃NNa [M+Na]⁺: *m/z* 258.1101, found *m/z* 258.1095.
- 4-(α -Benzoyloxyethyl)-N,N-dimethylbenzamide (*C₁₈H₁₉O₃N*): IR (Neat): 1112, 1270, 1633, 1715, 3471 cm⁻¹; ¹H NMR (CDCl₃, TMS) δ = 1.67 (d, *J* = 6.6 Hz, 3H), 3.04 (br, 6H), 6.14 (q, *J* = 6.6 Hz, 1H), 7.41 (t, *J* = 7.1 Hz, 2H), 7.44 (t, *J* = 7.4 Hz, 4H), 7.56 (t, *J* = 7.1 Hz, 1H), 8.08 (d, *J* = 7.1 Hz, 2H); ¹³C NMR (CDCl₃, TMS) δ = 22.4, 35.3, 39.6, 72.4, 126.0, 127.3, 128.4, 129.6, 130.3, 133.0, 135.8, 143.2, 165.7, 171.2; HRMS (ESI) calcd for C₁₈H₁₉O₃NNa [M+Na]⁺: *m/z* 320.1257, found *m/z* 320.1261.
- (α -Acetoxypropylbenzene (*C₁₁H₁₄O₂*): IR (Neat): 1072, 1236, 1736 cm⁻¹; ¹H NMR (CDCl₃, TMS) δ = 0.87 (t, *J* = 7.4 Hz, 3H), 1.74–1.96 (m, 2H), 2.07 (s, 3H), 5.66 (t, *J* = 6.9 Hz, 1H), 7.27–7.36 (m, 5H); ¹³C NMR (CDCl₃, TMS) δ = 9.8, 21.2, 29.2, 77.3, 126.5, 127.8, 128.3, 140.5, 170.4; HRMS (ESI) calcd for C₁₁H₁₄O₂Na [M+Na]⁺: *m/z* 201.0886, found *m/z* 201.0887.
- (α -Benzoyloxypropylbenzene (*C₁₆H₁₆O₂*): IR (Neat): 1111, 1271, 1719 cm⁻¹; ¹H NMR (CDCl₃, TMS) δ = 0.97 (t, *J* = 7.4 Hz, 3H), 1.91–2.01 (m, 1H), 2.01–2.12 (m, 1H), 5.92 (t, *J* = 6.9 Hz, 1H), 7.28 (t, *J* = 7.3 Hz, 1H), 7.34 (t, *J* = 7.3 Hz, 2H), 7.42 (d, *J* = 7.3 Hz, 2H), 7.44 (t, *J* = 7.4 Hz, 2H), 7.56 (t, *J* = 7.4 Hz, 1H), 8.09 (d, *J* = 7.4 Hz, 2H); ¹³C NMR (CDCl₃, TMS) δ = 10.0, 29.5, 77.9, 126.4, 127.8, 128.3, 129.6, 130.5, 132.9, 140.6, 165.9; HRMS (ESI) calcd for C₁₆H₁₆O₂Na [M+Na]⁺: *m/z* 263.1043, found *m/z* 263.1043.
- (α -Acetoxybutylbenzene (*C₁₂H₁₆O₂*): IR (Neat): 1025, 1237, 1736 cm⁻¹; ¹H NMR (CDCl₃, TMS) δ = 0.92 (t, *J* = 7.4 Hz, 3H), 1.21–1.41 (m, 2H), 1.70–1.80 (m, 1H), 1.86–1.93 (m, 1H), 2.07 (s, 3H), 5.74 (t, *J* = 6.3 Hz, 1H), 7.25–7.36 (m, 5H); ¹³C NMR (CDCl₃, TMS) δ = 13.8, 18.8, 21.3, 38.4, 75.9, 126.5, 127.8, 128.4, 140.8, 170.4; HRMS (ESI) calcd for C₁₂H₁₆O₂Na [M+Na]⁺: *m/z* 215.1043, found *m/z* 215.1041.
- (α -Benzoyloxybutylbenzene (*C₁₇H₁₈O₂*): IR (Neat): 1111, 1272, 1718 cm⁻¹; ¹H NMR (CDCl₃, TMS) δ = 0.95 (t, *J* = 7.4 Hz, 3H), 1.31–1.51 (m, 2H), 1.83–1.92 (m, 1H), 2.02–2.11 (m, 1H), 5.99 (t, *J* = 6.1 Hz, 1H), 7.28 (t, *J* = 7.3 Hz, 1H), 7.35 (t, *J* = 7.3 Hz, 2H), 7.42 (d, *J* = 7.3 Hz, 2H), 7.44 (t, *J* = 7.5 Hz, 2H), 7.58 (t, *J* = 7.5 Hz, 1H), 8.08 (d, *J* = 7.5 Hz, 2H); ¹³C NMR (CDCl₃, TMS) δ = 13.8, 18.8, 21.3, 38.7, 75.9, 126.4, 127.8, 128.3, 129.6, 130.5, 132.9, 140.9, 165.8; HRMS (ESI) calcd for C₁₇H₁₈O₂Na [M+Na]⁺: *m/z* 277.1199, found *m/z* 277.1202.
- 1,3-Diacetoxy-1-phenylpropane (*C₁₃H₁₆O₄*): IR (Neat): 1043, 1236, 1743 cm⁻¹; ¹H NMR (CDCl₃, TMS) δ = 2.04 (s, 3H), 2.06 (s, 3H), 2.06–2.15 (m, 1H), 2.21–2.30 (m, 1H), 3.98–4.04 (m, 1H), 4.10–4.19 (m, 1H), 5.85 (dd, *J* = 8.3 and 5.7 Hz, 1H), 7.27–7.36 (m,

3-Acetoxy-1-benzoyloxy-1-phenylpropane ($C_{18}H_{18}O_4$): IR (Neat): 1111, 1272, 1720 cm^{-1} ; ^1H NMR (CDCl_3 , TMS) δ = 2.00 (s, 3H), 2.20–2.30 (m, 1H), 2.38–2.47 (m, 1H), 4.09–4.16 (m, 1H), 4.19–4.28 (m, 1H), 6.12 (dd, J = 8.1 and 5.4 Hz, 1H), 7.31 (t, J = 7.2 Hz, 1H), 7.36 (t, J = 7.2 Hz, 2H), 7.40–7.48 (m, 4H), 7.60 (t, J = 7.3 Hz, 1H), 8.07 (d, J = 7.3 Hz, 2H); ^{13}C NMR (CDCl_3 , TMS) δ = 20.9, 35.4, 60.8, 73.5, 126.3, 128.2, 128.4, 128.6, 129.6, 130.1, 133.1, 139.9, 165.6, 171.0; HRMS (ESI) calcd for $C_{18}H_{18}O_4\text{Na}$ [M+Na] $^+$: m/z 321.1097, found m/z 321.1092.

1-Acetoxy-4-bromo-1-phenylbutane ($C_{12}H_{15}O_2\text{Br}$): IR (Neat): 1026, 1235, 1735 cm^{-1} ; ^1H NMR (CDCl_3 , TMS) δ = 1.77–1.86 (m, 1H), 1.86–2.01 (m, 2H), 2.01–2.08 (m, 1H), 2.09 (s, 3H), 3.40 (t, J = 6.5 Hz, 2H), 5.76 (dd, J = 7.6 and 5.6 Hz, 1H), 7.28–7.39 (m, 5H); ^{13}C NMR (CDCl_3 , TMS) δ = 21.2, 28.7, 33.1, 34.8, 75.0, 126.4, 128.1, 128.5, 140.1, 170.3; HRMS(APPI) calcd for $C_{12}H_{14}O_2\text{Br}$ [M-H] $^+$: m/z 269.0172, found m/z 269.0196.

1-Benzoyloxy-4-bromo-1-phenylbutane ($C_{17}H_{17}O_2\text{Br}$): IR (Neat): 1109, 1271, 1719 cm^{-1} ; ^1H NMR (CDCl_3 , TMS) δ = 1.87–2.29 (m, 4H), 3.43 (t, J = 6.6 Hz, 2H), 6.03 (dd, J = 7.8 and 5.5 Hz, 1H), 7.30 (t, J = 7.0 Hz, 1H), 7.36 (t, J = 7.0 Hz, 2H), 7.42 (d, J = 7.0 Hz, 2H), 7.45 (t, J = 7.6 Hz, 2H), 7.57 (t, J = 7.6 Hz, 1H), 8.08 (d, J = 7.6 Hz, 2H); ^{13}C NMR (CDCl_3 , TMS) δ = 28.7, 33.2, 35.1, 75.6, 126.3, 128.1, 128.4, 128.6, 129.6, 130.2, 133.1, 140.2, 165.7; HRMS(APPI) calcd for $C_{17}H_{16}O_2\text{Br}$ [M-H] $^+$: m/z 331.0328, found m/z 331.0351.

(α -Acetoxy)octylbenzene ($C_{16}H_{24}O_2$): IR (Neat): 1022, 1237, 1738 cm^{-1} ; ^1H NMR (CDCl_3 , TMS) δ = 0.86 (t, J = 7.0 Hz, 3H), 1.15–1.38 (m, 10H), 1.71–1.79 (m, 1H), 1.84–1.94 (m, 1H), 2.06 (s, 3H), 5.72 (t, J = 6.5 Hz, 1H), 7.26–7.35 (m, 5H); ^{13}C NMR (CDCl_3 , TMS) δ = 14.0, 21.3, 22.6, 25.5, 29.1, 29.3, 31.7, 36.3, 76.1, 126.5, 127.8, 128.4, 140.8, 170.4; HRMS (ESI) calcd for $C_{16}H_{24}O_2\text{Na}$ [M+Na] $^+$: m/z 271.1669, found m/z 271.1666.

(α -Benzoyloxy)octylbenzene ($C_{21}H_{26}O_2$): IR (Neat): 1109, 1271, 1720 cm^{-1} ; ^1H NMR (CDCl_3 , TMS) δ = 0.86 (t, J = 6.9 Hz, 3H), 1.17–1.46 (m, 10H), 1.86–1.92 (m, 1H), 2.01–2.10 (m, 1H), 5.97 (t, J = 6.9 Hz, 1H), 7.29 (t, J = 7.2 Hz, 1H), 7.35 (t, J = 7.2 Hz, 2H), 7.42 (d, J = 7.2 Hz, 2H), 7.45 (t, J = 7.3 Hz, 2H), 7.56 (t, J = 7.3 Hz, 1H), 8.08 (d, J = 7.3 Hz, 2H); ^{13}C NMR (CDCl_3 , TMS) δ = 14.1, 22.6, 25.5, 29.1, 29.3, 31.8, 36.5, 76.7, 126.4, 127.8, 128.3, 128.4, 129.6, 130.5, 132.9, 140.9, 165.9; HRMS (ESI) calcd for $C_{21}H_{26}O_2\text{Na}$ [M+Na] $^+$: m/z 333.1825, found m/z 333.1819.

4-(α -Acetoxy)pentylbiphenyl ($C_{19}H_{22}O_2$): IR (Neat): 1108, 1271, 1719 cm^{-1} ; ^1H NMR (CDCl_3 , TMS) δ = 0.89 (t, J = 7.1 Hz, 3H), 1.20–1.30 (m, 1H), 1.30–1.39 (m, 3H), 1.76–1.84 (m, 1H), 1.90–1.98 (m, 1H), 2.09 (s, 3H), 5.76 (t, J = 7.0 Hz, 1H), 7.35 (t, J = 7.5 Hz, 1H), 7.40 (d, J = 7.5 Hz, 2H), 7.44 (t, J = 7.5 Hz, 2H), 7.58 (t, J = 7.9 Hz, 4H); ^{13}C NMR (CDCl_3 , TMS) δ = 13.9, 21.3, 22.4, 27.7, 36.0, 75.9, 127.0, 127.1, 127.2, 127.3, 128.7, 137.8, 139.8, 140.7, 170.4; HRMS (ESI) calcd for $C_{19}H_{22}O_2\text{Na}$ [M+Na] $^+$: m/z 305.1512, found m/z 305.1510.

4-(α -Benzoyloxy)pentylbiphenyl ($C_{24}H_{24}O_2$): IR (Neat): 1108, 1271, 1719 cm^{-1} ; ^1H NMR (CDCl_3 , TMS) δ = 0.91 (t, J = 7.1 Hz, 3H), 1.30–1.48 (m, 4H), 1.90–1.99 (m, 1H), 2.06–2.15 (m, 1H), 6.02 (t, J = 6.1 Hz, 1H), 7.34 (t, J = 7.3 Hz, 1H), 7.40–7.51 (m, 6H), 7.54–7.60 (m, 5H), 8.10 (d, J = 7.3 Hz, 2H); ^{13}C NMR (CDCl_3 , TMS) δ = 14.0, 22.5, 27.7, 36.3, 76.5, 126.9, 127.0, 127.1, 127.2, 127.3, 128.3, 128.7, 129.6, 130.5, 132.9, 140.0, 140.8, 165.9; HRMS (ESI) calcd for $C_{24}H_{24}O_2\text{Na}$ [M+Na] $^+$: m/z 367.1669, found m/z 367.1667.

9-Acetoxy-9H-fluorene ($C_{15}H_{12}O_2$): IR (Neat): 1024, 1234, 1737 cm^{-1} ; ^1H NMR (CDCl_3 , TMS) δ = 2.19 (s, 3H), 6.80 (s, 1H), 7.30 (t, J = 7.6 Hz, 2H), 7.41 (t, J = 7.6 Hz, 2H), 7.55 (d, J = 7.6 Hz, 2H), 7.67 (d, J = 7.6 Hz, 2H); ^{13}C NMR (CDCl_3 , TMS) δ = 21.2, 75.1, 120.1, 125.8, 127.8, 129.4, 141.0, 142.0, 171.7; HRMS (ESI) calcd for $C_{15}H_{12}O_2\text{Na}$ [M+Na] $^+$: m/z 240.0730, found m/z 247.0733.

9-Benzoyloxy-9H-fluorene ($C_{20}H_{14}O_2$): IR (Neat): 1262, 1451, 1724 cm^{-1} ; ^1H NMR (CDCl_3 , TMS) δ = 7.05 (s, 1H), 7.31 (t, J = 7.8 Hz, 2H), 7.41–7.47 (m, 4H), 7.57 (t, J = 7.4 Hz, 1H), 7.63 (d, J = 7.8 Hz, 2H), 7.71 (d, J = 7.8 Hz, 2H), 8.09 (d, J = 7.4 Hz, 2H); ^{13}C NMR (CDCl_3 , TMS) δ = 75.7, 120.2, 126.2, 128.0, 128.5, 128.6, 129.6, 130.1, 133.3, 141.2, 142.3, 167.4; HRMS (ESI) calcd for $C_{20}H_{14}O_2\text{Na}$ [M+Na] $^+$: m/z 309.0886, found m/z 309.0888.

Diphenylmethyl acetate ($C_{15}H_{14}O_2$): IR (Neat): 1023, 1232, 1741 cm^{-1} ; ^1H NMR (CDCl_3 , TMS) δ = 2.16 (s, 3H), 6.88 (s, 1H), 7.26–7.36 (m, 10H); ^{13}C NMR (CDCl_3 , TMS) δ = 21.3, 76.8, 127.0, 127.9, 128.5, 140.2, 170.0; HRMS (ESI) calcd for $C_{15}H_{14}O_2\text{Na}$ [M+Na] $^+$: m/z 249.0886, found m/z 249.0886.

Diphenylmethyl benzoate ($C_{20}H_{16}O_2$): IR (Neat): 1107, 1265, 1712 cm^{-1} ; ^1H NMR (CDCl_3 , TMS) δ = 7.12 (s, 1H), 7.29 (t, J = 7.3 Hz, 2H), 7.32–7.38 (m, 4H), 7.42–7.47 (m, 6H), 7.57 (t, J = 7.3 Hz, 1H), 8.14 (d, J = 7.3 Hz, 2H); ^{13}C NMR (CDCl_3 , TMS) δ = 77.3, 127.0, 127.8, 128.3, 128.4, 129.6, 129.9, 133.0, 140.2, 165.4; HRMS (ESI) calcd for $C_{20}H_{16}O_2\text{Na}$ [M+Na] $^+$: m/z 311.1043, found m/z 311.1041.