

# An Improved Protocol for Regioselective Ring-Opening of Sulfonyl Aziridines with Iodine Promoted by $\text{PPh}_3$

Jinfeng Zhang, Lingguo Meng,\* Chuntao Li, and Guoyuan Xiao

*Department of Chemistry, Huaipei Normal University, Huaipei, Anhui 235000, China*

A new route to synthesize  $\beta$ -iodo amines from sulfonyl aziridines and iodine was developed in the presence of  $\text{PPh}_3$ . This ring-opening reaction was an efficient and simple process to give  $\beta$ -iodo amines in excellent yields with high chemoselectivity.

**Keywords** sulfonyl aziridines, iodine, ring-opening,  $\beta$ -iodo amines,  $\text{PPh}_3$

## Introduction

Aziridines are one kind of the most important intermediates in organic synthesis and have served as key building blocks in the preparation of various nitrogen-containing molecules.<sup>[1]</sup> On the other hand, aziridines have covered many interesting reactions, such as ring expansion,<sup>[2]</sup> cyclization,<sup>[3]</sup> isomerization,<sup>[4]</sup> *N*-arylation,<sup>[5]</sup> etc. Apart from above reactions, the ring-opening of aziridines has been also studied intensively and numerous synthetic methods have been developed to realize the opening of aziridine rings.<sup>[6]</sup> Here, we focused on the preparation of  $\beta$ -iodo amines<sup>[7]</sup> by using sulfonyl aziridines as starting materials in the presence of  $\text{PPh}_3$  and iodine.

To our best knowledge, the usual preparative methods are regioselective ring-opening of aziridines with tetraalkylammonium halides by using the following reagents, such as  $\beta$ -cyclodextrin,<sup>[8a]</sup> ammonium-12-molybdophosphate<sup>[8b]</sup> and  $\text{BF}_3\cdot\text{OEt}_2$ ,<sup>[8c]</sup> as catalyst or promoter. On the other hand, the  $\beta$ -iodo amines also could be obtained through ring-opening of aziridines in the presence of  $\text{CeCl}_3$ ,<sup>[9a]</sup>  $\text{ZnI}_2$ ,<sup>[9b]</sup> and ionic liquid.<sup>[9c]</sup> Except above synthetic methods, Wu also reported thiophenol promoted efficient ring-opening of aziridines with iodine to synthesize  $\beta$ -iodo amines, but thiophenol is one of disagreeable, sick and harmful materials.<sup>[10]</sup> Herein, we describe an alternative methodology to synthesize  $\beta$ -iodo amines by iodination of aziridines using molecular iodine and  $\text{PPh}_3$ . It should be noted that better regioselectivity was observed than Wu's methods in some extent.

## Results and Discussion

Our studies were initiated by the addition of  $\text{PPh}_3$  to

a solution of 2-phenyl-1-tosylaziridine (**1a**) and iodine under various reaction conditions, and the results are shown in Table 1. Treatment of **1a** with 0.5 equiv. of iodine and 0.5 equiv. of  $\text{PPh}_3$  in  $\text{CH}_2\text{Cl}_2$  at room temperature in 1 min afforded *N*-(2-iodo-2-phenylethyl)-4-methylbenzenesulfonamide (**2a**) in 96% yield (Table 1, Entry 1). The yield of **2a** could not be further improved by increasing the amount of  $\text{PPh}_3$ , but decreased sharply when 0.2 equiv. of  $\text{PPh}_3$  was added in the reaction (Table 1, Entries 2 and 3). 35% yield of **2a** was obtained when 0.2 equiv. of iodine was used and the yield was failure to further increase when 1.0 equiv. of iodine was used. Solvent screening revealed a significant solvent effect. When shifting the solvent to  $\text{CHCl}_3$ , toluene, dioxane, or THF, the desired product **2a** was obtained in 85%, 81%, 75% or 70% yield, respectively (Table 1, Entries 6–9). With use of  $\text{CCl}_4$ ,  $\text{CH}_3\text{CN}$  or DMF as a solvent, product **2a** was isolated in low yields (Table 1, Entries 10–12). Only a trace amount of the **2a** was detected by TLC when DMSO was used as solvent (Table 1, Entry 13).

With the optimized reaction conditions in hand, a variety of substituted aziridines reacted with iodine smoothly to generate the corresponding ring-opening products in excellent yields with high chemoselectivity and the results were summarized in Table 2. Clearly, aziridines bearing the substitutes on the *para*-positions of benzene rings have no obvious effect on the yields of the reactions. Aziridines with electron-withdrawing groups, such as F, Cl, and Br on the *para*-positions of benzene rings reacted with iodine to give the corresponding products **2b**–**2d** in 95%–97% yields (Table 2, Entries 2–4). However, the reaction became complicated and no desired product was isolated when 2-(4-methoxyphenyl)-1-tosylaziridine was used as substrate, which might be ascribed to the unstable chemical

\* E-mail: milig@126.com; Tel.: 0086-0561-3802047

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**Table 1** Ring-opening of aziridine **1a** with iodine in the presence of  $\text{PPh}_3^a$ 

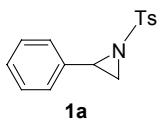
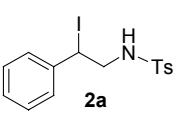
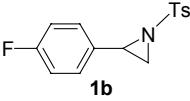
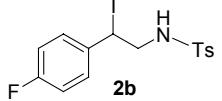
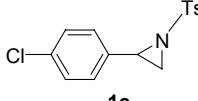
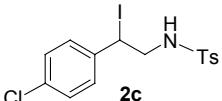
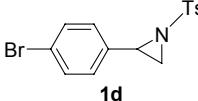
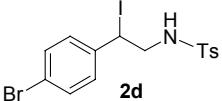
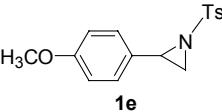
Entry	$\text{I}_2$ (equiv.)	$\text{PPh}_3$ (equiv.)	Solvent	Yield <sup>b</sup> /%	
				2a	
1	1/2	1/2	$\text{CH}_2\text{Cl}_2$	96	
2	1/2	1/5	$\text{CH}_2\text{Cl}_2$	48	
3	1/2	1.0	$\text{CH}_2\text{Cl}_2$	96	
4	1/5	1/2	$\text{CH}_2\text{Cl}_2$	35	
5	1.0	1/2	$\text{CH}_2\text{Cl}_2$	93	
6	1/2	1/2	$\text{CHCl}_3$	85	
7	1/2	1/2	Toluene	81	
8	1/2	1/2	Dioxane	75	
9	1/2	1/2	THF	70	
10	1/2	1/2	$\text{CCl}_4$	35	
11	1/2	1/2	$\text{CH}_3\text{CN}$	30	
12	1/2	1/2	DMF	20	
13	1/2	1/2	DMSO	trace	

<sup>a</sup> Reaction conditions: **1a** (0.2 mmol),  $\text{I}_2$  (amount indicated in Table 1),  $\text{PPh}_3$  (amount indicated in Table 1), solvent (2.0 mL), r.t., in air, <1 min. <sup>b</sup> Isolated yields.

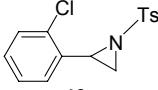
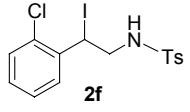
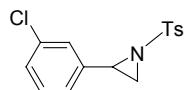
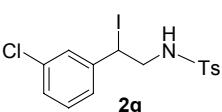
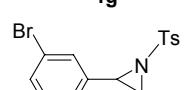
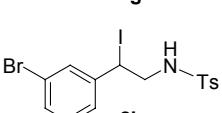
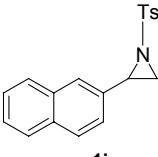
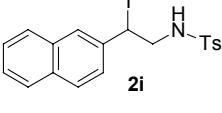
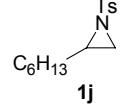
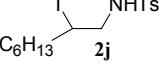
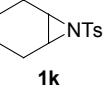
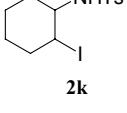
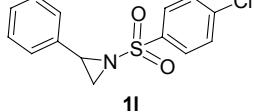
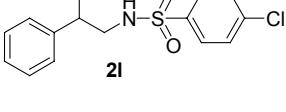
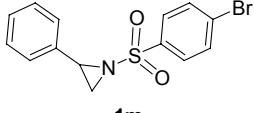
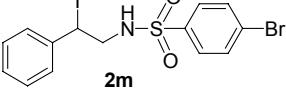
property of the desired product. The *ortho*-position effect of aziridine was not observed in the reaction of 2-(2-chlorophenyl)-1-tosylaziridine with iodine, which generated the product **2f** in 90% yield. 97% and 94% yields were obtained when 3-chlorophenyl- and 3-bromophenyl-1-tosylaziridine were proceeded in the reaction, respectively. Furthermore, 2-(naphthalen-2-yl)-1-tosylaziridine could be converted to desired products in 96% yield. Aliphatic aziridines, such as 2-hexyl-1-tosylaziridine and 7-tosyl-7-azabicyclo[4.1.0]heptane, also afforded corresponding ring-opening products in excellent yields, and it is important to note that only single ring-opening product was found and better chemoselectivity was observed compared to previous synthetic methods (Table 2, Entry 10).<sup>[10]</sup> On the other hand, the reactions of 4-bromophenylsulfonyl- and 4-chlorophenylsulfonyl-2-phenylaziridine with iodine also gave corresponding products in 95% and 93% yields, respectively.

To further evaluate the scope of the ring-opening reaction, epoxides were examined under the above optimized reaction conditions. It would be seen from Scheme 1, 7-oxabicyclo[4.1.0]heptane and 2-phenyl-oxirane reacted with iodine smoothly to generate iodo-hydrins products in 85% and 95% yields, respectively.

**Table 2**  $\text{PPh}_3$ -mediated ring-opening reaction of aziridines<sup>a</sup>

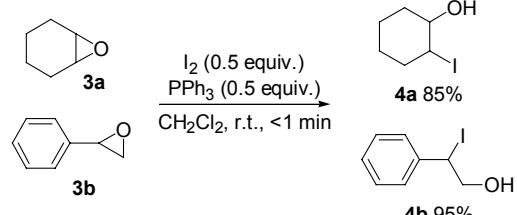
Entry	Aziridine, <b>1</b>	$\text{I}_2$ (0.5 equiv.)	$\text{PPh}_3$ (0.5 equiv.)	Product, <b>2</b>	Yield <sup>b</sup> /%	
					$\text{CH}_2\text{Cl}_2$ , r.t. <1 min	
1	 <b>1a</b>			 <b>2a</b>	96	
2	 <b>1b</b>			 <b>2b</b>	97	
3	 <b>1c</b>			 <b>2c</b>	96	
4	 <b>1d</b>			 <b>2d</b>	95	
5	 <b>1e</b>			<b>2e</b> was not detected		complicated

Continued

Entry	Aziridine, <b>1</b>	Product, <b>2</b>	Yield <sup>b</sup> /%
6			90
7			97
8			94
9			96
10			93
11			91
12			95
13			93

<sup>a</sup> **1a** (0.2 mmol), I<sub>2</sub> (0.1 mmol), PPh<sub>3</sub> (0.1 mmol), CH<sub>2</sub>Cl<sub>2</sub> (2.0 mL), r.t., in air, <1 min. <sup>b</sup> Isolated yield.

Scheme 1 PPh<sub>3</sub>-mediated ring-opening reaction of epoxides



## Conclusions

In conclusion, we have described a new route to synthesize  $\beta$ -iodo amines from sulfonyl aziridines and iodine in the presence of PPh<sub>3</sub> and these methods are also suitable for the ring-opening of epoxides. Com-

pared with previous reports, the synthetic method has some merits: (1) easier to manipulate; (2) without any metals; (3) better regioselectivity in some extent than Wu's methods.

## Experimental

All reactions were conducted in oven-dried glassware with magnetic stirring. Chromatographic purification was performed on silica gel (100–200 mesh) and analytical thin layer chromatography (TLC) on silica gel 60-F<sub>254</sub> (Qindao), which was detected by fluorescence. <sup>1</sup>H NMR (400 MHz) and <sup>13</sup>C NMR (100 MHz) spectra were measured with a Bruker AC 400 spectrometer with CDCl<sub>3</sub> or d<sub>6</sub>-DMSO as solvent and recorded relative to internal tetramethylsilane standard. High resolution mass spectra were obtained with a Micromass GCT-

TOF mass spectrometer. IR spectra were recorded as thin films or as solids in KBr pellets on a Perkin-Elmer FT210 spectrophotometer. Melting points were determined on a digital melting point apparatus and temperatures were uncorrected.

### General procedure for the preparation of $\beta$ -iodoamines or iodohydrins

To a solution of aziridines **1** or epoxide **3** (0.20 mmol) and iodine (0.10 mmol) in  $\text{CH}_2\text{Cl}_2$  (2.0 mL) was added  $\text{PPh}_3$  (0.10 mmol), and the reaction mixture was stirred at room temperature in 1 min. Then the mixture was purified directly by flash column chromatography on silica gel to afford the corresponding product.

**N-(2-Iodo-2-phenylethyl)-4-methylbenzenesulfonamide (2a)**<sup>[9a]</sup>  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$ : 7.72 (d,  $J=8.0$  Hz, 2H), 7.33 (d,  $J=8.0$  Hz, 2H), 7.26 (s, 5H), 5.04 (t,  $J=7.6$  Hz, 1H), 4.97 (t,  $J=6.4$  Hz, 1H), 3.73–3.66 (m, 1H), 3.55–3.48 (m, 1H), 2.45 (s, 3H);  $^{13}\text{C}$  NMR (75 MHz,  $\text{CDCl}_3$ )  $\delta$ : 143.8, 139.9, 137.0, 129.9, 129.0, 128.7, 127.5, 127.0, 51.2, 29.9, 21.6.

**N-(2-(4-Fluorophenyl)-2-iodoethyl)-4-methylbenzenesulfonamide (2b)** White solid 141–143 °C;  $^1\text{H}$  NMR (400 MHz,  $d_6\text{-DMSO}$ )  $\delta$ : 8.05 (t,  $J=5.6$  Hz, 1H), 7.65 (d,  $J=8.4$  Hz, 2H), 7.44–7.40 (m, 2H), 7.36 (d,  $J=8.4$  Hz, 2H), 7.13–7.08 (m, 2H), 5.20–5.16 (m, 1H), 3.49–3.36 (m, 2H), 2.36 (s, 3H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$ : 163.1 (d,  $J=243.8$  Hz), 143.2, 138.0, 137.8 (d,  $J=3.2$  Hz), 130.3 (d,  $J=8.4$  Hz), 130.1, 126.8, 115.9 (d,  $J=21.4$  Hz), 51.1, 30.5, 21.4; IR (KBr)  $\nu$ : 3280, 2925  $\text{cm}^{-1}$ ; HRMS (EI) calcd for  $\text{C}_{15}\text{H}_{15}\text{FNO}_2\text{S}$  ( $\text{M}^+ - \text{I}$ ): 292.0808; found 292.0803.

**N-(2-(4-Chlorophenyl)-2-iodoethyl)-4-methylbenzenesulfonamide (2c)**<sup>[9a]</sup>  $^1\text{H}$  NMR (400 MHz,  $d_6\text{-DMSO}$ )  $\delta$ : 8.05 (t,  $J=5.6$  Hz, 1H), 7.64 (d,  $J=8.4$  Hz, 2H), 7.40 (d,  $J=8.4$  Hz, 2H), 7.35–7.31 (m, 4H), 5.17–5.13 (m, 1H), 3.47–3.39 (m, 2H), 2.36 (s, 3H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$ : 143.2, 140.5, 138.0, 132.9, 130.1, 129.0, 126.8, 50.9, 29.9, 21.4.

**N-(2-(4-Bromophenyl)-2-iodoethyl)-4-methylbenzenesulfonamide (2d)**<sup>[8a]</sup>  $^1\text{H}$  NMR (400 MHz,  $d_6\text{-DMSO}$ )  $\delta$ : 8.05 (t,  $J=6.0$  Hz, 1H), 7.64 (d,  $J=8.4$  Hz, 2H), 7.47 (d,  $J=8.4$  Hz, 2H), 7.39–7.31 (m, 4H), 5.17–5.13 (m, 1H), 3.50–3.38 (m, 2H), 2.36 (s, 3H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$ : 143.2, 140.9, 138.0, 131.9, 130.3, 130.1, 126.8, 121.5, 50.8, 29.9, 21.4.

**N-(2-(2-Chlorophenyl)-2-iodoethyl)-4-methylbenzenesulfonamide (2f)** Colorless oil.  $^1\text{H}$  NMR (400 MHz,  $d_6\text{-DMSO}$ )  $\delta$ : 8.14 (t,  $J=6.0$  Hz, 1H), 7.67 (d,  $J=8.0$  Hz, 2H), 7.60 (dd,  $J=8.4, 1.2$  Hz, 1H), 7.41–7.28 (m, 5H), 5.42–5.38 (m, 1H), 3.55–3.51 (m, 2H), 2.36 (s, 3H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$ : 143.3, 138.4, 137.9, 132.6, 130.2, 130.1, 129.1, 128.3, 126.8, 49.6, 26.0, 21.4; IR (neat)  $\nu$ : 3285, 2920  $\text{cm}^{-1}$ ; HRMS (EI) calcd for  $\text{C}_{15}\text{H}_{15}^{35}\text{ClNO}_2\text{S}$  ( $\text{M}^+ - \text{I}$ ): 308.0512; found 308.0525.

**N-(2-(3-Chlorophenyl)-2-iodoethyl)-4-methylbenzenesulfonamide (2g)** White solid 124–126 °C;  $^1\text{H}$

NMR (400 MHz,  $d_6\text{-DMSO}$ )  $\delta$ : 8.09 (t,  $J=6.0$  Hz, 1H), 7.65 (d,  $J=8.0$  Hz, 2H), 7.43 (s, 1H), 7.36–7.16 (m, 5H), 5.16–5.13 (m, 1H), 3.49–3.45 (m, 2H), 2.36 (s, 3H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$ : 143.8, 143.2, 138.0, 133.5, 130.8, 130.1, 128.4, 128.0, 127.0, 126.8, 50.6, 29.4, 21.4; IR (KBr)  $\nu$ : 3287, 2925  $\text{cm}^{-1}$ ; HRMS (ESI) calcd for  $\text{C}_{15}\text{H}_{16}^{35}\text{ClNO}_2\text{S}$  ( $\text{M}^+ + \text{H}$ ): 435.9635; found 435.9631.

**N-(2-(3-Bromophenyl)-2-iodoethyl)-4-methylbenzenesulfonamide (2h)** White solid 122–125 °C;  $^1\text{H}$  NMR (400 MHz,  $d_6\text{-DMSO}$ )  $\delta$ : 8.09 (t,  $J=6.0$  Hz, 1H), 7.65 (d,  $J=8.0$  Hz, 2H), 7.55 (s, 1H), 7.43–7.34 (m, 4H), 7.24 (d,  $J=8.0$  Hz, 1H), 5.15–5.11 (m, 1H), 3.48–3.45 (m, 2H), 2.36 (s, 3H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$ : 144.1, 143.2, 138.0, 131.3, 131.1, 130.8, 130.1, 127.4, 126.8, 122.0, 50.7, 29.4, 21.4; IR (KBr)  $\nu$ : 3287, 2925  $\text{cm}^{-1}$ ; HRMS (ESI) calcd for  $\text{C}_{15}\text{H}_{16}^{79}\text{BrNO}_2\text{S}$  ( $\text{M}^+ + \text{H}$ ): 479.9130; found 479.9126.

**N-(2-Iodo-2-(naphthalen-2-yl)ethyl)-4-methylbenzenesulfonamide (2i)** White solid 141–142 °C;  $^1\text{H}$  NMR (400 MHz,  $d_6\text{-DMSO}$ )  $\delta$ : 8.12 (t,  $J=6.0$  Hz, 1H), 7.86–7.84 (m, 4H), 7.66 (d,  $J=8.0$  Hz, 2H), 7.51–7.49 (m, 3H), 7.32 (d,  $J=8.0$  Hz, 2H), 5.38–5.34 (m, 1H), 3.62–3.55 (m, 2H), 2.33 (s, 3H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$ : 143.2, 138.7, 138.1, 133.0, 130.0, 128.8, 128.3, 127.9, 126.9, 126.9, 126.8, 126.5, 126.3, 50.9, 32.3, 21.4; IR (KBr)  $\nu$ : 3278, 2930  $\text{cm}^{-1}$ ; HRMS (EI) calcd for  $\text{C}_{19}\text{H}_{18}\text{NO}_2\text{S}$  ( $\text{M}^+ - \text{I}$ ): 324.1058; found 324.1054.

**N-(2-Iodoctyl)-4-methylbenzenesulfonamide (2j)**<sup>[11]</sup>  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$ : 7.79 (d,  $J=8.0$  Hz, 2H), 7.31 (d,  $J=8.0$  Hz, 2H), 5.00 (d,  $J=8.0$  Hz, 1H), 3.26–3.15 (m, 2H), 2.98–2.94 (m, 2H), 2.42 (s, 3H), 1.49–1.34 (m, 2H), 1.25–1.05 (m, 8H), 0.85 (t,  $J=7.2$  Hz, 3H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$ : 143.5, 137.8, 129.7, 127.0, 52.7, 35.3, 31.5, 28.6, 25.1, 22.4, 21.5, 14.4, 14.0.

**N-(2-Iodocyclohexyl)-4-methylbenzenesulfonamide (2k)**<sup>[9a]</sup>  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$ : 7.81 (d,  $J=8.0$  Hz, 2H), 7.32 (d,  $J=8.0$  Hz, 2H), 5.26 (d,  $J=6.4$  Hz, 1H), 4.02–3.97 (m, 1H), 3.31–3.27 (m, 1H), 2.42 (s, 3H), 2.35–2.33 (m, 1H), 2.19–2.16 (m, 1H), 1.99–1.89 (m, 1H), 1.68–1.66 (m, 1H), 1.53–1.50 (m, 1H), 1.41–1.22 (m, 3H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$ : 143.5, 137.4, 129.6, 127.3, 59.1, 34.9, 32.9, 29.6, 26.5, 21.5.

**4-Chloro-N-(2-iodo-2-phenylethyl)benzenesulfonamide (2l)** White solid 151–153 °C;  $^1\text{H}$  NMR (400 MHz,  $d_6\text{-DMSO}$ )  $\delta$ : 8.26 (t,  $J=6.0$  Hz, 1H), 7.78 (d,  $J=8.8$  Hz, 2H), 7.62 (d,  $J=8.8$  Hz, 2H), 7.38 (d,  $J=7.2$  Hz, 2H), 7.30–7.21 (m, 3H), 5.19–5.15 (m, 1H), 3.55–3.40 (m, 2H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$ : 141.3, 139.8, 137.8, 129.8, 129.0, 128.8, 128.6, 128.1, 50.9, 31.7; IR (KBr)  $\nu$ : 3291, 2923  $\text{cm}^{-1}$ ; HRMS (EI) calcd for  $\text{C}_{14}\text{H}_{13}^{35}\text{ClNO}_2\text{S}$  ( $\text{M}^+ - \text{I}$ ): 294.0356; found 294.0354.

**4-Bromo-N-(2-iodo-2-phenylethyl)benzenesulfonamide (2m)** White solid 150–152 °C;  $^1\text{H}$  NMR (400

MHz,  $d_6$ -DMSO)  $\delta$ : 8.26 (t,  $J=6.0$  Hz, 1H), 7.77 (d,  $J=8.8$  Hz, 2H), 7.70 (d,  $J=8.8$  Hz, 2H), 7.38 (d,  $J=7.2$  Hz, 2H), 7.30–7.23 (m, 3H), 5.18–5.14 (m, 1H), 3.55–3.39 (m, 2H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$ : 141.3, 140.2, 132.7, 129.0, 128.9, 128.6, 128.1, 126.7, 50.9, 31.7; IR (KBr)  $\nu$ : 3290, 2920  $\text{cm}^{-1}$ ; HRMS (EI) calcd for  $\text{C}_{14}\text{H}_{13}{^{79}\text{BrNO}_2\text{S}}$  ( $\text{M}^+ - \text{I}$ ) 337.9850; found 337.9844.

**2-Iodocyclohexanol (4a)**<sup>[10]</sup>  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$ : 4.06–4.00 (m, 1H), 3.68–3.62 (m, 1H), 2.48–2.42 (m, 2H), 2.12–1.98 (m, 2H), 1.85–1.82 (m, 1H), 1.44–1.20 (m, 3H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$ : 75.8, 43.3, 38.4, 33.5, 27.8, 24.3.

**2-Iodo-2-phenylethanol (4b)**<sup>[10]</sup>  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$ : 7.44 (d,  $J=7.2$  Hz, 2H), 7.32–7.27 (m, 3H), 5.21–5.18 (m, 1H), 4.11–4.06 (m, 1H), 2.21 (s, 1H);  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$ : 140.0, 128.9, 128.5, 127.9, 68.6, 35.6.

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(Pan, B.; Qin, X.)