Accepted Manuscript

A three-step total synthesis of goniothalesdiol A using a one-pot Sharpless epoxidation/regioselective epoxide ring-opening

Perla Ramesh, Yarram Narasimha Reddy

PII:	S0040-4039(17)30132-6
DOI:	http://dx.doi.org/10.1016/j.tetlet.2017.01.100
Reference:	TETL 48603
To appear in:	Tetrahedron Letters
Received Date:	3 January 2017
Revised Date:	21 January 2017
Accepted Date:	25 January 2017



Please cite this article as: Ramesh, P., Reddy, Y.N., A three-step total synthesis of goniothalesdiol A using a onepot Sharpless epoxidation/regioselective epoxide ring-opening, *Tetrahedron Letters* (2017), doi: http://dx.doi.org/ 10.1016/j.tetlet.2017.01.100

This is a PDF file of an unedited manuscript that has been accepted for publication. As a service to our customers we are providing this early version of the manuscript. The manuscript will undergo copyediting, typesetting, and review of the resulting proof before it is published in its final form. Please note that during the production process errors may be discovered which could affect the content, and all legal disclaimers that apply to the journal pertain.

ACCEPTED MANUSCRIPT

Graphical Abstract

A three-step total synthesis of goniothalesdiol A using a one-pot Sharpless	Leave this area blank for abstract info.
epoxidation/regioselective epoxide ring- opening	
Perla Ramesh, Yarram Narasimha Reddy	
$Ph \longrightarrow O \xrightarrow{4 \text{ Steps}} Ph \xrightarrow{OH} OH \xrightarrow{OH} OCH_3 \xrightarrow{HO} Ph \xrightarrow{HO} Ph \xrightarrow{HO} OCH_3 \xrightarrow{HO} Ph \xrightarrow{HO} Ph \xrightarrow{HO} OCH_3 \xrightarrow{HO} Ph \xrightarrow{HO} $	
$Ph \longrightarrow O \xrightarrow{3 \text{ Steps}} Ph \longrightarrow OH OH O \xrightarrow{Ph} OCH_3 \xrightarrow{Ph}$	OCH3 * Only three steps * Highly efficient * Protecting Group-Free * High overall yield 55.1%



Tetrahedron Letters

journal homepage: www.elsevier.com

A three-step total synthesis of goniothalesdiol A using a one-pot Sharpless epoxidation/regioselective epoxide ring-opening

Perla Ramesh* and Yarram Narasimha Reddy*

Natural Products Chemistry Division, Indian Institute of Chemical Technology, Uppal Road, Hyderabad-500007, India

ARTICLE INFO

ABSTRACT

Article history: Received Received in revised form Accepted Available online

Keywords: Goniothalesdiol A Sharpless epoxidation Asymmetric synthesis Protecting-group-free Tetrahydropyran

1. Introduction

The tetrahydropyran backbone is a ubiquitous heterocyclic unit found in numerous bioactive natural products and pharmaceutically active compounds. Studies on natural products isolated from several species of the genus *Goniothalamus* have led to the discovery of various secondary metabolites including acetogenins, styryllactones, alkaloids, and flavonoids with interesting biological properties, such as pesticidal, teratogenic and cytotoxic.¹ Goniothalesdiol A (1) (Figure 1), a new natural 3,4-dihydroxy 2,6-disubstituted tetrahydropyran derivative, was isolated in 2006 from the stem of a southern Taiwan tree *Goniothalamus amuyon* by Wu and co-workers. The gross structure and relative stereochemistry of 1 were assigned on the basis of

¹H NMR, ¹³C NMR and HRMS experiments, and the absolute configuration was confirmed by biosynthesis.²

To date, there have been six reports describing elegant synthetic efforts toward this natural product. In 2009, Yadav et al.³ reported the first total synthesis of goniothalesdiol A (1) from homopropargyl alcohol in 11 steps with 35.1% overall yield by employing a Sharpless kinetic resolution and intramolecular oxa-Michael addition for the construction of the pyran ring. More recently, the same author has reported the synthesis of 1 that proceeded in nine steps using an intramolecular oxa-Michael reaction as a key step.⁴ Fadnavis et al.⁵ achieved the asymmetric synthesis of 1 from (*R*)-2,3-*O*-cyclohexylidine glyceraldehyde in 11 steps with an overall yield of 22%. Protecting-group-free total synthesis of 1 was accomplished by She et al. in six steps with

Using a one-pot Sharpless asymmetric epoxidation/regioselective epoxide ring-opening as a key step, the protecting-group-free total synthesis of goniothalesdiol A was accomplished in only three steps starting from commercially available *trans*-cinnamaldehyde in 55.1% overall yield. In 6 previously reported total syntheses, 6-11 steps were required.

2017 Elsevier Ltd. All rights reserved.

30% overall yield.⁶ Recently, Sabitha and co-workers⁷ reported the application of the Tandem α -aminoxylation–allylation reaction, resulting in the concise synthesis of **1**. Quite recently,⁸ Rakeshwar has described the last synthesis of **1** in six steps with 31% overall yield, starting from the commercially available 2-deoxy-D-ribose. Most of these syntheses rely on the use of protecting groups and as a consequence are lengthy sequences. As part of our ongoing projects focused on improving the efficiency of natural products synthesis,⁹ herein we describe the asymmetric total synthesis of goniothalesdiol A (**1**) by a unified strategy that requires no use of protecting groups.



Figure 1. The structure of goniothalesdiol A (1).

2. Results and discussion

Our first retrosynthetic analysis for the synthesis of 1 is shown in Scheme 1. Goniothalesdiol A (1) would be obtained from triol 2 through an intramolecular oxa-Michael addition of the hydroxyl group present at the active benzylic position. Triol 2 would be built by using a Sharpless asymmetric epoxidation applied to compound 3, followed by a regioselective epoxide ring opening with water to control the stereogenic centers at C2 and C3. Finally, the α,β -unsaturated ester 3 would be synthesized

* Corresponding authors. Tel.: +91-40-27191631; fax: +91-40-27160512; e-mails: perlaramesh@yahoo.co.in (P. Ramesh); nyarram20@gmail.com (Y. Narasimha Reddy)

ACCEPTED MANUSCRIP

Tetrahedron Letters

from commercially available *trans*-cinnamaldehyde **4** using an asymmetric allylation followed by an olefin cross-metathesis.





The synthesis of goniothalesdiol A (1) started with the known homoallylic alcohol 5,^{9d,e} which is readily available in high enantiomeric excess from (E)-cinnamaldehyde by an asymmetric allylation. Chemoselective olefin cross-metathesis (OCM)¹⁰ of compound 5 with methyl acrylate in the presence of secondgeneration Grubbs catalyst provided the desired (E)- α , β unsaturated ester 3 in 91% yield as a single stereoisomer (judged by ¹H NMR analysis). Sharpless asymmetric epoxidation¹¹ of allylic alcohol 3 using (+)-diethyl tartrate as a chiral ligand proceeded efficiently to produce epoxy alcohol 6 in 90% yield. Regioselective epoxide ring opening of epoxy alcohol 6 using hot water under the catalyst-free condition at 60 °C provided triol 2,^{12,9e} which was then subjected to intramolecular oxa-Michael addition using the acid catalyst. At this stage, we anticipated that a desired THP product 1 would be obtained from triol 2 through an intramolecular 1,4-addition of the hydroxyl group present at the active benzylic position on the unsaturated ester. Unfortunately, when compound 2 was treated with TBAF in dry THF at 0 °C, an excellent yield in the unexpected intramolecular 1,4-addition product trans-THF 7 was obtained with 97:3 diastereomeric ratio.¹³ The 2,5-trans configuration of the tetrahydrofuran ring in undesired product 2 was confirmed by NOE data (Scheme 2).

Due to the first strategy failure, a second strategy was then envisaged, utilizing asymmetric reactions in order to reduce the number of steps. In this strategy (Scheme 3), the required *cis*-THP **1** would be achieved from **8** by using an intramolecular regioselective epoxide ring opening at the active benzylic position. Compound **8** would be emerged from **9** by a Sharpless asymmetric epoxidation, while **9** would be derived from commercially available *trans*-cinnamaldehyde by the successful implementation of asymmetric aldol addition followed by 1,3-*syn* reduction.



Scheme 3. Second retrosynthetic analysis for 1.

As delineated in scheme 4, the synthesis of 1 commenced with methyl(S,E)-5-hydroxy-3-oxo-7-phenylhept-6-enoate (11), which is a known intermediate easily available from transcinnamaldehyde by an asymmetric Mukaiyama aldol reaction.¹⁴ As we reported previously,^{14a} the stereoselective syn reduction of **11** under Narasaka's conditions¹⁵ using Et₂BOMe/NaBH₄ in THF-MeOH (4:1) at -78 °C gave syn-diol 9, which was then subjected to Sharpless asymmetric epoxidation using (+)-diethyl tartrate as a chiral ligand. Surprisingly, under Sharpless conditions, asymmetric epoxidation (formation of 8) and concomitant cyclization via intramolecular regioselective epoxide ring opening proceeded smoothly to form a tetrahydropyran ring, and the desired cis-THP product (1) was obtained in 84% yield for a single step. Goniothalesdiol A (1) was identical in all respect with the spectral data (¹H NMR, ¹³C NMR and HRMS) and the specific rotation reported previously.²⁻ ⁸ It is noteworthy that this three-step synthesis of **1** compares well with those reported in the literature that ranges in length from 6 to 11 steps.

ACCEPTED MANUSCRIPT



3. Conclusions

In summary, two approaches of the goniothalesdiol A (1) were examined. One of them failed in the formation of the *cis*-THP ring, utilizing an intramolecular oxa-Michael addition of active benzylic hydroxyl group on an unsaturated ester. On the contrary, the second approach allowed us to minimize the number of steps by using a one-pot Sharpless epoxidation/regioselective epoxide ring-opening strategy. This highly concise synthesis relied on the use of a key one-pot Sharpless epoxidation/regioselective epoxide ring-opening reaction to rapidly construct the unique sixmembered *cis*-tetrahydropyran ring and two continuous stereocenters in a single step. In contrast to previous approaches,²⁻⁸ this protocol is atom economical, step-economical and devoid of any protecting group manipulations. Our route to goniothalesdiol A (1) is significantly shorter than the previous approaches, requiring only three steps instead of 6-11 steps.

Acknowledgments

We thank CSIR New Delhi for financial support under XII Five Year Plan CSC0108-ORIGIN.

References and notes

- (a) Zeng, L.; Ye, Q.; Oberlies, N. H.; Shi, G.; Gu, Z. M.; He, K.; Mc Laughlin, J. L. Nat. Prod. Rep. **1996**, *13*, 275-306; (b) Din, L. B.; Colegate, S. M.; Razak, D. A. Phytochemistry **1990**, *29*, 346-348; (c) Omar, S.; Chee, C. L.; Ahmad, F.; Ni, J. X.; Jaber, H.; Huang, J.; Nakatsu, T. Phytochemistry **1992**, *31*, 4395-4397; (d) Sam, T. W.; Chew, S. Y.; Matsjeh, S.; Gan, E. K.; Razak, D.; Mohamed, A. L. Tetrahedron Lett. **1987**, *28*, 2541-2544; (e) Fang, X.-P.; Anderson, J. E.; Chang, C. J.; McLauglin, J. L. Tetrahedron **1991**, *4*, 9751-9758.
- Lan, Y.-H.; Chang, F.-R.; Yang, Y.-L.; Wu, Y.-C. Chem. Pharm. Bull. 2006, 54, 1040–1043.
- 3. Yadav, J. S.; Rami Reddy, N.; Harikrishna, V.; Subba Reddy, B. V. *Tetrahedron Lett.* **2009**, *50*, 1318–1320.

- Yadav, J. S.; Nageshwar Rao, R.; Somaiah, R.; Harikrishna, V.; Subba Reddy, B. V. *Helv. Chim. Acta* 2010, 93, 1362–1368.
- 5. Venkataiah, M.; Somaiah, P.; Reddipalli, G.; Fadnavis, N. W. *Tetrahedron: Asymmetry* **2009**, *20*, 2230–2233.
- Li, J.; Zheng, H.; Su, Y.; Xie, X.; She, X. Synlett 2010, 2283– 2284.
- Sabitha, G.; Rammohan Reddy, T.; Yadav, J. S. *Tetrahedron Lett.* 2011, *52*, 6550–6553.
- Pallavi, T.; Kumaraswamy, B.; Raji, R. G.; Rakeshwar B.; Khagga M. *Tetrahedron: Asymmetry* 2012, 23, 659–661.
- (a) Ramesh, P.; Raju, A.; Fadnavis, N. W. Tetrahedron: Asymmetry 2015, 26, 1251-1255; (b) Ramesh, P. Synthesis 2016, 48, 4300-4304; (c) Ramesh, P.; Anjibabu, R.; Raju, A. Tetrahedron Lett. 2016, 57, 2100-2102; (d) Ramesh, P. ChemistrySelect, 2016, 1, 3244-3246; (e) Ramesh, P.; Rao, T. P. J. Nat. Prod. 2016, 79, 2060-2065; (f) Ramesh, P; Reddy, Y. N; Reddy, T. N; Srinivasu, N. Tetrahedron: Asymmetry 2017, DOI: org/10.1016/j.tetasy.2017.01.005.
- 10. Connon, S. J.; Blechert, S. Angew. Chem., Int. Ed. 2003, 42, 1900–1923.
- Martin, V. S.; Woodard, S. S.; Katsuki, T.; Yamada, Y.; Ikeda, M.; Sharpless, K. B. J. Am. Chem. Soc. **1981**, 103, 6237-6240.
- 12. Wang, Z.; Cui, Y. T.; Xu, Z. B.; Qu, J. J. Org. Chem. 2008, 73, 2270-2274.
- Jun-Ling, C.; Zheng-Wei, Y.; Feng-Ling, Q. J. Fluorine Chem. 2013, 155, 143–150.
- (a) Ramesh, P.; Raju, A.; Paramesh, J.; Ramisetti, A. *Tetrahedron Lett.* **2016**, *57*, 1087-1089; (b) Brownbridge, P.; Chan, T. H.; Brook, M. A.; Kang, G. J. *Can. J. Chem.* **1983**, *61*, 688-693; (c) Rayala, N. K.; Meshram, H. M. *Tetrahedron Lett.* **2011**, *52*, 1003-1007; (d) Xu, Q.; Yu, J.; Han, F.; Hu, J.; Chen, W.; Yang, L. *Tetrahedron: Asymmetry* **2010**, *21*, 156-158; (e) Ramesh, P.; Raju, A.; Fadnavis, N. W. *Helv. Chim. Acta* **2016**, *99*, 70-72.
- (a) Narasaka, K.; Pai, F.-C. *Tetrahedron* **1984**, *40*, 2233-2238. (b)
 Chen, K.-M.; Hardtmann, G. E.; Prasad, K.; Repic, O.; Shapiro, M. J. *Tetrahedron Lett.* **1987**, *28*, 155-158.

Supplementary Material

¹H and ¹³C spectra of all compounds have been provided in a separate electronic file as a supplementary data.

CRIPT CC

Tetrahedron

Highlights:

- The protecting-group-free total synthesis of goniothalesdiol A is described.
- Only three steps starting from *trans*-cinnamaldehyde in 55.1% overall yield. •
- A one-pot Sharpless epoxidation/regioselective epoxide opening is a key step. •
- Atom economical, step-economical and devoid of protecting group manipulations.

roup m.