

# A Facile Synthesis of 1,5-Diphenyl-3-(substituted oxy)-1*H*-1,2,4-triazoles

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A series of 1,5-diphenyl-3-alkoxy (or acyloxy)-1*H*-1,2,4-triazoles were synthesized and assayed in the rat adjuvant induced arthritis model. Some compounds show significant anti-inflammatory activity.

1,2,4-Triazole, 5. Mitt.<sup>1)</sup>:

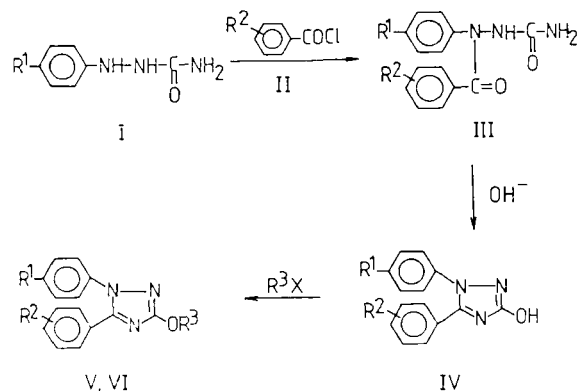
Einfache Synthese von 1,5-Diphenyl-3-(O-substituierten)-1*H*-1,2,4-triazolen

Eine Serie von 1,5-Diphenyl-3-alkoxy (oder acyloxy)-1*H*-1,2,4-triazolen wurde synthetisiert und auf ihre antiphlogistische Aktivität an Ratten im Adjuvans-Arthritis-Test untersucht. Einige dieser Verbindungen zeigen günstige entzündungshemmende Aktivität.

The potent and long-lasting anti-inflammatory activity of some diarylhetereocycles<sup>2,3)</sup> encouraged us the investigation of modified analogues. In the case of triazoles various 3-substituted derivatives with alkylthio<sup>2)</sup>, CF<sub>3</sub><sup>4)</sup>, and NH<sup>5)</sup> groups have been synthesized, but only a few of them possess high biological activity. Compounds with RO groups had received little attention. This paper reports the synthesis and anti-inflammatory activity of new 1,5-diphenyl-3-alkoxy (or acyloxy)-1*H*-1,2,4-triazoles.

Using of our experiences with thio ether compounds<sup>2)</sup> the intermediate hydroxy compounds VI were prepared from 1-phenyl-semicarbazides<sup>6,7)</sup> I (obtained from phenylhydrazines with KOCN) by acylation with benzoyl chlorides II followed by cyclization of III under base conditions.

3-Methoxy-derivatives V were obtained by alkylation of VI with Me<sub>2</sub>SO<sub>4</sub>, the 3-alkoxy-ones with basic side chains V from IV with alkylchlorides in dioxane/K<sub>2</sub>CO<sub>3</sub>. Acylation of IV with Ac<sub>2</sub>O, benzoyl chloride and sulfonyl chlorides lead to 3-(O-acylated)-derivatives VI (Scheme).



R<sup>1</sup> = Cl, F,

R<sup>2</sup> = 4-Cl, 4-F, 2-SMe

R<sup>3</sup> = Me, basic side chain, Ac, C<sub>6</sub>H<sub>5</sub>CO, R'SO<sub>2</sub>

The structures of 1-31 were characterized by satisfactory microanalyses and by IR- and <sup>1</sup>H-NMR-spectral data. Chemical data of the compounds are summarized in Tables 2 and 3.

## Adjuvant-induced arthritis (Newbould's preventive test<sup>8)</sup>)

Adult Long Evans male rats weighing 200-250 g were used. Arthritis was induced in rats by injecting heat-killed *Mycobacterium tuberculosis* ravenel (Freund adjuvant) in 0.1 ml of liquid paraffin suspension into the left hind paw on day 1. Compounds (dissolved in water containing Tween 80) were administered orally at low doses (6.25 and 12.5 mg/kg for 21 days starting on day 1). Control animals were given vehiculum.

Plethysmographic measurements of the primary lesion (injected paw) and the secondary lesion (non-injected paw) were made by mercury displacement on day 0, 3, 6, 10, 13, 17, 20, and 22 after injection of phlogist compared with controls.

## Results

Table 1 contains the biological data of the most active compounds.

Important structure-activity relationships can be summarized as follows.

1. Active compounds contain halogen substituents (as previously found<sup>2,5)</sup> in the para position on the phenyl rings (MeS-ones are inactive).
2. the highest effect resides in the methoxy and other alkoxy compounds having basic side chain
3. the hydroxy, benzoylated and sulfonylated derivatives exhibit no biological activity.

## Conclusion

12 proved to be the most active compound in comparison with naproxen but the representatives of other series<sup>2,5</sup> (SR, SO<sub>2</sub>R, NHR) were superior to 12 in detailed studies.

**Table 1:** Anti-inflammatory activity

Comp.	Inhibition of adjuvant induced arthritis in rats* (preventive adm.) % in 6.25 mg/kg po	Comp.	Inhibition of adjuvant induced arthritis in rats (preventive adm.) % in 6.25 mg/kg po.
7	33	17	30 <sup>+</sup>
8	33	18	37
12	67	22	42
14	31	Naproxen	44
15	49 <sup>+</sup>		

\* The % inhibitory effect shows the decrease of the hind-paw volumen after treatment in comparison with the increase of the hind-paw volume of controls

+ In 12.5 mg/kg po.

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## Experimental Part

Mp: Boetius apparatus, uncorr. - Analyses: C, H, N, S, and halogen were within  $\pm 0.5$  % of the theoretical values. - IR: Bruker IFS 85. - <sup>1</sup>H-NMR: Varian XL-100 FT. - TLC: Kieselgel 60 F<sub>254</sub> (Merck), benzene/methanol 10:1.

### General procedure for the preparation of 1,5-diphenyl-3-hydroxy-1,2,4-triazoles IV (1-6)

A mixture of 25 mmole of I, 28 mmole of II, and 2.4 ml of pyridine in 63 ml of benzene was heated for 3 h. After cooling the product was filtered off, washed with water and heated in a mixture of 62 ml 10 % KOH (aqueous) and 31 ml of ethanol at 60°C under N<sub>2</sub> for 1 h. The warm solution was acidified to pH 4, and after cooling the product was washed with water and recrystallized from dioxane. Products (R<sup>1</sup>; R<sup>2</sup>; mp. °C; yield %): 1: Cl, 4-Cl, 278-280, 95; 2: F, 4-Cl, 260-263, 87; 3: Cl, 4-F, 303-305, 70; 4: F, 4-F, 297-300, 85; 5: Cl, 2-SMe, 267-270, 93; 6: F, 2-SMe, 267-270, 86.

**Table 2:** Chemical data of compounds V

Comp. V	R <sup>1</sup>	R <sup>2</sup>	R <sup>3</sup>	Mp (°C)	Yield (%)	Molecular formula	Mol.W.
7	Cl	4-Cl	Me	120-122	67	C <sub>15</sub> H <sub>11</sub> Cl <sub>2</sub> N <sub>3</sub> O	320.1
8	F	4-Cl	Me	85-87	30	C <sub>15</sub> H <sub>11</sub> ClFN <sub>3</sub> O	303.7
9	F	4-F	Me	93-95	50	C <sub>15</sub> H <sub>11</sub> F <sub>2</sub> N <sub>3</sub> O	287.2
10	Cl	2-SMe	Me	98-100	44	C <sub>16</sub> H <sub>14</sub> ClN <sub>3</sub> OS	331.8
11	F	2-SMe	Me	118-120	41	C <sub>16</sub> H <sub>14</sub> FN <sub>3</sub> OS	315.3
12*	Cl	4-Cl	-(CH <sub>2</sub> ) <sub>2</sub> NMe <sub>2</sub>	255-257	30	C <sub>18</sub> H <sub>19</sub> Cl <sub>2</sub> N <sub>4</sub> O	410.7
13*	F	4-Cl	-(CH <sub>2</sub> ) <sub>2</sub> NMe <sub>2</sub>	233-236	28	C <sub>18</sub> H <sub>19</sub> Cl <sub>2</sub> FN <sub>4</sub> O	397.2
14	Cl	4-Cl	-(CH <sub>2</sub> ) <sub>3</sub> -Mf	109-111	70	C <sub>21</sub> H <sub>22</sub> Cl <sub>2</sub> N <sub>4</sub> O <sub>2</sub>	433.3
15	F	4-Cl	-(CH <sub>2</sub> ) <sub>3</sub> -Mf	88-89	83	C <sub>21</sub> H <sub>22</sub> ClFN <sub>4</sub> O <sub>2</sub>	416.8
16	Cl	4-F	-(CH <sub>2</sub> ) <sub>3</sub> -Mf	87-89	70	C <sub>21</sub> H <sub>22</sub> ClFN <sub>4</sub> O <sub>2</sub>	416.8
17	F	4-F	-(CH <sub>2</sub> ) <sub>3</sub> -Mf	71-74	68	C <sub>21</sub> H <sub>22</sub> F <sub>2</sub> N <sub>4</sub> O <sub>2</sub>	400.42
18 <sup>+</sup>	Cl	4-Cl	-(CH <sub>2</sub> ) <sub>3</sub> -Mp	237-240	41	C <sub>22</sub> H <sub>27</sub> Cl <sub>2</sub> N <sub>5</sub> O	519.2
19 <sup>+</sup>	F	4-Cl	-(CH <sub>2</sub> ) <sub>3</sub> -Mp	227-230	86	C <sub>22</sub> H <sub>27</sub> Cl <sub>2</sub> FN <sub>5</sub> O	502.8
20 <sup>+</sup>	Cl	4-F	-(CH <sub>2</sub> ) <sub>3</sub> -Mp	221-224	68	C <sub>22</sub> H <sub>27</sub> Cl <sub>2</sub> FN <sub>5</sub> O	502.8

Mf = morpholino; Mp = 1-Me-4-piperazino; \* = mono HCl salt; + = di-HCl salt

**Table 3:** Chemical data of compounds VI

Comp. VI	R <sup>1</sup>	R <sup>2</sup>	R <sup>3</sup>	Mp (°C)	Yield (%)	Molecular formula	Mol.W.
21	Cl	4-Cl	MeCO	139-141	82	C <sub>16</sub> H <sub>11</sub> Cl <sub>2</sub> N <sub>3</sub> O <sub>2</sub>	348.1
22	F	4-Cl	MeCO	84-86	70	C <sub>16</sub> H <sub>11</sub> ClFN <sub>3</sub> O <sub>2</sub>	331.7
23	Cl	4-F	MeCO	113-114	80	C <sub>16</sub> H <sub>11</sub> ClFN <sub>3</sub> O <sub>2</sub>	331.7
24	F	4-F	MeCO	113-115	66	C <sub>16</sub> H <sub>11</sub> F <sub>2</sub> N <sub>3</sub> O <sub>2</sub>	315.2
25	Cl	4-Cl	C <sub>6</sub> H <sub>5</sub> -CO	160-162	89	C <sub>21</sub> H <sub>13</sub> Cl <sub>2</sub> N <sub>3</sub> O <sub>2</sub>	410.2
26	F	4-Cl	C <sub>6</sub> H <sub>5</sub> -CO	145-148	73	C <sub>21</sub> H <sub>13</sub> ClFN <sub>3</sub> O <sub>2</sub>	393.7
27	Cl	4-F	C <sub>6</sub> H <sub>5</sub> -CO	124-126	90	C <sub>21</sub> H <sub>13</sub> ClFN <sub>3</sub> O <sub>2</sub>	393.7
28	F	4-F	C <sub>6</sub> H <sub>5</sub> -CO	104-106	81	C <sub>21</sub> H <sub>13</sub> F <sub>2</sub> N <sub>3</sub> O <sub>2</sub>	377.3
29	Cl	4-Cl	2-MeS-C <sub>5</sub> H <sub>4</sub> -CO	177-179	67	C <sub>22</sub> H <sub>15</sub> Cl <sub>2</sub> N <sub>3</sub> O <sub>5</sub> S	456.32
30	Cl	4-Cl	MeSO <sub>2</sub>	137-139	63	C <sub>15</sub> H <sub>11</sub> Cl <sub>2</sub> N <sub>3</sub> O <sub>3</sub> S	384.16
31	Cl	4-Cl	3,4-(MeO) <sub>2</sub> -C <sub>6</sub> H <sub>3</sub> -SO <sub>2</sub>	163-165	69	C <sub>22</sub> H <sub>17</sub> Cl <sub>2</sub> N <sub>3</sub> O <sub>5</sub> S	506.3

*General procedure for the preparation of  
1,5-diphenyl-3-alkoxy-1,2,4-triazoles V (7-11)*

*Method for 7-11:* A mixture of 10 mmole of IV, 11 mmole of  $\text{Me}_2\text{SO}_4$  and 1.6 g  $\text{K}_2\text{CO}_3$  in 60 ml of acetone was refluxed under stirring for 8 h. After evaporation the residue was poured into water, extracted with  $\text{CH}_2\text{Cl}_2$ , dried over  $\text{Na}_2\text{SO}_4$  and chromatographed on Kieselgel (benzene/methanol 10:1). After evaporation the first fractions gave V from a mixture of ethanol-petrolether.

*Method for 12-20:* A stirred mixture of 10 mmole of IV, 12 mmole of alkylchloride and 4.0 g of  $\text{K}_2\text{CO}_3$  in 50 ml of dioxane was refluxed for 6-8 h, then poured into water and extracted with  $\text{CH}_2\text{Cl}_2$ , finally dried. 14-17 were isolated as base, 12, 13 as HCl-salts in crystalline form.

*General procedure for the preparation of  
1,5-diphenyl-5-acyloxy-1,2,4-triazoles VI*

*Method for 21-24:* A solution of 10 mmole of IV in 5 ml of  $\text{Ac}_2\text{O}$  was heated for 30 min, then poured into water giving the products (recryst. from ethanol).

*Method for 25-29:* A stirred mixture of 10 mmole of IV, 12 mmole (substituted) benzoyl chloride and 4.0 g of  $\text{K}_2\text{CO}_3$  in 40 ml of dioxane was heated for 8-10 h. The mixture was poured into water, filtered and recrystallized from MeCN.

*Method for 30-31:* A mixture of 2 mmole of IV and 0.2 g KOH in 9 ml of dioxane was heated for 1 h, then 2 mmole of sulfonyl chloride in 3 ml of dioxane were dropped in. Then the mixture was heated for 5 h and finally poured into water yielding 30 (from ethanol) and 31 (from MeCN), respectively.

*References*

- 1 Part IV: É.K. Bozó, G. Szilágyi, J. Langó and I. Pelczér, *Heterocycles*, under publication
- 2 G. Szilágyi, T. Somorai, É. Bozó, J. Langó, G. Nagy, J. Reiter, J. Janáky, and F. András, *Eur. J. Med. Chem.*, in press.
- 3 T.R. Sharpe, S.C. Cherkofsky, W.C. Howes, D.H. Smith, W.A. Gregory, S.B. Haber, M.R. Leadbetter, and J. Whitney, *J. Med. Chem.* 28, 1188 (1985).
- 4 L. Czollner, G. Szilágyi, J. Langó and J. Janáky, *Monatsh. Chemie* 119, 349 (1988).
- 5 É. Bozó, G. Szilágyi, and J. Janáky, *Arch. Pharm. (Weinheim)* 322, 583 (1989).
- 6 G. Hevitt, *J. Chem. Soc.* 1893, 872.
- 7 O. Danek und S. Nouzova, *Coll. Czech. Chem. Comm.* 33, 425 (1968).
- 8 B.B. Newbould, *Brit. J. Pharm.* 21, 127 (1963).

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