Synthesis of 1,2:5,6-Di-O-isopropylidene-3-O-[3-(5-fluorouracil-1-yl)-propionoyl]-α-D-glucofuranose and its Antitumor Activity

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In order to provide a less toxic 5-fluorouracil derivative, 1,2:5,6-di-O-isopropylidene-3-O-[3-(5-fluorouracil-1-yl)-propionoyl]- α -D-glucofuranose, which was the derivative of 5-fluorouracil combining indirectly to 3-position of diacetoneglucose, was synthesized, and its antitumor activity was tested.

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It has been reported that 5-fluorouracil has not only remarkable antitumor activity but also has strong side-effects [1,2]. In order to reduce the latter effects, 1-(tetrahydro-2-furanyl)-5-fluorouracil (ftorafur) [3] and 5-fluorouridine [4] were synthesized.

In the previous paper [5], as a non-toxic model of 5-fluorouracil, we synthesized 1,2:5,6-di-O-isopropylidene-3-O-[3-(uracil-1-yl)]- α -D-glucofuranose (3a), which did not contain a fluoro group at the 5-position of uracil. The present paper is concerned with the synthesis of 1,2:5,6-di-O-isopropylidene-3-O-[3-(5-fluorouracil-1-yl)propionoyl]- α -D-glucofuranose (3b), which is expected to hold a remarkable antitumor activity, via the synthetic route shown in Scheme 1.

We have synthesized **3b** by the Michael addition of 1,2:5,6-di-O-isopropylidene-3-O-acryloyl- α -D-glucofuranose (1) with 5-fluorouracil (2), according to the method established in the previous paper [5].

Table 1 Antitumor Activities of 3b and 4 Doses Tested T/C (%) 2),соон (mg/kg) 3b 240 122 100 120 113 60 108 101 30 104 100

EXPERIMENTAL

Melting points were determined in capillary tubes on a warm plate and are uncorrected. The ir spectrum was recorded on a JASCO A-202 spectrophotometer, and the ¹H-nmr spectrum was measured with a JEOL PMX-60 spectrometer using tetramethylsilane as the internal standard. The ¹³C-nmr spectrum was obtained on a JEOL JNM-PS-100 spectrometer equipped with a JNM-PET-100 Fourier transform accessory. The mass spectrum was obtained on a JEOL JMS-01SG spectrometer. Optical rotation was determined with a Union Digital PM-101 polarimeter. The reaction was monitored via tlc with Merck F₂₅₄ silica gel plate, which was developed with petroleum ether-butanol (9:1 v/v).

1,2:5,6-Di-O-isopropylidene-3-O-acryloyl- α - D-glucofuranose (1) and 5-Fluorouracil (2b).

Compound 1 was prepared by the reaction of Schotten-Bauman of 1,2:5,6-di-O-isopropylidene- α -D-glucofuranose with acryloyl chloride in 5N sodium hydroxide solution, according to the method reported in the previous paper [5], mp 76.5-77.0°.

5-Fluorouracil (2b) of commercial feagent grade was used without further purification.

1,2:5,6-Di-O-isopropylidene-3-O-[3-(5-fluorouracil-1-yl)propionoyl]- α -D-glucofuranose (3b).

Metallic sodium (0.10 g) and **2b** (0.20 g, 1.5 mmoles) were added into dried ethanol (10 cm³). After evolution of hydrogen ceased, **1** (0.47 g, 1.5 mmoles) was added to the solution. The reaction mixture was refluxed in oil bath until the system became homogeneous. The reactant was evaporated under reduced pressure to give the crude product as a syrup. The residual syrup was subjected to column chromatography on silica gel. Elution with benzene-methanol (49:1 v/v) afforded the product as a white powder. The crude product was purified by reversed phase chromatogaphy on Ci³-ODS (eluting agent: water-methanol) (50:50 v/v), 0.39 g (58%), mp 103.5-104.8; ¹H-nmr (deuteriochloroform): δ 1.4-1.5 (t, 12H, 4CH³), 5.8 (d, 1H, H-1'), 7.7 (d, 1H, H-6 uracil ring), 10.0 (broad, 1H, NH); ¹³C-nmr (deuteriochloroform): δ 170 (C=O, ester), 130 (5-C, uracil ring), 72.5 (5'-C, furan), 67 (6'-C, furan); ms: m/e 429 (M*-15); uv (methanol): δ max 270.5 nm; ϵ max 9280; ir (potassium bromide): 3180 (NH), 2980 (CH), 1710 (ester C=O) and 1680 cm⁻¹ (amide C=O); $[\alpha]_D^{2s}$ -22° (c 0.1, methanol).

Anal. Calcd. for C₁₉H₂₅FN₂O₆: C, 51.35; H, 5.67; N, 6.30; F, 4.28. Found: C, 50.98; H, 5.68; N, 6.16; F, 4.46.

Antitumor Activity.

The antitumor effect of compound 3b obtained was tested in vivo against P-388 leukemia in mice according to a typical NCI protocol [6]. Compound 3b has anticancer activity, as shown in Table 1. However, it

was not at such a high level. On the contrary, 5-fluorouracil-1-propionic acid (4) has practically no activity under the same testing conditions, as shown in Table 1.

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