



Iodine-catalyzed three-component reaction of quinazoline-2,5-diones with aldehydes and styrenes for the synthesis of allylamine derivatives

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ABSTRACT

Iodine-catalyzed three component reaction of quinazoline-2,5-diones, aldehydes, and styrenes provides allylamine derivatives was described. Both paraformaldehyde and ethyl glyoxalate can be applied in this reaction with simple styrenes with high regioselectivity.

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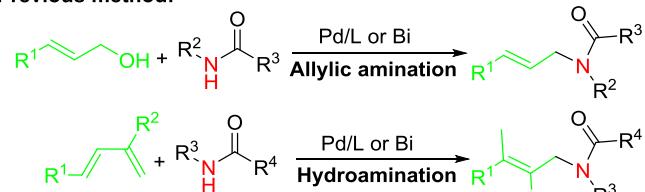
C–N formation

1. Introduction

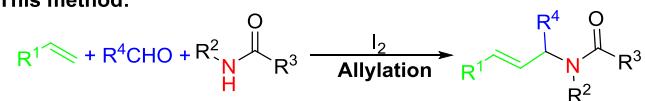
Allylamines represent an important structural motif frequently found in a wide variety of natural products, pharmaceuticals,¹ bioactive and antimycotic activity products.^{2,3} They are one of the most valuable chemical intermediates for a wide range of organic compounds.⁴ Among the different types of allylic substitution reactions, the synthesis of allylamines is of special interest. The transition-metal-catalyzed allylic aminations of allylalcohols⁵ and their derivatives^{6–9} have become a powerful tool for the construction of allylamines. Alternatively, the Lewis acid-promoted reductive amination of cinnamaldehyde giving allylamines has also been reported.^{10–12} The sequential hydrozirconation trans-metallation imine addition of alkynes establishes an efficient new route for the preparation of synthetically useful allylic amine building blocks.¹³ However, such a reaction is a synthetic challenge due to the intrinsic poor nucleophilicity of simple alkenes. Recently, some groups have reported the palladium-catalyzed Heck arylation of allylamine derivatives employing aryl iodide or arenediazonium salts.¹⁴ Huang et al. have developed a novel palladium-catalyzed

vinylation reaction that enables the catalytic functionalization of alkenes with amines.¹⁵ As compared to analogous reactions, the selective N-allylation of electron-deficient *N*-heterocycles and amides with olefins has limitedly been investigated. In this respect, notable progress was made respectively by Beller and Shibasaki groups via the allylic substitution of allylic alcohols with amides to afford allylic amines (**Scheme 1**).¹⁶ More recently, Beller et al. described a general and regioselective 1,4-addition of electron deficient amides into 1,3-dienes catalyzed by $\{[\text{Pd}(\pi\text{-cinnamyl})\text{Cl}]_2\}$ in

Previous method:



This method:



Scheme 1. Synthesis of allylamine derivatives.

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the presence of 1,3-bis(diphenylphosphino)propane to deliver allylamines.¹⁷

We have recently reported the synthesis of allylamine derivatives via I₂-catalyzed one-pot multi-component reaction of 3,4-dihydropyrimidinones (or amides), paraformaldehyde, and styrenes.¹⁸ This direct method from unactivated styrenes does not need any additives and proceeds with high regioselectivity, which forms water as the sole by-product.

Herein, we became interested in the formation of more challenging allylic amines. Glycinate are fascinating precursors for the synthesis of multifunctionalized heterocycles.¹⁹ Therefore, we present the iodine-catalyzed direct N-allylation of quinazoline-2,5-dione with aldehydes (including ethyl glyoxalate, and paraformaldehyde) and non-activated alkenes (**Scheme 1**).

2. Results and discussion

Initially, the model reaction of octahydroquinazoline-2,5-dione (**1a**), ethyl glyoxalate (**2a**, 50 wt % solution in toluene), and styrene (**3a**) was used to optimize the reaction conditions (**Table 1**). Only trace amount of product **4a** was obtained when sulfuric acid-modified polyethylene glycol (PEG-OSO₃H) or trimethylsilyl chloride (TMSCl) was utilized as catalyst (entries 1 and 2). The application of trifluoroacetic acid (TFA) resulted in no conversion (entry 3). When trifluoromethanesulfonic acid (TfOH), the desired product **4a** was isolated in 17% yield (entry 4). Also, some typical Lewis acids such as ZnCl₂, FeCl₃ were tested in this reaction, and only FeCl₃ promoted the reaction in 32% yield of **4a** (entries 5 and 6). Notably, good result was obtained using 0.2 equiv of molecular iodine as the catalyst in 1,4-dioxane at 110 °C for 24 h, and the desired product **4a** was isolated in 60% yield (entry 7). Good reaction yield and activity were also observed using I₂ or FeCl₃ combining with TfOH (entries 8 and 9), which provided the desired product **4a** 48% and 58% yield, respectively. Then, the effects of solvent and temperature in this reaction were also investigated. Among the solvents examined, dioxane, which was used in an earlier study gave good result, in contrast to the unsatisfactory result obtained with DMF and DMSO as solvent (entries 12 and 13). Other nonpolar solvents, such as xylene and toluene (entries 10 and 11) only gave a low yield even at higher temperature. When the catalyst loading of I₂ was

Table 1
Optimization of the three-component reaction conditions^a

Entry	Catalysis (mol%)	Solvent	Temp (°C)	Time (h)	Yield (%) ^b
1	PEG-OSO ₃ H (20)	Dioxane	110	24	Trace
2	TMSCl (20)	Dioxane	110	24	Trace
3	TFA (20)	Dioxane	110	24	nd ^c
4	TfOH (20)	Dioxane	110	24	17
5	ZnCl ₂ (20)	Dioxane	110	24	Trace
6	FeCl ₃ (20)	Dioxane	110	24	32
7	I ₂ (20)	Dioxane	110	24	60
8	FeCl ₃ (20)/TfOH (2)	Dioxane	110	24	48
9	I ₂ (20)/TfOH (2)	Dioxane	110	24	58
10	I ₂ (20)	Xylene	120	24	35
11	I ₂ (20)	Toluene	110	24	42
12	I ₂ (20)	DMF	150	24	nd ^c
13	I ₂ (20)	DMSO	150	24	Trace
14	I ₂ (40)	Dioxane	110	36	64
15	I ₂ (30)	Dioxane	110	36	57
16	I ₂ (50)	Dioxane	110	36	65

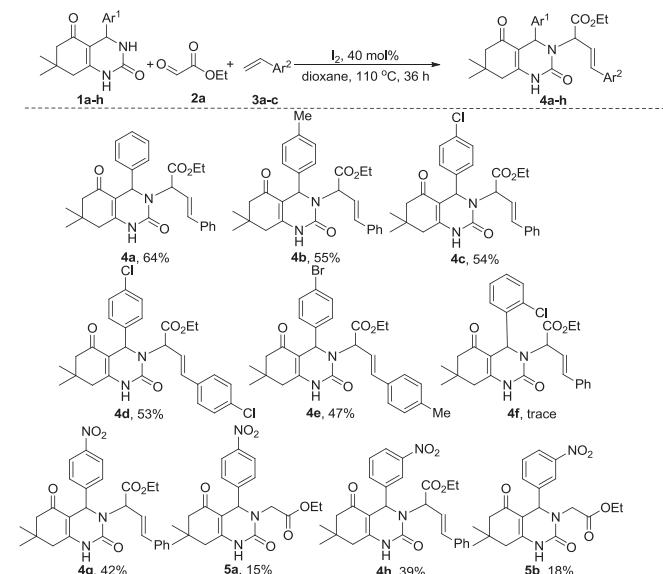
^a Reaction conditions: quinazoline-2,5-dione (**1a**, 0.5 mmol), ethyl glyoxalate (**2a**, 1.5 mmol), styrene (**3a**, 0.75 mmol), catalyst, solvent (3 mL), 110 °C.

^b Isolated yield of **4a** after purification by flash chromatography.

^c Not detected in the reaction mixture.

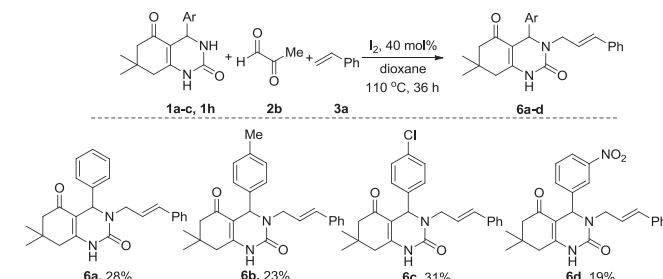
increased to 40 mol %, a considerably higher yield was observed (entry 14). Reducing the I₂ loading to 30 mol % lead to a low yield of 57% (entry 15), and increasing the I₂ loading to 50 mol %, the yield have no obvious change (entry 16). It is worth mentioning that the reaction proceeds with highly selective, delivering exclusively the linear *E* isomer of the allylic amine product **4a** in all cases.

With the optimized conditions in hand, a wide variety of substituted quinazoline-2,5-diones (**1**, **2a**, and styrenes (**3**) were submitted to investigate its substrate scope and generality (**Scheme 2**). A series of allylamine derivatives **4a–h** were obtained with moderate yield and high regioselectivity. No significant difference in yield was observed with electron-donating or electron-withdrawing substituents on the phenyl group of substrates **1**. However, steric effects has a negative effect on the reaction. When introducing steric hindrance ortho to the aromatic ring, trace desired product **4f** with a mixture was detected. The reaction of 4-NO₂ or 3-NO₂ substituted quinazoline-2,5-diones **1g–h** with **2a** and **3a** resulted in the formation of **4g–h** in 39–42% yield, respectively, whereas a minor amount of by-products as **5a** and **5b** were also isolated. Notably, in all reactions in which we observed lower yields of **4**, starting materials were recovered. In some experiments, we also observed the formation of trace amount of by-products.



Scheme 2. Three-component reaction of quinazoline-2,5-diones, ethyl glyoxalate, and styrenes.

Next, we tried to extend the scope of the methodology to the use of methylglyoxal instead of ethyl glyoxalate in the three-component reaction (**Scheme 3**). To our surprise, the reaction only provide product (**6a**) with recovered **1a**. Similarly, 4-Me, 4-Cl or 3-NO₂ substituted quinazoline-2,5-diones reacted well to deliver the corresponding products **6c–d**.²⁰ The mechanism studies on the deacylation are underway.



Scheme 3. Reaction between quinazoline-2,5-dione, methylglyoxal, and styrene.

To further expand the scope of substrates, we examined paraformaldehyde (**2c**) as the aldehyde partner in the three-component reaction (Table 2). When decreasing the catalyst of I₂ loading to 20 mol % in dioxane at 110 °C within 12 h, the three-component reaction proceeded to deliver the allylamine derivatives **6a–p** in good yields. We found that quinazoline-2,5-diones bearing electron-withdrawing groups (4-NO₂, 4-Cl, 4-F, and 4-Br) provided higher yields than those containing electron-donating groups (4-Me) on the phenyl ring. Styrenes bearing electron-withdrawing groups provided higher yields than those containing electron-donating groups (entries 1–15). Steric effects due to the installation of *ortho*-substituents (2-Cl) were negligible (entry 16). When the styrene was replaced with 2-vinylnaphthalene, the reaction also resulted in the formation of allylamine derivatives **6q–r** in 60–64% yield (entries 17 and 18). However, some limitations were noted in case of alkyl olefin (1-octene) or heteroaromatic olefin (4-vinyl pyridine or 2-vinyl pyridine); they were unreactive in this reaction.

Table 2

Three-component reaction of quinazoline-2,5-diones, paraformaldehyde, and styrenes^a

Entry	R ¹	R ²	Product	Yield ^b (%)
1	H	C ₆ H ₅	6a	70
2	4-CH ₃	C ₆ H ₅	6b	52
3	4-Cl	C ₆ H ₅	6c	61
4	3-NO ₂	C ₆ H ₅	6d	62
5	H	4-ClC ₆ H ₅	6e	53
6	H	4-BrC ₆ H ₅	6f	56
7	H	4-CH ₃ C ₆ H ₅	6g	41
8	4-CH ₃	4-CH ₃ C ₆ H ₅	6h	39
9	4-CH ₃	4-ClC ₆ H ₅	6i	51
10	4-F	C ₆ H ₅	6j	64
11	4-Br	C ₆ H ₅	6k	63
12	4-NO ₂	C ₆ H ₅	6l	60
13	4-NO ₂	4-CH ₃ C ₆ H ₅	6m	42
14	4-NO ₂	4-ClC ₆ H ₅	6n	53
15	3-NO ₂	4-BrC ₆ H ₅	6o	61
16	2-Cl	C ₆ H ₅	6p	65
17	H	C ₁₂ H ₁₀	6q	64
18	4-Cl	C ₁₂ H ₁₀	6r	60

^a Reaction conditions: quinazoline-2,5-dione (0.5 mmol), paraformaldehyde (3.0 mmol), styrene (0.75 mmol), I₂ (20% mmol), dioxane (3 mL), 110 °C, 12 h.

^b Isolated yield of **4a** after purification by flash chromatography.

3. Conclusions

In summary, we have documented the I₂-catalyzed three-component reaction of octahydroquinazoline-2,5-diones with ethyl glyoxylate or paraformaldehyde and styrenes. The reaction proceeds with high regioselectivity to deliver allylic amine derivatives and shows good tolerance towards wide substrate scope, as demonstrated by the application of paraformaldehyde, and ethyl glyoxylate.

4. Experimental section

4.1. General procedure

4.1.1. Preparation of 4-phenyl-7,7-dimethyl-1,2,3,4,5,6,7,8-octahydroquinazoline-2,5-dione. The reaction of benzaldehyde (10 mmol) with urea (12 mmol), and 5,5-dimethylcyclohexane-1,3-dione (10 mmol) in the presence of H₂SO₄ (10% mmol) as catalyst in ethanol (15 mL) at 80 °C for 8 h produced 4-phenyl-7,7-dimethyl-

1,2,3,4,5,6,7,8-octahydroquinazoline-2,5-dione. After the reaction was completed (monitored by TLC), pure product was obtained from the crude product by filtration and recrystallization to afford.²¹

4.2. General procedure for three-component reactions

The mixture of quinazoline-2,5-dione (**1a**, 0.5 mmol, 135.0 mg), ethyl glyoxylate (**2a**, 3.0 equiv, 1.5 mmol, 306 mg), styrene (**3a**, 1.5 equiv, 0.75 mmol, 78.0 mg), I₂ (0.2 equiv, 0.1 mmol, 25.4 mg) and 1,4-dioxane (3 mL) was stirred at 110 °C. After the starting material **1a** had been consumed (monitored by TLC), 15 mL water were added and the mixture was extracted with EtOAc (3×15 mL). The extract was washed with 10% Na₂S₂O₃ solution (w/w), dried over anhydrous Na₂SO₄ and evaporation. The product was purified by flash chromatography on silica gel using ethyl acetate/petroleum ether as eluent to provide the corresponding allylamine.

4.2.1. (E)-Ethyl-2-(7,7-dimethyl-2,5-dioxo-4-phenyl-1,2,5,6,7,8-hexahydroquinolin-3(4H)-yl)-4-phenyl-3-butenoate (4a**).** White solid, mp 95–96 °C; yield: 64% (ethyl acetate/petroleum ether: 1/2.5); ¹H NMR (400 MHz, CDCl₃) δ=8.75 (br, 1H, NH), 7.40 (d, J=8.0 Hz, 2H, ArH), 7.30–7.22 (m, 8H, ArH), 6.48 (d, J=16.0 Hz, 1H, CH), 6.24–6.18 (q, 1H, CH), 5.56 (s, 1H, CH), 4.80 (d, J=8.0 Hz, 1H, CH), 4.16–4.11 (q, 2H, CH₂), 2.41–2.11 (m, 4H, 2CH₂), 2.20 (t, J=8.0 Hz, 3H, CH₃), 1.08 (s, 3H, CH₃), 0.90 (s, 3H, CH₃); ¹³C NMR (100 MHz, CDCl₃) δ=193.4, 169.2, 153.1, 149.5, 141.5, 135.8, 135.7, 128.5, 128.4, 128.2, 127.8, 127.4, 126.7, 121.7, 110.1, 61.7, 61.3, 58.0, 50.3, 39.8, 32.8, 29.2, 27.1, 14.1; HRMS [M+H]⁺ calcd for C₂₈H₃₀N₂O₄ 459.2278, found 459.2287.

4.2.2. (E)-Ethyl-2-(7,7-dimethyl-2,5-dioxo-4-(*p*-tolyl)-1,2,5,6,7,8-hexahydroquinolin-3(4H)-yl)-4-phenyl-3-butenoate (4b**).** White solid, mp 196–197 °C; yield: 55% (ethyl acetate/petroleum ether: 1/3); ¹H NMR (400 MHz, CDCl₃) δ=9.36 (br, 1H, NH), 7.24–7.17 (m, 7H, ArH), 7.02 (d, J=8.0 Hz, 2H, ArH), 6.42 (d, J=16.0 Hz, 1H, CH), 6.20–6.13 (q, 1H, CH), 5.45 (s, 1H, CH), 4.70 (d, J=8.0 Hz, 1H, CH), 4.10–4.04 (q, 2H, CH₂), 2.34–2.03 (m, 7H, CH₃, 2CH₂), 1.14 (t, J=8.0 Hz, 3H, CH₃), 1.01 (s, 3H, CH₃), 0.85 (s, 3H, CH₃); ¹³C NMR (100 MHz, CDCl₃) δ=193.4, 169.2, 153.1, 149.3, 138.5, 137.4, 135.9, 135.4, 129.0, 128.4, 128.2, 127.4, 126.7, 121.8, 110.1, 61.6, 61.3, 57.9, 50.3, 39.7, 32.7, 29.2, 27.2, 21.1, 14.1; HRMS [M+H]⁺ calcd for C₂₉H₃₂N₂O₄ 473.2435, found 473.2444.

4.2.3. (E)-Ethyl-2-(4-(4-chlorophenyl)-7,7-dimethyl-2,5-dioxo-1,2,5,6,7,8-hexahydroquinolin-3(4H)-yl)-4-phenyl-3-butenoate (4c**).** White solid, mp 197–199 °C; yield: 54% (ethyl acetate/petroleum ether: 1/3); ¹H NMR (400 MHz, CDCl₃) δ=8.78 (br, 1H, NH), 7.36–7.22 (m, 9H, ArH), 6.48 (d, J=16.0 Hz, 1H, CH), 6.22–6.15 (q, 1H, CH), 5.55 (s, 1H, CH), 4.78 (d, J=8.0 Hz, 1H, CH), 4.18–4.13 (q, 1H, CH₂), 2.41–2.11 (m, 4H, 2CH₂), 1.22 (t, J=8.0 Hz, 1H, CH₃), 1.09 (s, 3H, CH₃), 0.90 (s, 3H, CH₃); ¹³C NMR (100 MHz, CDCl₃) δ=193.3, 169.0, 153.0, 149.6, 140.1, 135.9, 135.7, 133.6, 128.8, 128.6, 128.4, 126.7, 121.4, 109.7, 61.6, 61.4, 57.5, 50.3, 39.8, 32.8, 29.3, 27.1, 14.1; HRMS [M+H]⁺ calcd for C₂₈H₂₉ClN₂O₄ 493.1889, found 493.1898.

4.2.4. (E)-Ethyl-4-(4-chlorophenyl)-2-(4-(4-chlorophenyl)-7,7-dimethyl-2,5-dioxo-1,2,5,6,7,8-hexahydroquinolin-3(4H)-yl)-3-butenoate (4d**).** White solid, mp 203–204 °C; yield: 53% (ethyl acetate/petroleum ether: 1/3); ¹H NMR (400 MHz, CDCl₃) δ=9.17 (br, 1H, NH), 7.27 (d, J=8.0 Hz, 2H, ArH), 7.19 (t, J=8.0 Hz, 4H, ArH), 7.07 (d, J=8.0 Hz, 1H, ArH), 6.36 (d, J=16.0 Hz, 1H, CH), 6.11–6.05 (q, 1H, CH), 5.46 (s, 1H, CH), 4.74 (d, 8.0 Hz, 1H, CH), 4.11–4.06 (q, 2H, CH₂), 2.34–2.04 (m, 4H, 2CH₂), 1.15 (t, J=8.0 Hz, 1H, CH₃), 1.02 (s, 3H, CH₃), 0.83 (s, 3H, CH₃); ¹³C NMR (100 MHz, CDCl₃) δ=193.3, 168.9, 152.9, 149.5, 140.1, 134.5, 134.1, 133.7, 128.8, 128.7, 128.6,

127.9, 122.2, 109.7, 61.6, 61.4, 57.5, 50.2, 39.8, 32.8, 29.2, 27.1, 14.1; HRMS [M+H]⁺ calcd for C₂₈H₂₈Cl₂N₂O₄ 527.1499, found 527.1510.

4.2.5. (*E*)-Ethyl-2-(4-(4-bromophenyl)-7,7-dimethyl-2,5-dioxo-1,2,5,6,7,8-hexahydroquinazolin-3(4H-yl)-4-(*p*-tolyl)-3-butenoate (4e**)**. White solid, mp 213–214 °C; yield: 47% (ethyl acetate/petroleum ether: 1/3); ¹H NMR (400 MHz, CDCl₃) δ=9.03 (br, 1H, NH), 7.39 (d, J=8.0 Hz, 2H, ArH), 7.27 (d, J=8.0 Hz, 2H, ArH), 7.11 (s, 5H, ArH), 6.43 (d, J=16.0 Hz, 1H, CH), 6.13–6.07 (q, 1H, CH), 5.54 (s, 1H, CH), 4.79 (d, J=8.0 Hz, 1H), 4.18–4.12 (q, 2H, CH₂), 2.41–2.10 (m, 7H, CH₃, 2CH₂), 1.22 (t, J=7.0 Hz, 3H, CH₃), 1.08 (s, 3H, CH₃), 0.89 (s, 3H, CH₃); ¹³C NMR (100 MHz, CDCl₃) δ=193.2, 169.1, 152.9, 149.5, 140.6, 138.3, 135.9, 132.8, 131.4, 129.2, 129.1, 126.6, 121.6, 120.2, 109.7, 61.6, 61.4, 57.2, 50.2, 39.8, 32.7, 29.2, 27.0, 21.2, 14.1; HRMS [M+H]⁺ calcd for C₂₉H₃₁BrN₂O₄ 551.1540, found 551.1546.

4.2.6. (*E*)-Ethyl-2-(7,7-dimethyl-4-(4-nitrophenyl)-2,5-dioxo-1,2,5,6,7,8-hexahydroquinazolin-3(4H-yl)-4-phenyl-3-butenoate (4g**)**. White solid, mp 207–209 °C; yield: 42% (ethyl acetate/petroleum ether: 1/2.5); ¹H NMR (400 MHz, CDCl₃) δ=9.17 (br, 1H, NH), 8.05 (d, J=8.0 Hz, 2H, ArH), 7.53 (d, J=8.0 Hz, 2H, ArH), 7.22–7.11 (m, 5H, ArH), 6.41 (d, J=16.0 Hz, 1H, CH), 6.10–6.04 (q, 1H, CH), 5.66 (s, 1H, CH), 4.82 (d, J=8.0 Hz, 1H, CH), 4.13–4.08 (q, J=16.0, 2H, CH₂), 2.38–2.04 (m, 4H, 2CH₂), 1.17 (t, J=8.0 Hz, 3H, CH₃), 1.03 (s, 3H, CH₃), 0.80 (s, 3H, CH₃); ¹³C NMR (100 MHz, CDCl₃) δ=193.2, 168.9, 152.9, 150.2, 148.7, 147.3, 136.6, 135.3, 128.7, 128.6, 128.2, 126.6, 123.7, 120.9, 109.1, 61.7, 61.6, 57.3, 50.2, 39.9, 32.8, 29.2, 27.0, 14.2; HRMS [M+H]⁺ calcd for C₂₈H₂₉N₃O₆ 504.2129, found 504.2138.

4.2.7. (*E*)-Ethyl-2-(7,7-dimethyl-4-(3-nitrophenyl)-2,5-dioxo-1,2,5,6,7,8-hexahydroquinazolin-3(4H-yl)-4-phenyl-3-butenoate (4h**)**. White solid, mp 174–176 °C; yield: 39% (ethyl acetate/petroleum ether: 1/2.5); ¹H NMR (400 MHz, CDCl₃) δ=8.63 (br, 1H, NH), 8.28 (s, 1H, ArH), 8.07 (d, J=8.0 Hz, 1H, ArH), 7.79 (d, J=8.0 Hz, 1H, ArH), 7.45 (t, J=8.0 Hz, 1H, ArH), 7.29–7.22 (m, 5H, ArH), 6.50 (d, J=16.0 Hz, 1H, CH), 6.24–6.18 (q, 1H, CH), 5.69 (s, 1H, CH), 4.79 (d, J=8.0 Hz, 1H, CH), 4.18–4.12 (q, J=16.0, 1H, CH₂), 2.45–2.11 (m, 4H, 2CH₂), 1.20 (t, J=8.0 Hz, 3H, CH₃), 1.10 (s, 3H, CH₃), 0.90 (s, 3H, CH₃); ¹³C NMR (100 MHz, CDCl₃) δ=193.3, 168.8, 152.4, 150.1, 148.3, 143.7, 136.2, 135.4, 133.8, 129.3, 128.5, 126.6, 122.8, 122.4, 121.1, 109.1, 61.6, 61.5, 57.8, 50.1, 39.8, 32.8, 29.1, 27.1, 14.0; HRMS [M+H]⁺ calcd for C₂₈H₂₉N₃O₆ 504.2129, found 504.2137.

4.2.8. Ethyl 2-(7,7-dimethyl-4-(4-nitrophenyl)-2,5-dioxo-1,2,5,6,7,8-hexahydroquinazolin-3(4H-yl)acetate (5a**)**. White solid, mp 221–222 °C; yield: 15% (ethyl acetate/petroleum ether: 1/2); ¹H NMR (400 MHz, CDCl₃) δ=9.53 (br, 1H, NH), 8.20 (d, J=8.0 Hz, 2H, ArH), 7.59 (d, J=8.0 Hz, 2H, ArH), 5.52 (s, 1H, CH), 4.49 (d, J=16.0 Hz, 1H, CH), 4.21–4.16 (q, 2H, CH₂), 3.51 (d, J=16.0 Hz, 1H, CH), 2.47–2.11 (m, 4H, 2CH₂), 1.26 (t, J=8.0 Hz, 3H, CH₃), 1.12 (s, 3H, CH₃), 0.92 (s, 3H, CH₃); ¹³C NMR (100 MHz, CDCl₃) δ=193.4, 168.1, 152.9, 150.5, 147.6, 147.5, 128.2, 123.9, 108.3, 61.5, 59.3, 50.1, 47.0, 39.8, 32.7, 29.2, 27.1, 14.1; HRMS [M+H]⁺ calcd for C₂₀H₂₃N₃O₆ 402.1660, found 402.1665.

4.2.9. Ethyl 2-(7,7-dimethyl-4-(3-nitrophenyl)-2,5-dioxo-1,2,5,6,7,8-hexahydroquinazolin-3(4H-yl)acetate (5b**)**. White solid, mp 202–203 °C; yield: 18% (ethyl acetate/petroleum ether: 1/2); ¹H NMR (400 MHz, CDCl₃) δ=9.40 (br, 1H, NH), 8.24 (s, 1H, ArH), 8.15 (d, J=8.0 Hz, 1H, ArH), 7.78 (d, J=8.0 Hz, 1H, ArH), 7.54 (t, J=8.0 Hz, 1H, ArH), 5.52 (s, 1H, CH), 4.46 (d, J=16.0 Hz, 1H, CH), 4.18–4.17 (q, 2H, CH₂), 3.55 (d, J=16.0 Hz, 1H, CH), 2.48–2.11 (m, 4H, 2CH₂), 1.25 (t, J=8.0 Hz, 3H, CH₃), 1.12 (s, 3H, CH₃), 0.95 (s, 3H, CH₃); ¹³C NMR (100 MHz, CDCl₃) δ=193.5, 168.1, 152.7, 150.6, 148.5, 142.8, 133.7, 129.7, 123.2, 122.3, 108.4, 61.5, 59.5, 50.2, 47.0, 39.8, 32.8, 29.2, 27.2,

14.1; HRMS [M+H]⁺ calcd for C₂₀H₂₃N₃O₆ 402.1660, found 402.1666.

4.2.10. (*E*)-3-Cinnamyl-7,7-dimethyl-4-phenyl-3,4,7,8-tetrahydroquinazoline-2,5(1H,6H)-dione (6a**)**. White solid, mp 176–178 °C; yield: 70% (ethyl acetate/petroleum ether: 1/3); ¹H NMR (400 MHz, CDCl₃) δ=9.32 (br, 1H, NH), 7.41 (t, J=4.0 Hz, 2H, ArH), 7.34–7.23 (m, 8H, ArH), 6.48 (d, J=16.0 Hz, 1H, CH), 6.11–6.05 (m, 1H, CH), 5.46 (s, 1H, CH), 4.67 (dd, J=15.0 Hz, 5.0 Hz, 1H, CH₂), 3.49 (dd, J=16.0 Hz, 8.0 Hz, 1H, CH₂), 2.43–2.10 (m, 4H, 2CH₂), 1.08 (s, 3H, CH₃), 0.92 (s, 3H, CH₃); ¹³C NMR (100 MHz, CDCl₃) δ=193.6, 153.1, 150.0, 141.3, 136.4, 133.8, 128.6, 128.5, 127.9, 127.8, 127.2, 126.4, 123.5, 109.3, 57.4, 50.4, 46.9, 39.8, 32.8, 29.2, 27.2; HRMS [M+H]⁺ calcd for C₂₅H₂₆N₂O₂ 387.2067, found 387.2072.

4.2.11. (*E*)-3-Cinnamyl-7,7-dimethyl-4-p-tolyl-3,4,7,8-tetrahydroquinazoline-2,5(1H,6H)-dione (6b**)**. White solid, mp 204–205 °C; yield: 52% (ethyl acetate/petroleum ether: 1/4); ¹H NMR (400 MHz, CDCl₃) δ=9.43 (br, 1H, NH), 7.36–7.24 (m, 7H, ArH), 7.14 (d, J=8.0 Hz, 2H, ArH), 6.50 (d, J=16.0 Hz, 1H, CH), 6.13–6.06 (m, 1H, CH), 5.44 (s, 1H, CH), 4.68 (dd, J=14.6 Hz, 4.0 Hz, 1H, CH₂), 3.50 (dd, J=16.0 Hz, 8.0 Hz, 1H, CH₂), 2.45–2.11 (m, 7H, CH₃, 2CH₂), 1.09 (s, 3H, CH₃), 0.95 (s, 3H, CH₃); ¹³C NMR (100 MHz, CDCl₃) δ=193.7, 153.1, 149.9, 138.3, 137.6, 136.4, 133.7, 129.3, 128.5, 127.8, 127.2, 126.4, 123.5, 109.4, 57.1, 50.4, 46.7, 39.8, 32.8, 29.2, 27.2, 21.1; HRMS [M+H]⁺ calcd for C₂₆H₂₈N₂O₂ 401.2224, found 401.2229.

4.2.12. (*E*)-3-Cinnamyl-7,7-dimethyl-4-(4-chlorophenyl)-3,4,7,8-tetrahydroquinazoline-2,5(1H,6H)-dione (6c**)**. White solid, mp 216–218 °C; yield: 61% (ethyl acetate/petroleum ether: 1/3); ¹H NMR (400 MHz, CDCl₃) δ=9.38 (br, 1H, NH), 7.35–7.23 (m, 9H, ArH), 6.46 (d, J=16.0 Hz, 1H, CH), 6.09–6.03 (m, 1H, CH), 5.45 (s, 1H, CH), 4.66 (dd, J=16.0 Hz, 4.0 Hz, 1H, CH₂), 3.47 (dd, J=16.0 Hz, 8.0 Hz, 1H, CH₂), 2.43–2.01 (m, 4H, 2CH₂), 1.08 (s, 3H, CH₃), 0.92 (s, 3H, CH₃); ¹³C NMR (100 MHz, CDCl₃) δ=193.5, 153.0, 150.1, 139.8, 136.2, 134.0, 133.7, 128.8, 128.6, 128.5, 127.9, 126.4, 123.2, 109.0, 56.8, 50.3, 47.0, 39.8, 32.8, 29.2, 27.1; HRMS [M+H]⁺ calcd for C₂₅H₂₅ClN₂O₂ 421.1677, found 421.1684.

4.2.13. (*E*)-3-Cinnamyl-7,7-dimethyl-4-(3-nitrophenyl)-3,4,7,8-tetrahydroquinazoline-2,5(1H,6H)-dione (6d**)**. White solid, mp 233–235 °C; yield: 62% (ethyl acetate/petroleum ether: 1/3); ¹H NMR (400 MHz, CDCl₃) δ=9.55 (br, 1H, NH), 8.25 (s, 1H, ArH), 8.12 (t, J=4.0 Hz, 1H, ArH), 7.79 (d, J=8.0 Hz, 1H, ArH), 7.54–7.49 (m, 1H, ArH), 7.32–7.25 (m, 5H, ArH), 6.48 (d, J=12.0 Hz, 1H, CH), 6.10–6.08 (m, 1H, CH), 5.60 (s, 1H, CH), 4.69 (dd, J=16.0 Hz, 8.0 Hz, 1H, CH₂), 3.51 (dd, J=16.0 Hz, 8.0 Hz, 1H, CH₂), 2.49–2.11 (m, 4H, 2CH₂), 1.11 (s, 3H, CH₃), 0.93 (s, 3H, CH₃); ¹³C NMR (100 MHz, CDCl₃) δ=193.5, 152.7, 150.8, 148.5, 143.5, 136.0, 134.4, 133.6, 129.6, 128.6, 128.0, 126.4, 123.0, 122.8, 122.0, 108.3, 57.1, 50.2, 47.3, 39.8, 32.8, 29.1, 27.2; HRMS [M+H]⁺ calcd for C₂₅H₂₅N₃O₄ 432.1918, found 432.1924.

4.2.14. (*E*)-7,7-Dimethyl-4-phenyl-3-(3-(*p*-chlorophenyl)allyl)-3,4,7,8-tetrahydroquinazoline-2,5(1H,6H)-dione (6e**)**. White solid, mp 206–208 °C; yield: 53% (ethyl acetate/petroleum ether: 1/3); ¹H NMR (400 MHz, CDCl₃) δ=9.28 (br, 1H, NH), 7.41 (d, J=8.0 Hz, 2H, ArH), 7.35–7.23 (m, 7H, ArH), 6.44 (d, J=16.0 Hz, 1H, CH), 6.09–6.04 (m, 1H, CH), 5.45 (s, 1H, CH), 4.64 (dd, J=15.0 Hz, 5.0 Hz, 1H, CH₂), 3.53 (dd, J=16.0 Hz, 8.0 Hz, 1H, CH₂), 2.45–2.12 (m, 4H, 2CH₂), 1.10 (s, 3H, CH₃), 0.94 (s, 3H, CH₂); ¹³C NMR (100 MHz, CDCl₃) δ=193.6, 153.0, 149.9, 141.2, 134.9, 133.4, 132.4, 128.7, 128.6, 128.0, 127.6, 127.2, 124.3, 109.4, 57.5, 50.3, 46.9, 39.9, 32.8, 29.2, 27.2; HRMS [M+H]⁺ calcd for C₂₅H₂₅ClN₂O₂ 421.1677, found 421.1683.

4.2.15. (*E*)-7,7-Dimethyl-4-phenyl-3-(3-(*p*-bromophenyl)allyl)-3,4,7,8-tetrahydroquinazoline-2,5(1H,6H)-dione (6f**)**. White solid, mp

209–210 °C; yield: 56% (ethyl acetate/petroleum ether: 1/4); ¹H NMR (400 MHz, CDCl₃) δ=9.33 (br, 1H, NH), 7.40 (t, J=8.6 Hz, 4H, ArH), 7.34–7.24 (m, 3H, ArH), 7.17 (d, J=8.0 Hz, 2H, ArH), 6.41 (d, J=16.0 Hz, 1H, CH), 6.09–6.02 (m, 1H, CH), 5.44 (s, 1H, CH), 4.62 (dd, J=16.0 Hz, 4.0 Hz, 1H, CH₂), 3.45 (dd, J=16.0 Hz, 8.0 Hz, 1H, CH₂), 2.43–2.11 (m, 4H, 2CH₂), 1.08 (s, 3H, CH₃), 0.92 (s, 3H, CH₃); ¹³C NMR (100 MHz, CDCl₃) δ=193.6, 153.0, 150.0, 141.2, 135.3, 132.4, 131.6, 128.6, 128.0, 127.9, 127.1, 124.5, 121.5, 109.3, 57.5, 50.3, 46.9, 39.8, 32.8, 29.2, 27.2; HRMS [M+H]⁺ calcd for C₂₅H₂₅BrN₂O₂ 465.1172, found 465.1179.

4.2.16. (E)-7,7-Dimethyl-4-phenyl-3-(3-(*p*-tolyl)allyl)-3,4,7,8-tetr-*ahydroquinazoline-2,5(1H,6H)-dione (6g)*. White solid, mp 216–218 °C; yield: 41% (ethyl acetate/petroleum ether: 1/3); ¹H NMR (400 MHz, CDCl₃) δ=9.72 (br, 1H, NH), 7.41 (d, J=8.0 Hz, 2H, ArH), 7.34–7.21 (m, 5H, ArH), 7.11 (d, J=8.0 Hz, 2H, ArH), 6.46 (d, J=16.0 Hz, 1H, CH), 6.07–6.00 (m, 1H, CH), 5.46 (s, 1H, CH), 4.69 (dd, J=16.0 Hz, 4.0 Hz, 1H, CH₂), 3.45 (dd, J=16.0 Hz, 8.0 Hz, 1H, CH₂), 2.44–2.09 (m, 7H, CH₃, 2CH₂), 1.08 (s, 3H, CH₃), 0.92 (s, 3H, CH₃); ¹³C NMR (100 MHz, CDCl₃) δ=193.6, 153.2, 150.2, 141.3, 137.6, 133.7, 133.6, 129.2, 128.6, 127.8, 127.2, 126.3, 122.3, 109.2, 57.3, 50.3, 46.8, 39.7, 32.7, 27.1, 21.1; HRMS [M+H]⁺ calcd for C₂₆H₂₈N₂O₂ 401.2224, found 401.2230.

4.2.17. (E)-7,7-Dimethyl-4-*p*-tolyl-3-(3-(*p*-tolyl)allyl)-3,4,7,8-tetr-*ahydroquinazoline-2,5(1H,6H)-dione (6h)*. White solid, mp 222–223 °C; yield: 39% (ethyl acetate/petroleum ether: 1/4); ¹H NMR (400 MHz, CDCl₃) δ=9.63 (br, 1H, NH), 7.29 (d, J=8.0 Hz, 2H, ArH), 7.23 (t, J=6.0 Hz, 2H, ArH), 7.11 (t, J=8.0 Hz, 4H, ArH), 6.45 (d, J=16.0 Hz, 1H, CH), 6.06–5.99 (m, 1H, CH), 5.42 (s, 1H, CH), 4.66 (dd, J=15.2 Hz, 4.8 Hz, 1H, CH₂), 3.45 (dd, J=16.0 Hz, 8.0 Hz, 1H, CH₂), 2.43–2.09 (m, 10H, 2CH₃, 2CH₂), 1.07 (s, 3H, CH₃), 0.93 (s, 3H, CH₃); ¹³C NMR (100 MHz, CDCl₃) δ=193.6, 153.2, 153.1, 150.1, 150.0, 138.4, 137.5, 137.5, 133.7, 133.6, 129.2, 129.1, 127.2, 126.3, 122.4, 109.3, 57.1, 50.4, 46.7, 39.7, 32.7, 29.2, 27.2, 21.1, 21.0; HRMS [M+H]⁺ calcd for C₂₇H₃₀N₂O₂ 415.2380, found 415.2386.

4.2.18. (E)-7,7-Dimethyl-4-*p*-tolyl-3-(3-(*p*-chlorophenyl)allyl)-3,4,7,8-tetr-*hydroquinazoline-2,5(1H,6H)-dione (6i)*. White solid, mp 228–230 °C; yield: 51% (ethyl acetate/petroleum ether: 1/3); ¹H NMR (400 MHz, CDCl₃) δ=9.30 (br, 1H, NH), 7.28–7.21 (m, 6H, ArH), 7.12 (d, J=4.0 Hz, 2H, ArH), 6.41 (d, J=16.0 Hz, 1H, CH), 6.07–6.00 (m, 1H, CH), 5.40 (s, 1H, CH), 4.61 (dd, J=15.4 Hz, 4.8 Hz, 1H, CH₂), 3.51 (dd, J=16.0 Hz, 8.0 Hz, 1H, CH₂), 2.42–2.10 (m, 7H, 2CH₂, CH₃), 1.08 (s, 3H, CH₃), 0.93 (s, 3H, CH₃); ¹³C NMR (100 MHz, CDCl₃) δ=193.6, 153.0, 150.0, 138.2, 137.7, 134.9, 133.4, 132.2, 129.3, 128.6, 127.6, 127.1, 124.4, 109.4, 57.3, 50.3, 46.7, 39.8, 32.8, 29.2, 27.2, 21.1; HRMS [M+H]⁺ calcd for C₂₆H₂₇ClN₂O₂ 435.1834, found 435.1838.

4.2.19. (E)-3-Cinnamyl-7,7-dimethyl-4-(4-fluorophenyl)-3,4,7,8-tetr-*hydroquinazoline-2,5(1H,6H)-dione (6j)*. White solid, mp 215–217 °C; yield: 64% (ethyl acetate/petroleum ether: 1/3); ¹H NMR (400 MHz, CDCl₃) δ=9.44 (br, 1H, NH), 7.40–7.23 (m, 7H, ArH), 7.04–6.98 (q, 2H, ArH), 6.48 (d, J=16.0 Hz, 1H, CH), 6.11–6.05 (m, 1H, CH), 5.45 (s, 1H, CH), 4.69 (dd, J=15.4 Hz, 5.0 Hz, 1H, CH₂), 3.45 (dd, J=16.0 Hz, 8.0 Hz, 1H, CH₂), 2.44–2.10 (m, 4H, 2CH₂), 1.09 (s, 3H, CH₃), 0.92 (s, 3H, CH₃); ¹³C NMR (100 MHz, CDCl₃) δ=193.6, 163.5, 161.1, 153.0, 150.0, 137.2, 136.3, 133.9, 128.9, 128.8, 128.6, 127.9, 126.4, 123.3, 115.6, 115.4, 109.3, 56.8, 50.3, 46.9, 39.8, 32.8, 29.2, 27.1; HRMS [M+H]⁺ calcd for C₂₅H₂₅FN₂O₂ 405.1973, found 405.1979.

4.2.20. (E)-3-Cinnamyl-7,7-dimethyl-4-(*p*-bromophenyl)-3,4,7,8-tetr-*hydroquinazoline-2,5(1H,6H)-dione (6k)*. White solid, mp 213–214 °C; yield: 63% (ethyl acetate/petroleum ether: 1/4); ¹H NMR (400 MHz, CDCl₃) δ=9.35 (br, 1H, NH), 7.45 (d, J=8.0 Hz, 2H, ArH), 7.32–7.24 (m, 7H, ArH), 6.46 (d, J=16.0 Hz, 1H, CH), 6.09–6.02

(m, 1H, CH), 5.43 (s, 1H, CH), 4.67 (dd, J=16.0 Hz, 4.0 Hz, 1H, CH₂), 3.45 (dd, J=15.2 Hz, 8.0 Hz, 1H, CH₂), 2.44–2.10 (m, 4H, 2CH₂), 1.09 (s, 3H, CH₃), 0.92 (s, 3H, CH₃); ¹³C NMR (100 MHz, CDCl₃) δ=193.5, 152.9, 150.1, 140.3, 136.2, 134.0, 131.8, 128.9, 128.6, 127.9, 126.4, 123.1, 121.9, 108.9, 56.9, 50.3, 46.9, 39.8, 32.8, 29.2, 27.2; HRMS [M+H]⁺ calcd for C₂₅H₂₅BrN₂O₂ 465.1172, found 465.1179.

4.2.21. (E)-3-Cinnamyl-7,7-dimethyl-4-(4-nitrophenyl)-3,4,7,8-tetr-*hydroquinazoline-2,5(1H,6H)-dione (6l)*. White solid, mp 231–233 °C; yield: 60% (ethyl acetate/petroleum ether: 1/2); ¹H NMR (400 MHz, CDCl₃) δ=9.50 (br, 1H, NH), 8.20–8.17 (m, 2H, ArH), 7.61–7.58 (m, 2H, ArH), 7.32–7.25 (m, 5H, ArH), 6.46 (d, J=16.0 Hz, 1H, CH), 6.08–6.02 (m, 1H, CH), 5.60 (s, 1H, CH), 4.70 (dd, J=15.0 Hz, 4.6 Hz, 1H, CH₂), 3.49 (dd, J=15.2 Hz, 8.0 Hz, 1H, CH₂), 2.48–2.11 (m, 4H, 2CH₂), 1.11 (s, 3H, CH₃), 0.91 (s, 3H, CH₃); ¹³C NMR (100 MHz, CDCl₃) δ=193.4, 152.9, 150.7, 148.4, 147.5, 136.0, 134.4, 128.6, 128.1, 128.1, 126.4, 124.0, 122.7, 108.3, 57.0, 50.2, 47.5, 39.9, 32.8, 29.2, 27.1; HRMS [M+H]⁺ calcd for C₂₅H₂₅N₃O₄ 432.1918, found 432.1923.

4.2.22. (E)-7,7-Dimethyl-4-(4-nitrophenyl)-3-(3-(*p*-tolyl)allyl)-3,4,7,8-tetr-*hydroquinazoline-2,5(1H,6H)-dione (6m)*. White solid, mp 243–245 °C; yield: 42% (ethyl acetate/petroleum ether: 1/3); ¹H NMR (400 MHz, CDCl₃) δ=9.43 (br, 1H, NH), 8.18 (t, J=4.0 Hz, 2H, ArH), 7.59 (t, J=4.0 Hz, 2H, ArH), 7.19 (d, J=8.0 Hz, 2H, ArH), 7.11 (d, J=8.0 Hz, 2H, ArH), 6.42 (d, J=16.0 Hz, 1H, CH), 6.01–5.97 (m, 1H, CH), 5.60 (s, 1H, CH), 4.67 (dd, J=16.0 Hz, 4.0 Hz, 1H, CH₂), 3.47 (dd, J=15.4 Hz, 8.0 Hz, 1H, CH₂), 2.47–2.11 (m, 7H, CH₃, 2CH₂), 1.10 (s, 3H, CH₃), 0.91 (s, 3H, CH₃); ¹³C NMR (100 MHz, CDCl₃) δ=193.4, 152.9, 150.7, 148.4, 147.5, 138.0, 134.4, 133.2, 129.3, 128.1, 126.3, 123.9, 121.6, 108.3, 56.9, 50.2, 47.5, 39.8, 32.8, 29.2, 27.1, 21.1; HRMS [M+H]⁺ calcd for C₂₆H₂₇N₃O₄ 446.2074, found 446.2081.

4.2.23. (E)-7,7-Dimethyl-4-(4-nitrophenyl)-3-(3-(*p*-chlorophenyl)allyl)-3,4,7,8-tetr-*hydroquinazoline-2,5(1H,6H)-dione (6n)*. White solid, mp 253–255 °C; yield: 53% (ethyl acetate/petroleum ether: 1/3); ¹H NMR (400 MHz, CDCl₃) δ=9.21 (br, 1H, NH), 8.19 (d, J=8.0 Hz, 2H, ArH), 7.58 (d, J=8.0 Hz, 2H, ArH), 7.29–7.21 (q, 4H, ArH), 6.41 (d, J=16.0 Hz, 1H, CH), 6.07–6.00 (m, 1H, CH), 5.58 (s, 1H, CH), 4.68 (dd, J=15.2 Hz, 5.2 Hz, 1H, CH₂), 3.47 (dd, J=15.2 Hz, 7.6 Hz, 1H, CH₂), 2.47–2.12 (m, 4H, 2CH₂), 1.11 (s, 3H, CH₃), 0.91 (s, 3H, CH₃); ¹³C NMR (100 MHz, DMSO-*d*₆) δ=192.5, 152.1, 151.0, 149.3, 146.7, 135.0, 131.9, 131.2, 128.4, 128.1, 127.8, 125.2, 123.7, 106.7, 56.4, 49.5, 46.5, 40.0, 32.2, 28.7, 26.3; HRMS [M+H]⁺ calcd for C₂₅H₂₄ClN₃O₄ 466.1528, found 466.1532.

4.2.24. (E)-7,7-Dimethyl-4-(3-nitrophenyl)-3-(3-(*p*-bromophenyl)allyl)-3,4,7,8-tetr-*hydroquinazoline-2,5(1H,6H)-dione (6o)*. White solid, mp 242–244 °C; yield: 61% (ethyl acetate/petroleum ether: 1/3); ¹H NMR (400 MHz, CDCl₃) δ=9.32 (br, 1H, NH), 8.24 (s, 1H, ArH), 8.12 (d, J=8.0 Hz, 1H, ArH), 7.77 (d, J=8.0 Hz, 1H, ArH), 7.51 (t, J=8.0 Hz, 1H, ArH), 7.43 (d, J=8.0 Hz, 2H, ArH), 7.17 (d, J=8.0 Hz, 2H, ArH), 6.40 (d, J=16.0 Hz, 1H, CH), 6.10–6.03 (m, 1H, CH), 5.58 (s, 1H, CH), 4.65 (dd, J=16.0 Hz, 8.0 Hz, 1H, CH₂), 3.52 (dd, J=16.0 Hz, 8.0 Hz, 1H, CH₂), 2.48–2.12 (m, 4H, 2CH₂), 1.11 (s, 3H, CH₃), 0.93 (s, 3H, CH₃); ¹³C NMR (100 MHz, CDCl₃) δ=193.5, 152.6, 150.7, 148.6, 143.4, 134.9, 133.5, 133.1, 131.7, 129.7, 127.9, 123.7, 123.1, 122.0, 121.9, 108.3, 57.2, 50.2, 47.3, 39.8, 32.9, 29.2, 27.2; HRMS [M+H]⁺ calcd for C₂₅H₂₄BrN₃O₄ 510.1023, found 510.1029.

4.2.25. (E)-3-Cinnamyl-7,7-dimethyl-4-(2-chlorophenyl)-3,4,7,8-tetr-*hydroquinazoline-2,5(1H,6H)-dione (6p)*. White solid, mp 97–99 °C; yield: 65% (ethyl acetate/petroleum ether: 1/4); ¹H NMR (400 MHz, CDCl₃) δ=9.57 (br, 1H, NH), 7.47 (t, J=4.0 Hz, 1H, ArH), 7.34–7.16 (m, 8H, ArH), 6.50 (d, J=16.0 Hz, 1H, CH), 6.10–6.03 (m, 1H, CH), 5.89 (s, 1H, CH), 4.47 (dd, J=16.0 Hz, 4.0 Hz, 1H, CH₂), 3.56

(dd, $J=16.0$ Hz, 8.0 Hz, 1H, CH₂), 2.45–2.08 (m, 4H, 2CH₂), 1.08 (s, 3H, CH₃), 0.98 (s, 3H, CH₃); ¹³C NMR (100 MHz, CDCl₃) δ =193.6, 152.4, 150.7, 138.5, 136.5, 133.6, 133.3, 130.6, 129.9, 129.3, 128.4, 127.7, 127.2, 126.4, 123.3, 107.9, 56.2, 50.3, 46.9, 39.8, 32.7, 29.2, 27.3; HRMS [M+H]⁺ calcd for C₂₅H₂₅ClN₂O₂ 421.1677, found 421.1682.

4.2.26. (E)-7,7-Dimethyl-3-(3-(2-naphthalenyl)allyl)-4-phenyl-3,4,7,8-tetrahydroquinazoline-2,5(1H,6H)-dione (6q). White solid, mp 221–223 °C; yield: 64% (ethyl acetate/petroleum ether: 1/4); ¹H NMR (400 MHz, CDCl₃) δ =9.50 (br, 1H, NH), 7.73 (d, $J=40.0$ Hz, 4H, ArH), 7.53–7.27 (m, 8H, ArH), 6.65 (d, $J=16.0$ Hz, 1H, CH), 6.23–6.19 (m, 1H, CH), 5.50 (s, 1H, CH), 4.72 (d, $J=16.0$ Hz, 1H, CH₂), 3.56 (dd, $J=16.0$ Hz, 8.0 Hz, 1H, CH₂), 2.45–2.09 (m, 4H, 2CH₂), 1.08 (s, 3H, CH₃), 0.92 (s, 3H, CH₃); ¹³C NMR (100 MHz, CDCl₃) δ =193.6, 153.1, 150.1, 141.3, 133.9, 133.7, 133.5, 133.0, 128.6, 128.2, 127.9, 127.6, 127.2, 126.4, 126.3, 125.9, 123.9, 123.5, 109.3, 57.5, 50.3, 47.0, 39.8, 32.8, 29.2, 27.1; HRMS [M+H]⁺ calcd for C₂₉H₂₈N₂O₂ 437.2224, found 437.2228.

4.2.27. (E)-4-(4-Chlorophenyl)-7,7-dimethyl-3-(3-(2-naphthalenyl)allyl)-3,4,7,8-tetrahydroquinazoline-2,5(1H,6H)-dione (6r). White solid, mp 249–250 °C; yield: 60% (ethyl acetate/petroleum ether: 1/4); ¹H NMR (400 MHz, CDCl₃) δ =9.54 (br, 1H, NH), 7.78–7.67 (m, 4H, ArH), 7.52–7.30 (m, 7H, ArH), 6.63 (d, $J=16.0$ Hz, 1H, CH), 6.22–6.15 (m, 1H, CH), 5.49 (s, 1H, CH), 4.71 (dd, $J=16.0$ Hz, 4.0 Hz, 1H, CH₂), 3.55 (dd, $J=15.2$ Hz, 8.0 Hz, 1H, CH₂), 2.45–2.10 (m, 4H, 2CH₂), 1.08 (s, 3H, CH₃), 0.92 (s, 3H, CH₃); ¹³C NMR (100 MHz, CDCl₃) δ =193.6, 153.0, 150.2, 139.9, 134.0, 133.7, 133.7, 133.5, 133.1, 128.8, 128.6, 128.2, 127.9, 127.6, 126.5, 126.3, 126.0, 123.6, 123.4, 109.0, 57.0, 50.3, 47.1, 39.8, 32.8, 29.2, 27.1; HRMS [M+H]⁺ calcd for C₂₉H₂₇ClN₂O₂ 472.1912, found 472.1917.

4.2.28. (E)-Ethyl 4-(4-chlorophenyl)-3-(1-ethoxy-1-oxo-4-phenyl-3-buten-2-yl)-6-methyl-2-oxo-1,2,3,4-tetrahydropyrimidine-5-carboxylate (7a). Colorless oil; yield: 41% (acetic acid/petroleum ether: 1/45); ¹H NMR (400 MHz, CDCl₃) δ =8.96 (br, 1H, NH), 7.24–7.14 (m, 9H, ArH), 6.42 (d, $J=16.0$ Hz, 1H, CH), 6.12–6.06 (q, 1H, CH), 5.42 (s, 1H, CH), 4.76 (d, $J=8.0$ Hz, 1H, CH), 4.09–4.01 (m, 4H, 2CH₂), 2.23 (s, 3H, CH₃), 1.16–1.11 (m, 6H, 2CH₃); ¹³C NMR (100 MHz, CDCl₃) δ =169.2, 165.3, 153.2, 146.1, 140.8, 135.7, 135.5, 133.6, 129.0, 128.8, 128.5, 128.3, 126.6, 121.9, 102.2, 61.4, 61.1, 60.1, 59.3, 18.4, 14.2, 14.0; HRMS [M+H]⁺ calcd for C₂₆H₂₇ClN₂O₅ 483.1681, found 483.1688.

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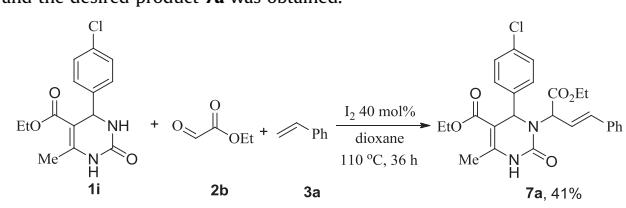
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Supplementary data

Supplementary data related to this article can be found at <http://dx.doi.org/10.1016/j.tet.2014.10.006>.

References and notes

- (a) Johannsen, M.; Jørgensen, K. A. *Chem. Rev.* **1998**, *98*, 1689–1708; (b) Petranay, G.; Ryder, N. S.; Stutz, A. *Science* **1984**, *224*, 1239–1241; (c) Kanno, H.; Taylor, R. J. K. *Tetrahedron Lett.* **2002**, *43*, 7337–7340.
- (a) Stutz, A.; Georgopoulos, A.; Granitzer, W.; Petranay, G.; Berney, D. *J. Med. Chem.* **1986**, *29*, 112–125; (b) Stutz, A. *Angew. Chem.* **1987**, *99*, 323–331; *Angew. Chem., Int. Ed. Engl.* **1987**, *26*, 320–328; (c) Nanavati, S. M.; Silverman, R. B. *J. Am. Chem. Soc.* **1991**, *113*, 9341–9349.
- For reviews, see: (a) Cheikh, R. B.; Chaabouni, R.; Laurent, A.; Mison, P.; Nafti, A. *Synthesis* **1983**, 685–699; (b) Trost, B. M.; Crawley, M. L. *Chem. Rev.* **2003**, *103*, 2921–2943.
- (a) Trost, B. M.; Vranken, D. L. *V. J. Am. Chem. Soc.* **1993**, *115*, 444–458; (b) Burgess, K.; Liu, L. T.; Pal, B. *J. Org. Chem.* **1993**, *58*, 4758–4763; (c) Paquette, L. A.; Leit, S. M. *J. Am. Chem. Soc.* **1999**, *121*, 8126–8127; (d) Liu, H.; Liang, X.; Søhnel, H.; Bulow, A.; Bols, M. *J. Am. Chem. Soc.* **2001**, *123*, 5116–5117; (e) Nagashima, H.; Isono, Y.; Iwamatsu, S. *J. Org. Chem.* **2001**, *66*, 315–319; (f) Ghorai, M. K.; Kumar, A.; Das, K. *Org. Lett.* **2007**, *9*, 5441–5444; (g) Hayashi, S.; Yorimitsu, H.; Oshima, K. *Angew. Chem., Int. Ed.* **2009**, *48*, 7224–7226; (h) Farwick, A.; Helmchen, G. *Org. Lett.* **2010**, *12*, 1108–1111; (i) Gartner, M.; Weihofen, R.; Helmchen, G. *Chem.–Eur. J.* **2011**, *17*, 7605–7622.
- (a) Muzart, J. *Eur. J. Org. Chem.* **2007**, 3077–3089; (b) Ohshima, T.; Miyamoto, Y.; Ipposhi, J.; Nakahara, Y.; Utsunomiya, M.; Mashima, K. *J. Am. Chem. Soc.* **2009**, *131*, 14317–14328; (c) Bandini, M. *Angew. Chem., Int. Ed.* **2011**, *50*, 994–995; (d) Biannic, A.; Aponick, A. *Eur. J. Org. Chem.* **2011**, *6605*–6617; (e) Das, K.; Shibuya, R.; Nakahara, Y.; Germain, N.; Ohshima, T.; Mashima, K. *Angew. Chem., Int. Ed.* **2012**, *51*, 150–154.
- (a) Ohmura, T. J.; Hartwig, F. J. *J. Am. Chem. Soc.* **2002**, *124*, 15164–15165; (b) Trost, B. M.; Bunt, R. C. *J. Am. Chem. Soc.* **1994**, *116*, 4089–4090.
- (a) Stanley, L. M.; Hartwig, J. F. *J. Am. Chem. Soc.* **2009**, *131*, 8971–8983; (b) Spiess, S.; Welter, C.; Frank, G.; Taquet, J. P.; Helmchen, G. *Angew. Chem., Int. Ed.* **2008**, *47*, 7652–7655; (c) Shi, C.; Ojima, I. *Tetrahedron* **2007**, *63*, 8563–8570; (d) Singh, O. V.; Han, H. *J. Am. Chem. Soc.* **2007**, *129*, 774–775; (e) Faller, J. W.; Wilt, J. C. *Org. Lett.* **2005**, *7*, 633–636; (f) Tissot-Croset, K.; Polet, D.; Alexakis, A. *Angew. Chem., Int. Ed.* **2004**, *43*, 2426–2428; (g) Berkowitz, D. B.; Maiti, G. *Org. Lett.* **2004**, *6*, 2661–2664; (h) Matsushima, Y.; Onitsuka, K.; Kondo, T.; Mitsudo, T.; Takahashi, S. *J. Am. Chem. Soc.* **2001**, *123*, 10405–10406.
- Ye, K. Y.; He, H.; Liu, W. B.; Dai, L. X.; Helmchen, G.; You, S. *J. Am. Chem. Soc.* **2011**, *133*, 19006–19014.
- (a) Weihofen, R.; Tverskoy, O.; Helmchen, G. *Angew. Chem., Int. Ed.* **2006**, *45*, 5546–5549; (b) Kuhn, O.; Mayr, H. *Angew. Chem., Int. Ed.* **1999**, *38*, 343–345; (c) Molander, G. A.; Gormisky, P. E.; Sandrock, D. L. *J. Org. Chem.* **2008**, *73*, 2052–2057; (e) Baruah, J. B.; Samuelson, A. G. *Tetrahedron* **1991**, *47*, 9449–9454.
- (a) Lee, O. Y.; Law, K. L.; Yang, D. *Org. Lett.* **2009**, *11*, 3302–3305; (b) Lee, O. Y.; Law, K. L.; Ho, C. Y.; Yang, D. *J. Org. Chem.* **2008**, *73*, 8829–8837.
- Yamauchi, T.; Sugiyama, J.; Higashiyama, K. *Heterocycles* **2002**, *58*, 431–447.
- Ramachandran, P. V.; Sakavuyi, P. D. K.; Clark, P. *Tetrahedron Lett.* **2010**, *51*, 3167–3169.
- (a) Wipf, P.; Kendal, C.; Stephenson, C. R. *J. Am. Chem. Soc.* **2003**, *125*, 761–768; (b) Wipf, P.; Stephenson, C. R. *J. Org. Lett.* **2003**, *5*, 2449–2452; (c) Patel, S. J.; Jamison, T. F. *Angew. Chem., Int. Ed.* **2004**, *43*, 3941–3944; (d) Boezio, A. A.; Charette, A. B. *J. Am. Chem. Soc.* **2003**, *125*, 1692–1693.
- (a) Prediger, P.; Barbosa, L. F.; Gnissou, Y.; Correia, C. R. *D. J. Org. Chem.* **2011**, *76*, 7737–7749; (b) Wu, J.; Marcoux, J. F.; Davies, I. W.; Reider, P. J. *Tetrahedron Lett.* **2001**, *42*, 159–162; (c) Dong, Y.; Busacca, C. A. *J. Org. Chem.* **1997**, *62*, 6464–6465; (d) Ripin, D. H.; Bourassa, D. E.; Brandt, T.; Heather, N. F.; Castaldi, M. J.; Hawkins, J.; Johnson, P. J.; Massett, S. S.; Neumsnn, K.; Phil-lips, J.; Raggan, J. W.; Rose, P. R.; Rutherford, J. L.; Sitter, B.; Stewart, A. M.; Vetelino, M. G.; Wei, L. *Org. Process Res. Dev.* **2005**, *9*, 440–450; (e) Olofsson, K.; Sahlin, H.; Larhed, M.; Hallberg, A. J. *Org. Chem.* **2001**, *66*, 544–549; (f) Olofsson, K.; Sahlin, H.; Larhed, M.; Hallberg, A. J. *Org. Chem.* **2000**, *65*, 7235–7239; (g) Alvisi, D.; Blart, E.; Bonini, B. F.; Mazzanti, G.; Ricci, A.; Zani, P. *J. Org. Chem.* **1996**, *61*, 7139–7146; (h) Reddington, M. V.; Bryant, D. C. *Tetrahedron Lett.* **2011**, *52*, 181–183; (i) Cacchi, S.; Fabrizi, G.; Goggiamani, A.; Sferrazza, A. *Org. Biomol. Chem.* **2011**, *9*, 1727–1730.
- (a) Xie, Y. J.; Hu, J. H.; Wang, Y. Y.; Xia, C. G.; Huang, H. M. *J. Am. Chem. Soc.* **2012**, *134*, 20613–20616; Hu, J. H.; Xie, Y. J.; Huang, H. M. *Angew. Chem., Int. Ed.* **2014**, *53*, 7272–7276.
- (a) Qin, H.; Yamagawa, N.; Matsunaga, S.; Shibasaki, M. *J. Am. Chem. Soc.* **2006**, *128*, 1611–1614; (b) Banerjee, D.; Jagadeesh, R. V.; Junge, K.; Junge, H.; Beller, M. *Angew. Chem., Int. Ed.* **2012**, *51*, 11156–11160; (c) Qin, H.; Yamagawa, N.; Matsunaga, S.; Shibasaki, M. *Angew. Chem., Int. Ed.* **2007**, *46*, 409–413.
- (a) Banerjee, D.; Junge, K.; Beller, M. *Angew. Chem., Int. Ed.* **2014**, *53*, 1630–1635; (b) Banerjee, D.; Junge, K.; Beller, M. *Org. Chem. Front.* **2014**, *1*, 368–372.
- Quan, Z. J.; Hu, W. H.; Zhang, Z.; Da, Y. X.; Jia, X. D.; Wang, X. C. *Adv. Synth. Catal.* **2013**, *355*, 891–900.
- (a) Yamamoto, Y.; Hayashi, H.; Saigoku, T.; Nishiyama, H. *J. Am. Chem. Soc.* **2005**, *127*, 10804–10805; (b) Luo, Y.; Lu, X.; Ye, Y.; Guo, Y.; Jiang, H.; Zeng, W. *Org. Lett.* **2012**, *14*, 5640–5643.
- In addition, the reaction between 3,4-dihydropyrimidinone (DHPMs, **1i**) with ethyl glyoxylate (**2a**) and styrene (**3a**) under the optimal conditions was tested, and the desired product **7a** was obtained.



- (a) Gowravaram, S.; Kiran Kumar Reddy, G. S.; Bhaskar, K. R.; Yadav, J. S. *Tetrahedron Lett.* **2003**, *44*, 6497–6499; (b) Kidwai, M.; Saxena, S.; Khalilur Rahman Khan, M.; Thukral, S. S. *Eur. J. Med. Chem.* **2005**, *40*, 816–819; (c) Mazaahir, K.; Divya, B.; Rakesh, K.; Pratibha, M. L. *Chem. Pharm. Bull.* **2010**, *58*, 1320–1323.