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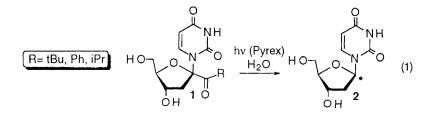
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# C1' ACYLATED DERIVATIVES OF 2'-DEOXYURIDINE. PHOTOLABILE PRECURSORS OF 2'-DEOXYURIDIN-1'-YL.

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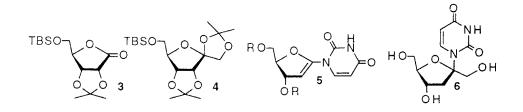
**Abstract.** C1' acylated derivatives of 2'-deoxyuridine (1a-c) were synthesized from 1-[3-deoxy- $\beta$ -D-psicofuranosyl]uracil (6). The acyl group is introduced via the C1' aldehyde (11). Following nucleophilic addition, the ketones (1a-c) are obtained via periodinane oxidation and desilylation with NH<sub>4</sub>F.

Modified nucleosides have the potential of exhibiting interesting and useful biological activity. Several groups have reported on the synthesis of novel nucleosides that are substituted at the C1' position. A variety of general approaches have been employed in order to produce alkyl, cyano, phosphono, and even O-acyl C1' substituted nucleosides.<sup>1-6</sup> To our knowledge, no examples of C1' acylated nucleosides have been reported. While C1' acylated nucleosides could be useful as therapeutic agents, we were interested in 1 as potential photolabile precursors for mechanistic studies concerned with the role of 2 in nucleic acid strand damage (eqn. 1).<sup>7</sup>

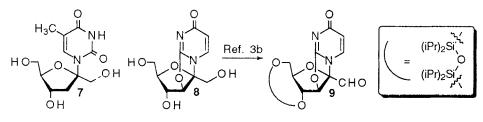


Based upon previous reports, we anticipated being able to prepare 1 from a variety of modified nucleoside synthons, including 3, 4 and  $5^{24.5}$  However, we chose to use

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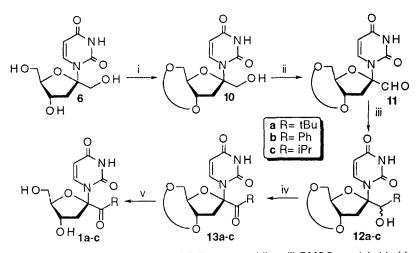


1-[3-deoxy- $\beta$ -D-psicofuranosyl]uracil (6), which was available in large quantities, as the starting point for 1.<sup>1</sup> The greatest stumbling block in the synthesis of 1 from 6 was expected to be the selective protection of the 4' and 6' (fructose numbering) hydroxyl groups. Moderate selectivity has been reported for the dimethoxytritylation of 7, as well as protection of 8 using 1,3-dichloro-1,1,3,3-tetrakisisopropyldisiloxane.<sup>3b,8</sup> In principle, 1 could have been synthesized from 9. However, we were concerned that conditions needed to effect opening of the cyclonucleoside might be incompatible with the functionality (e.g. disiloxane, aldehyde) present later in the synthesis.



Consequently, we chose to synthesize 1 from 6 via intermediate 11 (Scheme 1). Although disiloxane formation proceeds in low yield (26%), 47% of 6 is recoverable by desilylating the unwanted silyloxy products. Swern oxidation of 10 proceeds quantitatively, and in practice, 11 can be used without purification. Reduction of 11 with *tert*-BuLi yielded 12a as an inseparable mixture of diastereomers in moderate yield. Similarly, reduction with PhLi resulted in a separable mixture of benzyl alcohols (12b) in 85% yield. Less satisfactory results were obtained for the formation of the isopropyl ketone (1c). Reduction of 11 with isopropyl magnesium bromide produced 12c as a separable mixture of diastereomers in a combined yield of 42%. A significant amount of 10 (58%) was also obtained, due to  $\beta$ -hydride transfer from the Grignard reagent.

Carrying out the reduction with isopropyl lithium eliminated the formation of 10, but did not increase the yield of 12c. Each of the alcohols (12a-c) were readily oxidized with either  $CrO_3$ -pyridine- $Ac_2O$ , or Dess-Martin periodinane. However, the latter was found to be more convenient. Deprotection of the silylated ketones was effected using  $NH_4F$  in refluxing MeOH.



<sup>a</sup>Key: i) 1,3-Dichlorotetraisopropyldisiloxane, pyridine ii) DMSO, oxalyl chloride, Et<sub>3</sub>N, CH<sub>2</sub>Cl<sub>2</sub> iii) RLi, THF iv) Dess-Martin periodinane, CH<sub>2</sub>Cl<sub>2</sub> v) NH<sub>4</sub>F, MeOH

#### Scheme 1<sup>a</sup>

Irradiation of **1a-c** in the presence of (0.2 M) cyclohexa-1,4-diene, in CD<sub>3</sub>CN, under anaerobic conditions, results in high yields of a mixture of  $\alpha$ , $\beta$ -deoxyuridine.<sup>7</sup> The  $\alpha$ , $\beta$ -deoxyuridine formed is completely protonated, which is consistent with the trapping of 2'-deoxyuridin-1'-yl (2). Ketones **1a-c** are stable to the reagents used to chemically synthesize and deprotect oligonucleotides. Hence, **1a-c** should be useful for carrying out mechanistic studies concerning the involvement of **2** in nucleic acid damage.

## Experimental.

General Procedures: <sup>1</sup>H and <sup>13</sup>C NMR spectra were recorded on Bruker AM 270 and 300 spectrometers. Chemical shifts are reported in parts per million (ppm) relative to the respective solvent. IR spectra were recorded on a Perkin-Elmer series 1600 FTIR. Elemental analyses were performed by M-H-W laboratories. Et<sub>3</sub>N, DMSO, pyridine,  $CH_2Cl_2$  and  $CH_3CN$  were freshly distilled from  $CaH_2$  prior to use. THF was freshly distilled from  $Na^0$ /benzophenone ketyl. All reactions were conducted under  $N_2$  atmosphere. Flash chromatography was carried out using silica gel (230-400 mesh).<sup>9</sup>

**Preparation of 10.** Psicofuranosyluracil (6, 2.87 g, 11.1 mmol) was dried by azeotroping from pyridine (2 x 5 mL). Dried 6 was dissolved in pyridine (120 mL), and 1,3-dichloro-1,1,3,3-tetrakisisopropyldisiloxane (3.85 g, 12.3 mmol) in pyridine (15 mL) was added at -20°C over the course of 2.5 h. After allowing the reaction to stir and warm

to room temperature overnight, it was quenched with saturated NaHCO<sub>3</sub>. The solvents were removed in vacuo, and the residue was taken up in EtOH (100 mL). The solids were filtered, and the salts washed with EtOH (3 x 25 mL). Silica gel (30 g) was added, and after removing the solvent in vacuo, it was applied to a column of silica (150 g). The column was eluted with EtOAc: Hex (1:1), which were increased to EtOAc: Hex (3:2) upon commencement of product elution. The column was washed with EtOH in order to elute polar products. Disiloxane 10 (1.43 g, 26%) was obtained as an oil. Psicofuranosyluracil (6) could be recovered from the mixture of undesired silvlated nucleosides by desilvlation with NH<sub>4</sub>F in MeOH. <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  8.56 (bd s, 1H), 7.93 (d, 1H, J= 8.5 Hz), 5.64 (d, 1H, J= 8.5 Hz), 4.31 (m, 1H), 4.02 (m, 4H), 3.82 (dd, 1H, J= 2.5, 8 Hz), 2.92 (dd, 1H, J= 6.8, 13.5 Hz), 2.47 (t, 1H, J= 6 Hz), 2.40 (dd, 1H, J= 10.5, 13.5 Hz), 0.98 (m, 28 H); <sup>13</sup>C NMR (CDCl<sub>3</sub>)  $\delta$  164.8, 150.3, 141.9, 100.6, 97.2, 85.1, 67.9, 65.2, 60.1, 39.4, 17.4, 17.2, 17.1, 16.9, 16.8, 13.4, 13.0, 12.8, 12.4; IR (film) 3445, 2944, 2868, 1709, 1691, 1679, 1462, 1300, 1118, 1038 cm<sup>-1</sup>; Anal. caled. for C22H40N2O7Si2: C, 52.77; H, 8.05; N, 5.59. Found: C, 52.52; H, 7.89; N, 5.60.

Preparation of 11. DMSO (0.22 g, 2.8 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (9 mL) was added dropwise to oxalyl chloride (0.175 g, 1.38 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (7 mL) at -60°C. After stirring for 10 min, 10 (0.62 g, 1.22 mmol) was added in CH<sub>2</sub>Cl<sub>2</sub> (9 mL). After stirring for 35 min at -60°C, Et<sub>3</sub>N (0.66 g, 6.43 mmol) was added. The mixture was stirred briefly at -60°C. before warming to room temperature. The reaction mixture was quenched with  $H_2O$  (30) mL), and extracted with CH<sub>2</sub>Cl<sub>2</sub> (2 x 100 mL). The organic layers were combined, and washed with 1% HCl (75 mL), followed by 5% NaHCO<sub>3</sub> (75 mL), and brine (100 mL). After drying over MgSO<sub>4</sub>, the solvents were removed in vacuo to yield 0.56 g (91%) of 11. An analytical sample could be obtained via flash chromatography (EtOAc: CH<sub>2</sub>Cl<sub>2</sub>; 1:1). However, in practice 11 was carried on without further purification. mp 148-148.5°C; <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 9.22 (s, 1H), 8.78 (bd s, 1H), 7.90 (d, 1H, J= 8 Hz), 5.75 (d, 1H, J= 8 Hz), 4.39 (m, 1H), 4.13 (d, 1H, J= 14 Hz), 3.95 (dd, 1H, J= 2.6, 14 Hz), 3.74 (dd, 1H, J= 1.5, 9 Hz), 3.09 (dd, 1H, J= 10.3, 13.4 Hz), 2.33 (dd, 1H, J= 8, 13.4 Hz), 0.91 - 1.08 (m, 28 H); <sup>13</sup>C NMR (CDCl<sub>3</sub>) δ 187.7, 163.7, 150.4, 139.1, 102.2, 92.6, 85.2, 66.7, 58.8, 37.0, 25.4, 24.6, 17.4, 17.3, 17.2, 17.1, 17.0, 16.9, 16.7, 13.4, 12.8, 12.4. IR (film) 2946, 2894, 2868, 1745, 1700, 1464, 1386, 1301, 1276, 1118, 1077, 1056, 1038 cm<sup>-1</sup>; Anal. calcd. for C<sub>22</sub>H<sub>38</sub>N<sub>2</sub>O<sub>7</sub>Si<sub>2</sub>: C, 52.98; H, 7.68; N, 5.62. Found: C, 53.13; H, 7.82; N, 5.61.

**Preparation of 12a.** To **11** (402 mg, 0.81 mmol) in THF (5 mL) at -78°C was added a solution of *tert*-BuLi (1.08 mL, 1.5 M) in hexanes. The solution was stirred for 3 h, at

which time it was quenched by the addition of 0.3 N NaOAc (1 mL), and allowed to warm to room temperature. The solution was diluted with EtOAc (45 mL), washed with H<sub>2</sub>O (30 mL), and brine (40 mL). After drying over MgSO<sub>4</sub>, the secondary alcohol (243 mg, 54%) was obtained as a mixture of diastereomers by flash chromatography (EtOAc: Hexanes; 1:4). The alcohol (**12a**) was very unstable, and was typically oxidized immediately. <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  9.15 (bd s, 1H), 8.08 (d, 1H, J= 8 Hz, minor), 8.00 (d, 1H, J= 8 Hz, major), 5.69 (dd, 1H, J= 2,8 Hz), 4.21 - 3.85 (m, 4H), 3.12 - 2.95 (m, 1H), 2.91 (d, 1H, J= 9 Hz, major), 2.78 - 2.55 (m, 1H), 2.40 (d, 1H, J= 9.5 Hz), 2.03 (d, 1H, J= 9.5 Hz), 1.26 - 0.82 (m, 37H).

**Preparation of 12b.** Aldehyde **11** (450 mg, 0.90 mmol) was reduced with PhLi (2 eq.) as described above for the preparation of **12a**. Flash chromatography (EtOAc:CH<sub>2</sub>Cl<sub>2</sub>; 1:4) yielded two separate diastereomers of **12b**. Diastereomer A (160 mg, 31%): <sup>1</sup>H NMR  $(CDCl_3) \delta 8.17 \text{ (bd s, 1H)}, 7.75 \text{ (d, 1H, J= 8 Hz)}, 7.39 - 7.36 \text{ (m, 2H)}, 7.32 - 7.26 \text{ (m, 2H)}, 7.32 - 7.2$ 3H), 5.50 (dd, 1H, J= 2.5, 8 Hz), 5.19 (d, 1H, J= 9 Hz), 4.24 (m, 1H), 4.15 (m, 2H), 3.92 (dd, 1H, J= 2.5, 13.5 Hz), 3.56 (td, 1H, J= 2.5, 8 Hz), 3.09 (dd, 1H, J= 6.5, 13.5 Hz), 2.51 (dd, 1H, J= 11.2, 13.5 Hz), 1.04 - 0.88 (m, 28H);  $^{13}$ C NMR (CDCl<sub>3</sub>)  $\delta$  164.4, 151.3, 142.1, 138.5, 133.9, 128.2, 127.7, 100.5, 99.3, 85.4, 76.3, 67.0, 59.4, 39.8, 17.4, 17.2, 17.1, 17.0, 16.9, 16.8, 16.7, 13.4, 13.3, 12.9, 12.8, 12.3; IR(film) 3350, 3036, 2968, 1684, 1456, 1286, 1148, 1119, 1039 cm<sup>-1</sup>; Anal calcd for C<sub>28</sub>H<sub>44</sub>N<sub>2</sub>O<sub>7</sub>Si<sub>2</sub>: C, 58.30; H, 7.69; N, 4.86. Found: C, 58.46; H, 7.81; N, 4.67. Diastereomer B (280 mg, 54%): <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  7.95 (bd s, 1H), 7.41 (d, 1H, J= 8 Hz), 7.23 - 7.20 (m, 5H), 5.43 (d, 1H, J= 8 Hz), 5.27 (dd, 1H, J= 2.5, 8 Hz), 4.28 (m, 1H), 4.10 (m, 1H), 3.96 (m, 2H), 3.10 (dd, 1H, J= 7, 13 Hz), 2.95 (dd, 1H, J= 10.5, 13 Hz), 2.69 (d, 1H, J= 8 Hz), 1.046 - 0.88 (m, 28H);  $^{13}$ C NMR (CDCl<sub>3</sub>)  $\delta$  163.8, 150.5, 140.8, 138.3, 128.9, 128.6, 127.6, 100.7, 98.3, 87.2, 74.5, 67.9, 60.1, 39.9, 17.8, 17.6, 17.5, 17.4, 17.3, 17.2, 13.8, 13.4, 13.2, 12.8; IR(film) 3398, 3195, 3059, 2946, 2894, 2868, 1682, 1464, 1408, 1383, 1287, 1273, 1148, 1119, 1039 cm<sup>-1</sup>; Anal. calcd. for C<sub>28</sub>H<sub>44</sub>N<sub>2</sub>O<sub>7</sub>Si<sub>2</sub>: C, 58.30; H, 7.69; N, 4.86. Found: C, 58.16; H, 7.69; N, 4.66.

**Preparation of 12c.** Aldehyde **11** (180 mg, 0.36 mmol) was reduced with isopropyl magnesium chloride (2 eq.) as described above for the preparation of **12a**. Flash chromatography (EtOAc:CH<sub>2</sub>Cl<sub>2</sub>; 1:3) yielded two separate diastereomers of **12c**. In addition, 105 mg (58%) of **10** was isolated from the reaction. Diastereomer A (41 mg, 22%): <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  8.47 (bd s, 1H), 8.00 (d, 1H, J= 8 Hz), 5.65 (d, 1H, J= 8 Hz), 4.31 (dd, 1H, J= 2, 10 Hz), 4.23 (m, 1H), 4.19 (m, 1H), 3.95 (dd, 1H, J= 3, 13 Hz), 3.85 (dd, 1H, J= 2, 13 Hz), 2.91 - 2.73 (m, 2H), 1.58 (m, 1H), 1.06 - 0.95 (m, 34H); IR (film) 3400, 3191, 2945, 2868, 1693, 1464, 1385, 1119, 1038 cm<sup>-1</sup>.

Diastereomer B (38 mg, 20%): <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  9.51 (bd s, 1H), 8.05 (d, 1H, J= 8 Hz), 5.69 (d, 1H, J= 8 Hz), 4.18 (m, 1H), 4.08 (m, 1H), 3.97 (dd, 1H, J= 2, 15 Hz), 3.75 (dd, 1H, J= 2, 9 Hz), 3.49 (dd, 1H, J= 6, 9 Hz), 2.98 (dd, 1H, J= 6, 14 Hz), 2.65 (dd, 1H, J= 12, 14 Hz), 1.63 (m, 1H), 1.05 - 0.94 (m, 34H); IR (film) 3430, 3190, 2870, 1694, 1465, 1385, 1300, 1250, 1118, 1039, 921.

**Preparation of 13a.** Dess-Martin periodinane (526 mg, 1.10 mmol) was added to **12a** (243 mg, 0.44 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (20 mL) at 0°C. After stirring for 5 h at room temperature, additional Dess-Martin reagent (526 mg, 1.10 mmol) was added, and the reaction was stirred overnight. The mixture was diluted with EtOAc (75 mL), and washed with cold, saturated NaHCO<sub>3</sub> (50 mL), to which Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> (2.5 g) was added. The organic layer was washed with brine (25 mL), and dried over MgSO<sub>4</sub>. The residue was purified by flash chromatography (EtOAc: Hex; 1:4) to yield 213 mg of **13a** (88%). <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  8.68 (bd s, 1H), 8.03 (d, 1H, J= 8 Hz), 5.73 (dd, 1H, J= 1.8, 8.2 Hz), 4.34 (m, 1H), 4.16 (d, 1H, J= 14 Hz), 3.96 (dd, 1H, J= 2.6, 13.5 Hz), 3.69 (dd, 1H, J= 1.7, 9 Hz), 3.42 (dd, 1H, J= 11.5, 13.5 Hz), 2.23 (dd, 1H, J= 6.7, 13.5 Hz), 1.15 - 0.97 (m, 37 Hz); IR (soln) 3184, 3054, 2945, 2867, 1682, 1622, 1463, 1384, 1366, 1299, 1272, 1203, 1149, 1116, 1039 cm<sup>-1</sup>.

**Preparation of 13b.** A mixture of diastereomers of alcohol **12b** (205 mg, 0.36 mmol) was oxidized as described above. Flash chromatography (EtOAc:CH<sub>2</sub>Cl<sub>2</sub>; 1:9) yielded 147 mg of **13b** (72%). <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  8.25 (d, 1H, J= 8 Hz), 7.69 (dd, 2H, J= 1.5, 7 Hz), 7.64 (bd s, 1H), 7.45 (d, 1H, J= 7 Hz), 7.33 (dt, 1H, J= 1.5, 7 Hz), 5.75 (dd, 1H, J= 2.4, 8.3 Hz), 4.45 (m, 1H), 4.25 (d, 1H, J= 14 Hz), 4.01 (dd, 1H, J= 2.5, 14 Hz), 3.86 (dd, 1H, J= 2, 9 Hz), 3.62 (dd, 1H, J= 11.5, 13.5 Hz), 2.35 (dd, 1H, J= 7, 13.5 Hz), 1.11 - 0.96 (m, 28 H); <sup>13</sup>C NMR (CDCl<sub>3</sub>)  $\delta$  189.7, 163.7, 149.8, 139.2, 134.1, 132.8, 129.1, 128.5, 102.4, 95.7, 86.2, 66.7, 59.2, 39.1, 17.8, 17.7, 17.6, 17.5, 17.3, 17.2, 17.1, 13.8, 13.4, 13.2, 12.7; IR (film) 2946, 2868, 1710, 1462, 1446, 1277, 1249, 1114, 1037 cm<sup>-1</sup>; Anal. calcd. for: C<sub>28</sub>H<sub>42</sub>N<sub>2</sub>O<sub>7</sub>Si<sub>2</sub>: C, 58.51; H, 7.36; N, 4.87. Found: C, 58.40; H, 7.61; N, 4.84.

**Preparation of 13c.** A mixture of diastereomers of alcohol **12c** (92 mg, 0.18 mmol) was oxidized as described above. Flash chromatography (EtOAc: $CH_2Cl_2$ ; 1:4) yielded 54 mg of **13c** (59%). <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  8.99 (bd s, 1H), 7.99 (d, 1H, J= 8 Hz), 5.72 (d, 1H, J= 8 Hz), 4.34 (m, 1H), 4.12 (dd, 1H, J= 1, 14 Hz), 3.95 (dd, 1H, J= 2, 14 Hz), 3.74 (dd, 1H, J= 1, 9 Hz), 3.24 (dd, 1H, J= 12, 14 Hz), 2.75 (sept, 1H, J= 7 Hz), 2.39 (dd, 1H, J= 7, 14 Hz), 1.09 - 0.94 (m, 34H); IR (film) 3255, 2946, 2869, 1688, 1465, 1383, 1297, 1273, 1210, 1148, 1117, 1038, 993 cm<sup>-1</sup>.

**Preparation of 1a.** NH<sub>4</sub>F (73 mg, 1.96 mmol), and **13a** (51.4 mg, 0.09 mmol) were refluxed in MeOH (4 mL) for 2 h. The reaction mixture was cooled to room temperature, and quenched with saturated NaHCO<sub>3</sub> (1 mL). The solvents were removed in vacuo, and the salts were washed with EtOAc (25 mL), filtered, and concentrated. The residue was chromatographed (MeOH: EtOAc: Hexanes; 1:4:5) to yield 28 mg of **1a** (96%). mp: 195 - 196°C; <sup>1</sup>H NMR (CD<sub>3</sub>CN) & 8.09 (d, 1H, J= 8 Hz), 5.65 (d, 1H, J= 8 Hz), 4.25 (m, 1H), 3.84 - 3.65 (m, 3H), 3.14 (dd, 1H, J= 7, 14 Hz), 2.25 (dd, 1H, J= 7.5, 14 Hz), 1.15 (s, 9H); IR (film) 3390 (bd), 2962, 1684, 1457, 1419, 1365, 1299, 1227, 1091, 1042 cm<sup>-1</sup>; Anal. Calcd. for C<sub>14</sub>H<sub>20</sub>N<sub>2</sub>O<sub>6</sub>: C; 53.84; H, 6.45; N, 8.97. Found: C, 53.71; H, 6.47; N, 8.89.

**Preparation of 1b.** Deprotection of **13b** (90 mg, 0.16 mmol) with NH<sub>4</sub>F in MeOH, as described above for the preparation of **1a**, yielded 52 mg of **1b** (99%). mp: 188-190.5°C; <sup>1</sup>H NMR (MeOH-d<sub>4</sub>)  $\delta$  8.54 (d, 1H, J= 8 Hz), 7.72 (dd, 2H, J= 1.5, 7 Hz), 7.48 (dd, 1H, J= 1.5, 8 Hz), 7.36 (dd, 2H, J= 7, 8 Hz), 5.73 (d, 1H, J= 8 Hz), 4.42 (dd, 1H, J= 6, 7 Hz), 3.97 (m, 2H), 3.82 (dd, 1H, J= 3, 12 Hz), 3.42 (dd, 1H, J= 9, 14 Hz), 2.47 (dd, 1H, J= 7, 14 Hz); <sup>13</sup>C NMR (MeOH-d<sub>4</sub>)  $\delta$  192.2, 166.1, 151.8, 141.2, 135.6, 133.5, 129.9, 129.2, 102.2, 98.1, 89.7, 69.0, 60.5, 42.1; IR (film) 3361, 2957, 2862, 1708, 1691, 1679, 1447, 1306, 1093, 1038 cm<sup>-1</sup>; Anal. calcd. for C<sub>16</sub>H<sub>16</sub>N<sub>2</sub>O<sub>6</sub>: C, 57.83; H, 4.85; N, 8.43. Found: C, 56.13; H, 5.29; N, 7.79.

**Preparation of 1c.** Deprotection of **13c** (32 mg, 0.06 mmol) with NH<sub>4</sub>F in MeOH, as described above for the preparation of **1a**, yielded 12 mg of **1c** (67%). mp: 162-164°C; <sup>1</sup>H NMR (MeOH-d<sub>4</sub>)  $\delta$  8.21 (d, 1H, J= 8 Hz), 5.70 (d, 1H, J= 8 Hz), 4.32 (dd, 1H, J= 7, 14 Hz), 3.93 (m, 1H), 3.83 (dd, 1H, J= 3, 12 Hz), 3.69 (dd, 1H, J= 4, 12 Hz), 3.08 - 2.95 (m, 2H), 2.55 (dd, 1H, J= 7, 14 Hz), 1.12 (d, 3H, J= 6.5 Hz), 1.03 (d, 3H, J= 6.5 Hz); IR (film) 3350, 2880, 1680, 1453, 1262, 1170, 1048 cm<sup>-1</sup>.

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## REFERENCES

- 1. Hol'y, A. Nucleic Acids Research 1974, 1, 289.
- a) Yoshimura, Y.; Kano, F.; Miyazaki, S.; Ashida, N.; Sakata, S.; Haraguchi, K.; Itoh, Y.; Tanaka, H.; Miyasaka, T. *Nucleosides & Nucleotides* 1996, 15, 305.

Haraguchi, K.; Itoh, Y.; Tanaka, H.; Miyasaka, T. *Nucleosides & Nucleotides* **1995**, *14*, 417.

- a) Yoshimura, Y.; Otter, B. A.; Ueda, T.; Matsuda, A. Chem. Pharm. Bull. 1992, 40, 1761.
  b) Tatsuoka, T.; Imao, K.; Suzuki, K. Heterocycles 1986, 24, 617.
- Sarma, M. S. P.; Megati, S.; Klein, R. S.; Otter, B. A. Nucleosides & Nucleotides 1995, 14, 393.
- 5. Faivre-Buet, V.; Grouiller, A.; Descotes, G.; *Nucleosides & Nucleotides* 1992, 11, 1651.
- 6. Uteza, V.; Chen, G-R.; Tuoi, J. L. Q.; Descotes, G.; Fenet, B.; Grouiller, A. *Tetrahedron* 1993, 49, 8579.
- 7. Goodman, B. K.; Greenberg, M. M. J. Org. Chem. 1996, 61, 2.
- 8. Azhayev, A.; Gouzaev, A.; Hovinen, J.; Azhayeva, E.; Lönnberg, H. Tetrahedron Lett. 1993, 34, 6435.
- 9. Still, W. C.; Kahn, M.; Mitra, A. J. Org. Chem. 1978, 43, 2923.

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