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## Organocatalytic Asymmetric Michael Addition of Oxazolones to Arylsulfonyl Indoles: Facile Access to *syn*-Configured α,β-Disubstituted Tryptophan Derivatives

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Enantioselective Michael addition of oxazolones to in situ generated vinylogous imine intermediates is reported. A series of optically active 3-alkylindole derivatives with adjacent quaternary and tertiary stereocenters was obtained. The

#### resulting adducts can readily be converted into syn-configured $\alpha$ , $\beta$ -disubstituted tryptophan derivatives without compromising the stereoselectivities.

### Introduction

Tryptophan, an essential amino acid for many organisms, is an important structural unit and synthetic intermediate of various bioactive compounds and natural products.<sup>[1]</sup> Likewise,  $\alpha,\beta$ -disubstituted tryptophan structural motifs are widely found in biologically active molecules and naturally occurring compounds such as stephacidins,<sup>[2a-2d]</sup> cycloaplysinopsins,<sup>[2e,2f]</sup> tubastrindoles,<sup>[2g]</sup> dictazolines,<sup>[2g-2i]</sup> dictazoles,<sup>[2i]</sup> and waikialoid.<sup>[2j]</sup> Although this structural scaffold is very important, to the best of our knowledge, only one asymmetric catalytic version has been reported for the synthesis of  $\alpha,\beta$ -disubstituted tryptophan derivatives with methyl or ethyl groups at the  $\beta$ -position.<sup>[3]</sup> Furthermore, enantioselective routes to these compounds have been less explored.<sup>[4]</sup> Therefore, to explore an efficient approach for the synthesis of  $\alpha,\beta$ -disubstituted tryptophan derivatives with various substituents at the  $\alpha$ - and  $\beta$ -positions is still highly desired. In 2006, an innovative solution to the indole skeleton by using arylsulfonyl indole 1 was developed by the Petrini group.<sup>[5]</sup> In that work, compound 1<sup>[6]</sup> was used to generate highly active vinylogous imine intermediates in situ under basic conditions, and these species were then treated with nucleophiles to afford 3-substituted indole derivatives (Scheme 1).<sup>[7,8]</sup> Inspired by these elegant studies, and considering that oxazolone<sup>[9]</sup> is an excellent masked amino acid fragment,<sup>[3,10]</sup> herein we wish to describe the first stereoselective Michael addition of oxazolones to vinylogous imine intermediates generated in situ from arylsulfonyl-substituted indoles under simple chiral organocata-

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lysis. The resulting adducts can be further conveniently converted into *syn*-configured  $\alpha$ , $\beta$ -disubstituted tryptophan derivatives with various substituents at the  $\alpha$ - and  $\beta$ -positions without compromising the stereoselectivities.



Scheme 1. Nucleophilic Michael addition to vinylogous imine intermediates generated in situ from arylsulfonyl indoles 1.

### **Results and Discussion**

At the outset of our studies, arylsulfonyl indole 1a and oxazolone 2a were chosen as model substrates to optimize the reaction conditions, and the results are listed in Table 1. The natural cinchona alkaloids quinine (3a), cinchonidine (3b), and cinchonine (3c) only gave poor yields and enantioselectivities and various levels of diastereoselectivities (Table 1, entries 1-3). Cinchona alkaloid derived Hatakeyama catalyst 3d showed unsuccessful results under the reaction conditions (Table 1, entry 4).<sup>[11]</sup> Subsequently, our attention turned to bifunctional thiourea catalysts.<sup>[12]</sup> Takemoto catalyst 3e provided an excellent ee value, although with poor diastereoselectivity (Table 1, entry 5).<sup>[13]</sup> No satisfying result was achieved when thiourea catalyst 3f bearing a chiral diaminocyclohexane skeleton was used (Table 1, entry 6).<sup>[13c]</sup> Then, we began to investigate cinchona alkaloid derived thiourea catalysts 3g-i.<sup>[14]</sup> To our delight, catalyst **3** gave the best *dr* and *ee* value (Table 1, entry 10).<sup>[14e]</sup> To improve the yield, the reaction concentration was increased

Table 1. Optimization of the reaction conditions.<sup>[a]</sup>



Entry	Cat.	Base	Solvent	Yield <sup>[b]</sup>	$dr^{[c]}$	ee <sup>[d]</sup>
				[%]		[%]
1 <sup>[e,f]</sup>	3a	K <sub>3</sub> PO <sub>4</sub>	toluene	30	63:37	43
2 <sup>[e,f]</sup>	3b	K <sub>3</sub> PO <sub>4</sub>	toluene	35	71:29	36
3 <sup>[e,f]</sup>	3c	K <sub>3</sub> PO <sub>4</sub>	toluene	40	50:50	-19
4 <sup>[e,f]</sup>	3d	$K_3PO_4$	toluene	29	71:29	< -5
5[e,f]	3e	$K_3PO_4$	toluene	58	57:43	94
6 <sup>[e,f]</sup>	3f	$K_3PO_4$	toluene	50	62:38	89
7[e,f]	3g	$K_3PO_4$	toluene	64	57:43	-87
8 <sup>[e,f]</sup>	3h	$K_3PO_4$	toluene	44	77:23	-94
9 <sup>[e,f]</sup>	3i	$K_3PO_4$	toluene	60	69:31	95
10 <sup>[e,f]</sup>	3j	$K_3PO_4$	toluene	52	79:21	97
11 <sup>[f]</sup>	3j	$K_3PO_4$	toluene	62	79:21	96
12	3j	$K_3PO_4$	toluene	63	82:18	97
13	3j	Na <sub>2</sub> CO <sub>3</sub>	toluene	45	78:22	91
14	3j	$K_2CO_3$	toluene	56	78:22	95
15	3j	KF	toluene	30	70:30	81
16	3j	KF/Al <sub>2</sub> O <sub>3</sub>	toluene	45	76:24	90
17	3j	NaOH	toluene	50	71:29	91
18	3j	KOH	toluene	72	70:30	92
19	3j	Et <sub>3</sub> N	toluene	31	62:38	49
20	3j	$K_3PO_4$	benzene	60	82:18	97
21	3j	$K_3PO_4$	<i>p</i> -xylene	53	73:27	98
22	3j	$K_3PO_4$	mesitylene	50	71:29	96
23	3j	$K_3PO_4$	chlorobenzene	55	72:28	97
24	3j	$K_3PO_4$	chloroform	40	73:27	80
25	3j	$K_3PO_4$	DCM	40	80:20	95
26	3j	$K_3PO_4$	DCE	41	77:23	94

[a] Unless otherwise noted, all reactions were carried out with **1a** (0.11 mmol), **2a** (0.1 mmol), catalyst (0.02 mmol), and base (0.11 mmol) in solvent (1.0 mL) at 30 °C for 2 h. [b] Isolated yield. [c] Determined by analysis of the crude mixture by <sup>1</sup>H NMR spectroscopy. [d] Determined by HPLC analysis. [e] Solvent: 2 mL. [f] Reaction time: 5 h.

onefold by reducing the volume of solvent, and the product was obtained in 62% yield (Table 1, entry 11). Further study showed that the yield did not decrease upon shortening the reaction time from 5 to 2 h (Table 1, entry 12). To further optimize the reaction, other parameters such the base and solvent were examined. Nevertheless, better results in terms of yield, diastereoselectivity, and enantioselectivity (Table 1, entries 13–26) were not obtained relative to the results obtained with K<sub>3</sub>PO<sub>4</sub> and toluene.<sup>[8d]</sup>



With the optimized reaction conditions in hand, we then screened a series of arylsulfonyl indoles 1 and oxazolones 2 to establish the general utility of this asymmetric transformation. As summarized in Table 2, most arylsulfonyl indoles and oxazolones underwent the reaction smoothly to afford syn-selective addition products with good results. Substrates with a substituent in the para position of the phenyl ring of  $R^1$  gave a slightly higher yield and enantioselectivity than did substrates with a substituent in the meta position. For example, 4-F-substituted 1b gave the product in 69% yield with 90% ee, whereas 3-F-substituted 1e gave the product in only 64% yield with 88% ee (Table 2, entry 2 vs. 5). A similar phenomenon was also observed in the reaction of 1c compared with that of 1f (Table 2, entry 3 vs. 6). However, an even greater difference was achieved when ortho-substituted 1q was used (Table 2, entry 17 vs. 19). It is noteworthy that when 1m was treated with 2a under the optimized reaction conditions, only 50% ee could be obtained (Table 2, entry 13). This seems to imply that sterics at the 2-position of the indole core play an important role in the enantioselectivity of the reaction, which is in line with previous reports.<sup>[8a,8c,8d,8h]</sup> Nevertheless, if the R group was changed from methyl to phenyl, relatively low stereoselecti-

Table 2. Asymmetric Michael addition of oxazolones to ary lsulfonyl indoles,  $^{\left[ a\right] }$ 



Entry	1: R, R <sup>1</sup>	2	4	Yield <sup>[b]</sup>	$dr^{[c]}$	$ee^{[d]}$
				[%]		[%]
1	1a: Me, Ph	2a	4a	63	82:18	97
2	<b>1b</b> Me, 4-FC <sub>6</sub> H <sub>4</sub>	2a	4b	69	81:19	90
3	1c: Me, $4$ -ClC <sub>6</sub> H <sub>4</sub>	2a	4c	65	83:17	97
4	<b>1d</b> : Me, $4$ -BrC <sub>6</sub> H <sub>4</sub>	2a	4d	70	82:18	96
5	<b>1e</b> : Me, $3 - FC_6H_4$	2a	4e	64	79:21	88
6	1f Me, $3-ClC_6H_4$	2a	<b>4</b> f	52	79:21	91
7	<b>1g</b> : Me, 4-MeC <sub>6</sub> H <sub>4</sub>	2a	4g	81	82:18	84
8	<b>1h</b> : Me, 4-OMeC <sub>6</sub> H <sub>4</sub>	2a	4h	77	82:18	91
9	<b>1i</b> : Me, 3,4-(OMe) <sub>2</sub> C <sub>6</sub> H <sub>4</sub>	2a	<b>4</b> i	50	90:10	97
10	<b>1j</b> : Me, 3-CF <sub>3</sub> C <sub>6</sub> H <sub>4</sub>	2a	4j	65	77:23	89
11	<b>1k</b> : Ph, 4-BrC <sub>6</sub> H <sub>4</sub>	2a	4k	55	67:33	90
12	<b>11</b> : Ph, 2-OMeC <sub>6</sub> H <sub>4</sub>	2a	41	40	87:13	93
13	1m: H, Ph	2a	4m	60	75:25	50
14 <sup>[e]</sup>	<b>1n</b> : Me, 3,4-Cl <sub>2</sub> C <sub>6</sub> H <sub>4</sub>	2b	4n	88	58:42	95
15 <sup>[e]</sup>	10: Me, 1-naphthyl	2b	<b>4</b> 0	74	87:13	80
16 <sup>[e]</sup>	<b>1p</b> : Me, 3-MeC <sub>6</sub> H <sub>4</sub>	2b	4p	51	58:42	91
17 <sup>[e]</sup>	1q: Me, 2-OMeC <sub>6</sub> H <sub>4</sub>	2b	4q	45	87:13	81
18 <sup>[e]</sup>	1c: Me, $4$ -ClC <sub>6</sub> H <sub>4</sub>	2b	4r	62	69:31	95
19 <sup>[e]</sup>	<b>1h</b> : Me, 4-OMeC <sub>6</sub> H <sub>4</sub>	2b	4s	71	75:25	86

[a] Unless otherwise noted, all reactions were carried out with 1 (0.11 mmol), 2 (0.1 mmol), 3j (0.02 mmol), and  $K_3PO_4$  (0.11 mmol) in toluene (1.0 mL) at 30 °C for 2 h. [b] Isolated yield. [c] Determined by analysis of the crude mixture by <sup>1</sup>H NMR spectroscopy. [d] Measured by using chiral stationary phase HPLC. [e] Reaction time: 4 h.

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vity was observed (Table 2, entry 4 vs. 11), likely because larger steric hindrance weakens the positive  $\pi$ - $\pi$  interaction of the "closed" conformation of the transition state (see the Supporting Information).<sup>[15]</sup> As described in the literature,<sup>[16]</sup> benzyl-substituted oxazolone **2b** gave slightly inferior results compared with methyl-substituted **2a** (Table 2, entry 3 vs. 18 and entry 8 vs. 19).

The absolute configuration of the two contiguous stereocenters of Michael addition product **4a** was unambiguously assigned as (10R,11S) by X-ray diffraction analysis (Figure 1).<sup>[17]</sup> The absolute configurations of the other products were assigned by analogy.



Figure 1. X-ray crystal structure of compound 4a.

The synthetic versatility of the adducts was illustrated by a further transformation. Adduct **4d** was readily converted into *syn*-selective  $\alpha$ , $\beta$ -disubstituted tryptophan derivative **5** in 88% yield by ring opening of the oxazolone subunit with sodium methoxide<sup>[3,10h]</sup> without a decrease in the diastereoselectivity and enantioselectivity (Scheme 2).



Scheme 2. Transformation of adduct 4d into 5.

### Conclusions

In conclusion, we have developed the first highly stereoselective Michael addition reaction of oxazolones to vinylogous imine intermediates generated in situ from arylsulfonyl indoles under chiral thiourea catalysis. A series of optically active C-3 alkyl-substituted indole derivatives with adjacent quaternary and tertiary stereocenters was obtained, and the resulting adducts could be readily converted into *syn*-configured  $\alpha,\beta$ -disubstituted tryptophan derivatives with various substituents at the  $\alpha$ - and  $\beta$ -positions without a decrease in the diastereoselectivities and enantioselectivities. Further investigations to broaden the scope of this type of transformation are currently underway.

## **Experimental Section**

**General Procedure:** An ordinary test tube equipped with a magnetic stirring bar was charged with catalyst **3j** (0.02 mmol), arylsulfonyl indole **1** (0.11 mmol), oxazolone **2** (0.1 mmol),  $K_3PO_4$  (0.11 mmol), and toluene (1 mL) and then sealed in air. After stirred at 30 °C for 2 h, the reaction mixture was purified by flash chromatography on silica gel (ethyl acetate/petroleum ether = 1:5–1:20) to afford product **4**. The product was confirmed by <sup>1</sup>H NMR and <sup>13</sup>C NMR spectroscopy. The diastereomeric ratio was determined by analysis of the crude material by NMR spectroscopy, and the enantiomeric excess was determined by chiral-phase HPLC analysis.

**Supporting Information** (see footnote on the first page of this article): Analytical data for all prepared compounds with copies of the <sup>1</sup>H NMR and <sup>13</sup>C NMR spectra and HPLC chromatographs.

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