

An Improved Method for the Synthesis of 1-Iodoalkynes

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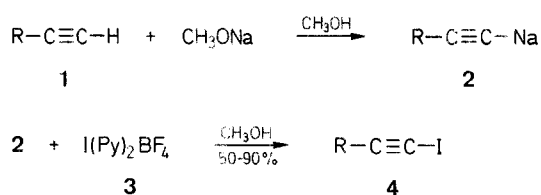
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A simple and general one-pot procedure for the synthesis of 1-iodoalkynes is described based on the reaction of terminal acetylenes with bis(pyridine)iodine(I) tetrafluoroborate in the presence of sodium methoxide. The method now reported is superior to the literature procedures in scope, handling, and often yield.

1-Haloalkynes are an interesting group of compounds due to their particular reactivity in substitution, and Cadiot-Chodkiewicz coupling reactions and also to the unexpected effect of the halogens in directing the addition of electrophiles to the triple bond.¹ To date the most general method for the synthesis of 1-iodoalkynes **4** is the reaction of iodine with the corresponding acetylide in liquid ammonia at ca. -30°C .^{1,2}

1-Iodoalkynes **4** also have been produced from the reaction of terminal alkynes and sodium hypoiodite in some instances, but low yields are reported due to side reactions caused by the instability of these compounds in alkaline medium and also due to problems in the purification step.^{1,3} On the other hand, the reaction of terminal acetylenes with iodine in methanol assisted by silver ion leads to a mixture of 1-iodoalkynes, 1,2-diiodoalkenes and 1,1-diiodoketones.⁴ Recently, we have reported the reaction of alkynes with (**3**)⁵ leading to the addition of I^+Nu^- to the triple bond.⁶ On this basis it seemed of interest to check the reactivity of **3** towards terminal acetylenes as a potential reagent for the synthesis of compounds **4**.

We have now found that sodium acetylides **2** (prepared *in situ* by reaction of sodium methoxide with the corresponding acetylene) react with **3** at room temperature in methanol to afford 1-iodoalkynes **4** with good to excellent yields in a clean process (Table 1).



All compounds were fully characterized by their IR, ^1H -, ^{13}C -NMR and in some cases MS spectra (Table 2). Data for known compounds were found to be in good agreement with those reported in the literature.

The mildness of the reaction conditions allows a variety of functional groups to be present in R in the starting acetylene **1** (Table 1). Since no excess of alkali is used, the 1-iodoalkynes **4** synthesized are stable in the reaction medium and can be easily isolated after an usual work-up procedure.

The general method reported herein for the synthesis of compounds **4** gives similar results as those in the bibliography^{1,2} when applied to simple acetylenes but gives much superior yields when these contain a functionalized chain R. On the other hand, our method is superior since the synthesis is carried out in methanol at room temperature avoiding the always cumbersome use of liquid ammonia as solvent. For the above reasons it could become the method of choice for compounds **4**.

Table 1. 1-Iodoalkynes **4** Prepared

Prod- uct	R	Yield ^a (%)	m.p. (°C) or b.p. (°C)/mbar	Molecular Formula ^b or Lit. m.p. (°C) or b.p. (°C)/mbar
4a	<i>n</i> -C ₄ H ₉	88	76/27	80/33 ⁴
4b	<i>n</i> -C ₆ H ₁₃	76	103/13	95/10 ⁷
4c	CH ₂ =CHCH ₂	95	80/93	C ₅ H ₅ I (192.0)
4d	CH ₂ =C(CH ₃)	75	83/93	C ₅ H ₅ I (192.0)
4e	C ₆ H ₅	83	135/13	98/1.6 ⁴
4f	HOCH ₂	50	44–45	43–44 ⁸
4g	CH ₃ OCH ₂	61	77/27	74/27 ⁸
4h	CH ₂ =CHCH(CH ₃)OCH ₂	70	oil	C ₇ H ₉ IO (236.1)
4i	ClCH ₂	85	75/27	47/5.3 ³
4j	BrCH ₂	88	98/27	60/2.7 ³

^a Based on starting acetylene.^b Satisfactory microanalyses obtained: C ± 0.35, H ± 0.24.**Table 2.** Spectral Data of Compounds **4**

Prod- uct	IR (Film) ^a ν _{C≡C} (cm ⁻¹)	¹ H-NMR (CDCl ₃ /TMS) ^b δ (ppm)	¹³ C-NMR (CDCl ₃) ^b δ (ppm) ^c
4a	2190	0.9 (t, 3H, <i>J</i> = 6 Hz, CH ₃); 1.2–2.5 (m, 4H, CH ₂ CH ₂); 2.3 (t, 2H, <i>J</i> = 6 Hz, CH ₂ C≡C)	3.6; 15.7; 22.6; 23.9; 32.6; 96.3
4b	2190	0.8–1.6 [m, 11H, (CH ₂) ₄ CH ₃]; 2.3 (t, 2H, <i>J</i> = 6 Hz, CH ₂ C≡C)	4.9; 15.3; 22.0; 23.8; 29.7; 29.7; 32.5; 95.5
4c	2190	3.0 (d, 2H, <i>J</i> = 5 Hz, CH ₂ C≡C); 5.2 (br s, 2H, CH ₂ =C); 5.5 (m, 1H, C=CH) ^d	0.6; 25.1; 91.2; 116.7; 131.9 ^e
4d	2150	1.9 (s, 3H, CH ₃); 5.1 (br s, 1H, CH=C); 5.3 (br s, 1H, CH=C)	9.1; 25.1; 97.4; 125.8; 128.8
4e	2170	6.8–7.4 (m, H _{arom})	10.6; 96.3; 124.1; 130.1; 130.6; 134.0
4f	2195 ^f	3.5 (br s, 1H, OH); 4.4 (s, 2H, CH ₂ O)	7.4; 54.3; 94.7
4g	2195	3.4 (s, 3H, CH ₃); 4.2 (s, 2H, CH ₂ O)	7.9; 59.6; 63.1; 92.5
4h	2195 ^f	1.2 (d, 3H, <i>J</i> = 10 Hz, CH ₃); 4.3 (s, 2H, OCH ₂ C≡C); 4.9–5.7 (m, 4H, CH ₂ =CHCHO)	6.6; 23.3; 58.6; 77.6; 93.0; 119.1; 140.9
4i	2190	4.2 (s, CH ₂)	9.2; 34.7; 90.8
4j	2190	4.0 (s, CH ₂)	10.6; 19.6; 91.8

^a Recorded on a Perkin Elmer 298 IR spectrometer.^b Recorded on a Varian FT-80A spectrometer.^c With reference to the solvent as an internal standard.^d Measured in CCl₄ using D₂O capillary lock.^e Measured in CCl₄.^f Recorded as nujol null.

organic layer successively washed with 1 normal sulfuric acid (25 ml) and a 5% aqueous solution of sodium thiosulfate (25 ml). The organic layer is dried with sodium sulfate, and evaporated under vacuum to yield the corresponding 1-iodoalkynes **4** as the residue (GC purity of the crude product > 90%). Products **4** are purified by distillation or recrystallization in methanol (Table 1).

This work was supported in part by the Comisión Asesora de Investigación Científica y Técnica (CAICYT Project No. 2571/83). Two of us (JMG and MAR) acknowledge the Ministerio de Educación y Ciencia for a fellowship.

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- (2) Jäger, V., Viehe, H.G., in: Houben-Weyl, *Methoden der Organischen Chemie*, 4th ed., Müller, E. (ed.), Vol. 5/2a, Georg Thieme Verlag, Stuttgart, 1977, pp. 601–610.
- (3) Hatch, L.J., Mangold, D.J. *J. Am. Chem. Soc.* **1955**, 77, 176.
- (4) Heasley, V.L., Shellhamer, D.F., Heasley, L.E., Yaeger, D.B., Heasley, G.E. *J. Org. Chem.* **1980**, 45, 4649.
- (5) Barluenga, J., González, J.M., Campos, P.J., Asensio, G. *Angew. Chem.* **1985**, 97, 341; *Angew. Chem. Int. Ed. Engl.* **1985**, 24, 319.
- (6) Barluenga, J., Rodríguez, M.A., González, J.M., Campos, P.J., Asensio, G. *Tetrahedron Lett.* **1986**, 27, 3303.
- (7) Vaughn, T.H. *J. Am. Chem. Soc.* **1933**, 55, 3453.
- (8) Beilsteins Handbuch der Organischen Chemie **1**, 455.

1-Iodoalkynes **4**; General Procedure:

To a two necked flask containing dry methanol (25 ml) under argon atmosphere, sodium (0.125 g, 5.4 mmol) is added and the mixture stirred until total dissolution. The corresponding acetylene **1** (5 mmol) is then slowly added followed by a stoichiometric amount of bis(pyridine)iodine(I) tetrafluoroborate (**3**; 1.86 g, 5 mmol). After stirring at room temperature for 30 min, the resulting solution is hydrolyzed with water (25 ml), extracted with dichloromethane (3 × 25 ml) and the

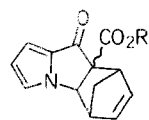
Errata and Addenda 1987

Hall, G., Sugden, J. K., Waghela, M. B.

Page 10. Line 3 of the Abstract should read: dropyrolizines ...

Page 14. The first word of Section 3.11. should be: Benzo[*b*]pyrroli-
zines.

Page 15. Formula 27 should be:



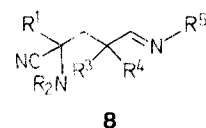
27

Page 15. The product referred to in Section 4.6., lines 4-5, should be:
10*H*-pyrrolizino[1,2-*b*]quinoline

Page 17. In Section 7., line 4 of the second paragraph should read:
34.¹⁸² ...

Ahlbrecht, H., von Daacke, A.

Page 24. Formula 8 should be:



8

Costisella, B., Keitel, I.

Page 45. In the heading of the experimental procedure, 6 should read 3
and 8 should read 7.

Stoss, P., Merrath, P., Schlüter, G.

Page 174. Numbers 1 and 3 should be exchanged in formula 2a-f.

Singh, G., Deb, B., Ila, H., Junappa, H.

Page 286. Compounds 1 are 2-aryl-2-arylthio ketene dithioacetals.

Asaad, F. M., Becher, J., Möller, J., Varma, K. S.

Page 301. Under the reaction scheme, the X group in compounds 3b,d
and 4b,d should be CO₂C₂H₅.

Legrel, P., Baudy-Floc'h, M., Robert, A.

Page 306. The title should read: A One-Pot Synthesis of α -Halohydrazides
from 2,2-Dicyano oxiranes.

Page 306. In the table under the reaction scheme, the second heading R¹
should be R².

van der Goorbergh, J. A. M., van der Steeg, M., van der Gen, A.

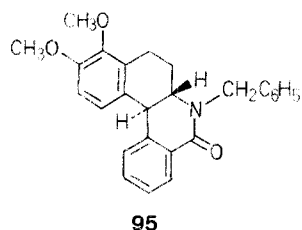
Pages 314-317. The systematic names for the heterocycles involved are:
4,5-dioxo-3,4-dihydro-2*H*,5*H*-thiopyrano[3,2-*c*][1]benzopyrans 4 (RF
24756), 4,5-dioxo-2*H*,5*H*-thiopyrano[3,2-*c*][1]benzopyrans 7 (RF
24756j), and 4,5-dioxo-1,3,4,4a,5,10b-hexahydro-2*H*-[1]benzopy-
rano[4,3-*b*]pyridines 8 (RF 24539).

Atanasi, O. A., Filippone, P., Santensanio, S., Serra-Zanetti, F.

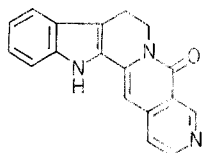
Page 382. In the table under the reaction scheme, R³ for 1b should be
CO₂C₂H₅ and R³ for 1c should be CO₂CH₃.

Campbell, A. I., Lenz, G. R.

Pages 428 and 446. Formulae 95 and 298 should be:



95

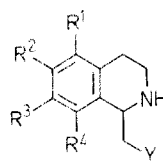


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Page 437. The heading for Table 3 should be: Intermolecular ...

Pelletier, J. C., Cava, M. P.

Page 476. Formula 1a-m should be:



1a-m

L'abbé, G.

Page 528. Compound 45 should be named: 3-(2-pyridyl)-2,4-dithioxo-
3,4-dihydro-2*H*-pyrido[1,2-*a*][1,3,5]triazine (RF 9177).

Evans, R. D., Schauble, J. H.

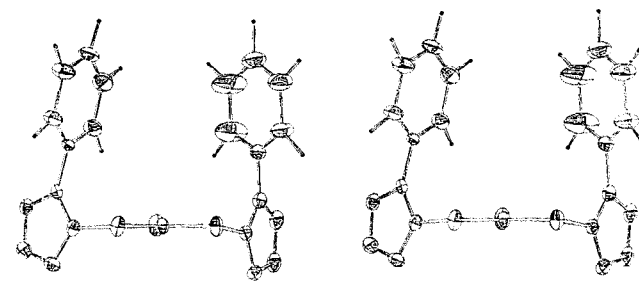
Page 551. Compounds 10 and 11 are tricyclo[2.2.1.0^{2,6}]heptane deriva-
tives.

Takeda, K., Tsuboyama, K., Hoshino, M., Kishino, M., Ogura, H.

Page 559. The Y-group for 2g and 2j should be furfuryloxy.

Takeda, K., Tsuboyama, K., Takayanagi, H., Ogura, H.

Page 560. The following figure should appear after the 4th paragraph:

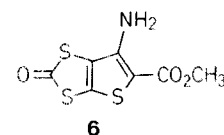


Eicher, T., Stapperferne, U.

Page 625. Compounds 13a,b are 6,7-dihydrofuro[2,3-*b*]pyridines
(RF 7431), and compounds 15a,b are 1,4-dihydrocyclopentimidazoles
(RF 5892).

Dölling, W., Augustin, M., Ihrke, R.

Page 655. Formula 6 should be:



6

Mikolajczyk, M., Balczewski, P.

Page 661. The second paragraph of ref. 21 should be ref. 22; refs. 22 and
23 should be 23 and 24, respectively.

Rösch, W., Regitz, M.

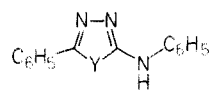
Page 692. Compounds 21a,b are 2*H*-1,2,3-diazaphospholes.

Tietze, L.-F., Brumby, T., Pretor, M.

Page 702. Compounds 8 and 9 are 4a,10b-dihydro-4*H*,5*H*-pyrano[3,4-
c][1]benzopyran-2-carboxylic esters.

Wamhoff, H., Zahran, M.

Page 877. Formula 18a,b should be:



18a,b

Castaldi, G., Giordano, C.

Page 1039. The target compounds 3 are 1-bromoalkyl aryl ketones.