



Original article

Synthesis and biological evaluation of substituted (thieno[2,3-d]pyrimidin-4-ylthio)carboxylic acids as inhibitors of human protein kinase CK2

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ABSTRACT

A novel series of substituted (thieno[2,3-d]pyrimidin-4-ylthio)carboxylic acids has been synthesized and tested in vitro towards human protein kinase CK2. It was revealed that the most active compounds inhibiting CK2 are 3-[[5-(4-methylphenyl)thieno[2,3-d]pyrimidin-4-yl]thio]propanoic acid and 3-[[5-(4-ethoxyphenyl)thieno[2,3-d]pyrimidin-4-yl]thio]propanoic acid (IC_{50} values are 0.1 μ M and 0.125 μ M, respectively). Structure–activity relationships of 28 tested thienopyrimidine derivatives have been studied and binding mode of this chemical class has been predicted. Evaluation of the inhibitors on seven protein kinases revealed considerable selectivity towards CK2.

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1. Introduction

Protein kinase CK2 has been intensively studied for more than 50 years, however, its biological role and regulation of its activity is still remains unclear [1]. Possessing constitutive catalytic activity with the ability to phosphorylate more than 300 physiological substrates (their number is growing continuously), this kinase stands out noticeably against a background of the kinase family [2]. Naturally, these features make CK2 appear at extremely diverse points of cell signaling pathways and be involved in processes leading to the development of various disorders, especially cancer. Increased CK2 expression and/or activity both have been shown in malignant cells and cells of intensively proliferating tissues [3]. This is related to the fact that CK2 can function as a general antiapoptotic agent [4], and its activity is quite important for determining whether cell will survive or die through apoptosis. Moreover, some viruses use this enzyme to phosphorylate their own proteins [5–8]; also this kinase is involved in development of inflammation processes [9]. Therefore, nowadays CK2 is considered as druggable

protein kinase target and could be used for the development of antitumor, antiviral and anti-inflammatory drugs [10].

Effective approach to study the function of protein kinases is the use of small molecule inhibitors. Taking into account that CK2 is not activated in response to external stimuli and exists in cell only in active form the mentioned approach is especially useful for CK2 targeting [11]. There are many compounds inhibiting CK2 that have been discovered [12]; nevertheless, none of them is in clinical use. So far, the only one orally administered highly selective and potent CK2 inhibitor CX-4945 has entered phase I clinical trials due to its broad spectrum antiproliferative activity in multiple cancer cell lines including inflammatory breast cancer [13]. Thus, development of new potent and selective CK2 inhibitors is a task of great importance which provides a powerful tool to extend our knowledge about CK2 function as well as to regulate its activity both in case of health and disease.

The goal of our research was to identify potent and selective inhibitors of CK2 among thieno[2,3-d]pyrimidines. Compounds from this chemical class are known to inhibit activity of ErbB, PDGF, VEGFR-2, FLT3 and Tie2 kinases [14–16], but there are no data available on their activity towards CK2. Data regarding synthesis and study of (thieno[2,3-d]pyrimidin-4-ylthio)carboxylic acids are rare in literature [17]. Taking into account that many potent CK2 inhibitors published at the moment contain carboxyl group in their

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structure [18–21], we focused our studies on (thieno[2,3-d]pyrimidin-4-ylthio)carboxylic acids.

2. Experimental section

2.1. Chemistry

General synthetic procedure for compounds **5a–u**, **6a–g** was performed according to Scheme 1.

Nuclear magnetic resonance spectra were recorded on a Varian Mercury VRX-400 spectrometer using DMSO- d_6 as solvent and tetramethylsilane as internal standard. Chemical shift values (δ) are quoted in ppm, and coupling constants (J) in Hz.

2.1.1. Substituted 2-aminothiophenes **1**

A mixture 108 mL (1 mol) of ethyl cyanoacetate, 32 g (1 mol) of elemental sulfur and 1 mol of corresponding ketone in 200 ml of diethyl ether was stirred at room temperature. To this mixture 80 ml of diethylamine was added for 12 h. The solid product was collected by filtration and washed with aqueous alcohol (1:1), yield 50–80%.

2.1.2. Substituted thieno[2,3-d]pyrimidin-4(3H)-ones **2**

1 mol formamide was added to 1 mol 2-aminothiophene and the mixture was heated at 160–170 °C for 24 h. The mixture was diluted with isopropyl alcohol. The solid product was collected by filtration and washed with isopropyl alcohol and water and dried, yield 80%.

2.1.3. Substituted 4-chlorothieno[2,3-d]pyrimidines **4**

A mixture of 0.33 mol thieno[2,3-d]pyrimidin-4(3H)-one, 333 mL POCl₃ and 0.33 mol PCl₅ was boiled until a homogeneous solution was formed. POCl₃ was distilled off in vacuum and equal volume of chloroform was slowly added into the warm mixture and pH was adjusted to 7 with 10% aqueous NaOH. The organic phase was separated and dried over Na₂SO₄, filtered, and then solvent was evaporated in vacuum. The residue was crystallized from isopropyl alcohol or hexane. The yield of this product was 80%.

2.1.4. Substituted (thieno[2,3-d]pyrimidin-4-ylthio)carboxylic acids **5a–u**

Substituted thieno[2,3-d]pyrimidine-4(3H)-thiones **3** were synthesized as described in Reference [22]. Compounds **5a–u** shown in Table 1 were synthesized on the base of **3**. To 0.01 mol of thieno

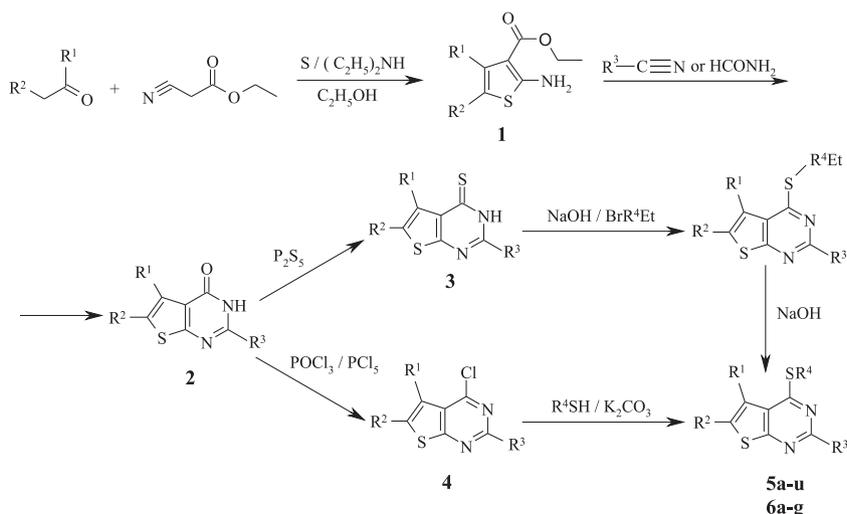
[2,3-d]pyrimidine-4(3H)-thione in 5 mL ethanol and 0.01 mol 2 M of aqueous sodium hydroxide the 0.0105 mol of bromine substituted carboxylic acid ethyl ester was added. The mixture was refluxed for 5 min; in some reactions with ethyl 3-bromopropionate the concentration of this reagent was slightly elevated up to 0.012 mol and the mixture was heated under reflux for 30 min. Water and dichloromethane were added to the mixture and the resulted phases were separated. The aqueous phase was extracted 2 times with CH₂Cl₂, the combined organic dried over Na₂SO₄, filtered, and concentrated in vacuum to afford the ester. The ester was hydrolyzed to solid acid.

2.1.4.1. 2-(5,6,7,8-Tetrahydro[1]benzothieno[2,3-d]pyrimidin-4-ylthio)propanoic acid (5a). Yield 80%; m.p. 162–163 °C; ¹H NMR (400 MHz, DMSO- d_6): δ : 1.58 (3H, d, CH₃, J = 7.3 Hz), 1.83 (4H, m, CH₂, -(CH₂)₄-), 2.82–2.98 (4H, m, CH₂, -(CH₂)₄-), 4.73 (1H, q, J = 7.3 Hz, J = 7.3 Hz), 8.69 (1H, s, 2-H). ¹³C NMR (100 MHz, DMSO- d_6): δ : 17.50, 21.98, 25.20, 26.16, 41.00, 95.41, 126.83, 127.23, 136.57, 150.92, 161.08, 164.58, 172.57. Anal. Calc. for C₁₃H₁₄N₂O₂S₂: C, 53.04; H, 4.79; S, 21.78. Found: C, 53.02; H, 4.76; S, 21.77.

2.1.4.2. 2-[(7-Methyl-5,6,7,8-tetrahydro[1]benzothieno[2,3-d]pyrimidin-4-ylthio)propanoic acid (5b). Yield 81%; m.p. 134–135 °C; ¹H NMR (400 MHz, DMSO- d_6): δ : 1.06 (3H, d, CH₃, J = 6.5 Hz), 1.46 (1H, m), 1.57 (3H, d, CH₃, J = 7.3 Hz), 1.92 (2H, m), 2.42 (1H, m), 2.91 (2H, m), 3.14 (1H, m), 4.73 (1H, q, J = 7.3 Hz, J = 7.3 Hz), 8.69 (1H, s, 2-H), 12.92 (bs, OH). ¹³C NMR (100 MHz, DMSO- d_6): δ : 17.58, 20.95, 25.84, 28.25, 30.08, 33.12, 40.90, 95.50, 126.45, 127.10, 135.94, 150.76, 161.09, 164.75, 172.49. Anal. Calc. for C₁₄H₁₆N₂O₂S₂: C, 54.52; H, 5.23; S, 20.79. Found: C, 54.50; H, 5.21; S, 20.77.

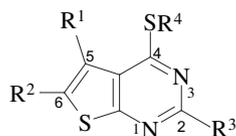
2.1.4.3. 2-[(5,6-Dimethylthieno[2,3-d]pyrimidin-4-ylthio)propanoic acid (5c). Yield 83%; m.p. 208–209 °C; ¹H NMR (400 MHz, DMSO- d_6): δ : 1.58 (3H, d, CH₃, J = 7.3 Hz), 2.47 (3H, s, CH₃), 2.5 (3H, s, CH₃), 4.73 (1H, q, J = 7.3 Hz, J = 7.3 Hz), 8.69 (1H, s, 2-H). ¹³C NMR (100 MHz, DMSO- d_6): δ : 13.49, 14.27, 17.45, 41.00, 95.47, 124.73, 128.14, 133.40, 150.62, 161.27, 163.93, 172.55. Anal. Calc. for C₁₁H₁₂N₂O₂S₂: C, 49.23; H, 4.51; S, 23.90. Found: C, 49.21; H, 4.50; S, 23.88.

2.1.4.4. [(5-Phenylthieno[2,3-d]pyrimidin-4-ylthio)acetic acid (5d). Yield 80%; m.p. 174–175 °C; ¹H NMR (400 MHz, DMSO- d_6): δ : 3.95 (2H, s), 7.48 (5H, m), 7.80 (1H, s, C₆H), 8.84 (1H, s, 2-H). ¹³C NMR (100 MHz, DMSO- d_6): δ : 31.76, 95.41, 125.15, 125.67, 127.81, 128.38,



Scheme 1. The synthesis pathway for compounds **5a–u**, **6a–g**.

Table 1
Structure and inhibitory activity towards CK2 of the (thieno[2,3-d]pyrimidin-4-ylthio)carboxylic acids.



Compound	R ¹	R ²	R ³	R ⁴	IC ₅₀ (μM)
5a	–CH ₂ –CH ₂ –CH ₂ –CH ₂ –		H	CH(CH ₃)COOH	17
5b	–CH ₂ –CH ₂ –CH(CH ₃)–CH ₂ –		H	CH(CH ₃)COOH	20
5c	CH ₃	CH ₃	H	CH(CH ₃)COOH	>33
5d	C ₆ H ₅	H	H	CH ₂ COOH	30
5e	–CH ₂ –CH ₂ –CH(CH ₃)–CH ₂ –		H	CH ₂ COOH	1.78
5f	–CH ₂ –CH ₂ –CH ₂ –CH ₂ –		H	CH(C ₂ H ₅)COOH	>33
5g	CH ₃	CH ₃	H	CH ₂ COOH	17.75
5h	C ₆ H ₅	H	H	CH(CH ₃)COOH	23
5i	4-ClC ₆ H ₄	H	H	CH ₂ COOH	32
5j	4-ClC ₆ H ₄	H	H	CH(C ₂ H ₅)COOH	>33
5k	H	C ₆ H ₅	H	CH ₂ COOH	1.25–1.75
5l	4-CH ₃ C ₆ H ₄	H	H	CH(CH ₃)COOH	31
5m	4-CH ₃ C ₆ H ₄	H	H	CH(C ₂ H ₅)COOH	>33
5n	4-BrC ₆ H ₄	H	H	CH ₂ COOH	14
5o	–CH ₂ –CH ₂ –CH ₂ –CH ₂ –		H	CH ₂ C ₆ H ₄ COOH– <i>p</i>	>33
5p	4-ClC ₆ H ₄	H	H	CH ₂ –CH ₂ COOH	0.175
5q	CH ₃	CH ₃	H	CH ₂ C ₆ H ₄ COOH– <i>m</i>	17.75
5r	–CH ₂ –CH ₂ –CH ₂ –CH ₂ –		CH ₃	CH ₂ –CH ₂ COOH	15.8
5s	–CH ₂ –CH ₂ –CH ₂ –CH ₂ –		H	CH ₂ –(CH ₂) ₂ COOH	>33
5t	CH ₃	CH ₃	H	CH ₂ –CH ₂ COOH	11.3
5u	C ₆ H ₅	CH ₃	H	CH ₂ COOH	>33
6a	4-CH ₃ C ₆ H ₄	H	H	CH ₂ –CH ₂ COOH	0.1
6b	3,4-(CH ₃) ₂ C ₆ H ₃	H	H	CH ₂ –CH ₂ COOH	0.175
6c	H	CH ₃	H	CH ₂ COOH	4
6d	4-C ₂ H ₅ OC ₆ H ₄	H	H	CH ₂ –CH ₂ COOH	0.125
6e	4-FC ₆ H ₄	H	H	CH ₂ –CH ₂ COOH	1.2
6f	H	CH ₃	H	CH ₂ –CH ₂ COOH	7
6g	H	C ₆ H ₅	CH ₃	CH ₂ –CH ₂ COOH	3.5

129.93, 134.82, 135.04, 151.49, 163.29, 165.80, 169.33. Anal. Calc. for C₁₄H₁₀N₂O₂S₂: C, 55.61; H, 3.33; S, 21.21. Found: C, 55.60; H, 3.31; S, 21.20.

2.1.4.5. [(7-Methyl-5,6,7,8-tetrahydro[1]benzothieno[2,3-d]pyrimidin-4-yl)thio]acetic acid (**5e**). Yield 85%; m.p. 209–208 °C; ¹H NMR (400 MHz, DMSO-*d*₆): δ: 1.06 (3H, d, CH₃, *J* = 6.5 Hz), 1.48 (1H, m), 1.94 (2H, m), 2.43 (1H, m), 2.93 (2H, m), 3.18 (1H, m), 4.12 (2H, s), 8.68 (1H, s). ¹³C NMR (100 MHz, DMSO-*d*₆): δ: 20.97, 25.76, 28.24, 30.06, 31.52, 33.07, 95.44, 126.47, 127.23, 136.03, 150.90, 161.21, 164.52, 169.35. Anal. Calc. for C₁₃H₁₄N₂O₂S₂: C, 53.04; H, 4.79; S, 21.78. Found: C, 53.03; H, 4.76; S, 21.75.

2.1.4.6. 2-(5,6,7,8-Tetrahydro[1]benzothieno[2,3-d]pyrimidin-4-ylthio)butanoic acid (**5f**). Yield 82%; m.p. 110–111 °C; ¹H NMR (400 MHz, DMSO-*d*₆): δ: 1.02 (3H, t, CH₃), 1.84 (4H, m, CH₂, –(CH₂)₄–), 1.89–2.04 (2H, m, CH₂), 2.84–3.01 (4H, m), 4.74 (1H, t), 8.69 (1H, s, 2-H), 12.93 (bs, OH). ¹³C NMR (100 MHz, DMSO-*d*₆): δ: 11.48, 22.00, 24.84, 25.22, 26.22, 47.25, 95.44, 126.85, 127.32, 136.51, 150.84, 161.19, 164.59, 171.93. Anal. Calc. for C₁₄H₁₆N₂O₂S₂: C, 54.52; H, 5.23; S, 20.79. Found: C, 54.50; H, 5.22; S, 20.78.

2.1.4.7. [(5,6-Dimethylthieno[2,3-d]pyrimidin-4-yl)thio]acetic acid (**5g**). Yield 84%; m.p. 209–210 °C; ¹H NMR (400 MHz, DMSO-*d*₆): δ: 2.47 (3H, s, CH₃), 2.53 (3H, s, CH₃), 4.12 (2H, s, CH₂), 8.68 (1H, s, 2-H), 12.79 (bs, OH). ¹³C NMR (100 MHz, DMSO-*d*₆): δ: 13.46, 14.20, 31.60, 95.44, 124.73, 128.29, 133.42, 150.70, 161.37, 163.71, 169.37. Anal. Calc. for C₁₀H₁₀N₂O₂S₂: C, 47.23; H, 3.96; S, 25.21. Found: C, 47.21; H, 3.95; S, 25.20.

2.1.4.8. 2-[(5-Phenylthieno[2,3-d]pyrimidin-4-yl)thio]propanoic acid (**5h**). Yield 80%; m.p. 111–112 °C; ¹H NMR (400 MHz, DMSO-*d*₆): δ: 1.40 (3H, d, CH₃, *J* = 7.3 Hz), 4.61 (1H, q, *J* = 7.3 Hz, *J* = 7.3 Hz), 7.42–7.49 (5H, m, C₆H₅), 7.78 (1H, s, 6-H), 8.84 (1H, s, 2-H). ¹³C NMR (100 MHz, DMSO-*d*₆): δ: 17.24, 41.18, 95.44, 125.16, 127.76, 128.36, 129.86, 134.81, 151.38, 163.19, 166.02, 172.54. Anal. Calc. for C₁₅H₁₂N₂O₂S₂: C, 56.94; H, 3.82; S, 20.27. Found: C, 56.93; H, 3.80; S, 20.25.

2.1.4.9. [(5-(4-Chlorophenyl)thieno[2,3-d]pyrimidin-4-yl)thio]acetic acid (**5i**). Yield 85%; m.p. 203–204 °C; ¹H NMR (400 MHz, DMSO-*d*₆): δ: 3.89 (2H, s, CH₂), 7.45 (2H, d, *J* = 8.5 Hz, 4-ClC₆H₄), 7.48 (2H, d, *J* = 8.5 Hz, 4-ClC₆H₄), 7.66 (1H, s, 6-H), 8.76 (1H, s, 2-H), 12.51 (bs, OH). ¹³C NMR (100 MHz, DMSO-*d*₆): δ: 31.78, 95.42, 125.70, 127.85, 131.70, 133.48, 151.57, 163.16, 165.79, 169.29. Anal. Calc. for C₁₄H₉ClN₂O₂S₂: C, 49.92; H, 2.69; S, 19.04. Found: C, 49.90; H, 2.67; S, 19.00.

2.1.4.10. 2-[(5-(4-Chlorophenyl)thieno[2,3-d]pyrimidin-4-yl)thio]butanoic acid (**5j**). Yield 82%; m.p. 144–145 °C; ¹H NMR (400 MHz, DMSO-*d*₆): δ: 0.94 (3H, t, CH₃), 1.74–1.90 (2H, m), 4.59 (1H, t), 7.44 (2H, d, *J* = 8.5 Hz, 4-ClC₆H₄), 7.49 (2H, d, *J* = 8.5 Hz, 4-ClC₆H₄), 7.70 (1H, s, 6-H), 8.77 (1H, s, 2-H). ¹³C NMR (100 MHz, DMSO-*d*₆): δ: 11.56, 24.62, 47.56, 95.47, 125.32, 125.75, 127.72, 131.57, 133.62, 151.39, 163.13, 166.07, 171.78. Anal. Calc. for C₁₆H₁₃ClN₂O₂S₂: C, 52.67; H, 3.59; S, 17.58. Found: C, 52.66; H, 3.57; S, 17.56.

2.1.4.11. [(6-Phenylthieno[2,3-d]pyrimidin-4-yl)thio]acetic acid (**5k**). Yield 86%; m.p. 171–172 °C; ¹H NMR (400 MHz, DMSO-*d*₆): δ: 4.15 (2H, s,

CH₂), 7.40–7.50 (3H, m), 7.79–7.84 (3H, m), 8.75 (1H, s, 2-H), 12.80 (bs, OH). ¹³C NMR (100 MHz, DMSO-*d*₆): δ: 31.38, 95.42, 114.05, 126.36, 128.59, 129.10, 132.13, 143.41, 151.99, 161.96, 164.38, 169.28. Anal. Calc. for C₁₄H₁₀N₂O₂S₂: C, 55.61; H, 3.33; S, 21.21. Found: C, 55.60; H, 3.31; S, 21.20.

2.1.4.12. 2-[[5-(4-Methylphenyl)thieno[2,3-*d*]pyrimidin-4-yl]thio]propanoic acid (**5l**). Yield 81%; m.p. 159–160 °C; ¹H NMR (400 MHz, DMSO-*d*₆): δ: 1.43 (3H, d, *J* = 7.3 Hz, CH₃), 2.42 (3H, s, CH₃), 4.60 (1H, q, *J* = 7.3 Hz, *J* = 7.3 Hz), 7.26 (2H, d, *J* = 8.1 Hz, 4-CH₃C₆H₄), 7.3 (2H, d, *J* = 8.1 Hz, 4-CH₃C₆H₄), 7.58 (1H, s, 6-H), 8.75 (1H, s, 2-H). ¹³C NMR (100 MHz, DMSO-*d*₆): δ: 17.24, 20.98, 41.10, 95.44, 124.95, 125.46, 128.36, 129.71, 131.89, 135.06, 137.64, 151.33, 163.14, 166.05, 172.57. Anal. Calc. for C₁₆H₁₄N₂O₂S₂: C, 58.16; H, 4.27; S, 19.41. Found: C, 58.15; H, 4.26; S, 19.40.

2.1.4.13. 2-[[5-(4-Methylphenyl)thieno[2,3-*d*]pyrimidin-4-yl]thio]butanoic acid (**5m**). Yield 84%; m.p. 133–134 °C; ¹H NMR (400 MHz, DMSO-*d*₆): δ: 0.94 (3H, t, CH₃), 1.71–1.89 (2H, m), 2.42 (3H, s, CH₃), 4.59 (1H, t), 7.25 (2H, d, *J* = 7.8, 4-CH₃C₆H₄), 7.3 (2H, d, *J* = 7.8, 4-CH₃C₆H₄), 7.58 (1H, s, 6-H), 8.75 (1H, s, 2-H). ¹³C NMR (100 MHz, DMSO-*d*₆): δ: 11.59, 20.95, 24.64, 47.42, 95.40, 125.08, 125.50, 128.33, 129.76, 131.89, 135.07, 137.68, 151.39, 163.21, 166.02, 172.00. Anal. Calc. for C₁₇H₁₆N₂O₂S₂: C, 59.28; H, 4.68; S, 18.62. Found: C, 59.27; H, 4.66; S, 18.60.

2.1.4.14. [[5-(4-Bromophenyl)thieno[2,3-*d*]pyrimidin-4-yl]thio]acetic acid (**5n**). Yield 80%; m.p. 171–172 °C; ¹H NMR (400 MHz, DMSO-*d*₆): δ: 3.98 (2H, s, CH₂), 7.44 (2H, d, *J* = 8.2, 4-BrC₆H₄), 7.69 (2H, d, *J* = 8.2, 4-BrC₆H₄), 7.84 (1H, s, 6-H), 8.84 (1H, s, 2-H), 12.67 (bs, OH). ¹³C NMR (100 MHz, DMSO-*d*₆): δ: 31.77, 95.43, 122.08, 125.67, 130.76, 131.98, 133.64, 134.04, 151.56, 163.15, 165.83, 169.26. Anal. Calc. for C₁₄H₉BrN₂O₂S₂: C, 44.10; H, 2.38; S, 16.82. Found: C, 44.07; H, 2.36; S, 16.81.

2.1.4.15. 4-(5,6,7,8-Tetrahydro[1]benzothieno[2,3-*d*]pyrimidin-4-ylthio)methylbenzoic acid (**5o**). Yield 83%; m.p. 257–258 °C; ¹H NMR (400 MHz, DMSO-*d*₆): δ: 1.80 (4H, m, -(CH₂)₄-), 2.82–2.98 (4H, m, -(CH₂)₄-), 4.67 (2H, s, CH₂), 7.58 (2H, d, *J* = 8.2), 7.88 (2H, d, *J* = 8.2), 8.76 (1H, s, 2-H), 12.92 (bs, OH). ¹³C NMR (100 MHz, DMSO-*d*₆): δ: 21.96, 25.22, 26.17, 32.22, 95.46, 126.87, 127.35, 128.96, 129.28, 129.55, 136.35, 142.28, 150.90, 161.36, 164.55, 166.71. Anal. Calc. for C₁₈H₁₆N₂O₂S₂: C, 60.65; H, 4.52; S, 17.99. Found: C, 60.64; H, 4.50; S, 17.97.

2.1.4.16. 3-[[5-(4-Chlorophenyl)thieno[2,3-*d*]pyrimidin-4-yl]thio]propanoic acid (**5p**). Yield 85%; m.p. 168–169 °C; ¹H NMR (400 MHz, DMSO-*d*₆): δ: 2.57 (2H, t, CH₂), 3.32 (2H, t, CH₂), 7.39 (2H, d, *J* = 7.8, 4-ClC₆H₄), 7.44 (2H, d, *J* = 7.8, 4-ClC₆H₄), 7.62 (1H, s, 6-H), 8.78 (1H, s, 2-H), 12.14 (bs, OH). ¹³C NMR (100 MHz, DMSO-*d*₆): δ: 24.43, 33.54, 95.49, 125.45, 127.67, 131.56, 133.37, 133.77, 151.60, 163.85, 165.94, 172.38. Anal. Calc. for C₁₅H₁₁ClN₂O₂S₂: C, 51.53; H, 3.16; S, 18.28. Found: C, 51.50; H, 3.15; S, 18.27.

2.1.4.17. 3-[[5-(3,4-Dimethylthieno[2,3-*d*]pyrimidin-4-yl)thio]methyl]benzoic acid (**5q**). Yield 80%; m.p. 116–117 °C; ¹H NMR (400 MHz, DMSO-*d*₆): δ: 2.47 (3H, s, CH₃), 2.52 (3H, s, CH₃), 4.64 (2H, s, CH₂), 7.39 (1H, t), 7.63 (1H, d), 7.82 (1H, d), 8.04 (1H, s), 8.65 (1H, s, 2-H), 12.60 (bs, OH). ¹³C NMR (100 MHz, DMSO-*d*₆): δ: 19.51, 14.37, 95.37, 124.90, 128.04, 128.56, 129.85, 130.95, 133.45, 137.77, 150.90, 161.82, 163.85, 166.90. Anal. Calc. for C₁₆H₁₄N₂O₂S₂: C, 58.16; H, 4.27; S, 19.41. Found: C, 58.15; H, 4.26; S, 19.40.

2.1.4.18. 3-[[2-Methyl-5,6,7,8-tetrahydro[1]benzothieno[2,3-*d*]pyrimidin-4-yl]thio]propanoic acid (**5r**). Yield 83%; m.p. 177–178 °C; ¹H NMR

(400 MHz, DMSO-*d*₆): δ: 1.79 (4H, m, -(CH₂)₄-), 2.50 (2H, t), 2.58 (3H, s), 2.77–2.93 (4H, m, -(CH₂)₄-), 3.39 (2H, t). Anal. Calc. for C₁₄H₁₆N₂O₂S₂: C, 54.52; H, 5.23; S, 20.79. Found: C, 54.51; H, 5.22; S, 20.76.

2.1.4.19. 4-(5,6,7,8-Tetrahydro[1]benzothieno[2,3-*d*]pyrimidin-4-ylthio)butanoic acid (**5s**). Yield 81%; m.p. 153–154 °C; ¹H NMR (400 MHz, DMSO-*d*₆): δ: 1.88 (4H, m, -(CH₂)₄-), 1.95 (2H, t, CH₂), 2.36 (2H, t, CH₂), 2.84–3.05 (4H, m, -(CH₂)₄-), 3.32 (2H, t, CH₂), 8.58 (1H, s, 2-H), 11.94 (bs, OH). ¹³C NMR (100 MHz, DMSO-*d*₆): δ: 22.01, 22.08, 24.17, 25.26, 26.27, 28.05, 32.51, 95.46, 127.07, 127.70, 136.07, 151.01, 162.26, 164.40, 173.46. Anal. Calc. for C₁₄H₁₆N₂O₂S₂: C, 54.52; H, 5.23; S, 20.79. Found: C, 54.51; H, 5.23; S, 20.76.

2.1.4.20. 3-[[5,6-Dimethylthieno[2,3-*d*]pyrimidin-4-yl]thio]propanoic acid (**5t**). Yield 87%; m.p. 189–190 °C; ¹H NMR (400 MHz, DMSO-*d*₆): δ: 2.47 (3H, s, CH₃), 2.53 (3H, s, CH₃), 2.69 (2H, t, CH₂), 3.46 (2H, CH₂), 8.61 (1H, s, 2-H), 12.19 (bs, OH). ¹³C NMR (100 MHz, DMSO-*d*₆): δ: 13.52, 14.37, 24.17, 33.55, 95.44, 124.92, 128.53, 133.18, 150.88, 162.21, 163.77, 172.54. Anal. Calc. for C₁₁H₁₂N₂O₂S₂: C, 49.53; H, 4.51; S, 23.9. Found: C, 49.52; H, 4.50; S, 23.7.

2.1.4.21. [(6-Methyl-5-phenylthieno[2,3-*d*]pyrimidin-4-yl)thio]acetic acid (**5u**). Yield 82%; m.p. 174–175 °C; ¹H NMR (400 MHz, DMSO-*d*₆): δ: 2.35 (3H, s, CH₃), 3.78 (2H, s, CH₂), 7.32 (2H, m, C₆H₅), 7.47 (3H, m, C₆H₅), 8.66 (1H, s, 2-H). ¹³C NMR (100 MHz, DMSO-*d*₆): δ: 14.09, 31.71, 95.40, 127.02, 128.10, 128.45, 130.58, 130.91, 133.60, 136.39, 150.98, 161.75, 163.43, 169.44. Anal. Calc. for C₁₅H₁₂N₂O₂S₂: C, 56.94; H, 3.82; S, 20.27. Found: C, 56.93; H, 3.80; S, 20.26.

2.1.5. Substituted (thieno[2,3-*d*]pyrimidin-4-ylthio)carboxylic acids **6a–g**

Compounds **6a–g** shown in Table 1 were synthesized on the base of substituted 4-chlorothieno[2,3-*d*]pyrimidines **4**. To a suspension of 0.01 mol of 3-mercapto propanoic acid or mercaptoacetic acid and 0.024 mol of potassium carbonate in 10 mL DMF the 4-chlorothieno[2,3-*d*]pyrimidine was added. The mixture was heated on a boiling water bath for 10 min. The resulting mixture was evaporated in vacuum. The residue was diluted with water and acetic acid slowly to pH = 7. The solid acid was filtrated and crystallized from isopropyl alcohol.

2.1.5.1. 3-[[5-(4-Methylphenyl)thieno[2,3-*d*]pyrimidin-4-yl]thio]propanoic acid (**6a**). Yield 85%; m.p. 161–162 °C; ¹H NMR (400 MHz, DMSO-*d*₆): δ: 2.42 (3H, s, CH₃), 2.56 (2H, t, CH₂), 3.29 (2H, t, CH₂), 7.21 (2H, d, *J* = 7.8, 4-CH₃C₆H₄), 7.26 (2H, d, *J* = 7.8, 4-CH₃C₆H₄), 7.50 (1H, s, 6-H), 8.76 (1H, s, 2-H), 12.12 (bs, OH). ¹³C NMR (100 MHz, DMSO-*d*₆): δ: 20.95, 24.42, 33.54, 95.39, 124.80, 125.79, 125.29, 129.76, 132.00, 135.19, 137.53, 151.62, 163.99, 165.85, 172.56. Anal. Calc. for C₁₆H₁₄N₂O₂S₂: C, 58.16; H, 4.27; S, 19.41. Found: C, 58.15; H, 4.26; S, 19.40.

2.1.5.2. 3-[[5-(3,4-Dimethylphenyl)thieno[2,3-*d*]pyrimidin-4-yl]thio]propanoic acid (**6b**). Yield 84%; m.p. 133–134 °C; ¹H NMR (400 MHz, DMSO-*d*₆): δ: 2.30 (3H, s, CH₃), 2.32 (3H, s, CH₃), 2.57 (2H, t, CH₂), 3.29 (2H, t, CH₂), 7.07–7.17 (3H, m, 3,4-(CH₃)₂C₆H₃), 7.46 (1H, s, 6-H), 8.75 (1H, s, 2-H), 12.10 (bs, OH). ¹³C NMR (100 MHz, DMSO-*d*₆): δ: 19.27, 19.37, 24.42, 33.53, 95.39, 124.60, 125.78, 127.27, 128.78, 130.93, 130.93, 132.29, 135.45, 136.21, 150.78, 151.59, 164.00, 165.86, 172.58. Anal. Calc. for C₁₇H₁₆N₂O₂S₂: C, 59.28; H, 4.68; S, 18.62. Found: C, 59.27; H, 4.67; S, 18.60.

2.1.5.3. [(6-Methylthieno[2,3-*d*]pyrimidin-4-yl)thio]acetic acid (**6c**). Yield 80%; m.p. 171–172 °C; ¹H NMR (400 MHz, DMSO-*d*₆): δ: 2.64 (3H, s, CH₃), 4.09 (2H, s, CH₂), 7.11 (1H, s, 5-H), 8.67 (1H, s, 2-H), 12.69 (bs, OH).

^{13}C NMR (100 MHz, DMSO- d_6): δ : 16.13, 31.22, 95.38, 116.18, 127.86, 141.87, 151.44, 160.44, 164.75, 169.34, 177.40. Anal. Calc. for $\text{C}_9\text{H}_8\text{N}_2\text{O}_2\text{S}_2$: C, 44.98; 3.36; S, 26.69. Found: C, 44.97; H, 3.34; S, 26.68.

2.1.5.4. 3-[[5-(4-Ethoxyphenyl)thieno[2,3-d]pyrimidin-4-yl]thio]propanoic acid (6d). Yield 86%; m.p. 138–139 °C; ^1H NMR (400 MHz, DMSO- d_6): δ : 1.42 (3H, t, CH_3), 2.56 (2H, t, CH_2), 3.29 (2H, t, CH_2), 4.08 (2H, q, CH_2), 6.91 (2H, d, $J = 8.5$, 4- $\text{CH}_3\text{CH}_2\text{OC}_6\text{H}_4$), 7.26 (2H, d, $J = 8.5$, 4- $\text{CH}_3\text{CH}_2\text{OC}_6\text{H}_4$), 7.84 (1H, s, 6-H), 8.75 (1H, s, 2-H). ^{13}C NMR (100 MHz, DMSO- d_6): δ : 14.69, 24.37, 33.53, 63.00, 95.41, 113.52, 124.50, 125.95, 126.82, 131.09, 134.99, 151.57, 158.57, 164.02, 165.77, 172.52. Anal. Calc. for $\text{C}_{17}\text{H}_{16}\text{N}_2\text{O}_2\text{S}_2$: C, 56.65; H, 4.47; S, 17.79. Found: C, 56.64; H, 4.45; S, 17.78.

2.1.5.5. 3-[[5-(4-Fluorophenyl)thieno[2,3-d]pyrimidin-4-yl]thio]propanoic acid (6e). Yield 83%; m.p. 166–167 °C; ^1H NMR (400 MHz, DMSO- d_6) δ 2.57 (2H, t, CH_2), 3.30 (2H, t, CH_2), 7.16–7.21 (2H, m, 4- FC_6H_4), 7.39–7.42 (2H, m, 4- FC_6H_4), 7.59 (1H, s, 6-H), 8.77 (1H, s, 2-H), 12.13 (bs, OH). ^{13}C NMR (100 MHz, DMSO- d_6): δ : 24.41, 33.49, 95.39, 114.52, 114.73, 125.34, 125.77, 131.20, 132.05, 133.97, 151.71, 160.99, 163.43, 163.91, 165.75, 172.53. Anal. Calc. for $\text{C}_{15}\text{H}_{11}\text{FN}_2\text{O}_2\text{S}_2$: C, 53.88; H, 3.32; S, 19.18. Found: C, 53.86; H, 3.31; S, 19.16.

2.1.5.6. 3-[(6-Methylthieno[2,3-d]pyrimidin-4-yl)thio]propanoic acid (6f). Yield 80%; m.p. 168–169 °C; ^1H NMR (400 MHz, DMSO- d_6): δ : 2.62 (3H, s, CH_3), 2.70 (2H, t, CH_2), 3.48 (2H, t, CH_2), 7.04 (1H, s, 5-H), 8.68 (1H, s, 2-H), 12.13 (OH, bs). ^{13}C NMR (100 MHz, DMSO- d_6): δ : 16.10, 23.91, 33.77, 95.37, 116.27, 128.14, 141.62, 151.58, 161.14, 164.68, 172.58. Anal. Calc. for $\text{C}_{10}\text{H}_{10}\text{N}_2\text{O}_2\text{S}_2$: C, 47.23; H, 3.96; S, 25.21. Found: C, 47.22; H, 3.94; S, 25.20.

2.1.5.7. 3-[(2-Methyl-6-phenylthieno[2,3-d]pyrimidin-4-yl)thio]propanoic acid (6g). Yield 82%; m.p. 215–216 °C; ^1H NMR (400 MHz, DMSO- d_6): δ : 2.68 (3H, s, CH_3), 2.72 (2H, t, CH_2), 3.50 (2H, t, CH_2), 7.36–7.46 (3H, m, C_6H_5), 7.59 (1H, s, 5-H), 7.74–7.76 (2H, m, C_6H_5), 12.22 (bs, OH). ^{13}C NMR (100 MHz, DMSO- d_6): δ : 23.87, 25.62, 33.72, 95.46, 113.80, 126.10, 126.33, 128.90, 128.98, 132.37, 141.69, 161.36, 162.19, 165.02, 172.55. Anal. Calc. for $\text{C}_{16}\text{H}_{14}\text{N}_2\text{O}_2\text{S}_2$: C, 56.94; H, 3.82; S, 20.27. Found: C, 56.92; H, 3.79; S, 20.19.

2.2. Biology

Selected compounds were tested using in vitro kinase assay. Each test was performed twice in a total reaction volume of 30 μL , containing 1 μg of peptide substrate RRRDDSDDDD (Custom synthesized by Genscript); 10 units of recombinant human CK2 (New England Biolabs); 50 μM ATP and γ -labeled ^{32}P ATP, diluted to specific activity 100 $\mu\text{Ci}/\mu\text{M}$; CK2 buffer (20 mM Tris–HCl, pH 7.5; 50 mM KCl; 10 mM MgCl_2) and inhibitor in varying concentrations. Incubation time was 20 min at 30 °C. The reaction was stopped by adding an equal volume of 10% o-phosphoric acid and the reaction mixture was loaded onto 20-mm discs of P-31 phosphocellulose paper (Whatman). Disks were washed three times with 1% o-phosphoric acid solution, air-dried at room temperature, and counted by Cherenkov method using a beta-counter (LKB). As negative control an equal volume of DMSO was added to the reaction mixture. Quercetin, a known CK2 inhibitor [23], was used as inhibition positive control. The final concentration of quercetin was 0.55 μM (which inhibited CK2 activity by 50%). Percent inhibition was calculated from substrate-incorporated radioactivity in the reaction relative to the radioactivity incorporated in control reactions, i.e. in the absence of inhibitor. Serial dilutions of inhibitor stock solution were used in reactions to determine their IC_{50} concentrations.

Protein kinases ASK1, JNK3, Aurora A, Rock 1, FGFR1, Met and Tie2 were assayed according to the supplier's suggestions (Millipore).

2.3. Computational methodology

For molecular docking of substituted (thieno[2,3-d]pyrimidin-4-ylthio)carboxylic acids DOCK 4.0 package was applied [24–27]. Molecular docking was performed as described previously [21].

3. Results and discussion

3.1. Biological effect and structure–activity relationships

To study CK2 inhibitory activity of substituted (thieno[2,3-d]pyrimidin-4-ylthio)carboxylic acids, 28 compounds of this class were synthesized and tested in vitro. The results showed that 21 compounds of the studied group inhibited the enzyme activity more than by 50% in the concentration ranging from 0.1 μM to 33 μM (IC_{50} values are shown in Table 1). Compounds displaying highest activity are **5e**, **5k**, **5p**, **6a**, **6b**, **6d**, **6e** and **6g**.

Kinetic studies of the most active compound **6a** (IC_{50} is 0.1 μM) showed that activity of (thieno[2,3-d]pyrimidin-4-ylthio)carboxylic acids is a result of their competition with ATP molecule for the binding site (Fig. 1). Inhibition constant for **6a** is 40 nM.

We studied structure–activity relationship of thieno[2,3-d]pyrimidine heterocycle considering substituents R^1 , R^2 and R^4 . It turned out that substituent R^4 affects compound's inhibitory activity the most. We observed that the heterocycles with $\text{R}^4 = \text{propionic acid}$ are more potent inhibitors than that with $\text{R}^4 = \text{acetic acid}$. To compare this effect one can refer to compound pairs such as **5t**, **5g**; **5p**, **5i** (corresponding IC_{50} values are 11.3 μM , 17.75 μM ; 0.175 μM , 32 μM). At the same time, $\text{R}^4 = \text{acetic acid}$ is more active than $\text{R}^4 = 2\text{-propionic acid}$ (see **5e**, **5b**; **5g**, **5c**, corresponding IC_{50} values are 1.78 μM , 20 μM ; 17.75 μM , >33 μM), while $\text{R}^4 = 2\text{-propionic acid}$ is more active than $\text{R}^4 = 2\text{-butyric acid}$ and $n\text{-butyric acid}$ (IC_{50} values of compounds **5a**, **5f**, **5s** are 17 μM , 33 μM and 33 μM , respectively). Substituents $\text{CH}_2\text{C}_6\text{H}_4\text{COOH-}m$ and CH_2COOH equally affect the inhibitory properties of studied heterocycles.

From the obtained results we concluded that the substituent R^4 of the heterocycle could be ranked by its effect as the following series: $\text{CH}_2\text{CH}_2\text{COOH} > \text{CH}_2\text{COOH} > \text{CH}(\text{CH}_3)\text{COOH} > \text{CH}_2\text{CH}_2\text{CH}_2\text{COOH} = \text{CH}(\text{C}_2\text{H}_5)\text{COOH}$. This points out that $\text{R}^4 = \text{propionic acid}$ causes the highest inhibitory activity of the studied compounds, and

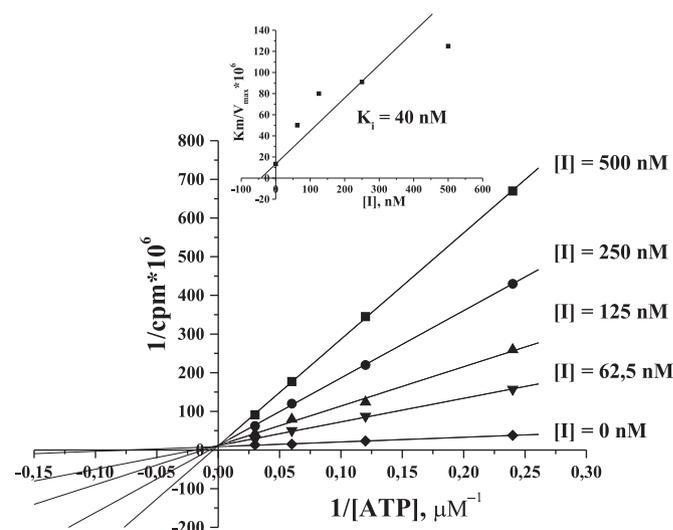


Fig. 1. Lineweaver–Burk plots of CK2 inhibition by compound **6a**. K_i value is 40 nM. Concentration of inhibitors varied from 0 nM to 500 nM.

Table 2
Specificity profile of CK2 inhibitors **6a** and **6d**^a.

Kinase	Kinase residual activity at presence of 6a , %	Kinase residual activity at presence of 6d , %
CK2	0.72	0.86
Jnk3	104	94
Rock1	126	115
Tie2	70	76
Ask1	92	100
Aurora A	23	48
Met	114	115
FGFR1	92	99

^a Residual activity, determined in the presence of 10 μ M of inhibitor is expressed as a percentage of the control without inhibitor. Final concentration of ATP in the experiment was 100 μ M.

it should be noticed that elongation and branching of the investigated substituents decreases their ability to inhibit CK2.

Substituents R¹ and R² of (thieno[2,3-d]pyrimidin-4-ylthio)acetic acid also significantly affect the inhibitory activity. Compounds containing R¹ = para-substituted phenyl (**5p**, **6a**, **6b**, **6d**, **6e**) display the lowest IC₅₀ values. It can be deduced that activity of these compounds is arranged as 4-FC₆H₄ < 3,4-CH₃C₆H₃ = 4-ClC₆H₄ < 4-C₂H₅OC₆H₄ < 4-CH₃C₆H₄ (IC₅₀ = 1.2 μ M, 0.175 μ M, 0.175 μ M, 0.125 μ M, 0.1 μ M, respectively).

3.2. Selectivity

Initial in vitro tests of discovered thienopyrimidine inhibitors **6a** and **6d** on four serine/threonine (ASK1, JNK3, Aurora A and Rock 1) and three tyrosine protein kinases (FGFR1, Met and Tie2) revealed remarkable specificity towards CK2 (Table 2). It can be noticed that the activity of Aurora A kinase is inhibited considerably by compound **6d**; nevertheless, we suppose that these thienopyrimidine derivatives are promising for extended selectivity evaluation.

3.3. Molecular modeling

In order to suppose the intermolecular interactions causing inhibitory activity of thienopyrimidines we inspected molecular docking complexes of the most active compounds displaying IC₅₀ less than 1 μ M (Fig. 2). It has been shown that these compounds

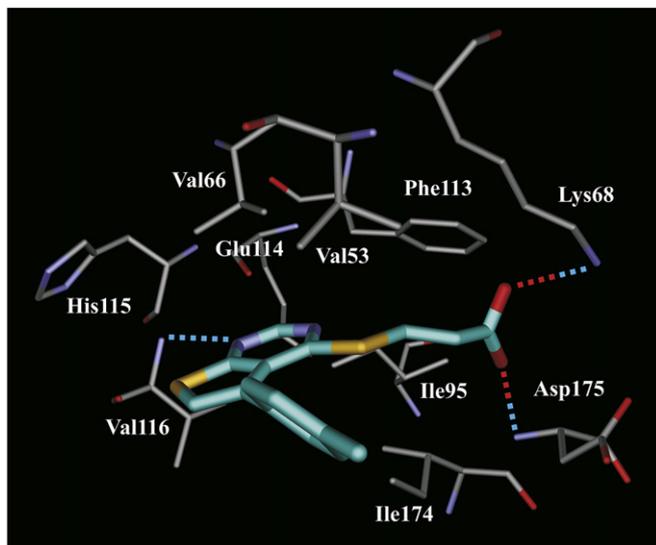


Fig. 2. **6a** bound to the active site of the CK2 catalytic subunit. The complex has been obtained with molecular docking. Intermolecular hydrogen bonds are shown as dotted lines.

have quite similar positions in the CK2 ATP-binding site, presumably because their R¹ and R⁴ substituents of the thieno[2,3-d]pyrimidine heterocycle are structurally alike. Key contacts that contribute significantly to the ligand binding are Van der Waals interactions of thieno[2,3-d]pyrimidine heterocycle and aromatic substituent R¹ with a number of hydrophobic residues (Leu45, Val53, Val66, Val116 and Ile174).

In addition, these compounds form three intermolecular hydrogen bonds in CK2 ATP-binding site. One of the bonds forms between nitrogen in 1-position of the thieno[2,3-d]pyrimidine heterocycle and Val116 of the CK2 hinge region. Two other bonds are formed by the substituent R⁴ (propionic acid) with side chain of Lys68 and main chain of Asp175 that are located in the hydrophobic region 1 of the active site.

Comparing SAR and modeling data it can be assumed that R⁴ = propionic acid provides an optimal length between carboxyl group and heterocycle of thieno[2,3-d]pyrimidine allowing inhibitor **6a** bind effectively in the CK2 active site due to formation of hydrogen bonds in two areas (hinge region and hydrophobic region 1) of catalytic site simultaneously. Replacing this substituent with a shorter R⁴ like acetic acid, branched 2-propionic and 2-butyric acids as well as with a longer 4-butyric acid resulted in decreasing of the compounds activity.

4. Conclusions

Substituted (thieno[2,3-d]pyrimidin-4-ylthio)carboxylic acids represent novel and selective class of CK2 inhibitors. The most active compounds obtained are **6a**, **6d** having IC₅₀ values of 0.1 μ M and 0.125 μ M, respectively. These inhibitors display remarkable selectivity towards CK2, and provide an opportunity to examine the therapeutic validity against CK2 and other kinases. The importance of carboxyl group in the structure of CK2 inhibitors and role of sterical effects in their binding interactions has been shown providing the basis for further structural optimization of this chemical class.

Author contribution

Andriy G. Golub: Molecular modeling, preparation of manuscript.

Volodymyr G. Bdzhol: Molecular modeling, SAR studies, preparation of manuscript.

Nadiia V. Briukhovetska: Organic synthesis, preparation of manuscript.

Anatoliiy O. Balanda: Organic synthesis.

Olexander P. Kukharenko: In vitro tests.

Igor M. Kotey: Organic synthesis.

Olga V. Ostrynska: In vitro tests.

Sergiy M. Yarmoluk: Supervision, research planning, preparation of manuscript.

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