

Two New Phenolic Compounds from *Artemisia iwayomogi*

by Xi-Tao Yan^{a)}, Yan Ding^{a)b)}, Wei Li^{a)}, Ya-Nan Sun^{a)}, Seo-Young Yang^{a)}, Young-Sang Koh^{c)},
and Young-Ho Kim^{*a)}

^{a)} College of Pharmacy, Chungnam National University, Daejeon 305-764, Korea
(phone: + 82-42-821-5933; fax: +82-42-823-6566; e-mail: yhk@cnu.ac.kr)

^{b)} School of Food Science and Technology, Dalian Polytechnic University, Dalian 116-034, P. R. China

^{c)} School of Medicine, Jeju National University, Jeju 690-756, Korea

Two new phenolic compounds, (*Z*)-5'-hydroxyjasnone 5'-*O*-[6''-*O*-[(*E*)-caffeoyl]- β -D-glucopyranoside] (**1**) and quercetin-7-*O*- β -D-glucuronide methyl ester (**2**), along with ten known phenolic compounds, **3**–**12**, were isolated from the aerial parts of *Artemisia iwayomogi*. Their structures were elucidated by spectroscopic methods, including 1D- and 2D-NMR, and HR-ESI-TOF-MS techniques. The inhibitory effects of compounds **1**–**12** on the LPS-stimulated production of IL-12 p40, IL-6, and TNF- α in bone marrow-derived dendritic cells were evaluated.

Introduction. – Since the discovery of artemisinin in the leaves of *Artemisia annua* in the early 1970s, plants belonging to the *Artemisia* genus have attracted considerable attention for their chemical constituents. *Artemisia iwayomogi* (Compositae), a member of the *Artemisia* genus, is a perennial herb widely distributed in Northeast Asian, especially Korea. The aerial parts of *A. iwayomogi* have long been used in traditional Korean medicine (called '*Han In Jin*') to cure various infectious diseases such as carbuncle, sores, cholecystitis, and hepatitis, and to treat fever, inflammation, and jaundice [1][2]. Previous phytochemical investigations on this plant led to the isolation of phenolic compounds, terpenes, and coumarins as major constituents [3–6].

In the course of our ongoing search for novel anti-inflammatory compounds from medicinal plants, twelve compounds were isolated from a MeOH extract of the aerial parts of *A. iwayomogi*, including two new phenolic compounds, (*Z*)-5'-hydroxyjasnone 5'-*O*-[6''-*O*-[(*E*)-caffeoyl]- β -D-glucopyranoside] (**1**) and quercetin-7-*O*- β -D-glucuronide methyl ester (**2**), and ten known phenolic compounds: patuletin 3-*O*- β -D-glucopyranoside (**3**) [7], quercetin (**4**) [8], citrusin C (**5**) [9], myrciaphenone A (**6**) [10], annphenone (**7**) [11], *erythro*-1,2-bis(4-hydroxy-3-methoxyphenyl)propane-1,3-diol (**8**) [12][13], *threo*-1,2-bis(4-hydroxy-3-methoxyphenyl)propane-1,3-diol (**9**) [13], protocatechuic aldehyde (**10**) [14], chlorogenic acid methyl ester (**11**) [15], and 3,4-di(*O*-caffeoyl)isoquinic acid (**12**) [16] (Fig. 1). Among them, compounds **3** and **8**–**12** were isolated from this plant for the first time. Herein, we describe the isolation and structure elucidation of the two new phenolic compounds, as well as the anti-inflammatory activities of the isolates in LPS-stimulated bone marrow-derived dendritic cells.

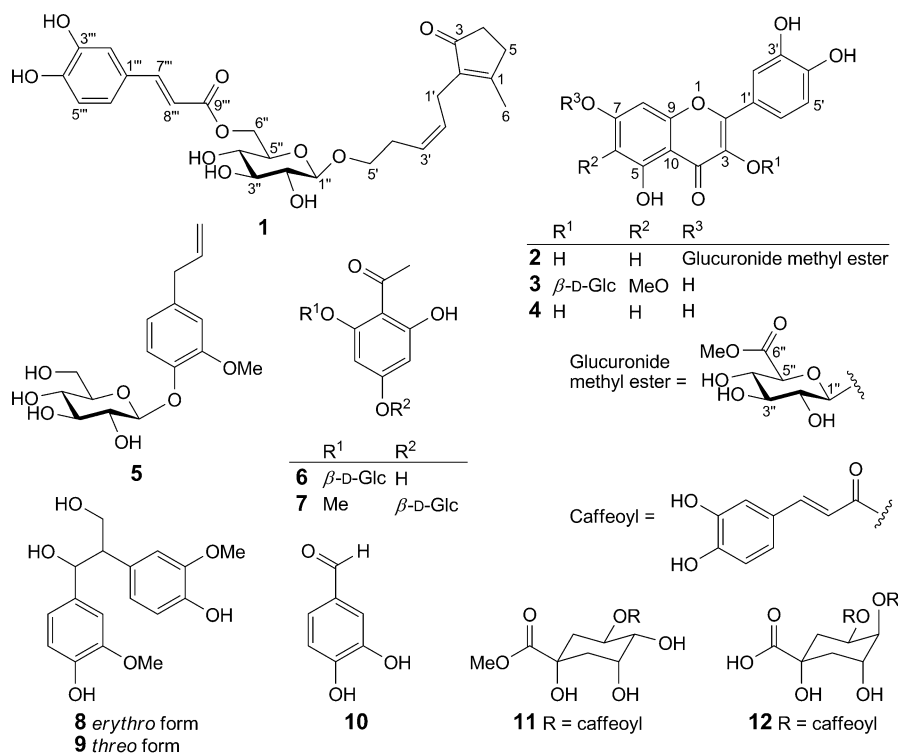
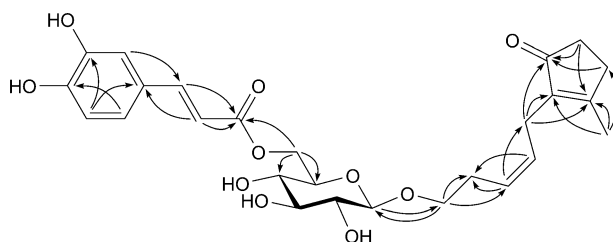


Fig. 1. The structures of compounds 1–12

Results and Discussion. – Compound **1** was obtained as a white amorphous powder. The molecular formula was determined as $C_{26}H_{32}O_{10}$ by high-resolution electrospray-ionization time-of-flight mass spectrometry (HR-ESI-TOF-MS) (m/z 503.1923 ($[M - H]^-$)). The IR spectrum showed strong absorption bands for OH (3365 cm^{-1}) and conjugated CO groups (1684 cm^{-1}), and conjugated olefinic bonds (1634 cm^{-1}). The ^1H -NMR spectrum of **1** (Table) exhibited downfield signals of three aromatic H-atoms at $\delta(\text{H})$ 6.99 ($d, J = 2.1, \text{H}-\text{C}(2'')$), 6.89 ($dd, J = 8.3, 2.1, \text{H}-\text{C}(6'')$), and 6.74 ($d, J = 8.3, \text{H}-\text{C}(5'')$), revealing one typical ABX coupling system, and of four olefinic H-atoms at $\delta(\text{H})$ 7.52 ($d, J = 15.8, \text{H}-\text{C}(7''')$), 6.24 ($d, J = 15.8, \text{H}-\text{C}(8''')$), 5.38 ($dt, J = 10.9, 7.6, 1.4, \text{H}-\text{C}(3')$), and 5.27 ($dt, J = 10.9, 6.9, 1.4, \text{H}-\text{C}(2')$), suggesting the presence of both (*E*)- and (*Z*)-form C=C bonds. Moreover, the ^1H -NMR spectrum of **1** showed signals corresponding to five O-bearing CH groups at $\delta(\text{H})$ 4.31 ($d, J = 8.3, \text{H}-\text{C}(1'')$) and 3.19–3.54 ($m, \text{H}-\text{C}(2'', 3'', 4'', 5'')$), two pairs of O-bearing CH_2 groups at $\delta(\text{H})$ 4.48 ($dd, J = 11.6, 2.1, \text{H}_a-\text{C}(6'')$) and 4.30 ($dd, J = 11.6, 6.2, \text{H}_b-\text{C}(6'')$), and 3.78 ($dt, J = 9.6, 7.6, \text{H}_a-\text{C}(5')$) and 3.61 ($dt, J = 9.6, 6.9, \text{H}_b-\text{C}(5')$), which were supported by DEPT-135 and HMQC data, four CH_2 groups in the range of $\delta(\text{H})$ 2.91–2.24 (m), and one Me group at $\delta(\text{H})$ 2.01 ($s, \text{Me}(6)$). Analyses of ^{13}C -NMR, DEPT, and HMQC data revealed that **1** contains 26 C-atoms comprising one Me, six CH_2 , and twelve CH groups, and seven quaternary C-atoms (Table). These spectral data implied the presence of one caffeoyl,

Table. ^1H - and ^{13}C -NMR Data (600 and 150 MHz, resp.) of **1** (CD_3OD) and **2** ($(\text{D}_6)\text{DMSO}$). δ in ppm, J in Hz. Atom numbering as indicated in Fig. 1.

Position 1		Position 2	
	$\delta(\text{H})$	$\delta(\text{C})$	
1		174.8	2
2		139.9	3
3		212.2	4
4	2.24–2.26 (<i>m</i>)	35.3	5
5	2.42–2.44 (<i>m</i>)	32.7	6
6	2.01 (<i>s</i>)	17.6	7
1'	2.88–2.91 (<i>m</i>)	22.3	8
2'	5.27 (<i>dt</i> , $J = 10.9, 6.9, 1.4$)	129.0	9
3'	5.38 (<i>dt</i> , $J = 10.9, 7.6, 1.4$)	127.5	10
4'	2.46–2.49 (<i>m</i>)	29.4	1'
5'	3.61 (<i>dt</i> , $J = 9.6, 6.9$), 3.78 (<i>dt</i> , $J = 9.6, 7.6$)	70.8	2'
1''	4.31 (<i>d</i> , $J = 8.3$)	104.8	3'
2''	3.19 (<i>dd</i> , $J = 8.3, 7.6$)	75.2	4'
3''	3.34–3.39 (<i>m</i>)	78.1	5'
4''	3.31–3.36 (<i>m</i>)	72.0	6'
5''	3.51–3.54 (<i>m</i>)	75.5	1''
6''	4.30 (<i>dd</i> , $J = 11.6, 6.2$), 4.48 (<i>dd</i> , $J = 11.6, 2.1$)	64.9	2''
1'''		127.8	3''
2'''	6.99 (<i>d</i> , $J = 2.1$)	115.2	4''
3'''		147.0	5''
4'''		149.8	6''
5'''	6.74 (<i>d</i> , $J = 8.3$)	116.6	MeO
6'''	6.89 (<i>dd</i> , $J = 8.3, 2.1$)	123.2	
7'''	7.52 (<i>d</i> , $J = 15.8$)	147.2	
8'''	6.24 (<i>d</i> , $J = 15.8$)	115.0	
9'''		169.2	

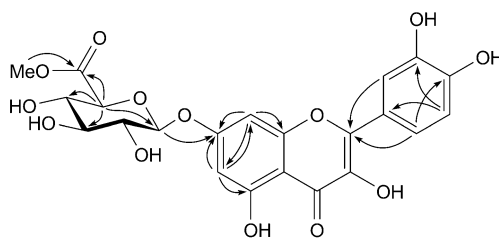
Fig. 2. Key HMBCs ($\text{H} \rightarrow \text{C}$) of **1**

one 5'-hydroxyjasmonate, and one glucosyl moiety in the molecule of **1**, which was confirmed by HMBCs and comparison of the NMR data with those reported in [17][18]. The $\text{C}=\text{C}$ bond in the caffeoyl moiety was determined to be (*E*)-configured based on its coupling constant ($J = 15.8$ Hz) between the $\text{H}-\text{C}(7''')$ and $\text{H}-\text{C}(8''')$, while the $\text{C}=\text{C}$ bond in the 5'-hydroxyjasmonate moiety was determined to be (*Z*)-configured due to the smaller coupling constant ($J = 10.9$ Hz) between the $\text{H}-\text{C}(2')$ and

H–C(3'). Correlations in the HMBC spectrum (Fig. 2) observed from the signal at $\delta(\text{H})$ 4.31 (H–C(1'')) to that at $\delta(\text{C})$ 70.8 (CH₂(5')), and from the signals at $\delta(\text{H})$ 3.61 and 3.78 (CH₂(5')) to that at $\delta(\text{C})$ 104.8 (CH(1')), indicated that the O–CH₂(5') group of the (*Z*)-5'-hydroxyjasnone moiety was connected to the anomeric C-atom (H–C(1'')) of the glucosyl moiety. Correlations from the signals at $\delta(\text{H})$ 4.30 and 4.48 (CH₂(6'')) to that at $\delta(\text{C})$ 169.2 (C(9'')) suggested that the caffeoyl moiety was linked to CH₂(6'') of the glucosyl moiety. The β -configuration at the anomeric center of the glucosyl moiety was supported by the relatively large *J* value (*J* = 8.3 Hz). The absolute D-configuration of the glucosyl moiety was determined by GC analysis. Consequently, the structure of **1** was elucidated as (*Z*)-5'-hydroxyjasnone 5'-O-[6''-O-[(*E*)-caffeoyl]- β -D-glucopyranoside].

Compound **2** was obtained as a yellow amorphous powder. Its molecular formula was determined as C₂₂H₂₀O₁₃ on the basis of HR-ESI-TOF-MS data (*m/z* 491.0828 ([*M* – H][–])). The IR spectrum showed absorption bands for OH groups (3308 cm^{–1}), an ester CO group (1740 cm^{–1}), an α,β -unsaturated CO group (1654 cm^{–1}), and a conjugated olefinic bond (1612 cm^{–1}). The ¹H-NMR spectrum of **2** (Table) exhibited signals for three *ABX* coupling aromatic H-atoms at $\delta(\text{H})$ 7.72 (*d*, *J* = 1.8, H–C(2'')), 7.56 (*dd*, *J* = 8.2, 1.8, H–C(6')), and 6.90 (*d*, *J* = 8.2, H–C(5')), two *meta*-coupling aromatic H-atoms at $\delta(\text{H})$ 6.81 (*d*, *J* = 1.8, H–C(8)) and 6.44 (*d*, *J* = 1.8, H–C(6)), for five H-atoms in the range of $\delta(\text{H})$ 3.28–5.33, and one MeO group at $\delta(\text{H})$ 3.67 (*s*). The ¹³C-NMR spectrum of **2**, combined with the HMQC spectrum, showed 22 C-atom signals, including those of one Me group and ten CH groups, and eleven quaternary C-atoms. The quaternary C-atom signals at $\delta(\text{C})$ 176.0, 147.7, 136.1, and 104.8 were typical of C(4), C(2), C(3), and C(10) of a 3-*O*-substituted flavonoid moiety. Additional signals of five *O*-bearing aromatic C-atoms at $\delta(\text{C})$ 162.2 (C(7)), 160.5 (C(5)), 155.8 (C(9)), 148.0 (C(4')), and 145.1 (C(3')) suggested that this 3-*O*-substituted flavonoid moiety corresponded to a quercetin (2-(3',4'-dihydroxyphenyl)-3,5,7-trihydroxy-4*H*-chromen-4-one) group [8]. Based on comparison of the NMR spectral data with those reported in literature [19][20], the signals of seven *O*-bearing C-atoms at $\delta(\text{C})$ 99.0 (C(1'')), 72.8 (C(2'')), 75.4 (C(3'')), 71.3 (C(4'')), 75.2 (C(5'')), 169.3 (C(6'')), and 52.0 (Me(7'')) were consistent with the β -D-glucuronide methyl ester group. The larger coupling constant of the anomeric H-atom signal at $\delta(\text{H})$ 5.33 (*d*, *J* = 7.3) indicated the β -configuration of the glucuronide methyl ester group. This group was also evidenced by the HMBCs (Fig. 3), especially the key HMBCs H–C(5'')/C(6'') and Me(7'')/C(6''). The HMBC from the signal at $\delta(\text{H})$ 5.33 (*d*, *J* = 7.3, H–C(1'')) to that at $\delta(\text{C})$ 162.2 (C(7)) indicated that the β -D-glucuronide methyl ester group was linked to the quercetin group through C(1'')–O–C(7). Thus, the structure of **2** was determined as quercetin-7-*O*- β -D-glucuronide methyl ester.

Inhibitory effects of the isolated compounds, **1**–**12**, on proinflammatory cytokines were evaluated by evaluating the production of IL-12 p40, IL-6, and TNF- α in LPS-stimulated bone marrow-derived dendritic cells. SB203580, an inhibitor of cytokine suppressive binding protein/p38 kinase, was used as a positive control, which inhibited IL-12 p40, IL-6, and TNF- α production with *IC*₅₀ values of 5.0 \pm 0.2, 3.5 \pm 0.1, and 7.2 \pm 0.3 μM , respectively. Among the compounds, a mixture of compounds **8** and **9** (a pair of diastereoisomers) and compound **4** exhibited potent inhibitory activities against IL-12 p40 production with *IC*₅₀ values of 0.03 \pm 0.001 and 5.1 \pm 0.2 μM , respectively.

Fig. 3. Key HMBCs (H \rightarrow C) of **2**

Moreover, compound **7** showed moderate inhibitory activity against IL-12 p40 with an IC_{50} value of $21.6 \pm 1.5 \mu\text{M}$. However, only compound **4** inhibited IL-6 and TNF- α production with IC_{50} values of $52.93 \pm 2.2 \mu\text{M}$ and $50.11 \pm 2.1 \mu\text{M}$, respectively. The remaining compounds did not show any significant activity against the production of IL-12 p40, IL-6, and TNF- α in LPS-stimulated bone marrow-derived dendritic cells ($IC_{50} > 60 \mu\text{M}$).

This work was financially supported by the *Technology Development Program for Agriculture and Forestry* (No. 108079-3), the *Ministry for Agriculture, Forestry and Fisheries*, Korea, and the *Priority Research Centers Program* through the *National Research Foundation of Korea* (NRF) funded by the *Ministry of Education, Science and Technology* (2009-0093815), Korea.

Experimental Part

General. Column chromatography (CC): silica gel (SiO_2 ; 40–75 and 75–200 μm particle size; *Fuji Silysia Chemical Ltd.*, Japan) and YMC RP-18 resins (30–50 μm particle size; *Fuji Silysia Chemical Ltd.*, Japan). TLC: Silica gel 60 F_{254} and RP-18 F_{254S} plates (*Merck*, DE-Darmstadt). GC: *Shimadzu GC-2010* spectrometer. Optical rotations: *Jasco P-2000* digital polarimeter. UV Spectra: *Shimadzu UV-1800* spectrophotometer; λ_{max} (log ϵ) in nm. IR Spectra: *Nicolet 380 FT-IR* spectrometer; KBr pellets; $\tilde{\nu}$ in cm^{-1} . NMR Spectra: *Jeol ECA 600* spectrometer (^1H - and ^{13}C -NMR at 600 and 125 MHz, resp.); δ in ppm rel. to Me_4Si as an internal standard, J in Hz. ESI-MS: *Agilent 1100 LC-MSD trap* spectrometer. HR-ESI-TOF-MS: *Agilent 6530 Accurate-Mass Q-TOF* LC/MS system; in m/z .

Plant Material. The aerial parts of *Artemisia iwayomogi* were collected on Jeju Island in June 2007, and taxonomically identified by Y.-H. K. at the College of Pharmacy, Chungnam National University, Daejeon, Korea. A voucher specimen (CNU07105) has been deposited with the herbarium of the above college.

Extraction and Isolation. The dried aerial parts of *Artemisia iwayomogi* (3 kg) were extracted with MeOH under reflux ($3 \times 18\text{ l}$, 12 h each). The combined extract was concentrated under reduced pressure to yield a residue (465 g), which was suspended in H_2O (3 l), and partitioned successively with CHCl_3 ($4 \times 3\text{ l}$) and AcOEt ($4 \times 3\text{ l}$) to yield a CHCl_3 -soluble fraction (94 g), an AcOEt-soluble fraction (42 g), and a H_2O fraction, resp. The AcOEt-soluble fraction was subjected to column chromatography (CC; SiO_2 ; $\text{CHCl}_3/\text{MeOH}$ 50:1 to 10:1) to afford six subfractions, *Fr.* 1–6. *Fr.* 2 (31 g) was further subjected to repeated CC (SiO_2 ; hexane/acetone 5:1 to 0:1), and then to CC (YMC C-18; $\text{H}_2\text{O}/\text{MeOH}$ 2:1 to 1:1) to yield compounds **1** (26 mg), **4** (45 mg), **5** (13 mg), **6** (10 mg), **7** (30 mg), a mixture **8/9** (8 mg), **10** (24 mg), and **12** (420 mg). The H_2O fraction was submitted to CC (*Diaion HP-20*; $\text{H}_2\text{O}/\text{MeOH}$ 100:0, 75:25, 50:50, 25:75, 0:100) to yield five fractions, *Fr.* A–E. *Fr.* C (eluted with $\text{H}_2\text{O}/\text{MeOH}$ 1:1, 56 g) was subjected to CC (SiO_2 ; $\text{CH}_2\text{Cl}_2/\text{MeOH}/\text{H}_2\text{O}$ 12:1:0.05 to 1:1:0.1) to afford four subfractions, *Fr.* C1–C4. *Fr.* C2 was purified by repeated CC (YMC C-18; $\text{H}_2\text{O}/\text{MeOH}$ 2.5:1 to 1:1) to

yield compounds **2** (22 mg) and **11** (26 mg). Fr. C3 was further purified by CC (YMC C-18; H₂O/MeOH 1.5 : 1 to 1 : 1) to yield compound **3** (10 mg).

(Z)-5'-Hydroxyjasnone 5'-O-{6'-O-[(E)-caffeoyl]-β-D-glucopyranoside} (= 2-[(2Z)-5-[[6-O-[(2E)-3-(3,4-Dihydroxyphenyl)-1-oxoprop-2-en-1-yl]-β-D-glucopyranosyl]oxy]pent-2-en-1-yl]-3-methyl-2-cyclopenten-1-one; **1**). White powder. $[\alpha]_D^{25} = -69.9$ ($c = 1.0$, MeOH). UV (MeOH): 234 (2.76), 330 (1.72). IR (KBr): 3365, 1684, 1634, 1605, 1525, 1440, 1386, 1283, 1164, 1039, 851, 803. ¹H- and ¹³C-NMR: see the Table. HR-ESI-MS (neg.): 503.1923 ($[M - H]^-$, C₂₆H₃₁O₁₀; calc. 503.1917).

Quercetin-7-O-β-D-glucuronide methyl ester (= 2-(3,4-Dihydroxyphenyl)-3,5-dihydroxy-7-[(6-methyl-β-D-glucopyranuronosyl)oxy]-4H-1-benzopyran-4-one; **2**): Yellow powder. $[\alpha]_D^{25} = -189.1$ ($c = 1.0$, MeOH). UV (MeOH): 256 (1.86), 372 (1.60). IR (KBr): 3308, 1740, 1654, 1612, 1597, 1497, 1317, 1247, 1212, 1173, 1088, 1044, 1024, 1000. ¹H- and ¹³C-NMR: see the Table. HR-ESI-MS (neg.): 491.0828 ($[M - H]^-$, C₂₂H₁₉O₁₃; calc. 491.0826).

Acid Hydrolysis. Compound **1** (2.0 mg) was dissolved in 1N HCl (dioxane/H₂O 1 : 1, v/v, 1 ml) and then heated to 80° in a water bath for 3 h. The cooled mixture was diluted with H₂O (4 ml) and extracted with AcOEt (3 × 5 ml). The aq. layer was thoroughly dried under N₂ after neutralization with Ag₂CO₃. The residues were dissolved in 0.1 ml of dry pyridine, and then L-cysteine methyl ester hydrochloride in pyridine (0.06M, 0.1 ml) was added to the solns. The mixture was heated at 60° for 2 h, and 0.1 ml of TMSCl (Me₃SiCl) soln. was added, followed by heating at 60° for 1.5 h. The dried product were partitioned with hexane and H₂O (0.1 ml each), and the org. layer was analyzed by GC (column: SPB-1, 0.25 mm × 30 m; detector, FID; detector temp., 300°; column temp., 210°; injector temp., 270°; carrier gas, He, 2 ml/min). The monosaccharide was confirmed as D-glucose by comparison of the retention time of the monosaccharide derivative (t_R 14.09 min) with that of authentic sugar derivatives (D-glucose derivative: t_R 14.11 min and L-glucose derivative: t_R 14.26 min), which were prepared by the same reaction from the standard glucoses.

Biological Assay. Bone marrow-derived dendritic cells (BMDCs) were grown from wild-type C57BL/6 mice (Taconic Farm, NY, USA). Briefly, the mouse tibia and femur were obtained by flushing with Dulbecco's modified Eagle's medium to yield bone marrow cells. The cells were cultured in RPMI 1640 medium containing 10% heat-inactivated fetal bovine serum (FBS) (Gibco, NY, USA), 50 μM 2-sulfanylethanol, and 2 mM glutamine supplemented with a 3% J558L hybridoma cell culture supernatant containing granulocyte-macrophage colony-stimulating factor. The culture medium was replaced with fresh medium every second day. At day 6 of culture, nonadherent cells and loosely adherent DC aggregates were harvested, washed, and resuspended in RPMI 1640 supplemented with 5% FBS. The BMDCs were incubated in 48-well plates at a density of 2×10^5 cells/ml, and then treated with the test compounds in DMSO (2, 10, 25, and 50 μM) for 1 h before stimulation with 10 ng/ml LPS from *Salmonella minnesota* (Alexis, NY, USA). Supernatants were harvested 16 h after stimulation. Concentrations of murine IL-12 p40, IL-6, and TNF-α in the culture supernatant fraction were determined by enzyme-linked immune-sorbent assay (BD Pharmingen, CA, USA) according to the manufacturer's instructions. The data are presented as mean ± SD of at least three independent experiments performed in triplicate.

REFERENCES

- [1] J. K. Kim, 'Illustrated Natural Drugs Encyclopedia', Namsandang Publishers, Seoul, 1989, p. 79.
- [2] J. H. Park, 'Korean Folk Medicine', Busan National University Publishers, Busan, 1999, p. 68.
- [3] Y. Ding, C. Liang, S. Y. Yang, J. C. Ra, E. M. Choi, J.-A. Kim, Y. H. Kim, *Biol. Pharm. Bull.* **2010**, *33*, 1448.
- [4] A. R. Kim, Y. N. Zou, T. H. Park, K. H. Shim, M. S. Kim, N. D. Kim, J. D. Kim, S. J. Bae, J. S. Choi, H. Y. Chung, *Phytother. Res.* **2004**, *18*, 1.
- [5] H.-H. Yu, Y.-H. Kim, B.-S. Kil, K.-J. Kim, S.-I. Jeong, Y.-O. You, *Planta Med.* **2003**, *69*, 1159.
- [6] H. Greger, C. Zdero, F. Bohlmann, *Phytochemistry* **1986**, *25*, 891.
- [7] W. Bylka, *Acta Physiol. Plant.* **2004**, *26*, 393.
- [8] H. Wagner, V. M. Chari, J. Sonnenbichler, *Tetrahedron Lett.* **1976**, *17*, 1799.

- [9] M.-R. Kim, S.-K. Lee, C.-S. Kim, K.-S. Kim, D.-C. Moon, *Arch. Pharmacol. Res.* **2004**, *27*, 1029.
- [10] M. Yoshikawa, H. Shimada, N. Nishida, Y. Li, I. Toguchida, J. Yamahara, H. Matsuda, *Chem. Pharm. Bull.* **1998**, *46*, 113.
- [11] A. K. Singh, V. Pathak, P. K. Agrawal, *Phytochemistry* **1997**, *44*, 555.
- [12] K. Lundquist, *Acta Chem. Scand., B* **1979**, *33*, 418.
- [13] S. G. Guan, W. B. Yu, S. H. Guan, *Shizhen Guoyi Guoyao* **2010**, *21*, 905.
- [14] E. T. Kolehmainen, K. P. Laihia, J. M. I. Hyötyläinen, R. T. Kauppinen, *Spectrochim. Acta, Part A* **1995**, *51*, 419.
- [15] X. Zhu, X. Dong, Y. Wang, P. Ju, S. Luo, *Helv. Chim. Acta* **2005**, *88*, 339.
- [16] M. Ono, C. Masuoka, Y. Odake, S. Ikegashira, Y. Ito, T. Nohara, *Food Sci. Technol. Res.* **2000**, *6*, 106.
- [17] J. Kitajima, T. Ishikawa, A. Urabe, *Chem. Pharm. Bull.* **2004**, *52*, 1013.
- [18] P. Zhao, T. Tanaka, K. Hirabayashi, Y.-J. Zhang, C.-R. Yang, I. Kouno, *Phytochemistry* **2008**, *69*, 3087.
- [19] E.-R. Woo, M. S. Piao, *Arch. Pharmacol. Res.* **2004**, *27*, 173.
- [20] D. P. Allais, A. Simon, B. Bennini, A. J. Chulia, M. Kaouadji, C. Delage, *Phytochemistry* **1991**, *30*, 3099.

Received April 23, 2013