The Journal of Organic Chemistry

Article

Subscriber access provided by BIU Pharmacie | Faculté de Pharmacie, Université Paris V

Cu(I)-Catalyzed Ligand-free Tandem One Pot or Sequential Annulation via Knoevenagel Intermediate: An Entry into Multifunctional Naphthalenes, Phenanthrenes, Quinolines and Benzo[b]carbazoles

K C Kumara Swamy, Rajnikanth Sunke, and ADULA KALYANI

J. Org. Chem., Just Accepted Manuscript • DOI: 10.1021/acs.joc.9b02991 • Publication Date (Web): 11 Dec 2019 Downloaded from pubs.acs.org on December 12, 2019

Just Accepted

"Just Accepted" manuscripts have been peer-reviewed and accepted for publication. They are posted online prior to technical editing, formatting for publication and author proofing. The American Chemical Society provides "Just Accepted" as a service to the research community to expedite the dissemination of scientific material as soon as possible after acceptance. "Just Accepted" manuscripts appear in full in PDF format accompanied by an HTML abstract. "Just Accepted" manuscripts have been fully peer reviewed, but should not be considered the official version of record. They are citable by the Digital Object Identifier (DOI®). "Just Accepted" is an optional service offered to authors. Therefore, the "Just Accepted" Web site may not include all articles that will be published in the journal. After a manuscript is technically edited and formatted, it will be removed from the "Just Accepted" Web site and published as an ASAP article. Note that technical editing may introduce minor changes to the manuscript text and/or graphics which could affect content, and all legal disclaimers and ethical guidelines that apply to the journal pertain. ACS cannot be held responsible for errors or consequences arising from the use of information contained in these "Just Accepted" manuscripts.

is published by the American Chemical Society. 1155 Sixteenth Street N.W., Washington, DC 20036

Published by American Chemical Society. Copyright © American Chemical Society. However, no copyright claim is made to original U.S. Government works, or works produced by employees of any Commonwealth realm Crown government in the course of their duties.

Cu(I)-Catalyzed Ligand-free Tandem One Pot or Sequential Annulation via Knoevenagel Intermediate: An Entry into Multifunctional Naphthalenes,

Phenanthrenes, Quinolines and Benzo[b]carbazoles

Rajnikanth Sunke, Adula Kalyani and K. C. Kumara Swamy*

School of Chemistry, University of Hyderabad, Hyderabad 500 046, Telangana, India



ABSTRACT

A simple but efficient one-pot or sequential copper-catalyzed protocol using 2-bromoaldehydes and active methylene group containing substrates that affords multifunctional naphthalenes, phenanthrenes, quinolines and benzo[b]carbazoles via Knoevenagel condensation, C-arylation and decarboxylation followed by aromatization, has been developed. The reaction utilizes the potential of Knoevenagel intermediates and does not require ancillary ligand. The phenanthrene products thus obtained show moderate fluorescence activity. Structural elaboration of the products obtain to dihydrobenzoquinazolines is also highlighted.

INTRODUCTION

The development of new and efficient methodologies for the construction of highly substituted polycyclic arenes and N-heteroarenes with flexible substituent patterns is a worthwhile exercise as it allows access to the diversity based chemical space useful for pharmaceuticals and electronic materials.¹ Among polycyclic arenes, multifunctional naphthalenes and phenanthrenes are important classes of organic compounds because they have not only found widespread applications in pharmaceutical chemistry (cf. Figure $1)^2$ but also play a significant role in supramolecular chemistry, nanotechnology, electronic materials and the design of chiral catalysts.³ Therefore, the development of efficient strategies to access such structural motifs from readily accessible starting materials has immense impact and value. Consequently, a number of synthetic strategies toward polysubstituted naphthalenes or phenanthrenes have been developed. The most important avenues include Diels-Alder reaction,⁴ annulation reaction,⁵ metathesis,⁶ rearrangements of strained rings,⁷ catalytic cyclization,⁸ C-H activation,⁹ cycloisomerization¹⁰ and cross-coupling protocol.¹¹ In addition, N-heteroarenes such as quinolines and benzocarbazoles also feature prominently among many natural products and drug candidates.¹² As a result, a number of synthetic platforms are used for the construction of such structural systems.¹³ While significant progress has been achieved for the construction of naphthalenes, phenanthrenes, quinolines and benzo[b]carbazoleswith diverse substitutions, the number of approaches based on Cu(I)-catalyzed reactions is rather limited.



Figure 1. Selected examples of natural products and biologically active molecules containing substituted naphthalenes, phenanthrenes, quinolines and benzo[*b*]carbazoles.

Although Cu(I)-catalyzed cross-coupling reactions are well-studied in domino and multi-component reactions,¹⁴ their application for the construction of polysubstituted naphthalene derivatives is very limited. Panda *et al.* developed an efficient method for the synthesis of polysubstituted α -naphthols from simple monocyclic enol esters (Scheme 1a).^{15a} Jiang and co-workers disclosed a copper-catalyzed tandem reaction of methyl-3-(2-bromophenyl)-3-oxopropanoate with methyl-3-oxobutanoate for the synthesis of dimethyl-4-hydroxy-2-methylnaphthalene-1,3-dicarboxylate (Scheme 1b),^{15b} Liu and co-workers described a copper-catalyzed borylative cyclization strategy to construct 3-boryl-1-naphthylamines from *o*-(cyano)phenyl propargyl carbonates (Scheme 1c).^{15c} In the year 2011, Beifuss and coworkers developed an efficient copper-catalyzed domino method for the synthesis of dialkyl-2-alkylnaphthalene-1,3-dicarboxylates via the reaction of *o*-halobenzyl halide with two

molecules of a β -ketoester (Scheme 1d).^{15d} Later, the same group extended this protocol to *o*-halobenzaldehydes as substrates to access same substituted naphthalenes (Scheme 1e).^{15e} Inspite of these outstanding efforts, it is still challenging to prepare polysubstituted aromatic compounds (PACs) possessing diverse substituents directly from readily available building blocks in a single synthetic operation. In this context, we report herein a convenient and efficient copper-catalyzed tandem or sequential synthesis of multifunctional naphthalenes, phenanthrenes, quinolines and benzo[*b*]carbazoles by using commercially available starting materials such as *o*-bromoarylaldehydes, alkyl cyanoacetates, dialkyl malonates and malononitrile (Scheme 1f). Our reaction works under ligand-free conditions and further elaboration of the products can be readily achieved, as shown by the isolation of dihydrobenzoquinazolines. Also, the phenanthrene products are fluorescence active, a property that may be useful in applications.

Scheme 1. Selected Related Reports on [Cu]-Mediated Approaches to Polysubstituted Naphthalenes and Our Present Work



RESULTS AND DISCUSSION

Initially, 2-bromobenzaldehyde 1a was treated with ethyl 2-cyanoacetate 2a in the presence of CuBr (10 mol%) and K₂CO₃ (3.0 equiv) in DMSO at 120 °C for 3 h. TLC analysis showed that 1a was consumed completely to give ethyl 3-amino-4-cyano-2naphthoate **3aa** in 74% yield (Table1, entry1). To identify the best reaction condition, we have varied the catalysts, bases and solvents as shown in Table 1. First, copper catalysts CuBr, CuCl, CuI, CuCl₂, Cu(OAc)₂ and Cu(OTf)₂ (0.1equiv) were checked by using 3.0 equiv of K_2CO_3 as the base and DMSO as the solvent at 120 °C (Table 1, entries 1-6). CuI provided the highest yield and the target product **3aa** was isolated in 81% yield (entry 4). In the absence of copper catalyst, product **3aa** was not at all observed (entry 7). Lowering of catalyst loading to 5 mol % decreased the yield (entry 8) whereas increasing the catalyst loading to (20 mol %) did not improve yield of the reaction (entry 8). Among the bases K₂CO₃, Cs₂CO₃, Na₂CO₃, K₃PO₄, t-BuOK, Et₃N and DBU (entries 4, 9-14) that were checked, K_2CO_3 provided the highest yield. Among the solvents DMSO, N,Ndimethylacetamide (DMA), 1,4-dioxane, toluene, and PEG-400 (entries 4, 16-19) that were tested, DMSO was found to be the solvent of choice, although DMF (entry 15) also gave a comparable yield. Conducting the reaction at 70 °C/10 h or 150 °C/ 3 h afforded **3aa** in nearly the same yield (entry 20). It was also noteworthy that this reaction protocol required neither an inert atmosphere nor any additional ligand to produce alkyl 3-amino-4-cyano-2-naphthoate **3aa** in good yield (entry 22). Eventually, reaction conditions in entry 4 was determined as optimal [1a (1.0 mmol), 2b (2.0 mmol), CuI (10 mol %), and K_2CO_3 (3.0 mmol) in DMSO at 120 °C/3 h].

Table 1. Optimization of the Reaction Conditions^a

	H + 2	CN EtO EtO CU]-catalyst, Base Solvent 120 °C		H ₂ ,OEt
	1a	2a	3aa	
Entry	catalyst	base	solvent	yield of 3aa $(\%)^b$
1	CuBr	K ₂ CO ₃	DMSO	74
2	CuCl	K ₂ CO ₃	DMSO	52
3	CuCl ₂ (anh.)	K ₂ CO ₃	DMSO	47
4	CuI	K ₂ CO ₃	DMSO	81 ^c
5	Cu(OAc) ₂ (anh.)	K ₂ CO ₃	DMSO	54
6	Cu(OTf) ₂	K ₂ CO ₃	DMSO	52
7	-	K ₂ CO ₃	DMSO	_d
8	CuI	K ₂ CO ₃	DMSO	$54(80)^{e}$
9	CuI	Cs ₂ CO ₃	DMSO	76
10	CuI	Na ₂ CO ₃	DMSO	71
11	CuI	K ₃ PO ₄	DMSO	60
12	CuI	t-BuOK	DMSO	42
13	CuI	Et ₃ N	DMSO	14
14	CuI	DBU	DMSO	25
15	CuI	K ₂ CO ₃	DMF	75
16	CuI	K ₂ CO ₃	DMA	37
17	CuI	K ₂ CO ₃	Dioxane	10
18	CuI	K ₂ CO ₃	Toluene	11

19	CuI	K ₂ CO ₃	PEG-400	7
20	CuI	K_2CO_3	DMSO	77 (80) ^f
21	CuI	-	DMSO	_g
22	CuI	K ₂ CO ₃	DMSO	$80(77)^{h}$

^{*a*}Reaction conditions: **1a** (0.54 mmol), **2a** (1.09 mmol), base (1.62 mmol), catalyst (0.054 mmol), solvent (2 mL), 120 °C (oil bath) for 3 h in a stoppered Schlenk tube in air. ^{*b*}Isolated yield of **3aa**. ^{*c*}Reaction performed under nitrogen gave the same yield but under oxygen (balloon), the yield was only 45%. ^{*d*}No catalyst was added. ^{*e*}5 mol % and 20 mol % of CuI were used. ^{*f*}The yields are for the reaction performed at 70 °C/10 h (77%) and 150 °C/ 3 h (80%). ^{*g*}No base was added; only Knoevenagel product (38%) was isolated. ^{*h*}10 mol % of L-proline (or 1,10-phenanthroline) was used.

As shown in Table 2, 2-bromobenzaldehydes **1a-1f** possessing electron-neutral, electron rich or electron-withdrawing substituents smoothly converted to the corresponding products **3** in good yields. The nitriles **2a**, **2c** and **2d** containing CO_2R^4 [R^4 = Me, Et, *n*-Bu, *n*-octyl) could also be successfully utilized. We were pleased to find that the tandem reaction of 1-bromo-2-naphthaldehyde **1g** with alkyl-2-cyanoacetates **2a-2d** afforded the fluorescence active (vide infra) alkyl 3-amino-4-cyanophenanthrene-2-carboxylates **3ga** (X-ray in Figure S2, SI), **3gb**, and **3gd** in good yields. We subsequently examined the reaction of 2-bromo-3-pyridinecarboxaldehyde **1h** with alkyl cyanoacetates **2a**, **2b** and **2d**; satisfyingly, the reaction furnished the desired products **3ha**, **3hb** and **3hd** in moderate to good yields of 72-52% (Table 3) along with < 17% yield of **3ha'** and **3hd'**. We successfully extended this chemistry to the synthesis of alkyl 7-amino-8-cyano-5-



^aReaction conditions: 1 (0.5 mmol), 2 (1.0 mmol), K₂CO₃ (1.6 mmol), CuI (0.05 mmol), DMSO (2 mL) in a stoppered Schlenk tube in air at 120 °C (oil bath) for 3h. ^bYield of product **3**.

E



 Table 3. Cu(I)-Catalyzed Tandem Synthesis of Multifunctional Quinolines and

 Benzo[b]carbazoles 3^{a,b}

^{*a*}Reaction conditions: **1** (0.5 mmol), **2** (1.0 mmol), K₂CO₃ (1.6 mmol), CuI (0.05 mmol), DMSO (2 mL), in a stoppered Schlenk tube in air at 120 °C (oil bath) for 3 h. ^{*b*}Yield of product **3**.

In order to decipher the reaction pathway, we performed the reaction of 2bromobenzaldehyde 1a (1 mmol) with ethyl cyanoacetate 2a (1 mmol) using K₂CO₃ (1.0

mmol) in DMSO at rt (25 °C) for 1 h that provided the Knoevenagel adduct I in 82% yield (Scheme 2, eq 1). In the absence of K_2CO_3 , the Knoevenagel adduct I was obtained in only 38% yield, along with unreacted starting material **1a** (Scheme 2, eq 2). When we treated I (1 mmol) with ethyl cyanoacetate **2a** (1 mmol), CuI (0.1 mmol), and K_2CO_3 (2.0 mmol) at 120 °C for 2 h, the desired product **3aa** was obtained in excellent yield (Scheme 2, eq 3). In the absence of CuI, the reaction did not proceed (Scheme 2, eq 4). These simple observations clearly showed that copper mediated C-arylation is the key step in this tandem process.

Scheme 2. Control Experiments



On the basis of the above control experiments and literature reports,^{17, 15b, 5i} a plausible pathway for the formation of **3aa** is illustrated in Scheme 3. Initially, the reaction of 2-bromobenzaldehyde **1** and alkyl cyanoacetate **2** generated Knoevenagel adduct **I**. Copper-catalyzed coupling of **I** with **2** leads to C-arylation product **II**. Then, the base abstracts a

proton from an active methine carbon of **II** forming the highly nucleophilic carbanion **III** that attacks the nitrile leading to the cyclized intermediate **IV** which undergoes hydrolysis to give **V**. Decarboxylation, followed by aromatization of **V** affords alkyl 3-amino-4-cyano-2-naphthoate (**3**). We wish to emphasize here that (i) the later stages of this reaction pathway as well as the resulting products differ from the one reported using β -ketoesters wherein an oxetane intermediate and elimination of carboxylate species is proposed^{15e} and (ii) the formation of **3ha'** and **3hd'** albeit in very low yields perhaps offers a different type of mechanism for their formation but currently it is a puzzling observation.

Scheme 3. Plausible Pathway for the Formation of Products 3



For expanding the scope of substrates, we treated 2-bromobenzaldehyde 1a (1.0 mmol) with methyl cyanoacetate 2b (1.0 mmol) and K₂CO₃ (1.0 mmol) in DMSO at rt for 1 h to generate intermediate I; after consumption of all of 1a (TLC), CuI (10 mol%), K₂CO₃

(2.0 mmol) and dimethylmalonate (**4b**, 1.0 mmol) were added and the contents heated at 120 °C for 2 h. The overall reaction proceeded smoothly to afford dimethyl 2aminonaphthalene-1,3-dicarboxylate **5abb** (X-ray in Figure S6, SI) in 68% yield (Table 4). The scope of this protocol was examined using diverse 2-bromobenzaldehydes (**1a**, **1c**, **1d** and **1g**), alkyl 2-cyanoacetates (**2a-b**) and dialkylmalonates (**4a-c**; $\mathbb{R}^5 = \mathbb{E}t$, Me, *i*-Pr). The corresponding products **5aaa-5aac**, **5abb**, **5cbb**, and **5dab** were isolated in good yields (Table 4). This new one-pot sequence also allows for the synthesis of polysubstituted phenanthrene **5gaa** in 80% yield (Table 4).





^{*a*}Reaction conditions: **1** (~0.5 mmol), **2** (0.5 mmol), K₂CO₃ (0.5 mmol), DMSO (2 mL), at rt for 1 h in a stoppered Schlenk tube in air; then CuI (0.05 mmol), **4** (0.5 mmol), and K₂CO₃ (1.0 mmol) were added in air and contents heated at 120 °C (oil bath) for 2 h. ^{*b*}Yield of product **5**.

In order to explore the above one pot sequential method further, malononitrile (6; 1 mmol) was used to replace dialkylmalonate 4 in Table 4. Very interestingly, this reaction afforded 2-aminonaphthalene-1,3-dicarbonitriles **7aa**, **7ca and 7da** in 67-82% yield (Table 5). Mechanistically, the reaction seems to involve an initial C-arylation via copper-catalyzed coupling of I with 6 to give intermediate VI. Hydrolysis of ester on VI leads to VII. Decarboxylation followed by cyclization of VII leads to intermediate VIII. Finally, aromatization of VIII afforded product **7**. Thus the later stages of this pathway appears to be slightly different from that shown in Scheme 3 or the one involving β -ketoesters.^{15e}

Table 5. Cu(I)-Catalyzed One-Pot Sequential Synthesis of 2-Aminonaphthalene-1,3-

dicarbonitrile 7^{*a,b*}



^{*a*}Reaction conditions: **1** (0.5 mmol), **2** (0.5 mmol), K_2CO_3 (0.5 mmol), DMSO (2 mL), at rt for 1h, in a stoppered Schlenk tube in air; then CuI (0.05 mmol), **6** (0.5 mmol), and K_2CO_3 (1.0 mmol) were added in air and the contents heated at 120 °C (oil bath) for 2h.^{*b*} Yield of product 7.

To demonstrate the utility of the present protocol, further structural elaboration of compounds **3aa** and **3da** was performed via the reaction with formamide **8** in the presence of K_2CO_3 to give the corresponding benzo[g]quinazoline **9**. Later, the compound **9aa** was converted to 4-oxo-3-phenyl-3,4-dihydrobenzo[g]quinazoline-10-carbonitrile **11aa** by using a reported Cu-mediated C-N bond forming reaction (Scheme 4);¹⁸ there was no perceptible reaction when CuI was used in place of Cu(OAc)₂ under the conditions given in Table 2.

Scheme 4. Structural Elaboration of 3



To examine the scalability of the optimized method, the tandem reaction between 2bromobenzaldehyde **1a** and ethyl cyanoacetate **2a** was carried out under standard conditions on a gram scale. The expected product **3aa** could be obtained in 68% isolated yield (Scheme 5).

Scheme 5. Gram-Scale Transformation



Photophysical Properties of Phenanthrenes 3ga, 3gd and 5gaa.

We expected phenanthrene-2-carboxylates **3ga**, **3gd** and **5gaa** to be photo chemically active and perhaps could be used later. Hence the UV-visible and fluorescence spectra were recorded. The emission spectra of compounds **3ga**, **3gd**, and **5gaa** were recorded upon excitation at 330 nm. As expected, these compounds show moderate fluorescence emission intensity (cf. Figure 2). The absorption peak wavelengths (λ_{abs}), molar extinction coefficient (ϵ), emission peak wavelengths (λ_{em}) in solution are summarized in Table 6 and Figure 2.

Table	6.	UV-Visible	Absorption	and	Fluorescence	Emission	Properties	of
Phenanthrene Carboxylates 3ga, 3gd and 5gaa								

entry	compound	$\lambda_{abs} (nm)^a$	\mathcal{E} (10 ⁴ M ⁻¹ cm ⁻¹)	$\lambda_{em} (nm)^b$	
1	3ga	435	1.0	461	
2	3gd	435	3.3	467	
3	5gaa	427	1.9	488	

^{*a*}UV-visible absorption wavelengths, ^{*b*}Emission wavelengths at rt in toluene at concentration of 1×10^{-5} M.



Figure 2. (A) Absorbance and (B) Fluorescence emission spectra of compounds 3ga, 3gd and 5gaa in toluene (10⁻⁵ M) at rt (25 °C).

CONCLUSIONS

We have developed a simple and highly efficient ligand-free one pot Cu(I)-mediated tandem or sequential protocol for the synthesis of multifunctional naphthalenes, phenanthrenes, quinolines and benzo[*b*]carbazoles. The protocol entails readily available substrates such as *o*-bromoarylaldehydes, alkyl cyanoacetates, dialkylmalonates and active methylene nitriles. The developed methodology was utilized further for the synthesis of 4-oxo-3-phenyl-3,4-dihydrobenzo[g]quinazoline-10-carbonitrile. The photophysical properties of phenanthrene carboxylates reveal possible applications for the highly fluorescent materials.

EXPERIMENTAL SECTION

General Methods. All reactions were carried out in air, unless specified otherwise. All required chemicals were procured and used as purchased without further purification, unless noted otherwise. Melting points were determined using a SUPERFIT hot stage apparatus and were uncorrected. ¹H and ¹³C NMR spectra were recorded using 5 mm tubes on a Bruker 400 and 500 MHz NMR spectrometer [field strengths: 400/100 or 500/125 MHz respectively] in CDCl₃/DMSO-d₆ solution with shifts referenced to TMS $({}^{1}H, {}^{13}C; \ddot{o} = 0)$. All J values are in Hz. Infrared spectra were recorded neat or by using KBr pellets on an FT/IR spectrometer. Mass spectra were recorded using HRMS (ESI-TOF analyzer) equipment. Crystallographic data were collected at 293 K on an X-ray diffractometer system using Mo-K_a radiation ($\lambda = 0.71073$ Å) or Cu-K_a ($\lambda = 1.54184$ Å) radiation. Structures were solved and refined using standard methods.¹⁹ Thin-layer chromatography was performed on silica/alumina plates and components were visualized by observation under iodine/UV light at 254 nm. Column chromatography was performed on silica gel (100-200 mesh)/neutral alumina, for column elution process hexane-EtOAc mixture was used as the eluent unless otherwise stated.

o-Bromoarylaldehydes (**1a-1h**) were purchased from Aldrich and used without further purification and **1i** was prepared by using a literature report.¹⁶

(i). General procedure for the synthesis of compounds 3aa-3ib. To an oven dried Schlenk tube, with a magnetic stirrer bar was added *o*-bromoaryl aldehyde 1 (~0.5 mmol, 1 equiv, individual quantities given below), CuI (0.05 mmol, 10 mol %), K₂CO₃ (1.6 mmol, 3 equiv), alkyl cyanoacetate 2 (1.0 mmol, 2 equiv) and DMSO (2 mL). The tube with the contents was stoppered and heated at 120 °C (oil bath) for 3h. After completion of the reaction as monitored by TLC, the crude reaction mixture was cooled to rt (25 °C), diluted with ethyl acetate (20 mL) and passed through a short celite column. The resulting solution was washed with water (2 x 20 mL) and the aqueous part extracted twice with ethyl acetate (20 mL). The combined organic layer was washed with brine solution (2 x 20 mL), dried over anhydrous Na₂SO₄, and concentrated in vacuum. The residue was then purified by using silica gel column chromatography using hexane-ethyl acetate (9:1) as eluent to afford the pure desired compounds 3. Compounds 3aa-3ib were prepared from the appropriate *o*-bromoaryl aldehyde 1 and alkyl cyanoacetate 2 by using the same procedure using similar molar quantities/ratio.

(ii). Procedure for the Scale-up Reaction. To an oven dried Schlenk tube, with a magnetic stirrer bar was added *o*-bromoaryl aldehyde 1 (1.000 g, 5.40 mmol), CuI (0.103 g, 0.54 mmol), K_2CO_3 (2.200 g, 16.20 mmol), ethyl cyanoacetate 2 (1.15 mL, 10.81 mmol) and DMSO (20 mL). The tube with the contents was stoppered and heated at 120 °C (oil bath) for 3h. After completion of the reaction as monitored by TLC, the crude reaction mixture was cooled to rt (25 °C), diluted with ethyl acetate (100 mL) and passed through a short celite (ca 10 g) column. The resulting solution was washed with water and the aqueous part extracted twice with ethyl acetate (2 x 100 mL). The combined organic layer was washed with brine solution (2 x 100 mL), dried over anhydrous

 Na_2SO_4 , and concentrated in vacuum. The residue was then purified by using silica gel column chromatography using hexane-ethyl acetate (9:1) as eluent to afford the desired compound **3aa** (0.88 g, 68%) as a vellow solid).

Ethyl 3-amino-4-cyano-2-naphthoate (3aa). This compound was prepared by using precursors **1a** (100 mg, 0.540 mmol) and **2a**. Yellow solid, yield 0.106 g (81%); mp 146-148 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.68 (s, 1H), 7.90 (d, J = 8.4 Hz, 1H), 7.79 (d, J = 8.0 Hz, 1H), 7.66-7.63 (m, 1H), 7.34 (t, J = 7.4 Hz, 1H), 6.71 (br s, 2H), 4.46 (q, J = 7.2 Hz, 2H), 1.48 (t, J = 7.2 Hz, 3H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 166.7, 151.0, 138.8, 136.1, 131.5, 130.1, 124.8, 123.8, 122.9, 116.9, 114.1, 88.9, 61.6, 14.3; IR (neat) v_{max} 3427, 3327, 2967, 2926, 2203, 1739, 1613, 1366, 1210, 1072, 1029, 787, 747 cm⁻¹; HRMS (ESI) calcd for C₁₄H₁₂N₂NaO₂ [M⁺+Na] *m/z* 263.0796, found 263.0792.

Methyl 3-amino-4-cyano-2-naphthoate (3ab): This compound was prepared by using precursors **1a** (100 mg, 0.540 mmol) and **2b**. Yellow solid, yield 0.095 g (77%); mp 152-154 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.69 (s, 1H), 7.91 (d, J = 8.4 Hz, 1H), 7.79 (d, J = 8.0 Hz, 1H), 7.68-7.64 (m, 1H), 7.36 (t, J = 8.0 Hz, 1H), 6.70 (br s, 2H), 4.01 (s, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 167.1, 151.0, 138.9, 136.2, 131.6, 130.1, 124.8, 123.9, 123.0, 116.8, 113.8, 89.0, 52.5; IR (neat) v_{max} 3440, 3337, 3015, 2962, 2218, 1739, 1625, 1366, 1213, 1076, 790, 744 cm⁻¹; HRMS (ESI) calcd for C₁₃H₁₁N₂O₂ [M⁺+H] *m/z* 227.0821, found 227.0815.

Butyl 3-amino-4-cyano-2-naphthoate (3ac). This compound was prepared by using precursors **1a** (100 mg, 0.540 mmol) and **2c**. Yellow solid, yield 0.105 g (72%); mp 142-

144 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.65 (s, 1H), 7.89 (d, J = 7.6 Hz, 1H), 7.79 (d, J = 8.4 Hz, 1H), 7.66-7.62 (m, 1H), 7.35-7.31 (m, 1H), 6.70 (br s, 2H), 4.40 (t, J = 6.4 Hz, 2H), 1.87-1.80 (m, 2H), 1.58-1.49 (m, 2H), 1.04 (t, J = 7.6Hz, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 166.8, 151.1, 138.7, 136.1, 131.5, 130.1, 124.8, 123.8, 122.9, 116.9, 114.1, 89.0, 65.4, 30.7, 29.7, 19.3, 13.8; IR (neat) v_{max} 3433, 3330, 2957, 2927, 2864, 2207, 1695, 1613, 1455, 1304, 1210, 1077, 794, 750 cm⁻¹; HRMS (ESI) calcd for C₁₆H₁₆N₂O₂ [M⁺+H] *m/z* 269.1290, found 269.1283.

Octyl 3-amino-4-cyano-2-naphthoate (3ad). This compound was prepared by using precursors **1a** (100 mg, 0.540 mmol) and **2d**. Yellow solid, yield 0.120 g (68 %); mp 149-151 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.67 (s, 1H), 7.90 (d, J = 8.4 Hz, 1H), 7.80 (d, J = 8.0 Hz, 1H), 7.67-7.63 (m, 1H), 7.36-7.32 (m, 1H), 6.71 (br s, 2H), 4.40 (t, J = 6.8 Hz, 2H), 1.88-1.81 (m, 2H), 1.53-1.32 (m, 10H), 0.92 (t, J = 6.4 Hz, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 166.8, 151.1, 138.8, 136.1, 131.6, 130.1, 124.8, 123.9, 123.0, 116.9, 114.2, 88.9, 65.8, 31.8, 29.3, 29.2, 28.6, 26.1, 22.7, 14.1; IR (neat) v_{max} 3436, 3334, 2924, 2857, 2208, 1737, 1699, 1616, 1573, 1456, 1367, 1303, 1211, 1170, 1083, 794, 755 cm⁻¹; HRMS (ESI) calcd for C₂₀H₂₅N₂O₂ [M⁺+H] *m/z* 325.1916, found 325.1917.

Ethyl 3-amino-4-cyano-6,7-*dimethoxy-2-naphthoate(3ba)*. This compound was prepared by using precursors **1b** (100 mg, 0.408 mmol) and **2a**. Yellow solid, yield 0.104 g (85%); mp 161-163 °C; ¹H NMR (500 MHz, CDCl₃) δ 8.01 (s, 1H), 7.96 (s, 1H), 6.96 (s, 1H), 6.28 (s, 2H), 4.53 (q, J = 7.0 Hz, 2H), 4.02 (s, 3H), 3.98 (s, 3H), 1.53 (t, J = 7.0 Hz, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 168.5, 153.2, 147.7, 147.3, 137.8, 131.8, 121.6, 117.3, 107.6, 105.1, 104.9, 98.4, 61.2, 55.9, 55.8, 14.5; IR (neat) v_{max} 3481, 3352, 2924,

2854, 2214, 1673, 1614, 1435, 1323, 1295, 1013, 814 cm⁻¹; HRMS (ESI) calcd for $C_{16}H_{16}N_2O_4$ [M⁺+H]: *m/z* 301.1188, found 301.1189.

Butyl 3-amino-4-cyano-6,7*-dimethoxy-2-naphthoate (3bc)*. This compound was prepared by using precursors **1b** (100 mg, 0.408 mmol) and **2c**. Yellow solid, yield 0.107 g (80%); mp 157-159 °C; ¹H NMR (500 MHz, CDCl₃) δ 8.48 (s, 1H), 7.15 (s, 1H), 7.06 (s, 1H), 6.57 (s, 2H), 4.38 (t, J = 6.5 Hz, 2H), 4.07 (s, 3H), 3.99 (s, 3H), 1.85-1.79 (m, 2H), 1.57-1.50 (m, 2H), 1.03 (t, J = 7.5 Hz, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 167.0, 154.3, 150.3, 147.9, 136.3, 133.3, 119.9, 117.3, 111.5, 108.2, 102.0, 88.4, 65.0, 56.2, 56.0, 30.7, 19.3, 13.8; IR (neat) v_{max} 3437, 3336, 2959, 2926, 2873, 2203, 1686, 1614, 1461, 1299, 1094, 1039, 974,752 cm⁻¹; HRMS (ESI) calcd for C₁₈H₂₀N₂O4 [M⁺+H] *m/z* 329.1501, found 329.1502.

Octyl 3-amino-4-cyano-6,7*-dimethoxy-2-naphthoate (3bd)*. This compound was prepared by using precursors **1b** (100 mg, 0.408 mmol) and **2d**. Yellow solid, yield 0.129 g (82%); mp 149-151 °C; ¹H NMR (500 MHz, CDCl₃) δ 8.48 (s, 1H), 7.15 (s, 1H), 7.06 (s, 1H), 6.57 (s, 2H), 4.36 (t, J = 6.5 Hz, 2H), 4.07 (s, 3H), 3.99 (s, 3H), 1.85-1.80 (m, 2H), 1.51-1.32 (m, 10H), 0.91 (t, J = 7.0 Hz, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 167.0, 154.3, 150.3, 147.9, 136.4, 133.3, 119.9, 117.3, 111.6, 108.3, 102.1, 88.4, 65.4, 56.3, 56.0, 31.8, 29.3, 29.2, 28.7, 26.1, 22.7, 14.1; IR (neat) v_{max} 3440, 3343, 2955, 2925, 2855, 2204, 1688, 1611, 1509, 1331, 1298, 1039, 1022, 787, 754 cm⁻¹; HRMS (ESI) calcd for $C_{22}H_{29}N_2O_4$ [M⁺+H] *m/z* 385.2127, found 385.2128.

Ethyl 3-amino-4-cyano-7-methoxy-2-naphthoate (3ca). This compound was prepared by using precursors **1c** (100 mg, 0.465 mmol) and **2a**. Yellow solid, yield 0.100 g (79 %); mp 165-167 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.60 (s, 1H), 7.83 (d, *J* = 8.8 Hz, 1H),

7.33 (dd, J = 8.8, 2.4 Hz, 1H), 7.11 (d, J = 2.8 Hz, 1H), 6.54 (s, 2H), 4.46 (q, J = 7.2 Hz, 2H), 3.91 (s, 3H). 1.49 (t, J = 7.2 Hz, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 166.7, 155.9, 150.1, 137.8, 131.3, 125.5, 124.6, 123.7, 117.2, 114.7, 109.3, 88.0, 61.8, 55.8, 14.5; IR (neat) v_{max} 3429, 3329, 2979, 2205, 1739, 1614, 1366, 1211, 1074, 1031, 787, 749 cm⁻¹; HRMS (ESI) calcd for C₁₅H₁₅N₂O₃ [M⁺+H] *m/z* 271.1083, found 271.1077. *Methyl 3-amino-4-cyano-7-methoxy-2-naphthoate* (*3cb*): This compound was prepared by using precursors **1c** (100 mg, 0.465 mmol) and **2b**. Yellow solid, yield 0.085 g (71 %); mp 170-172 °C; ¹H NMR (500 MHz, CDCl₃) δ 8.59 (s, 1H), 7.83 (d, J = 9.0 Hz, 1H), 7.33 (dd, J = 9.0, 2.5 Hz, 1H), 7.10 (d, J = 3.0 Hz, 1H), 6.52 (br s, 2H), 3.99 (s, 3H), 3.91 (s, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 167.2, 156.2, 149.5, 137.3, 131.5, 125.7, 124.5, 124.3, 116.9, 114.2, 108.0, 89.5, 55.4, 52.4; IR (neat) v_{max} 3446, 3345, 2959, 2922, 2209, 1738, 1704, 1617, 1505, 1374, 1289, 1215, 1142, 1089, 1012, 824, 787 cm⁻¹; HRMS (ESI) calcd for C₁₄H₁₃N₂O₃ [M⁺+H] *m/z* 257.0926, found 257.0929.

Butyl 3-amino-4-cyano-7-methoxy-2-naphthoate (3cc). This compound was prepared by using precursors **1c** (100 mg, 0.465 mmol) and **2c**. Yellow solid, yield 0.095 g (68 %); mp 161-163 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.57 (s, 1H), 7.82 (d, *J* = 9.2 Hz, 1H), 7.33 (dd, *J* = 9.2, 2.4 Hz, 1H), 7.11 (d, *J* = 2.8 Hz, 1H), 6.53 (br s, 2H), 4.40 (t, *J* = 6.4 Hz, 2H), 3.92 (s, 3H), 1.87-1.80 (m, 2H), 1.57-1.49 (m, 2H), 1.05 (t, *J* = 7.2 Hz, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 166.8, 156.1, 149.6, 137.1, 131.4, 125.7, 124.4, 124.2, 117.0, 114.5, 107.9, 89.5, 65.3, 55.4, 30.7, 19.3, 13.8; IR (neat) *v*_{max} 3437, 3335, 2925, 2858, 2207, 1738, 1691, 1609, 1459, 1375, 1227, 1080, 1024, 799 cm⁻¹; HRMS (ESI) calcd. for C₁₇H₁₉N₂O₃ [M⁺+H] *m/z* 299.1396, found: 299.1390.

Octyl 3-amino-4-cyano-7-methoxy-2-naphthoate (3cd). This compound was prepared by using precursors **1c** (100 mg, 0.465 mmol) and **2d**. Yellow solid, yield 0.121 g (73%); mp 153-155 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.56 (s, 1H), 7.82 (d, *J* = 9.2 Hz, 1H), 7.33 (dd, *J* = 8.8, 2.4 Hz, 1H), 7.10 (d, *J* = 2.4 Hz, 1H), 6.53 (br s, 2H), 4.39 (t, *J* = 6.8 Hz, 2H), 3.92 (s, 3H), 1.88-1.81 (m, 2H), 1.53-1.27 (m, 10H), 0.92 (t, *J* = 6.8 Hz, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 166.8, 156.1, 149.6, 137.1, 131.4, 125.6, 124.4, 124.2, 117.0, 114.4, 107.9, 89.4, 65.7, 55.4, 31.8, 29.2₄, 29.1₈, 28.6, 26.0, 22.6, 14.1; IR (neat) v_{max} 3440, 3337, 2924, 2854, 2207, 1694, 1611, 1505, 1463, 1380, 1296, 1233, 1184, 1170, 1086, 1005, 818, 731 cm⁻¹; HRMS (ESI) calcd for C₂₁H₂₆N₂O₃ [M⁺+H] *m/z* 355.2022, found 355.2019.

Ethyl 3-amino-4-cyano-7-fluoro-2-naphthoate (3da). This compound was prepared by using precursors **1d** (100 mg, 0.493 mmol) and **2a**. Yellow solid, yield 0.089 g (70 %); mp 161-163 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.59 (s, 1H), 7.90-7.86 (m, 1H), 7.45-7.40 (m, 2H), 6.65 (br s, 2H), 4.46 (q, J = 7.2 Hz, 2H), 1.49 (t, J = 7.2 Hz, 3H); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 166.4, 159.0 (J = 243.6 Hz), 150.5, 137.6 (J = 5.0 Hz), 132.9, 125.2 (J = 8.3 Hz), 125.1,121.6 (J = 25.2 Hz), 116.6, 115.2, 113.1 (J = 20.8 Hz), 89.2, 61.7, 14.3; IR (neat) ν_{max} 3429, 3330, 2958, 2922, 2858, 2209, 1739, 1696, 1616, 1506, 1372, 1281, 1220, 1162, 1087, 1025, 823, 794 cm⁻¹; HRMS (ESI) calcd for C₁₄H₁₁FN₂NaO₂ [M⁺+Na] *m/z* 281.0697, found 281.0696.

Butyl 3-amino-4-cyano-7-fluoro-2-naphthoate (3dc). This compound was prepared by using precursors **1d** (100 mg, 0.493 mmol) and **2c**. Yellow solid, yield 0.095 g (67 %); mp 154-156 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.60 (s, 1H), 7.92-7.88 (m, 1H), 7.46-7.41 (m, 2H), 6.67 (s, 2H), 4.41 (t, J = 6.8 Hz, 2H), 1.87-1.80 (m, 2H), 1.56-1.48 (m, 2H),

1.04 (t, J = 7.2 Hz, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 166.5, 159.0 (J = 243.7 Hz), 150.5, 137.6 (J = 5.1 Hz), 132.9, 125.2 (J = 8.2 Hz), 125.1 (J = 8.8 Hz), 121.6 (J = 25.3 Hz), 116.7, 115.2, 113.2 (J = 20.1 Hz), 89.2, 65.6, 30.6, 19.3, 13.8; IR (neat) v_{max} 3440, 3338, 2961, 2208, 1738, 1609, 1506, 1373, 1297, 1225, 1082, 1036, 818, 740 cm⁻¹; HRMS (ESI) calcd for C₁₆H₁₆FN₂O₂ (M⁺ + H) m/z 287.1196, found 287.1183.

Octyl 3-amino-4-cyano-7-fluoro-2-naphthoate (3dd). This compound was prepared by using precursors **1d** (100 mg, 0.493 mmol) and **2d**. Yellow solid, yield 0.110 mg (65 %); mp 149-151 °C; ¹H NMR (500 MHz, CDCl₃) δ 8.59 (s, 1H), 7.91-7.88 (m, 1H), 7.45-7.41 (m, 2H), 6.66 (s, 2H), 4.40 (t, J = 6.5 Hz, 2H), 1.87-1.81 (m, 2H), 1.51-1.45 (m, 2H), 1.43-1.31 (m, 8H), 0.92 (t, J = 7.0 Hz, 3H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 166.5, 159.0 (J = 243.6 Hz), 150.5, 137.6 (J = 5.0 Hz), 132.9, 125.2 (J = 8.4 Hz), 125.1 (J = 8.8 Hz), 121.6 (J = 25.1 Hz), 116.7, 115.2, 113.2 (J = 20.1 Hz), 89.2, 65.9, 31.8, 29.24, 29.20, 28.6, 26.0, 22.7, 14.1; IR (neat) v_{max} 3483, 3363, 2958, 2927, 2856, 2204, 1796, 1693, 1609, 1572, 1376, 1279, 1216, 1157, 1077, 1024, 831, 732 cm⁻¹; HRMS (ESI) calcd for C₂₀H₂₄FN₂O₂ [M⁺+H] *m/z* 343.1822, found 343.1814.

Ethyl 3-amino-4-cyano-8-fluoro-2-naphthoate (3ea). This compound was prepared by using precursors **1e** (100 mg, 0.493 mmol) and **2a**. Yellow solid, yield 0.078 g (61%); mp 160-162 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.95 (s, 1H), 7.68 (d, J = 8.4 Hz, 1H), 7.61-7.55 (m, 1H), 7.01-6.97 (m, 1H), 6.83 (s, 2H), 4.48 (q, J = 7.2 Hz, 2H), 1.50 (t, J = 7.2 Hz, 3H); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 166.5, 160.0 (J = 254.2 Hz), 151.6, 137.3 (J = 3.4 Hz), 132.0 (J = 9.3 Hz), 131.5 (J = 5.6 Hz), 118.8 (J = 4.1 Hz), 116.5, 115.1, 114.3, 107.5 (J = 19.6 Hz), 88.9, 61.8, 14.3; IR (neat) v_{max} 3429, 3327, 2984, 2926, 2207,

1735, 1696, 1616, 1574, 1438, 1376, 1273, 1181, 1122, 1048, 866, 796, 747 cm⁻¹; HRMS (ESI) calcd for $C_{14}H_{11}FNaN_2O_2$ [M⁺+Na] *m/z* 281.0702, found 281.0700.

Methyl 3-amino-4-cyano-8-fluoro-2-naphthoate (3eb). This compound was prepared by using precursors **1e** (100 mg, 0.493 mmol) and **2b**. Yellow solid, yield 0.082 g (68 %); mp 165-167 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.95 (s, 1H), 7.68 (d, J = 8.4 Hz, 1H), 7.61-7.55 (m, 1H), 7.01-6.97 (m, 1H), 6.81 (s, 2H), 4.02 (s, 3H); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 166.9, 160.0 (J = 254.5 Hz), 151.5, 137.3, 132.1 (J = 9.3 Hz), 131.7 (J = 5.8 Hz), 118.8 (J = 4.2 Hz),116.5, 115.2 (J = 17.2 Hz), 114.0, 107.1 (J = 19.6 Hz), 88.9, 52.6; IR (neat) v_{max} 3440, 3335, 2960, 2922, 2846, 2215, 1710, 1624, 1571, 1443, 1365, 1306, 1208, 1077, 791 cm⁻¹; HRMS (ESI) calcd for C₁₃H₁₀FN₂O₂ [M⁺+H] *m/z* 245.0726, found 245.0726.

Ethyl 3-amino-4-cyano-7-methyl-2-naphthoate (3fa). This compound was prepared by using precursors **1f** (100 mg, 0.502 mmol) and **2a**. Yellow solid, yield 0.103 g (80 %); mp 140-142 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.63 (s, 1H), 7.69-7.67 (m, 2H), 7.17(dd, J = 8.4, 1.6 Hz, 1H), 6.68 (s, 2H), 4.44 (q, J = 7.2 Hz, 2H), 2.54 (s, 3H), 1.48 (t, J = 7.2 Hz, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 166.8, 151.2, 142.4, 138.5, 136.4, 129.9, 126.2, 123.1, 122.2, 117.1, 113.1, 88.4, 61.4, 22.2, 14.3; IR (neat) v_{max} 3427, 3328, 2974, 2931, 2203, 1739, 1705, 1612, 1572, 1366, 1209, 1151, 1071, 1030, 786, 749 cm⁻¹; HRMS (ESI) calcd for C₁₅H₁₅N₂O₂ [M⁺+H] *m/z* 255.1134, found 255.1137.

Ethyl 2-amino-3-cyanophenanthrene-1-carboxylate (3ga). This compound was prepared by using precursors **1g** (100 mg, 0.425 mmol) and **2a**. Yellow solid, yield 0.089 g (72%); mp 176-180 °C; ¹H NMR (400 MHz, CDCl₃) δ 9.74-9.71 (m, 1H), 8.61 (s, 1H), 7.87-7.85 (m, 1H), 7.70-7.66(m, 2H), 7.59-7.54 (m, 2H), 7.00 (br s, 2H), 4.47 (q, *J* = 7.2 Hz, 2H),

1.49 (t, J = 7.2 Hz, 3H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 166.8, 153.1, 138.5, 134.9, 134.3, 128.9, 127.5, 127.3, 126.6, 125.3, 123.2, 119.9, 112.3, 88.7, 61.5, 14.3; IR (neat): v_{max} 3464, 3316, 2922, 2190, 1731, 1601, 1370, 1215, 1093, 1064, 804, 747 cm⁻¹; HRMS (ESI) calcd for C₁₈H₁₄N₂NaO₂ [M⁺+Na] *m/z* 313.0953, found: 313.0945.

Methyl 2-amino-3-cyanophenanthrene-1-carboxylate (**3gb**). This compound was prepared by using precursors **1g** (100 mg, 0.425 mmol) and **2b**. Yellow solid, yield 0.073 g (62%); mp 178-182 °C; ¹H NMR (400 MHz, CDCl₃) δ 9.74-9.72 (m, 1H), 8.61 (s, 1H), 7.88-7.86 (m, 1H), 7.71-7.68 (m, 2H), 7.56 (s, 2H), 6.99 (br s, 2H), 4.01 (s, 3H); ¹³C{¹H} NMR(100 MHz, CDCl₃) δ 167.2, 153.0, 138.5, 134.9, 134.3, 128.94, 128.91, 127.4, 127.2, 126.6, 125.34, 125.28, 123.2, 120.0, 111.9, 88.7, 52.4; IR (neat) v_{max} 3447, 3339, 2958, 2921, 2010, 1702, 1610, 1459, 1609, 1459, 1316, 1279, 1260, 1219, 1092, 1026, 909, 803, 748, 563 cm⁻¹; HRMS (ESI) calcd for C₁₇H₁₃N₂O₂ [M⁺+H] *m/z* 277.0977, found 277.0979.

Octyl 3-amino-4-cyanophenanthrene-2-carboxylate (3gd). This compound was prepared by using precursors **1g** (100 mg, 0.425 mmol) and **2d**. Yellow solid, yield 0.104 g (65%); mp 120-124 °C;¹H NMR (500 MHz, CDCl₃) δ 9.75-9.74 (m, 1H), 8.61 (s, 1H), 7.89-7.87 (m, 1H), 7.71-7.69 (m, 2H), 7.61-7.57 (m, 2H), 7.02 (br s, 2H), 4.41 (t, *J* = 6.5 Hz, 2H), 1.89-1.83 (m, 2H), 1.54-1.48 (m, 2H), 1.44-1.32 (m, 8H), 0.93 (t, *J* = 7.0 Hz, 3H); ¹³C{¹H} NMR(125 MHz, CDCl₃) δ 166.8, 153.1, 138.4, 134.9, 134.3, 128.9, 127.4, 127.3, 126.6, 125.3, 123.2, 120.0, 112.3, 88.6, 65.7, 31.8, 29.3, 29.2, 28.7, 26.1, 22.7, 14.1; IR (neat) *v*_{max} 3453, 3335, 2923, 2851, 2193, 1737, 1692, 1608, 1581, 1517, 1313, 1282, 1217, 1199, 1122, 1094, 1024, 971, 807, 752, 729, 682, 659 cm⁻¹; HRMS (ESI) calcd for C₂₄H₂₇N₂O₂ [M⁺+H] *m/z* 375.2073, found: 375.2091.

Ethyl 7-amino-8-cyanoquinoline-6-carboxylate (3ha). This compound (along with **3ha'**; see below) was prepared by using precursors **1h** (100 mg, 0.537 mmol) and **2a**. R_f 0.3 (ethyl acetate:hexane 1:9).Yellow solid, yield 0.086 g (66%); mp 151-153 °C; ¹H NMR (500 MHz, CDCl₃) δ 8.98 (dd, J = 4.0, 1.5 Hz, 1H), 8.65 (s, 1H), 8.08 (dd, J = 8.0, 1.5 Hz, 1H), 7.29-7.26 (m, 1H), 6.95 (v br, 2H), 4.48 (q, J = 7.0 Hz, 2H), 1.48 (t, J = 7.0 Hz, 3H); ¹³C {¹H} NMR (125 MHz, CDCl₃) δ 166.2, 155.0, 153.2, 151.3, 138.1, 137.6, 119.5, 119.4, 116.2, 114.8, 92.0, 61.8, 14.2; IR (neat) v_{max} 3416, 3325, 3301, 2215, 1738, 1628, 1581, 1565, 1370, 1310, 1206, 1083, 1033, 924, 794, 727, 510 cm⁻¹; HRMS (ESI) calcd for C₁₃H₁₂N₃O₂ [M⁺+H] *m/z* 242.0930, found 242.0930.

Ethyl 7-amino-6-cyanoquinoline-8-carboxylate (3ha'): $R_f 0.5$ (ethyl acetate:hexane 1:9). This compound was prepared by using precursors **1h** and **2a** along with **3ha** as described above. Yield 0.015 g (12%) as yellow solid; mp: 142-145 °C; ¹H NMR (500 MHz, CDCl₃) δ : 8.74 (s, 1H), 8.01 (dd, J = 7.0, 1.0 Hz, 1H), 7.90 (dd, J = 8.0, 1.5 Hz, 1H), 7.28 (t, J = 6.0 Hz, 1H), 5.88 (s, 2H), 4.46 (q, J = 7.0 Hz, 2H), 1.48 (t, J = 7.0 Hz, 3H); ¹³C{¹H} NMR (125 MHz, CDCl₃): 165.9, 157.6, 150.2, 142.5, 137.9, 133.8, 122.4, 121.6, 117.6, 111.8, 109.0, 61.8, 14.2; IR (neat): v_{max} 3416, 3294, 3183, 2906, 2221, 1696, 1607, 1564, 1478, 1305, 1281, 1203, 1100, 1020, 795, 755 cm⁻¹; HRMS (ESI): Calcd. for C₁₃H₁₂N₃O₂ [M⁺+H]: m/z 242.0930. Found: 242.0930.

Methyl 7-amino-8-cyanoquinoline-6-carboxylate (3hb). This compound was prepared by using precursors **1h** (100 mg, 0.537 mmol) and **2b**. Yellow solid, yield 0.088 g (72%); mp 134-141 °C; ¹H NMR (500 MHz, CDCl₃) δ 8.97 (dd, J = 4.5, 1.5 Hz, 1H), 8.65 (s, 1H), 8.07 (dd, J = 8.0, 1.5 Hz, 1H), 7.29-7.26 (m, 1H), 6.94 (v br, 2H), 4.01 (s, 3H); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 166.7, 155.0, 153.2, 151.2, 138.3, 137.7, 119.5,

116.2, 114.5, 91.8, 52.7; IR (neat) v_{max} 3479, 3410, 3299, 2949, 2210, 1701, 1619, 1581, 1562, 1488, 1428, 1310, 1197, 1071, 1020, 918, 794, 698, 605 cm⁻¹; HRMS (ESI) calcd for C₁₂H₁₀N₃O₂ [M⁺+H] *m/z* 228.0773, found 228.0774.

Octyl 7-amino-8-cyanoquinoline-6-carboxylate (3hd). This compound (along with **3hd**'; see below) was prepared by using precursors **1h** (100 mg, 0.537 mmol) and **2d**. R_f 0.3 (ethyl acetate:hexane 1:9). Yellow solid, yield 0.107 g (61%); mp 162-164 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.99-8.98 (m, 1H), 8.63 (s, 1H), 8.10-8.08 (m, 1H), 7.30-7.27 (m, 1H), 6.92 (v br, 2H), 4.41 (t, *J* = 6.8 Hz, 2H), 1.88-1.81 (m, 2H), 1.49-1.32 (m, 10H), 0.91 (t, *J* = 6.0 Hz, 3H); ¹³C{¹H} NMR(125 MHz, CDCl₃) δ 166.3, 155.0, 153.2, 151.3, 138.1, 137.7, 119.5, 119.4, 116.3, 114.8, 91.9, 66.0, 31.8, 29.2₃, 29.1₉, 28.6, 26.0, 22.7, 14.1; IR (neat) v_{max} 3425, 3310, 3059, 2961, 2227, 1702, 1628, 1566, 1488, 1463, 1439, 1278, 1198, 1015, 966, 755, 691 cm⁻¹; HRMS (ESI), calcd for C₁₉H₂₄N₃O₂ [M⁺+H] *m/z* 326.1869, found 326.1869.

Octyl 7-amino-8-cyanoquinoline-6-carboxylate (3hd'): This compound was prepared by using precursors **1h** and **2d** along with **3hd** as described above. R_f 0.5 (ethyl acetate:hexane 1:9). Yield 0.030 g (17%) as yellow solid; mp: 169-171 °C; ¹H NMR (500 MHz, CDCl₃) δ : 8.71 (s, 1H), 8.00 (dd, J = 7.5, 1.5 Hz, 1H), 7.93 (dd, J = 8.0, 1.5 Hz, 1H), 7.29-7.26 (m, 1H), 5.95 (s, 2H), 4.38 (t, J = 6.5 Hz, 2H), 1.86-1.80 (m, 2H), 1.50-1.45 (m, 2H), 1.40-1.31 (m, 8H), 0.91 (t, J = 6.5 Hz, 3H); ¹³C{¹H} NMR (125 MHz, CDCl₃): 166.0, 157.6, 150.2, 142.5, 138.0, 133.9, 122.3, 121.6, 117.7, 111.8, 108.9, 66.0, 31.8, 29.2₄, 29.1₉, 28.6, 26.0, 22.7, 14.1; IR (neat): v_{max} 3426, 3311, 2954, 2919, 2868, 2226, 1701, 1628, 1565, 1458, 1377, 1305, 1278, 1197, 1081, 966, 892, 799, 758, 590 cm⁻¹; HRMS (ESI): Calcd. for C₁₉H₂₄N₃O₂ [M⁺+H]: m/z 326.1869. Found: 326.1869.

Ethvl 8-amino-3-bromo-7-cvano-5-ethvl-5H-benzo[b]carbazole-9-carboxvlate (3ia). This compound was prepared by using precursors 1i (100 mg, 0.262 mmol) and 2a. Yellow solid, vield 0.078 g (68%); mp 208-210 °C; ¹H NMR (500 MHz, CDCl₃) δ 8.77 (s, 1H),8.34 (s, 1H),7.86 (d, J = 8.0 Hz, 1H),7.52 (s, 1H),7.44 (d, J = 1.5 Hz, 1H), 7.31 (dd, J = 8.0, 1.5 Hz, 1H), 6.68 (br, 2H), 4.38 (q, J = 7.0 Hz, 2H), 4.27 (q, J = 7.0 Hz, 2H), 1.42 (t, J = 8.0 Hz, 6H); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 166.9, 151.3, 143.8, 143.1, $139.9, 135.1, 122.9_3, 122.8_9, 122.3, 121.7_0, 121.6_6, 121.0, 119.8, 118.0, 111.8, 111.3,$ 98.5, 86.9, 61.3, 37.9, 14.4, 13.3; IR (neat) v_{max}3343, 2921, 2852, 2203, 1734, 1682, 1460, 1254, 1084, 799, 582 cm⁻¹; HRMS (ESI) calcd for C₂₂H₁₈⁷⁹BrN₃O₂ and $C_{22}H_{18}^{81}BrN_{3}O_{2}[M^{+}+H] m/z$ 436.0661 and 438.0661, found 436.0658 and 438.0640. *Methyl* 8-amino-3-bromo-7-cyano-5-ethyl-5H-benzo[b]carbazole-9-carboxylate (3ib). This compound was prepared by using precursors 1i (100 mg, 0.262 mmol) and 2b. Yellow solid, yield 0.071 g (64%); mp 195-197 °C; ¹H NMR (500 MHz, CDCl₃) δ 8.80 (s, 1H), 8.35 (s, 1H), 7.92 (d, J = 8.0 Hz, 1H), 7.56 (s, 1H), 7.51 (d, J = 1.5 Hz, 1H), 7.38 (dd, J = 8.0, 1.5 Hz, 1H), 6.73 (br, 2H), 4.32 (t, J = 7.0 Hz, 2H), 4.01(s, 3H), 1.49 (t, J = 7.0 Hz, 2H), 1.49 (t, J = 7.0 Hz, 2H), 1.40 (t, J =7.0 Hz, 3H); ${}^{13}C{}^{1}H{}$ NMR (125 MHz, CDCl₃) δ 167.3, 151.2, 143.8, 143.1, 139.9, 135.2, 128.8, 123.0, 122.2, 121.7, 121.6, 121.1, 119.8, 117.8, 111.8, 111.0, 98.5, 87.0, 52.2, 37.9, 13.2; IR (neat): v_{max} 3427, 3338, 2921, 2852, 2207, 1707, 1632, 1517, 1433, 1332, 1216, 1187, 1060, 827, 582 cm⁻¹; HRMS (ESI) calcd for $C_{21}H_{17}^{79}BrN_3O_2$ and $C_{21}H_{17}^{81}BrN_{3}O_{2}$ [M⁺+H] *m/z* 422.0504 and 424.04, found 422.0505 and 424.0470. Octyl 8-amino-3-bromo-7-cyano-5-ethyl-5H-benzo[b]carbazole-9-carboxylate (3id). This

compound was prepared by using precursors 1i (100 mg, 0.262 mmol) and 2d. Yellow solid, yield 0.075 g (55%); mp 229-231 °C; ¹H NMR (500 MHz, CDCl₃) δ 8.85 (s, 1H),

ACS Paragon Plus Environment

 8.45 (s, 1H), 7.96 (d, J = 8.0 Hz, 1H), 7.62 (s, 1H), 7.54 (d, J = 1.0 Hz, 1H), 7.40 (dd, J = 8.0, 1.5 Hz, 1H), 6.77 (s, 2H),4.41-4.35 (m, 4H), 1.89-1.83 (m, 2H), 1.52-1.49 (m, 4H), 1.44-1.27 (m, 9H), 0.93 (t, J = 7.0 Hz, 3H); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 167.0, 151.4, 143.9, 143.2, 139.9, 135.2, 123.0, 122.3, 121.7, 121.0, 119.9, 118.0, 111.8, 111.4, 98.6, 87.0, 65.5, 38.0, 31.8, 29.3, 29.2, 28.7, 26.1, 22.7, 14.1,13.3; IR (neat) v_{max} 3349, 2923, 2854, 2195, 1699, 1608, 1460, 1332, 1193, 1057, 586 cm⁻¹; HRMS (ESI) calcd for C₂₈H₃₁⁷⁹BrN₃O₂ and C₂₈H₃₁⁸¹BrN₃O₂ [M⁺+H] *m/z* 520.1600 and 522.1579, found 520.1600 and 522.1585.

(*iii*). General procedure for the synthesis of compounds 5aaa-5gaa and 7aa-7da. To an oven dried Schlenk tube with a magnetic stirrer bar was added *o*-bromo aldehyde 1 (~0.5 mmol, 1 equiv), K_2CO_3 (0.5 mmol, 1 equiv), alkyl cyanoacetate 2 (0.5 mmol, 1 equiv) and DMSO (2 mL). The contents were sealed and heated at rt for 1h. Then CuI (0.05 mmol, 10 mol%), K_2CO_3 (1.0 mmol, 2 equiv) and 4 or 6 (0.5 mmol, 1 equiv) was added and the reaction was performed at 120 °C for 2 h. After completion of the reaction as monitored by TLC, the crude reaction mixture was cooled to rt. The mixture was diluted with ethyl acetate (20 mL) and passed through celite. The resulting solution was washed with water and the aqueous part extracted twice with ethyl acetate (20 mL). The combined organic layer was washed with brine solution (2 x 20 mL), dried over anhydrous Na₂SO₄, and concentrated in vacuum. The residue was then purified by using silica gel column chromatography using hexane-ethyl acetate (9:1) as eluent to afford the pure compounds **5** and **7**. A similar molar stoichiometry/ratio was used in all the cases.

Diethyl 2-aminonaphthalene-1,3-dicarboxylate (5aaa). This compound was prepared by using precursors **1a** (100 mg, 0.540 mmol), **2a** and **4a**. Yellow oil, yield 0.125 g (80%); ¹H NMR (500 MHz, CDCl₃) δ 8.62 (s, 1H), 8.32 (d, J = 9.0 Hz, 1H), 7.77 (br s, 2H), 7.71 (d, J = 8.0 Hz, 1H), 7.53-7.50 (m, 1H), 7.23 (t, J = 7.0 Hz, 1H), 4.52 (q, J = 7.0 Hz, 2H), 4.43 (q, J = 7.0 Hz, 2H), 1.51-1.45 (m, 6H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 169.2, 167.4, 149.8, 139.0, 135.7, 130.3, 130.2, 125.1, 124.5, 122.4, 114.9, 105.6, 61.2, 60.8, 14.4, 14.3;IR (neat): v_{max} 3453, 3352, 2926, 1702, 1621, 1561, 1366, 1264, 1204, 1160, 1074, 1027, 804, 747 cm⁻¹; HRMS (ESI) calcd for C₁₆H₁₈NO₄ [M⁺+H] *m/z* 288.1236, found 288.1238.

3-Ethyl 1-methyl 2-aminonaphthalene-1,3-dicarboxylate (Saab). This compound was prepared by using precursors **1a** (100 mg, 0.540 mmol), **2a** and **4b**. Yellow oil, yield 0.117 g (79%); ¹H NMR (500 MHz, CDCl₃) δ 8.63 (s, 1H), 8.29 (d, J = 9.0 Hz, 1H), 7.84 (br s, 2H), 7.71 (d, J = 8.0Hz, 1H), 7.54-7.51 (m, 1H), 7.25 (t, J = 7.5 Hz, 1H), 4.44 (q, J = 7.0 Hz, 2H), 4.04 (s, 3H), 1.48 (t, J = 7.0 Hz, 3H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 169.7, 167.4, 150.1, 139.3, 135.7, 130.4, 130.3, 125.1, 124.5, 122.4, 114.8, 105.0, 61.2, 51.7, 14.3; IR (KBr) ν_{max} 3450, 3347, 2981, 2950, 2203, 1697, 1619, 1601, 1556, 1260, 1201, 1158, 1071, 1018, 802, 744 cm⁻¹; HRMS (ESI) calcd for C₁₅H₁₆NO₄ [M⁺+H] *m/z* 274.1079, found 274.1078.

3-Ethyl 1-isopropyl 2-aminonaphthalene-1,3-dicarboxylate (**5aac**). This compound was prepared by using precursors **1a** (100 mg, 0.540 mmol), **2a** and **4c**. Yellow oil, yield 0.121 g (74%); ¹H NMR (500 MHz, CDCl₃) δ 8.62 (s, 1H), 8.31 (dd, *J* = 8.5, 0.5 Hz, 1H), 7.73-7.71 (m, 1H), 7.68 (br s, 2H), 7.53-7.50 (m, 1H), 7.25-7.22 (m, 1H), 5.48-5.43 (m, 1H), 4.44 (q, *J* = 7.0 Hz, 2H), 1.50-1.46 (m, 9H); ¹³C{¹H} NMR (100 MHz, CDCl₃)

δ 168.8, 167.5, 149.4, 138.8, 135.6, 130.2, 125.1, 124.3, 122.4, 114.8, 106.1, 68.6, 61.2, 22.2, 14.3; IR (KBr) v_{max} 2981, 2936, 2203, 1728, 1703, 1622, 1561, 1381, 1204, 1145, 1098, 1018, 805, 750 cm⁻¹; HRMS (ESI) calcd for C₁₇H₁₉NO₄ [M⁺+H] *m/z* 302.1392, found 302.1393.

3-Methyl 1-methyl 2-aminonaphthalene-1,3-dicarboxylate (5abb): This compound was prepared by using precursors **1a** (100 mg, 0.540 mmol), **2b** and **4b**. Yellow oil, yield 0.096 g (68%); ¹H NMR (500 MHz, CDCl₃) δ 8.64 (s, 1H), 8.29 (d, *J* = 8.5 Hz, 1H), 7.83 (br s, 2H), 7.72-7.70 (m, 1H), 7.55-7.51 (m, 1H), 7.26-7.23 (m, 1H), 4.04 (s, 3H), 3.98 (s, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 169.7, 167.8, 150.0, 139.4, 135.7, 130.5, 130.3, 125.1, 124.5, 122.5, 114.5, 105.1, 52.2, 51.7; IR (KBr) v_{max} 3337, 2926, 2856, 2216, 1707, 1623, 1572, 1445, 1306, 1207, 1158, 1076, 948, 745 cm⁻¹; HRMS (ESI) calcd for C₁₄H₁₄NO₄ [M⁺+H] *m/z* 260.0923, found 260.0924.

Dimethyl 2-amino-6-methoxynaphthalene-1,3-dicarboxylate (5cbb). This compound was prepared by using precursors **1c** (100 mg, 0.465 mmol), **2b** and **4b**. Yellow oil, yield 0.118 g (87 %); ¹H NMR (500 MHz, CDCl₃) δ 8.57 (s, 1H), 8.22 (d, *J* = 9.5 Hz, 1H), 7.63 (br s, 2H), 7.22 (dd, *J* = 9.5, 2.5 Hz, 1H), 7.04 (d, *J* = 3.0 Hz, 1H), 4.03 (s, 3H), 3.97 (s, 3H), 3.90 (s, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 169.6, 167.9, 154.9, 148.5, 137.9, 131.1, 126.3, 126.0, 122.8, 114.9, 108.0, 105.8, 55.3, 52.2, 51.7; IR (KBr) ν_{max} 3391, 3337, 3181, 2918, 2848, 2360, 1692, 1645, 1608, 1560, 1467, 1428, 1259, 1191, 1038, 950, 824. cm⁻¹; HRMS (ESI) calcd for C₁₅H₁₅NO₅ [M⁺+H] *m/z* 290.1028, found 290.1027.

3-Ethyl 1-methyl 2-amino-6-fluoronaphthalene-1,3-dicarboxylate (5dab). This compound was prepared by using precursors 1d (100 mg, 0.493 mmol), 2b and 4b.

Yellow oil, yield 0.125 g (87%); ¹H NMR (500 MHz, CDCl₃) δ 8.56 (s, 1H), 8.30 (dd, J = 9.5, 5.0 Hz, 1H), 7.77 (br s, 2H), 7.36 (dd, J = 8.5, 2.5 Hz, 1H), 7.32-7.30 (m, 1H), 4.44 (q, J = 7.0 Hz, 2H), 4.03 (s, 3H), 1.48 (t, J = 7.0 Hz, 3H); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 169.4, 167.2, 158.1(J = 242.3 Hz), 149.5, 138.0 (J = 4.8 Hz), 132.5, 127.0 (J = 7.5 Hz), 125.5 (J = 8.3 Hz), 120.0 (J = 24.1 Hz), 116.1, 112.8 (J = 20.0 Hz), 105.4, 61.3, 51.7, 14.3; IR (KBr) v_{max} 3451, 3348, 2923, 2852, 1704, 1612, 1568, 1504, 1462, 1371, 1264, 1222, 1204, 1136, 853, 731; HRMS (ESI) calcd for C₁₅H₁₅FNO₄ [M⁺+H] *m/z* 292.0985, found 292.0982.

Diethyl 2-aminophenanthrene-1,3-dicarboxylate (5gaa). This compound was prepared by using precursors **1g** (100 mg, 0.425 mmol), **2a** and **4a**. Yellow oil, yield 0.115 g (80%);¹H NMR (500 MHz, CDCl₃) δ 8.57 (s, 1H), 8.11 (d, *J* = 8.5 Hz, 1H), 7.83 (d, *J* = 8.0 Hz, 1H), 7.60-7.58 (m, 2H), 7.53-7.46 (m, 2H), 7.03 (br s, 2H), 4.45 (q, *J* = 7.0 Hz, 2H), 4.37 (q, *J* = 7.0 Hz, 2H), 1.48 (t, *J* = 7.0 Hz, 3H), 1.18 (t, *J* = 7.5 Hz, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 171.5, 167.5, 147.6, 135.9, 134.7, 134.0, 128.3, 128.2, 127.8, 127.6, 127.0, 125.0, 124.3, 123.4, 113.3, 110.9, 61.5, 61.1, 14.4, 13.8; IR (neat) *v*_{max} 3646, 2919, 2850, 1738, 1608, 1367, 1260, 1216, 1023, 799, 750 cm⁻¹; HRMS (ESI) calcd for C₂₀H₂₀NO₄ [M⁺+H] *m/z* 338.1392, found 338.1394.

2-Aminonaphthalene-1,3-dicarbonitrile (7aa). This compound was prepared by using precursors **1a** (100 mg, 0.540 mmol), **2a** and **6**. Yellow solid, yield 0.073 g (70%); mp 140-142 °C; ¹H NMR (500 MHz, CDCl₃) δ 8.24 (s, 1H), 7.97 (dd, J = 8.5, 0.5 Hz, 1H), 7.79 (d, J = 8.0 Hz, 1H), 7.74-7.71 (m, 1H), 7.46-7.43 (m, 1H), 5.30 (br s, 2H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 148.8, 139.9, 135.2, 132.3, 129.3, 125.4, 125.2, 123.6, 115.6₀, 115.5₇, 99.8, 89.5; IR (KBr) v_{max} 3473, 3359, 3242, 2923, 2853, 2349, 2229, 2208, 1638,

1504, 1220, 770 cm⁻¹; HRMS (ESI) calcd for $C_{12}H_7N_3$ [M⁺] *m/z* 193.0640, found 193.0642.

2-Amino-6-methoxynaphthalene-1,3-dicarbonitrile (7ca). This compound was prepared by using precursors **1c** (100 mg, 0.465 mmol), **2a** and **6**. Yellow solid, yield 0.085 g (82%); mp 147-149 °C; ¹H NMR (500 MHz, CDCl₃) δ 8.50 (s, 1H), 7.69 (d, *J* = 9.0 Hz, 1H), 7.43-7.38 (m, 2H), 6.74 (br s, 2H), 3.84 (s, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 156.3, 149.2, 140.5, 130.8, 126.0, 124.9, 124.3, 116.5₅, 116.4₆, 108.6, 100.6, 87.7, 55.9; IR (neat) v_{max} 3450, 3320, 3240, 2923, 2850, 2221, 2217, 1680, 1612, 1243, 1093, 1020, 792 cm⁻¹ HRMS (ESI) calcd for C₁₃H₁₀N₃O [M⁺+H] *m/z* 224.0824, found 224.0826.

2-*Amino-6-fluoronaphthalene-1,3-dicarbonitrile (7da)*. This compound was prepared by using precursors **1d** (100 mg, 0.493 mmol), **2a** and **6**. Yellow solid, yield 0.070 g (67%); mp 165-169 °C; ¹HNMR (500 MHz, CDCl₃) δ 8.18 (s, 1H), 7.99-7.96 (m, 1H), 7.53-7.49 (m, 1H), 7.44 (dd, J = 8.5, 2.5 Hz, 1H), 5.28 (br s, 2H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 159.7 (J = 246.5 Hz), 148.3, 138.7 (J = 5.1 Hz), 132.0, 126.0 (J = 8.6 Hz), 125.9 (J = 9.3 Hz), 122.5, 122.3, 115.3 (J = 11.4 Hz), 112.7 (J = 21.6 Hz), 101.2, 89.9; IR (KBr) v_{max} 3472, 3348, 3247, 3050, 2211, 2207, 2027, 1983, 1601, 1224, 1181, 1147, 967, 816, 721 cm⁻¹; HRMS (ESI) calcd for C₁₂H₇FN₃ [M⁺+H] *m/z* 212.0624, found 212.0624.

(iv). Synthesis of compounds 9aa, 9da, and 11aa. 4-Oxo-3,4dihydrobenzo[g]quinazoline-10-carbonitrile (9aa). This compound was prepared by using precursors **3aa** (100 mg, 0.416 mmol) and **8**. White solid, yield 0.083 g (91%); mp 176-178 °C; ¹H NMR (500 MHz, CDCl₃) δ 12.52 (s, 1H), 9.13 (s, 1H), 8.40 (d, *J* = 8.5 Hz, 1H), 8.31(s, 1H), 8.20 (d, *J* = 8.5 Hz, 1H), 7.97-7.93 (m, 1H), 7.77-7.74 (m, 1H);

¹³C{¹H} NMR (125 MHz, CDCl₃) δ 160.7, 149.1, 148.7, 135.9, 133.8, 132.4, 131.2, 130.6, 127.9, 124.7, 121.9, 116.2, 105.2; IR (neat) v_{max} 3480, 2923, 2853, 2159, 2030, 1976, 1453, 1367, 1228, 1216, 798 cm⁻¹; HRMS (ESI) calcd for C₁₃H₈N₃O [M⁺+H] *m/z* 222.0667, found: 222.0663.

7-*Fluoro-4-oxo-3,4-dihydrobenzo[g]quinazoline-10-carbonitrile* (**9***da*). This compound was prepared by using precursors **3da** (100 mg, 0.418 mmol) and **8**. White solid, yield 0.077 g (87%); mp 185-187 °C; ¹H NMR (500 MHz, CDCl₃) δ 12.55 (s, 1H), 9.14-9.08 (m, 1H), 8.31-8.19 (m, 3H), 7.88-7.87 (m, 1H); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 161.0 (*J* = 246.2 Hz), 159.7,148.6, 148.5, 133.0 (*J* = 6.1 Hz), 132.9, 131.4 (*J* = 10.2 Hz), 127.9 (*J* = 9.0 Hz), 122.9 (*J* = 26.4 Hz), 122.7, 116.0, 113.9 (*J* = 21.5 Hz), 105.6; IR (neat) v_{max} 3390, 3057, 2920, 2222, 1701, 1609, 1482, 1302, 1231, 1166, 1102, 866, 799, 556 cm⁻¹; HRMS (ESI) calcd for C₁₃H₇FN₃O [M⁺+H] *m/z* 240.0568, found 240.0541.

4-Oxo-3-phenyl-3,4-dihydrobenzo[g] quinazoline-10-carbonitrile (**11aa**). This compound was prepared by using precursors **9aa** (100 mg, 0.452 mmol) and **10**. White solid, yield 0.096 g (72%); mp 146-148 °C; ¹H NMR (500 MHz, CDCl₃) δ 9.17 (s, 1H), 8.44 (d, J = 8.5 Hz, 1H), 8.31(s, 1H), 8.18 (d, J = 8.5 Hz, 1H), 7.92-7.89 (m, 1H),7.75-7.72 (m, 1H), 7.64-7.61(m, 2H), 7.58-7.55 (m, 1H) 7.50-7.48 (m, 2H); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 160.1, 148.2, 147.3, 137.0, 136.3, 134.0, 131.7, 131.2, 130.3, 129.9, 129.5, 127.9, 126.9, 125.6, 120.6, 115.5, 107.2; IR (neat): v_{max} 3066, 2916, 2214, 1691, 1606, 1581, 1259, 1017, 935, 796 cm⁻¹; HRMS (ESI) calcd for C₁₉H₁₂N₃O [M⁺+H] *m*/*z* 298.0980, found 298.0974.

2,7-*Dibromo-9-ethyl-9H-carbazole-3-carbaldehyde* (*1i*). White solid, yield 0.081 g (75%); mp 139-141 °C; ¹H NMR (400 MHz, CDCl₃) δ 10.47 (s, 1H), 8.69 (s, 1H), 7.99

(d, J = 8.4 Hz, 1H), 7.63 (s, 1H), 7.61 (d, J = 1.6 Hz, 1H), 7.46 (dd, J = 8.0, 1.6 Hz, 1H), 4.34 (q, J = 7.2 Hz, 2H), 1.50 (t, J = 7.6 Hz, 3H); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 191.5, 143.9, 141.7, 125.5, 124.4, 124.1, 122.7, 122.4, 122.2, 121.9, 121.0, 113.1, 112.5, 38.2, 13.7; IR (neat) v_{max} 3073, 2974, 2864, 2786, 1670, 1579, 1438, 1339, 1229, 1053, 840, 787 cm⁻¹; HRMS (ESI) calcd for C₁₅H₁₂Br₂NO [M + H]⁺ *m/z* 379.9286, found 379.9279.

(*E*)-*Ethyl* 3-(2-*bromophenyl*)-2-*cyanoacrylate* (*I*). This compound was prepared by using precursors **1a** (100 mg, 0.540 mmol) and **2a**. Colorless oil, yield 0.113 g (75%); ¹H NMR (500 MHz, CDCl₃) δ : 8.65 (s, 1H), 8.18 (dd, *J* = 7.5, 1.0 Hz, 1H), 7.71 (dd, *J* = 8.0, 1.0 Hz, 1H), 7.47 (t, *J* = 7.5 Hz, 1H), 7.41-7.38 (m, 1H), 4.42 (q, *J* = 7.0 Hz, 2H), 1.42 (t, *J* = 7.0 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃): 161.7, 153.7, 133.7, 133.6, 131.7, 130.1, 128.1, 126.5, 114.7, 106.4, 62.9, 14.1; IR (KBr): 2995, 2222, 1726, 1609, 1460, 1424, 1283, 1255, 1199, 1119, 1024, 756, 738 cm⁻¹; HRMS (ESI) calcd for C₁₂H₁₀⁷⁹BrNaO₂ and C₁₂H₁₀⁸¹BrNaO₂ [M⁺+Na] *m*/*z* 301.9793 and 303.9793, found 301.9795 and 301.9774.

(v). X-ray data and crystals structures of 3da, 3ga, 3hb, 3ia, 5abb, 7aa and 3ha'.

Compound 3da: C₁₄H₁₁FN₂O₂, M = 258.25, Monoclinic, Space group *P21/c*, a = 8.2304(11), b = 13.6434(17), c = 10.9178(12) Å, V = 1225.3(30) Å³, $\beta = 91.930(4)^{\circ}$, Z = 4, $\mu = 0.106$ mm⁻¹, data/restraints/parameters: 2145/0/181, R indices (I> 2 σ \(I)): R1 = 0.0363, *w*R2 (all data) = 0.1043. CCDC No: 1583383

Compound 3ga: C₁₈H₁₄N₂O₂, M = 290.31, Triclinic, Space group *P-1*, a = 7.300(2), b = 9.493(4), c = 10.119(5) Å, V = 707.32(50) Å³, $\alpha = 103.93(12)^{\circ}$, $\beta = 98.12(12)^{\circ}$, $\gamma = 10.119(5)$ Å, V = 707.32(50) Å³, $\alpha = 103.93(12)^{\circ}$, $\beta = 98.12(12)^{\circ}$, $\gamma = 10.119(5)$ Å, V = 707.32(50) Å³, $\alpha = 103.93(12)^{\circ}$, $\beta = 98.12(12)^{\circ}$, $\gamma = 10.119(5)$ Å, V = 707.32(50) Å³, $\alpha = 103.93(12)^{\circ}$, $\beta = 98.12(12)^{\circ}$, $\gamma = 10.119(5)$ Å, V = 707.32(50) Å³, $\alpha = 103.93(12)^{\circ}$, $\beta = 98.12(12)^{\circ}$, $\gamma = 10.119(5)$ Å

97.41(16)°, Z = 2, $\mu = 0.090$ mm⁻¹, data/restraints/parameters 2407/0/208, R indices (I> $2\sigma(I)$) R1 = 0.0448, wR2 (all data) = 0.1591. CCDC No: 1963460.

Compound 3ha': $C_{13}H_{11}N_{3}O_{2}$, M = 241.25, triclinic, Space group *P*-1, a = 7.6992(12), b = 8.7622(4), c = 9.7258(6) Å, V = 579.93(11) Å³, $\alpha = 66.365(5)^{\circ}$, $\beta = 74.766(11)^{\circ}$, $\gamma = 84.533(8)^{\circ}$, Z = 2, $\mu = 0.097$ mm⁻¹, data/restraints/parameters: 2033/0/165, R indices (I> 2σ \(I)): R1 = 0.0923, *w*R2 (all data) = 0.3195. CCDC No: 1963461.

Compound 3hb: $C_{12}H_9N_3O_2$, M = 906.87, Monoclinic, Space group *PC*, a = 10.1364 (8), b = 30.3806 (19), c = 7.3026 (5) Å, V = 2159.9 (3) Å³, $\beta = 106.248$ (7)°, Z = 8, $\mu = 0.099$ mm⁻¹, data/restraints/parameters: 7033/2/617, R indices (I> 2 σ \(I)): R1 = 0.0697, *w*R2 (all data) = 0.1606. CCDC No: 1963462.

Compound 3ia: $C_{22}H_{18}BrN_{3}O_{2}$, M = 436.30, Orthorhombic, Space group *Pbca*, a = 20.5404(6), b = 8.9927(3), c = 20.6948(6) Å, V = 3822.61(20) Å³, $\alpha = \beta = \gamma = 90^{\circ}$, Z = 8, $\mu = 2.174$ mm⁻¹, data/restraints/parameters 3379/0/255, R indices (I> 2σ \(I)) R1 = 0.0547, *w*R2 (all data) = 0.1170. CCDC No: 1963463.

Compound 5abb: $C_{14}H_{13}NO_4$, M = 259.25, Triclinic, Space group *P-1*, a = 7.7957(2), b = 8.5442(3), c = 10.0862(3) Å, V = 618.81(3) Å³, $a = 68.141(3)^{\circ}$, $\beta = 83.048(3)^{\circ}$, $\gamma = 88.471(3)^{\circ}$, Z = 2, $\mu = 0.102$ mm⁻¹, data/restraints/parameters 2177/0/182, R indices (I> 2σ \(I)) R1 = 0.0535, wR2 (all data) = 0.1784. CCDC No: 1963464.

Compound 7aa: C₁₂H₇N₃, M = 193.21, monoclinic, Space group $P2_1/c$, a = 3.8196(5), b = 16.069(2), c = 15.350(2) Å, V = 941.12(20) Å³, $\beta = 92.66(1)^{\circ}$, Z = 4, $\mu = 0.086$ mm⁻¹, data/restraints/parameters 1664/0/124, R indices (I> 2 σ \(I)) R1 = 0.0433, wR2 (all data) = 0.1072. CCDC No: 1963465.

ASSOCIATED CONTENT

Supporting Information

X-ray crystallographic data (cif files), ORTEPs of compounds **3da**, **3ga**, **3hb**, **3ha'**, **3ia**, **5abb** and **7aa** (Figures S1-S7; CCDC 1583383 and 1963460-1963465), and 1 H/ 13 C NMR spectra. This material is available free of charge via the Internet at <u>http://pubs.acs.org</u>

AUTHOR INFORMATION

Corresponding Author

*E-mail: kckssc@uohyd.ac.in; kckssc@yahoo.com.

ACKNOWLEDGMENTS

We thank the Department of Science & Technology (DST, New Delhi) for single crystal X-ray diffractometer and HRMS facility (PURSE, IRHPA and FIST grants). We also thank UGC for the UPE-II and NRC programs. KCK thanks SERB for a J. C. Bose fellowship (SR/S2/JCB-53/2010) for funding. RK thanks UGC-DSKPDF (No.F.4-2/2006 (BSR)/CH/18-19/0026) and AK thanks DST-INSPIRE (IF160070) for fellowship.

REFERENCES

(1) (a) Kong, W. -J.; Finger, L. H.; Oliveira, J. C. A.; Ackermann, L. Rhodaelecrocatalysis for Annulative C-H Activation: Polycyclic Aromatic Hydrocarbons through Versatile Double Electrocatalysis. *Angew. Chem. Int. Ed.* **2019**, *58*, 6342. (b) Weil, T.; Vosch, T.; Hofkens, J.; Peneva, K.; Müllen, K. The

Rylene Colorant Family-Tailored Nanoemitters for Photonics Research and Applications. *Angew. Chem. Int. Ed.* 2010, *49*, 9068. (c) Feng, X.; Pisula, W.;
Müllen, K. Large Polycyclic Aromatic Hydrocarbons: Synthesis and Discotic Organization. *Pure Appl. Chem.* 2009, *81*, 2203. (d) Anthony, J. E. The Larger Acenes: Versatile Organic Semiconductors. *Angew. Chem. Int. Ed.* 2008, *47*, 452. (e) Brasholz, M. "Super-Reducing" Photocatalysis: Consecutive Energy and Electron Transfers with Polycyclic Aromatic Hydrocarbons. *Angew. Chem. Int. Ed.* 2017, *56*, 10280. (f) Anthony, E. J. Functionalized Acenes and Heteroacenes for Organic Electronics. *Chem. Rev.* 2006, *106*, 5028.

- (2) (a) Abdissa, N.; Pan, F.; Gruhonjic, A.; Grafenstein, J.; Fitzpatrick, P. A.; Landberg, G.; Rissanen, K.; Yenesew, A.; Erdelyi, M. Naphthalene Derivatives from the Roots of *Pentas parvifolia* and *Pentasbussei. J. Nat. Prod.* 2016, *79*, 2181.
 (b) Yang, X.; Shi, Q.; Liu, Y.-N.; Zhao, G.; Bastow, K. F.; Lin, J.-C.; Yang, S.-C.; Yang, P.-C.; Lee, K.-H. Antitumor Agents 268. Design, Synthesis, and Mechanistic Studies of New 9-Substituted Phenanthrene-Based Tylophorine Analogues as Potent Cytotoxic Agents. *J. Med. Chem.* 2009, *52*, 5262. (c) Bao, L.; Liu, W.; Li, Y.; Wang, X.; Xu, F.; Yang, Z.; Yue, Y.; Zuo, C.; Zhang, Q.; Wang, W. Carcinogenic Metabolic Activation Process of Naphthalene by the Cytochrome P450 Enzyme 1B1: A Computational Study. *Chem. Res. Toxicol.* 2019, *32*, 603. (d) Makar, S.; Saha, T.; Singh, S. K. Naphthalene, a Versatile Platform in Medicinal Chemistry: Sky-High Perspective. *Eur. J. Med. Chem.* 2019, *161*, 252.
- (3) (a) Georghiou, P. E.; Li, Z.; Ashram, M.; Chowdhury, S.; Mizyed, S.; Tran, A. H.Calixnaphthalenes: Deep, Electron-Rich Naphthalene Ring Containing Calixarenes.
 - ACS Paragon Plus Environment

The First Decade. *Synlett* **2005**, *6*, 879. (b) Watson, M. D.; Fechtenkötter, A.; Müllen, K. Big is Beautiful- "Aromaticity" Revised from the Viewpoint of Macromolecular and Supramolecular Benzene Chemistry. *Chem. Rev.* **2001**, *101*, 1267. (c) Kumobayashi, H.; Miura, T.; Sayo, N.; Saito, T.; Zhang, X. Recent Advances of BINAP Chemistry in the Industrial Aspects. *Synlett* **2001**, *37*, 1055. (d) Bringmann, G.; Price Mortimer, A. J.; Keller, P. A.; Gresser, M. J.; Garner, J.; Breuning, M. Atroposelective synthesis of axially chiral biaryl compounds. *Angew. Chem. Int. Ed.* **2005**, *44*, 5384. (e) Kobaisi, M. A.; Bhosale, S. V.; Latham, K.; Raynor, A. M.; Bhosale, S. V. Functional Naphthalene Diimides: Synthesis, Properties, and Applications. *Chem. Rev.* **2016**, *116*, 11685. (f) Liu, R.; Farinha, J. P. S.; Winnik, M. A. Preparation and Spectroscopic Properties of Phenanthrene-Labeled SEBS Triblock Copolymers. *Macromolecules* **1999**, *32*, 3957.

- (4) (a) Cochran, J. E.; Padwa, A. Tandem Pummerer Diels-Alder Sequence for the Preparation of α-Thio Substituted Naphthalene Derivatives. *Tetrahedron Lett.* 1995, 36, 3495. (b) Ma, Y.; Lv, J.; Liu, C.; Yao, X.; Yan, G.; Yu, W.; Ye, J. Electrochemical [4+2] Annulation-Rearrangement-Aromatization of Styrenes: Synthesis of Naphthalene Derivatives. *Angew. Chem. Int. Ed.* 2019, 58, 6756. (c) Kocsis, L. S.; Brummond, K. M. Intramolecular Dehydro-Diels-Alder Reaction Affords Selective Entry to Arylnaphthalene or Aryldihydronapthalene Lignans. *Org. Lett.* 2014, 16, 4158.
- (5) (a) Mal, D.; Pahari, P. Recent Advances in the Hauser Annulation. *Chem. Rev.*2007, 107, 1892. (b) Shu, W. M.; Liu, S.; He, J. -X.; Wang, S.; Wu, A. -X. Sequential σ-Bond Insertion/ Benzannulation Involving Arynes: Selective Synthesis

of Polysubstituted Naphthalenes. J. Org. Chem. 2018, 83, 9156. (c) Ponra, S.; Vitale, M. R.; Michelet, V.; Ratovelomanana-Vidal, V. HNTf₂-Catalyzed Regioselective Preparation of Polysubstituted Naphthalene Derivatives through Alkyne-Aldehyde Coupling. J. Org. Chem. 2015, 80, 3250. (d) Chen, L. L.; Zhang, J. -W.; Yang, W. -W.; Fu, J. Y.; Zhu, J. Y.; Wang, Y. B. Synthesis of 1-Cyano-3-Benzannulation acylnaphthalenes via Formal [4+2]of 2-(2 -Alkylphenyl)acetonitriles and Akynones. J. Org. Chem. 2019, 84, 8090. (e) Nagataa, T.; Satoh, T.; Nishii, Y.; Miura, M. Rhodium-Catalyzed Oxidative Annulation of (2-Arylphenyl)boronic Acids with Alkynes: Selective Synthesis of Phenanthrene Derivatives. Synlett. 2016, 27, 1707. (f) Nagata, T.; Hirano, K.; Satoh, T.; Miura, M. Iridium-Catalyzed Annulative Coupling of 2-Arylbenzoyl Chlorides with Alkynes; Selective Formation of Phenanthrene Derivatives. J. Org. Chem. 2014, 79, 8960. (g) Matsumoto, A.; Ilies, L.; Nakamura, Eiichi. Phenanthrene Synthesis by Iron-Catalyzed [4+2] Benzannulation between Alkyne and Biaryl or 2-Alkenylphenyl Grignard Reagent. J. Am. Chem. Soc. 2011, 133, 6557. (h) Wu, Y.; Wu, F.; Zhu, D.; Luo, B.; Wang, H.; Hu, Y.; Wen, S.; Huang, P. Pd catalyzed Insertion of Alkynes into Cyclic Diaryliodoniums: A Direct Access to Multi-substituted Phenanthrenes. Org. Biomol. Chem. 2015, 13, 10386. (i) Koppanathi, N.; Swamy, K. C. K. Regioselective Carboannulation of Electron deficient Allenes with Dialkyl (2-formylphenyl) Malonates Leading to Multisubstituted Naphalenes. Org. Biomol. Chem. 2016, 14, 5079.

(6) (a) Barluenga, J.; Andina, F.; Aznar, F.; Valdés, C. New Cascade Processes onGroup 6 Fischer-Type Carbene Complexes: Cyclopropanation and Metathesis

Reactions. Org. Lett. 2007, 9, 4143. (b) McAtee, C. C.; Riehl, P. S.; Schindler, C. S.
Polycyclic Aromatic Hydrocarbons via Iron(III)-Catalyzed Carbonyl-Olefin
Metathesis. J. Am. Chem. Soc. 2017, 139, 2960. (c) Bera, K.; Srakar, S.; Jalal, S.;
Jana, U. Synthesis of Substituted Phenanthrene by Iron(III)-Catalyzed
Intramolecular Alkyne-Carbonyl Metathesis. J. Org. Chem. 2012, 77, 8780.

- (7) (a) Hamura, T.; Suzuki, T.; Matsumoto, T.; Suzuki, K. Tandem Ring Expansion of Alkenyl Benzocyclobutenol Derivatives into Substituted Naphthols. *Angew. Chem. Int. Ed.* 2006, *45*, 6294. (b) Glass, A. C.; Morris, B. B.; Zakharov, L. N.; Liu, S. -Y. Synthesis of Substituted Naphthalenes via a Catalytic Ring- Expansion Rearrangement. *Org. Lett.* 2008, *10*, 4855.
- (a) Wang, Z. -Q.; Liang, Y.; Lei, Y.; Zhou, M. -B.; Li, J. -H. Iron-Catalyzed Annulations of 2-(2-alkynyl)phenoxy)-1-arylethanones Leading to Substituted Naphthalen-1-ols. *Chem. Commun.* 2009, 5242. (b) Jagdale, A. R.; Park, J. H.; Youn, S. W. Cyclization Reaction for the Synthesis of Polysubstituted Naphthalenes in the Presence of Au(I) Precatalysts. *J. Org. Chem.* 2011, *76*, 7204. (c) Kim, H. Y.; Oh, K. A Facile Access to 4-Substituted-2-naphthols via a Tandem Friedel-Crafts Reaction: A β-Chlorovinyl Ketone Pathway. *Org. Lett.* 2014, *16*, 5934. (d) Balamurugan, R.; Gudla, V. Gold-Catalyzed Electrophilic Addition to Arylalkynes. A Facile Method for the Regioselective Synthesis of Substituted Naphthalenes. *Org. Lett.* 2009, *11*, 3116. (e) Manojveer, S.; Balamurugan, R. Synthesis of Naphthalene Derivatives from ortho-Alkynylacetophenone derivatives via Tandem in situ Incorporation of Acetal and Intramolecular Heteoalkyne Metathesis/Annulation. *Org. Lett.* 2014, *16*, 1712.

- (9) (a) Wang, Z.; Xu, H. Rhodium-Catalyzed C-H activation/cyclization of Enaminones with Sulfoxonium Yields toward Polysubstituted Naphthalenes. *Tetrahedron. Lett.* 2019, *60*, 664. (b) Chen, H.; Ouyang, L.; Liu, J.; Shi, J. W.; Chen, G.; Zheng, L. Synthesis of Multisubstituted 1-Naphthic Acids via Ru-Catalyzed C-H Activation and Double-Alkyne Annulation Under Air. *J. Org. Chem.* 2019. 0000 (DOI: 10.1021/acs.joc.9b00926). (c) Hu, P.; Zhang, Y.; Xu, Y.; Yang, S.; Liu, B.; Li, X. Construction of (Dihydro)naphtha[1,8-*bc*]pyrnas via Rh(III)-Catalyzed Two fold C-H Activation of Benzoylacetonitriles. *Org. Lett.* 2018, *20*, 2160. (d) Zhou, S.; Wang, J.; Wang, L.; Song, C.; Chen, K.; Zhu, J. Enaminones as Synthons for a Directed C- H Functionalization Rh^{III}-Catalyzed Synthesis of Naphthalenes. *Angew. Chem. Int. Ed.* 2016, *55*, 9384. (e) Tan, X.; Liu, B.; Li, X.; Li, B.; Xu, S.; Song, H.; Wang, B. Rhodium-Catalyzed Cascade Oxidative Annulation Leading to Substituted Naphtho[1,8-*bc*]pyrnas by Sequential Cleavage of C(sp²)-H/C (sp³)-H and C(sp²)-H/ O-H Bonds. *J. Am. Chem. Soc.* 2012, *134*, 16163.
- (10) (a) Iwasaki, M.; Araki, Y.; Nishihara, Y. Phenanthrene Synthesis by Palladium-Catalyzed Benzannulation with *o*-Bromobenzyl Alcohols through Multiple Carbon-Carbon Bond Formations. *J. Org. Chem.* 2017, *82*, 6242. (b) Komeyama, K.; Igawa, R.; Takaki, K. Cationic Iron-Catalyzed Intramolecular Alkyne-Hydroarylation with Electron-Deficient Arenes. *Chem. Commun.* 2010, *46*, 1748.
 (c) Fürstner, A.; Mamne, V. Flexible Synthesis of Phenanthrenes by a PtCl₂-Catalyzed Cycloisomerization Reaction. *J. Org. Chem.* 2002, *67*, 6264. (d) Matsuda, T.; Kato, K.; Goya, T.; Shimada, S.; Murakami, M. Ruthenium-Catalyzed Cycloisomerization of 2, 2'- Diethynylbiphenyls Involving Cleavage of a Carbon-

carbon Triple Bond. *Chem. Eur. J.* **2016**, *22*, 1941. (e) Kwon, Y.; Cho, H.; Kim, S. Expedient Synthesis of Phenanthrenes via In(III)-Catalyzed 6-Exo- Dig Cycloisomerization. *Org. Lett.* **2013**, *15*, 920.

- (11) (a) Kim, Y. H.; Lee, H.; Kim, Y. J.; Kim, B. T.; Heo, J. -N. Direct One-Pot Synthesis of Phenanthrenes via Suzuki-Miyaura Coupling/Aldol Condensation Cascade Reaction. J. Org. Chem. 2008, 73, 495. (b) Shimizu, M.; Nagao, I.; Tomioka, Y.; Hiyama, Τ. Palladium-Catalyzed Annulation of vic-Bis(pinacolatoboryl)alkenes and Phenathrenes with 2,2'-Dibromobiaryls: Facile Synthesis of Functionalized Phenathrenes and Dibenzo[g, p]chrysenes. Angew. Chem. Int. Ed. 2008, 47, 8096. (c) Shimizu, M.; Hiyama, T. Palladium-Catalyzed Double Cross-Coupling Reactions of Organodimetallic Reagents Leading to Polycyclic Aromatic Hydrocarbons. Eur. J. Org. Chem. 2013, 8069.
- (12) (a) Adachi, Y.; Matsuki, M.; Watanabe, H.; Takase, K.; Kodama, K.; Matsui, J.; Funahashi, Y.; Nomoto, K. Antitumor and Antiangiogenic Activities of Lenvatinib in Mouse Xenograft Models of Vascular Endothelial Growth Factor-Induced Hypervascular Human Hepatocellular Carcinoma. *Cancer Investigation* 2019, *37*, 185. (b) Asche, C.; Frank, W.; Albert, A.; Kucklaender, U. Synthesis, Antitumour Activity and Structure-Activity Relationships of 5*H*-benzo[*b*]carbazoles. *Bioorg. Med. Chem.* 2005, *13*, 819.
- (13) (a) Upadhyay, A.; Chandrakar, P.; Gupta, S.; Parmar, N.; Singh, S. K.; Rashid, M.;
 Kushwaha, P.; Wahajuddin, M.; Sashidhara, K. V.; Kar, S. Synthesis, Biological Evaluation, Structure-Activity Relationship, and Mechanism of Action Studies of Quinoline-Metronidazole Derivatives Against Experimental Visceral

Leishmaniasis, J. Med. Chem. 2019, 62, 5655. (b) Chakraborty, G.; Sikari, R.; Das, S.; Mondal, R. Sinha, S.; Banerjee, S.; Paul, N. D. Dehydrogenative Synthesis of Quinolines. 2-Aminoquinolines, and Quinazolines Using Singlet Diradical Ni(II)-Catalysts. J. Org. Chem. 2019, 84, 2626. (c) Qiu, Y. -F.; Niu, Y.-J.; Wei, X.; Cao, Z. -J. AgSCF₃/Na₂S₂O₈-Promoted B. -0.: Wang, X. -C.; Ouan, Trifluoromethylthiolation/Cyclization of o-Propargyl Arylazides/o-AlkynylBenzylazides: Synthesis of SCF₃-Substituted Quinolines and Isoquinolines. J. Org. Chem. 2019, 84, 4165. (d) Zhang, X.; Ma, X.; Qiu, W.; Evans, J.; Zhang, W. Cascade Knoevenagel and aza-Wittig Reactions for the Synthesis of Substituted Quinolines and Quinolin-4-ols. Green Chem. 2019, 21, 349. (e) Xu, W.; Wang, G.; Xie, X.; Liu, Y. Gold(I)-Catalyzed Formal Intramolecular Dehydro-Diels-Alder Reaction of Ynamide-ynes: Synthesis of Functionalized Benzo[b]carbazoles. Org. Lett. 2018, 20, 3273. (f) Li, D. -Y.; Wang, A.; Zhu, X-P.; Feng, W.; Liu, P. -N. Direct Access to Substituted Benzo[b]carbazoles through Cascade Annulation of 2-Vinylbenzaldehydes with Indoles. Chem. Commun. 2019, 55, 3339. (g) Boominathan, S. S. K.; Senadi, G. C.; Vandavasi, J. K.; Chen, J. Y. -Fu.; Wang, J. -J. An Iron-Catalyzed Cascade Approach to Benzo[b]carbazole Synthesis Followed by 1,4-Sulfonyl Migration. Chem. Eur. J. 2015, 21, 3193. (h) Wei, D.; Dorcet, V.; Darcel, C.; Sortais, J.-B. Synthesis of Quinolines Through Acceptorless Dehydrogenative Coupling Catalyzed by Rhenium PN(H)P Complexes. ChemSusChem. 2019, 12, 3078. (i) Parua, S.; Sikari, R.; Sinha, S.; Das, S.; Chakraborty, G.; Paul, N. D. A nickel catalyzed acceptorless dehydrogenative approach to quinolines. Org. Biomol. Chem. 2018, 16, 274.

2	
3	
4	
5	
7	
8	
9	
10	
11	
13	
14	
15	
16	
17	
19	
20	
21	
22	
23 24	
25	
26	
27	
28	
29 30	
31	
32	
33	
34 35	
36	
37	
38	
39 40	
40	
42	
43	
44	
45 46	
47	
48	
49	
50	
51 52	
52	
54	
55	
56	
5/ 50	
59	
60	

(14)	(a) Jiang, H.; Yang, J.; Tang, X.; Li, J.; Wu, W. Cu-Catalyzed Three-Component
	Cascade Annulation Reaction: An Entry to Functionalized Pyridines. J. Org. Chem.
	2015, 80, 8763. (b) Tang, J.; Xu, B.; Mao, X.; Yang, H.; Wang, X.; Lv, X. One-Pot
	Synthesis of Pyrrolo[3,2,1-kl]phenothiazines through Copper-Catalyzed Tandem
	Coupling/Double Cyclization Reaction. J. Org. Chem. 2015, 80, 11108. (c) Ma, Y.
	-G.; Huang, MQ.; Liu, Z.; Liu, JQ.; Wang, XS. Synthesis of Substituted 4H-
	Thiochromen-4-imines via Copper-Catalyzed Cyclization Cascades of o-
	Bromobenzothioamides with Terminal Alkynes. J. Org. Chem. 2018, 83, 9504. (d)
	Villuri, B. K.; Ichake, S.S.; Reddy, S.R.; Kavala, V.; Bandi, V.; Kuo, CW.; Yao,
	CF. Copper-Catalyzed Cascade Synthesis of 2-Aryl-3-cyanobenzofuran and
	Dibenzo[b,f] oxepine-10-carbonitrile Derivatives. J. Org. Chem. 2018, 83, 10241.
	(e) Liu,Y.; Wan, JP.; Tandem Reactions Initiated by Copper-catalyzed Cross-
	coupling: A New Strategy towards Heterocycle Synthesis. Org. Biomol. Chem.
	2011, 9, 6873. (f) Lu, J.; Fu, H. Copper-Catalyzed Cascade Synthesis of Alkyl-6-
	Aminobenzimidazo[2,1-a]isoquinoline-5-carboxylates. J. Org. Chem. 2011, 76,
	4600. (g) Cai, Q.; Li, Z.; Wei, J.; Fu, L.; Ha, C.; Pei, D.; Ding, K. Synthesis of Aza-
	Fused Polycyclic Quinolines through Copper-Catalyzed Cascade Reactions. Org.
	Lett. 2010, 12, 1500. (h) Jiang. H.; Yang. J.; Tang. X.; Wu. W. Divergent Syntheses
	of Isoquinolines and Indolo[1,2-a]quinazolines by Copper-Catalyzed Cascade
	Annulation from 2-Haloaryloxime Acetates with Active Methylene Compounds

and Indoles. J. Org. Chem. 2016, 81, 2053.

(15) (a) Panda, N.; Mothkuri, R.; Pal, A.; Patal, A. R. Copper-Catalyzed Synthesis of α-naphthols from Enol Esters. *Adv. Synth. Catal.* 2013, 355, 2809. (b) Xu, H.; Li, S.;

Liu, H.; Fu, H.; Jiang, Y. Simple and Efficient Copper-Catalyzed Cascade Synthesis of Naphthols Containing Multifunctional Groups under Mild Conditions. *Chem. Commun.* **2010**, *46*, 7617. (c) Xiong, M.; Hu, H.; Hu, X.; Liu, Y. Copper-Catalyzed Borylative Cyclization of *o*-(Cyano)phenylPropargyl Carbonates: Synthesis of Functionalized 1-Naphthylamines. *Org. Lett.* **2018**, *20*, 3661. (d) Malakar, C. C.; Schmidt, D.; Conard, J.; Beifuss, U. Cu(I)-Catalyzed Domino Reactions: Efficient and Selective Synthesis of 4*H*-Chromenes and Naphthalenes. *Org. Lett.* **2011**, *13*, 1972. (e) Malakar, C. C.; Sudheendran, K.; Imrich, H. G.; Mika, S.; Beifuss, U. Cu(I)-Catalyzed Annulation for the Synthesis of Substituted Naphthalanes using *o*-Bromobenzaldehydes and β -Ketoesters as Substrates. *Org. Biomol. Chem.* **2012**, *10*, 3899.

- (16) (a) Colin-Molina, A.; Jellen, M. J.; Garcia-Quezada, E.; Cifuentes-Quintal, M. E.; Murillo, F.; Barroso, J.; Pérez-Estrada, S. P.; Toscano, R. A.; Merino, Gabriel.; Rodriguez-Molina, B. Origin of the Isotropic Motion in Crystalline Molecular Rotors with Carbazole Stators. *Chem. Sci.* 2019, *10*, 4422. (b) Chaitanya, K. T.; Nagarajan, R. Synthesis of Functionalized Ellipticinium and Ellipticine Derivatives via Electrophilic Cyclization. *Org. Biomol. Chem.* 2011, *9*, 4662.
- (17) (a) Lu, J.; Gong, X.; Yang, H.; Fu, H. Concise Copper-catalyzed One-pot Tandem Synthesis of Benzimidazo[1,2-*b*]isoquinolin-11-one Derivatives. *Chem. Commun.* **2010**, *46*, 4172. (b) Kavala, V.; Cheng, -C. W.; Barange, D. K.; Kuo, C.-W.; Lei, P. -M. Synthesis of Isocoumarin Derivatives via the Copper-Catalyzed Tandem Sequential Cyclization of 2-Halo-*N*-phenylbenzamides and Acyclic1,3-Diketones. *J. Org. Chem.* **2012**, *77*, 5022.

- (18) Adepu, R.; Kumar, S. K.; Sandra, S.; Rambabu, D.; Krishna, R. G.; Reddy, M. C. Kandale, A.; Misra, P.; Pal, M. C-N Bond Formation under Cu-Catalysis Synthesis and in vitro Evaluation of *N*-aryl Substituted Thieno[2,3-*d*]pyrimidin-4(3*H*)-ones against Chorismate Mutase. *Bioorg. Med. Chem.* 2012, 20, 5127.
 - (19) (a) Sheldrick, G. M. SADABS, Siemens Area Detector Absorption Correction;
 University of Gottingen, Germany, 1996. (b) Sheldrick, G. M. SHELX-97- A program for crystal structure solution and refinement; University of Gottingen, 1997. (c) Sheldrick, G. M. SHELXTL NT Crystal Structure Analysis Package;
 Bruker AXS, Analytical X-ray System, WI, 1999, version 5.10.