This article was downloaded by: [Duke University Libraries] On: 01 June 2012, At: 01:57 Publisher: Taylor & Francis Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Synthetic Communications: An International Journal for Rapid Communication of Synthetic Organic Chemistry

Publication details, including instructions for authors and subscription information: http://www.tandfonline.com/loi/lsyc20

Efficient One-Pot Condensation of β-Naphthol, Aldehydes, and Cyclic 1,3-Dicarbonyl Compounds Catalyzed by p-TSA Under Solvent-Free and Sonication Conditions

Jianjun Li^a, Jia Li^a, Jin Fang^a & Weike Su^a

^a Key Laboratory of Pharmaceutical Engineering of Ministry of Education, College of Pharmaceutical Sciences, Zhejiang University of Technology, Hangzhou, China

Available online: 04 Mar 2010

To cite this article: Jianjun Li, Jia Li, Jin Fang & Weike Su (2010): Efficient One-Pot Condensation of β-Naphthol, Aldehydes, and Cyclic 1,3-Dicarbonyl Compounds Catalyzed by p-TSA Under Solvent-Free and Sonication Conditions, Synthetic Communications: An International Journal for Rapid Communication of Synthetic Organic Chemistry, 40:7, 1029-1039

To link to this article: http://dx.doi.org/10.1080/00397910903029966

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <u>http://www.tandfonline.com/page/terms-and-conditions</u>

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae, and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings,

demand, or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.



Synthetic Communications[®], 40: 1029–1039, 2010 Copyright © Taylor & Francis Group, LLC ISSN: 0039-7911 print/1532-2432 online DOI: 10.1080/00397910903029966

EFFICIENT ONE-POT CONDENSATION OF β -NAPHTHOL, ALDEHYDES, AND CYCLIC 1,3-DICARBONYL COMPOUNDS CATALYZED BY p-TSA UNDER SOLVENT-FREE AND SONICATION CONDITIONS

Jianjun Li, Jia Li, Jin Fang, and Weike Su

Key Laboratory of Pharmaceutical Engineering of Ministry of Education, College of Pharmaceutical Sciences, Zhejiang University of Technology, Hangzhou, China

A facile and efficient procedure has been developed by one-pot condensation of β-naphthol, aldehydes, and cyclic 1,3-dicarbonyl compounds for the synthesis of 8,9,10,12tetrahydrobenzo[a]xanthen-11-one or 8,9-dihydrobenzo-[f]cyclopenta[b]chromen-10(11H)one derivatives catalyzed by p-toluenesulfonic acid under solvent-free and sonication conditions.

Keywords: One-pot condensation; p-TSA; solvent-free; ultrasound irradiation

Xanthenes and benzoxanthenes are of considerable interest in industry as well as in academia because of their promising biological and pharmaceutical activities such as antiviral,^[1] antibacterial,^[2] and anti-inflammatory actions,^[3] and they are being used as zoxazolamine^[4,5] and in photodynamic therapy.^[6,7] Furthermore, these active oxygen-containing heterocyclic compounds can also be employed as dyes,^[8,9] in laser technology,^[10,11] as pH-sensitive fluorescent materials for the visualization of biomolecules.^[12] Thus, the synthesis of this heterocyclic nucleus is of great significance. Lately, only a few procedures for the preparation of 8,9,10,12-tetrahydrobenzo[a]-xanthen-11-one or 8,9-dihydrobenzo-[f]cyclopenta[b]chromen-10(11*H*)-one derivatives have been reported. Catalysts such as NaHSO₄ · SiO₂^[13] and Sr(OTf)₂ have been used for these reactions,^[14] but these methods require long reaction times.

Recently, the utilization of multicomponent reactions (MCRs) to generate novel, drug-like scaffolds has received great attention from organic chemists because products can be prepared directly in a single step and diversity can be achieved simply by varying the reaction substrates.^[15–19] Previously, it occured to us that dihydropyrimidiones,^[20] 1-substituted-1*H*-1,2,3,4-tetrazoles,^[21] 2-hydroxy-7,8-dihydroquinolin-5(6*H*)-ones, and 7,8-dihydroquinolin-2,5-(1*H*,6*H*)-diones^[22] could

Received March 13, 2009.

Address correspondence to Weike Su, Key Laboratory of Pharmaceutical Engineering of Ministry of Education, College of Pharmaceutical Sciences, Zhejiang University of Technology, Hangzhou 310014, China. E-mail: pharmlab@zjut.edu.cn

successfully be prepared by MCRs. More recently, we also reported the reaction of the β -naphthol, aldehydes, and amides to form amidoalkyl naphthols.^[23]

p-Toluenesulfonic acid (*p*-TSA) is a commercially available and very cheap chemical with good stability. Recently, it has been shown that *p*-TSA can be used as a substitute for conventional acidic catalytic materials. The toxicity and volatile nature of many organic solvents, particularly chlorinated hydrocarbons, that are widely used in huge amounts for organic reactions pose a serious threat to the environment. Thus, design of solventless catalytic reactions has received tremendous attention recently.

Taking these points into consideration, in continuation of our investigations on these active oxygen-containing heterocyclic compounds,^[24,14] herein we report a new route to 8,9,10,12-tetrahydrobenzo[a]xanthen-11-one or 8,9-dihydrobenzo-[f]cyclopenta[b]chromen-10(11*H*)-one derivatives with *p*-TSA as catalyst under solvent-free conditions. In addition, all the reactions were carried out under ultrasonic irradiation because it could accelerate the progress of various organic transformations with mild reaction conditions, short reaction times, and good yields.

We initially studied the reaction of benzaldehyde 1a, β -naphthol 2, and 5,5-dimethyl-cyclohexane-1,3-dione 3 along with a catalytic amount of *p*-TSA (10 mol%) in several classic solvents chosen as the medium for comparison. The results are summarized in Table 1. 1,2-Dichloroethane (Table 1, entry 5, 93%) and solvent-free conditions (Table 1, entry 6, 95%) afforded the product 9,9-dimethyl-12-phenyl-9,10-dihydro-8*H*-benzo[a]xanthen-11(12*H*)-one 4a in excellent yields. Table 1 (entry 7) shows that the reaction time is reduced and yields improve in the absence of

PhCHO +	ОН	+ 0 0	<i>p</i> -TSA Solvent-free, 70°C,))))	
1a	2	3		4a
Entry	Solvent		Time (h)	Yields $(\%)^b$
1	Water		3	30
2	THF		3	45
3	CH ₃ CH ₂ OH		3	<10
4	[bpy]BF ₄		3	60
5	(CH ₂ Cl) ₂		1.5	93
6	None		0.5	95
7	None		1.0	90^c

Table 1. Optimization of conditions for the condensation reaction of aldehyde, β -naphthol, and dimedone^{*a*}

 \sim

^{*a*}Reaction conditions: benzaldehyde **1a** (1 mmol), β-naphthol **2** (1 mmol), 5,5-dimethyl-cyclohexane-1,3-dione **3** (1.2 mmol), *p*-TSA (0.1 mmol), solvent (3 mL), 70°C.

^bIsolated yield.

^cThe reaction was carried out by stirring instead of sonic irradiation.

ultrasound irradiation conditions. Taking the reaction yields and environmental impact into consideration, we carried out this reaction under ultrasonic irradiation and solvent-free conditions at 70 °C.

Encouraged by this success, we extended this reaction to a range of aldehydes in the presence of 10 mol% p-TSA under solvent-free and ultrasonic irradiation conditions (Table 2). Varoius aromatic aldehydes with electron-donating and electron-withdrawing substituents reacted successfully, and the corresponding products **4a**-**k** were obtained in good to excellent yields in 0.5–3 h. These results were also obtained in the case of the aliphatic and heterocyclic aldehydes (Table 2, entries 12–14). The desired products were characterized by NMR, infrared (IR), and mass spectroscopy (MS) and also by comparison with authentic samples.

We have not established an exact mechanism for the formation of this kind of compound; however, a reasonable pathway is shown in Scheme 1. The products may be formed either through step $1 \rightarrow$ step 2 or through step $3 \rightarrow$ step 4 by referring to literature.^[13] The role of *p*-TSA comes in step 1, when it catalyzes the Knoevenagel-type coupling of aldehyde 1 with active methylene compound 3, and in step 2, when it catalyzes the Michael-type addition of β -naphthol 2 and hetero-diene 7 to afford the corresponding products 4.

Table 2. *p*-TSA-promoted one-pot condensation of β -naphthol, aldehydes, and 5,5-dimethyl-cyclohexane-1,3-dione^{*a*}

RCHO	+ 0 + 0 + 0		<i>p</i> -TSA Solvent-free, 70°C,			
1	2	3			4	
Entry	1	R	Time (h)	Product	Yields (%) ^b	
1	1a	C_6H_5	0.5	4 a	95	
2	1b	$3-F-C_6H_4$	0.5	4b	96	
3	1c	3-Cl-C ₆ H ₄	0.5	4c	95	
4	1d	$3-OH-C_6H_4$	0.5	4d	97	
5	1e	$3-NO_2-C_6H_4$	0.5	4 e	94	
6	1f	3-OCH ₃ -C ₆ H ₄	0.5	4 f	91	
7	1g	3-OMe-4-OH-C ₆ H ₃	2.5	4g	90	
8	1h	3-Me-4-Me-C ₆ H ₃	1.0	4h	92	
9	1I	4-OMe-C ₆ H ₄	0.5	4i	94	
10	1j	$4-Cl-C_6H_4$	0.5	4j	94	
11	1k	CH ₃ CH ₂	2.5	4k	87	
12	11	$CH_3C(CH_3)_2$	3.0	41	86	
13	1m	2-thiophene	2.5	4 m	85	
14	1a	C_6H_5	0.5	а	95 ^c	
15	1j	$4-Cl-C_6H_4$	0.5	j	96 ^c	

^{*a*}Reaction conditions: aldehyde 1 (1 mmol), β-naphthol 2 (1 mmol), 5,5-dimethyl-cyclohexane-1,3-dione 3 (1.2 mmol), p-TSA (0.1 mmol), 70°C.

^bIsolated yield.

^cCyclopentane-1,3-dione was substituted for 5,5-dimethyl-cyclohexane-1,3-dione 3.



Scheme 1. Possible mechanism for the formation of compound 4.

Next, we extended this reaction to phenols and α -naphthol under similar conditions. However, in the case of phenol and α -naphthol, the reaction did not proceed or gave only traces of the expected products under similar conditions after 24 h. A by-product was characterized, which was formed from the condensation of 2 equiv. of 5,5-dimethyl-cyclohexane-1,3-dione 3 and 1 equiv. of aldehydes because of their weaker nucleophilicity compared to β -naphthol.

RCHO 1	+	^{OH} + ^O + ^O	Solvent-free, 7	0°C,)))))	
Entry	1	R	Time (h)	Product	Yields (%) ^b
1	1a	C ₆ H ₅	0.5	6a	95
2	1d	$3-OH-C_6H_4$	0.5	6d	96
3	1f	3-OCH ₃ -C ₆ H ₄	0.5	6f	93
4	1h	3-Me-4-Me-C ₆ H ₃	0.5	6h	92
5	1j	$4-Cl-C_6H_4$	0.5	6j	94
6	1k	CH ₃ CH ₂	2.5	6k	85
7	1n	2-Furan	2.5	6n	80
8	10	2-OMe-4-OMe- C_6H_3	1.0	60	90
9	lp		0.5	бр	91
10	1q	$C_6H_5CH_2$	1.0	6q	86

Table 3. *p*-TSA-promoted one-pot condensation of β -naphthol, aldehydes, and cyclopentane-1,3-dione^{*a*}

^aReaction conditions: aldehyde 1 (1 mmol), β -naphthol 2 (1 mmol), cyclopentane-1,3-dione 5 (1.2 mmol), p-TSA (0.1 mmol), 70°C.

^bIsolated yield.

We next tried to observe the effect of substituents in 5,5-dimethyl-1,3cyclohexanedione 3 using cyclopentane-1,3-dione 5. Both aromatic and aliphatic aldehydes 1 reacted well with 2 and 5 in the presence of $10 \mod \%$ of *p*-TSA to afford 6 in good to high yields under similar conditions (Table 3).

In summary, we have developed a facile and efficient method for the synthesis of 8,9,10,12-tetrahydrobenzo[a]xanthen-11-one or 8,9-dihydrobenzo-[f]cyclopenta[b]chromen-10(11*H*)-one derivatives. The easy workup procedure, short reaction times, solvent-free conditions, and good yields make this method an important addition to the existing methodologies.

EXPERIMENTAL

All reagents are commercially available and were used without any purification. Melting points were recorded on a Büchi B-540 capillary melting-point apparatus and are uncorrected. ¹H NMR and ¹³C NMR spectra were recorded on a Varian Mercury Plus-400 instrument using CDCl₃ as the solvent with tetramethylsilane (TMS) as an internal standard at room temperature. Chemical shifts are given in δ relative to TMS; the coupling constants *J* are given in hertz. IR spectra were recorded on an Avatar 370 Fourier transform (FT)–IR spectrophotometer. MS were measured with Finnigan Trace DSQ (EI, CI) or Thermo Finnigan LCQ-Advantage (ESI).

General Procedure for the Synthesis of Product 4 or 6

A mixture of aldehyde 1 (1.0 mmol), β -naphthol 2 (1.0 mmol), dimedone 3 or 5 (1.2 mmol), and *p*-TSA (10 mol%) were irradiated at 70 °C in the presence of ultrasonic waves for a certain time as shown in Tables 2 and 3. The progress of the reaction was monitored by thin-layer chromatography (TLC) until the disappearance of aldehyde. After the completion of the reaction, water (15 mL) was added and the crude product was extracted with ethyl acetate (3 × 10 mL). The organic layer was dried (MgSO₄) and evaporated, and the crude product was purified through crystallization from ethanol to afford a pure product.

Data

9,9-Dimethyl-12-phenyl-9,10-dihydro-8*H***-benzo[a]xanthen-11(12***H***)-one (4a). Mp 154–155 °C. \nu_{max} (KBr)/cm⁻¹ 3125, 2954, 1651, 1595, 1398, 1375, 1228, 1176, 1025, 808, 743, 699, 510. ¹H NMR (400 MHz, CDCl₃): \delta_{\rm H} 7.99 (1H, d, J = 8.4 Hz, Ar-H), 7.79–7.75 (2H, m, Ar-H), 7.45–7.31 (5H, m, Ar-H), 7.19–7.15 (2H, m, Ar-H), 7.07–7.03 (1H, m, Ar-H), 5.71 (1H, s, CH), 2.57 (2H, s, CH₂), 2.27 (2H, m, CH₂), 1.12 (3H, s, CH₃), 0.96 (3H, s, CH₃). ¹³C NMR (100 MHz, CDCl₃): \delta_{\rm C} 196.9, 163.9, 147.8, 144.7, 131.5, 131.4, 128.8, 128.4, 128.2, 127.0, 126.2, 124.9, 123.7, 117.7, 117.0, 114.3, 50.9, 41.4, 34.7, 32.2, 29.3, 27.1. MS (ESI) m/z 355 ([M + 1]⁺, 100), 274 (20).**

12-(3-Fluorophenyl)-9,9-dimethyl-9,10-dihydro-8*H***-benzo[a]xanthen-11(12***H***)-one (4b).** Mp 157–158 °C. ¹H NMR (400 MHz, CDCl₃): $\delta_{\rm H}$ 7.93 (1H, d, J = 8.0 Hz, Ar-H), 7.80–7.77 (2H, m, Ar-H), 7.46–7.32 (3H, m, Ar-H), 7.26–7.11 (2H, m, Ar-H), 7.00–6.96 (1H, m, Ar-H), 6.77–6.75 (1H, m, Ar-H), 5.72 (1H, s, CH), 2.57 (2H, s, CH₂), 2.28 (2H, m, CH₂), 1.12 (3H, s, CH₃), 0.97 (3H, s, CH₃). ¹³C NMR (100 MHz, CDCl₃): $\delta_{\rm C}$ 196.8, 164.1, 164.0, 161.6, 147.8, 147.2, 147.1, 131.5, 131.2, 129.6, 129.5, 129.1, 127.1, 125.0, 124.1, 123.4, 117.0, 116.9, 115.4, 115.2, 113.7, 113.3, 113.1. MS (EI) m/z 372.2 (M⁺, 30), 277.1 (100). HRMS (EI): calcd. for C₂₅H₂₁FO₂ [M⁺]: 372.1526; found: 372.1515.

12-(3-Chlorophenyl)-9,9-dimethyl-9,10-dihydro-8*H***-benzo[a]xanthen-11(12***H***)-one (4c).** Mp 180–181 °C. ¹H NMR (400 MHz, CDCl₃): $\delta_{\rm H}$ 7.91 (1H, d, J = 8.0 Hz, Ar-H), 7.80–7.77 (2H, m, Ar-H), 7.47–7.25 (5H, m, Ar-H), 7.11 (1H, t, J = 8.0 Hz, Ar-H), 7.05–7.02 (1H, m, Ar-H), 5.69 (1H, s, CH), 2.57 (2H, s, CH₂), 2.28 (2H, m, CH₂), 1.12 (3H, s, CH₃), 0.97 (3H, s, CH₃). ¹³C NMR (100 MHz, CDCl₃): $\delta_{\rm C}$ 196.7, 164.1, 147.8, 146.7, 134.1, 131.5, 131.2, 129.4, 129.1, 128.4, 127.1, 126.8, 126.5, 125.0, 123.4, 117.0, 116.8, 113.7, 50.8, 41.4, 36.2, 34.5, 32.2, 29.2, 27.1, 18.4. MS (ESI) m/z 389.1 ([M + 1]⁺, 100). HRMS (EI): calcd. for C₂₅H₂₁ClO₂ [M⁺]: 388.1230; found: 388.1238.

12-(3-Hydroxyphenyl)-9,9-dimethyl-9,10-dihydro-8*H*-benzo[a]xanthen-**11(12***H*)-one (4d). Mp 240–242 °C. ¹H NMR (400 MHz, CDCl₃): $\delta_{\rm H}$ 9.22 (1H, s, OH), 8.02 (1H, d, *J* 8.8 Hz, Ar-H), 7.92 (2H, d, *J* 9.2 Hz, Ar-H), 7.53–7.42 (3H, m, Ar-H), 6.93 (1H, m, Ar-H), 6.73 (1H, d, *J*=8.4 Hz, Ar-H), 6.65 (1H, s, Ar-H), 6.44 (1H, d, *J*=8.4 Hz, Ar-H), 5.49 (1H, s, CH), 2.63 (2H, m, CH₂), 2.33 (1H, d, *J*=15.6 Hz, CH₂), 2.15 (1H, d, *J*=16.8 Hz, CH₂), 1.06 (3H, s, CH₃), 0.91 (3H, s, CH₃). ¹³C NMR (100 MHz, CDCl₃): $\delta_{\rm C}$ 195.8, 163.7, 157.1, 147.2, 146.2, 131.0, 130.7, 129.0, 128.5, 127.1, 124.9, 123.2, 118.9, 117.4, 117.1, 115.1, 113.2, 50.1, 33.9, 31.9, 28.8, 26.3. MS (ESI) m/z 369.3 ([M – 1]⁺, 100). HRMS (EI): calcd. for C₂₅H₂₂O₃ [M⁺]: 370.1569; found: 370.1577.

9,9-Dimethyl-12-(3-nitrophenyl)-9,10-dihydro-8H-benzo[a]xanthen-11(12H)-one (4e). Mp 169–170 °C. ν_{max} (KBr)/cm⁻¹ 3125, 2954, 2864, 1649, 1596, 1529, 1375, 1344, 1225, 1025, 812, 748, 683, 510. ¹H NMR (500 MHz, CDCl₃): $\delta_{\rm H}$ 8.12–7.79 (6H, m, 7.47–7.35 (4H, m, Ar-H), 5.82 (1H, s, CH), 2.61 (2H, s, CH₂), 2.29 (2H, m, CH₂), 1.13 (3H, s, CH₃), 0.95 (3H, s, CH₃). ¹³C NMR (125 MHz, CDCl₃): $\delta_{\rm C}$ 196.8, 164.6, 148.4, 147.8, 146.8, 134.9, 131.6, 131.0, 129.7, 129.1, 128.7, 127.4, 125.2, 123.3, 123.1, 121.6, 117.3, 116.0, 113.1, 50.8, 41.4, 34.8, 32.3, 29.3, 27.1. MS (EI) m/z 399 (M⁺, 8), 165 (56), 221 (40), 277 (100).

12-(3-Methoxyphenyl)-9,9-dimethyl-9,10-dihydro-8*H*-benzo[a]xanthen-**11(12***H*)-one (4f). Mp 169–170 °C. ¹H NMR (400 MHz, CDCl₃): $\delta_{\rm H}$ 8.00 (1H, d, J = 8.4 Hz, Ar-H), 7.76 (2H, m, Ar-H), 7.45–7.30 (7.08 (1H, t, J = 8.0 Hz, Ar-H), 6.94–6.89 (2H, m, Ar-H), 5.69 (1H, s, CH), 3.70 (3H, s, CH₃), 2.56 (2H, s, CH₂), 2.27 (2H, d, J = 5.6 Hz, CH₂), 1.11 (3H, s, CH₃), 0.97 (3H, s, CH₃). ¹³C NMR (100 MHz, CDCl₃): $\delta_{\rm C}$ 196.8, 163.9, 159.4, 147.7, 146.3, 131.4, 129.0, 128.8, 128.3, 126.9, 124.8, 123.6, 121.0, 117.5, 117.0, 114.6, 114.1, 111.2, 55.0, 50.9, 41.4, 34.6, 32.2, 29.2, 27.2. MS (ESI) m/z 385.3 ([M + 1]⁺, 100). HRMS (EI): calcd. for C₂₆H₂₄O₃ [M⁺]: 384.1725; found: 384.1735.

12-(4-Hydroxy-3-methoxyphenyl)-9,9-dimethyl-9,10-dihydro-8*H*-benzo-[a]xanthen-11(12*H*)-one (4g). Mp 164–166 °C. ¹H NMR (400 MHz, CDCl₃): $\delta_{\rm H}$ 7.98 (1H, d, J = 8.4 Hz, Ar-H), 7.79–7.74 (2H, m, Ar-H), 7.45–7.35 (2H, m, Ar-H), 7.31 (1H, d, J = 9.2 Hz, Ar-H), 7.00 (1H, m, Ar-H), 6.68–6.63 (2H, m, Ar-H), 5.64 (1H, s, CH), 3.82 (3H, s, CH₃), 2.56 (2H, s, CH₂), 2.28 (2H, d, J = 5.2 Hz, CH₂), 1.11 (3H, s, CH₃), 0.98 (3H, s, CH₃). ¹³C NMR (100 MHz, CDCl₃): $\delta_{\rm C}$ 197.2, 163.8, 147.7, 146.1, 143.8, 137.0, 131.5, 131.4, 128.7, 128.4, 126.9, 124.9, 123.7, 121.0, 117.8, 117.0, 114.1, 111.4, 55.9, 50.9, 41.4, 34.1, 32.2, 29.3, 27.1. MS (ESI) m/z 399.3 ([M – 1]⁺, 100).

12-(3,4-Dimethylphenyl)-9,9-dimethyl-9,10-dihydro-8*H***-benzo[a]xanthen-11(12***H***)-one (4h).** Mp 181–182 °C. ν_{max} (KBr)/cm⁻¹ 3125, 2958, 1650, 1593, 1398, 1371, 1237, 1226, 1172, 819, 747, 478. ¹H NMR (500 MHz, CDCl₃): $\delta_{\rm H}$ 8.04 (1H, d, J = 8.5 Hz, Ar-H), 7.77–7.72 (2H, m, Ar-H), 7.43–7.25 (3H, m, 7.10–7.03 (2H, m, Ar-H), 6.91 (1H, d, J = 8.0 Hz, Ar-H), 5.63 (1H, s, CH), 2.56 (2H, m, CH₂), 2.27 (2H, m, CH₂), 2.13 (3H, s, CH₃), 2.09 (3H, s, CH₃), 1.11 (3H, s, CH₃), 0.99 (3H, s, CH₃). ¹³C NMR (125 MHz, CDCl₃): $\delta_{\rm C}$ 196.2, 163.8, 147.7, 142.3, 136.2, 134.4, 131.5, 129.7, 129.5, 128.7, 128.4, 127.0, 125.9, 124.9, 123.8, 118.1, 117.1, 114.5, 51.0, 41.4, 34.3, 32.4, 29.2, 27.4, 20.0, 19.4. MS (EI) m/z 382 (M⁺, 40), 277 (100).

12-(4-Methoxyphenyl)-9,9-dimethyl-9,10-dihydro-8*H*-benzo[a]xanthen-**11(12***H*)-one (4i). Mp 205–206 °C. ν_{max} (KBr)/cm⁻¹ 3121, 2958, 1652, 1607, 1507, 1462, 1382, 1218, 1143, 1028, 834, 747, 661, 539. ¹H NMR (400 MHz, CDCl₃): $\delta_{\rm H}$ 7.98 (1H, d, J=8.4 Hz, Ar-H), 7.78–7.73 (2H, m, Ar-H), 7.45–7.24 (5H, m, Ar-H), 6.71–6.69 (2H, m, Ar-H), 5.66 (1H, s, CH), 3.68 (3H, s, CH₃), 2.56 (2H, s, CH₂), 2.27 (2H, m, CH₂), 1.11 (3H, s, CH₃), 0.97 (3H, s, CH₃). ¹³C NMR (100 MHz, CDCl₃): $\delta_{\rm C}$ 197.0, 163.7, 157.8, 147.7, 137.2, 131.5, 131.4, 129.4, 128.7, 128.4, 127.0, 124.9, 123.7, 117.9, 117.1, 114.4, 113.6, 55.1, 50.9, 41.4, 33.9, 32.3, 29.3, 27.2. MS (EI) m/z 384 (M⁺, 40), 277 (100).

12-(4-Chlorophenyl)-9,9-dimethyl-9,10-dihydro-8*H*-benzo[a]xanthen-**11(12***H*)-one (4j). Mp 181–182 °C. ν_{max} (KBr)/cm⁻¹ 3133, 2958, 1648, 1596, 1483, 1400, 1375, 1224, 1139, 1009, 841, 747, 535. ¹H NMR (500 MHz, CDCl₃): $\delta_{\rm H}$ 7.90 (1H, d, J = 8.5 Hz, Ar-H), 7.79–7.76 (2H, m, Ar-H), 7.45–7.25 (5H, m, Ar-H), 7.14–7.12 (2H, m, Ar-H), 5.68 (1H, s, CH), 2.56 (2H, s, CH₂), 2.28 (2H, m, CH₂), 1.12 (3H, s, CH₃), 0.96 (3H, s, CH₃). ¹³C NMR (125 MHz, CDCl₃): $\delta_{\rm C}$ 196.9, 164.1, 147.7, 143.3, 131.9, 131.5, 131.2, 129.8, 129.1, 128.5, 128.4, 127.2, 125.0, 123.5, 117.1, 113.9, 50.9, 41.4, 34.2, 32.3, 29.3, 27.1. MS (EI) m/z 388 (M⁺, 15), 221 (20), 277 (100).

12-Ethyl-9,9-dimethyl-9,10-dihydro-8*H***-benzo[a]xanthen-11(12***H***)-one (4k).** ν_{max} (KBr)/cm⁻¹ 3130, 2960, 1651, 1595, 1394, 1225, 1177, 1145, 813, 748, 649, 480. ¹H NMR (400 MHz, CDCl₃): $\delta_{\rm H}$ 8.10 (1H, d, J=8.0 Hz, Ar-H), 7.82 (1H, d, J=8.0 Hz, Ar-H), 7.70 (1H, d, J=9.0 Hz, Ar-H), 7.56–7.53 (1H, m, Ar-H), 7.45–7.42 (1H, m, Ar-H), 7.20 (1H, d, J=8.5 Hz, Ar-H), 4.74 (1H, t, J=4.0 Hz, CH), 2.55 (2H, d, J=3.5 Hz, Ar-H), 2.37 (2H, d, J=4.5 Hz, Ar-H), 1.86–1.83 (2H, m, CH₂), 1.20 (3H, s, CH₃), 1.16 (3H, s, CH₃), 0.61 (3H, t, J=7.5 Hz, CH₃). ¹³C NMR (100 MHz, CDCl₃): $\delta_{\rm C}$ 197.6, 166.3, 148.7, 131.5, 131.2, 128.6, 128.0, 126.7, 124.8, 123.3, 117.7, 116.8, 112.1, 51.1, 41.4, 32.2, 29.7, 28.7, 27.4, 27.3, 9.0. MS (EI) m/z 305 ([M – 1]⁺, 2), 85 (100), 124 (60), 221 (65), 275 (90). **12**-*tert*-Butyl-9,9-dimethyl-9,10-dihydro-8*H*-benzo[a]xanthen-11(12*H*)-one (4l). Mp 110–111 °C. ν_{max} (KBr)/cm⁻¹ 3125, 2962, 1642, 1592, 1394, 1220, 1176, 1005, 812, 750, 616, 490. ¹H NMR (400 MHz, CDCl₃): $\delta_{\rm H}$ 8.21 (1H, d, J=9.0 Hz, Ar-H), 7.80 (1H, d, J=8.0 Hz, Ar-H), 7.72 (1H, d, J=9.0 Hz, Ar-H), 7.52–7.49 (1H, m, Ar-H), 7.43–7.40 (1H, m, Ar-H), 7.28 (1H, d, J=9.0 Hz, Ar-H), 4.62 (1H, s, CH), 2.65 (1H, d, J=18.0 Hz, CH₂), 2.52 (1H, d, J=18.0 Hz, CH₂), 2.42 (1H, d, J=16.5 Hz, CH₂), 2.28 (1H, d, J=16.5 Hz, CH₂), 1.27 (3H, s, CH₃), 1.14 (3H, s, CH₃), 0.78 (9H, s, CH₃). ¹³C NMR (100 MHz, CDCl₃): $\delta_{\rm C}$ 197.2, 167.6, 150.6, 132.7, 131.3, 128.2 (CH), 127.8, 126.0, 124.6, 118.4, 116.8, 113.9, 51.0, 41.6, 40.0, 35.9, 31.7, 30.1, 27.8, 27.4. MS (EI) m/z 277 ([M – 57]⁺, 100), 165 (45), 221 (30).

9,9-Dimethyl-12-(thiophen-2-yl)-9,10-dihydro-8*H***-benzo[a]xanthen-11(12***H***)-one (4m).** Mp 180–181 °C. ν_{max} (KBr)/cm⁻¹ 3105, 2958, 1651, 1593, 1376, 1224, 1177, 1147, 1009, 813, 746, 700, 661, 507. ¹H NMR (500 MHz, CDCl₃): $\delta_{\rm H}$ 8.04 (1H, d, J = 8.5 Hz, Ar-H), 7.82–7.78 (2H, m, Ar-H), 7.51–7.30 (3H, m, Ar-H), 7.01–7.00 (1H, m, CH), 6.77–6.74 (2H, m, CH), 6.04 (1H, s, CH), 2.57 (2H, s, CH₂), 2.35 (2H, s, CH₂), 1.14 (3H, s, CH₃), 1.05 (3H, s, CH₃). ¹³C NMR (125 MHz, CDCl₃): $\delta_{\rm C}$ 196.8, 164.6, 148.6, 147.8, 131.4, 129.1, 128.4, 127.2, 126.3, 125.1, 125.0, 124.0, 123.5, 117.2, 117.1, 113.8, 50.9, 41.4, 32.3, 29.4, 29.3, 27.2. MS (EI) m/z 360 (M⁺, 75), 165 (55), 221 (50), 277 (100), 327 (60).

11-Phenyl-8,9-dihydrobenzo[f]cyclopenta[b]chromen-10(11*H***)-one (6a). Mp 237–238 °C. \nu_{max} (KBr)/cm⁻¹ 3391, 3125, 1705, 1667, 1596, 1377, 1232, 1101, 1011, 942, 811, 747, 696, 528, 509. ¹H NMR (400 MHz, CDCl₃): \delta_{\rm H} 7.84–7.77 (3H, m, Ar-H), 7.40–7.38 (3H, m, Ar-H), 7.28–7.26 (2H, m, Ar-H), 7.21–7.18 (2H, m, Ar-H), 7.12–7.09 (1H, m, Ar-H), 5.58 (1H, s, CH), 2.84–2.73 (2H, m, CH₂), 2.55–2.45 (2H, m, CH₂). ¹³C NMR (125 MHz, CDCl₃): \delta_{\rm C} 202.5, 177.1, 149.2, 143.6, 131.8, 131.7, 129.6, 128.5, 128.4, 128.2, 127.2, 126.6, 125.2, 124.2, 118.9, 117.4, 116.1, 36.0, 33.8, 25.4. MS (EI) m/z 312 (M⁺, 20), 235 (100).**

11-(3-Hydroxyphenyl)-8,9-dihydrobenzo[f]cyclopenta[b]chromen-10(11*H***)one (6d). Mp > 300 °C. ¹H NMR (400 MHz, CDCl₃): \delta_{\rm H} 9.23 (1H, s, OH), 7.99–7.86 (3H, m, Ar-H), 7.54–7.42 (4H, m, Ar-H), 7.00–6.96 (1H, m, Ar-H), 6.69 (1H, d,** *J* **= 8.0 Hz, Ar-H), 6.55–6.46 (2H, m, Ar-H), 5.41 (1H, s, CH), 2.90–2.76 (2H, m, CH₂), 2.51–2.41 (2H, m, CH₂). ¹³C NMR (100 MHz, CDCl₃): \delta_{\rm C} 201.4, 176.8, 157.2, 148.4, 145.3, 131.3, 131.1, 129.6, 129.1, 128.4, 127.0, 125.1, 123.9, 118.8, 117.9, 117.4, 116.0, 114.9, 113.4, 35.1, 33.5, 24.7. MS (ESI) m/z 329.1 ([M + 1]⁺, 100). HRMS (EI): calcd. for C₂₂H₁₆O₃ [M⁺]: 328.1099; found: 328.1106.**

11-(3-Methoxyphenyl)-8,9-dihydrobenzo[f]cyclopenta[b]chromen-10(11*H***)one (6f). Mp 197–198 °C. ¹H NMR (400 MHz, CDCl₃): \delta_{\rm H} 7.83–7.78 (3H, m, Ar-H), 7.41–7.36 (3H, m, Ar-H), 7.13–7.09 (1H, m, Ar-H), 6.85–6.83 (2H, m, Ar-H), 6.66–6.64 (1H, m, Ar-H), 5.55 (1H, s, CH), 3.71 (3H, s, CH₃), 2.80–2.75 (2H, m, CH₂), 2.51–2.49 (2H, m, CH₂). ¹³C NMR (100 MHz, CDCl₃): \delta_{\rm C} 202.3, 177.1, 159.6, 149.2, 145.1, 131.8, 131.7, 129.5, 129.3, 128.4, 127.1, 125.1, 124.1, 120.6, 118.7, 117.3, 115.9, 114.3, 111.6, 55.1, 35.9, 33.7, 25.3. MS (ESI) m/z 343.2 ([M + 1]⁺, 100). HRMS (EI): calcd. for C₂₃H₁₈O₃ [M⁺]: 342.1256; found: 342.1267.** **11-(3,4-Dimethylphenyl)-8,9-dihydrobenzo[f]cyclopenta[b]chromen-10(11***H***)-one (6h). Mp 223–224 °C. \nu_{max} (KBr)/cm⁻¹ 3407, 3131, 1709, 1670, 1591, 1396, 1237, 943, 825, 746, 498. ¹H NMR (500 MHz, CDCl₃): \delta_{\rm H} 7.83–7.79 (3H, m, Ar-H), 7.40–7.36 (3H, m, Ar-H), 7.03–6.93 (3H, m, Ar-H), 5.50 (1H, s, CH), 2.79–2.75 (2H, m, CH₂), 2.50–2.47 (2H, m, CH₂), 2.13 (3H, s, CH₃), 2.12 (3H, s, CH₃). ¹³C NMR (125 MHz, CDCl₃): \delta_{\rm C} 202.5, 177.1, 149.2, 141.2, 136.6, 134.8, 131.9, 131.8, 129.7, 129.4, 129.3, 128.4, 127.1, 125.5, 125.1, 124.2, 119.2, 117.4, 116.5, 35.5, 33.8, 25.3, 19.9, 19.4. MS (EI) m/z 340 (M⁺, 10), 178 (22), 235 (100).**

11-(4-Chlorophenyl)-8,9-dihydrobenzo[f]cyclopenta[b]chromen-10(11*H***)one (6j). Mp 233–234 °C. \nu_{max} (KBr)/cm⁻¹ 3426, 3131, 1699, 1658, 1396, 1233, 1088, 1013, 9445, 819, 744, 527. ¹H NMR (500 MHz, CDCl₃): \delta_{\rm H} 7.85–7.82 (2H, m, Ar-H), 7.71–7.69 (1H, m, Ar-H), 7.41–7.38 (3H, m, Ar-H), 7.21–7.15 (4H, m, Ar-H), 5.55 (1H, s, CH), 2.84–2.75 (2H, m, CH₂), 2.56–2.45 (2H, m, CH₂). ¹³C NMR (125 MHz, CDCl₃): \delta_{\rm C} 202.4, 177.2, 149.2, 142.0, 132.4, 131.8, 131.6, 129.9, 129.5, 128.7, 128.6, 127.3, 125.3, 124.0, 118.3, 117.4, 115.5, 35.5, 33.8, 25.4. MS (EI) m/z 346 (M⁺, 10), 141 (20), 235 (100).**

11-Ethyl-8,9-dihydrobenzo[f]cyclopenta[b]chromen-10(11*H***)-one (6k). Mp 166–167 °C. ¹H NMR (400 MHz, CDCl₃): \delta_{\rm H} 8.02 (1H, d, J = 8.4 Hz, Ar-H), 7.83 (1H, d, J = 8.0 Hz, Ar-H), 7.73 (1H, d, J = 8.4 Hz, Ar-H), 7.57–7.44 (2H, m, Ar-H), 7.24 (1H, d, J = 8.0 Hz, Ar-H), 4.61 (1H, s, CH), 2.92–2.55 (4H, m, CH₂), 2.00–1.89 (2H, m, CH₂), 0.61 (3H, t, J = 7.6 Hz, CH₃). ¹³C NMR (100 MHz, CDCl₃): \delta_{\rm C} 203.3, 179.8, 149.4, 131.7, 131.4, 128.6, 126.8, 125.0, 123.3, 117.4, 117.1, 116.8, 33.9, 29.8, 26.6, 25.2, 9.0. MS (ESI) m/z 265.2 ([M + 1]⁺, 100). HRMS (EI): calcd. for C₁₈H₁₆O₂ [M⁺]: 264.1150; found: 264.1156.**

11-(Furan-2-yl)-8,9-dihydrobenzo[f]cyclopenta[b]chromen-10(11*H***)-one (6n**). Mp 220–222 °C. ¹H NMR (400 MHz, CDCl₃): $\delta_{\rm H}$ 7.97 (1H, d, J = 8.4 Hz, Ar-H), 7.84–7.81 (2H, m, Ar-H), 7.52–7.42 (2H, m, Ar-H), 7.35 (1H, d, J = 8.8 Hz, Ar-H), 7.18 (1H, m, CH), 6.20–6.19 (1H, m, CH), 6.08 (1H, d, J = 7.2 Hz, CH), 5.68 (1H, s, CH), 2.93–2.76 (2H, m, CH₂), 2.64–2.54 (2H, m, CH₂). ¹³C NMR (100 MHz, CDCl₃): $\delta_{\rm C}$ 202.4, 178.6, 155.2, 149.0, 141.8, 132.1, 131.8, 129.9, 128.7, 127.4, 125.5, 123.8, 117.7, 114.2, 110.5, 106.9, 34.0, 29.7, 25.6. MS (EI) m/z 302.2 (M⁺, 30). HRMS (EI): calcd. for C₂₀H₁₄O₃ [M⁺]: 302. 0943; found: 302.0951.

11-(2,4-Dimethoxyphenyl)-8,9-dihydrobenzo[f]cyclopenta[b]chromen-10(11*H***)-one (6o). Mp 192–193 °C. ¹H NMR (400 MHz, CDCl₃): \delta_{\rm H} 8.00 (1H, d,** *J* **8.4 Hz, Ar-H), 7.77–7.73 (2H, m, Ar-H), 7.42–7.31 (3H, m, Ar-H), 6.93 (1H, d,** *J* **= 8.4 Hz, Ar-H), 6.42–6.29 (2H, m, Ar-H), 5.80 (1H, s, CH), 3.93 (3H, s, OCH₃), 3.69 (3H, s, OCH₃), 2.80 (2H, s, CH₂), 2.47 (2H, s, CH₂). ¹³C NMR (100 MHz, CDCl₃): \delta_{\rm C} 202.5, 177.3, 159.4, 157.3, 148.8, 132.0, 131.6, 130.5, 128.8, 128.2, 126.9, 125.4, 124.9, 124.0, 119.3, 117.7, 117.2, 105.3, 99.0, 56.1, 55.1, 33.8, 29.1, 25.3. MS (ESI) m/z 373.1 ([M + 1]⁺, 100). HRMS (EI): calcd. for C₂₄H₂₀O₄ [M⁺]: 372.1362; found: 372.1371.**

11-(Benzo[d][1,3]dioxol-5-yl)-8,9-dihydrobenzo[f]cyclopenta[b]chromen-10(11*H***)-one (6p). Mp 228–230 °C. ¹H NMR (400 MHz, CDCl₃): \delta_{\rm H} 7.83–7.80 (3H, m, Ar-H), 7.41–7.35 (3H, m, Ar-H), 6.81–6.78 (1H, m, Ar-H), 6.69–6.63 (2H, m,** Ar-H), 5.83 (2H, d, J = 13.6 Hz, Ar-H), 5.49 (1H, s, CH), 2.84–2.72 (2H, m, CH₂), 2.52–2.49 (2H, m, CH₂). ¹³C NMR (100 MHz, CDCl₃): $\delta_{\rm C}$ 202.4, 176.9, 149.0, 147.7, 146.1, 137.7, 131.8, 129.5, 128.4, 127.1, 125.1, 124.1, 121.4, 118.8, 117.3, 116.1, 108.7, 108.0, 100.8, 35.6, 33.7, 25.3. MS (ESI) m/z 357.1 ([M + 1]⁺, 100). HRMS (EI): calcd. for C₂₃H₁₆O₄ [M⁺]: 356.1049; found: 356.1054.

11-Benzyl-8,9-dihydrobenzo[f]cyclopenta[b]chromen-10(11*H***)-one (6q). Mp 181–182 °C. ¹H NMR (400 MHz, CDCl₃): \delta_{\rm H} 8.14 (1H, d, J=8.4 Hz, Ar-H), 7.87 (1H, d, J=8.0 Hz, Ar-H), 7.72 (1H, d, J=8.8 Hz, Ar-H), 7.65–7.61 (1H, m, Ar-H), 7.53–7.49 (1H, m, Ar-H), 7.09–6.96 (4H, m, Ar-H), 6.32 (2H, d, J=7.2 Hz, Hz, Ar-H), 4.89 (1H, t, J=3.6 Hz, CH), 3.21–3.20 (2H, m, CH₂), 2.60–2.44 (4H, m, CH₂). ¹³C NMR (100 MHz, CDCl₃): \delta_{\rm C} 203.1, 179.8, 149.4, 137.4, 131.6, 131.2, 129.6, 128.8, 128.7, 127.3, 127.1, 126.1, 125.1, 123.1, 116.9, 116.4, 115.9, 38.6, 33.7, 30.6, 25.0. MS (ESI) m/z 327.2 ([M + 1]⁺, 100). HRMS (EI): calcd. for C₂₃H₁₈O₂ [M⁺]: 326.1307; found: 372.1297.**

ACKNOWLEDGMENT

We are grateful to the Science and Technology Projects of Zhejiang (Nos. 2008C11046 and 2007C21111) for financial support.

REFERENCES

- Lambert, R. W.; Martin, J. A.; Merrett, J. H.; Parkes, E. B. K.; Thomas, G. J. Preparation of pyrimidine nucleosides as thymidine kinase inhibitors and virucides. PCT Int. Appl. WO9706178, 1997.
- Hideo, T.; Teruomi, J. [1]Benzopyrano[2,3-b]xanthene derivatives. Jpn. Patent 56005480, 1981.
- Poupelin, J. P.; Saint-Ruf, G.; Foussard-Blanpin, O.; Marcisse, G.; Uchida-Ernouf, G.; Lacroix, R. Synthesis and antiinflammatory properties of bis(2-hydroxy-1-naphthyl) methane derivatives. *Eur. J. Med. Chem.* 1978, 13, 67–71.
- Buu-Hoi, N. P.; Saint-Ruf, G. De. A.; Hieu, H. T. Inducing or inhibitory activity of dibenzo[a,j]- and dibenzo[c,h]xanthenes or zoxazolamine hydroxylase synthesis in vivo: Molecular structure-activity relations. *Chim. Ther.* 1972, 7, 83–86.
- Saint-Ruf, G.; Hieu, H. T.; Poupelin, J. P. Effect of dibenzoxanthenes on the paralyzing action of zoxazolamine. *Naturwissenschaften* 1975, 62, 584–585.
- Ion, R. M. Photodynamic therapy of cancer—A photocatalytic process or a photosensibilization process? *Prog. Catal.* 1997, 2, 55–64.
- Ion, R. M.; Planner, A.; Wiktorowicz, K.; Frackowiak, D. The incorporation of various porphyrins into blood cells measured via flow cytometry, absorption, and emission spectroscopy. *Acta. Biochim. Pol.* **1998**, *45*, 833–845.
- Banerjee, A.; Mukherjee, A. K. Chemical aspects of santalin as a histological stain. *Stain Technol.* 1981, 56, 83–85.
- Menchen, S. M.; Benson, S. C.; Lam, J. Y. L.; Zhen, W.; Sun, D.; Rosenblum, B. B.; Khan, S. H.; Taing, M. Sulfonated diarylrhodamine dyes for labeling polynucleotides. US Patent 6583168, 2003.
- Sirkencioglu, O.; Talinli, N.; Akar, A. Synthesis of 14-alkyl-l4H-dibenzo[a,j]xanthenes. J. Chem. Res., Synop. 1995, 502.

- 11. Ahmad, M.; King, T. A.; Ko, D.-K.; Cha, B. H.; Lee, J. Performance and photostability of xanthene and pyrromethene laser dyes in sol-gel phases. *J. Phys. D: Appl. Phys.* **2002**, *35*, 1473–1476.
- 12. Knight, C. G.; Stephens, T. Xanthene-dye-labelled phosphatidylethanolamines as probes of interfacial pH: Studies in phospholipid vesicles. *Biochem. J.* **1989**, *258*, 683–687.
- Das, B.; Laxminarayana, K.; Krishnaiah, M.; Srinivas, Y. An efficient and convenient protocol for the synthesis of novel 12-aryl- or 12-alkyl-8,9,10,12-tetrahydrobenzo[a]xanthen-11-one derivatives. *Synlett* 2007, 20, 3107–3112.
- Li, J. J.; Tang, W. Y.; Lu, L. M.; Su, W. K. Strontium triflate-catalyzed one-pot condensation of β-naphthol, aldehydes, and cyclic 1,3-dicarbonyl compounds. *Tetrahedron Lett.* 2008, 49, 7117–7120.
- Orru, R. V. A.; de Greef, M. Recent advances in solution-phase multicomponent methodology for the synthesis of heterocyclic compounds. *Synthesis* 2003, 10, 1471–1499.
- Zhu, J. P. Recent developments in the isonitrile-based multicomponent synthesis of heterocycles. *Eur. J. Org. Chem.* 2003, 1133–1144.
- Hulme, C.; Gore, V. Multi-component reactions: Emerging chemistry in drug discovery from xylocain to crixivan. *Curr. Med. Chem.* 2003, 10, 51–80.
- Yadav, J. S.; Reddy, B. V. S.; Eeshwaraiah, B.; Gupta, M. K. Bi(OTf)₃/[bmim]BF₄ as novel and reusable catalytic system for the synthesis of furan, pyrrole, and thiophene derivatives. *Tetrahedron Lett.* 2004, 45, 5873–5876.
- Wang, B.; Gu, Y. L.; Luo, C.; Yang, T.; Yang, L. M.; Suo, J. S. Pyrrole synthesis in ionic liquids by Paal–Knorr condensation under mild conditions. *Tetrahedron Lett.* 2004, 45, 3417–3419.
- Su, W. K.; Li, J. J.; Zheng, Z. G.; Shen, Y. C. One-pot synthesis of dihydropyrimidiones catalyzed by strontium(II) triflate under solvent-free conditions. *Tetrahedron Lett.* 2005, 46, 6037–6040.
- Su, W. K.; Hong, Z.; Shan, W. G.; Zhang, X. X. A facile synthesis of 1-substituted-1H-1,2,3,4-tetrazoles catalyzed by ytterbium triflate hydrate. *Eur. J. Org. Chem.* 2006, 2723–2726.
- Zhong, W. H.; Lin, F. L.; Chen, R. E., Su, W. K. An efficient synthesis of 2-hydroxy-7,8dihydroquinolin-5(6H)-ones and 7,8-dihydroquinoline-2,5(1H,6H)-diones from Morita– Baylis–Hillman adduct acetates. Synthesis 2008, 16, 2561–2568.
- Su, W. K.; Tang, W. Y.; Li, J. J. Strontium(II) triflate-catalysed condensation of βnaphthol, aldehyde, and urea or amides: A facile synthesis of amidoalkyl naphthols. J. Chem. Res. 2008, 123–128.
- Su, W. K.; Yang, D.; Jin, C.; Zhang, B. Yb(OTf)₃-catalyzed condensation reaction of naphthol and aldehyde in ionic liquids: A green synthesis of aryl-14H-dibenzo[a,j]xanthenes. *Tetrahedron Lett.* 2008, 49, 3391–3394.