## 1415

## Remarkably Mild and Efficient Cetrimonium Bromide Catalyzed Friedel–Crafts Amidoalkylation of Sesamols with *N*-Boc Imines Generated In Situ in Aqueous Medium

Hui Zhang,<sup>a,b</sup> Ming-Gui Chen,<sup>c</sup> Chun-Xia Lian,<sup>a</sup> Wei-Cheng Yuan,<sup>a</sup> Xiao-Mei Zhang<sup>\*a</sup>

<sup>b</sup> Graduate School of Chinese Academy of Sciences, Beijing 100049, P. R. of China

<sup>c</sup> School of Chemistry and Chemical Engineering, Southwest Petroleum University, Chengdu 610500, P. R. of China

Received 2 February 2010

**Abstract:** A remarkably mild and efficient cetrimonium bromide (CTAB) catalyzed Friedel–Crafts amidoalkylation of sesamol or its 2-substituted derivatives with *N*-Boc imines generated in situ was developed. A wide range of 6-amidoalkyl sesamols were synthesized in high yields. This methodology was also applicable to 2-naphthols.

Key words: sesamol, Friedel–Crafts amidoalkylation,  $\alpha$ -amido sulfones, CTAB, 6-amidoalkyl sesamols

The Friedel–Crafts aminoalkylation or amidoalkylation of electron-rich phenols with imines represents an important transformation for the synthesis of biologically active compounds and chiral amino alcohol ligands. For example, synthesis of 1-aminoalkyl naphthols via Friedel–Crafts amino- or amidoalkylation of 2-naphthols attracted much attention due to the importance of the products.<sup>1</sup>

Sesamol is one of the representative structural motifs often observed in natural alkaloids and biologically active compounds.<sup>2</sup> As an electron-rich phenol, sesamol and its derivatives are good nucleophiles that can be employed in the Friedel–Crafts reaction to construct complex natural products.<sup>3</sup> Jurd has reported the Mannich type Friedel– Crafts aminoalkylation of sesamol and the application of the resulting Mannich bases in the synthesis of anti-tumor agents.<sup>4</sup> Recently, Frackenpohl also reported the same reaction and its application in the synthesis of insecticidal heterolignans.<sup>5</sup> However, to the best of our knowledge, a systematic study on Friedel–Crafts amidoalkylation of sesamol with imines has not been reported.

*N*-Boc imines are of great importance in nucleophilic addition because they are very reactive electrophiles and the Boc group can be easily removed from the resulting adducts, however, *N*-Boc imines are known to be unstable and difficult to handle. Gratifyingly, this problem can be circumvented by generating *N*-Boc imines in situ from stable  $\alpha$ -amido sulfones.<sup>6</sup> Herein, we report a remarkably mild and efficient Friedel–Crafts amidoalkylation of sesamol or its 2-substituted derivatives with *N*-Boc imines

SYNLETT 2010, No. 9, pp 1415–1417 Advanced online publication: 25.03.2010 DOI: 10.1055/s-0029-1219802; Art ID: W01910ST © Georg Thieme Verlag Stuttgart · New York generated in situ, through which a wide variety of 6-amidoalkyl sesamols were synthesized in high yields.

First, we investigated the Friedel–Crafts amidoalkylation of sesamol **1a** with *N*-Boc imine generated in situ from  $\alpha$ amido sulfone **2a** in water at 30 °C, employing 1.5 equivalents of CsOH as the base (Table 1, entry 1). The reaction proceeded in two days to afford 64% of the desired product 6-amidoalkyl sesamol **3a**. Encouraged by this finding, we next examined the effects of various bases. Among these bases, Na<sub>2</sub>CO<sub>3</sub> gave a promising yield of 80%. However, no increase in yield was observed by prolonging the reaction time to three days (Table 1, entry 10). When the reaction was performed at higher temperature, many side products arose and the yield dropped significantly (Table 1, entry 11).

Recently, Kobayashi and other groups reported several surfactant-promoted reactions in aqueous medium.<sup>7</sup> Therefore, in order to improve the efficiency of this reaction, we introduced surfactants into the reaction mixture. To our delight, several surfactants accelerated the reaction dramatically. In the presence of 10 mol% of surfactants SDS (sodium 1-dodecanesulfonate) and 12-2-12 [ethanediyl-1,2-bis(dodecyldimethyl-ammonium bromide)], the reaction proceeded smoothly to afford more than 80% yield of the product in five hours (Table 1, entries 12 and 13). Moreover, 10 mol% of CTAB (cetrimonium bromide) provided almost quantitative amounts of product in five hours (Table 1, entry 14). Further research indicated that the reaction was actually complete within only three hours (Table 1, entry 15). However, reducing the catalyst loading to 5 mol% resulted in an obvious decrease of product yield (Table 1, entry 16).

Having established the optimal reaction conditions, the generality of this reaction was examined. In the presence of 10 mol% of CTAB, sesamol or its 2-substituted derivatives underwent Friedel–Crafts amidoalkylation with a wide variety of  $\alpha$ -amido sulfones in water at 30 °C.<sup>8,9</sup> The results are summarized in Table 2.

As can be seen in Table 2, it is clear that the methodology is quite general. All aryl or alkyl  $\alpha$ -amido sulfones tested underwent the reaction smoothly to provide the corresponding 6-amidoalkyl sesamols in good yields. For aryl  $\alpha$ -amido sulfones, the substituents on the aromatic rings

<sup>&</sup>lt;sup>a</sup> Key Laboratory for Asymmetric Synthesis & Chirotechnology of Sichuan Province, Chengdu Institute of Organic Chemistry, Chinese Academy of Sciences, Chengdu 610041, P. R. of China Fax +86(28)85229250; E-mail: xmzhang@cioc.ac.cn

 
 Table 1
 Screening of Bases and Catalysts in the Friedel–Crafts
 Amidoalkylation of Sesamol 1a with  $\alpha$ -Amido Sulfone 2a<sup>a</sup>



2a							
Entry	Base	Catalyst (mol%)	Time (h)	Yield (%)b			
1	CsOH·H <sub>2</sub> O	none	48	64			
2	КОН	none	48	76			
3	NaOH	none	48	51			
4	LiOH·H <sub>2</sub> O	none	48	66			
5	Cs <sub>2</sub> CO <sub>3</sub>	none	48	73			
6	K <sub>2</sub> CO <sub>3</sub>	none	48	61			
7	Na <sub>2</sub> CO <sub>3</sub>	none	48	80			
8	Na <sub>3</sub> PO <sub>4</sub>	none	48	73			

4	$LiOH \cdot H_2O$	none	48	66	
5	Cs <sub>2</sub> CO <sub>3</sub>	none	48	73	
6	K <sub>2</sub> CO <sub>3</sub>	none	48	61	
7	Na <sub>2</sub> CO <sub>3</sub>	none	48	80	
8	Na <sub>3</sub> PO <sub>4</sub>	none	48	73	
9	NaOAc	none	48	33	
10	Na <sub>2</sub> CO <sub>3</sub>	none	72	79	
11°	Na <sub>2</sub> CO <sub>3</sub>	none	48	70	
12	Na <sub>2</sub> CO <sub>3</sub>	SDS (10)	5	82	
13	Na <sub>2</sub> CO <sub>3</sub>	12-2-12 (10)	5	84	
14	Na <sub>2</sub> CO <sub>3</sub>	CTAB (10)	5	97	
15	Na <sub>2</sub> CO <sub>3</sub>	CTAB (10)	3	97	
16	$Na_2CO_3$	CTAB (5)	5	67	

<sup>a</sup> Unless indicated otherwise, the reaction was conducted with sesamol 1a (0.24 mmol),  $\alpha$ -amido sulfone 2a (0.2 mmol) and base (0.3 mmol) in H<sub>2</sub>O (2.0 mL) with or without catalyst at 30 °C. <sup>b</sup> Isolated yield.

<sup>c</sup> The reaction was performed at 50 °C.

affected the reaction rate to some degree. The reactions of aryl  $\alpha$ -amido sulfones bearing electron-donating groups on the para-positions completed within only three hours (Table 2, entries 2 and 4), while reactions of aryl  $\alpha$ -amido sulfones bearing electron-withdrawing groups on the para-positions proceeded slower (Table 2, entries 6, 8 and 9). Prolonged reaction time was needed for aryl  $\alpha$ -amido sulfones bearing meta- or ortho-substituents on the aromatic rings (Table 2, entries 3, 5, 7, and 10-12). In addition, electron-withdrawing substituents on the meta and ortho positions of the aromatic rings retarded the rate and the reaction required to ten hours to complete (Table 2, entries 7, and 10–12). 1-Naphthyl and heteroaromatic  $\alpha$ amidosulfones provided the corresponding products in Table 2 Friedel–Crafts Amidoalkylation of Sesamol 1 with α-Amido Sulfone 2 Catalyzed by CTAB<sup>a</sup>



<sup>a</sup> Unless indicated otherwise, the reaction was conducted with sesamol 1 (0.24 mmol),  $\alpha$ -amido sulfone 2 (0.2 mmol) and Na<sub>2</sub>CO<sub>3</sub> (0.3 mmol) in H<sub>2</sub>O (2.0 mL) with CTAB (10 mol%) at 30 °C. Ar = Ph. <sup>b</sup> Isolated yield.

<sup>c</sup> Ar =  $4 - MeC_6H_4$ 

high yields within five hours (Table 2, entries 13–15). It is noteworthy that even alkyl  $\alpha$ -amidosulfones underwent the reaction smoothly to afford excellent yields of the products (Table 2, entries 16 and 17). In addition, 2-methyl sesamol and 2-iodo sesamol reacted with N-Boc imine generated in situ from  $\alpha$ -amido sulfone **2a** to give the products in good yields after an extended reaction time (Table 2, entries 18 and 19).

Furthermore, when several naphthols were also subjected to the CTAB-catalyzed Friedel–Crafts amidoalkylation with imines generated in situ from  $\alpha$ -amido sulfone **2a** under the optimized reaction conditions, both 2-naphthol and 6-bromo-2-naphthol underwent the reaction rapidly to generate the corresponding products **3t** and **3u** in high yields (Figure 1). However, the reaction of 1-naphthol proceeded sluggishly to provide the product **3v** in poor yield.



**Figure 1** Products of Friedel–Crafts amidoalkylation of naphthols with *N*-Boc imine generated in situ from  $\alpha$ -amido sulfone **1a** 

In conclusion, we have developed a remarkably mild and efficient CTAB-catalyzed Friedel–Crafts amidoalkylation of sesamol or its 2-substituted derivatives with *N*-Boc imines generated in situ. A wide range of 6-amidoalkyl sesamols were synthesized in high yields through this transformation. This methodology was also appropriate for 2-naphthols.

**Supporting Information** for this article is available online at http://www.thieme-connect.com/ejournals/toc/synlett.

## Acknowledgment

We are grateful for financial support from the National Natural Science Foundation of China (20772122) and the National Basic Research Program of China (973 Program) (2010CB833301).

## **References and Notes**

- For selected examples, see: (a) Macleod, P. D.; Li, Z.; Li, C. *Tetrahedron* **2010**, *66*, 1045. (b) Feng, J.; Sarim, D.; Li, C. *Tetrahedron Lett.* **2008**, *49*, 668. (c) Nandi, G. C.; Samai, S.; Kumar, R.; Singh, M. S. *Tetrahedron Lett.* **2009**, *50*, 7220. (d) Shaterion, H. R.; Hosseinian, A.; Ghashang, M. *Synth. Commun.* **2008**, *38*, 3375. (e) Shaterion, H. R.; Yarahmadi, H.; Ghashang, M. *Tetrahedron* **2008**, *64*, 1263. (f) Shaterian, H. R.; Yarahmadi, H. *Tetrahedron Lett.* **2008**, *49*, 1297. (g) Patil, S. B.; Singh, P. R.; Surpur, M. P.; Samant, S. D. *Synth. Commun.* **2007**, *37*, 1659. (h) Selvam, N. P.; Perumal, P. T. *Tetrahedron Lett.* **2006**, *47*, 7481. (i) Khodaei, M. M.; Khosropour, A. R.; Moghanian, H. *Synlett* **2006**, 916.
- (2) (a) Zheng, S.; Chan, C.; Furuuchi, T.; Wright, B. J. D.; Zhou, B.; Guo, J.; Danishefsky, S. J. Angew. Chem. Int. Ed. 2006, 45, 1754. (b) Chen, J.; Chen, X.; Willot, M.; Zhu, J. Angew. Chem. Int. Ed. 2006, 45, 8028. (c) Paolis, M. D.; Chiaroni, A.; Zhu, J. P. Chem. Commun. 2003, 2896.
  (d) Hitotsuyanagi, Y.; Ichihara, Y.; Takeya, K.; Itokawa, H. Tetrahedron Lett. 1994, 35, 9401.

- (3) (a) Chen, X.; Zhu, J. Angew. Chem. Int. Ed. 2007, 46, 3962.
  (b) Rondot, C.; Zhu, J. Org. Lett. 2005, 7, 1641. (c) Bailly, F.; Queffelec, C.; Mbemba, G.; Mouscadet, J.-F.; Cotelle, P. Bioorg. Med. Chem. Lett. 2005, 15, 5053. (d) Alvarez-Manzaneda, E.; Chahboun, R.; Cabrera, E.; Alvarez, E.; Haidour, A.; Ramos, J. M.; Alvarez-Manzaneda, R.; Tapia, R.; Es-Samti, H.; Fernández, A.; Barranco, I. Eur. J. Org. Chem. 2009, 1139.
- (4) (a) Jurd, L. J. Heterocycl. Chem. 1997, 34, 601. (b) Jurd, L. J. Heterocycl. Chem. 1996, 33, 1919. (c) Jurd, L. J. Heterocycl. Chem. 1988, 25, 89. (d) Jurd, L. J. Heterocycl. Chem. 1985, 22, 993.
- (5) Frackenpohl, J.; Adelt, I.; Antonicek, H.; Arnold, C.; Behrmann, P.; Blaha, N.; Böhmer, J.; Gutbrod, O.; Hanke, R.; Hohmann, S.; Houtdreve, M.; Lösel, P.; Malsam, O.; Melchers, M.; Neufert, V.; Peschel, E.; Reckmann, U.; Schenke, T.; Thiesen, H.-P.; Velten, R.; Vogelsang, K.; Weiss, H.-C. *Bioorg. Med. Chem. Lett.* **2009**, *17*, 4160.
- (6) For examples, see: (a) Fini, F.; Sgarzani, V.; Pettersen, D.; Herrera, R. P.; Bernardi, L.; Ricci, A. Angew. Chem. Int. Ed. 2005, 44, 7975. (b) Palomo, C.; Oiarbide, M.; Laso, A.; Lopez, R. J. Am. Chem. Soc. 2005, 127, 17622.
  (c) Marianacci, O.; Micheletti, G.; Bernardi, L.; Fini, F.; Fochi, M.; Pettersen, D.; Sgarzani, V.; Ricci, A. Chem. Eur. J. 2007, 13, 8338. (d) Song, J.; Shih, H.-W.; Deng, L. Org. Lett. 2007, 9, 603. (e) Gianelli, C.; Sambri, L.; Carlone, A.; Bartoli, G.; Melchiorre, P. Angew. Chem. Int. Ed. 2008, 47, 8700. (f) Číhalová, S.; Remeš, M.; Císařová, I.; Veselý, J. Eur. J. Org. Chem. 2009, 6277. (g) Jiang, X. X.; Zhang, Y. F.; Wu, L. P.; Zhang, G.; Liu, X.; Zhang, H. L.; Fu, D.; Wang, R. Adv. Synth. Catal. 2009, 351, 2096.
- (7) (a) Mori, Y.; Kakumoto, K.; Manabe, K.; Kobayashi, S. *Tetrahedron Lett.* **2000**, *41*, 3107. (b) Otto, S.; Engberts, J. B. F. N.; Kwak, J. C. T. *J. Am. Chem. Soc.* **1998**, *120*, 9517. (c) Manabe, K.; Kobayashi, S. *Chem. Commun.* **2000**, 669. (d) Kobayashi, S.; Lam, W. W. L.; Manabe, K. *Tetrahedron Lett.* **2000**, *41*, 6115.
- (8) Friedel–Crafts Amidoalkylation; General Procedure: *N*-Boc  $\alpha$ -amido sulfone (0.2 mmol, 1.0 equiv), Na<sub>2</sub>CO<sub>3</sub> (0.3 mmol, 1.5 equiv), CTAB (0.02 mmol, 10%) and H<sub>2</sub>O (2 mL) were added to a 10-mL glass vial equipped with a small magnetic stirring bar. Sesamol (0.24 mmol, 1.2 equiv) was added and, after stirring for the stipulated time at 30 °C, the mixture was diluted with H<sub>2</sub>O (3 mL) and extracted with Et<sub>2</sub>O (3 × 25 mL). The organic layers were combined, dried over anhydrous sodium sulfate and concentrated under reduced pressure. The residue was subjected to silica gel flash chromatography (EtOAc–Hexanes, 1:10) to give the pure product.
- (9) *Tert*-butyl (6-Hydroxybenzo[d][1,3]dioxol-5-yl)(phenyl)methylcarbamate (3a): Yield: 97%; white solid; mp 168.5– 168.9 °C; <sup>1</sup>H NMR (300 MHz, DMSO- $d_6$ ):  $\delta = 9.31$  (s, 1 H), 7.61 (d, J = 9.6 Hz, 1 H), 7.29–7.22 (m, 4 H), 7.19–7.14 (m, 1 H), 6.92 (s, 1 H), 6.42 (s, 1 H), 6.11 (d, J = 9.6 Hz, 1 H), 5.88 (s, J = 0.6 Hz, 1 H), 5.84 (s, J = 0.6 Hz, 1 H), 1.39 (s, 9 H); <sup>13</sup>C NMR (75 MHz, DMSO- $d_6$ ):  $\delta = 154.99$ , 148.49, 146.09, 143.67, 139.83, 128.03, 126.63, 126.36, 121.09, 107.39, 100.60, 97.39, 77.99, 51.21, 28.28; HRMS (ESI): m/z [M + Na]<sup>+</sup> calcd for C<sub>19</sub>H<sub>21</sub>NNaO<sub>5</sub>: 366.1317; found: 366.1312

Copyright of Synlett is the property of Georg Thieme Verlag Stuttgart and its content may not be copied or emailed to multiple sites or posted to a listserv without the copyright holder's express written permission. However, users may print, download, or email articles for individual use.