Gold-Catalyzed Skeletal Rearrangement of 1-[2-(1*H*-Isochromen-3-yl)aryl]ethanones with Alcohols

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Abstract: A novel gold-catalyzed skeletal rearrangement of 1-[2-(1*H*-isochromen-3-yl)aryl]ethanones with alcohols to construct a five-membered carbocycle ring onto aromatic systems has been developed. This method serves as the first example of synthesizing inden-1-ones by a gold-catalyzed skeletal rearrangement strategy.

Key words: gold, skeletal rearrangement, 1-[2-(1*H*-isochromen-3-yl)aryl]ethanones, alcohols, inden-1-ones

Inden-1-one and its derivatives are commonly found in natural compounds, pharmaceuticals, and functional materials, as well as being versatile synthetic blocks and catalysts in organic synthesis.¹ Most often, the intramolecular Friedel-Crafts acylation of 3-arylpropanoic acids and their derivatives is the most direct method for their construction.^{2,3} However, the classical Friedel-Crafts transformation generally require the use of at least an equivalent amount of either a Lewis or Brønsted acid.³ Thus, it is desirable to develop a catalytic system for these purposes. Recently, many catalytic methods have been disclosed,3-7 including the Friedel-Crafts reaction,3 Nazarov cyclization,⁴ transition-metal-catalyzed annulation,⁵ and ring-closing metathesis.⁶ Here, we report a novel route to prepare polysubstituted inden-1-ones by goldcatalyzed skeletal rearrangement of 1-[2-(1H-isochromen-3-yl)aryl]ethanones with alcohols (Scheme 1).



Scheme 1 Gold-catalyzed skeletal rearrangement

Gold is one of the most important transition metal catalysts in organic synthesis, and has been widely used in organic synthesis for constructing numerous chemical bonds.^{8,9} Particularly, gold-catalyzed rearrangement reac-

SYNTHESIS 2012, 44, 2049–2057 Advanced online publication: 01.06.2012 DOI: 10.1055/s-0031-1291152; Art ID: SS-2012-H0240-OP © Georg Thieme Verlag Stuttgart · New York tions have attracted significant interest because of their synthetic utility and concept: these transformations are highly efficient and atom-economical and provide opportunities to rapidly construct complex molecules and discover new reactions.^{8,9} Interestingly, this strategy was applied to construct various carbocycles,⁸ including indene derivatives.⁹ To the best of our knowledge, however, skeletal rearrangement of 1-[2-(1*H*-isochromen-3-yl)ar-yl]ethanones with alcohols for the synthesis of inden-1-ones has not been reported.

Our study began with the reaction of 1-[2-(4-methyl-1Hisochromen-3-yl)phenyl]ethanone (1a) with 5 mol% AuBr₃ in THF at 80 °C; however, no reaction was observed (Table 1, entry 1). Gratifyingly, an unexpected skeletal rearrangement reaction took place in the presence of MeOH (2a): the desired rearrangement product, 2-[2-(methoxymethyl)phenyl]-2-methyl-3-methylene-2,3-dihydro-1*H*-inden-1-one (3), was obtained in 10% yield at a loading of 10 equivalents of MeOH (entry 2).¹⁰ In light of the results, MeOH was employed as the solvent, and the vield of product **3** was enhanced sharply to 78% (entry 3). The amount of AuBr₃ was subsequently investigated (entries 3-5): 10 mol% AuBr₃ gave identical results to those of 5 mol% AuBr₃ (entry 4), whereas the yield was lowered to 59% when 2 mol% AuBr₃ was used (entry 5). Notably, the reaction cannot take place without gold catalysts (entry 6). Screening revealed that gold catalysts, such as AuCl₃, HAuCl₄ AuCl, or AuI, have reactivity for the reaction, but they were less effective than AuBr₃ (entries 7– 10). Among the reaction temperature examined, it turned out the reaction at 80 °C gave the best results (entries 3, 11, and 12). However, other metal catalysts, PtBr₂ and PdBr₂, had no effect on the reaction (entries 13 and 14).

With the optimal reaction conditions in hand, the scope of both 1-[2-(1*H*-isochromen-3-yl)aryl]ethanones 1 and alcohols 2 was explored (Table 2).¹⁰ Initially, a number of alcohols 2b–f were investigated for the reaction with 1-[2-(4-methyl-1*H*-isochromen-3-yl)phenyl]ethanone (1a) and AuBr₃ (Table 2, entries 1–5). The results disclosed that both primary and secondary alcohols were suitable for the reaction (entries 1–4), but tertiary alcohol (*t*-BuOH) could not furnish the desired product 8 (entry 5). It was found that BnOH (2g) was also less reactive for the reaction (entry 6). Subsequently, a variety of 1-[2-(1*H*-isochromen-3-

 Table 1
 Screening Conditions for Gold-Catalyzed Skeletal Rearrangement^a

	o la	+ MeOH ∑ 2a		Me
Entry	[M] (mol%)	Solvent	Temp (°C)	Yield (%) ^t
1	$\operatorname{AuBr}_{3}(5)$	THF	80	0
2°	$\operatorname{AuBr}_{3}(5)$	MeOH-THF	80	10
3	$\operatorname{AuBr}_{3}(5)$	МеОН	80	78
4	AuBr ₃ (10)	MeOH	80	79
5	$\operatorname{AuBr}_{3}(2)$	MeOH	80	59
6	-	МеОН	80	0
7	$\operatorname{AuCl}_{3}(5)$	MeOH	80	20
8	$HAuCl_4(5)$	MeOH	80	41
9	AuCl (5)	MeOH	80	30
10	AuI (5)	MeOH	80	33
11	$\operatorname{AuBr}_{3}(5)$	МеОН	100	78
12	$\operatorname{AuBr}_{3}(5)$	MeOH	60	40
13	$PtBr_{2}(5)$	МеОН	80	trace
14	$PdBr_{2}\left(5\right)$	MeOH	80	trace

^a Reaction conditions: **1a** (0.3 mmol), [M], and solvent (2 mL) for 12 h under argon atmosphere.

^b Isolated yield.

^c Amount of MeOH added: 10 equiv.

yl)aryl]ethanones **1b–l** were treated with MeOH (**2a**) and AuBr₃ (entries 7–17). The results disclosed that 1-[2-(1H-isochromen-3-yl)phenyl]ethanone (**1b**) smoothly underwent the rearrangement reaction to afford the correspond-

2-[2-(methoxymethyl)phenyl]-3-methylene-2,3ing dihydro-1H-inden-1-one (10) in 41% yield (entry 7). Interestingly, substrates 1c-l with a methyl group on the 4position of the 1H-isochromene moiety were also successful in the rearrangement reaction (entries 8-17). Moreover, screening revealed that several functional groups, including Me, MeO, OCH₂O, F, Cl, and NO₂ on the aromatic ring of substrates 1 were tolerated well under the optimal conditions. For example, substrates with three Me groups on different aryl rings smoothly underwent the reaction with MeOH and AuBr₃, providing the corresponding products 11 and 12 in good yields (entries 8 and 9). MeO-substituted substrates were also consistent with the optimal conditions: Substrate with two MeO group formed the target product 14 in 86% yield (entry 11), and four MeO-substituted substrate still gave a moderate yield (entry 12). Gratifyingly, a heterocycle and a protected diol, which were introduced into this system, also make this methodology more useful for the preparation of pharmaceuticals and natural products (entry 13). Importantly, substituents, Cl and F, were found to be compatible with the optimal conditions, thereby facilitating additional modifications at the halogenated positions (entries 14-16). Interestingly, substrate 11 with two electron-withdrawing NO2 groups could also react with MeOH and AuBr₃ to give the desired product in moderate yield (entry 17).

As shown in Scheme 2, the reactions some other isochromenes 1 were also investigated in the presence of AuBr₃ and MeOH (2a). The results demonstrated that 2-(1*H*-isochromen-3-yl)benzaldehyde (1m) was not a suitable substrate for the rearrangement reaction: only <5%yield of the desired product 21 was observed by GC-MS analysis (Scheme 2, equation 1). It is noteworthy that a carbonyl group on the 2-position of the 3-phenyl ring in 1*H*-isochromenes is the key for the occurrence of the reaction; no reaction takes place using 3-phenyl-1*H*-isochromene (1n) (Scheme 2, equation 2).

Table 2 Gold-Catalyzed Rearrangement of 1-[2-(1H-Isochromen-3-yl)aryl]ethanones 1 with Alcohols 2^a



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 Table 2
 Gold-Catalyzed Rearrangement of 1-[2-(1H-Isochromen-3-yl)aryl]ethanones 1 with Alcohols 2^a (continued)



 Table 2
 Gold-Catalyzed Rearrangement of 1-[2-(1H-Isochromen-3-yl)aryl]ethanones 1 with Alcohols 2^a (continued)



Entry Substrate 1

0

1d

1e







10

11

12





OMe

1g

0

1h

1i























18 (81)



Table 2Gold-Catalyzed Rearrangement of 1-[2-(1H-Isochromen-3-yl)aryl]ethanones 1 with Alcohols 2^a (continued)

^a Reaction conditions: **1** (0.3 mmol), alcohol **2** (2 mL), and AuBr₃ (5 mol%) at 80 °C for 12 h under argon atmosphere. ^b Isolated yield.



Scheme 2 AuBr₃-catalyzed rearrangement reactions of other isochromenes 1m,n with MeOH (2a)



Scheme 3 Possible mechanism for the gold-catalyzed skeletal rearrangement

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A possible mechanism as outlined in Scheme 3 is proposed.^{8,9} Complexation of substrate **1a** with AuBr₃ yields intermediate A, in which the C=C bond is activated by the AuBr₃ Lewis acid catalyst. Subsequently, the addition of the oxygen atom of the carbonyl group to the C=C bond takes place to offer intermediate **B**, followed by addition of methanol to intermediate **B** to form a six-membered oxocycle intermediate C.^{9e} Intermediate C undergoes the oxygen-migration-ring-opening-cyclization sequence to afford intermediate **D**. Finally, the desired product **3** is obtained from intermediate **D** together with the regeneration of the active gold species.

In summary, we have described a novel and general method for the construction of 2,3-dihydro-1H-inden-1-ones by alcohol-induced gold-catalyzed skeletal rearrangement. Importantly, this rearrangement method provides a new synthetic utility for gold catalysts. Further study of the detailed mechanism and applications of this goldcatalyzed rearrangement transformation in organic synthesis are currently underway in our laboratory.

NMR spectroscopy was performed on a Bruker Avance spectrometer operating at 500 MHz (¹H NMR) and 125 MHz (¹³C NMR). Mass spectrometric analysis was performed using GC-MS technique (Shimadzu GCMS-QP2010) and ESI-Q-TOF (Bruker Micro-QTOF-II. All melting points are uncorrected.

All starting materials were synthesized according to the literature procedure.^{9e} The analytical and spectral data for **1a** are given below.

1-[2-(4-Methyl-1*H*-isochromen-3-yl)phenyl]ethanone (1a) IR (KBr): 2925, 1768 cm⁻¹

¹H NMR (500 MHz, CDCl₃): δ = 7.58 (d, *J* = 7.5 Hz, 1 H), 7.49– 7.46 (m, 1 H), 7.44–7.41 (m, 2 H), 7.31 (t, J = 7.5 Hz, 1 H), 7.22– 7.20 (m, 2 H), 7.05 (d, J = 7.5 Hz, 1 H), 5.07 (s, 2 H), 2.52 (s, 3 H), 1.99 (s, 3 H).

¹³C NMR (125 MHz, CDCl₃): δ = 202.5, 149.6, 141.4, 133.6, 133.5, 131.3, 130.3, 128.8, 128.7, 128.0, 127.5, 126.7, 123.6, 121.0, 109.8, 69.2, 29.4, 13.6.

LRMS (EI, 70 eV): *m/z* (%): 265 (M⁺ + 1, 19), 264 (100), 115 (44).

HRMS (EI): m/z calcd for $C_{18}H_{16}O_2$ (M⁺): 264.1223; found: 264.1226.

Gold-Catalyzed Skeletal Rearrangement Reaction; General Procedure

To a Schlenk tube were added 1-[2-(4-methyl-1H-isochromen-3yl)phenyl]ethanone 1 (0.3 mmol), AuBr₃ (5 mol%), and alcohol 2 (2 mL). Then, the tube was charged with argon and the mixture was stirred at 80 °C (oil bath temperature) for 12 h until complete consumption of starting material as monitored by TLC (eluent: hexane-EtOAc, 10:1) and GC-MS analysis. After completion of the reaction, the mixture was cooled to r.t., diluted with EtOAc (10 mL), and the EtOAc layer was washed with brine $(3 \times 3 \text{ mL})$. The aqueous phase was re-extracted with EtOAc (3×5 mL). The combined organic extracts were dried (Na2SO4) and concentrated under vacuum. The resulting residue was purified by silica gel column chromatography (hexane-EtOAc, 5:1) to afford the desired product.

2-[2-(Methoxymethyl)phenyl]-2-methyl-3-methylene-2,3-dihydro-1*H*-inden-1-one (3)

Yield: 65.9 mg (79%); colorless oil.

IR (KBr): 2928, 1748 cm⁻¹.

¹H NMR (500 MHz, CDCl₃): δ = 7.52–7.49 (m, 3 H), 7.32–7.28 (m, 2 H), 7.18 (d, J = 7.5 Hz, 1 H), 7.10–7.07 (m, 1 H), 6.88 (d, J = 7.0

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Hz, 1 H), 5.64 (s, 1 H), 5.27 (s, 1 H), 4.90 (d, J = 14.5 Hz, 1 H), 4.70 (d, J = 15.0 Hz, 1 H), 3.35 (s, 3 H), 1.59 (s, 3 H).

 13 C NMR (125 MHz, CDCl₃): δ = 202.5, 153.1, 140.5, 140.1, 138.7, 132.9, 129.5, 127.6, 127.0, 125.9, 124.8, 123.7, 121.9, 107.5, 105.2, 64.9, 52.6, 50.4, 21.8.

LRMS (EI, 70 eV): m/z (%) = 278 (M⁺, 4), 231 (100), 202 (40).

HRMS (EI): *m/z* calcd for C₁₉H₁₈O₂ (M⁺): 278.1307; found: 278.1311.

2-[2-(Ethoxymethyl)phenyl]-2-methyl-3-methylene-2,3-dihydro-1*H*-inden-1-one (4)

Yield: 52.6 mg (60%); colorless oil.

IR (KBr): 2926, 1747 cm⁻¹.

¹H NMR (500 MHz, CDCl₃): δ = 7.44–7.41 (m, 3 H), 7.23–7.21 (m, 2 H), 7.12 (s, 1 H), 7.02 (s, 1 H), 6.82 (d, *J* = 7.5 Hz, 1 H), 5.56 (s, 1 H), 5.15 (s, 1 H), 4.78 (d, J = 14.5 Hz, 1 H), 4.61 (d, J = 14.5 Hz, 1 H), 3.78–3.75 (m, 1 H), 3.33–3.30 (m, 1 H), 1.52 (s, 3 H), 1.05 (t, J = 7.0 Hz, 3 H).

 13 C NMR (125 MHz, CDCl₃): $\delta = 207.3$, 153.0, 140.5, 140.0, 138.7, 132.8, 129.5, 127.6, 127.0, 125.9, 124.8, 123.7, 121.9, 107.5, 105.2, 64.9, 52.6, 50.4, 21.8, 14.2.

LRMS (EI, 70 eV): *m*/*z* (%) = 292 (M⁺, 8), 247 (40), 157 (82).

HRMS (EI): *m*/*z* for C₂₀H₂₀O₂ (M⁺): 292.1463; found: 292.1467.

2-Methyl-3-methylene-2-[2-(propoxymethyl)phenyl]-2,3-dihydro-1*H*-inden-1-one (5) Yield: 66.1 mg (72%); colorless oil.

IR (KBr): 2931, 1692 cm⁻¹.

¹H NMR (500 MHz, CDCl₃): $\delta = 7.44 - 7.40$ (m, 3 H), 7.21 - 7.20 (m, 2 H), 7.10 (t, J = 7.5 Hz, 1 H), 7.00 (t, J = 7.0 Hz, 1 H), 6.81 (d, *J* = 7.5 Hz, 1 H), 5.54 (s, 1 H), 5.17 (s, 1 H), 4.80 (d, *J* = 14.5 Hz, 1 H), 4.62 (d, J = 14.5 Hz, 1 H), 3.68 (t, J = 7.5 Hz, 1 H), 3.22 (t, J = 7.5 Hz, 1 H), 1.52 (s, 3 H), 1.46–1.41 (m, 2 H), 0.77 (t, J = 7.5 Hz. 3 H).

¹³C NMR (125 MHz, CDCl₃): δ = 200.8, 153.4, 140.5, 140.2, 133.1, 129.3, 127.6, 127.0, 126.9, 125.8, 124.7, 123.7, 121.7, 107.1, 105.0, 64.8, 64.0, 52.7, 23.1, 22.0, 10.6.

LRMS (EI, 70 eV): m/z (%) = 306 (M⁺, 4), 231 (100), 217 (22).

HRMS (EI): m/z calcd for $C_{21}H_{22}O_2$ (M⁺): 306.1620; found: 306.1624.

2-[2-(Isopropoxymethyl)phenyl]-2-methyl-3-methylene-2,3-dihydro-1*H*-inden-1-one (6)

Yield: 36.7 mg (40%); colorless oil.

IR (KBr): 2919, 1684 cm⁻¹.

¹H NMR (500 MHz, CDCl₂): $\delta = 7.49$ (d, J = 7.5 Hz, 1 H), 7.42 (t, *J* = 8.5 Hz, 2 H), 7.28 (t, *J* = 4.5 Hz, 1 H), 7.22 (t, *J* = 6.0 Hz, 1 H), 7.09 (t, J = 8.5 Hz, 1 H), 6.88 (d, J = 2.5 Hz, 1 H), 6.82 (d, J = 7.5 Hz, 1 H), 5.54 (s, 1H), 5.14 (s, 1 H), 4.84 (d, J = 11.0 Hz, 1 H), 4.62 (d, J = 13.0 Hz, 1 H), 3.28–3.23 (m, 1 H), 1.51 (s, 3 H), 1.49 (d, J = 7.0 Hz, 3 H), 1.46 (d, J = 6.5 Hz, 3 H).

 13 C NMR (125 MHz, CDCl₃): δ = 197.2, 153.1, 140.5, 140.1, 138.8, 132.9, 129.5, 127.6, 127.0, 125.9, 124.8, 123.7, 121.9, 107.5, 105.2, 71.5, 64.9, 52.6, 21.8, 21.3, 20.7.

LRMS (EI, 70 eV): m/z (%) = 306 (M⁺, 16), 247 (28), 231 (100).

HRMS (EI): *m*/*z* calcd for C₂₁H₂₂O₂ (M⁺): 306.1620; found: 306.1624.

2-[2-(Butoxymethyl)phenyl]-2-methyl-3-methylene-2,3-dihydro-1*H*-inden-1-one (7)

Yield: 67.2 mg (70%); colorless oil.

IR (KBr): 2925, 1740 cm⁻¹.

¹H NMR (500 MHz, DMSO- d_6): δ = 7.57–7.55 (m, 1 H), 7.50–7.48 (m, 2 H), 7.33–7.31 (m, 2 H), 7.17 (s, 1 H), 7.09–7.07 (m, 1 H), 6.93 (d, J = 7.5 Hz, 1 H), 5.70 (s, 1 H), 5.36 (s, 1 H), 4.88 (d, J = 15.0 Hz, 1 H), 4.60 (d, J = 15.0 Hz, 1 H), 3.79 (d, J = 9.0 Hz, 1 H), 3.21 (d, J = 9.5 Hz, 1 H), 1.48 (s, 3 H), 1.36–1.32 (m, 2 H), 1.24–1.20 (m, 2 H), 0.77 (t, J = 7.0 Hz, 3 H).

¹³C NMR (125 MHz, DMSO- d_6): $\delta = 202.9$, 153.1, 140.6, 140.4, 139.3, 133.2, 130.1, 128.2, 127.5, 127.4, 126.5, 125.3, 124.3, 107.0, 106.2, 64.5, 61.6, 52.7, 31.9, 22.2, 19.4, 14.2.

LRMS (EI, 70 eV): m/z (%) = 320 (M⁺, 1), 246 (6), 40 (100).

HRMS (EI): m/z calcd for $C_{22}H_{24}O_2$ (M⁺): 320.1776; found: 320.1780.

2-[2-(Methoxymethyl)phenyl]-3-methylene-2,3-dihydro-1*H*-inden-1-one (10)

Yield: 32.5 mg (41%); yellow oil.

IR (KBr): 2927, 1752 cm⁻¹.

¹H NMR (500 MHz, CDCl₃): δ = 7.52–7.49 (m, 3 H), 7.32–7.28 (m, 2 H), 7.19 (t, *J* = 7.5 Hz, 1 H), 7.10–7.07 (m, 1 H), 6.88 (d, *J* = 7.0 Hz, 1 H), 5.64 (s, 1 H), 5.27 (s, 1 H), 4.89 (d, *J* = 14.5 Hz, 1 H), 4.71 (d, *J* = 15.0 Hz, 1 H), 4.41 (s, 1 H), 3.11 (s, 3 H).

¹³C NMR (125 MHz, CDCl₃): δ = 202.5, 153.1, 140.5, 140.1, 138.7, 132.9, 129.5, 127.6, 127.0, 125.9, 124.8, 123.7, 121.9, 107.5, 105.2, 59.0, 52.6, 50.4.

LRMS (EI, 70 eV): m/z (%) = 264 (M⁺, 27), 246 (100), 176 (11).

HRMS (EI): m/z calcd for $C_{18}H_{16}O_2$ (M⁺): 264.1150; found: 264.1154.

2-[2-(Methoxymethyl)-4-methylphenyl]-2,6-dimethyl-3-methylene-2,3-dihydro-1*H***-inden-1-one (11) Yield: 79.9 mg (87%); yellow oil.**

IR (KBr): 2931, 1745 cm⁻¹.

¹H NMR (500 MHz, CDCl₃): δ = 7.45 (d, *J* = 8.0 Hz, 1 H), 7.31– 7.29 (m, 1 H), 7.23 (s, 1 H), 7.03 (s, 1 H), 6.95–6.88 (m, 2 H), 5.47 (s, 1 H), 5.10 (s, 1 H), 4.75 (d, *J* = 14.5 Hz, 1 H), 4.60 (d, *J* = 11.0 Hz, 1 H), 3.27 (s, 3 H), 2.30 (s, 3 H), 2.16 (s, 3 H), 1.48 (s, 3 H).

 ^{13}C NMR (125 MHz, CDCl₃): δ = 203.9, 153.2, 142.8, 139.4, 137.6, 135.5, 132.7, 130.5, 127.8, 126.8, 125.1, 124.1, 121.6, 107.6, 104.0, 64.9, 52.5, 50.5, 21.8, 21.6, 20.9.

LRMS (EI, 70 eV): m/z (%) = 306 (M⁺, 22), 171 (100), 40 (82).

HRMS (EI): *m/z* for C₂₁H₂₂O₂ (M⁺): 306.1620; found: 306.1624.

2-[2-(Methoxymethyl)-5-methylphenyl]-2,5-dimethyl-3-methylene-2,3-dihydro-1*H*-inden-1-one (12) Yield: 73.4 mg (80%); colorless oil.

IR (KBr): 2930, 1744 cm^{-1} .

¹H NMR (500 MHz, DMSO- d_6): δ = 7.40–7.35 (m, 2 H), 7.27–7.22 (m, 2 H), 7.12 (d, J = 8.0 Hz, 1 H), 6.87 (d, J = 8.0 Hz, 1 H), 5.64 (s, 1 H), 5.34 (s, 1 H), 4.84 (d, J = 15.0 Hz, 1 H), 4.55 (d, J = 16.0 Hz, 1 H), 3.19 (s, 3 H), 2.27 (s, 3 H), 2.21 (s, 3 H), 1.43 (s, 3 H).

¹³C NMR (125 MHz, DMSO- d_6): δ = 200.4, 153.1, 140.9, 140.4, 139.8, 136.4, 130.1, 129.0, 127.9, 127.4, 125.3, 124.2, 122.9, 107.3, 106.0, 64.6, 52.8, 50.2, 22.1, 21.6, 21.5.

LRMS (EI, 70 eV): m/z (%) = 306 (M⁺, 2), 273 (14), 40 (100).

HRMS (EI): m/z calcd for $C_{21}H_{22}O_2$ (M⁺): 306.1620; found: 306.1624.

2-[2-(Methoxymethyl)-4,5-dimethylphenyl]-2,5,6-trimethyl-3methylene-2,3-dihydro-1*H***-inden-1-one (13) Yield: 46.1 mg (46%); colorless oil.**

IR (KBr): 2945, 1746 cm⁻¹.

¹H NMR (500 MHz, CDCl₃): δ = 7.50 (s, 1 H), 7.39 (s, 1 H), 7.07 (s, 1 H), 7.00 (s, 1 H), 5.42 (s, 1 H), 5.17 (s, 1 H), 5.05 (d, *J* = 11.5

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Hz, 1 H), 4.75 (d, *J* = 13.5 Hz, 1 H), 3.40 (s, 3 H), 2.29 (s, 3 H), 2.27 (s, 3 H), 2.24 (s, 3 H), 2.20 (s, 3 H), 1.53 (s, 3 H).

 ^{13}C NMR (125 MHz, CDCl₃): δ = 201.5, 155.4, 153.5, 143.5, 139.1, 137.4, 136.7, 136.1, 136.0, 135.8, 135.5, 134.8, 133.9, 106.9, 103.7, 64.8, 52.5, 50.4, 20.2, 20.1, 20.0, 19.8, 19.6.

LRMS (EI, 70 eV): m/z (%) = 334 (M⁺, 9), 297 (100), 149 (21).

HRMS (EI): m/z calcd for $C_{23}H_{26}O_2$ (M⁺): 334.1933; found: 334.1937.

6-Methoxy-2-[4-methoxy-2-(methoxymethyl)phenyl]-2-methyl-3-methylene-2,3-dihydro-1*H***-inden-1-one (14) Yield: 87.2 mg (86%); yellow oil.**

IR (KBr): 2922, 1697 cm⁻¹.

¹H NMR (500 MHz, DMSO- d_6): δ = 7.46 (d, J = 8.5 Hz, 1 H), 7.37 (d, J = 8.5 Hz, 1 H), 7.02 (d, J = 2.0 Hz, 1 H), 6.93–6.91 (m, 1 H), 6.75–6.73 (m, 1 H), 6.51 (d, J = 2.5 Hz, 1 H), 5.49 (s, 1 H), 5.18 (s, 1 H), 4.84 (d, J = 15.0 Hz, 1 H), 4.61 (d, J = 15.0 Hz, 1 H), 3.80 (s, 3 H), 3.64 (s, 3 H), 3.23 (s, 3 H), 1.41 (s, 3 H).

¹³C NMR (125 MHz, DMSO- d_6): $\delta = 202.2$, 159.7, 157.8, 152.8, 140.4, 134.3, 133.2, 132.4, 128.8, 123.6, 116.9, 114.1, 110.0, 108.5, 107.3, 64.8, 56.1, 55.6, 52.4, 50.4, 22.1.

LRMS (EI, 70 eV): *m*/*z* (%) = 338 (M⁺, 19), 291 (88), 166 (42).

HRMS (EI): m/z calcd for $C_{21}H_{22}O_4$ (M⁺): 338.1518; found: 338.1522.

2-[4,5-Dimethoxy-2-(methoxymethyl)phenyl]-5,6-dimethoxy-2methyl-3-methylene-2,3-dihydro-1*H***-inden-1-one (15) Yield: 60.9 mg (51%); white solid; mp 99.6–100.7 °C.**

IR (KBr): 2928, 1701 cm⁻¹.

¹H NMR (500 MHz, CDCl₃): δ = 7.21 (s, 1 H), 7.10 (s, 1 H), 7.09 (s, 1 H), 6.90 (s, 1 H), 5.51 (s, 1 H), 5.23 (s, 1 H), 5.01 (d, *J* = 6.5 Hz, 1 H), 4.78 (d, *J* = 6.5 Hz, 1 H), 3.98 (s, 3 H), 3.91 (s, 3 H), 3.90 (s, 3 H), 3.81 (s, 3 H), 2.88 (s, 3 H), 1.60 (s, 3 H).

 ^{13}C NMR (125 MHz, CDCl₃): δ = 204.6, 155.7, 152.2, 151.6, 148.1, 147.6, 142.0, 130.6, 129.6, 127.4, 112.6, 112.2, 105.5, 104.6, 102.2, 60.4, 57.6, 56.8, 56.3, 56.1, 55.9, 50.9, 22.7.

LRMS (EI, 70 eV): m/z (%) = 398 (M⁺, 12), 217 (100), 40 (72).

HRMS (EI): m/z calcd for $C_{23}H_{26}O_6$ (M⁺): 398.1729; found: 398.1733.

6-[4-(Methoxymethyl)benzo[*d***][1,3]dioxol-5-yl]-6-methyl-7methylene-6,7-dihydro-5***H***-indeno[5,6-d][1,3]dioxol-5-one (16)** Yield: 97.7 mg (89%); white solid; mp 102.5–103.4 °C.

IR (KBr): 2930, 1752 cm⁻¹.

¹H NMR (500 MHz, CDCl₃): δ = 7.12 (s, 1 H), 7.07 (s, 1 H), 7.06 (s, 1 H), 6.85 (s, 1 H), 6.07 (d, *J* = 2.5 Hz, 2 H), 5.92 (d, *J* = 1.5 Hz, 1 H), 5.89 (d, *J* = 1.0 Hz, 1 H), 5.46 (s, 1 H), 4.79 (s, 1 H), 3.79 (d, *J* = 13.0 Hz, 1 H), 3.73 (d, *J* = 13.0 Hz, 1 H), 2.88 (s, 3 H), 1.54 (s, 3 H).

 ^{13}C NMR (125 MHz, CDCl₃): δ = 203.6, 154.6, 151.7, 150.2, 146.9, 146.8, 144.3, 131.8, 131.0, 129.3, 114.7, 109.2, 106.0, 102.7, 102.5, 101.2, 100.7, 71.7, 57.5, 57.2, 26.6.

LRMS (EI, 70 eV): m/z (%) = 366 (M⁺, 7), 335 (42), 40 (52).

HRMS (EI): m/z calcd for $C_{21}H_{18}O_6$ (M⁺): 366.1103; found: 366.1107.

6-Fluoro-2-[4-fluoro-2-(methoxymethyl)phenyl]-2-methyl-3methylene-2,3-dihydro-1*H***-inden-1-one (17) Yield: 70.7 mg (75%); light yellow oil.**

IR (KBr): 2920, 1715 cm⁻¹.

¹H NMR (500 MHz, DMSO- d_6): δ = 7.63–7.61 (m, 1 H), 7.53–7.50 (m, 1 H), 7.40–7.38 (m, 1 H), 7.22–7.18 (m, 1 H), 7.03–7.00 (m, 1

H), 6.84–6.82 (m, 1 H), 5.67 (s, 1 H), 5.35 (s, 1 H), 4.89 (d, *J* = 15.5 Hz, 1 H), 4.61 (d, *J* = 15.5 Hz, 1 H), 3.24 (s, 3 H), 1.44 (s, 3 H).

¹³C NMR (125 MHz, DMSO- d_6): $\delta = 202.3$, 151.8, 141.0, 136.7, 136.1, 135.5 (d, J = 7.0 Hz, 1 C), 129.9 (d, J = 8.0 Hz, 1 C), 124.5 (d, J = 9.0 Hz, 1 C), 117.8 (d, J = 23.0 Hz, 1 C), 114.7 (d, J = 21.0 Hz, 1 C), 112.5 (d, J = 23.0 Hz, 1 C), 110.9, 110.7, 106.9, 106.5, 64.3, 52.7, 50.5, 22.2.

LRMS (EI, 70 eV): m/z (%) = 314 (M⁺, 4), 267 (18), 40 (100). HRMS (EI): m/z for $C_{19}H_{16}F_2O_2$ (M⁺): 314.1118; found: 314.1122.

5-Fluoro-2-[5-fluoro-2-(methoxymethyl)phenyl]-2-methyl-3methylene-2,3-dihydro-1*H*-inden-1-one (18)

Yield: 76.3 mg (81%); colorless oil.

IR (KBr): 2919, 1716 cm⁻¹.

¹H NMR (500 MHz, DMSO- d_6): δ = 7.63–7.61 (m, 1 H), 7.53–7.50 (m, 1 H), 7.40–7.38 (m, 1 H), 7.20–7.18 (m, 1 H), 7.03–6.99 (m, 1 H), 6.84–6.82 (m, 1 H), 5.67 (s, 1 H), 5.35 (s, 1 H), 4.90 (d, *J* = 15.5 Hz, 1 H), 4.61 (d, *J* = 15.5 Hz, 1 H), 3.24 (s, 3 H), 1.44 (s, 3 H).

¹³C NMR (125 MHz, DMSO-*d*₆): δ = 201.2, 151.7, 140.9, 136.7 (d, *J* = 77.0 Hz, 1 C), 135.5, 135.4, 130.0, 124.5 (d, *J* = 9.0 Hz, 1 C), 117.8 (d, *J* = 23.0 Hz, 1 C), 114.7 (d, *J* = 21.5 Hz, 1 C), 112.5 (d, *J* = 23.0 Hz, 1 C), 110.9, 110.7, 106.9, 106.5, 64.3, 52.7, 50.5, 22.2.

LRMS (EI, 70 eV): m/z (%) = 314 (M⁺, 6), 267 (22), 40 (100).

HRMS (EI): m/z calcd for $C_{19}H_{16}F_2O_2$ (M⁺): 314.1118; found: 314.1122.

6-Chloro-2-[4-chloro-2-(methoxymethyl)phenyl]-2-methyl-3methylene-2,3-dihydro-1*H*-inden-1-one (19) Yield: 62.5 mg (60%); light yellow oil.

IR (KBr): 2919, 1720 cm⁻¹.

¹H NMR (500 MHz, CDCl₃): $\delta = 8.23-8.17$ (m, 2 H), 8.08 (d, J = 17.5 Hz, 2 H), 7.51 (d, J = 8.0 Hz, 1 H), 7.40 (d, J = 8.5 Hz, 1 H), 5.65 (s, 1 H), 5.17 (s, 1 H), 5.13 (d, J = 16.0 Hz, 1 H), 5.06 (d, J = 13.0 Hz, 1 H), 3.00 (s, 3 H), 1.67 (s, 3 H).

 ^{13}C NMR (125 MHz, CDCl₃): δ = 197.2, 153.1, 140.5, 140.1, 138.8, 132.9, 129.5, 127.6, 127.0, 126.9, 125.9, 124.8, 123.7, 121.9, 105.2, 64.9, 52.6, 50.4, 21.8.

LRMS (EI, 70 eV): m/z (%) = 346 (M⁺, 25), 316 (100), 276 (22).

HRMS (EI): m/z calcd for $C_{19}H_{16}Cl_2O_2$ (M^+): 346.0527; found: 346.0531.

2-[2-(Methoxymethyl)-4-nitrophenyl]-2-methyl-3-methylene-6nitro-2,3-dihydro-1*H*-inden-1-one (20)

Yield: 66.2 mg (60%); yellow oil.

IR (KBr): 2925, 1745 cm⁻¹.

¹H NMR (500 MHz, CDCl₃): $\delta = 8.29-8.23$ (m, 2 H), 8.15 (d, J = 17.0 Hz, 2 H), 7.57 (d, J = 8.5 Hz, 1 H), 7.45 (d, J = 8.0 Hz, 1 H), 5.65 (s, 1 H), 5.23 (s, 1 H), 5.20 (d, J = 16.0 Hz, 1 H), 5.13 (d, J = 13.5 Hz, 1 H), 3.06 (s, 3 H), 1.73 (s, 3 H).

 ^{13}C NMR (125 MHz, CDCl₃): δ = 201.3, 149.5, 149.0, 146.0, 142.4, 142.3, 141.5, 132.1, 124.0, 123.9, 123.8, 123.2, 116.9, 110.7, 110.6, 60.4, 51.1, 50.2, 21.0.

LRMS (EI, 70 eV): *m*/*z* (%) = 368 (M⁺, 9), 202 (23), 40 (100).

HRMS (EI): m/z calcd for $C_{19}H_{16}N_2O_6$ (M⁺): 368.1008; found: 368.1012.

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Supporting Information for this article is available online at http://www.thieme-connect.com/ejournals/toc/synthesis.

References

- (1) For selected papers, see: (a) de Solms, S. J.; Woltersdorf, O. W. Jr.; Cragoe, E. J. Jr.; Watson, L. S.; Fanelli, G. M. Jr. J. Med. Chem. 1978, 21, 437. (b) Dolling, U.-H.; Davis, P.; Grabowski, E. J. J. J. Am. Chem. Soc. 1984, 106, 446. (c) Lopes, L. M. X.; Yoshida, M.; Gottlieb, O. R. Phytochemistry 1984, 23, 2021. (d) Anstead, G. M.; Wilson, S. R.; Katzenellenbogen, J. A. J. Med. Chem. 1989, 32, 2163. (e) Hajela, K.; Kapil, R. S. Ind. J. Chem., Sect. B: Org. Chem. Incl. Med. Chem. 1995, 34, 361. (f) Sugimoto, H.; Iimura, Y.; Yamanishi, Y.; Yamatsu, K. J. Med. Chem. 1995, 38, 4821. (g) Harrowven, D. C.; Newman, N. A.; Knight, C. A. Tetrahedron Lett. 1998, 39, 6757. (h) Russell, M. A.; Chai, C. L. L.; Wardlaw, J. H.; Elix, J. A. J. Nat. Prod. 2000, 63, 129. (i) Nagle, D. G.; Zhou, Y.-D.; Park, P. U.; Paul, V. J.; Rajbhandari, I.; Duncan, C. J. G.; Pasco, D. S. J. Nat. Prod. 2000, 63, 1431. (j) Park, C. H.; Siomboing, X.; Yous, S.; Gressier, B.; Luyckx, M.; Chavatte, P. Eur. J. Med. Chem. 2002, 37, 461. (k) Ito, T.; Tanaka, T.; Iinuma, M.; Nakaya, K.; Takahashi, Y.; Sawa, R.; Murata, J.; Darnaedi, D. J. Nat. Prod. 2004, 67, 932. (1) McDevitt, R. E.; Malamas, M. S.; Manas, E. S.; Unwalla, R. J.; Xu, Z. B.; Miller, C. P.; Harris, H. A. Bioorg. Med. Chem. Lett. 2005, 15, 3137. (m) Grimsdale, A. C.; Müllen, K. Angew. Chem. Int. Ed. 2005, 44, 5592. (n) Wessig, P.; Teubner, J. Synlett 2006, 1543. (o) Alcalde, E.; Mesquida, N.; López-Pérez, S.; Frigola, J.; Mercè, R. J. Med. Chem. 2009, 52, 675. (p) Zhu, X.; Tsuji, H.; Nakabayashi, K.; Ohkoshi, S.-i.; Nakamura, E. J. Am. Chem. Soc. 2011, 133, 16342.
- (2) (a) Larock, R. Comprehensive Organic Transformations, 2nd ed.; Wiley-VCH: New York, 1999, 1422–1433.
 (b) Floyd, M. B.; Allen, G. A. Jr. J. Org. Chem. 1970, 35, 2647. (c) Martens, H.; Hoornaert, G. Synth. Commun. 1972, 2, 147. (d) Martens, H.; Hoornaert, G. Tetrahedron 1974, 30, 3641. (e) Johnston, K. M.; Shotter, R. G. Tetrahedron 1974, 30, 4059. (f) Shotter, R. G.; Johnston, K. M.; Jones, J. F. Tetrahedron 1978, 34, 741. (g) Galatsis, P.; Manwell, J. J.; Blackwell, J. M. Can. J. Chem. 1994, 72, 1656.
- (3) For selected papers, see: (a) Cui, D.-M.; Zhang, C.; Kawamura, M.; Shimada, S. *Tetrahedron Lett.* 2004, 45, 1741. (b) Fillion, E.; Fishlock, D.; Wilsily, A.; Goll, J. M. *J. Org. Chem.* 2005, 70, 1316; and references cited therein.
- (4) For pioneering papers, see: (a) Lawrence, N. J.; Armitage, E. M. S.; Greedy, B.; Cook, D.; Ducki, S.; McGown, A. T. *Tetrahedron Lett.* 2006, *47*, 1637. (b) Yin, W.; Ma, Y.; Xu, J.; Zhao, Y. J. Org. Chem. 2006, 71, 4312.
- (5) (a) Liebeskind, L. S.; Gasdaska, J. R.; McCallum, J. S. J. Org. Chem. 1989, 54, 669. (b) Tao, W.; Silverberg, L. J.; Rheingold, A. L.; Heck, R. F. Organometallics 1989, 8, 2550. (c) Larock, R. C.; Doty, M. J.; Cacchi, S. J. Org. Chem. 1993, 58, 4579. (d) Padwa, A.; Austin, D. J.; Gareau, Y.; Kassir, J. M.; Xu, S. L. J. Am. Chem. Soc. 1993, 115, 2637. (e) Kokubo, K.; Matsumasa, K.; Miura, M.; Nomura, M. J. Org. Chem. 1996, 61, 6941. (f) Vincente, J.; Abad, J.-A.; Gil-Rubio, J. Organometallics 1996, 15, 3509. (g) Fukuyama, T.; Chatant, N.; Kakiuchi, F.; Murai, S. J. Org. Chem. 1997, 62, 5647. (h) Gevorgyan, V.; Quan, L. G.; Yamamoto, Y. Tetrahedron Lett. 1999, 40, 4089. (i) Quan, L. G.; Gevorgyan, V.; Yamamoto, Y. J. Am. Chem. Soc. 1999, 121, 3545. (j) Yamabe, H.; Mizuno, A.; Kusama, H.; Iwasawa, N. J. Am. Chem. Soc. 2005, 127, 3248.
- (6) For selected papers, see: (a) Suzuki, T.; Ohwada, T.; Shudo, K. J. Am. Chem. Soc. 1997, 119, 6774. (b) Ready, T. E.;

Chien, J. C. W.; Rausch, M. D. *J. Organomet. Chem.* **1999**, *583*, 11. (c) Clive, D. L. J.; Sannigrahi, M.; Hisaindee, S. *J. Org. Chem.* **2001**, *66*, 954. (d) Kerr, D. J.; Metje, C.; Flynn, B. L. *Chem. Commun.* **2003**, 1380. (e) Coyanis, E. M.; Panayides, J.-L.; Fernandes, M. A.; de Koning, C. B.; van Otterlo, W. A. L. *J. Organomet. Chem.* **2006**, *691*, 5222; and references cited therein.

(7) For other methods, see: (a) Marvel, C. S.; Hinman, C. W. J. Am. Chem. Soc. 1954, 76, 5435. (b) Wessig, P.; Glombitza, C.; Muller, G.; Teubner, J. J. Org. Chem. 2004, 69, 7582; and references cited therein. (c) Petrignet, J.; Roisnel, T.; Grée, R. Chem.-Eur. J. 2007, 13, 7374; and references cited therein.

(8) For selected recent reviews, see: (a) Hashmi, A. S. K.; Hutchings, G. J. Angew. Chem. Int. Ed. 2006, 45, 7896.
(b) Hashmi, A. S. K. Chem. Rev. 2007, 107, 3180.
(c) Jiménez-Núñez, E.; Echavarren, A. M. Chem. Rev. 2008, 108, 3326. (d) Gorin, D. J.; Sherry, B. D.; Toste, F. D. Chem. Rev. 2008, 108, 3351. (e) Michelet, V.; Toullec, P. Y.; Genêt, J.-P. Angew. Chem. Int. Ed. 2008, 47, 4268. (f) Li, Z.; Brouwer, C.; He, C. Chem. Rev. 2008, 108, 3239.
(g) Fürstner, A. Chem. Soc. Rev. 2009, 38, 3208. (h) Lee, S. I.; Chatani, N. Chem. Commun. 2009, 371. (i) Shapiro, D.; Toste, F. D. Synlett **2010**, 675. (j) Hashmi, A. S. K.; Bührle, M. Aldrichimica Acta **2010**, 43, 27. (k) Wang, S.; Zhang, G.; Zhang, L. Synlett **2010**, 692. (l) Hashimi, A. S. K. Angew. Chem. Int. Ed. **2010**, 49, 5232. (m) Corma, A.; Leyva-Pérez, A.; Sabater, M. J. Chem. Rev. **2011**, 111, 1657. (n) Bandini, M. Chem. Soc. Rev. **2011**, 40, 1358. (o) Leyva-Pérez, A.; Corma, A. Angew. Chem. Int. Ed. **2012**, 51, 614.

- (9) (a) Hashmi, A. S. K.; Frost, T. M.; Bats, J. W. J. Am. Chem. Soc. 2000, 122, 11553. (b) Fürstner, A.; Hannen, P. Chem.– Eur. J. 2006, 12, 3006. (c) Dubé, P.; Toste, D. J. Am. Chem. Soc. 2006, 128, 12062. (d) Peng, L.; Zhang, X.; Zhang, S.; Wang, J. J. Org. Chem. 2007, 72, 1192. (e) Hashmi, A. S. K.; Schafer, S.; Wölfle, M.; Gil, C. D.; Fischer, P.; Laguna, A.; Blanco, M. C.; Gimeno, M. C. Angew. Chem. Int. Ed. 2007, 46, 6184. (f) Leyva-Pérez, A.; Corma, A. J. Org. Chem. 2009, 74, 2067. (g) Jagdale, A. R.; Youn, S. W. Eur. J. Org. Chem. 2011, 3904. (h) Gómez-Suárez, A.; Ramón, R. S.; Songis, O.; Slawin, A. M. Z.; Cazin, C. S. J.; Nolan, S. P. Organometallics 2011, 30, 5463. (i) Corma, A.; Ruiz, V. R.; Leyva-Pérez, A.; Sabater, M. J. Adv. Synth. Catal. 2010, 352, 1701.
- (10) The X-ray single-crystal diffraction analysis of **16** is given in the Supporting Information.