

Palladium-Catalyzed Late-Stage ortho-C-H Bond Aroylation of Anilines Using 4-Methoxy-2-pyridinyl as a Removable Directing Group

Jean-Ho Chu,*^{,†}[®] Meng-Fan Chiang,[‡] Chin-Wei Li,[‡] Zhe-Hong Su,[†] Shao-Chi Lo,[‡] and Ming-Jung Wu*^{,‡}©

[†]Department of Applied Science, National Taitung University, Taitung 95092, Taiwan [‡]Department of Chemistry, National Sun Yat-sen University, Kaohsiung 804, Taiwan

S Supporting Information

ABSTRACT: A synthetic methodology for the late-stage ortho-C-H bond aroylation of anilines with aryl aldehydes led to a variety of ortho-aroylated anilines by the use of palladium(II) acetate, tert-butyl hydroperoxide, and 1,4dioxane as the catalyst, oxidant, and solvent, respectively, is presented. An N-phenylpyridin-2-amine palladacycle was isolated and characterized by X-ray crystallography. Controlled experiments, radical trapping experiments, and the experiments of the kinetic isotope effect were undertaken to support the proposed reaction mechanism. Syntheses of 2aminobenzophenone and 9(10H)-acridanone based on the developed methodology were successfully demonstrated.

INTRODUCTION

In the past decade, transition-metal-catalyzed C-H bond functionalization has seen an increased interest in organic synthesis.¹ In comparison with conventional methods, the synthetic strategy can efficiently reduce reaction steps and the formation of byproducts, and the preactivation of starting substrates is not required. Therefore, the way of C-H bond functionalization enables molecular synthesis to become more efficient and simpler. To date, a number of research works have been conducted in the literature.²

On the other hand, the so-called directing group often plays an important role in the transition-metal-catalyzed C-H bond functionalization.³ It is not only able to promote the reactivity but also to control the regioselectivity for the type of reaction, such as the ortho-, meta-, and para-position of a phenyl ring. The pyridinyl is believed to be the most common directing group due to its good affinity with transition metals, especially for the palladium ion. Therefore, it is often used in our studies with respect to the palladium-catalyzed selective ortho-C-H bond functionalization.⁵

In view of the importance of benzophenones in organic synthesis⁶ and biological activity,⁷ organic chemists are devoted to developing versatile synthetic methods to target this molecular skeleton. So far, numerous methodologies for the synthesis of benzophenones have been conducted in literature, especially for the ortho cross-dehydrogenative coupling (CDC) of arenes and aryl aldehydes.^{5a,8} In these methods, the palladium ion and tert-butyl hydroperoxide (TBHP) have been examined to be the best metal catalyst and



a radical initiator, respectively, in terms of the synthetic practicability. Moreover, various directing groups, such as the pyridinyl, acetyl, carbamate, isoxazole, and so on were involved in these reported reactions.

In 2015, we successfully demonstrated that the palladiumcatalyzed ortho-C-H bond aroylation of phenols with aryl aldehydes using the pyridinyl as a directing group. The reaction was undertaken via the intermolecular C_{Ar} -H/ C_{Aldehvde}-H bond cross-coupling to afford a variety of 2hydroxybenzophenones followed by the removal of the pyridinyl group.^{5a} Moreover, the synthesis of 1-hydroxy-9Hfluoren-9-ones⁹ was achieved by the intramolecular C_{Ar} -H/ CAr-H bond cross-coupling reaction of 2-hydroxybenzophenones. Finally, this work encouraged us to continuously study the palladium-catalyzed ortho-C-H bond aroylation of anilines.

In the presented study, the palladium-catalyzed ortho-C-H bond aroylation of 4-methoxy-N-arylpyridin-2-amines with aryl aldehydes in the presence of TBHP was undertaken to afford a variety of ortho-aroylated 4-methoxy-N-arylpyridin-2-amines in modest to good yields. Meanwhile, a possible reaction mechanism was presented based on the investigation of controlled and radical trapping experiments as well as the experiments of the kinetic isotope effect. Toward the synthetic application, 2-aminobenzophenone can be given by removing the pyridinyl group of ortho-phenylated 4-methoxy-N-phenyl-

Received: February 22, 2019

Aroylation of 1

Article

		Ph-C (1.5 10 mol 0 TBHP R ² N 1,4- 120	CHO (2a) 5 equiv) % Pd(OAc) ₂ (6 equiv) dioxane °C 24 h		R ¹ N O Ph R ² N O		
entry	substrate	Product 3 , 4 (ratio) ^a	yield (%) ^b	Ph 3 entry	substrate	Product 3,4 (ratio) ^a	yield (%) ^b
1	H H [·] N 1-H/H	3-H/H , $R^1 = H$; $R^2 = H$	0	6		$3a + 4a (67:33), R^1$ = OMe; R ² = Boc	87
2	Me N H ^{-N} 1-Me/H	3-Me/H , $R^1 = Me$; $R^2 = H$	0	7	MeO NeO NeO NeO NeO NeO NeO NeO NeO NeO N	3-OMe/Moc + 4- OMe/Moc (65:35), R ¹ = OMe; R ² = Moc	88
3	MeO N H ^{-N} 1-OMe/H	3-OMe/H , R ¹ = OMe; R ² = H	0	8	MeO N N O N O N O N O N O N O N	3-OMe/Piv + 4- OMe/Piv (77:23), R ¹ = OMe; R ² = Pivaloyl	45
4	N O 1-H/Boc	3-H/Boc , R ¹ = H; R ² = Boc	trace	9	MeO Me N O 1-OMe/Ac	3-OMe /Ac, $R^1 =$ OMe; $R^2 =$ Acetyl	33
5		3-Me/Boc , R ¹ = Me; R ² = Boc	40	10	MeO Ph N 1-OMe/Bn	3-OMe/Bn, $R^1 =$ OMe; $R^2 = Bn$	28

^aProduct ratio was determined as an average of three runs by GC-FID. ^bProduct yield was determined as an average of three runs by the isolation.

pyridin-2-amine. In addition, 9(10H)-acridanone¹⁰ can be directly synthesized through the palladium-catalyzed intramolecular C-H/N-H bond cross-coupling of 2-aminobenzophenone or ortho-phenylated 4-methoxy-N-phenylpyridin-2-amine followed by the removal of the pyridinyl group.

RESULTS AND DISCUSSION

Reactivity Investigation of the Directing Group and N-Protecting Group for the Palladium-Catalyzed ortho-C-H Bond Aroylation of 1. At the beginning of this study, we first carried out the reaction of N-phenylpyridin-2-amine (1-H/H) with benzaldehyde (2a) in the presence of palladium(II) acetate and TBHP as the catalyst and a radical initiator, respectively. 1,4-Dioxane was the solvent and the reaction was held at 120 °C for 24 h. However, none of the

desired product 3-H/H was found under the above reaction conditions (entry 1, Table 1). On the basis of our previous experience, the directing group bearing an electron-donating group (e.g., methyl or methoxy) can significantly promote the reactivity of the palladium-catalyzed site-selective (e.g., ortho) C-H bond activation.^{5b} Therefore, we turned to employ compounds 1-Me/H and 1-OMe/H as new starting substrates by using 4-methyl and 4-methoxy as the directing group, respectively. However, there was still none of the desired orthoaroylated products 3-Me/H and 3-OMe/H obtained from the reaction of 1-Me/H or 1-OMe/H with 2a, and the starting substrate (i.e., 1-Me/H or 1-OMe/H) was recovered (entries 2-3, Table 1). These results made us consider the role of the free amino group of 1 in the reaction. Thus, we tagged a protecting group, tert-butyloxycarbonyl (Boc) on the secondary amino groups of 1-H/H, 1-Me/H, and 1-OMe/H led to 1-H/Boc, 1-Me/Boc, and 1-OMe/Boc. The subsequent *ortho*-C-H bond aroylation of *N*-Boc protected substrates 1-H/Boc, 1-Me/Boc, and 1-OMe/Boc with 2a was carried out under the aforementioned reaction conditions. The experimental results showed that the desired *ortho*-benzoylated products 3-H/Boc, 3-Me/Boc, and 3a (accompanied dibenzoylated product 4a) were generated in the trace, 40%, and 87% yield, respectively (entries 4–6, Table 1). In other words, the electronic property of the directing group (i.e., pyridinyl) is important for the reaction, whereas the *N*-protecting group (e.g., Boc) is also critical for this case.

On the basis of these above results, we attempted to understand the influence of the N-protecting group of 1 in the presented reaction. Therefore, a series of N-protecting groups, such as methyloxycarbonyl (Moc), tert-butylcarbonyl (Piv), acetyl (Ac), and benzyl (Bn) were introduced to 1-OMe/H. Then, various N-protecting substrates 1-OMe/Moc, 1-OMe/ Piv, 1-OMe/Ac, and 1-OMe/Bn was generated and further examined under the aforementioned reaction conditions. Experimental results addressed that the yield of orthobenzoylated products 3-OMe/Moc, 3-OMe/Piv, 3-OMe/Ac, and 3-OMe/Bn accompanied dibenzoylated products 4-OMe/ Moc, 4-OMe/Piv, and 4-OMe/Ac was determined to be 88%, 45%, 33%, and 28% yields, respectively (entries 7-10, Table 1). Herein, we chose Boc as the optimal N-protecting group for the palladium-catalyzed ortho-C-H bond aroylation of 1 with 2. This is because Boc is relatively easy to remove compared to Moc. By varying the N-protecting group of 1-OMe/H, the oxygen atom (not the one on the carbonyl group) of Boc or Moc is critical for the reaction. This might be due to the electrostatic repulsion between the oxygen atom of the N-protecting group and the nitrogen atom of the pyridinyl group. The phenomenon forces the coordination of the palladium ion and pyridine to orient to the ortho-C-H bond as well as prevents the interaction between the amine and TBHP. These two factors might eventually result in the occurrence of the *ortho*-C–H bond aroylation of **1-OMe**/**Boc** in good yields. Moreover, the steric effect of the N-protecting group might also potentially provide the influence in the reaction (see entries 8-10, Table 1).

Optimization of the Reaction Conditions. According to the reactivity investigation of the directing group and *N*-protecting group, both of Boc and 4-methoxy-2-pyridinyl group were eventually chosen as the optimal *N*-protecting group and directing group for the palladium-catalyzed *ortho*-C-H bond aroylation of anilines 1 with aryl aldehyde 2. Next, the reaction of 1a with 2a in the presence of palladium(II) acetate was examined by the variation of the oxidant, solvent, the amount of reagents, and the reaction temperature (see Table S1).

First of all, a variety of oxidants (6 equiv) were screened in the presence of 10 mol % $Pd(OAc)_2$ using 1,4-dioxane as the reaction solvent in which these reactions were heated at 120 °C for 24 h (entries 1–10 and 18, Table S1). The screening results showed that the anticipated product 3a was generated only by the use of TBPB (*tert*-butyl peroxybenzoate) and TBHP as the oxidant. The product ratio (yield) of 3a and 4a was determined to be >99:1 (35%) and 67:33 (87%), respectively (entries 1 and 18, Table S1). In addition to the oxidant, we also evaluated the solvent effect in the presence of 10 mol % $Pd(OAc)_2$ and 6 equiv TBHP (entries 11–21, Table S1). The results of solvent screening showed that 1,4-dioxane still acted as the best solvent for the reaction. Then, the reaction temperature was varied from 80 to 140 °C to give 120 °C as the best temperature (entry 18 and 22–24, Table S1). Finally, the equivalent of TBHP was examined from 0 to 7 (entries 25–29, Table S1) to indicate 6 equiv as the optimal amount in which a mixture of **3a** and **4a** was isolated in 87% total yield with product ratio 67:33 (entry 18, Table S1). On the basis of the optimized experiments, we finally decided to use the following reaction conditions for the *ortho*-C–H bond aroylation of **1**: 10 mol % Pd(OAc)₂, 6 equiv TBHP, and 1.5 equiv aryl aldehyde **2** in 1,4-dioxane heated at 120 °C for 24 h.

Substrate Scope. With the optimized reaction conditions in hand, we subsequently carried out the demonstration of substrates for the palladium-catalyzed *ortho*-C-H bond aroylation of **1a** with a variety of aryl aldehyde **2** (Table 2).



MeO Boc ^{-N.}	10 mol % Pd(OAc TBHP (6 equiv) 3 4 R 120 °C , 24 h 2 (1.5 equiv)	$ \begin{array}{c} D_{2} \\ B_{0C} \\ \end{array} \\ \begin{array}{c} \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\$	
entry	aldehyde (2)	product 3, 4 $(ratio)^a$	yield (%) ^b
1	$\mathbf{R} = \mathbf{H} \ (\mathbf{2a})$	3a + 4a (68:32)	87
2	R = 4-OMe $(2b)$	3b + 4b (67:33)	89
3	$R = 4-NO_2 (2c)$	not detected	0
4	$\mathbf{R} = 4 - \mathbf{CF}_3 \ (\mathbf{2d})$	3d	24
5	R = 4-Ac (2e)	3e	44
6	R = 2-F (2f)	3f + 4f (98:2)	58
7	$\mathbf{R} = 3 - \mathbf{F} \ (\mathbf{2g})$	3g + 4g (88:12)	84
8	R = 4-F (2h)	3h + 4h (90:10)	86
9	R = 2-Cl (2i)	3i	39
10	R = 3-Cl (2j)	3j + 4j (98:2)	57
11	R = 4-Cl (2k)	3k + 4k (95:5)	62
12	R = 3-Br (2l)	31	31
13	R = 4-Br (2m)	3m	37
14	R = 3,5-diMe (2n)	3n + 4n (78:22)	88
15	R = 2,4-diMe (20)	30 + 40 (88:12)	61
16	R = 3,5-diOMe (2p)	3p	35
17	R = 2,4-diOMe (2q)	3q + 4q (90:10)	29
18	R = 3-CHO $(2r)$	3r	45
a .			

^{*a*}Product ratio was determined as an average of three runs by GC-FID. ^{*b*}Product yield was determined as an average of three runs by the isolation.

A series of *ortho* monoaroylated products 3a,b,f-h,j,k,n,o,q (major) as well as diaroylated products 3a,b,f-h,j,k,n,o,q (minor) were obtained as following product ratios and yields 3a:4a = 68:32 (87%), 3b:4b = 67:33 (89%), 3f:4f = 98:2 (58%), 3g:4g = 88:12 (84%), 3h:4h = 90:10 (86%), 3j:4j = 98:2 (57%), 3k:4k = 95:5 (64%), 3n:4n = 78:22 (88%), 3o:4o = 88:12 (61%), and 3q:4q = 90:10 (29%) (entries 1-2, 5-8, 10-11, 14-15, and 17, Table 2). On the other hand, only monoaroylated products 3d,e,i,l,m,p,r were isolated in 24-45% yields by the reaction of 1a with 2d,e,i,l,m,p,r (entries 4-5, 9, 12-13, 16, and 18, Table 2), whereas none of the anticipated product 3d was generated (entry 3, Table 2). These experimental results show that aryl aldehyde 2 bearing an electron-donating group (e.g., OMe) usually exhibits better reactivity than that of the electron-withdrawing group (e.g.,

 NO_2) in the presented reaction. This might be due to the electronic donating effect of the aroyl group could stabilize the palladium intermediate (see Figure 3). By contrast, the electronic withdrawing effect may destabilize it. Moreover, the steric hindrance of arvl aldehvdes 2 seems to potentially offer the influence on the reactivity, which resulted in low product yields. For instance, ortho-substituted aryl aldehydes **2f**, **2i**, **2o**, and **2q** (see entries 6, 9, 15 and 17, Table 2).¹¹ In addition, when isophthalaldehyde (2r) which contains two formyl groups at C1 and C3-position of benzene ring was used, none of the dual C-H/C-H bond coupling product was given. Only monoaroylated product 3r was isolated in 45% yield (entry 18, Table 2). Finally, the structure of dibenzoylated N-Boc protected 4-methoxy-N-phenylpyridin-2-amine (4a) was secured by X-ray crystallography as shown as Figure 1a.¹²



Figure 1. An ORTEP drawing of (a) compound 4a and (b) compound 5h.¹² All hydrogen atoms are omitted for clarity.

In addition to substrate 1a, we also conducted the reaction of 1b-i with 2a under the optimized reaction conditions to afford a variety of *ortho* monobenzoylated *N*-Boc protected 4methoxy-*N*-arylpyridin-2-amine 5 (a major product) and dibenzoylated products 6 (a minor product was produced in some cases) in 39–94% yields. These experimental results are summarized in Table 3. On the basis of the comparison of

Table 3. ortho-C-H Bond Benzoylation of 1 with 2a

MeO N Boc ^{-N} 1	10 mol % Pd(OA TBHP (6 equiv 3 2a 4 (1.5 equiv) 120 °C , 24 h R	MeO Boc ^{-N} Hen 5 ^R	N O Ph Boc N O Ph Boc R Boc R
entry	substrate (1)	product 5, 6 (ratio) ^a	yield (%) ^b
1	R = 2-OMe (1b)	5b	91
2	R = 3-OMe (1c)	5c	83
3	R = 4-OMe $(1d)$	5d + 6d (61:39)	93
4	R = 4-F (1e)	5e + 6e (71:29)	74
5	R = 3-F (1f)	5f	57
6	R = 2-F (1g)	5g	94
7	R = 4-Cl (1h)	5h + 6h (92:8)	79
8	R = 4-Br (1i)	5i	39

^{*a*}Product ratio was determined as an average of three runs by GC-FID. ^{*b*}Product yield was determined as an average of three runs by the isolation.

product yields, the substituent electronic effect in the *ortho*-C– H bond aroylation of substrates **1b**–i seems not to apparently influence the reactivity. The structure of *ortho* monobenzoylated product **5h** was eventually secured by X-ray crystallography as shown as Figure 1b.¹² In addition, *ortho* dibenzoylated products **6d**, **6e**, and **6h** were isolated in 39%, 29%, and 8% product ratios, respectively (entries 3–4, and 7, Table 3).

Mechanistic Investigation. In order to gain an insight of the reaction, we subsequently carried out a series of experiments that include the preparation of a palladacycle, controlled experiments, radical trapping experiments, and the experiments of kinetic isotope effect to support the proposed catalytic mechanism.

1. Preparation and Characterization of the Palladacycle I. The palladacycle I was prepared by the stoichiometric reaction of 1a with palladium(II) acetate in dichloromethane (see the Supporting Information). The reaction was heated at 60–70 °C for 1–2 h to afford palladacycle I in 80% yield (Scheme 1).





The structure of I was secured by X-ray crystallography as shown as Figure 2.¹² On the basis of the structural analysis of I, we further confirm that the pyridinyl group of 1a is the directing group, but not the Boc group. In other words, the pyridinyl group coordinates with the palladium(II) ion, which enables the *ortho*-C-H bond activation of aniline to generate I. By contrast, the Boc group herein plays only as a protecting group for the secondary amine of 1a. It does not participate in the coordination of the palladium(II) ion. The conformation of I looks like a U-shape geometry, as well as the two ligands (i.e., 1a) are aligned as a head-to-tail form. Finally, the reaction of I with 2a in the presence of TBHP (6 equiv) was done; however, none of the anticipated *ortho*-benzoylated product 3a



Figure 2. An ORTEP drawing of palladacycle **I**.¹² All hydrogen atoms are omitted for clarity.

was observed (Scheme 2). This result implies that the palladacycle I might not be the key intermediate for the palladium-catalyzed *ortho*-C-H bond aroylation of 1 with 2.





2. Radical Trapping Experiment. To examine whether the presented reaction occurs via a radical process, the reaction of 1a with 2a in the presence of BHT, DPE, and TEMPO, respectively, was undertaken (see Table 4) (BHT = butylated hydroxytoluene, DPE = 1,1-diphenylethylene, and TEMPO = (2,2,6,6-tetramethylpiperidin-1-yl)oxyl). The experimental results showed that the total yield of products 3a and 4a dramatically decreased down to 17%, 27%, and the trace amount from 87% (in the absence of a radical scavenger) when 3 equiv of BHT, DPE, and TEMPO was added to the reaction, respectively. This study indicates that a radical species at least could be involved in the course of the reaction.

3. Investigation of the Kinetic Isotope Effect. To evaluate the ortho-C-H bond cleavage of 1 in the presented reaction is whether a rate-determining step, the intra- and intermolecular kinetic isotope effect (KIE) experiments as well as the KIE from two parallel reactions were done by the use of 1a-D₁, 1a, and 1a-D₅ as starting substrates. Next, we carried out the reaction of 1a-D₁, 1a, and 1a-D₅ with 4-methoxybenzaldehyde (2b) under the optimized reaction conditions. Basically, the determination of KIE was undertaken on the basis of the ¹H NMR spectroscopy (see Scheme 3 and Supporting Information). The value of $(k_H/k_D)_{intra'}$, $(k_H/k_D)_{inter'}$, and $(k_H/k_D)_{parallel}$ was determined to be 5.4, 1.0, and 0.9, respectively.¹³ The analysis of experimental results indicates that the palladiumcatalyzed ortho-C-H bond cleavage of 1 is not involved in the

Table 4. Radical Trapping Experiments for the Palladium-Catalyzed *ortho*-C-H Bond Benzoylation of 1a with 2a

	Ph-CHO (2a, 1.5 equiv)	MeO.	
MeO	N N 10 mol % Pd(OAc) ₂ TBHP (6 equiv) Radical Scavenger 1,4-dioxane 120 °C, 24 h	Boc N HeO Bo Ph 3a	N O Ph O Ph 4a
entry	radical scavenger (equiv)	product ratio (3a:4a) ^a	yield (%) ^b
1	-	68:32	87
2	BHT (1)	67:33	65
3	BHT (2)	93:7	25
4	BHT (3)	95:5	17
5	DPE (1)	90:10	70
6	DPE (2)	95:5	54
7	DPE (3)	95:5	27
8	TEMPO (1)	90:10	52
9	TEMPO (2)	90:10	20
10	TEMPO (3)	100:0	trace

^{*a*}Product ratio was determined as an average of three runs by GC-FID. ^{*b*}Product yield was determined as an average of three runs by the isolation.





rate-determining steps of the presented reaction. By contrast, the binding of **1** with the palladium ion is strongly suggested to be the key rate-determining step in the reaction.¹⁴

Proposed Mechanism. On the basis of the above mechanistic studies, a possible catalytic process for the palladium-catalyzed *ortho*-C-H bond aroylation of 1 with 2 is presented as shown as Figure 3. In the beginning, the binding of substrate 1 and palladium(II) acetate enables the formation of π -complex II by release of acetate anion. The π -



Figure 3. A proposed catalytic mechanism for the palladium-catalyzed *ortho*-C–H bond aroylation of **1** with **2**.

complex II is then transformed to the key aroyl palladium(III or IV) π -complex III by the redox reaction of the π -complex II and the aroyl radical 2° (generated from aryl aldehydes 2 by the treatment of TBHP via one electron abstraction). Next, the π -complex III is fast transformed to intermediate IV via the palladium(III or IV) insertion at the *ortho*-C-H bond of aniline ring accompanied the release of acetic acid. Finally, the desired product 3 or 5 is generated through the reductive elimination of the intermediate IV. Meanwhile, the palladium(II) acetate is regenerated from the intermediate IV to continue the catalytic cycle.

Synthetic Applications. The synthetic extension for a developed methodology is important. To demonstrate the merit of *ortho*-aroylated *N*-Boc protected 4-methoxy-*N*-arylpyridin-2-amines 3 and 5 in organic synthesis, we herein chose compound 3a as a modeling substrate to approach the syntheses of 2-aminobenzophenone (7), 9(10H)-acridanone (8), and *N*-(4-methoxy-2-pyridinyl)-9(10H)-acridanone (9) (see Scheme 4). First, the pyridinyl and Boc groups of 3a were





concurrently removed by the treatment of methyl trifluoromethanesulfonate (MeOTf) in toluene heated at 100 °C for 6 h and followed by the addition of hydrazine/acetic acid $(3/1, v/v)^{5c}$ in methanol heated at 130 °C for 6 h to afford 2aminobenzophenone (7) in 70% yield. Afterward, compound 7 was directly transformed to 9(10*H*)-acridanone (8) in 37% yield under the reaction conditions as shown as follows: 10 mol % CuI/bipyridine (1/1, equiv/equiv) in the presence of oxygen gas in dimethylacetamide (DMAC) heated at 130 °C for 24 h.¹⁵ On the other hand, the synthesis of 8 can be obviously improved by another synthetic route. That is, the treatment of 3a by a solution of hydrazine/acetic acid (3/1, v/v) in methanol heated at 100 °C for 1 h, and then conducted to the same reaction conditions with the synthesis of 8 from 7, which underwent the intramolecular C–H/N–H bond coupling to give *N*-pyridinyl-9(10*H*)-acridanone (9) in 88% yield. Finally, the pyridinyl group of 9 was again removed to afford 8 in 70% yield by employing the same reaction conditions with the transformation of 3a to 7.

CONCLUSION

We have developed a simple and efficient methodology for the synthesis of ortho-aroylated N-Boc protected N-arylpyridin-2amines using palladium(II) acetate and aryl aldehyde as the catalyst and coupling reagent, respectively, in the presence of a key oxidant, tert-butyl hydroperoxide. The ortho-aroylated Narylpyridin-2-amines are easily transformed to the alternative of 2-aminobenzophenones and 9(10H)-acridanones. In addition, we present a possible catalytic mechanism to explain the reaction through the studies of the palladacycle, controlled experiments, radical trapping experiments, and kinetic isotope effects. The methodology provides a facile route to synthesize ortho-aroylated anilines directed from anilines and aryl aldehydes. It can serve as a useful tool to target 2aminobenzophenone and 9(10H)-acridanone analogue. Finally, we believe that the developed methodology is able to play an important role in the synthetic field of pharmaceuticals and materials.

EXPERIMENTAL SECTION

General Methods. Solvents and reagents were purchased from commercial suppliers, and used without purification. ¹H NMR spectra were measured on 300, 400, and 500 MHz NMR spectrometers. Natural abundance ¹³C NMR spectra were measured by using 300, 400, and 500 MHz NMR spectrometers operating at 75, 100, and 125 MHz, respectively. Chemical shifts are given in parts per million (ppm) and coupling constant *J* in hertz (Hz) for both nuclei, with the solvent (usually CDCl₃) peak as an internal standard. The reference peak for ¹H is δ 7.26 of chloroform, and for ¹³C it is the central peak at δ 77.0. Low- and high-resolution mass spectrometry was obtained by the following ionization method and mass analyzer type for most of the compounds: EI-magnetic sector. Melting points were measured by using open capillary tubes and uncorrected.

General Procedure for the Synthesis of Compounds 3–6. A well-stirred solution of substrate 1 with aryl aldehydes 2 by the addition of Pd(OAc)₂ (4.5 mg, 0.02 mmol), TBHP (108 mg, 1.2 mmol) in 1,4-dioxane (3 mL) was heated at 120 °C for 24 h. After cooling down to room temperature, the reaction solution was washed by water (20 mL). The solution was then extracted by ethyl acetate (15 mL \times 3). Organic layers were combined, dried over MgSO₄, filtered through a pad of Celite, and evaporated in a vacuum. The residue was purified by silica gel chromatography using n-hexane/ ethyl acetate (30/1 to 10/1) as the eluent to afford products 3-6. The amounts of 1 and 2 are listed as follows. 1-H/H: 27 mg, 0.20 mmol for 3-H/H; 1-Me/H: 37 mg, 0.20 mmol for 3-Me/H; 1-OMe/H: 48 mg, 0.20 mmol for 3-OMe/H; 1-H/Boc: 54 mg, 0.20 mmol for 3-H/ Boc; 1-Me/Boc: 57 mg, 0.20 mmol for 3-Me/Boc; 1-OMe/Moc: 52 mg, 0.20 mmol for 3-OMe/Moc and 3-OMe/Moc; 1-OMe/Piv: 57 mg, 0.20 mmol for 3-OMe/Piv and 3-OMe/Piv; 1-OMe/Ac: 49 mg, 0.02 mmol for 3-OMe/Ac and 3-OMe/Ac; 1-OMe/Bn: 58 mg, 0.20 mmol for 3-OMe/Bn and 3-OMe/Bn; 1a: 60 mg, 0.20 mmol for 3-4; 1b: 66 mg, 0.20 mmol for 5b; 1c: 66 mg, 0.20 mmol for 5c; 1d: 66 mg, 0.2 mmol for 5d and 6d; 1e: 64 mg, 0.20 mmol for 5e and 6e: 1f: 64 mg, 0.20 mmol for 5f; 1g: 64 mg, 0.20 mmol for 5g; 1h: 64 mg,

0.20 mmol for 5h and 6h; 1i: 76 mg, 0.20 mmol for 5i; 2a: 32 mg, 0.30 mmol for 3a and 4a; 2b: 41 mg, 0.30 mmol for 3b and 4b; 2c: 45 mg, 0.30 mmol for 3c; 2d: 52 mg, 0.30 mmol for 3d; 2e: 45 mg, 0.30 mmol for 3e; 2f: 37 mg, 0.30 mmol for 3f and 4f; 2g: 37 mg, 0.30 mmol for 3g and 4g; 2h: 37 mg, 0.30 mmol for 3h and 4h; 2i: 42 mg, 0.30 mmol for 3i; 2j: 42 mg, 0.30 mmol for 3j and 4j; 2k: 42 mg, 0.30 mmol for 3k and 4k; 2l: 56 mg, 0.30 mmol for 3l; 2m: 56 mg, 0.30 mmol for 3m; 2n: 40 mg, 0.30 mmol for 3n and 4n; 2o: 40 mg, 0.30 mmol for 30 and 40; 2p: 50 mg, 0.30 mmol for 3p; 2q: 50 mg, 0.30 mmol for 3q and 4q; 2r: 40 mg, 0.30 mmol for 3r. All product yields were determined by the isolation and shown as follows: 3-H/H: 0% (0 mg, 0 mmol); 3-Me/H: 0% (0 mg, 0 mmol); 3-OMe/H: 0% (0 mg, 0 mmol); 3-H/Boc: trace; 3-Me/Boc: 40% (31 mg, 0.080 mmol); 3-OMe/Moc: 57% (41 mg, 0.11 mmol); 3-OMe/Piv: 35% (27 mg, 0.070 mmol); 3-OMe/Ac: 33% (23 mg, 0.070 mmol); 3-OMe/Bn: 28% (22 mg, 0.060 mmol); 3a: 59% (49 mg, 0.12 mmol); 3b: 60% (52 mg, 0.12 mmol); 3c: 0% (0 mg, 0 mmol); 3d: 24% (24 mg, 0.050 mmol); 3e: 44% (40 mg, 0.090 mmol); 3f: 57% (47 mg, 0.11 mmol); 3g: 74% (63 mg, 0.15 mmol); 3h: 77% (65 mg, 0.15 mmol); 3i: 39% (34 mg, 0.080 mmol); 3j: 56% (49 mg, 0.11 mmol); 3k: 61% (54 mg, 0.12 mmol); 3l: 30% (29 mg, 0.060 mmol); 3m: 37% (36 mg, 0.070 mmol); 3n: 69% (60 mg, 0.14 mmol); 3o: 54% (47 mg, 0.11 mmol); 3p: 35% (33 mg, 0.070 mmol); 3q: 26% (24 mg, 0.050 mmol); 3r: 45% (39 mg, 0.090 mmol); 4-OMe/Moc: 31% (32 mg, 0.070 mmol); 4-OMe/Piv: 10% (10 mg, 0.020 mmol); 4a: 28% (29 mg, 0.060 mmol); 4b: 29% (33 mg, 0.060 mmol); 4f: 1% (1 mg, 0.002 mmol); 4g: 10% (11 mg, 0.020 mmol); 4h: 9% (10 mg, 0.020 mmol); 4j: 1% (1 mg, 0.002 mmol); 4k: 3% (4 mg, 0.006 mmol); 4n: 19% (22 mg, 0.040 mmol); 4o: 7% (8 mg, 0.01 mmol); 4q: 3% (4 mg, 0.006 mmol); 5b: 91% (79 mg, 0.18 mmol); 5c: 83% (72 mg, 0.17 mmol); 5d: 57% (50 mg, 0.11 mmol); 5e: 53% (45 mg, 0.11 mmol); 5f: 57% (48 mg, 0.11 mmol); 5g: 94% (79 mg, 0.19 mmol); 5h: 73% (64 mg, 0.15 mmol); 5i: 39% (38 mg, 0.080 mmol); 6d: 36% (39 mg, 0.070 mmol); 6e: 21% (22 mg, 0.040 mmol); 6h: 6% (7 mg, 0.01 mmol).

Synthesis of Compound 7. A well-stirred solution of **3a** (81 mg, 0.20 mmol) in toluene (2 mL) by the addition of MeOTf (45 μ L, 0.4 mmol) was heated at 100 °C under nitrogen gas for 6 h. After cooling down to room temperature, the reaction solution was evaporated in a vacuum. The residue was dissolved in methanol (2 mL) and then the mixed solution of hydrazine (0.3 mL) and acetic acid (0.9 mL) (1:3, v/v) was added to the above solution, which was further heated at 130 °C for 6 h. After cooling down to room temperature, water (20 mL) was added to the solution, which was further extracted by ethyl acetate (15 mL × 3). Organic layers were combined, dried over MgSO₄, filtered through a pad of Celite, and evaporated in a vacuum. The residue was purified by silica gel chromatography using *n*-hexane/ethyl acetate (30/1 to 10/1) as the eluent to give product 7 in 70% (28 mg, 0.14 mmol).

Synthesis of Compound 8. A well-stirred solution of 7 (500 mg, 2.50 mmol), copper(II) iodide (9.5 mg, 0.050 mmol), 2,2'-bipyridine (8 mg, 0.05 mmol) in dimethylacetamide (10 mL) was heated at 130 °C for 24 h. After cooling down to room temperature, water (20 mL) was added to the above solution that was further extracted by ethyl acetate (15 mL \times 3). Organic layers were combined, dried over MgSO₄, filtered through a pad of Celite, and evaporated in a vacuum. The residue was purified by silica gel chromatography using *n*-hexane/ethyl acetate (30/1 to 10/1) as the eluent to give product 8 in 37% (447 mg, 0.98 mmol).

Synthesis of (a) Compound 9 transformed to (b) Compound 8. (a) A well-stirred solution of 3a (500 mg, 2.50 mmol) in methanol (10 mL) was added by the mixed solution of hydrazine (1.8 mL) and acetic acid (5.4 mL) (1:3, v/v), which was heated at 100 °C for 3 h. After cooling down to room temperature, 15 mL of water was added to the above solution, which was extracted by ethyl acetate (15 mL \times 2). Organic layers were combined, dried over MgSO₄, filtered through a pad of Celite, and evaporated in vacuum. Next, the residue was dissolved in dimethylacetamide (5 mL) without purification, and subsequently, copper(II) iodide (9.5 mg, 0.050 mmol) and 2,2′-bipyridine (8 mg, 0.05 mmol) were added to the above solution,

which was heated at 130 °C for 24 h. After cooling down to room temperature, 20 mL of water was added to the above solution that was further extracted by ethyl acetate (15 mL \times 3). Organic layers were combined, dried over MgSO₄, filtered through a pad of Celite, and evaporated in a vacuum. The residue was purified by silica gel chromatography using *n*-hexane/ethyl acetate (30/1 to 10/1) as the eluent to give product **9** in 88% (447 mg, 0.98 mmol). (b) The operating procedure for the synthesis of **8** from **9** was the same as that of 7 to **8**, and the yield of **8** is 70% (133 mg, 0.69 mmol).

Characterization Data of Compounds 3–9.¹⁶ *tert-Butyl phenyl(pyridin-2-yl)carbamate (1-H/Boc).* White solid; mp 64–65 °C; $R_f = 0.53$ (*n*-hexane/ethyl acetate = 1/1); ¹H NMR (400 MHz, CDCl₃) δ 1.45 (s, 9 H), 7.07 (ddd, J = 5.6, 4.8, 0.8 Hz, 1 H), 7.20–7.25 (m, 3 H), 7.32–7.37 (m, 2 H), 7.50 (d, J = 8.0 Hz, 1 H), 7.70 (ddd, J = 8.6, 7, 2.0 Hz, 1 H), 8.38 (dd, J = 4.8 Hz, 1 H); ¹³C NMR (CDCl₃, 100 MHz) δ 28.2 (CH₃), 81.6 (C_q), 120.5 (CH), 121.0 (CH), 126.3 (CH), 127.6 (CH × 2), 128.8 (CH × 2), 137.5 (CH), 141.8 (C_q), 148.6 (CH), 153.6 (C_q), 155.4 (C_q); MS (EI, *m/z*) 270 (M⁺, 3), 169 (100), 57 (45); HRMS (EI-magnetic sector) *m/z* [M⁺] calcd for C₁₆H₁₈N₂O₂ 270.1368, found 270.1369.

tert-Butyl (4-methylpyridin-2-yl)(phenyl)carbamate (1-Me/Boc). White solid; mp 64–65 °C; $R_f = 0.53$ (*n*-hexane/ethyl acetate = 1/1); ¹H NMR (400 MHz, CDCl₃) δ 1.44 (s, 9 H), 2.36 (s, 3 H), 6.91 (dd, J = 5.2, 0.8 Hz, 1 H), 7.19–7.23 (m, 3 H), 7.30–7.36 (m, 3 H), 8.24 (d, J = 5.2 Hz, 1 H); ¹³C NMR (CDCl₃, 100 MHz) δ 21.0 (CH₃), 28.2 (CH₃ × 3), 81.5 (C_q), 121.6 (CH), 121.9 (CH), 126.1 (CH), 127.5 (CH × 2), 128.7 (CH × 2), 141.9 (C_q), 148.2 (CH), 148.9 (C_q), 153.7 (C_q), 155.4 (C_q); MS (EI, *m*/z) 284 (M⁺, 4), 183 (100), 57 (46); HRMS (EI-magnetic sector) *m*/*z* [M⁺] calcd for C₁₇H₂₀N₂O₂ 284.1525, found 284.1534.

Methyl (4-methoxypyridin-2-yl)(phenyl)carbamate (1-OMe/Moc). White solid; mp 64–65 °C; $R_f = 0.53$ (*n*-hexane/ethyl acetate = 1/1); ¹H NMR (400 MHz, CDCl₃) δ 3.76 (s, 3 H), 3.85 (s, 3 H), 6.66 (dd, J = 5.6, 2.4 Hz, 1 H), 7.07 (d, J = 2.0 Hz, 1 H), 7.26 (m, 3 H), 7.35–7.37 (m, 2 H), 8.20 (d, J = 5.6, 1 H); ¹³C NMR (100 MHz, CDCl₃) δ 53.2 (CH₃), 55.3 (CH₃), 106.2 (CH), 108.0 (CH), 126.8 (CH), 127.5 (CH × 2), 128.9 (CH × 2), 141.1 (C_q), 149.4 (CH), 155.1 (C_q), 156.4 (C_q), 167.1 (C_q); MS (EI, *m/z*) 258 (M⁺, 1), 257 (21), 177 (100), 164 (30), 161 (48); HRMS (EI-magnetic sector) *m*/*z* [M⁺] calcd for C₁₄H₁₄N₂O₃ 258.1004, found 258.1005.

N-(4-*Methoxypyridin*-2-*yl*)-*N*-*phenylpivalamide* (1-OMe/Piv). White solid; mp 66−67 °C; $R_f = 0.52$ (*n*-hexane/ethyl acetate = 1/1); ¹H NMR (CDCl₃, 400 MHz) δ 1.17 (s, 9 H), 3.80 (s, 3 H), 6.63 (d, *J* = 2.4 Hz, 1 H), 6.70 (dd, *J* = 6.0, 2.4 Hz, 1 H), 7.23−7.30 (m, 3 H), 7.33−7.37 (m, 2 H), 8.27 (d, *J* = 5.6 Hz, 1 H); ¹³C NMR (CDCl₃, 100 MHz) δ 29.3 (CH₃ × 3), 42.1 (C_q), 55.3 (CH₃), 107.7 (CH), 108.6 (CH), 127.1 (CH), 128.5 (CH × 2), 129.1 (CH × 2), 143.4 (C_q), 149.6 (CH), 158.6 (C_q), 167.3 (C_q), 181.1 (C_q); MS (EI, *m*/z) 284 (M⁺, 1), 199 (100), 200 (17), 283 (14); HRMS (EImagnetic sector) *m*/*z* [M⁺] calcd for C₁₇H₂₀N₂O₂ 284.1525, found 284.1524.

N-(4-*Methoxypyridin*-2-*yl*)-*N*-*phenylacetamide* (1-OMe/Ac). White solid; mp 58–59 °C; $R_f = 0.43$ (*n*-hexane/ethyl acetate = 1/2); ¹H NMR (400 MHz, CDCl₃) δ 2.10 (s, 3 H), 3.84 (s, 3 H), 6.69 (dd, J = 5.6, 2.4 Hz, 1 H), 6.99 (d, J = 1.2 Hz, 1 H), 7.29–7.33 (m, 3 H), 7.38–7.43 (m, 2 H), 8.24 (d, J = 5.6 Hz, 1 H); ¹³C NMR (100 MHz, CDCl₃) δ 24.2 (C_q), 55.4 (CH₃), 106.9 (CH), 108.5 (CH × 2), 127.5 (CH), 128.1 (CH), 129.4 (CH × 2), 142.0 (C_q), 149.6 (CH), 156.7 (C_q), 167.1 (C_q), 170.9 (C_q); MS (EI, *m*/z) 242 (M⁺, 1), 199 (100), 200 (47); HRMS (EI-magnetic sector) *m*/*z* [M⁺] calcd for C₁₄H₁₄N₂O₂ 242.1055, found 242.1056.

N-Benzyl-4-methoxy-N-phenylpyridin-2-amine (1-OMe/Bn). White solid; mp 81–82 °C; $R_f = 0.63$ (*n*-hexane/ethyl acetate = 1/1); ¹H NMR (400 MHz, CDCl₃) δ 3.65 (s, 3 H), 5.25 (s, 2 H), 5.99 (d, J = 2.0 Hz, 1 H), 6.27 (dd, J = 6.0, 2.0 Hz, 1 H), 7.14–7.35 (m, 10 H), 8.05 (d, J = 5.6 Hz, 1 H); ¹³C NMR (100 MHz, CDCl₃) δ 166.7 (C_q), 160.6 (C_q), 148.9 (CH), 145.4 (C_q), 139.4 (C_q), 129.6 (CH × 2), 128.2 (CH × 2), 127.5 (CH × 2), 126.9 (CH × 2), 126.6 (CH), 125.6 (CH), 101.8 (CH), 93.5 (CH), 54.8 (CH₃), 53.7 (CH₂); MS (EI, m/z) 290 (M⁺, 1), 289 (48), 213 (65), 182 (45); HRMS (EI- magnetic sector) m/z [M⁺] calcd for C₁₉H₁₈N₂O 290.1419, found 290.1421.

tert-Butyl (4-methoxypyridin-2-yl)(phenyl)carbamate (1a). White solid; mp 66–67 °C; $R_f = 0.40$ (*n*-hexane/ethyl acetate = 2/1); ¹H NMR (300 MHz, CDCl₃) δ 1.45 (s, 9 H), 3.86 (s, 3 H), 6.63 (dd, J = 5.5, 2.5 Hz, 1 H), 7.04 (d, J = 2.5 Hz, 1 H), 7.20–7.25 (m, 3 H), 7.36 (t, J = 8.0 Hz, 2 H), 8.19 (d, J = 6.0 Hz, 1 H); ¹³C NMR (125 MHz, CDCl₃) δ 28.2 (CH₃ × 3), 55.3 (CH₃), 81.5 (C_q), 106.4 (CH), 107.6 (CH), 126.2 (CH), 127.5 (CH × 2), 128.7 (CH × 2), 141.8 (C_q), 149.3 (CH), 153.5 (C_q), 156.9 (C_q), 166.8 (C_q); MS (EI, *m*/z) 300 (M⁺, 4); HRMS (EI-magnetic sector) *m*/*z* [M⁺] calcd for C₁₇H₂₀N₂O₃ 300.1474, found 300.1476.

tert-Butyl (2-methoxyphenyl)(4-methoxypyridin-2-yl)carbamate (**1b**). Yellow solid; mp 72–73 °C; $R_f = 0.59$ (*n*-hexane/ethyl acetate = 1/1); ¹H NMR (300 MHz, CDCl₃) δ 1.41 (s, 9 H), 3.80 (s, 3 H), 3.85 (s, 3 H), 6.55 (dd, J = 5.7, 2.1 Hz, 1 H), 6.93–6.97 (m, 2 H), 7.16–7.30 (m, 3 H), 8.10 (d, J = 5.7 Hz, 1 H); ¹³C NMR (125 MHz, CDCl₃) δ 28.1 (CH₃ × 3), 55.2 (CH₃), 55.6 (CH₃), 81.0 (C_q), 104.1 (CH), 107.0 (CH), 111.7 (CH), 120.6 (CH), 128.4 (CH), 129.7 (CH), 130.7 (C_q), 148.8 (CH), 153.7 (C_q), 155.4 (C_q), 156.8 (C_q), 166.5 (C_q); MS (EI, *m*/*z*) 330 (M⁺, 1), 299 (100), 243 (76), 199 (97), 89 (62), 57 (70); HRMS *m*/*z* calcd for C₁₈H₂₂N₂O₄ 330.1580, found 330.1579.

tert-Butyl (3-methoxyphenyl)(4-methoxypyridin-2-yl)carbamate (**1c**). Brown solid; mp 50–51 °C; $R_f = 0.56$ (*n*-hexane/ethyl acetate = 1/1); ¹H NMR (300 MHz, CDCl₃) δ 1.45 (s, 9 H), 3.77 (s, 3 H), 3.85 (s, 3 H), 6.64 (dd, J = 5.4, 1.8 Hz, 1 H), 6.79 (t, J = 7.8 Hz, 3 H), 7.00 (s, 1 H), 7.22 (d, J = 9.0 Hz, 1 H), 8.21 (d, J = 6.0 Hz, 1 H); ¹³C NMR (125 MHz, CDCl₃) δ 28.2 (CH₃ × 3), 55.2 (CH₃), 55.3 (CH₃), 81.6 (C_q), 106.6 (CH), 107.7 (CH), 111.8 (CH), 113.3 (CH), 119.8 (CH), 129.2 (CH), 142.8 (C_q), 149.4 (CH), 153.4 (C_q), 156.8 (C_q), 159.8 (C_q), 166.8 (C_q); MS (EI, m/z) 330 (M⁺, 22), 230 (100), 215 (55), 57 (100); HRMS m/z calcd for C₁₈H₂₂N₂O₄ 330.1580, found 330.1578.

tert-Butyl (4-*methoxyphenyl*)(4-*methoxypyridin-2-yl*)*carbamate* (**1d**). White solid; mp 108–109 °C; $R_f = 0.55$ (*n*-hexane/ethyl acetate = 1/1); ¹H NMR (300 MHz, CDCl₃) δ 1.44 (s, 9 H), 3.80 (s, 3 H), 3.85 (s, 3 H), 6.60 (dd, J = 5.7, 2.4 Hz, 1 H), 6.88 (d, J = 6.9 Hz, 2 H), 7.06 (d, J = 2.1 Hz, 1 H), 7.14 (d, J = 8.7 Hz, 2 H), 8.18 (d, J = 6.0 Hz, 1 H); ¹³C NMR (125 MHz, CDCl₃) δ 28.2 (CH₃ × 3), 55.3 (CH₃), 55.4 (CH₃), 81.4 (C_q), 105.9 (CH), 107.4 (CH), 114.0 (CH × 2), 128.9 (CH × 2), 134.7 (C_q), 149.2 (CH), 153.8 (C_q), 157.0 (C_q), 157.8 (C_q), 166.8 (C_q); MS (EI, *m*/*z*) 330 (M⁺, 8), 230 (100), 215 (60); HRMS *m*/*z* calcd for C₁₈H₂₂N₂O₄ 330.1580, found 330.1582.

tert-Butyl (4-fluorophenyl)(4-methoxypyridin-2-yl)carbamate (**1e**). Yellow solid; mp 58–59 °C; $R_f = 0.31$ (*n*-hexane/ethyl acetate = 2/1); ¹H NMR (300 MHz, CDCl₃) δ 1.44 (s, 9 H), 3.87 (s, 3 H), 6.63 (dd, J = 6.0, 2.4 Hz, 1 H), 7.00–7.10 (m, 3 H), 7.17–7.21 (m, 2 H), 8.17 (d, J = 5.7 Hz, 1 H); ¹³C NMR (125 MHz, CDCl₃) δ 28.2 (CH₃ × 3), 55.3 (CH₃), 81.8 (C_q), 106.0 (CH), 107.6 (CH), 115.6 (d, $J_{C-F} = 22.8$ Hz, CH), 129.4 (d, $J_{C-F} = 8.3$ Hz, CH), 137.7 (d, $J_{C-F} = 2.44.0$ Hz, C_q), 166.9 (C_q); MS (EI, *m*/z) 318 (M⁺, 4), 230 (48), 217 (100), 57 (54); HRMS *m*/z calcd for C₁₇H₁₉FN₂O₃ 318.1380, found 318.1377.

tert-Butyl (3-fluorophenyl)(4-methoxypyridin-2-yl)carbamate (1f). Yellow solid; mp 68–69 °C; $R_f = 0.28$ (*n*-hexane/ethyl acetate = 2/1); ¹H NMR (300 MHz, CDCl₃) δ 1.45 (s, 9 H), 3.87 (s, 3 H), 6.67 (dd, J = 6.0, 2.4 Hz, 1 H), 6.89–7.03 (m, 4 H), 7.24–7.32 (m, 1 H), 8.21 (d, J = 5.7 Hz, 1 H); ¹³C NMR (125 MHz, CDCl₃) δ 28.1 (CH₃ × 3), 55.4 (CH₃), 82.0 (C_q), 106.7 (CH), 108.0 (CH), 113.0 (d, $J_{C-F} = 21.0$ Hz, CH), 114.6 (d, $J_{C-F} = 23.8$ Hz, CH), 122.9 (d, $J_{C-F} = 2.8$ Hz, CH), 129.6 (d, $J_{C-F} = 9.1$ Hz, CH), 143.2 (d, $J_{C-F} = 245.1$ Hz, C_q), 167.0 (C_q); MS (EI, *m*/z) 318 (M⁺, 3), 217 (100), 57 (S0); HRMS *m*/z calcd for C₁₇H₁₉FN₂O₃ 318.1380, found 318.1377.

tert-Butyl (2-fluorophenyl)(4-methoxypyridin-2-yl)carbamate (**1g**). White solid; mp 61–62 °C; $R_f = 0.30$ (*n*-hexane/ethyl acetate = 2/1); ¹H NMR (300 MHz, CDCl₃) δ 1.44 (s, 9 H), 3.87 (s, 3 H),

6.59 (dd, J = 5.7, 2.1 Hz, 1 H), 7.10–7.16 (m, 2 H), 7.22–7.30 (m, 3 H), 8.10 (d, J = 5.7 Hz, 1 H); ¹³C NMR (125 MHz, CDCl₃) δ 28.0 (CH₃ × 3), 55.3 (CH₃), 81.9 (C_q), 104.3 (CH), 107.5 (CH), 116.0 (d, $J_{C-F} = 20.0$ Hz, CH), 124.1 (d, $J_{C-F} = 3.6$ Hz, CH), 128.6 (d, $J_{C-F} = 7.8$ Hz, CH), 129.4 (d, $J_{C-F} = 12.8$ Hz, C_q), 130.3 (d, $J_{C-F} = 0.9$ Hz, CH), 148.8 (CH), 153.0 (C_q), 156.0 (C_q), 158.3 (d, $J_{C-F} = 247.9$ Hz, C_q), 166.7 (C_q); MS (EI, m/z) 318 (M⁺, 2), 217 (22), 199 (100), 57 (38); HRMS m/z calcd for C₁₇H₁₉FN₂O₃ 318.1380, found 318.1380.

tert-Butyl (4-chlorophenyl)(4-methoxypyridin-2-yl)carbamate (1h). Yellow solid; mp 74–75 °C; $R_f = 0.50$ (*n*-hexane/ethyl acetate = 2/1); ¹H NMR (300 MHz, CDCl₃) δ 1.45 (s, 9 H), 3.86 (s, 3 H), 6.64 (dd, J = 5.5, 2.5 Hz, 1 H), 7.07 (d, J = 2.5 Hz, 1 H), 7.15 (d, J = 6.5 Hz, 2 H), 7.30 (d, J = 8.5 Hz, 2 H), 8.18 (d, J = 5.5 Hz, 1 H); ¹³C NMR (125 MHz, CDCl₃) δ 28.1 (CH₃ × 3), 55.3 (CH₃), 81.9 (C_q), 106.3 (CH), 107.8 (CH), 128.8 (CH × 2), 128.9 (CH × 2), 131.8 (C_q), 140.3 (C_q), 149.3 (CH), 153.2 (C_q), 156.5 (C_q), 166.9 (C_q); MS (EI, *m/z*) 334 (M⁺, 3), 233 (65), 57 (100); HRMS *m/z* calcd for C₁₇H₁₉ClN₂O₃ 334.1084, found 334.1086.

tert-Butyl (4-bromophenyl)(4-methoxypyridin-2-yl)carbamate (1i). Orange solid; mp 120–121 °C; $R_f = 0.50$ (*n*-hexane/ethyl acetate = 2/1); ¹H NMR (300 MHz, CDCl₃) δ 1.45 (s, 9 H), 3.87 (s, 3 H), 6.65 (dd, J = 5.7, 2.1 Hz, 1 H), 7.05–7.10 (m, 3 H), 7.45 (d, J = 8.7 Hz, 2 H), 8.18 (d, J = 5.1 Hz, 1 H); ¹³C NMR (125 MHz, CDCl₃) δ 28.2 (CH₃ × 3), 55.4 (CH₃), 82.0 (C_q), 106.3 (CH), 107.8 (CH), 119.7 (C_q), 129.1 (CH × 2), 131.8 (CH × 2), 140.9 (C_q), 149.3 (CH), 153.2 (C_q), 156.4 (C_q), 166.9 (C_q); MS (EI, *m/z*) 378 (M⁺, 5), 279 (100), 277 (90), 57 (57); HRMS *m/z* calcd for C₁₇H₁₉BrN₂O₃ 378.0579, found 378.0580.

tert-Butyl (2-benzoylphenyl)(4-methylpyridin-2-yl)carbamate (3-Me/Boc). White oil; $R_f = 0.53$ (*n*-hexane/ethyl acetate = 1/2); ¹H NMR (400 MHz, CDCl₃) δ 1.37 (s, 9 H), 2.21 (s, 3 H), 6.71 (dd, *J* = 5.2, 0.4 Hz, 1 H), 7.09 (s, 1 H), 7.33–7.57 (m, 8 H), 7.70 (dd, *J* = 8.0, 0.8 Hz, 1 H), 8.03 (d, *J* = 4.8 Hz, 1 H); ¹³C NMR (100 MHz, CDCl₃) δ 21.0 (CH₃), 28.1 (CH₃), 81.8 (C_q), 120.6 (CH), 121.1 (CH), 126.5 (CH), 127.9 (CH × 2), 129.5 (CH), 130.0 (CH × 2), 130.7 (CH), 130.9 (CH), 132.7 (CH), 136.9 (C_q), 137.2 (C_q), 139.8 (C_q), 147.2 (CH), 148.2 (C_q), 153.2 (C_q), 154.6 (C_q), 195.7 (C_q); MS (ESI+, *m/z*) 389 (M⁺, 100); HRMS (ESI+) *m/z* [M + H] calcd for C₂₄H₂₅N₂O₃ 389.1860, found 389.1865.

Methyl (2-benzoylphenyl)(4-methoxypyridin-2-yl)carbamate (3-OMe/Moc). White oil; $R_f = 0.59$ (*n*-hexane/ethyl acetate = 1/1); ¹H NMR (400 MHz, CDCl₃) δ 3.63 (s, 3 H), 3.77 (s, 3 H), 6.50 (dd, J = 5.6, 2.4 Hz, 1 H), 7.16 (d, J = 2.0 Hz, 1 H), 7.36–7.45 (m, 5 H), 7.50–7.57 (m, 2 H), 7.75 (td, J = 7.6, 2.0 Hz, 2 H), 8.01 (d, J = 6.0 Hz, 1 H); ¹³C NMR (100 MHz, CDCl₃) δ 53.2 (CH₃), 55.2 (CH₃), 104.6 (CH), 108.2 (CH), 126.7 (CH), 128.0 (CH × 2), 129.6 (CH), 130.2 (CH × 2), 131.2 (CH), 132.9 (CH), 139.3(C_q), 148.4 (CH), 154.7(C_q), 155.9(C_q), 166.8(C_q), 166.4(C_q), 195.6(C_q); HRMS (EImagnetic sector) m/z [M⁺] calcd for C₂₁H₁₈N₂O₄ 362.1267, found 362.1267.

N-(2-Benzoylphenyl)-*N*-(4-methoxypyridin-2-yl)pivalamide (3-OMe/Piv). White oil; $R_f = 0.63$ (*n*-hexane/ethyl acetate = 1/1); ¹H NMR (400 MHz, CDCl₃) δ 0.94 (s, 9 H), 3.79 (s, 3 H), 6.71 (dd, *J* = 6.0, 2.4 Hz, 1 H), 7.08 (d, *J* = 2.4 Hz, 1 H), 7.24–7.33 (m, 3 H), 7.44 (t, *J* = 7.6 Hz, 3 H), 7.56 (tt, *J* = 9.2, 1.2 Hz, 1 H), 7.85 (dd, *J* = 8.0, 0.8 Hz, 2 H), 8.22 (d, *J* = 5.6 Hz, 1 H); ¹³C NMR (100 MHz, CDCl₃) δ 28.6 (CH₃× 3), 41.7 (C_q), 55.4 (CH₃), 107.8 (CH), 109.9 (CH), 126.0 (CH), 128.2 (CH × 2), 128.7 (CH), 129.5 (CH), 130.6 (CH × 2), 131.1 (CH), 133.1 (CH), 136.8 (C_q), 137.7 (C_q), 148.7 (CH), 158.3 (C_q), 167.4 (C_q), 180.7 (C_q), 196.2 (C_q); MS (ESI, *m*/z) 411 (M⁺23, 1), 397 (80), 389 (100), 381 (90); HRMS (EI-magnetic sector) *m*/z [M⁺] calcd for C₂₄H₂₄N₂O₃ 388.1787, found 388.1786.

N-(2-Benzoylphenyl)-*N*-(4-methoxypyridin-2-yl)acetamide (3-OMe/Ac). White oil; $R_f = 0.53$ (*n*-hexane/ethyl acetate = 1/2); ¹H NMR (400 MHz, CDCl₃) δ 2.07 (s, 3 H), 3.76 (s, 3 H), 6.57 (bs, 1 H), 7.19 (d, J = 2.4 Hz, 1 H), 7.38–7.40 (m, 4 H), 7.49–7.57 (m, 3 H), 7.78 (d, J = 2.8 Hz, 2 H), 8.10 (bs, 1 H); ¹³C NMR (100 MHz, CDCl₃) δ 29.7 (CH₃), 55.2 (CH₃), 109.0 (CH), 126.9 (CH), 128.1 (CH × 2), 128.2 (CH), 129.3(CH), 130.4 (CH × 2), 131.3 (CH), 132.9 (CH), 134.4 (C_q), 136.7 (C_q), 137.8 (C_q), 140.1 (C_q), 148.6

(CH), 167.0 (C_q), 171.1 (C_q), 195.6 (Cq); MS (EI, m/z) 361 (M⁺, 1), 340 (30), 177 (41), 86 (58), 84 (100), 70 (28), 61 (44); HRMS (EI-magnetic sector) m/z [M⁺] calcd for C₂₁H₁₈N₂O₃ 346.1317, found 346.1318.

(2-(Benzyl(4-methoxypyridin-2-yl)amino)phenyl)(phenyl)-methanone (3-OMe/Bn). White oil; $R_f = 0.56$ (*n*-hexane/ethyl acetate = 1/ 1); ¹H NMR (400 MHz, CDCl₃) δ 3.61 (s, 3 H), 4.95 (s, 2 H), 5.78 (s, 1 H), 6.13 (d, J = 5.8 Hz, 1 H), 7.16–7.25 (m, 7 H), 7.30–7.35 (m, 3 H), 7.44–7.51 (m, 3 H), 7.63 (dd, J = 9.2, 0.8 Hz, 1 H), 7.74 (d, J = 3.2 Hz, 1 H); ¹³C NMR (100 MHz, CDCl₃) δ 173.1 (C_q), 166.6 (C_q), 159.6 (C_q), 148.6 (CH), 144.4 (C_q), 138.8(C_q), 137.4 (C_q), 132.6 (CH), 131.9 (CH), 130.2 (CH), 129.6 (CH), 128.3 (CH × 2), 127.9 (CH × 2), 127.3 (CH × 2), 126.8 (CH), 125.8 (CH), 101.8 (C_q), 93.5 (CH), 54.8 (CH2), 54.7 (CH₃); MS (EI, *m*/z) 394 (M⁺, 1), 289 (100), 213 (20); HRMS (EI-magnetic sector) *m*/*z* [M⁺] calcd for C₂₆H₂₂N₂O₂ 394.1681, found 394.1681.

tert-Butyl (2-benzoylphenyl)(4-methoxypyridin-2-)carbamate (**3a**). Yellow solid; mp 76–77 °C; $R_f = 0.40$ (*n*-hexane/ethyl acetate = 2/1); ¹H NMR (300 MHz, CDCl₃) δ 1.38 (s, 9 H), 3.74 (s, 3 H), 6.45 (dd, J = 6.0, 2.4 Hz, 1 H), 6.92 (d, J = 2.1 Hz, 1 H), 7.32–7.58 (m, 7 H), 7.72 (d, J = 7.2 Hz, 2 H), 7.97 (d, J = 5.7 Hz, 1 H); ¹³C NMR (125 MHz, CDCl₃) δ 28.1 (CH₃ × 3), 55.1 (CH₃), 81.9 (C_q), 104.6 (CH), 107.6 (CH), 126.6 (CH), 127.9 (CH × 2), 129.5 (CH), 130.1 (CH × 2), 130.7 (CH), 131.0 (CH), 132.7 (CH), 137.0 (C_q), 137.3 (C_q), 139.8 (C_q), 148.2 (CH), 153.1 (C_q), 156.1 (C_q), 166.4 (C_q), 195.7 (C_q); MS (EI, *m*/*z*) 405 (M⁺,1), 275 (43), 243 (64), 227 (53), 99 (100); HRMS *m*/*z* calcd for C₂₄H₂₄N₂O₄ 404.1736, found 404.1735.

tert-Butyl (2-(4-methoxybenzoyl)phenyl)(4-methoxypyridin-2-yl)carbamate (**3b**). Yellow solid; mp 122–123 °C; $R_f = 0.51$ (*n*-hexane/ethyl acetate = 1/1); ¹H NMR (300 MHz, CDCl₃) δ 1.38 (s, 9 H), 3.74 (s, 3 H), 3.84 (s, 3 H), 6.45 (dd, J = 5.7, 2.4 Hz, 1 H), 6.82 (d, J = 9.0 Hz, 2 H), 6.93 (d, J = 2.1 Hz, 1 H), 7.33–7.56 (m, 4 H), 7.70 (d, J = 8.7 Hz, 2 H), 7.98 (d, J = 5.7 Hz, 1 H); ¹³C NMR (125 MHz, CDCl₃) δ 28.1 (CH₃ × 3), 55.1 (CH₃), 55.4 (CH₃), 81.8 (C_q), 104.8 (CH), 107.6 (CH), 113.2 (CH × 2), 126.6 (CH), 129.2 (CH), 130.2 (C_q), 130.6 (CH), 130.7 (CH), 132.4 (CH × 2), 137.4 (C_q), 139.6 (C_q), 148.2 (CH), 153.1 (C_q), 156.2 (C_q), 163.3 (C_q), 166.4 (C_q), 194.4 (C_q); MS (EI, *m/z*) 434 (M⁺, 1), 243 (40), 199 (100), 57 (66); HRMS *m/z* calcd for C₂₅H₂₆N₂O₅ 434.1842, found 434.1844.

tert-Butyl (4-methoxypyridin-2-yl)(2-(4-(trifluoromethyl)benzoyl)phenyl)carbamate (**3d**). Yellow solid; mp 148–149 °C; R_f = 0.48 (*n*-hexane/ethyl acetate = 2/1); ¹H NMR (300 MHz, CDCl₃) δ 1.38 (s, 9 H), 3.76 (s, 3 H), 6.47 (dd, J = 5.7, 1.8 Hz, 1 H), 6.99 (s, 1 H), 7.36–7.43 (m, 3 H), 7.56–7.62 (m, 3 H), 7.83 (d, J = 8.1 Hz, 2 H), 7.95 (d, J = 5.7 Hz, 1 H); ¹³C NMR (125 MHz, CDCl₃) δ 28.0 (CH₃ × 3), 55.1 (CH₃), 82.1 (C_q), 104.3 (CH), 107.4 (CH), 124.9 (q, J_{C-F} = 4.8 Hz, CH × 2), 126.7 (CH), 129.4 (CH), 130.2 (CH × 2), 130.7 (CH), 131.5 (CH), 133.8 (C_q), 136.5 (C_q), 139.9 (C_q), 140.3 (C_q), 148.2 (CH), 153.1 (C_q), 155.9 (C_q), 166.5 (C_q), 194.7 (C_q); MS (EI, *m*/z) 472 (M⁺, 1), 343 (26), 227 (43), 199 (100); HRMS *m*/z calcd for C₂₅H₂₃F₃N₂O₄ 472.1610, found 472.1611.

tert-Butyl (2-(4-acetylbenzoyl)phenyl)(4-methoxypyridin-2-yl)carbamate (**3e**). Yellow solid; mp 118–119 °C; $R_f = 0.50$ (*n*-hexane/ethyl acetate = 1/1); ¹H NMR (300 MHz, CDCl₃) δ 1.38 (s, 9 H), 2.62 (s, 3 H), 3.75 (s, 3 H), 6.46 (dd, J = 6.0, 2.4 Hz, 1 H), 6.99 (d, J = 2.4 Hz, 1 H), 7.38–7.44 (m, 3 H), 7.58 (td, J = 6.9, 1.8 Hz, 1 H), 7.79 (d, J = 8.1, 2 H), 7.89–7.97 (m, 3 H); ¹³C NMR (125 MHz, CDCl₃) δ 26.8 (CH₃), 28.0 (CH₃ × 3), 55.1 (CH₃), 82.0 (C_q), 104.3 (CH), 107.4 (CH), 126.7 (CH), 127.8 (CH × 2), 129.4 (CH), 130.1 (CH × 2), 130.7 (CH), 131.4 (CH), 136.7 (C_q), 139.6 (C_q), 139.9 (C_q), 140.9 (C_q), 148.1 (CH), 153.1 (C_q), 155.9 (C_q), 166.5 (C_q), 195.1 (C_q), 197.5 (C_q); MS (EI, *m/z*) 446 (M⁺, 1), 345 (61), 317 (81), 243 (92), 227 (88), 199 (100), 147 (76); HRMS *m/z* calcd for C₂₆H₂₆N₂O₅ 446.1842, found 446.1840.

tert-Butyl (2-(2-fluorobenzoyl)phenyl)(4-methoxypyridin-2-yl)carbamate (**3f**). Yellow solid; mp 166–167 °C; $R_f = 0.54$ (*n*hexane/ethyl acetate = 1/1); ¹H NMR (300 MHz, CDCl₃) δ 1.39 (s, 9 H), 3.79 (s, 3 H), 6.48 (dd, J = 5.7, 2.4 Hz, 1 H), 7.00–7.12 (m, 3 H), 7.34–7.59 (m, 6 H), 7.99 (d, J = 5.7 Hz, 1 H); ¹³C NMR (125 MHz, CDCl₃) δ 28.0 (CH₃ × 3), 55.2 (CH₃), 81.9 (C_q), 103.8 (CH), 107.4 (CH), 116.3 (d, $J_{C-F} = 21.5$ Hz, CH), 123.6 (d, $J_{C-F} = 3.6$ Hz, CH), 126.9 (CH), 127.0 (d, $J_{C-F} = 11.4$ Hz, C_q), 129.9 (CH), 130.9 (CH), 131.8 (d, $J_{C-F} = 11.0$ Hz, CH), 133.6 (d, $J_{C-F} = 8.8$ Hz, CH), 137.1 (C_q), 139.8 (C_q), 148.1 (CH), 153.1 (C_q), 156.1 (C_q), 160.8 (d, $J_{C-F} = 256.0$ Hz, C_q), 166.5 (C_q), 192.5 (C_q); MS (EI, *m/z*) 422 (M⁺, 1), 301 (65), 243 (90), 227 (55), 199 (100), 123 (71); HRMS *m/z* calcd for C₂₄H₂₃FN₂O₄ 422.1642, found 422.1642.

tert-Butyl (2-(3-fluorobenzoyl)phenyl)(4-methoxypyridin-2-yl)carbamate (**3g**). Yellow solid; mp 76–77 °C; $R_f = 0.43$ (*n*-hexane/ ethyl acetate = 2/1); ¹H NMR (300 MHz, CDCl₃) δ 1.39 (s, 9 H), 3.77 (s, 3 H), 6.46 (dd, J = 3.9, 1.8 Hz, 1 H), 6.98 (s, 1 H), 7.20–7.44 (m, 6 H), 7.50–7.57 (m, 2 H), 7.97 (d, J = 6.0 Hz, 1 H); ¹³C NMR (125 MHz, CDCl₃) δ 28.0 (CH₃ × 3), 55.2 (CH₃), 82.0 (C_q), 104.4 (CH), 107.6 (CH), 116.6 (d, $J_{C-F} = 22.8$ Hz, CH), 119.7 (d, $J_{C-F} =$ 21.9 Hz, CH), 125.8 (d, $J_{C-F} = 3.6$ Hz, CH), 126.6 (CH), 129.3 (CH), 129.6 (d, $J_{C-F} = 8.1$ Hz, CH), 130.6 (CH), 131.3 (CH), 136.6 (C_q), 139.5 (d, $J_{C-F} = 6.4$ Hz, C_q), 139.9 (C_q), 148.2 (CH), 153.1 (C_q), 156.0 (C_q), 162.3 (d, $J_{C-F} = 246.1$ Hz, C_q), 166.5 (C_q), 194.4 (C_q); MS (EI, m/z) 422 (M⁺, 1), 340 (44), 177 (65), 61 (94), 57 (100); HRMS m/z calcd for C₂₄H₂₃FN₂O₄ 422.1642, found 422.1640.

tert-Butyl (2-(4-fluorobenzoyl)phenyl)(4-methoxypyridin-2-yl)carbamate (**3h**). Yellow solid; mp 84–85 °C; $R_f = 0.60$ (*n*-hexane/ ethyl acetate = 1/1); ¹H NMR (300 MHz, CDCl₃) δ 1.38 (s, 9 H), 3.76 (s, 3 H), 6.46 (dd, J = 5.7, 2.4 Hz, 1 H), 6.98–7.05 (m, 3 H), 7.37–7.43 (m, 3 H), 7.63 (td, J = 8.4, 2.1 Hz, 1 H), 7.76 (t, J = 5.4Hz, 2 H), 7.96 (d, J = 6.0 Hz, 1 H); ¹³C NMR (125 MHz, CDCl₃) δ 28.1 (CH₃ × 3), 55.2 (CH₃), 82.0 (C_q), 104.6 (CH), 107.5 (CH), 115.1 (d, $J_{C-F} = 21.9$ Hz, CH), 126.6 (CH), 129.2 (CH), 130.6 (CH), 131.1 (CH), 133.6 (d, $J_{C-F} = 2.8$ Hz, C_q), 136.9 (C_q), 139.8 (C_q), 148.2 (CH), 153.1 (C_q), 156.0 (C_q), 164.5 (C_q), 165.5 (d, $J_{C-F} = 242.2$ Hz, C_q), 166.5 (C_q), 194.2 (C_q); MS (EI, *m/z*) 422 (M⁺, 1), 243 (43), 199 (100), 85 (44), 71 (42), 57 (41); HRMS *m/z* calcd for C₂₄H₂₃FN₂O₄ 422.1642, found 422.1645.

tert-Butyl (2-(2-chlorobenzoyl)phenyl)(4-methoxypyridin-2-yl)carbamate (**3i**). Yellow solid; mp 94–95 °C; $R_f = 0.40$ (*n*-hexane/ ethyl acetate = 2/1); ¹H NMR (300 MHz, CDCl₃) δ 1.41 (s, 9 H), 3.81 (s, 3 H), 6.49 (dd, J = 6.0, 2.1 Hz, 1 H), 7.15–7.19 (m, 2 H), 7.29–7.41 (m, 5 H), 7.49 (d, J = 7.2 Hz, 1 H), 7.56 (t, J = 7.5 Hz, 1 H), 7.98 (d, J = 6.0 Hz, 1 H); ¹³C NMR (125 MHz, CDCl₃) δ 28.1 (CH₃ × 3), 55.2 (CH₃), 81.9 (C_q), 103.5 (CH), 107.4 (CH), 126.2 (CH), 127.1 (CH), 130.2 (CH), 130.7 (CH), 131.3 (CH), 131.4 (CH), 131.5 (CH), 132.1 (C_q), 132.4 (CH), 136.1 (C_q), 138.2 (C_q), 140.3 (C_q), 147.9 (CH), 153.2 (C_q), 156.0 (C_q), 166.5 (C_q), 194.2 (C_q); MS (EI, *m*/*z*) 438 (M⁺, 1), 340 (61), 275 (65), 243 (97), 199 (100), 177 (99), 161 (74); HRMS *m*/*z* calcd for C₂₄H₂₃ClN₂O₄ 438.1346, found 438.1347.

tert-Butyl (2-(3-chlorobenzoyl)phenyl)(4-methoxypyridin-2-yl)carbamate (**3***j*). Yellow solid; mp 88–89 °C; $R_f = 0.45$ (*n*-hexane/ ethyl acetate = 2/1); ¹H NMR (300 MHz, CDCl₃) δ 1.39 (s, 9 H), 3.77 (s, 3 H), 6.46 (dd, J = 5.7, 2.1 Hz, 1 H), 6.99 (d, J = 1.8 Hz, 1 H), 7.29 (t, J = 7.8 Hz, 1 H), 7.38–7.48 (m, 4 H), 7.54–7.62 (m, 2 H), 7.70 (s, 1 H), 7.96 (d, J = 5.7 Hz, 1 H); ¹³C NMR (125 MHz, CDCl₃) δ 28.1 (CH₃ × 3), 55.2 (CH₃), 82.1 (C_q), 104.3 (CH), 107.6 (CH), 126.7 (CH), 128.1 (CH), 129.3 (CH), 129.9 (CH), 130.6 (CH), 131.3 (CH), 132.6 (CH), 134.0 (C_q), 136.6 (C_q), 139.0 (C_q), 139.8 (C_q), 148.2 (CH), 153.1 (C_q), 155.9 (C_q), 166.5 (C_q), 194.5 (C_q); MS (EI, *m*/z) 438 (M⁺, 1), 309 (40), 227 (78), 199 (100); HRMS *m*/z calcd for C₂₄H₂₃ClN₂O₄ 438.1346, found 438.1345.

tert-Butyl (2-(4-*chlorobenzoyl*)*phenyl*)(4-*methoxypyridin*-2-*y*))*carbamate* (**3k**). Yellow solid; mp 106–107 °C; $R_f = 0.48$ (*n*hexane/ethyl acetate = 2/1); ¹H NMR (300 MHz, CDCl₃) δ 1.38 (*s*, 9 H), 3.78 (*s*, 3 H), 6.47 (dd, J = 5.7, 1.8 Hz, 1 H), 6.98 (d, J = 1.8Hz, 1 H), 7.30–7.43 (m, 5 H), 7.56 (td, J = 7.2, 1.8 Hz, 1 H), 7.67 (d, J = 8.4 Hz, 2 H), 7.95 (d, J = 6.0 Hz, 1 H); ¹³C NMR (125 MHz, CDCl₃) δ 28.1 (CH₃ × 3), 55.2 (CH₃), 82.0 (C_q), 104.5 (CH), 107.5 (CH), 126.6 (CH), 128.3 (CH × 2), 129.2 (CH), 130.7 (CH), 131.2 (CH), 131.4 (CH × 2), 135.7 (C_q), 136.8 (C_q), 139.1 (C_q), 139.8 (C_q), 148.2 (CH), 153.1 (C_q), 156.0 (C_q), 166.5 (C_q), 194.5 (C_q); MS (EI, m/z) 438 (M⁺, 1), 309 (49), 243 (81), 227 (64), 199 (100); HRMS m/z calcd for $C_{24}H_{23}^{35}ClN_2O_4$ 438.1346, found 438.1343.

tert-Butyl (2-(3-bromobenzoyl)phenyl)(4-methoxypyridin-2-yl)carbamate (**3**). Yellow solid; mp 78–79 °C; $R_f = 0.43$ (*n*-hexane/ ethyl acetate = 2/1); ¹H NMR (300 MHz, CDCl₃) δ 1.39 (s, 9 H), 3.77 (s, 3 H), 6.46 (dd, J = 5.7, 2.1 Hz, 1 H), 6.99 (s, 1 H), 7.23–7.25 (t, J = 7.8 Hz, 1 H), 7.38–7.43 (m, 3 H), 7.54–7.66 (m, 3 H), 7.84 (s, 1 H), 7.96 (d, J = 5.7 Hz, 1 H); ¹³C NMR (125 MHz, CDCl₃) δ 28.1 (CH₃ × 3), 55.2 (CH₃), 82.1 (C_q), 104.3 (CH), 107.6 (CH), 126.7 (CH), 128.5 (CH), 129.5 (CH), 130.6 (CH), 131.3 (CH), 132.8 (CH), 135.5 (CH), 136.6 (C_q), 139.2 (C_q), 139.8 (C_q), 148.2 (CH), 153.1 (C_q), 155.9 (C_q), 166.5 (C_q), 194.4 (C_q); MS (EI, m/z) 482 (M⁺, 1), 353 (39), 299 (37), 243 (87), 227 (87), 199 (100); HRMS m/z calcd for C₂₄H₂₃BrN₂O₄ 482.0841, found 482.0843.

tert-Butyl (2-(4-*bromobenzoyl*)*phenyl*)(4-*methoxypyridin*-2-*yl*)*carbamate* (**3m**). Yellow solid; mp 82–83 °C; $R_f = 0.48$ (*n*hexane/ethyl acetate = 2/1); ¹H NMR (300 MHz, CDCl₃) δ 1.38 (s, 9 H), 3.77 (s, 3 H), 6.47 (dd, J = 5.4, 1.8 Hz, 1 H), 6.98 (s, 1 H), 7.37–7.60 (m, 8 H), 7.95 (d, J = 5.7, 1 H); ¹³C NMR (125 MHz, CDCl₃) δ 28.1 (CH₃ × 3), 55.2 (CH₃), 82.0 (C_q), 104.5 (CH), 107.5 (CH), 126.7 (CH), 127.8 (C_q), 129.3 (CH), 130.7 (CH), 131.2 (CH × 2), 131.5 (CH × 2), 136.1 (C_q), 136.7 (C_q), 139.8 (C_q), 148.2 (CH), 153.1 (C_q), 156.0 (C_q), 166.5 (C_q), 194.7 (C_q); MS (EI, *m/z*) 482 (M⁺, 1), 355 (27), 227 (52), 199 (100). HRMS *m/z* calcd for C₂₄H₂₃BrN₂O₄ 482.0841, found 482.0844.

tert-Butyl (2-(3,5-dimethylbenzoyl)phenyl)(4-methoxypyridin-2yl)carbamate (**3n**). Yellow solid; mp 132–133 °C; $R_f = 0.63$ (*n*-hexane/ethyl acetate = 1/1); ¹H NMR (300 MHz, CDCl₃) δ 1.37 (s, 9 H), 2.26 (s, 6 H), 3.74 (s, 3 H), 6.44 (dd, J = 5.4, 2.1 Hz, 1 H), 6.95 (d, J = 2.4 Hz, 1 H), 7.12 (s, 1 H), 7.29 (s, 2 H), 7.34–7.46 (m, 3 H), 7.52–7.58 (m, 1 H), 7.98 (d, J = 5.7 Hz, 1 H); ¹³C NMR (125 MHz, CDCl₃) δ 21.1 (CH₃ × 2), 28.0 (CH₃ × 3), 55.1 (CH₃), 81.8 (C_q), 104.6 (CH), 107.6 (CH), 126.6 (CH), 127.8 (CH × 2), 129.4 (CH), 130.7 (CH), 130.8 (CH), 134.5 (CH), 137.4 (C_q), 137.5 (C_q), 137.6 (C_q), 139.6 (C_q), 148.1 (CH), 153.1 (C_q), 156.2 (C_q), 166.3 (C_q), 199 (100); HRMS *m*/*z* calcd for C₂₆H₂₈N₂O₄ 432.2049, found 432.2047.

tert-Butyl (2-(2,4-dimethylbenzoyl)phenyl)(4-methoxypyridin-2-yl)carbamate (**3o**). Yellow solid; mp 90–91 °C; $R_f = 0.48$ (*n*-hexane/ethyl acetate = 2/1); ¹H NMR (300 MHz, CDCl₃) δ 1.40 (s, 9 H), 2.29 (s, 3 H), 2.31 (s, 3 H), 3.77 (s, 3 H), 6.46 (dd, J = 5.7, 2.1 Hz, 1 H), 6.84–6.90 (m, 2 H), 6.98 (s, 1 H), 7.34 (d, J = 6.6 Hz 1 H), 7.32–7.46 (m, 3 H), 7.51 (td, J = 7.2, 1.5 Hz 1 H), 7.98 (d, J = 5.7 Hz, 1 H); ¹³C NMR (125 MHz, CDCl₃) δ 20.7 (CH₃), 21.3 (CH₃), 28.1 (CH₃ × 3), 55.1 (CH₃), 81.8 (C_q), 104.1 (CH), 107.2 (CH), 131.9 (CH), 132.0 (CH), 134.8 (C_q), 138.1 (C_q), 138.9 (C_q), 139.7 (C_q), 141.6 (C_q), 148.0 (CH), 153.1 (C_q), 156.1 (C_q), 166.3 (C_q), 197.3 (C_q); MS (EI, *m/z*) 432 (M⁺, 1), 303 (41), 243 (88), 199 (100), 124 (37); HRMS *m/z* calcd for C₂₆H₂₈N₂O₄ 432.2049, found 432.2046.

tert-Butyl (2-(3,5-dimethoxybenzoyl)phenyl)(4-methoxypyridin-2-yl)carbamate (**3p**). Yellow solid; mp 86–87 °C; $R_f = 0.63$ (*n*-hexane/ethyl acetate = 1/1); ¹H NMR (300 MHz, CDCl₃) δ 1.39 (s, 9 H), 3.75 (s, 9 H), 6.46 (dd, J = 5.7, 1.8 Hz, 1 H), 6.59 (s, 1 H), 6.82 (d, J = 1.8 Hz, 2 H), 6.92 (s, 1 H), 7.37–7.56 (m, 4 H), 7.98 (d, J = 6.0 Hz, 1 H); ¹³C NMR (125 MHz, CDCl₃) δ 28.0 (CH₃ × 3), 55.1 (CH₃), 55.5 (CH₃ × 2), 81.9 (C_q), 104.7 (CH), 105.5 (CH), 107.5 (CH), 107.6 (CH × 2), 126.6 (CH), 129.5 (CH), 130.8 (CH), 131.0 (CH), 137.0 (C_q), 139.4 (C_q), 139.8 (C_q), 148.1 (CH2), 153.1 (C_q), 156.1 (C_q), 160.3 (C_q × 2), 166.3 (C_q), 195.5 (C_q); MS (EI, *m/z*) 464 (M⁺, 1), 335 (38), 227 (100), 199 (91); HRMS *m/z* calcd for C₂₆H₂₈N₂O₆ 464.1947, found 464.1950.

tert-Butyl (2-(2,4-dimethoxybenzoyl)phenyl)(4-methoxypyridin-2-yl)carbamate (**3q**). White solid; mp 68–69 °C; $R_f = 0.47$ (*n*-hexane/ethyl acetate = 1/2); ¹H NMR (300 MHz, CDCl₃) δ 1.39 (s, 9 H), 3.59 (s, 3 H), 3.76 (s, 3 H), 3.80 (s, 3 H), 6.30–6.34 (m, 2 H), 6.46 (dd, J = 5.7, 2.1 Hz, 1 H), 6.96 (d, J = 2.1 Hz,1 H), 7.33 (t, J =8.1 Hz, 2 H), 7.42 (d, J = 8.1 Hz, 2 H), 7.51 (td, J = 6.8, 1.5 Hz, 1 H), 8.00 (d, J = 5.7 Hz, 1 H); ¹³C NMR (125 MHz, CDCl₃) δ 28.0 (CH₃ × 3), 55.0 (CH₃), 55.4 (CH₃ × 2), 81.6 (C_q), 98.3 (CH), 107.3 (CH), 121.2 (C_q), 126.7 (CH), 129.6 (CH), 130.5 (CH), 131.1 (CH), 134.0 (CH), 139.1 (C_q), 139.2 (C_q), 147.9 (CH), 153.2 (C_q), 156.1 (C_q), 160.5 (C_q), 163.8 (C_q), 166.2 (C_q), 194.0 (C_q); MS (EI, m/z) 464 (M⁺, 1), 305 (42), 243 (85), 199 (100), 165 (47), 57 (70); HRMS m/z calcd for C₂₆H₂₈N₂O₆ 464.1947, found 464.1949.

tert-Butyl (2-(3-formylbenzoyl)phenyl)(4-methoxypyridin-2-yl)carbamate (**3r**). Yellow solid; mp 88–89 °C; $R_f = 0.63$ (*n*-hexane/ ethyl acetate = 1/1); ¹H NMR (300 MHz, CDCl₃) δ 1.37 (s, 9 H), 3.75 (s, 3 H), 6.45 (dd, J = 5.7, 2.4 Hz, 1 H), 6.98 (d, J = 2.1 Hz, 1 H), 7.38–7.48 (m, 3 H), 7.55–7.61 (m, 2 H), 7.95 (d, J = 6.0 Hz, 1 H), 8.03 (dd, J = 7.5, 1.5 Hz, 2 H), 8.19 (s, 1 H), 9.98 (s, 1 H); ¹³C NMR (125 MHz, CDCl₃) δ 28.0 (CH₃ × 3), 55.2 (CH₃), 82.2 (Cq), 104.5 (CH), 107.4 (CH), 126.8 (CH), 128.8 (CH), 129.3 (CH), 130.6 (CH), 131.5 (CH), 132.2 (CH), 132.3 (CH), 135.4 (CH), 136.1 (C_q), 136.5 (C_q), 138.2 (C_q), 139.9 (C_q), 148.2 (CH), 153.1 (C_q), 155.9 (C_q), 166.5 (C_q), 191.5 (CH), 194.6 (C_q); MS (EI, m/z) 432 (M⁺, 1), 303 (27), 227 (66), 199 (100), 184 (28); HRMS m/zcalcd for C₂₅H₂₄N₂O₅ 432.1685, found 432.1683.

Methyl (2,6-dibenzoylphenyl)(4-methoxypyridin-2-yl)carbamate (4-OMe/Moc). White oil; $R_f = 0.63$ (*n*-hexane/ethyl acetate = 1/1); ¹H NMR (CDCl₃, 400 MHz) δ 3.62 (s, 3 H), 3.67 (s, 3 H), 6.32 (dd, J = 4.4, 2.4 Hz, 1 H), 6.99 (bs, 1 H), 7.35 (td, J = 7.6, 1.2 Hz, 4 H), 7.51 (td, J = 8.4, 1.2 Hz, 3 H), 7.62 (dd, J = 7.2, 0.8 Hz, 2 H), 7.71 (d, J = 5.6 Hz, 1 H), 7.74–7.77 (m, 4 H); ¹³C NMR (CDCl₃, 100 MHz) δ 53.2 (CH₃), 55.0 (CH₃), 102.1 (C_q), 107.7 (CH), 126.5 (CH), 127.9 (CH × 4), 130.0 (CH × 4), 131.2 (CH × 2), 132.9 (CH × 2), 137.1(C_q × 2), 139.3(C_q × 2), 147.1 (CH), 154.4 (C_q), 154.9 (C_q), 166.4 (C_q × 2), 195.2 (C_q); MS (EI.m/z) 466 (M⁺, 1), 362 (58), 361 (100), 317 (51); HRMS (EI-magnetic sector) m/z [M⁺] calcd for C₂₈H₂₂N₂O₅ 466.1529, found 466.1530.

N-(2,6-*Dibenzoylphenyl*)-*N*-(4-*methoxypyridin*-2-*yl*)*pivalamide* (4-OMe/Piv). White oil; $R_f = 0.66$ (*n*-hexane/ethyl acetate = 1/1); ¹H NMR (CDCl₃, 400 MHz) δ 0.76 (9 H), 3.73 (3 H), 6.52 (dd, *J* = 3.6, 1.6 Hz, 1 H), 7.01 (d, *J* = 1.2 Hz, 1 H), 7.39–7.47 (m, 7 H), 7.54 (t, *J* = 5.2 Hz, 2 H), 7.83 (d, *J* = 4.4 Hz, 4 H), 7.90 (d, *J* = 3.6 Hz, 1 H); ¹³C NMR (CDCl₃, 100 MHz) δ 28.1 (CH₃× 3), 41.3 (C_q), 55.2 (CH₃), 107.9 (CH), 109.2 (CH), 126.1 (CH), 128.1 (CH × 4), 130.5 (CH × 4), 130.9 (CH × 2), 133.0 (CH × 2), 136.9 (C_q × 2), 139.4 (C_q × 2), 140.2 (C_q), 147.9 (CH), 157.2 (C_q), 167.0 (C_q), 179.7 (C_q), 195.5 (C_q × 2); MS (EI, *m/z*) 492 (M⁺, 1), 369 (90), 353 (51), 347 (100), 305 (43); HRMS (EI-magnetic sector) *m/z* [M⁺] calcd for C₃₁H₂₈N₂O₄ 492.2049, found.492.2051.

tert-Butyl (2,6-*dibenzoylphenyl*)(4-*methoxypyridin-2-yl*)*carbamate* (4a). White solid; mp 126–127 °C; $R_f = 0.48$ (*n*-hexane/ethyl acetate = 2/1); ¹H NMR (300 MHz, CDCl₃) δ 1.36 (s, 9 H), 3.64 (s, 3 H), 6.28 (dd, J = 5.7, 2.1 Hz, 1 H), 6.81 (bs, 1 H), 7.32 (t, J = 7.5 Hz, 4 H), 7.48 (t, J = 7.8 Hz, 3 H), 7.62 (d, J = 7.5 Hz, 2 H), 7.66 (d, J = 5.7 Hz, 1 H), 7.76 (d, J = 7.2 Hz, 4 H); ¹³C NMR (125 MHz, CDCl₃) δ 27.9 (CH₃ × 3), 54.9 (CH₃), 82.2 (C_q), 101.8 (C_q), 107.2 (CH), 126.3 (CH), 127.9 (CH × 4), 130.0 (CH × 4), 131.2 (CH), 132.6 (CH × 2), 137.3 (C_q × 2), 139.2 (C_q × 2), 146.8 (CH), 152.6 (C_q), 155.1 (C_q), 166.1 (C_q), 195.2 (C_q × 2); MS (EI, *m/z*) 508 (M⁺, 1), 331 (39), 303 (100), 213 (21), 105 (22); HRMS *m/z* calcd for C₃₁H₂₈N₂O₅ 508.1998, found 508.2000.

tert-Butyl (2,6-bis(4-methoxybenzoyl)phenyl)(4-methoxypyridin-2-yl)carbamate (**4b**). Yellow solid; mp 138–139 °C; $R_f = 0.49$ (*n*-hexane/ethyl acetate = 1/1); ¹H NMR (300 MHz, CDCl₃) δ 1.35 (s, 9 H), 3.66 (s, 3 H), 3.83 (s, 6 H), 6.28 (dd, J = 5.7, 2.1 Hz, 1 H), 6.80 (d, J = 8.7 Hz, 4 H), 6.85–6.93 (m, 1 H), 7.45 (t, J = 6.6 Hz, 1 H), 7.57 (d, J = 7.5 Hz, 2 H), 7.68 (d, J = 7.5 Hz, 1 H), 7.74 (d, J = 8.7 Hz, 4 H); ¹³C NMR (125 MHz, CDCl₃) δ 27.9 (CH₃ × 3), 54.9 (CH₃), 55.4 (CH₃ × 2), 82.0 (C_q), 102.0 (CH), 107.1 (CH), 113.1 (CH × 4), 126.1 (CH), 130.2 (C_q × 2), 130.7 (CH × 2), 132.4 (CH × 4), 137.4 (C_q), 139.5 (C_q × 2), 147.0 (CH), 152.6 (C_q), 155.4 (C_q), 163.2 (C_q × 2), 166.1 (C_q), 193.9 (C_q × 2); MS (EI, *m*/z) 568 (M⁺, 1), 361 (93), 333 (100), 225 (94), 135 (93); HRMS *m*/z calcd for C₃₃H₃₂N₂O₇ 568.2210, found 568.2209. tert-Butyl (2,6-bis(2-fluorobenzoyl)phenyl)(4-methoxypyridin-2-yl)carbamate (4f). Yellow solid; mp 140–141 °C; $R_f = 0.61$ (*n*-hexane/ethyl acetate = 1/1); ¹H NMR (300 MHz, CDCl₃) δ 1.38 (s, 9 H), 3.68 (s, 3 H), 6.31 (dd, J = 5.7, 2.1 Hz, 1 H), 6.87 (s, 1 H), 6.95–7.08 (m, 4 H), 7.35–7.43 (m, 2 H), 7.46–7.56 (m, 3 H), 7.67–7.73 (m, 3 H); ¹³C NMR (125 MHz, CDCl₃) δ 27.9 (CH₃ × 3), 55.0 (CH₃), 82.3 (C_q), 107.1 (CH), 116.3 (d, $J_{C-F} = 21.5$ Hz, CH × 2), 123.6 (d, $J_{C-F} = 3.6$ Hz, CH × 2), 126.7 (d, $J_{C-F} = 11.5$ Hz, C_q × 2), 127.0 (CH), 131.6 (CH × 2), 131.9 (CH × 2), 133.5 (d, $J_{C-F} = 8.6$ Hz, CH × 2), 137.3 (C_q), 139.5 (C_q × 2), 146.7 (CH), 152.5 (C_q), 154.6 (C_q), 160.8 (d, $J_{C-F} = 256.0$ Hz, C_q), 166.1 (C_q × 2), 192.1 (C_q × 2); MS (EI, *m*/z) 544 (M⁺, 1), 321 (78), 264 (31), 158 (100); HRMS *m*/z calcd for C₃₁H₂₆F₂N₂O₅ 544.1810, found 544.1808.

tert-Butyl (2,6-bis(3-fluorobenzoyl)phenyl)(4-methoxypyridin-2yl)carbamate (**4g**). Yellow solid; mp 128–129 °C; $R_f = 0.55$ (*n*-hexane/ethyl acetate = 2/1); ¹H NMR (300 MHz, CDCl₃) δ 1.37 (s, 9 H), 3.67 (s, 3 H), 6.30 (dd, J = 5.7, 1.2 Hz, 1 H), 6.85 (s, 1 H), 7.17 (t, J = 4.5 Hz, 2 H), 7.26–7.33 (m, 2 H), 7.45 (d, J = 9.0 Hz, 2 H), 7.52–7.54 (m, 3 H), 7.61–7.67 (m, 3 H); ¹³C NMR (125 MHz, CDCl₃) δ 27.9 (CH₃ × 3), 55.0 (CH₃), 82.5 (C_q), 101.8 (CH), 107.3 (CH × 2), 116.6 (d, $J_{C-F} = 21.9$ Hz, CH × 2), 119.7 (d, $J_{C-F} = 21.9$ Hz, CH × 2), 125.7 (d, $J_{C-F} = 2.8$ Hz, CH × 2), 126.6 (CH), 129.6 (d, $J_{C-F} = 7.3$ Hz, CH × 2), 131.1 (CH × 2), 137.6 (C_q), 138.9 (C_q × 2), 139.4 (d, $J_{C-F} = 246.1$ Hz, C_q × 2), 166.3 (C_q), 193.8 (d, $J_{C-F} = 1.9$ Hz, C_q × 2); MS (EI, m/z) S44 (M⁺, 1), 340 (62), 177 (100), 161 (72), S7 (61); HRMS m/z calcd for C₃₁H₂₆F₂N₂O₅ S44.1810, found S44.1808.

tert-Butyl (2,6-bis(4-fluorobenzoyl)phenyl)(4-methoxypyridin-2-yl)carbamate (**4h**). Yellow solid; mp 132–133 °C; $R_f = 0.53$ (n-hexane/ethyl acetate = 2/1); ¹H NMR (300 MHz, CDCl₃) δ 1.36 (s, 9 H), 3.67 (s, 3 H), 6.31 (dd, J = 5.4, 1.5 Hz, 1 H), 6.85 (s, 1 H), 6.99 (t, J = 8.4 Hz, 4 H), 7.45 (t, J = 8.1 Hz, 1 H), 7.58–7.61 (m, 2 H), 7.65 (d, J = 5.7 Hz, 1 H), 7.79 (td, J = 5.7, 2.4 Hz, 4 H); ¹³C NMR (125 MHz, CDCl₃) δ 27.9 (CH₃ × 3), 55.0 (CH₃), 82.4 (C_q), 107.1 (CH), 115.0 (d, $J_{C-F} = 21.9$ Hz, CH × 4), 126.5 (CH), 131.0 (CH × 2), 132.6 (d, $J_{C-F} = 9.1$ Hz, CH × 4), 133.5 (d, $J_{C-F} = 2.8$ Hz, $C_q \times 2$), 137.6 (C_q), 139.1 (C_q × 2), 146.9 (CH), 152.5 (C_q), 155.0 (C_q), 164.5 (C_q), 166.4 ($J_{C-F} = 24.6$ Hz, $C_q \times 2$), 193.6 (C_q × 2); MS (EI, m/z) 544 (M⁺, 1), 349 (49), 321 (100), 123 (36); HRMS m/z calcd for C₃₁H₂₆F₂N₂O₅ 544.1810, found 544.1811.

tert-Butyl (2,6-bis(3-chlorobenzoyl)phenyl)(4-methoxypyridin-2yl)carbamate (4j). Yellow solid; mp 138–139 °C; $R_f = 0.55$ (*n*-hexane/ethyl acetate = 2/1); ¹H NMR (300 MHz, CDCl₃) δ 1.38 (s, 9 H), 3.67 (s, 3 H), 6.31 (dd, J = 5.7, 2.1 Hz, 1 H), 6.83 (s,1 H), 7.27 (t, J = 7.8 Hz, 2 H), 7.43 (dd, J = 2.1, 1.2 Hz, 2 H), 7.45–7.54 (m, 1 H), 7.60–7.67 (m, 5 H), 7.73 (s, 2 H); ¹³C NMR (125 MHz, CDCl₃) δ 27.9 (CH₃ × 3), 55.0 (CH₃), 82.5 (C_q), 101.6 (C_q), 107.4 (CH), 126.7 (CH), 127.9 (CH × 2), 129.3 (CH × 2), 129.9 (CH × 2), 131.3 (CH), 132.6 (CH × 2), 134.3 (C_q × 2), 137.6 (C_q), 138.9 (C_q × 2), 146.9 (CH), 152.5 (C_q), 154.7 (C_q), 166.3 (C_q), 193.8 (C_q × 2); MS (EI, *m/z*) 576 (M⁺, 1), 365 (100), 337 (98), 225 (32); HRMS *m/z* calcd for C₃₁H₂₆C₁₂N₂O₅ 576.1219, found 576.1217.

tert-Butyl (2,6-bis(4-chlorobenzoyl)phenyl)(4-methoxypyridin-2yl)carbamate (**4k**). Yellow solid; mp 180–181 °C; $R_f = 0.65$ (*n*-hexane/ethyl acetate = 2/1); ¹H NMR (300 MHz, CDCl₃) δ 1.36 (s, 9 H), 3.68 (s, 3 H), 6.32 (dd, J = 6.0, 2.1 Hz, 1 H), 6.79 (s, 1 H), 7.29 (d, J = 8.7 Hz, 4 H), 7.50 (t, J = 6.9 Hz, 1 H), 7.62–7.70 (m, 7 H); ¹³C NMR (125 MHz, CDCl₃) δ 27.9 (CH₃ × 3), 55.0 (CH₃), 82.4 (C_q), 107.1 (CH), 126.6 (CH), 128.2 (CH × 4), 131.0 (CH), 131.3 (CH × 4), 135.6 (C_q × 2), 139.0 (C_q × 2), 139.1 (C_q × 2), 146.8 (CH), 152.5 (C_q), 154.7 (C_q), 166.4 (C_q), 194.0 (C_q × 2); MS (EI, m/z) 576 (M⁺, 1), 365 (67), 337 (100), 139 (43); HRMS m/z calcd for C₃₁H₂₆C₁₂N₂O₅ 576.1219, found 576.1216.

tert-Butyl (2,6-bis(3,5-dimethylbenzoyl)phenyl)(4-methoxypyridin-2-yl)carbamate (**4n**). Yellow solid; mp 154–155 °C; $R_f = 0.60$ (*n*-hexane/ethyl acetate = 2/1); ¹H NMR (300 MHz, CDCl₃) δ 1.36 (s, 9 H), 2.26 (s, 12 H), 3.65 (s, 3 H), 6.27 (dd, J = 5.7, 2.1 Hz, 1 H), 6.90 (m, 1 H), 7.10 (m, 2 H), 7.39 (m, 4 H), 7.46 (t, J = 7.2 Hz, 1 H), 7.60 (d, J = 7.2 Hz, 2 H), 7.66 (d, J = 5.7 Hz, 1 H); ¹³C NMR (125 MHz, CDCl₃) δ 21.1 (CH₃ × 4), 27.9 (CH₃ × 3), 54.9 (CH₃), 81.9 (C_q), 107.2 (CH), 126.1 (CH), 127.8 (CH × 4), 131.0 (CH), 134.3 (CH × 2), 137.4 (C_q × 2), 137.5 (C_q × 4), 139.5 (C_q × 2), 146.7 (CH), 152.6 (C_q), 155.2 (C_q), 166.1 (C_q), 195.6 (C_q × 2); MS (EI, *m*/*z*) 564 (M⁺, 1), 331 (34), 242 (84), 133 (100); HRMS *m*/*z* calcd for C₃₅H₃₆N₂O₅ 564.2624, found 564.2626.

tert-Butyl (2,6-bis(2,4-dimethylbenzoyl)phenyl)(4-methoxypyridin-2-yl)carbamate (40). White solid; mp 156–157 °C; $R_f = 0.65$ (*n*-hexane/ethyl acetate = 2/1); ¹H NMR (300 MHz, CDCl₃) δ 1.38 (s, 9 H), 2.28 (s, 6 H), 2.40 (s, 6H), 3.67 (s, 3 H), 6.29 (dd, J = 5.7, 2.1 Hz, 1 H), 6.74–6.80 (m, 3 H), 6.97 (s, 2 H), 7.27 (d, J = 8.7 Hz, 2 H), 7.40 (t, J = 7.8 Hz, 1 H), 7.57 (d, J = 7.5 Hz, 2 H), 7.73 (d, J = 5.7 Hz, 1 H); ¹³C NMR (125 MHz, CDCl₃) δ 20.5 (CH₃ × 2), 21.3 (CH₃ × 2), 28.0 (CH₃ × 3), 54.8 (CH₃), 81.9 (C_q), 106.8 (CH), 132.0 (CH), 134.9 (C_q × 2), 137.7 (C_q), 138.7 (C_q), 140.3 (C_q), 141.3 (C_q × 2), 146.7 (CH), 152.9 (C_q), 155.0 (C_q), 166.0 (C_q × 2), 196.9 (C_q × 2); MS (EI, *m*/z) 564 (M⁺, 1), 359 (38), 331 (100), 225 (52), 133 (60); HRMS *m*/z calcd for C₃₅H₃₆N₂O₅ 564.2624, found 564.2626.

tert-Butyl (2,6-bis(2,4-dimethoxybenzoyl)phenyl)(4-methoxypyridin-2-yl)carbamate (4q). Yellow solid; mp 96–97 °C; $R_f = 0.30$ (*n*-hexane/ethyl acetate = 1/2); ¹H NMR (300 MHz, CDCl₃) δ 1.39 (s, 9 H), 3.61 (s, 6 H), 3.67 (s, 3 H), 3.79 (s, 6 H), 6.26–6.34 (m, 5 H), 6.83 (s, 1 H), 7.37 (t, J = 7.8 Hz, 1 H), 7.46 (d, J = 8.7 Hz, 2 H), 7.56 (d, J = 7.8 Hz, 2 H), 7.67 (d, J = 5.7 Hz, 1 H); ¹³C NMR (125 MHz, CDCl₃) δ 27.9 (CH₃ × 3), 54.8 (CH₃), 55.3 (CH₃ × 2), 55.4 (CH₃ × 2), 81.7 (C_q), 98.2 (CH × 2), 101.4 (CH), 103.8 (CH × 2), 106.7 (CH), 121.2 (C_q), 126.2 (CH), 131.1 (CH × 2), 134.1 (CH × 2), 136.5 (C_q), 141.0 (C_q × 2), 146.3 (CH), 152.6 (C_q), 155.0 (C_q), 160.5 (C_q × 2), 163.6 (C_q × 2), 165.8 (C_q), 193.5 (C_q × 2); MS (EI, *m/z*) 628 (M⁺, 1), 363 (100), 225 (96), 165 (32); HRMS *m/z* calcd for C₃₅H₃₆N₂O₉ 628.2421, found 628.2421.

tert-Butyl (2-benzoyl-6-methoxyphenyl)(4-methoxypyridin-2-yl)carbamate (**5b**). White solid; mp 112–113 °C; $R_f = 0.45$ (*n*-hexane/ ethyl acetate = 1/1); ¹H NMR (300 MHz, CDCl₃) δ 1.40 (s, 9 H), 3.72 (s, 3 H), 3.89 (s, 3 H), 6.38 (dd, J = 5.7, 1.8 Hz, 1 H), 6.92 (s, 1 H), 7.00 (d, J = 7.5 Hz, 1 H), 7.13 (d, J = 8.1 Hz, 1 H), 7.30–7.40 (m, 3 H), 7.48 (t, J = 7.2 Hz, 1 H), 7.67 (d, J = 7.5 Hz, 2 H), 7.90 (d, J = 5.7 Hz, 1 H); ¹³C NMR (125 MHz, CDCl₃) δ 28.0 (CH₃ × 3), 55.0 (CH₃), 56.0 (CH₃), 81.2 (C_q), 102.8 (CH), 107.0 (CH), 113.5 (CH), 120.8 (CH), 127.8 (CH × 2), 127.9 (CH), 128.8 (C_q), 129.9 (CH × 2), 132.7 (CH), 137.4 (C_q), 138.4 (C_q), 148.1 (CH), 152.9 (C_q), 155.9 (Cq), 156.1 (C_q), 166.1 (C_q), 195.8 (C_q); MS (EI, m/z) 434 (M⁺, 1), 329 (36), 273 (100), 257 (31), 229 (100); HRMS m/zcalcd for C₂₅H₂₆N₂O₅ 434.1842, found 434.1844.

tert-Butyl (2-benzoyl-5-methoxyphenyl)(4-methoxypyridin-2-yl)carbamate (5c). Yellow solid; mp 68–69 °C; $R_f = 0.54$ (*n*-hexane/ ethyl acetate = 1/1); ¹H NMR (300 MHz, CDCl₃) δ 1.39 (s, 9 H), 3.76 (s, 3 H), 3.87 (s, 3 H), 6.48 (dd, J = 5.7, 2.1 Hz, 1 H), 6.87 (dd, J = 8.4, 2.4 Hz, 1 H), 6.97 (dd, J = 5.7, 2.1 Hz, 2 H), 7.33–7.43 (m, 3 H), 7.50 (t, J = 7.2 Hz, 1 H), 7.70 (d, J = 7.2 Hz, 2 H), 8.01 (d, J = 5.7Hz, 1 H); ¹³C NMR (125 MHz, CDCl₃) δ 28.1 (CH₃ × 3), 55.1 (CH₃), 55.5 (CH₃), 81.8 (C_q), 104.7 (CH), 107.7 (CH), 112.0 (CH), 116.1 (CH), 127.9 (CH × 2), 129.3 (C_q), 130.0 (CH × 2), 131.8 (CH), 132.3 (CH), 138.1 (C_q), 142.0 (C_q), 148.3 (CH), 153.0 (C_q), 156.2 (C_q), 161.8 (C_q), 166.4 (C_q), 195.0 (C_q); MS (EI, m/z) 434 (M⁺, 1), 303 (100), 229 (47), 105 (39); HRMS m/z calcd for C₂₅H₂₆N₂O₅ 434.1842, found 434.1843.

tert-Butyl (2-benzoyl-4-methoxyphenyl)(4-methoxypyridin-2-yl)carbamate (**5d**). Yellow solid; mp 128–129 °C; $R_f = 0.30$ (*n*-hexane/ ethyl acetate = 3/1); ¹H NMR (300 MHz, CDCl₃) δ 1.38 (s, 9 H), 3.73 (s, 3 H), 3.82 (s, 3 H), 6.44 (dd, J = 5.7, 1.8 Hz, 1 H), 6.84 (d, J = 1.8 Hz, 1 H), 6.93 (d, J = 2.7 Hz, 1 H), 7.08 (dd, J = 8.7, 3.0 Hz, 1 H), 7.32–7.37 (m, 3 H), 7.48 (t, J = 7.2 Hz, 1 H), 7.72 (d, J = 7.2 Hz, 2 H), 7.98 (d, J = 5.7 Hz, 1 H); ¹³C NMR (125 MHz, CDCl₃) δ 28.1 (CH₃ × 3), 55.1 (CH₃), 55.6 (CH₃), 81.7 (C_q), 104.5 (CH), 107.5 (CH), 114.2 (CH), 116.6 (CH), 128.0 (CH × 2), 130.0 (CH × 2), 131.9 (CH), 132.4 (C_q), 132.8 (CH), 137.1 (C_q), 137.9 (C_q), 148.2 (CH), 153.3 (C_q), 156.2 (C_q), 157.8 (C_q), 166.3 (C_q), 195.6 (C_q); MS (EI, m/z) 434 (M⁺, 1), 273 (39), 257 (23), 229 (100), 105 (26), 57 (37); HRMS m/z calcd for $C_{25}H_{26}N_2O_5$ 434.1842, found 434.1844.

tert-Butyl (2-*benzoyl*-4-fluorophenyl)(4-methoxypyridin-2-yl)carbamate (**5e**). Yellow solid; mp 132–133 °C; $R_f = 0.50$ (*n*hexane/ethyl acetate = 3/1); ¹H NMR (300 MHz, CDCl₃) δ 1.38 (s, 9 H), 3.73 (s, 3 H), 6.45 (dd, J = 5.7, 1.8 Hz, 1 H), 6.93 (d, J = 1.2Hz, 1 H), 7.13 (dd, J = 8.1, 2.7 Hz, 1 H), 7.24–7.25 (m, 1 H), 7.33– 7.43 (m, 3 H), 7.51 (t, J = 4.5 Hz, 1 H), 7.70 (d, J = 7.8 Hz, 2 H), 7.94 (d, J = 5.7 Hz, 1 H); ¹³C NMR (125 MHz, CDCl₃) δ 28.1 (CH₃ × 3), 55.1 (CH₃), 82.1 (C_q), 104.3 (CH), 107.6 (CH), 116.1 (d, J_{C-F} = 23.6 Hz, CH), 117.8 (d, $J_{C-F} = 22.4$ Hz, CH), 128.1 (CH × 2), 130.0 (CH × 2), 132.6 (d, $J_{C-F} = 8.1$ Hz, CH), 133.0 (CH), 135.6 (d, $J_{C-F} = 3.3$ Hz, C_q), 136.7 (C_q), 138.7 (d, $J_{C-F} = 6.9$ Hz, C_q), 148.1 (CH), 153.0 (C_q); 155.8 (C_q), 160.1 (d, $J_{C-F} = 247.4$ Hz, C_q), 166.4 (C_q), 194.3 (C_q); MS (EI, *m*/z) 422 (M⁺, 1), 293 (35), 261 (45), 245 (44), 217 (100), 57 (44); HRMS *m*/z calcd for C₂₄H₂₃FN₂O₄ 422.1642, found 422.1642.

tert-Butyl (2-*benzoyl-5-fluorophenyl*)(4-*methoxypyridin-2-yl*)*carbamate* (**5f**). Yellow solid; mp 132–133 °C; R_f = 0.53 (*n*hexane/ethyl acetate = 3/1); ¹H NMR (300 MHz, CDCl₃) δ 1.39 (*s*, 9 H), 3.75 (*s*, 3 H), 6.47 (dd, *J* = 5.7, 2.4 Hz, 1 H), 6.93 (d, *J* = 2.4 Hz, 1 H), 7.07 (td, *J* = 8.1, 2.4 Hz, 1 H), 7.17 (dd, *J* = 9.3, 2.4 Hz, 1 H), 7.33–7.54 (m, 4 H), 7.69 (d, *J* = 7.2 Hz, 2 H), 7.97 (d, *J* = 5.7 Hz, 1 H); ¹³C NMR (125 MHz, CDCl₃) δ 28.1 (CH₃ × 3), 55.2 (CH₃), 82.3 (C_q), 104.6 (CH), 107.8 (CH), 113.8 (d, *J*_{C-F} = 21.1 Hz, CH), 118.0 (d, *J*_{C-F} = 9.1 Hz, CH), 128.0 (CH × 2), 130.0 (CH × 2), 131.3 (d, *J*_{C-F} = 9.1 Hz, CH), 132.8 (CH), 133.2 (d, *J*_{C-F} = 3.6 Hz, C_q), 137.3 (C_q), 142.0 (d, *J*_{C-F} = 11.0 Hz, C_q), 148.2 (CH), 152.7 (C_q), 155.7 (C_q), 163.7 (d, *J*_{C-F} = 250.0 Hz, C_q), 166.5 (C_q), 194.6 (C_q); MS (EI, *m*/z) 422 (M⁺, 2), 217 (100); HRMS *m*/z calcd for C₂₄H₂₃FN₂O₄ 422.1642, found 422.1643.

tert-Butyl ⁽²⁾-benzoyl-6-fluorophenyl)(4-methoxypyridin-2-yl)carbamate (**5g**). Yellow solid; mp 78–79 °C; $R_f = 0.48$ (*n*-hexane/ ethyl acetate = 3/1); ¹H NMR (300 MHz, CDCl₃) δ 1.42 (s, 9 H), 3.73 (s, 3 H), 6.41 (dd, J = 6, 2.4 Hz, 1 H), 6.99 (d, J = 1.5 Hz, 1 H), 7.24 (t, J = 6.9 Hz, 1 H), 7.26–7.43 (m, 4 H), 7.50 (t, J = 7.2 Hz, 1 H), 7.67 (d, J = 7.5 Hz, 2 H), 7.89 (d, J = 5.7 Hz, 1 H); ¹³C NMR (125 MHz, CDCl₃) δ 27.9 (CH₃ × 3), 55.1 (CH₃), 82.2 (C_q), 102.9 (CH), 107.4 (CH), 117.9 (d, $J_{C-F} = 21.0$ Hz, CH), 124.5 (d, $J_{C-F} =$ 3.6 Hz, CH), 127.9 (CH), 128.3 (d, $J_{C-F} = 8.1$ Hz, CH × 2), 129.8 (CH × 2), 132.9 (CH), 137.0 (C_q), 139.0 (C_q), 148.0 (CH), 152.3 (C_q), 155.2 (C_q), 159.0 (d, $J_{C-F} = 249.3$ Hz, C_q), 166.3 (C_q), 194.6 (d, $J_{C-F} = 2.8$ Hz, C_q); MS (EI, m/z) 422 (M⁺, 2), 303 (71), 245 (69), 217 (100); HRMS m/z calcd for C₂₄H₂₃FN₂O₄ 422.1642, found 422.1642.

tert-Butyl (2-*benzoyl-4-chlorophenyl*)(4-*methoxypyridin-2-yl*)*carbamate* (5*h*). Yellow solid; mp 130–131 °C; $R_f = 0.60$ (*n*hexane/ethyl acetate = 2/1); ¹H NMR (300 MHz, CDCl₃) δ 1.39 (s, 9 H), 3.74 (s, 3 H), 6.45 (dd, J = 5.7, 2.1 Hz, 1 H), 6.90 (d, J = 1.5Hz, 1 H), 7.34–7.40 (m, 4 H), 7.51 (d, J = 8.1 Hz, 2 H), 7.70 (d, J =7.5 Hz, 2 H), 7.95 (d, J = 5.7 Hz, 1 H); ¹³C NMR (125 MHz, CDCl₃) δ 28.1 (CH₃ × 3), 55.1 (CH₃), 82.3 (C_q), 104.4 (CH), 107.7 (CH), 128.1 (CH × 2), 129.2 (CH), 130.0 (CH × 2), 130.9 (CH), 132.1 (CH), 132.4 (C_q), 133.1 (CH), 136.6 (C_q), 138.3 (C_q), 138.4 (C_q), 148.2 (CH), 155.8 (C_q), 155.7 (C_q), 166.4 (C_q), 194.2 (C_q); MS (EI, *m/z*) 309 (M⁺, 1), 277 (62), 261 (63), 105 (54), 57 (73); HRMS *m*/ *z* calcd for C₂₄H₂₃ClN₂O₄ 438.1346, found 438.1345.

tert-Butyl ⁽²⁾ ⁽²⁾

tert-Butyl (2,6-dibenzoyl-4-methoxyphenyl)(4-methoxypyridin-2-yl)carbamate (6d). Yellow solid; mp 164–165 °C; $R_f = 0.40$ (*n*-hexane/ethyl acetate = 3/1); ¹H NMR (300 MHz, CDCl₃) δ 1.36 (s, 9 H), 3.62 (s, 3 H), 3.84 (s, 3 H), 6.27 (dd, J = 5.7, 2.1 Hz, 1 H), 6.72 (s, 1 H), 7.12 (s, 2 H), 7.32 (t, J = 7.5 Hz, 4 H), 7.47 (t, J = 7.5 Hz, 2 H),7.66 (d, J = 5.7 Hz, 1 H), 7.77 (d, J = 7.2 Hz, 4 H); ¹³C NMR (125 MHz, CDCl₃) δ 27.9 (CH₃ × 3), 54.9 (CH₃), 55.8 (CH₃), 82.0 (C_q), 102.0 (C_q), 107.1 (CH), 116.4 (CH × 2), 127.9 (CH × 4), 130.0 (CH × 4), 132.6 (CH × 2), 137.2 (C_q × 2), 140.2 (C_q × 2), 146.8 (CH), 152.9 (C_q), 155.2 (C_q), 157.3 (C_q), 166.1 (C_q), 195.0 (C_q × 2); MS (EI, *m*/z) 538 (M⁺, 1), 333 (100), 105 (39), 77 (30), 57 (32); HRMS *m*/z calcd for C₃₂H₃₀N₂O₆ 538.2104, found 538.2101.

tert-Butyl (2,6-dibenzoyl-4-fluorophenyl)(4-methoxypyridin-2yl)carbamate (**6e**). White solid; mp 148–149 °C; $R_f = 0.60$ (*n*-hexane/ethyl acetate = 3/1); ¹H NMR (300 MHz, CDCl₃) δ 1.40 (s, 9 H), 3.65 (s, 3 H), 6.30 (d, J = 4.2 Hz, 1 H), 6.79 (s, 1 H), 7.35 (t, J = 7.2 Hz, 6 H), 7.51 (t, J = 7.2 Hz, 2 H), 7.67 (d, J = 5.4 Hz, 1 H), 7.78 (d, J = 7.5 Hz, 4 H); ¹³C NMR (125 MHz, CDCl₃) δ 27.9 (CH₃ × 3), 55.0 (CH₃), 82.4 (C_q), 107.3 (CH), 117.9 (d, $J_{C-F} = 23.8$ Hz, CH), 128.0 (CH × 4), 129.9 (CH × 4), 132.9 (CH × 2), 136.7 (C_q × 2), 141.1 (d, $J_{C-F} = 6.0$ Hz, C_q), 146.7 (CH), 152.6 (C_q), 154.8 (C_q), 159.9 (d, $J_{C-F} = 249.6$ Hz, C_q), 166.2 (C_q × 2), 193.8 (C_q × 2); MS (EI, m/z) 526 (M⁺, 1), 349 (41), 321 (100), 57 (36); HRMS m/zcalcd for C₃₁H₂₇FN₂O₅ 526.1904, found 526.1902.

tert-Butyl (2,6-dibenzoyl-4-chlorophenyl)(4-methoxypyridin-2-yl)carbamate (6h). Yellow solid; mp 121–122 °C; $R_f = 0.68$ (n-hexane/ethyl acetate = 2/1); ¹H NMR (300 MHz, CDCl₃) δ 1.37 (s, 9 H), 3.62 (s, 3 H), 6.29 (dd, J = 5.7, 2.1 Hz, 1 H), 6.76 (s, 1 H), 7.33 (t, J = 7.5 Hz, 4 H), 7.49 (t, J = 7.2 Hz, 2 H), 7.58 (s, 2 H), 7.65 (d, J = 5.7 Hz, 1 H), 7.75 (d, J = 7.5 Hz, 4 H); ¹³C NMR (125 MHz, CDCl₃) δ 27.9 (CH₃ × 3), 55.0 (CH₃), 82.6 (C_q), 101.8 (C_q), 107.3 (CH), 128.1 (CH × 5), 129.9 (CH × 5), 130.8 (CH), 132.4 (C_q), 132.9 (CH × 2), 136.7 (C_q × 2), 140.7 (C_q × 2), 146.8 (CH), 152.3 (C_q), 154.7 (C_q), 166.2 (C_q), 193.7 (C_q × 2); MS (EI, m/z) 542 (M⁺, 1), 365 (84), 337 (100), 105 (82), 77 (49), 57 (71); HRMS m/z calcd for C₃₁H₂₇ClN₂O₅ 542.1608, found 542.1607.

(2-Aminophenyl)(phenyl)methanone (7). Yellow solid; $R_f = 0.63$ (*n*-hexane/ethyl acetate = 4/1); ¹H NMR (400 MHz, CDCl₃) δ 6.08 (bs, 2 H), 6.60 (td, *J* = 4.8, 0.8 Hz, 1 H), 6.73 (dd, *J* = 5.6, 0.8 Hz, 1 H), 7.29 (td, *J* = 4.8, 0.8 Hz, 1 H), 7.44 (tt, *J* = 4.4, 0.8 Hz, 3 H), 7.51 (tt, *J* = 4.4, 0.8 Hz, 1 H), 7.62–7.64 (m, 2 H); ¹³C NMR (100 MHz, CDCl₃) δ 115.5 (CH), 117.0 (CH), 118.2 (C_q), 128.1 (CH × 2), 129.1 (CH × 2), 131.0 (CH), 134.2 (CH), 134.5 (CH), 140.1 (C_q), 150.9 (C_q), 199.0 (C_q); MS (EI, *m*/*z*) 197 (M⁺, 1), 196 (100), 120 (21); HRMS (EI-magnetic sector) *m*/*z* [M⁺] calcd for C₁₃H₁₁NO 197.0841, found 197.0840.

9(10H)-Acridanone (**8**). Yellow solid; $R_f = 0.43$ (*n*-hexane/ethyl acetate = 2/1); ¹H NMR (400 MHz, DMSO- d_6) δ 7.25 (td, J = 6.8, 1.2 Hz, 2 H), 7.54 (d, J = 8.4 Hz, 2 H), 7.73 (td, J = 7.2, 1.6 Hz, 2 H), 8.23 (dd, J = 8.0, 1.6 Hz, 2 H), 11.73 (d, J = 0.4 Hz, 1 H); ¹³C NMR (100 MHz, DMSO- d_6) δ 117.3 (CH × 2), 120.5 (C_q × 2), 121.0 (CH × 2), 126.0 (CH × 2), 133.4 (CH × 2), 140.9 (C_q × 2), 176.8 (C_q); MS (EI, m/z) 196 (M⁺, 1); HRMS (EI-magnetic sector) m/z [M⁺] calcd for C₁₃H₉NO 195.0684, found 195.0686.

10-(4-Methoxypyridin-2-yl)acridin-9(10H)-one (9). White oil; $R_f = 0.43$ (*n*-hexane/ethyl acetate = 1/2); ¹H NMR (400 MHz, CDCl₃) δ 3.95 (s, 3 H), 6.69 (d, J = 5.6 Hz, 2 H), 6.97 (d, J = 1.2 Hz, 1 H), 7.12 (dd, J = 4.0, 1.6 Hz, 1 H), 7.29 (td, J = 4.8, 0.8 Hz, 2 H), 7.52 (td, J = 4.4, 0.8 Hz, 2 H), 8.58 (dd, J = 5.2, 0.8 Hz, 2 H), 8.68 (d, J = 4.0 Hz, 1 H); ¹³C NMR (100 MHz, CDCl₃) δ 55.9 (CH₃), 111.2 (CH), 116.2 (CH × 2), 121.8 (CH × 2), 127.5 (CH × 2), 130.0 (CH), 133.3 (CH × 2), 152.3 (CH), 142.2 (C_q × 2), 153.9 (C_q), 169.1 (C_q), 173.2 (C_q × 2), 178.2 (C_q); MS (EI, *m*/*z*) 302 (M⁺, 1), 88 (43), 70 (64), 61 (100); HRMS (EI-magnetic sector) *m*/*z* [M⁺] calcd for C₁₉H₁₄N₂O₂ 302.1055, found 302.1054.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.organo-met.9b00119.

Optimization of reaction conditions, synthesis and characterization of starting substrates, preparation of deuterium-labeled substrates $1a-D_1$ and $1a-D_5$, and KIEs operation and calculation, ¹H, ¹³C and DEPT (90° and 135°) NMR spectra of new and known compounds (PDF)

Accession Codes

CCDC 1897991–1897993 contain the supplementary crystallographic data for this paper. These data can be obtained free of charge via www.ccdc.cam.ac.uk/data_request/cif, or by emailing data_request@ccdc.cam.ac.uk, or by contacting The Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax: +44 1223 336033.

AUTHOR INFORMATION

Corresponding Authors

*E-mail: j_hchu@nttu.edu.tw. *E-mail: mijuwu@faculty.nsysu.edu.tw.

ORCID 💿

Jean-Ho Chu: 0000-0003-3366-2126 Ming-Jung Wu: 0000-0002-5507-1848

Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENTS

We thank the Ministry of Science and Technology of the Republic of China (MOST 104-2113-M-143-002-MY2 and MOST 106-2113-M-143-001-MY2) for financial support and Mr. Min-Yuan Hung for his technical support and services in GC-MS, Center for Research Resources and Development of Kaohsiung Medical University (Taiwan).

REFERENCES

(1) Selected reviews: (a) Abrams, D. J.; Provencher, P. A.; Sorensen, E. J. Recent Applications of C-H Functionalization in Complex Natural Product Synthesis. Chem. Soc. Rev. 2018, 47, 8925-8967. (b) Wang, C.-S.; Dixneuf, P. H.; Soulé, J.-F. Photoredox Catalysis for Building C-C Bonds from C(sp²)-H Bonds. Chem. Rev. 2018, 118, 7532–7585. (c) Leitch, J. A.; Frost, C. G. Ruthenium-Catalysed σ -Activation for Remote Meta-Selective C-H Functionalization. Chem. Soc. Rev. 2017, 46, 7145-7153. (d) Yang, Y.; Lan, J.; You, J. Oxidative C-H/C-H Coupling Reactions between Two (Hetero)arenes. Chem. Rev. 2017, 117, 8787-8863. (e) He, J.; Wasa, M.; Chan, K. S. L.; Shao, Q.; Yu, J.-Q. Palladium-Catalyzed Transformations of Alkyl C-H Bonds. Chem. Rev. 2017, 117, 8754-8786. (f) Yi, H.; Zhang, G.; Wang, H.; Huang, Z.; Wang, J.; Singh, A. K.; Lei, A. Recent Advances in Radical C-H Activation/Radical Cross-Coupling. Chem. Rev. 2017, 117, 9016-9085. (g) Hummel, J. R.; Boerth, J. A.; Ellman, J. A.; Transition-Metal-Catalyzed, C.-H. Bond Addition to Carbonyls, Imines, and Related Polarized π Bonds. Chem. Rev. 2017, 117, 9163-9227. (h) Agasti, S.; Dey, A.; Maiti, D. Palladium-Catalyzed Benzofuran and Indole Synthesis by Multiple C-H Functionalizations. Chem. Commun. 2017, 53, 6544-6556. (i) Wang, S.; Chen, S.-Y.; Yu, X.-Q. C-H Functionalization by High-Valent Cp*Co(iii) Catalysis. Chem. Commun. 2017, 53, 3165-3180. (j) Afewerki, S.; Cordova, A. Combinations of Aminocatalysts and Metal Catalysts: A Powerful Cooperative Approach in Selective Organic Synthesis. Chem. Rev. 2016, 116, 13512-13570. (k) Zhu, R.-Y.; Farmer, M. E.; Chen, Y.-Q.;

Yu, J.-Q. A Simple and Versatile Amide Directing Group for C-H Functionalizations. Angew. Chem., Int. Ed. 2016, 55, 10578-10599. (l) Della, C.; Fontana, M.; Motti, E.; Catellani, M. Pd/Norbornene: A Winning Combination for Selective Aromatic Functionalization via C-H Bond Activation. Acc. Chem. Res. 2016, 49, 1389-1400. (m) Rouquet, G.; Chatani, N. Catalytic Functionalization of C(sp²)-H and C(sp³)-H Bonds by Using Bidentate Directing Groups. Angew. Chem., Int. Ed. 2013, 52, 11726-11743. (n) Mousseau, J. J.; Charette, A. B. Direct Functionalization Processes: A Journey from Palladium to Copper to Iron to Nickel to Metal-Free Coupling Reactions. Acc. Chem. Res. 2013, 46, 412-424. (o) Arockian, P. B.; Bruneau, C.; Dixneuf, P. H. Ruthenium(II)-Catalyzed C-H Bond Activation and Functionalization. Chem. Rev. 2012, 112, 5879-5918. (p) Hartwig, J. F. Borylation and Silylation of C-H Bonds: A Platform for Diverse C-H Bond Functionalizations. Acc. Chem. Res. 2012, 45, 864-873. (q) Yeung, C. S.; Dong, V. M. Catalytic Dehydrogenative Cross-Coupling: Forming Carbon-Carbon Bonds by Oxidizing Two Carbon-Hydrogen Bonds. Chem. Rev. 2011, 111, 1215-1292. (r) Ackerman, L. Carboxylate-Assisted Transition-Metal-Catalyzed C-H Bond Functionalizations: Mechanism and Scope. Chem. Rev. 2011, 111, 1315-1345. (s) Lyons, T. W.; Sanford, M. S. Palladium-Catalyzed Ligand-Directed C-H Functionalization Reactions. Chem. Rev. 2010, 110, 1147-1169. (t) Colby, D. A.; Bergman, R. G.; Ellman, J. A. Rhodium-Catalyzed C-C Bond Formation via Heteroatom-Directed C-H Bond Activation. Chem. Rev. 2010, 110, 624-655. (u) Bellina, F.; Rossi, R. Transition Metal-Catalyzed Direct Arylation of Substrates with Activated sp³-Hybridized C-H Bonds and Some of Their Synthetic Equivalents with Aryl Halides and Pseudohalides. Chem. Rev. 2010, 110, 1082-1146. (v) Chen, X.; Engle, K. M.; Wang, D.-H.; Yu, J.-Q. Palladium(II)-Catalyzed C-H Activation/C-C Cross-Coupling Reactions: Versatility and Practicality. Angew. Chem., Int. Ed. 2009, 48, 5094-5115.

(2) Selected reviews: (a) Brady, P. B.; Bhat, V. Recent Applications of Rh- and Pd-Catalyzed C(sp³)-H Functionalization in Natural Product Total Synthesis. *Eur. J. Org. Chem.* 2017, 2017, 5179-5190.
(b) Nairoukh, Z.; Cormier, M.; Marek, I. Merging C-H and C-C Bond Cleavage in Organic Synthesis. *Nat. Rev. Chem.* 2017, 1, 0035.
(c) Shi, X.; Sasmal, A.; Soule, J. F.; Doucet, H. Metal-Catalyzed C-H Bond Activation of 5-Membered Carbocyclic Rings: A Powerful Access to Azulene, Acenaphthylene and Fulvene Derivatives. *Chem.* - Asian J. 2018, 13, 143-157.

(3) (a) Sambiagio, C.; Schönbauer, D.; Blieck, R.; Toan, D.-H.; Pototschnig, G.; Schaaf, P.; Wiesinger, T.; Zia, M. F.; Wencel-Delord, J.; Besset, T.; Maes, B. U. W.; Schnürch, M. A Comprehensive Overview of Directing Groups Applied in Metal-Catalysed C-H Functionalisation Chemistry. Chem. Soc. Rev. 2018, 47, 6603-6743. (b) Xu, Y.; Dong, G. sp³ C-H Activation via Exo-Type Directing Groups. Chem. Sci. 2018, 9, 1424-1432. (c) Swamy, T.; Reddy, B. V. S.; Gree, R.; Ravinder, V. Substrate-Directed C-H Functionalization of 2-Aryl Pyridines by Transition Metal Complexes. ChemistrySelect 2018, 3, 47-70. (d) Hirano, K.; Miura, M. A Lesson for Site-Selective C-H Functionalization on 2-Pyridones: Radical, Organometallic, Directing Group and Steric Controls. Chem. Sci. 2018, 9, 22-32. (e) Ping, Y.; Wang, L.; Ding, Q.; Peng, Y. Nitrile as a Versatile Directing Group for C(sp²)-H Functionalizations. Adv. Synth. Catal. 2017, 359, 3274-3291. (f) Zhao, Q.; Poisson, T.; Pannecoucke, X.; Besset, T. The Transient Directing Group Strategy: A New Trend in Transition-Metal-Catalyzed C-H Bond Functionalization. Synthesis 2017, 49, 4808-4826. (g) Chu, J.-H.; Chen, C.-C.; Wu, M.-J. Palladium-Catalyzed Arylation and Alkylation of 3,5-Diphenylisoxazole with Boronic Acids via C-H Activation. Organometallics 2008, 27, 5173-5176.

(4) The directing group assists the transition metal to catalyze siteselective C-H bond functionalization, *ortho*-position: (a) Shen, Y.; Lee, W.-C. C.; Gutierrez, D. A.; Li, J. J. Palladium-Catalyzed Direct $C(sp^2)$ -H ortho-Arylation of Anilides Using 2-Aminophenylpyrazole as the Directing Group. *J. Org. Chem.* **2017**, *82*, 11620–11625. (b) Reddy, D. M.; Wang, S. C.; Du, K.; Lee, C.-F. Palladium-Catalyzed ortho-C-H Arylation of Acetophenone Oxime Ethers with Aryl Pinacol Boronic Esters. J. Org. Chem. 2017, 82, 10070-10076. (c) Xu, J.; Liu, Y.; Wang, Y.; Li, Y.; Xu, X.; Jin, Z. Pd-Catalyzed Direct ortho-C-H Arylation of Aromatic Ketones Enabled by a Transient Directing Group. Org. Lett. 2017, 19, 1562-1565. meta-position: (d) Xu, H.-J.; Lu, Y.; Farmer, M. E.; Wang, H.-W.; Zhao, D.; Kang, Y.-S.; Sun, W.-Y.; Yu, J.-Q. Rh(III)-Catalyzed meta-C-H Olefination Directed by a Nitrile Template. J. Am. Chem. Soc. 2017, 139, 2200-2203. (e) Chu, L.; Shang, M.; Tanaka, K.; Chen, Q.; Pissarnitski, N.; Streckfuss, E.; Yu, J.-Q. Remote Meta-C-H Activation Using a Pyridine-Based Template: Achieving Site-Selectivity via the Recognition of Distance and Geometry. ACS Cent. Sci. 2015, 1, 394-399. (f) Leow, D.; Li, G.; Mei, T.-S.; Yu, J.-Q. Activation of Remote Meta-C-H Bonds Assisted by an End-On Template. Nature 2012, 486, 518-522. para-position: (g) Bag, S.; Patra, T.; Modak, A.; Deb, A.; Maity, S.; Dutta, U.; Dey, A.; Kancherla, R.; Maji, A.; Hazra, A.; Bera, M.; Maiti, D. Remote para-C-H Functionalization of Arenes by a D-Shaped Biphenyl Template-Based Assembly. J. Am. Chem. Soc. 2015, 137, 11888-11891. (h) Maji, A.; Dahiya, A.; Lu, G.; Bhattacharya, T.; Brochetta, M.; Zanoni, G.; Liu, P.; Maiti, D. H-Bonded Reusable Template Assisted para-Selective Ketonisation Using Soft Electrophilic Vinyl Ethers. Nat. Commun. 2018, 9, 3582.

(5) Our works on the palladium-catalyzed selective ortho-C-H functionalization using the pyridinyl as a directing group: (a) Chu, J.-H.; Chen, S.-T.; Chiang, M.-F.; Wu, M.-J. Palladium-Catalyzed Direct Ortho Aroylation of 2-Phenoxypyridines with Aldehydes and Catalytic Mechanism Investigation. Organometallics 2015, 34, 953-966. (b) Chu, J.-H.; Huang, H.-P.; Shu, W.-T.; Chen, S.-T.; Wu, M.-J. Palladium(II)-Catalyzed Direct Ortho Arylation of 4-Methyl-Nphenylpyridin-2-amines via C-H Activation/C-C Coupling and Synthetic Applications. Organometallics 2014, 33, 1190-1204. (c) Chu, J.-H.; Wu, C.-C.; Chang, D.-H.; Lee, Y.-M.; Wu, M.-J. Direct Ortho Arylation of 9-(Pyridin-2-yl)-9H-carbazoles Bearing a Removable Directing Group via Palladium(II)-Catalyzed C-H Bond Activation. Organometallics 2013, 32, 272-282. (d) Chu, J.-H.; Lin, P.-S.; Lee, Y.-M.; Shen, W.-T.; Wu, M.-J. Palladium(II)-Catalyzed One-Pot Syntheses of 9-(Pyridin-2-yl)-9H-carbazoles through a Tandem C-H Activation/C-X (X = C or N) Formation Process. Chem. - Eur. J. 2011, 17, 13613-13620. (e) Chu, J.-H.; Lin, P.-S.; Wu, M.-J. Palladium(II)-Catalyzed Ortho Arylation of 2-Phenoxypyridines with Potassium Aryltrifluoroborates via C-H Functionalization. Organometallics 2010, 29, 4058-4065. (f) Chu, J.-H.; Tsai, S.-L.; Wu, M.-J. Palladium(II)-Catalyzed ortho Arylation of 2-Phenylpyridines with Potassium Aryltrifluoroborates by C-H Functionalization. Synthesis 2009, 2009, 3757-3764.

(6) Synthesis of 9(10*H*)-acridanones using 2-aminobenzophenones as starting substrates: (a) Huang, P.-C.; Parthasarathy, K.; Cheng, C.-H. Copper-Catalyzed Intramolecular Oxidative C-H Functionalization and C-N Formation of 2-Aminobenzophenones: Unusual Pseudo-1,2-Shift of the Substituent on the Aryl Ring. *Chem. - Eur. J.* **2013**, *19*, 460–464. (b) Huang, J.; Wan, C.; Xu, M.-F.; Zhu, Q. Synthesis of 10-Methylacridin-9(10H)-ones through Cu-Catalyzed Intramolecular Oxidative C(sp²)–H Amination of 2-(Methylamino)benzophenones. *Eur. J. Org. Chem.* **2013**, *2013*, 1876–1880. (c) Zhou, W.; Liu, Y.; Yang, Y.; Deng, G.-J. Copper-Catalyzed Intramolecular Direct Amination of sp² C–H Bonds for the Synthesis of N-Aryl Acridones. *Chem. Commun.* **2012**, *48*, 10678–10680.

(7) (a) Cortez-Maya, S.; Cortes, E. C.; Hernandez-Ortega, S.; Apan, T. R.; Martinez-Garcia, M. Synthesis of 2-Aminobenzophenone Derivatives and their Anticancer Activity. *Synth. Commun.* **2012**, *42*, 46–54. (b) Singh, R. K.; Devi, S.; Prasad, D. N. Synthesis, Physicochemical and Biological Evaluation of 2-Amino-5-chlorobenzophenone Derivatives as Potent Skeletal Muscle Relaxants. *Arabian J. Chem.* **2015**, *8*, 307–312. (c) Li, P.; Sahore, K.; Liu, J.; Singh, R. K. Synthesis and Antimicrobial Evaluation of 2-Aminobenzophenone Linked 1,4-Dihydropyridine Derivatives. *Asian J. Chem.* **2014**, *26*, 5291–5294. (d) Liou, J.-P.; Chang, C.-W.; Song, J.-S.; Yang, Y.-N.; Yeh, C.-F.; Tseng, H.-Y.; Lo, Y.-K.; Chang, Y.-L.; Chang, C.-M.; Hsieh, H.-P. Synthesis and Structure–Activity Relationship of 2-

Aminobenzophenone Derivatives as Antimitotic Agents. J. Med. Chem. 2002, 45, 2556–2562.

(8) (a) Jia, X.; Zhang, S.; Wang, W.; Luo, F.; Cheng, J. Palladium-Catalyzed Acylation of sp² C-H bond: Direct Access to Ketones from Aldehydes. Org. Lett. 2009, 11, 3120-3123. (b) Park, J.; Park, E.; Kim, A.; Lee, Y.; Chi, K.-W.; Kwak, J. H.; Jung, Y. H.; Kim, I. S. Rhodium-Catalyzed Oxidative ortho-Acylation of Benzamides with Aldehydes: Direct Functionalization of the sp² C-H Bond. Org. Lett. 2011, 13, 4390-4393. (c) Sharma, S.; Park, E.; Park, J.; Kim, I. S. Tandem Rh(III)-Catalyzed Oxidative Acylation of Secondary Benzamides with Aldehydes and Intramolecular Cyclization: The Direct Synthesis of 3-Hydroxyisoindolin-1-ones. Org. Lett. 2012, 14, 906-909. (d) Sharma, S.; Park, J.; Park, E.; Kim, A.; Kim, M.; Kwak, J. H.; Jung, Y. H.; Kim, I. S. Palladium-Catalyzed Oxidative Acylation of N-Benzyltriflamides with Aldehydes via C-H Bond Activation. Adv. Synth. Catal. 2013, 355, 332-336. (e) Chan, C.-W.; Zhou, Z.; Chan, A. S. C.; Yu, W.-Y. Pd-Catalyzed Ortho-C-H Acylation/Cross Coupling of Aryl Ketone O-Methyl Oximes with Aldehydes Using tert-Butyl Hydroperoxide as Oxidant. Org. Lett. 2010, 12, 3926-3929. (f) Yang, Y.; Zhou, B.; Li, Y. Rhodium-Catalyzed Oxidative ortho-Acylation of Aryl Ketone O-Methyl Oximes with Aryl and Alkyl Aldehydes. Adv. Synth. Catal. 2012, 354, 2916-2920. (g) Sharma, S.; Kim, A.; Park, J.; Kim, M.; Kwak, J. H.; Jung, Y. H.; Park, J. S.; Kim, I. S. Pd(II)-Catalyzed Direct C-H Acylation of N-Boc Hydrazones with Aldehydes: One-Pot Synthesis of 1,2-Diacylbenzenes. Org. Biomol. Chem. 2013, 11, 7869-7876. (h) Yao, J.; Feng, R.; Wu, Z.; Liu, Z.; Zhang, Y. Palladium-Catalyzed Decarboxylative Coupling of α -Oxocarboxylic Acids with $C(sp^2)$ -H of 2-Aryloxypyridines. Adv. Synth. Catal. 2013, 355, 1517-1522. (i) Chan, C.-W.; Zhou, Z.; Yu, W.-Y. Palladium(II)-Catalyzed Direct ortho-C-H Acylation of Anilides by Oxidative Cross-Coupling with Aldehydes using tert-Butyl Hydroperoxide as Oxidant. Adv. Synth. Catal. 2011, 353, 2999-3006. (j) Banerjee, A.; Bera, A.; Santra, S. K.; Guin, S.; Patel, B. K. Palladium-Catalysed Regioselective Aroylation and Acetoxylation of 3,5-Diarylisoxazole via ortho C-H Functionalisations. RSC Adv. 2014, 4, 8558-8566. (k) Maiti, S.; Burgula, L.; Chakraborti, G.; Dash, J. Palladium-Catalyzed Pyridine-Directed Regioselective Oxidative C-H Acylation of Carbazoles by Using Aldehydes as the Acyl Source. Eur. J. Org. Chem. 2017, 2017, 332-340. (1) Deb, M.; Hazra, S.; Gupta, A.; Elias, A. J. Synthesis of Unsymmetrical Multi-Aroyl Derivatives of Ferrocene Using Palladium Catalysed Oxidative C-H Aroylation. Dalton Trans. 2018, 47, 7229-7236. (m) Shao, L.-Y.; Xu, Z.; Wang, C.-Y.; Fu, X.-P.; Chen, M.-M.; Liu, H.-W.; Ji, Y.-F. Palladium-Catalyzed Direct Mono-Aroylation of O-Arylmethyl and Aryl-Substituted Acetoxime Ethers. Org. Biomol. Chem. 2018, 16, 6284-6294. (n) Chen, M.-M.; Shao, L.-Y.; Lun, L.-J.; Wu, Y.-L.; Fu, X.-P.; Ji, Y.-F. Palladium-Catalyzed Late-Stage Mono-Aroylation of the Fully Substituted Pyrazoles via Aromatic C-H Bond Activation. Chin. Chem. Lett. 2019, 30, 702-706.

(9) 1-Hydroxy-9H-fluoren-9-one derivatives usually own potent biological activity: (a) Hu, Q. F.; Zhou, B.; Huang, J. M.; Gao, X. M.; Shu, L. D.; Yang, G. Y.; Che, C. T. Antiviral Phenolic Compounds from Arundina gramnifolia. J. Nat. Prod. 2013, 76, 292–296.
(b) Chen, Y.; Li, Y.; Qing, C.; Zhang, Y.; Wang, L.; Liu, Y. 1,4,5-Trihydroxy-7-methoxy-9H-fluoren-9-one, A New Cytotoxic Compound from Dendrobium Chrysotoxum. Food Chem. 2008, 108, 973–976.
(c) Sreenivas, D. K.; Nagarajan, R. Highly Regioselective Synthesis of Indenocarbazolones via Palladium-Catalyzed Intramolecular ortho Arylation. Synlett 2012, 23, 1007–1012.

(10) The structure of 9(10H)-acridanones is often observed in some of potent bioactive molecules: (a) Winter, R.; Hinrichs, D. J.; Riscoe, M. K. Discovery of Dual Function Acridones as a New Antimalarial Chemotype. *Nature* **2009**, *459*, 270–273. (b) Basavaiah, D.; Rao, J. S.; Reddy, R. J. Simple, Facile, and One-Pot Conversion of the Baylis–Hillman Adducts into Functionalized 1,2,3,4-Tetrahydroacridines and Cyclopenta[b]quinolones. J. Org. Chem. **2004**, *69*, 7379–7382.

(11) When the aryl aldehyde bearing an *ortho*-substituent (e.g., 2f, 2i, 2o, and 2q) is converted to the aroyl radical and subsequently

reacts with the palladium center, the *ortho* steric hindrance might suppress the reactivity. By contrast, the starting substrates 1b and 1g bearing an *ortho*-substituent might cause the steric hindrance between the pyridine and aniline to form an orthogonal molecular geometry, which might promote the ease of the formation of the palladium complex I or II. The phenomenon may eventually result in the observation of the excellent yield.

(12) Copies of the deposited crystallographic data CCDC-1897993 (4a), CCDC-1897992 (5h), and CCDC-1897991 (I) can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data request/cif.

(13) These results are consistent with that of our previous study, the palladium-catalyzed ortho-C-H bond aroylation of 2-phenoxypyridines, see ref 5a. In the radical reactions, large $k_{\rm H}/k_{\rm D}$ values are often observed because of the quantum tunneling effect, see: (a) Lewis, E. S.; Funderburk, L. H. Rates and Isotope Effects in the Proton Transfers from 2-Nitropropane to Pyridine Bases. J. Am. Chem. Soc. 1967, 89, 2322-2327. (b) Wu, A.; Mader, E. A.; Datta, A.; Hrovat, D. A.; Borden, W. T.; Mayer, J. M. Nitroxyl Radical Plus Hydroxylamine Pseudo Self-Exchange Reactions: Tunneling in Hydrogen Atom Transfer. J. Am. Chem. Soc. 2009, 131, 11985-11997. (c) Lamberson, C. R.; Xu, L.; Muchalski, H.; Montenegro-Burke, J. R.; Shmanai, V. V.; Bekish, A. V.; McLean, J. A.; Clarke, C. F.; Shchepinov, M. S.; Porter, N. A. Unusual Kinetic Isotope Effects of Deuterium Reinforced Polyunsaturated Fatty Acids in Tocopherol-Mediated Free Radical Chain Oxidations. J. Am. Chem. Soc. 2014, 136, 838-841.

(14) (a) The interpretation of kinetic isotope effects on the transition-metal-catalyzed C-H bond functionalization, see: Simmons, E. M.; Hartwig, J. F. On the Interpretation of Deuterium Kinetic Isotope Effects in C-H Bond Functionalizations by Transition-Metal Complexes. *Angew. Chem., Int. Ed.* **2012**, *51*, 3066–3072. (b) Bhalla, G.; Liu, X. Y.; Oxgaard, J.; Goddard, W. A., III; Periana, R. A. Synthesis, Structure, and Reactivity of O-Donor Ir(III) Complexes: C-H Activation Studies with Benzene. J. Am. Chem. Soc. **2005**, *127*, 11372–11389.

(15) The synthetic conditions was followed by the literature reported, see ref 6a.

(16) For the synthesis and characterization of substrates 1-H/H, 1-Me/H, and 1-OMe/H, see ref 5b.