



## RECOGNITION OF A G-C BASE PAIR BY $\alpha$ -N<sup>7</sup>-DEOXYINOSINE WITHIN THE PYRIMIDINE-PURINE-PYRIMIDINE DNA TRIPLE HELICAL MOTIF

Judith Marfurt, Jürg Hunziker and Christian Leumann\* <sup>1)</sup>

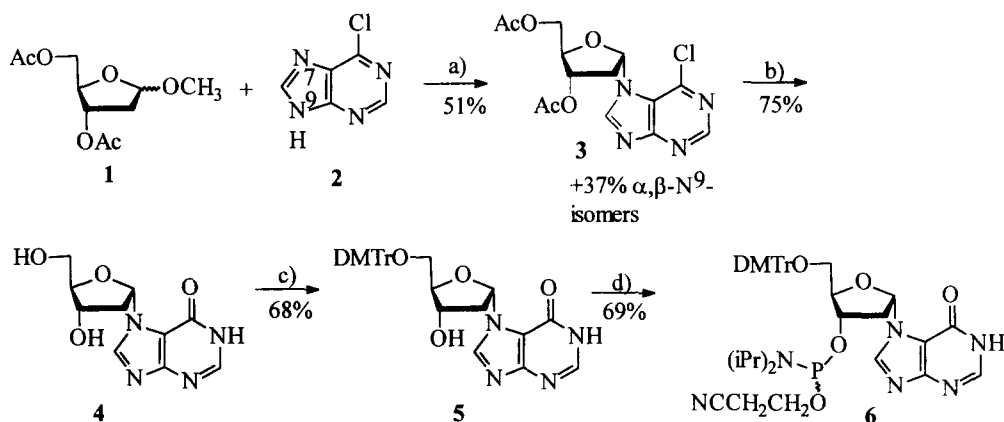
Department of Chemistry and Biochemistry, University of Bern, Freiestrasse 3, CH-3012 Bern

**Abstract:** The  $\alpha$ -nucleoside 7-(2'-deoxy- $\alpha$ -D-ribofuranosyl)hypoxanthine, incorporated into an otherwise  $\beta$ -configured oligodeoxynucleotide that is designed to bind to a DNA duplex in the parallel motif, recognizes selectively and efficiently a G-C base pair, presumably via monodentate  $\alpha$ -H<sup>7</sup>•G-C base-triple formation.

Copyright © 1996 Elsevier Science Ltd

Bidentate thymine-adenine (T-A) and protonated cytosine-guanine (C<sup>+</sup>-G) base recognition in the parallel Hoogsteen DNA triple helical motif (py•pu-py motif),<sup>1,2</sup> as well as A-A, T-A and G-G base recognition in the antiparallel reversed Hoogsteen motif (pu•pu-py motif)<sup>3,4</sup> are distinctly favored over other possible base-base combinations<sup>5,6</sup> and form the structural basis of the attractive interaction between a third DNA strand and a DNA duplex. Because of the restriction of both binding modes to homopurine and homopyrimidine sequences, much effort has recently been devoted to the search for a more general mode of DNA duplex complexation by oligonucleotides.<sup>7</sup> In this context we have embarked on a study of oligonucleotides containing the non-natural nucleoside 7-(2'-deoxy- $\alpha$ -D-ribofuranosyl)hypoxanthine ( $\alpha$ -<sup>7</sup>H) (**4**, Figure 1).

Figure 1



a)  $(\text{Me}_3\text{Si})_2\text{NH}$ ,  $\text{TMSCl}$ ,  $\text{SnCl}_4$ ,  $\text{MeCN}$ , r.t., 3h. b)  $\text{NaOH}$  in  $\text{THF}:\text{MeOH}:\text{H}_2\text{O}$  5:4:1, 0-65°C, 5h. c)  $\text{DMTrCl}$ ,  $\text{C}_5\text{H}_5\text{N}$ , r.t., 3h. d)  $(i\text{Pr})_2\text{N}(\text{NCCH}_2\text{CH}_2\text{O})\text{PCl}$ ,  $\text{THF}$ , r.t., 90 min.

<sup>1)</sup> E-mail: leumann@ioc.unibe.ch, Fax: +41 (31) 631 3422

By systematically exploring its properties in triple helix formation we also investigated its pairing in the context of the py•pu-py motif and found that a  $\alpha$ - $^7\text{H}$  residue within an otherwise  $\beta$ -configured pyrimidine oligonucleotide effectively recognizes a G-C base-pair with high selectivity.

$\alpha$ - $^7\text{H}$  (**4**), as well as its phosphoramidite building block **6**, were conveniently prepared from the methylglycoside **1** and 6-chloropurine (**2**) using standard procedures in nucleoside chemistry (*Figure 1*). The nature of the glycosidic bond in **4** as well as its conformational preferences were safely established by x-ray analysis.<sup>8</sup> Building block **6** was incorporated into oligomer **8** by solid phase DNA-synthesis and its composition analyzed by MALDI-TOF mass spectrometry after isolation (M-1 calc: 4507.1, found: 4508.7). Oligomer **7** was prepared using standard DNA chemistry and used as a reference.

Triple helix binding affinity and specificity was determined by DNase I footprint analysis in analogy to known procedures.<sup>9,10</sup> A plasmid was constructed containing four triplex target cassettes each spaced by 13 random nucleotides, and each containing the consensus sequence shown in *Figure 2* displaying one of the four possible canonical base-pairs in the center. A  $^{32}\text{P}$ -radiolabelled 229 bp fragment of this plasmid was used for the footprint assay with oligo **7** and **8** (100mM NaCl, 10 mM Bis-Tris.HCl, 0.25 mM spermine.4HCl, pH 7.0, 18°C).

The corresponding autoradiogram (*Figure 2*) clearly shows protection from DNase I activity in the cassette containing a central G-C base pair at concentrations of **8** as low as 0.2 $\mu\text{M}$ , indicating strong binding of **8** to this cassette. Triplex formation of **8** is selective. No binding of **8** to the cassettes containing the T-A or the C-G central base-pair was observed but weak binding to the cassette containing the A-T base-pair occurred. Quantitation of binding of **8** to the G-C containing cassette, determined as described,<sup>11</sup> revealed an association constant ( $K_{\text{ass}}$ ) of  $1.7 (\pm 0.9) \times 10^6 \text{ M}^{-1}$ . This compares to a  $K_{\text{ass}}$  of  $7.4 (\pm 3.0) \times 10^6 \text{ M}^{-1}$  for the reference oligomer **7** binding to the same target cassette (data not shown). Therefore the exchange of a  $\alpha$ - $^7\text{H}$  residue for a methylcytidine in the context of the 15-mers studied here resulted only in a fivefold decrease in binding efficiency.

The decrease in affinity due to the replacement of  $^{13}\text{C}$  for  $\alpha$ - $^7\text{H}$  in the third strand is less pronounced than for mismatches in purely  $\beta$ -configured oligomers containing only natural bases. It was shown previously that within the same sequence context, non canonical base triples (Z•G-C triple, Z=G, T, A) decrease binding efficiency by about 2 orders of magnitude with respect to the matched base triple  $^{13}\text{C}$ •G-C.<sup>5</sup> Energetically the  $\alpha$ - $^7\text{H}$ •G-C base triple contributes more to the stability of the triplex than any of the 14 possible non canonical natural ones, the best of which (G•T-A) showing reduced binding by a factor of ca. 15.

We assume that base-base recognition occurs via one H-bond between  $\text{N}^1$  of hypoxanthine and either  $\text{N}^7$  or  $\text{O}^6$  of guanine (*Figure 3*), favoring the  $\text{N}^1\text{H}\cdots\text{N}^7$  model (*Figure 3*, left) because of the observable weak binding of  $\alpha$ - $^7\text{H}$  to adenine (*Figure 2*). However, at this point we can not exclude other binding modes as e.g site selective intercalation. Computer model building within the given py•pu-py motif suggests that the sugar of the  $\alpha$ - $^7\text{H}$  residue adopts a 1'-endo conformation with the base in the (formal) syn orientation ( $\text{O}^6$  of the base

oriented towards the  $\alpha$ -face of the sugar). Additional factors that may contribute to the stability of the  $\alpha$ - $^7\text{H}\cdot\text{G}$  base-pair, derived from this model, might comprise favourable stacking interactions between the imidazole moiety of  $\alpha$ - $^7\text{H}$  with the next, 5'-located cytosine base in the third strand. Furthermore, no repulsive interactions between H-2 of  $\alpha$ - $^7\text{H}$  and O<sup>6</sup> of guanine is expected in either arrangement (Figure 3).

Figure 2:

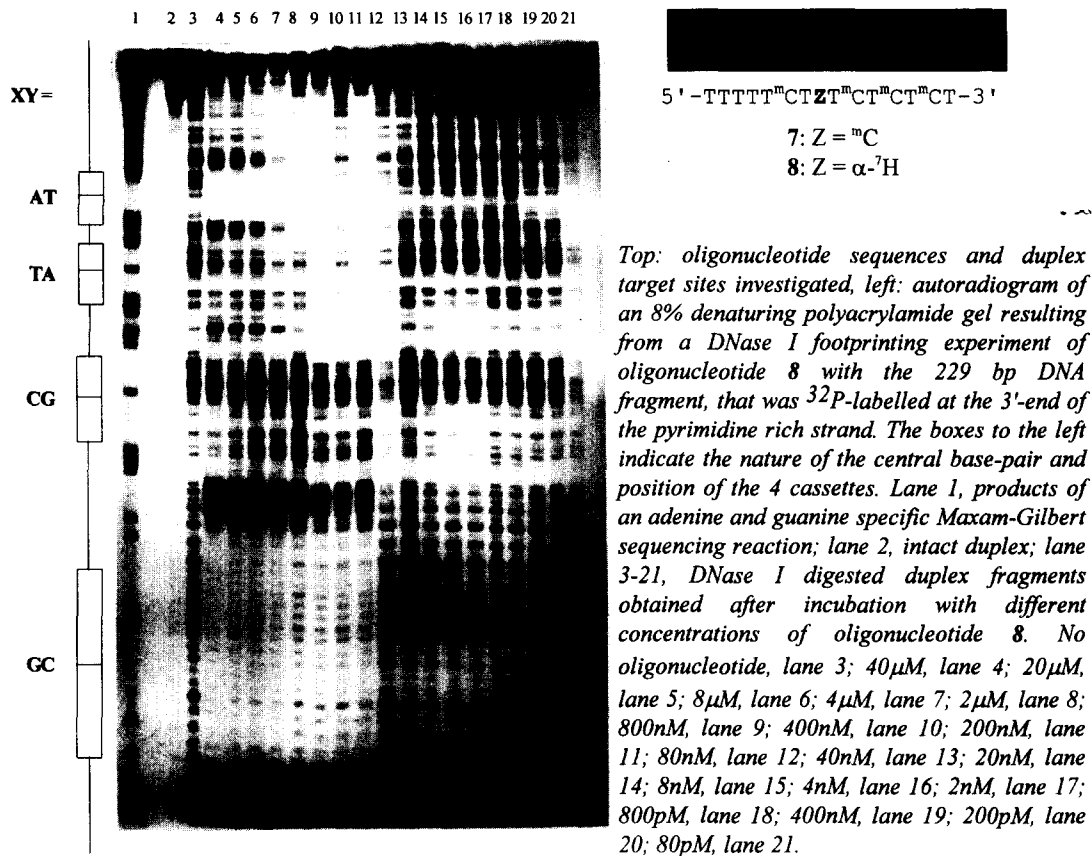
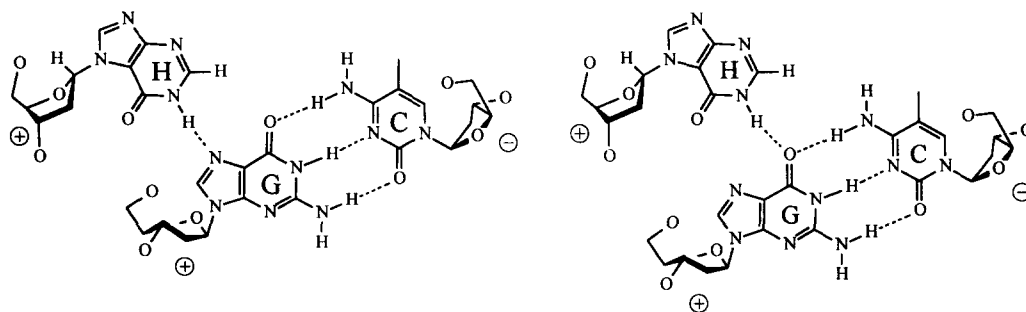


Figure 3:



In the series of N<sup>7</sup>-β-configured purine nucleosides, guanine and a structural relative of this base were investigated in triple helix formation and were shown to act as base neutral cytosine replacements in the py•pu-py motif<sup>10-12</sup> or to a mismatch in the pu•pu-py motif.<sup>13</sup> It is well known that completely α-configured pyrimidine rich and purine rich oligonucleotides form triplexes with natural DNA.<sup>14</sup> On the level of DNA-duplex formation it was reported recently that a single α-configured deoxyadenosine residue can replace its β-anomer without loss of pairing energy.<sup>15</sup> Here, we show that an α-nucleoside within an otherwise β-configured third strand can selectively and efficiently recognize a DNA base-pair. Related work on oligonucleotides containing the corresponding β-configured N<sup>7</sup>H-deoxynucleoside as well as both anomeric forms of it is currently in progress.

#### Acknowledgment:

Financial support from the Swiss National Science Foundation and from Ciba-Geigy AG, Basel is gratefully acknowledged.

#### References and Notes:

1. Moser, H. E.; Dervan, P. B. *Science* **1987**, *238*, 645-650.
2. François, J.; Saison-Behmoaras, T.; Hélène, C. *Nucleic Acids Res.* **1988**, *16*, 11431-11440.
3. Durland, R. H.; Kessler, D. J.; Gunnell, S.; Duvic, M.; Pettitt, B. M.; Hogan, M. E. *Biochemistry* **1991**, *30*, 9246-9255.
4. Beal, P. A.; Dervan, P. B. *Science* **1991**, *251*, 1360-1363.
5. Best, G. C.; Dervan, P. B. *J. Am. Chem. Soc.* **1995**, *117*, 1187-1193.
6. Greenberg, W. A.; Dervan, P. B. *J. Am. Chem. Soc.* **1995**, *117*, 5016-5022.
7. Thuong, N. T.; Hélène, C. *Angew. Chem. Intl. Ed. Engl.* **1993**, *32*, 666-690.
8. Marfurt, J.; Stulz, E.; Trafelet, H. U.; Zingg, A.; Leumann, C.; Hazenkamp, M.; Judd, R.; Schenker, S.; Strouse, G.; Ward, T. R.; Förtsch, M.; Hauser, J.; Bürgi, H. *Acta Crystallogr., Sect. C* **1996**, *C52*, 713-716.
9. Jones, R. J.; Lin, K.-Y.; Milligan, J. F.; Wadwani, S.; Matteucci, M. D. *J. Org. Chem.* **1993**, *58*, 2983-2991.
10. Hunziker, J.; Priestley, E. S.; Brunar, H.; Dervan, P. B. *J. Am. Chem. Soc.* **1995**, *117*, 2661-2662.
11. Priestley, E. S.; Dervan, P. B. *J. Am. Chem. Soc.* **1995**, *117*, 4761-4765.
12. Brunar, H.; Dervan, P. B. *Nucleic Acids Res.* **1996**, *24*, 1987-1991.
13. Rao, T. S.; Durland, R. H.; Revankar, G. R. *J. Heterocyclic Chem.* **1994**, *31*, 935-940.
14. Noonberg, S. B.; François, J.; Praseuth, D.; Guieysse-Peugeot, A.; Lacoste, J.; Garestier, T.; Hélène, C. *Nucleic Acids Res.* **1995**, *23*, 4042-4049.
15. Ide, H.; Shimizu, Y.; Kimura, Y.; Sakamoto, S.; Makino, K.; Glackin, M.; Wallace, S. S.; Nakamuta, H.; Sasaki, M.; Sugimoto, N. *Biochemistry* **1995**, *34*, 6947-6955.