

## Full Paper

# Synthesis and Cytotoxic Evaluation of Some New Phthalazinylpiperazine Derivatives

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A new series of 1,4-disubstituted phthalazinylpiperazine derivatives **7a–f**, **12a–f** and **20a–f** were designed and synthesized in order to develop potent and selective antitumor agents. The target compounds were screened for their cytotoxic activities against A549, HT-29 and MDA-MB-231 cancer cell lines *in vitro*. Among them, compounds **7a–f** exhibited excellent selectivity for MDA-MB-231 with IC<sub>50</sub> values ranging from 0.013  $\mu$ M to 0.079  $\mu$ M. The most promising compound, **7e** (IC<sub>50</sub> = 2.19  $\mu$ M, 2.19  $\mu$ M, 0.013  $\mu$ M), was 9.3, 10, and  $4.9 \times 10^3$  times more active than vatalanib (IC<sub>50</sub> = 20.27  $\mu$ M, 21.96  $\mu$ M, 63.90  $\mu$ M), respectively.

**Keywords:** Anti-cancer / Phthalazinylpiperazine derivatives / Synthesis

Received: July 12, 2011; Revised: August 23, 2011; Accepted: August 26, 2011

DOI 10.1002/ardp.201100250

## Introduction

The search for new anticancer chemotherapeutic agents continues to be an active area of research in many companies. During the last decade, many synthetic phthalazine derivatives have been reported as antitumor agents [1–5]. Among them, vatalanib (PTK-787), an anilinophthalazine compound, is currently undergoing phase III clinical trials for renal cell carcinoma cancer and glioblastoma. Moreover, several additional phase II studies are underway, such as myelodysplasia, meningioma and metastatic melanoma, *etc.* [6] Therefore, the broad spectrum and potent antitumor activity of vatalanib has attracted much attention in recent years.

In order to improve upon the antitumor efficacy of vatalanib, many new phthalazine derivatives were developed [7–9], such as compounds **I** and **II** (Figure 1) [7], which had been proved highly potent against human liver cancer *in vitro*. From these compounds, we could disclose that the introduction of a piperaziny group between phthalazine and aniline

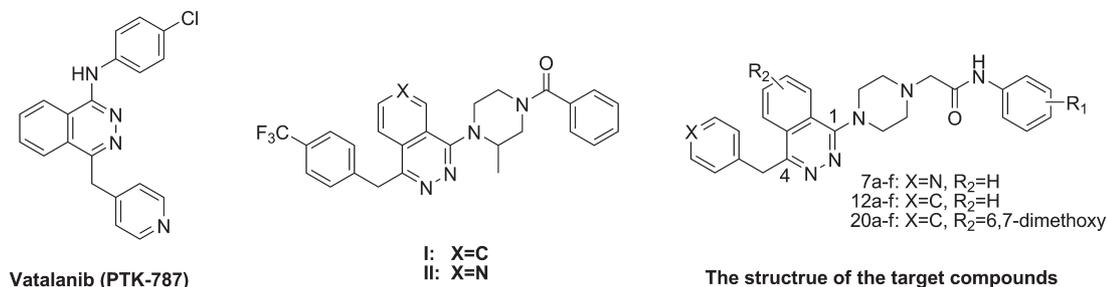
was suitable, and the 4-phenyl group was generally well tolerated. So we combined vatalanib with the piperaziny group in one structure, and inserted an acetyl-flexible linker into the structure to get the new chemical entities (Figure 1). Various substituents **R**<sub>1</sub> were introduced into the aniline ring to investigate their influence on cytotoxic activities to obtain the first series of compounds **7a–f**. Furthermore, the pyridinyl group at position 4 of compounds **7a–f** was replaced by phenyl group to produce compounds **12a–f**. For further research on the influence of electrical effects in the phthalazine nucleus, we introduced two electron-donating methoxy groups into **12a–f** to get another series of derivatives **20a–f**. As a result, eighteen novel 1,4-disubstituted phthalazinylpiperazine derivatives (Figure 1) were designed and synthesized, and their cytotoxicities against A549, HT-29 and MDA-MB-231 cancer cell lines were tested.

## Results and discussion

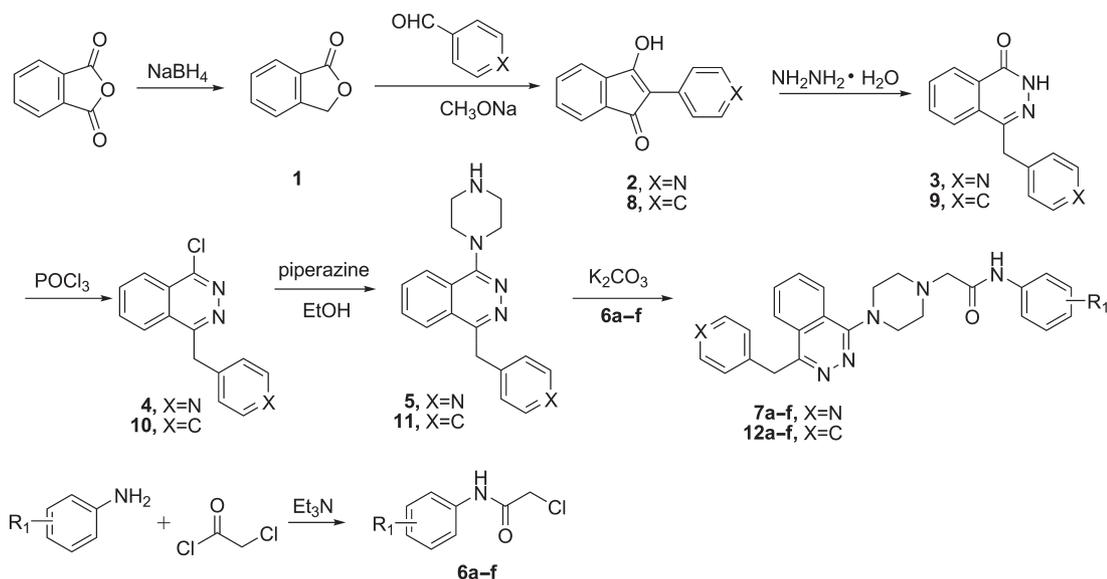
### Chemistry

The synthesis of target compounds **7a–f** and **12a–f** was achieved using a convenient seven-step procedure starting from phthalic anhydride depicted in Scheme 1. The commercially available phthalic anhydride was reduced by NaBH<sub>4</sub> in dry tetrahydrofuran to give compound **1**, which was then reacted with 4-pyridine carboxaldehyde in methanol and

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**Figure 1.** Structures of vatalanib, compound I, II and the target compounds.



**Scheme 1.** Synthesis of target compounds 7a-f and 12a-f.

ethyl propionate under basic condition to obtain compound **2**. The key intermediate, pyridinylphthalazone **3**, was prepared from compound **2** in 80% hydrazine hydrate at 100°C for 5 h in 75% yield. Then, compound **3** was chlorinated using phosphoryl chloride in acetonitrile to produce compound **4** [10]. Next, compound **5** was synthesized by the reaction of compound **4** with excess piperazine in EtOH. Furthermore, the side chains **6a-f** were prepared *via* a series of substituted anilines with 2-chloroacetyl chloride in presence of Et<sub>3</sub>N. Another phenylphthalazinylpiperazine derivative **11** was obtained according to the same method described for compound **5** except that 4-pyridine carboxaldehyde was replaced by benzaldehyde.

The synthesis of target compounds **20a-f** was summarized in Scheme 2. The starting material, ethyl 3,4-dimethoxybenzoate, was methylated with (CH<sub>3</sub>O)<sub>2</sub>SO<sub>2</sub> under basic condition to give compound **13**, which was subsequently hydrolyzed to get compound **14**. The 5,6-dimethoxyisobenzofuran-1(3H)-one **15** was prepared *via* cyclization of compound

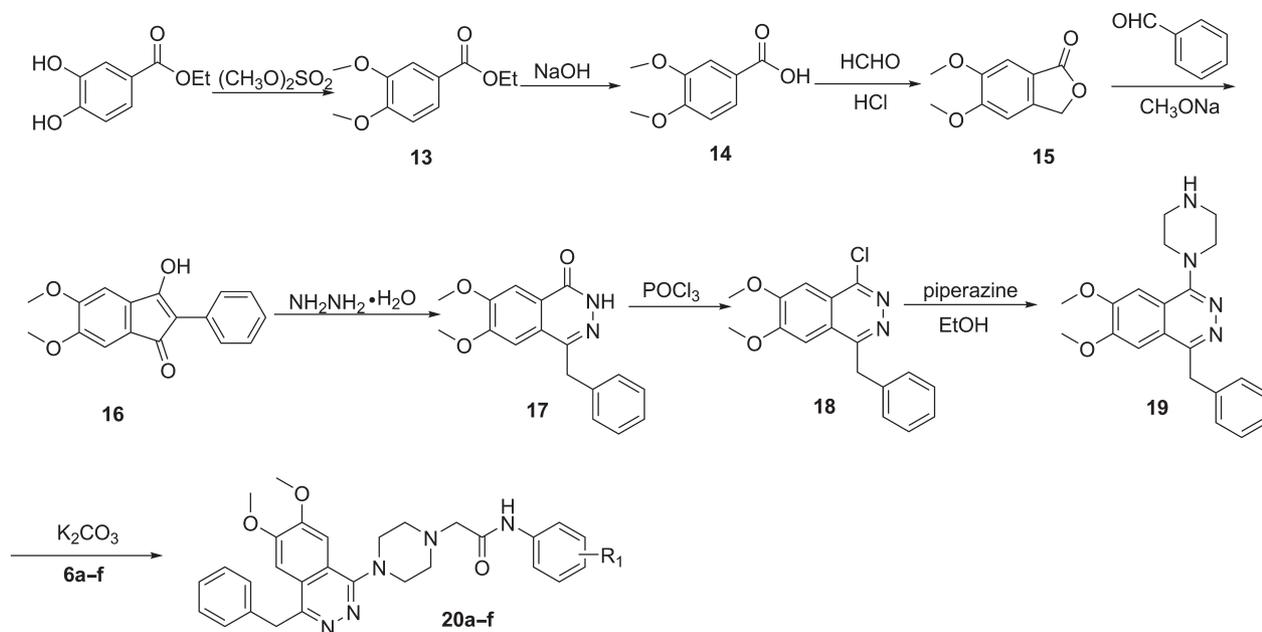
**14** with 37% formaldehyde solution in the presence of hydrogen chloride gas [11]. Similarly, the intermediate **19** was obtained as in the manner described for compound **5**, except that 4-pyridine carboxaldehyde was replaced with benzaldehyde.

Finally, the target compounds **7a-f**, **12a-f** and **20a-f** were successfully obtained *via* the reaction of the intermediate **5**, **11**, and **19** with **6a-f** in the presence of K<sub>2</sub>CO<sub>3</sub> in acetone, respectively.

The structures and the data of <sup>1</sup>H-NMR, MS and elemental analysis were outlined in Table 1 and Table 2.

### Evaluation of biological activity

The preliminary cytotoxic activities of compounds **7a-f**, **12a-f** and **20a-f** on A549 (human lung cancer cell lines), HT-29 (human colon cancer cell line) and MDA-MB-231 (human breast cancer cell line) were investigated *in vitro*. These compounds' properties are summarized in Table 3 and are compared to those of vatalanib.



**Scheme 2.** Synthesis of target compounds 20a–f.

The data in Table 3 indicated that the cytotoxic activities of compounds **7a–f**, **20a** and **20d** were more active than vatalanib against all three human cancer cell lines, and most of them were comparable to that of vatalanib.

Obviously, 4-pyridinylphthalazine derivatives (**7a–f**) were better than their 4-phenyl analogues (**12a–f**). As shown in Table 3, compounds **7a–f** displayed excellent cytotoxic activi-

ties for A549 and HT-29 cell lines with  $IC_{50}$  values in the single-digit  $\mu\text{M}$ . In particular, compounds **7a–f** showed  $IC_{50}$  values ranging from 0.013  $\mu\text{M}$  to 0.079  $\mu\text{M}$  against MDA-MB-231 cell line, which were  $10^2$  to  $10^3$ -fold more active than those of the corresponding compounds **12a–f**.

As well, most of the target compounds exhibited better activities against MDA-MB-231 cell line than did vatalanib.

**Table 1.** The structures, MS, elemental analysis and physical data of compounds **7a–f**, **12a–f** and **20a–f**.

Compd.	X	R <sub>1</sub>	m.p. (°C)	Elemental Analysis			[MH <sup>+</sup> ] (m/z)	Yields (%)
				C%	H%	N%		
<b>7a</b>	N	4-CH <sub>3</sub>	188–189	71.66(71.54)*	6.24 (6.19)	18.57 (18.66)	453.2	15%
<b>7b</b>	N	2-CF <sub>3</sub>	149–150	64.02 (63.99)	4.97 (4.98)	16.59 (16.63)	507.1	47%
<b>7c</b>	N	3-Cl	76–77	66.03 (66.00)	5.33 (5.35)	17.77 (17.76)	473.1	38%
<b>7d</b>	N	4-OCF <sub>3</sub>	76–77	62.06 (62.00)	4.82 (4.83)	16.08 (16.02)	475.1	44%
<b>7e</b>	N	4-F	215–216	68.40 (68.32)	5.52 (5.58)	18.41 (18.43)	523.3	47%
<b>7f</b>	N	3,5-diCF <sub>3</sub>	80–81	58.54 (58.51)	4.21 (4.23)	14.63(14.60)	457.5	48%
<b>12a</b>	C	4-CH <sub>3</sub>	165–167	74.47(74.43)	6.47 (6.51)	15.51 (15.54)	575.5	65%
<b>12b</b>	C	2-CF <sub>3</sub>	154–155	66.52 (66.46)	5.18(5.19)	13.85 (13.89)	452.2	66%
<b>12c</b>	C	3-Cl	80–81	68.71 (68.66)	5.55 (5.59)	14.84 (14.74)	506.1	33%
<b>12d</b>	C	4-OCF <sub>3</sub>	77–78	64.48 (64.55)	5.02 (4.94)	13.43 (13.45)	472.1	49%
<b>12e</b>	C	4-F	77–78	71.19 (71.22)	5.75 (5.64)	15.37 (15.45)	474.0	51%
<b>12f</b>	C	3,5-diCF <sub>3</sub>	78–79	60.73 (60.77)	4.39 (4.44)	12.21 (12.12)	456.3	37%
<b>20a</b>	C	4-CH <sub>3</sub>	90–92	70.43 (70.40)	6.50 (6.52)	13.69 (13.66)	574.3	25%
<b>20b</b>	C	2-CF <sub>3</sub>	70–71	63.71 (63.66)	5.35 (5.39)	12.38 (12.40)	512.3	33%
<b>20c</b>	C	3-Cl	100–101	65.47(65.53)	5.68(5.62)	13.16 (13.10)	566.3	41%
<b>20d</b>	C	4-OCF <sub>3</sub>	111–112	61.95(62.04)	5.20 (5.27)	12.04 (12.00)	532.4	36%
<b>20e</b>	C	4-F	106–107	67.56 (67.46)	5.86 (5.91)	13.58 (13.49)	582.4	45%
<b>20f</b>	C	3,5-diCF <sub>3</sub>	92–93	58.77 (58.86)	4.61 (4.54)	11.05 (11.00)	516.3	41%
							634.5	

**Table 2.** The IR, <sup>1</sup>H-NMR data of compounds **7a–f**, **12a–f** and **20a–f**.

Compd.	IR, $\bar{\nu}/\text{cm}^{-1}$	<sup>1</sup> H-NMR (300 MHz, DMSO- <i>d</i> <sub>6</sub> ), $\delta$
<b>7a</b>	3412.1 ( $\nu_{\text{NH}}$ ), 1675.8 ( $\nu_{\text{C=O}}$ ), 1601.4, 1521.6, 1409.9 ( $\nu_{\text{C=C}}$ ), 818.0, 779.1 ( $\delta_{\text{=CH}}$ )	9.78 (s, 1H, CNHC=O), 8.44 (d, <i>J</i> = 5.8 Hz, 2H, pyridinyl- <b>2H,6H</b> ), 8.15 (m, 2H, phthalazine- <b>5H,8H</b> ), 7.93 (m, 2H, phthalazine- <b>6H,7H</b> ), 7.54 (d, <i>J</i> = 8.3 Hz, 2H, Ph- <b>2H,6H</b> ), 7.31 (d, <i>J</i> = 5.6 Hz, 2H, pyridine- <b>3H,5H</b> ), 7.13 (d, <i>J</i> = 8.3 Hz, 2H, Ph- <b>3H,5H</b> ), 4.62 (s, 2H, CCH <sub>2</sub> C), 3.46 (s, 4H, (CH <sub>2</sub> CH <sub>2</sub> ) <sub>2</sub> N), 3.26 (s, 2H, NCH <sub>2</sub> C=O), 2.83 (s, 4H, CH <sub>2</sub> N(CH <sub>2</sub> ) <sub>2</sub> ), 2.25 (s, 3H, CCH <sub>3</sub> ).
<b>7b</b>	3298.7 ( $\nu_{\text{NH}}$ ), 1698.7 ( $\nu_{\text{C=O}}$ ), 1590.1, 1529.3 ( $\nu_{\text{C=C}}$ ), 766.5 ( $\delta_{\text{=CH}}$ )	9.96 (s, 1H, CNHC=O), 8.44 (d, <i>J</i> = 5.6 Hz, 2H, pyridinyl- <b>2H,6H</b> ), 8.27 (d, <i>J</i> = 8.1 Hz, 1H, Ph- <b>3H</b> ), 8.18 (m, 2H, phthalazine- <b>5H,8H</b> ), 7.94 (m, 2H, phthalazine- <b>6H,7H</b> ), 7.72 (m, 2H, Ph- <b>4H,5H</b> ), 7.34 (m, 3H, pyridinyl- <b>3H,5H</b> + Ph- <b>6H</b> ), 4.63 (s, 2H, CCH <sub>2</sub> C), 3.47 (s, 4H, (CH <sub>2</sub> CH <sub>2</sub> ) <sub>2</sub> N), 3.34 (s, 2H, NCH <sub>2</sub> C=O), 2.89 (s, 4H, CH <sub>2</sub> N(CH <sub>2</sub> ) <sub>2</sub> ).
<b>7c</b>	3269.7 ( $\nu_{\text{NH}}$ ), 1686.5 ( $\nu_{\text{C=O}}$ ), 1595.6, 1518.1 ( $\nu_{\text{C=C}}$ ), 877.0, 778.6, 680.2 ( $\delta_{\text{=CH}}$ )	10.10 (s, 1H, CNHC=O), 8.44 (d, <i>J</i> = 5.6 Hz, 2H, pyridinyl- <b>2H,6H</b> ), 8.16 (m, 2H, phthalazine- <b>5H,8H</b> ), 7.93 (m, 3H, phthalazine- <b>6H,7H</b> + Ph- <b>2H</b> ), 7.58 (d, <i>J</i> = 7.7 Hz, 1H, Ph- <b>6H</b> ), 7.34 (m, 3H, pyridinyl- <b>3H,5H</b> + Ph- <b>5H</b> ), 7.13 (d, <i>J</i> = 7.7 Hz, 1H, Ph- <b>4H</b> ), 4.62 (s, 2H, CCH <sub>2</sub> C), 3.47 (s, 4H, (CH <sub>2</sub> CH <sub>2</sub> ) <sub>2</sub> N), 3.30 (s, 2H, NCH <sub>2</sub> C=O), 2.83 (s, 4H, CH <sub>2</sub> N(CH <sub>2</sub> ) <sub>2</sub> ).
<b>7d</b>	3411.8 ( $\nu_{\text{NH}}$ ), 1688.3 ( $\nu_{\text{C=O}}$ ), 1602.4, 1512.9, 1411.6 ( $\nu_{\text{C=C}}$ ), 805.4, 778.5 ( $\delta_{\text{=CH}}$ )	10.10 (s, 1H, CNHC=O), 8.44 (d, <i>J</i> = 5.1 Hz, 2H, pyridinyl- <b>2H,6H</b> ), 8.15 (m, 2H, phthalazine- <b>5H,8H</b> ), 7.91 (m, 2H, phthalazine- <b>6H,7H</b> ), 7.82 (d, <i>J</i> = 9.0 Hz, 2H, Ph- <b>2H,6H</b> ), 7.32 (d, <i>J</i> = 7.9 Hz, 4H, pyridine- <b>3H,5H</b> + Ph- <b>3H,5H</b> ), 4.62 (s, 2H, CCH <sub>2</sub> C), 3.47 (s, 4H, (CH <sub>2</sub> CH <sub>2</sub> ) <sub>2</sub> N), 3.30 (s, 2H, NCH <sub>2</sub> C=O), 2.83 (s, 4H, CH <sub>2</sub> N(CH <sub>2</sub> ) <sub>2</sub> ).
<b>7e</b>	3263.4 ( $\nu_{\text{NH}}$ ), 1671.0 ( $\nu_{\text{C=O}}$ ), 1532.6, 1508.8 ( $\nu_{\text{C=C}}$ ), 836.1, 779.4 ( $\delta_{\text{=CH}}$ )	9.93 (s, 1H, CNHC=O), 8.44 (d, <i>J</i> = 5.6 Hz, 2H, pyridinyl- <b>2H,6H</b> ), 8.15 (m, 2H, phthalazine- <b>5H,8H</b> ), 7.92 (m, 2H, phthalazine- <b>6H,7H</b> ), 7.71 (m, 2H, Ph- <b>2H,6H</b> ), 7.31 (d, <i>J</i> = 5.7 Hz, 2H, pyridine- <b>3H,5H</b> ), 7.16 (t, <i>J</i> = 8.8 Hz, 2H, Ph- <b>3H,5H</b> ), 4.62 (s, 2H, CCH <sub>2</sub> C), 3.47 (s, 4H, (CH <sub>2</sub> CH <sub>2</sub> ) <sub>2</sub> N), 3.27 (s, 2H, NCH <sub>2</sub> C=O), 2.83 (s, 4H, CH <sub>2</sub> N(CH <sub>2</sub> ) <sub>2</sub> ).
<b>7f</b>	3422.1 ( $\nu_{\text{NH}}$ ), 1695.5 ( $\nu_{\text{C=O}}$ ), 1601.6, 1532.4 ( $\nu_{\text{C=C}}$ ), 778.3, 703.2, 680.8 ( $\delta_{\text{=CH}}$ )	10.55 (s, 1H, CNHC=O), 8.44 (m, 4H, pyridinyl- <b>2H,6H</b> + phthalazine- <b>5H,8H</b> ), 8.15 (m, 2H, Ph- <b>2H,6H</b> ), 7.93 (m, 2H, phthalazine- <b>6H,7H</b> ), 7.78 (s, 1H, Ph- <b>4H</b> ), 7.32 (d, <i>J</i> = 5.8 Hz, 2H, Ph- <b>3H,5H</b> ), 4.62 (s, 2H, CCH <sub>2</sub> C), 3.49 (s, 4H, (CH <sub>2</sub> CH <sub>2</sub> ) <sub>2</sub> N), 3.36 (s, 2H, NCH <sub>2</sub> C=O), 2.85 (s, 4H, CH <sub>2</sub> N(CH <sub>2</sub> ) <sub>2</sub> ).
<b>12a</b>	3391.2 ( $\nu_{\text{NH}}$ ), 1671.6 ( $\nu_{\text{C=O}}$ ), 1521.5, 1407.7 ( $\nu_{\text{C=C}}$ ), 818.6, 763.2, 713.6 ( $\delta_{\text{=CH}}$ )	9.78 (s, 1H, CNHC=O), 8.18 (m, 2H, phthalazine- <b>5H,8H</b> ), 7.90 (m, 2H, phthalazine- <b>6H,7H</b> ), 7.54 (d, <i>J</i> = 8.4 Hz, 2H, aniline- <b>2H,6H</b> ), 7.31 (m, 4H, Ph- <b>2H,3H,5H,6H</b> ), 7.10 (m, 3H, Ph- <b>4H</b> , aniline- <b>3H,5H</b> ), 4.58 (s, 2H, CCH <sub>2</sub> C), 3.46 (s, 4H, (CH <sub>2</sub> CH <sub>2</sub> ) <sub>2</sub> N), 3.26 (s, 2H, NCH <sub>2</sub> C=O), 2.83 (s, 4H, CH <sub>2</sub> N(CH <sub>2</sub> ) <sub>2</sub> ), 2.25 (s, 3H, CCH <sub>3</sub> ).
<b>12b</b>	3264.2 ( $\nu_{\text{NH}}$ ), 1698.4 ( $\nu_{\text{C=O}}$ ), 1588.1, 1532.7 ( $\nu_{\text{C=C}}$ ), 768.5 ( $\delta_{\text{=CH}}$ )	9.97 (s, 1H, CNHC=O), 8.18 (m, 3H, aniline- <b>3H</b> , phthalazine- <b>5H,8H</b> ), 7.88 (m, 2H, phthalazine- <b>6H,7H</b> ), 7.91 (m, 2H, aniline- <b>4H,5H</b> ), 7.32 (m, 5H, Ph- <b>2H,3H,5H,6H</b> + aniline- <b>6H</b> ), 7.18 (t, <i>J</i> = 7.2 Hz, 1H, Ph- <b>4H</b> ), 4.59 (s, 2H, CCH <sub>2</sub> C), 3.46 (s, 4H, (CH <sub>2</sub> CH <sub>2</sub> ) <sub>2</sub> N), 3.34 (s, 2H, NCH <sub>2</sub> C=O), 2.89 (s, 4H, CH <sub>2</sub> N(CH <sub>2</sub> ) <sub>2</sub> ).
<b>12c</b>	3276.2 ( $\nu_{\text{NH}}$ ), 1687.4 ( $\nu_{\text{C=O}}$ ), 1594.4, 1516.9 ( $\nu_{\text{C=C}}$ ), 778.8, 713.8, 675.1 ( $\delta_{\text{=CH}}$ )	10.10 (s, 1H, CNHC=O), 8.17 (m, 2H, phthalazine- <b>5H,8H</b> ), 7.92 (m, 3H, phthalazine- <b>6H,7H</b> + aniline- <b>2H</b> ), 7.61 (d, <i>J</i> = 8.3 Hz, 1H, aniline- <b>6H</b> ), 7.31 (m, 5H, Ph- <b>2H,3H,5H,6H</b> + aniline- <b>5H</b> ), 7.13 (m, 2H, Ph- <b>4H</b> , aniline- <b>4H</b> ), 4.58 (s, 2H, CCH <sub>2</sub> C), 3.46 (s, 4H, (CH <sub>2</sub> CH <sub>2</sub> ) <sub>2</sub> N), 3.30 (s, 2H, NCH <sub>2</sub> C=O), 2.83 (s, 4H, CH <sub>2</sub> N(CH <sub>2</sub> ) <sub>2</sub> ).
<b>12d</b>	3303.0 ( $\nu_{\text{NH}}$ ), 1688.4 ( $\nu_{\text{C=O}}$ ), 1513.8, 1410.8 ( $\nu_{\text{C=C}}$ ), 846.7, 762.8, 712.8 ( $\delta_{\text{=CH}}$ )	10.11 (s, 1H, CNHC=O), 8.18 (m, 2H, phthalazine- <b>5H,8H</b> ), 7.89 (m, 2H, phthalazine- <b>6H,7H</b> ), 7.82 (d, <i>J</i> = 8.6 Hz, 2H, aniline- <b>2H,6H</b> ), 7.29 (m, 6H, Ph- <b>2H,3H,5H,6H</b> + aniline- <b>3H,5H</b> ), 7.14 (t, <i>J</i> = 7.1 Hz, 1H, Ph- <b>4H</b> ), 4.58 (s, 2H, CCH <sub>2</sub> C), 3.46 (s, 4H, (CH <sub>2</sub> CH <sub>2</sub> ) <sub>2</sub> N), 3.31 (s, 2H, NCH <sub>2</sub> C=O), 2.84 (s, 4H, CH <sub>2</sub> N(CH <sub>2</sub> ) <sub>2</sub> ).
<b>12e</b>	3282.3 ( $\nu_{\text{NH}}$ ), 1695.8 ( $\nu_{\text{C=O}}$ ), 1506.3 ( $\nu_{\text{C=C}}$ ), 824.4, 762.9 ( $\delta_{\text{=CH}}$ )	9.93 (s, 1H, CNHC=O), 8.17 (m, 2H, phthalazine- <b>5H,8H</b> ), 7.93 (m, 2H, phthalazine- <b>6H,7H</b> ), 7.70 (m, 2H, aniline- <b>2H,6H</b> ), 7.31 (m, 4H, Ph- <b>2H,3H,5H,6H</b> ), 7.16 (t, <i>J</i> = 8.9 Hz, 3H, Ph- <b>4H</b> , aniline- <b>3H,5H</b> ), 4.58 (s, 2H, CCH <sub>2</sub> C), 3.46 (s, 4H, (CH <sub>2</sub> CH <sub>2</sub> ) <sub>2</sub> N), 3.27 (s, 2H, NCH <sub>2</sub> C=O), 2.83 (s, 4H, CH <sub>2</sub> N(CH <sub>2</sub> ) <sub>2</sub> ).
<b>12f</b>	3282.8 ( $\nu_{\text{NH}}$ ), 1696.7 ( $\nu_{\text{C=O}}$ ), 1532.4 ( $\nu_{\text{C=C}}$ ), 762.4, 700.4, 680.9 ( $\delta_{\text{=CH}}$ )	10.58 (s, 1H, CNHC=O), 8.47 (s, 2H, aniline- <b>2H,6H</b> ), 8.17 (m, 2H, phthalazine- <b>5H,8H</b> ), 7.89 (m, 2H, phthalazine- <b>6H,7H</b> ), 7.78 (s, 1H, aniline- <b>4H</b> ), 7.26 (m, 4H, Ph- <b>2H,3H,5H,6H</b> ), 7.16 (t, <i>J</i> = 6.9 Hz, 1H, Ph- <b>4H</b> ), 4.58 (s, 2H, CCH <sub>2</sub> C), 3.48 (s, 4H, (CH <sub>2</sub> CH <sub>2</sub> ) <sub>2</sub> N), 3.36 (s, 2H, NCH <sub>2</sub> C=O), 2.85 (s, 4H, CH <sub>2</sub> N(CH <sub>2</sub> ) <sub>2</sub> ).
<b>20a</b>	3419.7 ( $\nu_{\text{NH}}$ ), 1690.2 ( $\nu_{\text{C=O}}$ ), 1515.0, 1464.7, 1420.7 ( $\nu_{\text{C=C}}$ ), 817.4 ( $\delta_{\text{=CH}}$ )	9.76 (s, 1H, CNHC=O), 7.54 (d, <i>J</i> = 8.2 Hz, 2H, aniline- <b>2H,6H</b> ), 7.34 (m, 6H, phthalazine- <b>5H,8H</b> + Ph- <b>2H,3H,5H,6H</b> ), 7.13 (m, 3H, aniline- <b>3H,5H</b> + Ph- <b>4H</b> ), 4.55 (s, 2H, CCH <sub>2</sub> C), 3.94 (s, 3H, OCH <sub>3</sub> ), 3.87 (s, 3H, OCH <sub>3</sub> ), 3.40 (s, 4H, (CH <sub>2</sub> CH <sub>2</sub> ) <sub>2</sub> N), 3.25 (s, 2H, NCH <sub>2</sub> C=O), 2.82 (s, 4H, CH <sub>2</sub> N(CH <sub>2</sub> ) <sub>2</sub> ), 2.25 (s, 3H, CCH <sub>3</sub> ).

**Table 2.** (continued)

Compd.	IR, $\bar{\nu}/\text{cm}^{-1}$	$^1\text{H-NMR}$ (300 MHz, DMSO- $d_6$ ), $\delta$
20b	3317.8 ( $\nu_{\text{NH}}$ ), 1697.1 ( $\nu_{\text{C=O}}$ ), 1514.1, 1457.1 ( $\nu_{\text{C=C}}$ ), 766.4 ( $\delta_{\text{CH}}$ ).	9.97 (s, 1H, CNHC=O), 8.26 (d, $J = 8.3$ Hz, 1H, aniline-3H), 7.85 (t, $J = 7.3$ Hz, 1H, aniline-4H), 7.74 (m, 2H, aniline-5H,6H), 7.35 (m, 7H, phthalazine-5H,8H + Ph-2H,3H,4H,5H,6H), 4.56 (s, 2H, CCH <sub>2</sub> C), 3.95 (s, 3H, OCH <sub>3</sub> ), 3.88 (s, 3H, OCH <sub>3</sub> ), 3.38 (m, 6H, (CH <sub>2</sub> CH <sub>2</sub> ) <sub>2</sub> N + NCH <sub>2</sub> C=O), 2.88 (s, 4H, CH <sub>2</sub> N(CH <sub>2</sub> ) <sub>2</sub> ).
20c	3299.8 ( $\nu_{\text{NH}}$ ), 1687.3 ( $\nu_{\text{C=O}}$ ), 1513.2, 1421.0 ( $\nu_{\text{C=C}}$ ), 817.7, 778.7 ( $\delta_{\text{CH}}$ ).	10.06 (s, 1H, CNHC=O), 7.90 (s, 1H, aniline-2H), 7.59 (d, $J = 8.2$ Hz, 1H, aniline-6H), 7.37 (m, 7H, phthalazine-5H,8H + Ph-2H,3H,5H,6H + aniline-5H), 7.24 (m, 2H, Ph-4H + aniline-4H), 4.55 (s, 2H, CCH <sub>2</sub> C), 3.94 (s, 3H, OCH <sub>3</sub> ), 3.87 (s, 3H, OCH <sub>3</sub> ), 3.41 (m, 6H, (CH <sub>2</sub> CH <sub>2</sub> ) <sub>2</sub> N + NCH <sub>2</sub> C=O), 2.82 (s, 4H, CH <sub>2</sub> N(CH <sub>2</sub> ) <sub>2</sub> ).
20d	3411.5 ( $\nu_{\text{NH}}$ ), 1690.9 ( $\nu_{\text{C=O}}$ ), 1513.4, 1421.7 ( $\nu_{\text{C=C}}$ ), 817.3 ( $\delta_{\text{CH}}$ ).	10.07 (s, 1H, CNHC=O), 7.78 (d, $J = 8.8$ Hz, 2H, aniline-2H,6H), 7.34 (m, 8H, phthalazine-5H,8H + Ph-2H,3H,5H,6H + aniline-3H,5H), 7.18 (t, $J = 7.4$ Hz, 1H, Ph-4H), 4.55 (s, 2H, CCH <sub>2</sub> C), 3.96 (s, 3H, OCH <sub>3</sub> ), 3.88 (s, 3H, OCH <sub>3</sub> ), 3.34 (m, 6H, (CH <sub>2</sub> CH <sub>2</sub> ) <sub>2</sub> N + NCH <sub>2</sub> C=O), 2.82 (s, 4H, CH <sub>2</sub> N(CH <sub>2</sub> ) <sub>2</sub> ).
20e	3426.1 ( $\nu_{\text{NH}}$ ), 1687.5 ( $\nu_{\text{C=O}}$ ), 1511.2, 1421.6 ( $\nu_{\text{C=C}}$ ), 836.1 ( $\delta_{\text{CH}}$ ).	10.30 (brs, 1H, CNHC=O), 7.70 (dd, $J = 8.7$ Hz, 2H, aniline-2H,6H), 7.37 (m, 6H, phthalazine-5H,8H + Ph-2H,3H,5H,6H), 7.17 (m, 3H, Ph-4H + aniline-3H,5H), 4.57 (s, 2H, CCH <sub>2</sub> C), 3.96 (s, 3H, OCH <sub>3</sub> ), 3.88 (s, 3H, OCH <sub>3</sub> ), 3.49 (s, 4H, (CH <sub>2</sub> CH <sub>2</sub> ) <sub>2</sub> N), 3.37 (s, 2H, NCH <sub>2</sub> C=O), 3.03 (s, 4H, CH <sub>2</sub> N(CH <sub>2</sub> ) <sub>2</sub> ).
20f	3398.5 ( $\nu_{\text{NH}}$ ), 1699.5 ( $\nu_{\text{C=O}}$ ), 1514.1, 1470.3 ( $\nu_{\text{C=C}}$ ), 818.2 ( $\delta_{\text{CH}}$ ).	11.17 (brs, 1H, CNHC=O), 8.44 (s, 2H, aniline-2H,6H), 7.81 (s, 1H, aniline-4H), 7.38 (m, 6H, phthalazine-5H,8H + Ph-2H,3H,5H,6H), 7.18 (t, $J = 7.2$ Hz, 3H, Ph-4H + aniline-3H,5H), 4.61 (s, 2H, CCH <sub>2</sub> C), 3.98 (s, 3H, OCH <sub>3</sub> ), 3.90 (s, 3H, OCH <sub>3</sub> ), 3.56 (s, 4H, (CH <sub>2</sub> CH <sub>2</sub> ) <sub>2</sub> N), 3.15 (s, 2H, NCH <sub>2</sub> C=O), 2.84 (s, 4H, CH <sub>2</sub> N(CH <sub>2</sub> ) <sub>2</sub> ).

**Table 3.** The antitumor activities of compounds 7a–f, 12a–f and 20a–f.

Compd.	IC <sub>50</sub> <sup>a</sup> ( $\mu\text{M}$ )			Compd.	IC <sub>50</sub> ( $\mu\text{M}$ )		
	A549	HT-29	MDA-MB-231		A549	HT-29	MDA-MB-231
7a	11.05	2.32	0.044	12e	20.87	26.36	18.01
7b	1.58	2.57	0.079	12f	NA <sup>b</sup>	10.82	127.3
7c	1.75	2.22	0.021	20a	3.71	3.71	4.50
7d	3.44	2.49	0.038	20b	49.51	93.71	68.95
7e	2.19	2.19	0.013	20c	114.6	4.70	20.68
7f	4.53	0.82	0.052	20d	13.93	4.30	39.55
12a	68.70	59.84	17.29	20e	25.02	NA <sup>a</sup>	NA
12b	NA <sup>a</sup>	31.67	15.24	20f	162.6	26.26	75.76
12c	23.98	11.46	4.24				
12d	NA <sup>a</sup>	26.86	19.19	vatalanib	20.27	21.96	63.90

<sup>a</sup> IC<sub>50</sub>: The IC<sub>50</sub> was the average of at least two independent experiments.

<sup>b</sup> NA: Compounds having IC<sub>50</sub> value > 200  $\mu\text{M}$ .

Moreover, the pharmacological results indicated that the cytotoxicity of compounds 7a–f against MDA-MB-231 cell line were 10 to 10<sup>2</sup>-fold higher than A549 and HT-29 cell lines, reflecting excellent selectivity for a particular type of human breast cancer.

Compounds 20a–f were chosen as an example to study the effect of electron-donating groups on antitumor activities. Among the six compounds bearing 6,7-dimethoxy groups, the most promising compound, 20a (IC<sub>50</sub> = 3.71  $\mu\text{M}$ , 3.71  $\mu\text{M}$ , 4.50  $\mu\text{M}$ ), was 5.5, 5.5, and 14-fold more active than vatalanib (IC<sub>50</sub> = 20.27  $\mu\text{M}$ , 21.96  $\mu\text{M}$ , 63.90  $\mu\text{M}$ ), respectively. However, it was disappointing to note that

the potencies of compounds 20a–f were much lower than compounds 7a–f.

The cytotoxic activities were not influenced by the introduction of different groups with various substituted position, spatial and electrical effects on aniline moiety.

## Experimental

### Chemistry

Melting points were measured with a Büchi Melting Point B-540 apparatus (Büchi Labortechnik, Flawil, Switzerland) and are uncorrected. Mass spectra (MS) were taken in ESI mode on an

Agilent 1100 LC-MS (Agilent, Palo Alto, CA, USA). Proton ( $^1\text{H}$ ), nuclear magnetic resonance spectroscopy were performed using Bruker ARX-300, 300MHz spectrometers (Bruker Bioscience, Billerica, MA, USA) with TMS as an internal standard. Elemental analysis were conducted with a Carlo-Erba 1106 Elemental analysis instrument (Carlo Erba, Milan, Italy). Infrared spectra (KBr disks) were recorded with a Bruker IFS 55 instrument (Bruker). Unless otherwise noted, all solvents and reagents were commercially available and used without further purification.

#### Synthesis of isobenzofuran-1(3H)-one (1)

$\text{NaBH}_4$  (126.2 g, 3.32 mol) was added slowly to a stirred solution of phthalic anhydride (123 g, 0.83 mol) in dry THF (750 mL) at 0–10°C, and then anhydrous methanol was added drop-wise to the reaction mixture at 0–10°C. The reaction mixture was stirred at room temperature for 3 h, and concentrated under vacuum. The residue was poured into a 10% hydrochloric acid aqueous solution (1000 mL), stirred vigorously for 3 h and separated by filtration. The solid was alkalized with a 10%  $\text{Na}_2\text{CO}_3$  aqueous solution (1000 mL), stirred for 1 h, and separated by filtration to give **1** as a white crystal (78 g, 70%).

#### Synthesis of 3-hydroxy-2-(pyridin-4-yl)-1H-inden-1-one (2)

A solution of **1** (50 g, 0.37 mol) and 4-pyridine carboxaldehyde (42 g, 0.39 mol) in ethyl propionate (200 mL) was added to a stirred solution of  $\text{CH}_3\text{ONa}$  (82 g, 1.49 mol) in MeOH (800 mL) at 0–10°C. The reaction mixture was stirred at room temperature for 1 h, heated to reflux for 1 h, and concentrated under vacuum. The residue was poured into ice water (1000 mL), acidified with glacial acetic acid to pH 2–3, and separated by filtration to give **2** as an orange solid (43.8 g, 52.6%).

#### Synthesis of 4-(pyridin-4-ylmethyl)phthalazin-1(2H)-one (3)

A solution of 80% hydrazine hydrate ( $\text{NH}_2\text{NH}_2 \cdot \text{H}_2\text{O}$ ) (306.6 mL, 4.9 mol) and compound **2** (43.8 g, 0.20 mol) was stirred at 100°C for 5 h. The mixture was cooled, separated by filtration, and washed with EtOH to give **3** as a light yellow crystal (34.9 g, 75%).

#### Synthesis of 1-chloro-4-(pyridin-4-ylmethyl)phthalazine (4)

Compound **3** (17.5 g, 0.07 mol) was added to a stirred solution of  $\text{POCl}_3$  (170 mL) and  $\text{CH}_3\text{CN}$  (80 mL) at room temperature, and then 3 drops of DMF were added to the mixture. The reaction mixture was heated at 90°C for 3 h and concentrated under vacuum. The residue was poured into ice water (500 mL), alkalized with a 10%  $\text{Na}_2\text{CO}_3$  aqueous solution to pH 7–8, and separated by filtration to give **4** as a red solid (13.5 g, 71.6%).

#### Synthesis of 1-(piperazin-1-yl)-4-(pyridin-4-ylmethyl)-phthalazine (5)

Compound **4** (13.5 g, 0.05 mol) was added portion-wise to a stirred solution of piperazine (30.4 g, 0.4 mol) in anhydrous ethanol at room temperature. The mixture was heated at 60°C for 5 h and concentrated under vacuum. The residue was poured into ice water (200 mL), stirred vigorously for 1 h, and separated by filtration to give **5** as a light yellow solid (11.45 g, 71%).

#### General procedure for preparation of 2-chloro-N-(substituted-phenyl)acetamide (6a–f)

A solution of substituted aniline (0.2 mol), triethylamine (36.7 mL, 0.3 mol), and acetone (250 mL) was cooled to

0–10°C. Keeping the temperature, 2-chloroacetyl chloride (21.4 mL, 0.26 mol) was added drop-wise during the reaction at room temperature for 4–8 h. The mixture was concentrated under vacuum. The residue was poured into ice water (200 mL), stirred at room temperature for 1 h, and separated by filtration to obtain **6a–f**.

#### General procedure for preparation of compound (7a–f)

A mixture of compound **5** (0.2 g, 0.65 mmol), 2-chloro-N-(substituted-phenyl)acetamide (0.72 mmol),  $\text{K}_2\text{CO}_3$  (0.23 g, 1.64 mmol), and acetone (2 mL) was stirred at reflux for 7–12 h and then cooled to room temperature. The liquid was collected by filtration and evaporated. The crude product was purified by chromatography on silica gel using  $\text{MeOH}/\text{CH}_2\text{Cl}_2$  to obtain **7a–f** as light yellow crystals.

#### Synthesis of 3-hydroxy-2-phenyl-1H-inden-1-one (8)

Prepared in a similar procedure as described for **2** (68.4% yield).

#### Synthesis of 4-benzylphthalazin-1-ol (9)

Prepared in a similar procedure as described for **3** (73.6% yield).

#### Synthesis of 1-chloro-4-benzylphthalazine (10)

Prepared in a similar procedure as described for **4** (99% yield).

#### Synthesis of 1-(piperazin-1-yl)-4-benzylphthalazine (11)

Prepared in a similar procedure as described for **5** (61% yield).

#### General procedure for preparation of compound (12a–f)

Prepared in a similar procedure as described for **7a–f**.

#### Synthesis of Ethyl 3,4-dimethoxybenzoate (13)

A mixture of ethyl 3,4-dimethoxybenzoate (50g, 0.28 mol),  $\text{K}_2\text{CO}_3$  (110 g, 0.80 mol), and acetone (300 mL) was stirred at room temperature for 0.5 h. And then  $(\text{CH}_3\text{O})_2\text{SO}_2$  (57 mL, 0.60 mol) was added drop-wise to the reaction. The reaction mixture was heated at 50°C for 5 h and concentrated under vacuum. The residue was poured into a 10% hydrochloric acid aqueous solution (1000 mL), stirred intensively for 3 h, and then dissolved in EtOAc (700 mL). The resulting solution was washed with brine and water. The organic phase was dried ( $\text{Na}_2\text{SO}_4$ ) and evaporated to yield compound **13** as a light yellow oil (54.5 g, 94.5%).

#### Synthesis of 3,4-dimethoxybenzoic acid (14)

Compound **13** (154.5 g, 0.26 mol) was added to a stirred solution of NaOH (20.5 g, 0.513 mol) in water (150 mL) at 0–10°C. The mixture was heated at 70°C for 2 h. After cooling to room temperature, the resulting mixture was poured into water (100 mL). Then the mixture was acidified with concentrated hydrochloric acid to pH 2–3, stirred at room temperature for 1 h and separated by filtration to give **14** as a white crystal (43.5 g, 92.2%).

#### Synthesis of 5,6-dimethoxyisobenzofuran-1(3H)-one (15)

Hydrogen chloride gas was gradually passed into the 37% formaldehyde solution (250 mL, 3.31 mol) for 0.5 h at room temperature, and then compound **14** (33.5 g, 0.184 mol) was added to the reaction. The mixture was heated to 65°C for 14 h and concentrated under vacuum. The residue was poured into water

(250 mL) and alkalized with a 10% sodium hydroxide aqueous solution, stirred for 1 h at room temperature, and separated by filtration to give **15** as a light purple powder, which was recrystallized from ethanol to obtain a white crystal (23 g, 50%).

#### *Synthesis of 3-hydroxy-5,6-dimethoxy-2-phenyl-1H-inden-1-one (16)*

Prepared in a similar procedure as described for **2** (73.9% yield).

#### *Synthesis of 4-benzyl-6,7-dimethoxyphthalazin-1(2H)-one (17)*

Prepared in a similar procedure as described for **3** (64.6% yield).

#### *Synthesis of 1-chloro-4-benzyl-6,7-dimethoxyphthalazine (18)*

Prepared in a similar procedure as described for **4** (99% yield).

#### *Synthesis of 1-(piperazin-1-yl)-4-benzyl-6,7-dimethoxyphthalazine (19)*

Prepared in a similar procedure as described for **5** (85.7% yield).

#### *General procedure for preparation of compound (20a–f)*

Prepared in a similar procedure as described for **7a–f**.

#### **Evaluation of biological activity**

The cytotoxic activities of compounds **7a–f**, **12a–f** and **20a–f** were evaluated with A549, HT-29 and MDA-MB-231 cell lines by the standard MTT assay *in vitro* [12], with vatalanib as the positive control. The cancer cell lines were cultured in minimum essential medium (MEM) supplement with 10% fetal bovine serum (FBS).

Approximately  $4 \times 10^3$  cells, suspended in MEM medium, were plated onto each well of a 96-well plate and incubated in 5% CO<sub>2</sub> at 37°C for 24 h. The test compounds at indicated final concentrations were added to the culture medium and the cell cultures were continued for 72 h. Fresh MTT was added to each well at a terminal concentration of 5 µg/mL and incubated with cells at 37°C for 4 h. The formazan crystals were dissolved in 100 µL DMSO each well, and the absorbency at 492 nm (for

absorbance of MTT formazan) and 630 nm (for the reference wavelength) was measured with the ELISA reader. All of the compounds were tested twice in each of the cell lines. The results expressed as IC<sub>50</sub> (inhibitory concentration 50%) were the averages of two determinations and calculated using the Bacus Laboratories Incorporated Slide Scanner (Bliss) software.

*This work was supported by a grant from the Doctoral startup foundation of Liaoning province of China (No. 20101110).*

*The authors have declared no conflict of interest.*

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