



Contents lists available at ScienceDirect

## Spectrochimica Acta Part A: Molecular and Biomolecular Spectroscopy

journal homepage: [www.elsevier.com/locate/saa](http://www.elsevier.com/locate/saa)

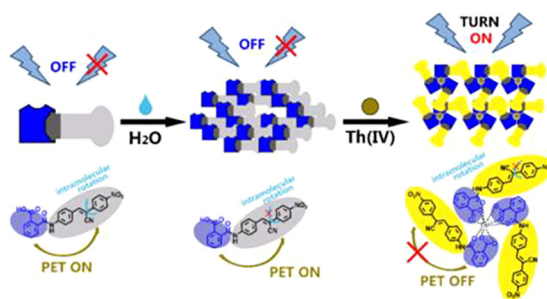
# “Turn-on” fluorescent sensor for Th<sup>4+</sup> in aqueous media based on a combination of PET-AIE effect

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## HIGHLIGHTS

- The first turn-on fluorescence probe for Th<sup>4+</sup> in aqueous media was reported.
- The detection limit for Th<sup>4+</sup> was as low as 0.074 μM.
- It was applied to detect Th<sup>4+</sup> on TLC plates and in real water samples.
- It displayed excellent living-cell imaging property for tracing Th<sup>4+</sup>.
- The detecting mechanism of AIE-PET effect was confirmed.

## GRAPHICAL ABSTRACT

The first “turn-on” fluorescent sensor for Th<sup>4+</sup> in aqueous media was achieved.

## ARTICLE INFO

## Article history:

Received 5 September 2020

Received in revised form 17 October 2020

Accepted 3 November 2020

Available online xxx

## Keywords:

Turn-on

Th<sup>4+</sup>

Sensor

Fluorescence

Aqueous media

## ABSTRACT

Previously reported fluorescent sensors for Th<sup>4+</sup> experienced emission quenching or generated false positive signal upon aggregate formation in aqueous media. Herein, a simple and novel thorium sensor (**CDB-BA**) based on cyanodistyrene structure was designed and synthesized, which integrated the highly emitting characteristic of AIE effect and off-on response of PET modulation for the first time to construct the “turn-on” fluorescent probe for Th<sup>4+</sup>. Besides excellent selectivity, **CDB-BA** exhibited remarkable fluorescent enhancement which was linearly related to the concentration of Th<sup>4+</sup> in the range of 0.25–8 μM. The detection limit was attained 0.074 μM, which was lower than that of most previously reported sensors. The mechanism of tris-chelate complex of **CDB-BA** with Th<sup>4+</sup> was confirmed by mass spectra, IR spectra and DFT calculation. The excellent Th<sup>4+</sup> sensing ability of **CDB-BA** was successfully applied to detecting Th<sup>4+</sup> on TLC plates, in real water samples and living-cell imaging. This work suggested that the combination of AIE and PET photophysical mechanism could offer the merits of minimized background and enhanced signal fidelity to develop novel “turn-on” fluorescent probe in complicated aqueous environment and biological research.

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## 1. Introduction

Thorium, a long-lived radionuclide, is very hazardous to human health and environment due to both its radiological and chemical toxicity [1–5]. Used as a X-ray contrast agent in the 1930s–1950s, ThO<sub>2</sub> was initially believed to have no side effects but was

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