A selective synthesis of 4-bromoisoquinoline and 4-bromoisquinolone Hong-Ping Zhang*, Hong-Yan Li and Hong-Fang Xiao

Department of Biology and Chemistry Engineering, Shaoyang University, Shaoyang 422000, P. R. China

2-Alkynyl benzyl azides smoothly underwent an electrocyclic reaction catalysed by palladium to selectively afford either 4-bromoisoquinoline and 4-bromoisquinolones under different conditions. 4-Bromoisoquinoline was synthesised in the presence of PdBr₂/CuBr₂/LiBr in MeCN, and 4-bromoisoquinolone was selectively produced with PdBr₂/CuBr₂/HOAc in CH₂CICH₂CI. A bromine was introduced into the products which makes the methodology more attractive for organic synthesis.

Keywords: 2-alkynyl benzyl azides, selectivity, cyclisation, isoquinoline, isoquinolone

Isoquinolines¹⁻³ and isoquinolin-1(2*H*)-ones⁴⁻⁷ are found in many natural products and synthetic pharmaceuticals and a number of efficient and selective methods have been developed for their synthesis.⁸⁻¹³ Although the classical routes to isoquinolines involve the Bischler-Napieralski¹⁴ or Pictet-Spengler reaction,¹⁵ these approaches often have drawbacks, for example, the use of strong acids and elevated temperature,¹⁶ which are not suitable for sensitive substrates. This has led to the development of new methodology. In recent years, new transition metal catalysed reactions have been developed for the synthesis of substituted isoquinolines from phenylacetylene substrates.¹⁷⁻²⁴ These reactions have proved to be extremely efficient in the synthesis of a wide variety of 3,4-substituted isoquinolines.

There are also a number of approaches to the synthesis of isoquinolin-1(2*H*)-ones reported in the literature,^{25–29} but most of them suffer from a poor precursor scope and less structural diversity. Recently, the transition metal catalysed synthesis of substituted isoquinolin-1(2*H*)-one derivatives has received attention, for example, Pd-catalysed multicomponent synthesis of oxazoline and benzoxazole,³⁰ Pd-catalysed synthesis of 4-aminophthalazin-1(2*H*)-one,³¹ and synthesis of quinazolino-[3,2-a]-quinazolines via a palladium-catalysed three-component reaction.³²

We recently reported the first successful application of the halopalladation cyclisation of alkynes with azides to the synthesis of 4-haloisoquinolines and 3-haloindoles.³³ As part of our continuing studies on the efficient preparation of biologically important heterocycles,³⁴ we describe herein strategies for the synthesis of 4-bromoisoquinolines and 4-bromoisoquinolin-1(2*H*)-one via a reagent-based cyclisation of 2-alkynyl benzyl azides (Scheme 1). To the best of our knowledge, there are no reports of an efficient synthesis of 4-bromoisoquinolines and 4-bromoisoquinolines and 4-bromoisoquinolines in the efficient synthesis of 4-bromoisoquinolines and 4-bromoisoquines and 4-bromoisoquines and 4-bromoisoquines and 4-bromoisoquines

In our earlier studies with 2-phenylethynyl benzyl azide (1a), we have devised route for the preparation of bromoisoquinolines in the presence of 5 mol% of PdBr₂, 3 equiv. of CuBr₂, and 2 equiv. of LiBr as additive in MeCN.³³ As shown in Table 1, exploratory experiments revealed that treatment of 2-phenylethynyl benzyl azide with PdBr₂/CuBr₂/LiBr in CH₂ClCH₂Cl was unsuccessful for the bromopalladation annulation of 1a to afford product 4-bromo-3-phenylisoquinoline (2a). However, unexpectedly 4-bromo-3-phenylisoquinolin-1(2H)-one (3a) was isolated (entry 2). During the course of our investigation, we observed that an identical yield of 3a was obtained by further decreasing the additive loadings of LiBr to 0.2 equiv, coupled with a decrease in the yield of 2a (entries 3 and 4). Encouraged by these results, we decided to focus on the development of appropriate protocols for reagent-based synthesis of 4-bromoisoquinolines and 4-bromoisoquinolin-1(2H)-one. The effect of additive on the reaction was examined, employing 5 mol% of PdBr₂ and 3 equiv. of CuBr₂ in CH₂ClCH₂Cl (entries 5–8). After a series of trials, 2 equiv. of LiBr as additive and MeCN as solvent provided the highest yield of 4-bromo-3-phenylisoquinoline (2a) (method A), and 2 equiv. of HOAc as additive and a mixture CH2ClCH2Cl/ H₂O(50:1) as solvent gave the highest yield of 4-bromo-3-phenylisoquinolin-1(2*H*)-one (**3a**) (method B) (entries 1 and 7).

As shown in Table 2, the scope of 2-alkynylbenzyl azides 2 for the halopalladation cyclisation reaction was explored under the standard reaction conditions. In the case of the method

Table 1 Annulation of 2-phenylethynyl benzyl azide (1a) in presence of PdBr_2^a

۲ ۱a	N_3 $PdX_2, CuX_2,$ additive	Br 2a	- ^{+ (}	X 3a	NH
Entry	Additive/equiv.	Solvent	Time/h	Yield/% ^b	
				2a	3a
1	LiBr (2)	MeCN	26	65	Trace
2	LiBr (2)	CICH ₂ CH ₂ CI	26	31	18
3	LiBr (1)	CICH ₂ CH ₂ CI	26	20	25
4	LiBr (0.2)	CICH ₂ CH ₂ CI	26	12	27
5	HOAc (1)	CICH ₂ CH ₂ CI	22	Trace	40
6	HOAc (2)	CICH ₂ CH ₂ CI	22	Trace	70
7	HOAc (2)/H ₂ O (0.1 mL)	CICH ₂ CH ₂ CI	22	Trace	83
8	HOAc (2)/H ₂ O (0.5 mL)	CICH ₂ CH ₂ CI	22	Trace	68

^aReaction conditions: **1a** (0.3 mmol), PdBr₂ (5 mol%), CuBr₂ (3 equiv.) and additive in the indicated solvent (5 mL) at 80 °C. ^bIsolated yield.



Scheme 1 Synthesis of bromoisoquinoline and bromoisoquinolone.

* Correspondent. E-mail: zhp4901@sina.com

A-based cyclisation, substrates **1b**, **1c** and **1h–j** successfully gave the corresponding bromoisoquinolines. Azides with an electron-donating group (compounds **1b** and **1c**) were subjected to method A, and gave 3-substituted 4-bromoisoquinolines **2b** and **2c** in good yields (entries 1 and 2). The aliphatic alkynyl benzyl azides **1h–j** provided good yields of 3-substituted 4-bromo-1,2-dihydroisoquinoline **4h–j** under the same conditions (entries 7–9). For example, under the reaction condition of 5 mol% of PdBr₂, 3 equiv. of CuBr₂, and 2 equiv. of LiBr, a 78% yield of 4-bromo-3-(4-methoxyphenyl)isoquinoli ne (**2b**) was obtained, and a 68% yield of 4-bromo-3-octyl-1,2dihydroisoquinoline (**4h**) was obtained (entries 2 and 7).

In the presence of 5 mol% of PdBr₂, 3 equiv. of CuBr₂, and 2 equiv. of HOAc, palladium-catalysed annulations of 2alkynylbenzyl azides (1b-g) provided moderate to good yields of the corresponding 3-substituted 4-bromoisoquinolin-1(2H)one (3b-g). As shown in Table 2, substituents affected the selectivity of reactions. The results showed that electronwithdrawing groups in the aromatic ring connecting with alkyne favored the desired reaction, but an electron-donating group suppressed it (entries 1-3). For example, the substrate 1d bearing a nitro group gave 81% yield of the target product 3d exclusively in the presence of PdBr₂ and CuBr₂, whereas the substrate 1c bearing a methoxyl group provided the desired product 3c in 42% yield (entries 1 and 3). The 2-heteroarylethynylbenzyl azides (1e) successfully underwent the cyclisation to afford the dibromo-addition product 3e in moderate yield (entry 4). Substrates with the aromatic ring bearing a substituted group 1f and 1g were compatible with annulations, provided products 3f and 3g in 57% and 59% yield, respectively. (entries 5 and 6). However, when the aliphatic alkynyl benzyl azides (1h-j) were subjected to method B the expected products were not obtained (entries 7-9).

In summary, an efficient selective synthesis method has been developed for the synthesis of 4-bromoisoquinolines and 4-bromoisoquinolin-1(2*H*)-ones. In the presence of PdBr₂, CuBr₂, and additive, a variety of 2-alkynyl benzyl azides smoothly underwent the halopalladation cyclisation to afford the corresponding 4-bromoisoquinolines and 4-bromoisoquinolin-1(2*H*)-ones in moderate to good yields. Most importantly, these isoquinolines and isoquinolones with a halo group would be useful introduction of various functional groups, which should be significant for drug discovery.

Experimental

CAUTION: Many low-molecular-weight azides are explosive. In this lab, no problems have been encountered, but great caution should be exercised when heating compounds of this type, especially neat. The reactions described here were run on only a few grams; an increase in the scale of these reactions will decrease the efficiency of heat dissipation and explosions may result.

NMR spectra were obtained with an INOVA-400 (Varian) and AMX-400 (Bruker) spectrometer, operating at 400 MHz (¹H NMR) and 100 MHz (¹³C NMR). TMS was used an internal standard and CDCl₃ was used as the solvent. Mass spectrometric analysis was performed on a GC-MS instrument (Shimadzu GCMS-QP2010 plus). Melting points were recorded on an Electrothermal type 9100 melting point apparatus. All melting points are uncorrected.

Synthesis of 3-substituted 4-bromoisoquinolines (**2**) and 3-substituted 4-bromo-1,2-dihydroisoquinoline (**4**); typical procedure

A mixture of the 2-alkynyl benzyl azide 1 (0.3 mmol), PdBr₂ (5 mol %), CuBr₂ (3 equiv.), LiBr (2 equiv.) and CH₃CN (5 mL) was stirred at the indicated reaction temperature until consumption of the starting material was complete as judged by TLC and GC analysis. Then the mixture was filtered and evaporated, the residue was purified by flash column chromatography to afford 2 and 4 (hexane/ethyl acetate).

Synthesis of 3-substituted 4-bromoisoquinoline 1(2H)-ones (3); *typical experimental procedure*

A mixture of the 2-alkynyl benzyl azides 1 (0.3 mmol), PdBr₂ (5 mol %), CuBr₂ (3 equiv.), HOAc (2 equiv.), H₂O (0.1 mL) and ClCH₂CH₂Cl (5 mL) was stirred at the indicated reaction temperature until consumption of the starting material was complete as judged by TLC and GC analysis. Then the mixture was filtered and evaporated, the residue was purified by flash column chromatography to afford **3** (hexane/ethyl acetate).

4-Bromo-3-phenylisoquinoline (**2a**):^{35,36} Colourless oil; ¹H NMR (500 MHz, CDCl₃) δ 9.23 (s, 1H), 8.33 (d, J = 8.5 Hz, 1H), 8.01 (d, J = 8.0 Hz, 1H), 7.84 (t, J = 8.0 Hz, 1H), 7.73 (d, J = 8.5 Hz, 2H), 7.68 (t, J = 8.0 Hz, 1H), 7.50 (t, J = 7.8 Hz, 2H), 7.45 (t, J = 8.0 Hz, 1H); ¹³C NMR (125 MHz, CDCl₃) δ 152.3, 151.1, 140.7, 136.0, 131.9, 129.9, 128.6, 128.3, 127.9, 127.7, 127.0, 118.3; LRMS (EI, 70 eV) m/z (%): 285 (M⁺+2, 23), 283 (M⁺, 25), 204 (-Br, 100).

4-Bromo-3-o-tolylisoquinoline (**2b**): Orange solid, m.p. 110.2–111.6 °C (uncorrected); ¹H NMR (500 MHz, CDCl₃) δ 9.24 (s, 1H), 8.30 (d, J = 8.5 Hz, 1H), 8.03 (d, J = 8.5 Hz, 1H), 7.85 (t, J = 8.3 Hz, 1H), 7.70 (t, J = 8.0 Hz, 1H), 7.38–7.30 (m, 4H), 2.16 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 153.3, 151.0, 140.7, 135.9, 135.6, 131.8, 130.1, 129.0, 128.6, 128.4, 127.9, 127.8, 126.6, 125.6, 119.5, 19.5; LRMS (EI, 70 eV) m/z (%): 299 (M⁺+2, 8), 297 (M⁺, 8), 218 (-Br, 100); HRMS (EI) for C₁₆H₁₂⁷⁹BrN (M⁺): calcd 297.0153, found 297.0150.



 Table 2
 Effect of 2-alkynylbenzyl azides on the reaction selectivity^a

^aReaction conditions of method A: **1** (0.3 mmol), PdBr₂ (5 mol%), CuBr₂ (3 equiv.) and LiBr (2 equiv.) in MeCN (5 mL) 80 °C. Reaction conditions of method B: **1** (0.3 mmol), PdBr₂ (5 mol%) and HOAc (2 equiv.) in CICH₂CH₂Cl (5 mL) and H₂O (0.1 mL) at 80 °C. ^bCalculated by using LC/MS. ^cIsolated yield. ^dAll of the new compounds were characterised by ¹H NMR and ¹³C NMR spectra as given in the Electronic Supplementary Information. ^e**3e** is 4-bromo-3-(5-bromothiophen-2-yl)isoquinolin-1(2H)-one.

4-Bromo-3-(4-methoxyphenyl)isoquinoline (2c): Orange solid, m.p. 109.8-110.9 °C (uncorrected); ¹H NMR (500 MHz, CDCl₃) δ 9.21 (s, 1H), 8.32 (d, J = 9.0 Hz, 1H), 7.98 (d, J = 3.0 Hz, 1H), 7.82 (t, J = 8.3 Hz, 1H), 7.72 (d, J = 8.5 Hz, 2H), 7.66 (t, J = 8.0 Hz, 1H), 7.03 (d, *J* = 9.5 Hz, 2H), 3.88 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 159.7, 152.0, 151.0, 136.1, 133.1, 131.8, 131.4, 128.4, 127.7, 127.6, 126.9, 118.0, 113.4, 55.3; LRMS (EI, 70 eV) m/z (%): 315 (M+2, 45), 313 (M⁺, 47), 234 (-Br, 62), 219 (41); HRMS (EI) for C₁₆H₁₂⁷⁹BrNO (M⁺): calcd 313.0102, found 313.0100.

4-Bromo-3-phenylisoquinolin-1(2H)-one (3a):37 White solid, m.p. 158.4–158.6 °C (uncorrected); ¹H NMR (500 MHz, CDCl₃) δ 7.91 (s, 1H), 7.83 (d, J = 7.5 Hz, 1H), 7.54–7.48 (m, 5H), 7.42 (t, J = 7.5 Hz, 1H), 7.27 (t, J = 7.5 Hz, 1H), 6.64 (d, J = 8.0 Hz, 1H); ¹³C NMR (125 MHz, CDCl₃) δ 167.1, 137.5, 135.2, 133.5, 132.2, 131.1, 130.2, 130.1, 129.3, 129.1, 123.9, 122.8, 103.6; LRMS (EI, 70 eV) m/z (%): 301 (M++2, 43.8), 299 (M+, 49.1), 220 (-Br, 100.0)

4-Bromo-3-o-tolylisoquinolin-1(2H)-one (3b): White solid, m.p. 172.4–172.6 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.86 (s, 1H), 7.83 (d, J = 7.6 Hz, 1H), 7.44–7.40 (m, 2H), 7.36–7.30 (m, 3H), 7.27–7.24 (m, 2H), 6.27 (d, J = 8.0 Hz, 1H), 2.30 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 167.2, 137.8, 136.5, 135.2, 133.9, 132.6, 131.0, 130,4, 130.2, 129.3, 128.5, 126.9, 123.9, 122.4, 102.8; LRMS (EI, 70 eV) *m*/*z* (%): 315 (M⁺+2, 13.1), 313 (M⁺, 15.8), 234 (-Br, 37.3), 216 (21.5). HRMS (EI) for C₁₆H₁₂⁷⁹BrNO (M⁺): calcd 313.0102, found 313.0100.

4-Bromo-3-(4-methoxyphenyl)isoquinolin-1(2H)-one (3c): White solid, m.p. 168.5-168.9 °C; 1H NMR (400 MHz, CDCl₃) & 7.80 (s, 1H), 7.83 (d, J = 7.6 Hz, 1H), 7.45–7.41 (m, 3H), 7.30 (t, J = 7.4 Hz, 1H), 6.70 (d, *J* = 8.4 Hz, 2H), 6.74 (d, *J* = 8.0 Hz, 1H), 3.90 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 167.1, 135.3, 134.3, 133.3, 132.2, 131.7, 131.1, 129.7, 129.2, 123.8, 122.8, 114.6, 104.3, 55.5; LRMS (EI, 70 eV) m/z (%): 331 (M++2, 46.9), 329 (M+, 40.9), 250 (-Br, 100). HRMS (EI) for C₁₆H₁₂⁷⁹BrNO₂ (M⁺): calcd 329.0051, found 329.0050.

4-Bromo-3-(4-nitrophenyl)isoquinolin-1(2H)-one (3d): Brown solid, m.p. 252.1-252.6 °C; ¹H NMR (500 MHz, CDCl₃) δ 8.36 (d, J = 11.0 Hz, 2H), 7.92 (s, 1H), 7.86 (d, J = 7.5 Hz, 1H), 7.73 (d, J = 10.5 Hz, 2H), 7.51–7.47 (m, 1H), 7.55–7.29 (m, 1H), 6.69 (d, J = 8.0 Hz, 1H); ¹³C NMR (125 MHz, CDCl₃) δ 166.8, 144.0, 134.9, 134.5, 132.6, 131.6, 131.1, 130.1, 130.0, 124.4, 124.1, 122.5, 99.4; LRMS (EI, 70 eV) m/z (%): 346 (M++2, 91.0), 344 (M+, 93.9), 316 (15.9), 236 (12.1), 219 (100). HRMS (EI) for C₁₅H₉⁷⁹BrN₂O₃ (M⁺): calcd 343.9797, found 343.9795.

4-Bromo-3-(5-bromothiophen-2-yl)isoquinolin-1(2H)-one (3e): Brown solid, m.p. 161.5–161.9 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.92 (d, J = 7.5 Hz, 1H), 7.89 (s, 1H), 7.84 (d, J = 7.5 Hz, 1H), 7.49 (t, J = 7.5 Hz, 1H), 7.42 (t, J = 8.0 Hz, 1H), 7.09 (d, J = 3.5 Hz, 1H), 7.01 (t, J = 5.75 Hz, 1H); ¹³C NMR (125 MHz, CDCl₃) δ 166.9, 140.6, 136.6, 134.7, 132.7, 130.9, 130.7, 130.5, 130.3, 130.0, 124.1, 123.0, 116.3; LRMS (EI, 70 eV) m/z (%): 387 (M++4, 8.2), 385 (M++2, 15.5), 383 (M⁺, 10.2), 306 (-Br, 11.6), 304 (-Br, 11.3), 225 (-2Br, 41.3). HRMS (EI) for C₁₃H₇⁷⁹Br₂NOS (M⁺): calcd 382.8615, found 382.8613.

4-Bromo-7-methoxy-3-phenylisoquinolin-1(2H)-one (3f): White solid, m.p. 168.4–168.8 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.90 (s, 1H), 7.58–7.51 (m, 5H), 7.33 (t, J = 2.0 Hz, 1H), 6.86 (d, J = 10.0 Hz, 1H), 6.70 (d, J = 9.0 Hz, 1H), 3.86 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 166.9, 160.8, 136.0, 132.4, 131.3, 130.2, 129.9, 129.1, 127.9, 123.9, 120.4, 111.7, 106.4, 55.7; LRMS (EI, 70 eV) m/z (%): 331 (M⁺+2, 11.8), 329 (M⁺, 11.1), 250 (-Br, 8.7). HRMS (EI) for $C_{16}H_{12}^{79}BrNO_2$ (M⁺): calcd 329.0051, found 329.0050.

4-Bromo-6,7-dimethoxy-3-phenylisoquinolin-1(2H)-one (**3g**): White solid, m.p. 171.4–171.9 °C; ¹H NMR (400 MHz, CDCl₃) δ 7.73 (s, 1H), 7.55–7.45 (m, 5H), 7.23 (s, 1H), 6.01 (s, 1H), 3.91 (s, 3H), 3.75 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 167.5, 152.5, 150.7, 137.7, 133.7, 130.4, 129.7, 129.2, 129.1, 124.0,104.9, 101.7, 77.3, 56.3, 55.6; LRMS (EI, 70eV) m/z (%): 361 (M++2, 14.8), 359 (M+, 13.1), 280 (-Br, 9.9). HRMS (EI) for C₁₇H₁₄⁷⁹BrNO₃ (M⁺): calcd 359.0157, found 359.0153.

4-Bromo-3-octyl-1,2-dihydroisoquinoline (4h): Colourless oil; ¹H NMR (500 MHz, CDCl₃) δ 7.41 (t, *J* = 8.75 Hz, 1H), 7.38–7.35 (m, 2H), 7.22 (t, *J* = 4.5 Hz, 1H), 4.42–4.34 (m, 2H), 2.90–2.82 (m, 2H), 1.72–1.68 (m, 2H), 1.45–1.31 (m, 10H), 0.90 (t, *J* = 7.0 Hz, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 140.1, 133.0, 129.4, 129.3, 128.7, 128.6, 126.5, 113.6, 52.0, 40.3, 31.8, 29.3, 29.2, 28.6, 27.5, 22.6, 14.1; LRMS (EI, 70 eV) m/z (%): 321 (M⁺+2, 2), 319 (M⁺, 2), 237 (2), 236 (11), 235 (3), 223 (97), 220 (100); HRMS (EI) for $C_{17}H_{24}^{79}BrN$ (M⁺): calcd 321.1092, found 321.1089.

4-Bromo-3-pentyl-1,2-dihydroisoquinoline (4i): Colourless oil; 1H NMR (500 MHz, CDCl₃) δ 7.43–7.34 (m, 3H), 7.22 (t, J = 4.5 Hz, 1H), 4.43-4.34 (m, 2H), 2.91-2.79 (m, 2H), 1.75-1.69 (m, 2H), 1.45-1.39 (m, 4H), 0.96 (t, J = 7.0 Hz, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 140.1, 133.0, 129.4, 129.3, 128.8, 128.6, 126.5, 113.6, 52.0, 40.3, 30.7, 27.2, 22.4, 14.0; LRMS (EI, 70 eV) m/z (%): 281 (M+2, 2), 279 (M⁺, 2), 223 (57), 221 (59), 199 (16), 198 (100), 197 (15); HRMS (EI) for C14H18BrN (M+): calcd 279.0623, found 279.0625

4-Bromo-3-cyclohexenyl-1,2-dihydroisoquinoline (4j): Pale yellow oil; ¹H NMR (500 MHz, CDCl₃) δ 7.44–7.36 (m, 3H), 7.28–7.26 (m, 1H), 6.06 (t, *J* = 7.5 Hz, 1H), 4.48–4.39 (m, 2H), 2.29–2.26 (m, 2H), 2.20-2.16 (m, 2H), 1.79-1.74 (m, 2H), 1.68-1.65 (m, 2H); ¹³C NMR (125 MHz, CDCl₃) δ 139.8, 137.1, 133.0, 131.8, 131.2, 129.3, 128.8, 1127.9, 124.6, 112.6, 52.0, 26.4, 25.2, 22.3, 21.6; LRMS (EI, 70 eV) m/z (%): 291 (M+2, 26), 289 (M+, 26), 260 (-Br, 11), 259 (29), 258 (10); HRMS (EI) for C15H1679BrN (M+): calcd 289.0466, found 289.0462.

Electronic Supplementary Information

¹H NMR and ¹³C NMR spectra of the new compounds have been deposited in the ESI available through stl.publisher. ingentaconnect.com/content/stl/jcr/supp-data.

The authors thank the Scientific and Technological Innovative team of Shaoyang University (2012) for financial support.

Received 2 May 2013; accepted 24 June 2013 Paper 1301924 doi: 10.3184/174751913X13744287541024 Published online: 6 September 2013

References

- 1 K.W. Bentley, Nat. Prod. Rep., 2005, 22, 249.
- 2
- K.W. Bentley, *Nat. Prod. Rep.*, 2006, **23**, 444. M.C. Delcey, A. Croisy, D. Carrez, C.Huel, A. Chiaroni, P.Ducrot, E. Bisagni, L. Jin and G. Leclercq, *Bioorg. Med. Chem.*, 2000, **8**, 2629. 3
- M. Jayaraman, B.M. Fox, M. Hollingshead, G. Kohlhagen, Y. Pommier and M. Cushman, J. Med. Chem., 2000, 43, 3688.
- 5 H. Zhang, D. Zembower and Z. Chen, Bioorg. Med. Chem. Lett., 1997, 7,
- 2687. V.S. Bernan, D.A. Montenegro, J.D. Korshalla, W.M. Maiese, D.A. 6 Steinberg and M.J. Greenstein, Antibiotics, 1994, 47, 1417. S.W. Li, M.G. Nair, D.M. Edwards, R.L. Kisluick, Y. Gaument, I.K. Dev,
- 7 D.S. Duch, J. Humphreys, G.K. Smith and R. Ferone, J. Med. Chem., 1991, 34, 2746.
- Q. Huang, J.A. Hunter and R.C. Larock, Org. Lett., 2001, 3, 2973
- 9 R.P. Korivi, W.-J. Wu and C.-H. Cheng, *Chem., Eur. J.*, 2009, 15, 10727.
 10 R.P. Korivi, W.-J. Wu and C.-H. Cheng, *Chem., Eur. J.*, 2010, 16, 282.
- 11 R.K. Chinnagolla, S. Pimparkar and M. Jeganmohan, Org. Lett., 2012, 14, 3032.
- 12 P.G. Jagtap, E. Baloglu, G. Southan, W. Williams, A. Roy, A. Nivorozhkin, N. Landrau, K. Desisto, A.L. Salzman and C. Szabo, Org. Lett., 2005, 7, 1753
- 13 L.E. Fisher, J.M. Muchowski and R.D. Clark, J. Org. Chem., 1992, 57, 2700
- A. Bischler and B.Napieralski, Ber. Dtsch. Chem. Ges., 1893, 26, 1903. 14
- A. Pictet and T. Spengler, Ber. Dtsch. Chem. Ges., 1911, 44, 2030. 15
- 16 F.W. Bergstrom, Chem. Rev., 1944, 35, 77
- 17 R.P. Korivi and C.-H. Cheng, Org. Lett., 2005, 7, 5179.
- G.B. Bajracharya, N.K. Pahadi, I.D. Gridnev and Y. Yamamoto, *J. Org. Chem.*, 2006, **71**, 6204.
 Y.-N. Niu, Z.-Y. Yan, G.-L. Gao, H.-L. Wang, X.-Z. Shu, K.-G. Ji and Y.-M.
- Liang, J. Org. Chem. 2009, **74**, 2893. S.-G. Lim, J.H. Lee, C.W. Moon, J.-B. Hong and C.-H. Jun, *Org. Lett.*,
- 20 2003, 5, 2759.
- N. Guimond and K. Fagnou, J. Am. Chem. Soc., 2009, 131, 12050.
 B. Wang, B. Lu, Y. Jiang, Y. Zhang and D. Ma, Org. Lett., 2008, 10, 2761.
 Z. Huo and Y. Yamamoto, Tetrahedron Lett., 2009, 50, 3651.
- 24 T.V.V. Ramakrishna and P.R. Sharp, Org. Lett., 2003, 5, 877 25 L. Pellegatti, E. Vedrenne, M.-A. Hiebel, F. Buron, S. Massip, J.-M. Leger,
- C. Jarry and S. Routier, Tetrahedron Lett., 2011, 52, 5224. S.K. Guchhait and C. Madaan, Org. Biomol. Chem., 2010, 8, 3631 26
- F. Mert-Balci, J. Conrad, K. Meindl, T. Schulz, D. Stalke and U. Beifuss, Synthesis, 2008, 22, 3649.
- M.I. Antczak and J.M. Ready, Chem. Sci., 2012, 3, 1450.
- V. Tyagi, S. Khan, A. Giri, H.M. Gauniyal, B. Sridhar and P.M.S. Chauhan, 29
- Y. Fydg, S. Khai, Y.A. Ola, H.M. Guanyar, D. Shahat and T.W.S. Channah, *Org. Lett.*, 2012, 14, 3126.
 P.J. Boissarie, Z.E. Hamilton, S. Lang, J.A. Murphy and C.J. Suckling, *Org. Lett.*, 2011, 13, 6256.
 T. Vlaar, E. Ruijter, A. Znabet, E. Janssen, F.J.J. de Kanter, U.W. Bert, T. Vlaar, E. Ruijter, A. Znabet, E. Janssen, F.J.J. de Kanter, U.W. Bert,
- B.U. Maes and R.V.A. Orru, Org. Lett., 2011, 13, 6496.
- 32 G. Qiu, Y. He and J. Wu, Chem. Commun., 2012, 48, 3836.
- 33 H.-P. Zhang, S.-C. Yu, Y. Liang, P. Peng, B.-X.Tang and J.-H. Li, Synlett, 2011, 0982.
- D. Fischer, H. Tomeba, N.K. Pahadi, N.T. Patil and Y. Yamamoto, Angew. 35 Chem. Int. Ed., 2007, 46, 4764.
- X.Yu and J. Wu, J. Comb. Chem., 2009, 11, 895. 36
- 37 C.O. Usifoh, J. Heterocycl. Chem., 2001, 38, 597.

Copyright of Journal of Chemical Research is the property of Science Reviews 2000 Ltd. and its content may not be copied or emailed to multiple sites or posted to a listserv without the copyright holder's express written permission. However, users may print, download, or email articles for individual use.