Lewis-Acid-Catalyzed Reactions of Bis(4-alkoxyphenyl)methanol with (Diarylmethylene)- and (Dialkylmethylene)cyclopropanes

Liang-Feng Yao^[a] and Min Shi*^[a,b]

Keywords: Lewis acids / Synthetic methods / Cycloalkenes / Strained molecules

Bis(4-methoxyphenyl)methanol (2a) can be transformed by BF₃·OEt₂-catalyzed reactions of (arylmethylene)cyclopropanes 1 to the corresponding polysubstituted cyclopentenes 3 as the major products along with methylenecyclobutanes and dienes as minor products. The reaction conditions are mild, yields are moderate to good. In the reactions of aliphatic methylenecyclopropanes with bis(4-methoxyphenyl)methanol (2a), cyclopentenes 3o-q were produced in good yields under the standard conditions. Interesting results were

Introduction

Methylenecyclopropanes (MCPs) are highly strained but readily accessible molecules, which can serve as useful building blocks in organic synthesis.^[1] Their most interesting aspect is that MCPs can undergo a variety of ring-opening and cycloaddition reactions to give interesting products in the presence of transition metals or Lewis acids. Indeed, the relief of ring strain can provide a powerful thermodynamic driving force.^[2,3] Thus far, a number of interesting cycloadditions and ring enlargements of MCPs have been explored. For example, Yamamoto et al. reported cycloaddition reactions of MCPs with aldehydes and imines using a palladium catalyst, affording the corresponding tetrahydrofuran (THF) and pyrrolidine skeletons in good yields.^[4] In addition, we as well as other groups have developed a number of heterocycle-forming reactions starting from MCPs and aldehydes or imines as well as ring enlargements of MCPs in the presence of Lewis or Brønsted acids.^[5,6] Recently, we have been investigating the Lewis-acidmediated ring-opening reactions of MCPs 1 with a number of electrophiles such as phenylsulfenyl chloride and phenylselenyl chloride as well as 3-methoxy-1,3,3-triarylprop-1yne or 1,1,3-triarylprop-2-yn-1-ol under mild conditions.^[7] A variety of novel products could be obtained in good yields by a simple one-step reaction. Herein, we report an

obtained from the reactions of 1-[cyclopropylidene(4-methoxyphenyl)methyl]-4-methoxybenzene (1e) with bis(4-alkoxyphenyl)methanols 2: polysubstituted cyclopentene derivatives 5 were obtained in good yields under mild reaction conditions. Plausible reaction mechanisms based on the deuterium-labeling and control experiments are proposed.

(© Wiley-VCH Verlag GmbH & Co. KGaA, 69451 Weinheim, Germany, 2009)

interesting Lewis-acid-catalyzed reaction of MCPs 1 with bis(4-alkoxyphenyl)methanols 2 to produce cyclopentenes 3 as the major products along with methylenecyclobutanes and diene derivatives 4a-c as the minor products. We could also prepare novel polysubstituted cyclopentenes 5 in moderate to good yields under mild conditions.

Results and Discussion

We performed initial examinations with (diphenylmethylene)cyclopropane 1a (0.2 mmol) and bis(4-methoxyphenyl)methanol 2a (0.24 mmol) as substrates in the presence of various Lewis acids (10 mol-%) in 1,2-dichloroethane (DCE) to determine the best catalyst for this intermolecular reaction. The results of these experiments are summarized in Table 1. Except in THF or using Yb(OTf)₃ as the Lewis acid, the reactions proceeded smoothly with various Lewis acids in a variety of solvents at room temperature (20 °C), giving cyclopentene 3a and methylenecyclobutane 4aa along with ring-opened diene products 4ab and 4ac as (E/Z) mixtures in 63-84% total yields, respectively (Table 1, Entries 1–7 and 10–12). Using BF₃·OEt₂ as the catalyst afforded 3a in the highest yield, perhaps due to its mild catalytic ability in the above reaction compared with that of $M(OTf)_n$ (n = 3 or 4, Table 1, Entry 1). Further increasing or decreasing the amount of BF3. OEt2 provided no improvement in the total yield (Table 1, Entries 8 and 9). In view of the total yield of this reaction, we used 10 mol-% of BF₃·OEt₂ as our optimal catalyst loading. We obtained the best result by using BF₃·OEt₂ (10 mol-%) as the catalyst in DCE at room temperature (20 °C), which afforded 3a in 52% yield and a 72% total yield including 4aa-ac (Table 1, Entry 1).

 [[]a] Laboratory for Advanced Materials & Institute of Fine Chemicals, East China University of Science and Technology, 130 Mei Long Lu, Shanghai 200237, P. R. China

[[]b] State Key Laboratory of Organometallic Chemistry, Shanghai Institute of Organic Chemistry, Chinese Academy of Sciences, 354 Fenglin Road, Shanghai 200032, P. R. China E-mail: Mshi@mail.sioc.ac.cn

Supporting information for this article is available on the WWW under http://dx.doi.org/10.1002/ejoc.200900546.

Table 1. Optimization of the reaction conditions.

Ph Ph 1a	+ Ar Ar Ar Ar Ar Ar Ar Ar	10 mol-%) .t., 3 h Ar 3a	Ph Ph Ar Ph	4aa ^{Ar} 4aa ^{Ar} Ar 4ab
			Ph Ph	Ar Ar 4ac
Entry ^[a]	Lewis acid	Solvent	3a	% Yield ^[b]
1	BF ₃ ·OEt ₂	DCE	52	20 (78:13:9)
2	Nd(OTf) ₃	DCE	15	69 (18:68:14)
3	In(OTf) ₃	DCE	10	67 (20:62:18)
4	Zr(OTf) ₄	DCE	30	40 (22:63:15)
5	TfOH	DCE	48	20 (21:60:19)
6	Yb(OTf) ₃ ^[e]	DCE	trace	trace
7	Sc(OTf) ₃	DCE	15	51 (29:51:20)
8	BF ₃ ·OEt ₂ ^[f]	DCE	52	7 (81:7:12)
9	BF ₃ ·OEt ₂ ^[g]	DCE	54	14 (79:12:9)
10	BF ₃ ·OEt ₂	toluene	35	40 (75:12:13)
11	BF ₃ ·OEt ₂	DCM	49	14 (76:14:10)
12	BF ₃ ·OEt ₂	THF	-	_

[a] All reactions were carried out with 1a (0.2 mmol), 2a (0.24 mmol) and Lewis acid (10 mol-%) in various solvents (2.0 mL) at room temperature. [b] Isolated yields. [c] Total yields of 4aa–ac in which the (*E*) configuration of 4ab has been determined by NOE spectroscopy (see the Supporting Information). [d] The ratios of 4aa/4ab/4ac were determined by ¹H NMR spectroscopy. [e] The reaction mixtures were stirred for 24 h. [f] 20 mol-% of BF₃·OEt₂ was used. [g] 5 mol-% of BF₃·OEt₂ was used.

Next, we attempted to improve the yield of 3a by adjusting the ratios of 1a and 2a. The results of these experiments are shown in Table 2. We found that using 1.0 equiv. of 1a and 1.5 equiv. of 2a produced 3a in 45% yield and a

mixture of **4aa–ac** in 13% total yield, respectively (Table 2, Entry 1). Increasing the amount of **1a** to 2.0 equiv. or 2.5 equiv. to react with **2a** (1.0 equiv.) afforded **3a** in 67 and 73% yields along with a mixture of **4aa–ac** in 28% and 14%

Table 2. Further optimization of the reaction conditions.



[a] Isolated yields. [b] Total yield of 4aa-ac. [c] The ratios of 4aa/4ab/4ac were determined by ¹H NMR spectroscopy.

total yields, respectively (Table 2, Entries 2 and 3). Further increases of the amount of **1a** provided no improvement in the yield (Table 2, Entry 4).

With these optimized reaction conditions in hand, we next turned our interest to the screening of the substrates. The results of these experiments are summarized in Table 3. We obtained the corresponding cyclopentene derivatives 3 in 25-78% yields along with the minor products 4 in 8-32% total yields. These studies highlighted the fact that substituents on the aromatic rings of 1 did not significantly affect the reaction outcomes (Table 3, Entries 1–7). Various strongly electron-donating alkoxy groups on the aromatic rings of 2 had little influence on the yields of 3 (Table 3, Entries 9–13). MCP 1i-having an ortho-chloro atom on the aromatic ring-produced the corresponding cyclopentene 3i in only 25% yield, presumably due to steric hindrance. We note that the strongly electron-donating group on the aromatic ring of 2 is essential in the reaction because when using diphenylmethanol or bis(4-chlorophenyl)methanol as the reactant, no reaction occurred under identical conditions. Moreover, it is noteworthy that, using triphenylmethanol as the substrate to react with 1, no reaction occurred under identical conditions perhaps due to a steric effect. Furthermore, using monosubstituted MCPs to react with 2a under identical conditions, complex product mixtures

formed (Scheme 1). Thus, these studies outlined the fact that the reaction is determined by the substituted pattern of 1 and the electronic nature and the steric effect of 2.

$$R \longrightarrow P^{H} + Ar \longrightarrow P^{H} Ar \xrightarrow{BF_3 \cdot OEt_2 (10 \text{ mol-}\%)}{DCE, r.t.} \text{ complex}$$

$$Ar = p \cdot MeOC_6H_4$$

$$R = p \cdot BrC_6H_4, o \cdot Bn, m \cdot MeOC_6H_3$$

Scheme 1. Reaction of monosubstituted MCPs with 2a.

We determined product structures by ¹H and ¹³C NMR spectroscopy and HRMS or microanalyses. We determined the crystal structure of **3a** by X-ray diffraction (Figure 1), and its CIF data were deposited in a data base.^[9] Moreover, we determined the (*E*) configuration of **4ab** by NOE studies (Figure 2, see the details in the Supporting Information).^[8]

Furthermore, we observed an interesting phenomenon on the basis of ¹H NMR spectroscopic data; when it was stored at -18 °C for two weeks, the diene product **4ab** could be completely transformed into another diene **4aB** by allylic isomerization (Scheme 2). We unambiguously determined the structure of **4aB** by X-ray diffraction, and the ORTEP drawing is depicted in Figure 3.^[9]

Table 3. Reactions of MCPs 1 with 2, catalyzed by BF_3 ·OEt₂, under the optimized conditions.



[a] Isolated yields. [b] For unsymmetrical substrates, diastereomers in a 1:1 ratio were obtained on the basis of ¹H NMR spectroscopy. [c] The total yields of **4a–c**. [d] The ratios of **4a/4b/4c** were determined by ¹H NMR spectroscopy. [e] For unsymmetrical substrates, the E/Z or Z/E ratio was 1:1.



Figure 1. ORTEP drawing of 3a.



Figure 2. NOE studies on 4ab.



Scheme 2. Transformation of 4ab to 4aB.



Figure 3. ORTEP drawing of 4aB.



Scheme 3. Deuterium-labeling experiment of the reaction.

On the basis of deuterium-labeling experiment, we outlined a plausible reaction mechanism as shown in Scheme 4. In the presence of BF₃·OEt₂, 2a produces cationic intermediate A, which reacts with MCP 1 to afford cationic intermediate B. Intermediate B undergoes a ring expansion to give cationic intermediate C, which can proceed through two different reaction pathways. The deprotonation of C gives the minor product 4a (path a). A 1,2-aryl shift produces intermediate D.^[10] Intermediate D undergoes a 1,2alkyl shift to intermediate E, which undergoes deprotonation to produce the major products **3a-n** (paths b and c). As for the products 4b and 4c, cationic intermediate C undergoes a 1,2-aryl shift to furnish cationic intermediate D, which undergoes a ring-opening process to afford cationic intermediate F. The deprotonation of intermediate F produces the products 4b and 4c.



Scheme 4. A plausible reaction mechanism for the formation of 3 and 4.

Eurjoean Journal of Organic Chemistry

Table 4. The reactions of aliphatic MCPs 1j-m with 2a in the presence of BF₃·OEt₂.



[a] Isolated yields.

Furthermore, we found that in the case of aliphatic MCP 1j, product 3o formed in 52% yield rather than the desired cyclopentene derivative of 3a-n, presumably because the deprotonation in this particular case proceeded through path d (Table 4, Entry 1 and Scheme 4). Moreover, we also examined aliphatic MCPs 1k-m in this reaction, and the results of these experiments are summarized in Table 4. The reactions proceeded smoothly to deliver the corresponding cyclopentene derivatives 3p-r in good yields varying from 61-67% under the standard conditions, suggesting that the substituents on the aromatic rings have little influence on the reaction outcome.

Interestingly, we found that using 1.0 equiv. of MCP 1e, having a strongly electron-donating methoxy group on the benzene ring, to react with 1.2 equiv. of 2a afforded a new product 5a in 55% yield under the standard conditions (Table 5, Entry 1). We also unambiguously determined the structure of 5a by X-ray diffraction, and the ORTEP drawing is depicted in Figure 4.^[9] Next, we attempted to improve the yield of 5a by adjusting the ratios of 1e and 2a similarly. The results are shown in Table 5. When we used 1.0 equiv. of 1e and 2.0 equiv. of 2a, 5a formed in 59% yield. Increasing the amount of 2a to 2.5 equiv. and 3.0 equiv. improved the isolated yields of 5a to 63% and 68%, respectively (Table 5, Entries 3 and 4). Further increasing the amount of 2a (4.0 equiv.) furnished no improvement in the yield.

Thus, the optimized reaction conditions constitute a reaction in DCE at room temperature (20 °C) with 1.0 equiv. of **1e** and 3.0 equiv. of **2a** in the presence of BF₃·OEt₂ (10 mol-%) for 3 h. Under these optimized conditions, we next examined the generality of this reaction using a variety of starting materials **2**, and the results are outlined in Table 6. The corresponding cyclopentene derivatives **5** formed in 50–57% yields (Table 6, Entries 1–4). Substituents on the aromatic rings of **2** have little influence on the reaction outcome. Moreover, we also determined the structure of **5b** by X-ray diffraction, and the ORTEP drawing is depicted in Figure 5.^[9] More interestingly, on the basis of crystal data, the similar products **5a** and **5b** have different space groups ($P2_1/c$ and P-1, respectively).

Table 5. Optimization of the reaction conditions.

Ar Ar +	OH Ar Ar	BF ₃ ·OEt ₂ (10 mol-%) DCE, r.t., 3 h	Ar Ar Ar
1e , 1 equiv.	2a , <i>y</i> equiv.	$Ar = p-MeOC_6H_4$	5a
Entry		y 2a	% Yield of 5a ^[a]
1		1.2	55
2		2	59
3		2.5	63
4		3	68
5		4	54

[a] Isolated yields.



Figure 4. ORTEP drawing of 5a.

In order to clarify the reaction route for the formation of **5**, we performed two control experiments (Scheme 5). We found that the reaction of **4eb** and **4ec** [(E/Z) mixtures, 0.1 mmol] with **2a** (0.2 mmol) under standard conditions

Table 6. Reactions of 1e with 2b-f under the optimized reaction conditions.

1e	$Ar \rightarrow Ar \rightarrow Ar \rightarrow Ar^{1} \rightarrow Ar^{1} \rightarrow BF_{3} \cdot OEt_{2} (10 \text{ mol-}\%) \rightarrow DCE, r.t., 3 \text{ h}$, 1.0 equiv., $Ar = p \cdot MeOC_{6}H_{4} \qquad 2, 3.0 \text{ equiv.}$	$ \begin{array}{c} Ar \\ Ar \\ Ar \\ Ar^{1} \\ Ar^{1} \\ 5 \end{array} $
Entry	2 (Ar ¹)	% Yield of 5 ^[a]
1	2b (<i>p</i> -EtOC ₆ H ₄)	5b , 50
2	2c [<i>p</i> -CH ₃ (CH ₂) ₅ OC ₆ H ₄]	5c , 50
3	2d $(p-CH_2=CHCH_2OC_6H_4)$	5d , 53
4	2f (<i>p</i> -CH≡CCH ₂ OC ₆ H ₄)	5e , 57

[a] Isolated yields.





Figure 5. ORTEP drawing of 5b.

delivered the desired product **5a** in 55% yield (Scheme 5). Moreover, the reaction of **3e** (0.1 mmol) with **2a** (0.2 mmol) under the standard conditions afforded a complex product mixture rather than the desired product **5a** (Scheme 5). On the basis of these above results, a plausible reaction mechanism to account for the formation of **5** is shown in Scheme 4. The addition of cationic intermediate **A** (Scheme 5) to **4eb** and **4ec** produces intermediate **G**, which undergoes an intramolecular electrophilic addition to give the cationic intermediate **H**. The deprotonation of intermediate **H** affords the corresponding cyclopentene **5a**.

In conclusion, we have found an interesting procedure wherein diaryl or aliphatic MCPs can react with bis(alkoxyphenyl)methanols to provide cyclopentenes as the major products along with cyclobutanes and diene derivatives as the minor products under mild conditions. In the presence of $BF_3 \cdot OEt_2$, MCP **1e**, with a strongly electron-donating methoxy group, reacts with bis(alkoxyphenyl)methanol to produce novel polysubstituted cyclopentene derivatives in good yields. Plausible reaction mechanisms based on the

Scheme 5. A control experiment and plausible reaction mechanism for the formation of **5**.

deuterium-labeling and control experiments have been proposed. Further studies regarding the mechanistic details and scope of this process are in progress.

Experimental Section

Typical Procedure for the Reaction of MCPs 1 with Bis(4-alk-oxyphenyl)methanol 2: MCP 1a (0.25 mmol, 52 mg) and bis(4-methoxyphenyl)methanol 2a (0.1 mmol, 24 mg) were dissolved in DCE (2.0 mL), and BF₃·OEt₂ (10 mol-%) was added. The mixture was stirred for 3.0 h at room temperature (25 °C). The solvent was removed in vacuo, and the residue was purified by flash column chromatography on silica gel with petroleum ether/EtOAc (40:1) as an eluent to afford a white solid 3a (32 mg, 73%) and the mixture of 4aa–4ac (6 mg, 14%). Since these reaction products, having electron-rich substituents on the aromatic rings, are quite labile during purification and recrystallization (the solid product will become an oily product along with some decomposed byproducts during recrystallization), we could not send all of them for elemental analyses.



MCP 1e (0.2 mmol, 53 mg) and bis(4-methoxyphenyl)methanol (2a) (0.6 mmol, 146 mg) were dissolved in DCE (2.0 mL), and BF₃·OEt₂ (10 mol-%) was added at room temperature. The mixture was stirred for 3 h. The solvent was removed in vacuo, and the residue was purified by flash column chromatography on silica gel with petroleum ether/EtOAc (10:1) as an eluent to give a white solid 5a (98 mg, 68%).

3a: A white solid (32 mg, 73%), m.p. 151–153 °C. ¹H NMR $(CDCl_3, 300 \text{ MHz}, TMS): \delta = 2.70 \text{ (dd, } J = 3.0, 16.2 \text{ Hz}, 1 \text{ H},$ CH₂), 3.61 (s, 3 H, OCH₃), 3.74 (s, 3 H, OCH₃), 3.77 (d, J =16.2 Hz, 1 H, CH₂), 4.79 (s, 1 H, CH), 6.19 (s, 1 H, =CH), 6.51 (d, J = 8.1 Hz, 2 H, Ar), 6.79 (d, J = 8.1 Hz, 2 H, Ar), 6.86 (d, J =8.7 Hz, 2 H, Ar), 6.94-7.04 (m, 5 H, Ar), 7.13-7.22 (m, 5 H, Ar), 7.37 (d, J = 8.7 Hz, 2 H, Ar) ppm. ¹³C NMR (CDCl₃, 75 MHz, TMS): $\delta = 43.8, 54.9, 55.1, 59.9, 61.6, 113.1, 113.7, 122.1, 125.3,$ 125.5, 127.2, 127.3, 127.5, 128.1, 128.2, 128.8, 130.0, 131.2, 145.3, 145.6, 151.9, 157.7, 158.8 ppm. IR (CH₂Cl₂): \tilde{v} = 3054, 3000, 2932, 2906, 2835, 2045, 1946, 1875, 1607, 1581, 1511, 1493, 1463, 1443, 1302, 1256, 1178, 1111, 1035, 830, 807, 789, 773, 756, 744, 700, 600, 531 cm⁻¹. MS (%): m/z = 432 (100) [M]⁺, 91 (7), 121 (12), 165 (9), 239 (9), 265 (24), 311 (27), 324 (8), 341 (8), 433 (35). C₃₁H₂₈O₂ (432.2089): calcd. C 86.08, H 6.52; found C 86.23, H 6.74. HRMS (EI): calcd. for C₃₁H₂₈O₂ 432.2089; found 432.2104.

3b: A colorless oil (28 mg, 61%). ¹H NMR (CDCl₃, 300 MHz, TMS): δ = 2.16 (s, 3 H, CH₃), 2.26 (s, 3 H, CH₃), 2.67 (dd, *J* = 3.3, 16.2 Hz, 1 H, CH₂), 3.61 (s, 3 H, OCH₃), 3.71 (d, *J* = 16.2 Hz, 1 H, CH₂), 3.73 (s, 3 H, OCH₃), 4.75 (s, 1 H, CH), 6.18 (s, 1 H, =CH), 6.51 (d, *J* = 8.7 Hz, 2 H, Ar), 6.76–6.88 (m, 8 H, Ar), 6.98 (d, *J* = 8.1 Hz, 2 H, Ar), 7.09 (d, *J* = 8.1 Hz, 2 H, Ar), 7.35 (d, *J* = 8.7 Hz, 2 H, Ar) ppm. ¹³C NMR (CDCl₃, 75 MHz, TMS): δ = 20.8, 20.9, 44.0, 54.9, 55.1, 59.9, 60.9, 113.1, 113.7, 122.2, 127.3, 127.8, 127.9, 128.2, 128.3, 128.6, 130.0, 131.4, 134.5, 134.9, 142.4, 145.7, 149.2, 157.6, 158.7 ppm. IR (CH₂Cl₂): \tilde{v} = 3050, 2998, 2922, 2835, 2045, 1891, 1607, 1582, 1573, 1512, 1463, 1442, 1302, 1246, 1177, 1111, 1036, 1003, 892, 864, 836, 808, 786, 750, 737, 702, 611, 575, 534 cm⁻¹. MS (%): *m/z* = 460 (100) [M]⁺, 105 (12), 121 (26), 239 (12), 265 (15), 339 (48), 355 (15), 461 (37). HRMS (EI): calcd. for C₃₃H₃₂O₂ 460.2402; found 460.2396.

3c: A colorless oil (29 mg, 62%). ¹H NMR (CDCl₃, 300 MHz, TMS): $\delta = 2.64$ (dd, J = 3.3, 16.5 Hz, 1 H, CH₂), 3.63 (s, 3 H, OCH₃), 3.70 (d, J = 16.5 Hz, 1 H, CH₂), 3.74 (s, 3 H, OCH₃), 4.68 (s, 1 H, CH), 6.17 (s, 1 H, =CH), 6.53 (d, J = 8.4 Hz, 2 H, Ar), 6.69 (d, J = 8.7 Hz, 1 H, Ar), 6.72 (d, J = 8.7 Hz, 1 H, Ar), 6.79 (d, J = 8.7 Hz, 2 H, Ar), 6.82–6.93 (m, 6 H, Ar), 7.12 (dd, J = 5.4, 8.7 Hz, 2 H, Ar), 7.35 (d, J = 8.4 Hz, 2 H, Ar) ppm. ¹³C NMR (CDCl₃, 75 MHz, TMS): δ = 44.1, 54.9, 55.1, 60.3, 60.6, 113.3, 113.8, 114.0 (d, $J_{C,F}$ = 20.6 Hz), 114.3 (d, $J_{C,F}$ = 20.6 Hz), 121.8, 127.3, 127.8, 129.4 (d, $J_{C,F}$ = 7.4 Hz), 129.8, 130.2 (d, $J_{C,F}$ = 8.0 Hz), 130.7, 140.9, 141.0, 145.5, 147.3, 157.9, 158.9, 160.7 (d, J = 242.7 Hz), 160.9 (d, J = 243.9 Hz) ppm. IR (CH₂Cl₂): \tilde{v} = 3058, 3000, 2933, 2907, 2836, 2043, 1962, 1885, 1757, 1607, 1583, 1573, 1514, 1463, 1442, 1418, 1406, 1305, 1261, 1163, 1140, 1108, 1035, 1015, 893, 865, 834, 787, 756, 737, 718, 701, 624, 610, 589, 573, 535, 508 cm⁻¹. MS (%): m/z = 468 (100) [M]⁺, 109 (15), 121 (36), 133 (11), 239 (16), 265 (27), 347 (24), 360 (12). HRMS (EI): calcd. for C₃₁H₂₆O₂F₂ 468.1901; found 468.1900.

3d: A colorless oil (28 mg, 55%). ¹H NMR (CDCl₃, 300 MHz, TMS): $\delta = 2.63$ (dd, J = 3.3, 16.5 Hz, 1 H, CH₂), 3.64 (s, 3 H, OCH₃), 3.69 (d, J = 16.5 Hz, 1 H, CH₂), 3.75 (s, 3 H, OCH₃), 4.67 (s, 1 H, CH), 6.16 (s, 1 H, =CH), 6.54 (d, J = 8.4 Hz, 2 H, Ar), 6.77–6.89 (m, 6 H, Ar), 7.00 (d, J = 8.4 Hz, 2 H, Ar), 7.08 (d, J = 8.7 Hz, 2 H, Ar), 7.14 (d, J = 8.7 Hz, 2 H, Ar), 7.34 (d, J = 8.7 Hz,

2 H, Ar) ppm. ¹³C NMR (CDCl₃, 75 MHz, TMS): δ = 43.8, 55.0, 55.2, 60.0, 60.8, 113.4, 113.8, 121.7, 127.3, 127.5, 127.70, 127.73, 129.4, 129.8, 130.1, 130.4, 131.3, 131.6, 143.5, 145.5, 149.8, 157.9, 158.9 ppm. IR (CH₂Cl₂): \tilde{v} = 3052, 3000, 2932, 2907, 2836, 2288, 2043, 1893, 1607, 1582, 1573, 1510, 1491, 1464, 1442, 1420, 1398, 1303, 1258, 1177, 1142, 1112, 1093, 1035, 1013, 1003, 972, 892, 864, 814, 787, 767, 737, 705, 673, 622, 607, 561, 535 cm⁻¹. MS (%): *m*/*z* = 500 (100) [M]⁺, 121 (40), 165 (15), 237 (15), 239 (21), 252 (37), 265 (61), 379 (22), 501 (34), 502 (67). HRMS (EI): calcd. for C₃₁H₂₆O₂Cl₂ 500.1310; found 500.1295.

3e: A colorless oil (28 mg, 57%). ¹H NMR (CDCl₃, 300 MHz, TMS): $\delta = 2.65$ (dd, J = 2.4, 16.2 Hz, 1 H, CH₂), 3.64 (s, 3 H, OCH3), 3.66 (s, 3 H, OCH3), 3.68 (s, 3 H, OCH3), 3.72 (s, 3 H, OCH₃), 3.78 (d, J = 16.2 Hz, 1 H, CH₂), 4.70 (s, 1 H, CH), 6.19 (s, 1 H, =CH), 6.53 (d, J = 8.4 Hz, 2 H, Ar), 6.56 (d, J = 8.4 Hz, 2 H, Ar), 6.72 (d, J = 8.4 Hz, 2 H, Ar), 6.79 (d, J = 8.4 Hz, 2 H, Ar), 6.85–6.89 (m, 4 H, Ar), 7.11 (d, J = 8.4 Hz, 2 H, Ar), 7.37 (d, J = 8.4 Hz, 2 H, Ar) ppm. ¹³C NMR (CDCl₃, 75 MHz, TMS): δ = 44.2, 54.9, 55.0, 55.1, 55.2, 60.2, 60.3, 112.5, 112.8, 113.2, 113.4, 113.5, 113.7, 122.2, 127.3, 128.3, 128.4, 128.9, 129.8, 130.0, 131.5, 137.8, 144.3, 145.7, 157.0, 157.3, 157.7, 158.7 ppm. IR (CH₂Cl₂): $\tilde{v} = 3035, 2999, 2932, 2906, 2835, 2058, 1607, 1581, 1510, 1463,$ 1441, 1301, 1249, 1178, 1111, 1036, 909, 822, 787, 754, 732, 611, 598, 579, 543 cm⁻¹. MS (%): m/z = 492 (100) [M]⁺, 121 (73), 371 (70), 227 (60), 253 (56), 252 (46), 493 (37), 385 (31). HRMS (EI): calcd. for C₃₃H₃₂O₄ 492.2300; found 492.2301.

3f: A colorless oil (40 mg, 68%). ¹H NMR (CDCl₃, 300 MHz, TMS): $\delta = 2.63$ (dd, J = 3.6, 16.5 Hz, 1 H, CH₂), 3.66 (s, 3 H, OCH₃), 3.68 (d, J = 16.5 Hz, 1 H, CH₂), 3.76 (s, 3 H, OCH₃), 3.68 (d, J = 16.5 Hz, 1 H, CH₂), 3.76 (s, 3 H, OCH₃), 4.66 (s, 1 H, CH), 6.16 (s, 1 H, =CH), 6.54 (d, J = 8.4 Hz, 2 H, Ar), 6.77–6.85 (m, 6 H, Ar), 7.02 (d, J = 8.4 Hz, 2 H, Ar), 7.15 (d, J = 8.4 Hz, 2 H, Ar), 7.30 (d, J = 8.4 Hz, 2 H, Ar), 7.15 (d, J = 8.4 Hz, 2 H, Ar), 7.30 (d, J = 8.4 Hz, 2 H, Ar), 7.33 (d, J = 8.4 Hz, 2 H, Ar) ppm. ¹³C NMR (CDCl₃, 75 MHz, TMS): $\delta = 43.8$, 55.0, 55.2, 59.9, 61.0, 113.5, 113.8, 119.6, 119.8, 121.7, 127.3, 127.7, 129.8, 129.9, 130.38, 130.43, 130.5, 130.7, 143.9, 145.6, 150.3, 158.0, 159.0 ppm. IR (CH₂Cl₂): $\tilde{v} = 2929$, 2835, 2359, 2342, 1683, 1652, 1606, 1558, 1509, 1488, 1472, 1457, 1437, 1418, 1394, 1302, 1256, 1177, 1111, 1075, 1035, 1009, 835, 808, 786, 763, 529 cm⁻¹. MS (%): mlz = 588 (51) [M]⁺, 121 (52), 239 (22), 265 (55), 388 (26), 390 (25), 589 (20), 590 (100). HRMS (EI): calcd. for C₃₁H₂₆O₂Br₂ 588.0297; found 588.0300.

3g: A colorless oil (30 mg, 64%); (syn or anti isomer) ¹H NMR $(CDCl_3, 300 \text{ MHz}, TMS): \delta = 2.66 \text{ (dd}, J = 3.3, 16.5 \text{ Hz}, 1 \text{ H},$ CH_2), 3.61 (s, 3 H, OCH₃), 3.71 (d, J = 16.5 Hz, 1 H, CH_2), 3.74 (s, 3 H, OCH₃), 4.70 (s, 1 H, CH), 6.17 (s, 1 H, =CH), 6.51 (d, J = 8.7 Hz, 2 H, Ar), 6.77–7.02 (m, 8 H, Ar), 7.12–7.22 (m, 4 H, Ar), 7.35 (d, J = 8.7 Hz, 2 H, Ar) ppm. (anti or syn isomer) ¹H NMR (CDCl₃, 300 MHz, TMS): $\delta = 2.70$ (dd, J = 3.3, 16.5 Hz, 1 H, CH₂), 3.63 (s, 3 H, OCH₃), 3.74 (s, 3 H, OCH₃), 3.76 (d, J =16.5 Hz, 1 H, CH₂), 4.76 (s, 1 H, CH), 6.18 (s, 1 H, =CH), 6.54 (d, J = 8.7 Hz, 2 H, Ar), 6.77–7.02 (m, 8 H, Ar), 7.12–7.22 (m, 4 H, Ar), 7.35 (d, J = 8.7 Hz, 2 H, Ar) ppm. (syn or anti isomer) ¹³C NMR (CDCl₃, 75 MHz, TMS): δ = 43.6, 54.9, 55.0, 59.8, 61.1, 113.2, 113.75, 121.9, 125.6, 127.3, 127.4, 127.6, 128.0, 128.7, 129.5, 130.2, 130.7, 131.0, 144.1, 145.5, 150.4, 157.8, 158.8 ppm. (anti or *syn* isomer) ¹³C NMR (CDCl₃, 75 MHz, TMS): δ = 44.0, 55.0, 55.1, 60.0, 61.3, 113.3, 113.8, 121.9, 125.7, 127.3, 127.6, 127.9, 128.7, 129.5, 129.9, 130.7, 130.8, 131.3, 144.7, 145.5, 151.3, 157.8, 158.9 ppm. IR (CH₂Cl₂): $\tilde{v} = 3054$, 2932, 2906, 2835, 1607, 1574, 1510, 1492, 1463, 1443, 1398, 1303, 1256, 1177, 1111, 1093, 1035, 1013, 832, 813, 787, 754, 737, 700, 605, 560, 534 cm⁻¹. MS (%): $m/z = 466 (100) [M]^+$, 121 (46), 165 (20), 237 (13), 252 (26), 265

(45), 310 (16), 344 (12), 345 (28), 467 (36), 468 (41). HRMS (EI): calcd. for $C_{31}H_{27}O_2Cl$ 466.1700; found 466.1707.

3h: A colorless oil (35 mg, 78%); (syn or anti isomer) ¹H NMR (CDCl₃, 300 MHz, TMS): δ = 2.66 (dd, J = 3.3, 16.5 Hz, 1 H, CH₂), 3.61 (s, 3 H, OCH₃), 3.71 (d, J = 16.5 Hz, 1 H, CH₂), 3.74 (s, 3 H, OCH₃), 4.70 (s, 1 H, CH), 6.17 (s, 1 H, =CH), 6.51 (d, J = 8.7 Hz, 2 H, Ar), 6.77–7.02 (m, 8 H, Ar), 7.12–7.22 (m, 4 H, Ar), 7.35 (d, J = 8.7 Hz, 2 H, Ar) ppm. (anti or syn isomer) ¹H NMR $(CDCl_3, 300 \text{ MHz}, TMS): \delta = 2.70 \text{ (dd, } J = 3.3, 16.5 \text{ Hz}, 1 \text{ H},$ CH₂), 3.63 (s, 3 H, OCH₃), 3.75 (s, 3 H, OCH₃), 3.76 (d, J =16.5 Hz, 1 H, CH₂), 4.76 (s, 1 H, CH), 6.18 (s, 1 H, =CH), 6.54 (d, J = 8.7 Hz, 2 H, Ar), 6.77–7.02 (m, 8 H, Ar), 7.12–7.22 (m, 4 H, Ar), 7.35 (d, J = 8.7 Hz, 2 H, Ar) ppm. (syn or anti isomer) ¹³C NMR (CDCl₃, 75 MHz, TMS): δ = 43.8, 54.9, 55.1, 60.0, 61.0, 113.1, 113.3, 113.9 (d, J = 22.6 Hz), 122.0, 125.5, 125.7, 127.3, 127.6, 128.0, 128.7, 129.5 (d, J = 7.6 Hz), 129.9, 130.9, 131.0, 141.2, 145.0, 145.5, 145.6, 147.58, 151.5, 157.7, 158.79, 160.6 (d, J =243.1 Hz) ppm. (anti or syn isomer) ¹³C NMR (CDCl₃, 75 MHz, TMS): $\delta = 44.1, 54.9, 55.1, 60.3, 61.1, 113.3, 113.7, 114.2$ (d, J =20.9 Hz), 122.0, 125.5, 125.7, 127.3, 127.6, 128.0, 128.7, 129.9, 130.2 (d, J = 7.4 Hz), 130.9, 131.0, 141.2, 145.0, 145.6, 147.6, 151.5, 157.8, 158.83, 160.8 (d, J = 242.9 Hz) ppm. IR (CH₂Cl₂): $\tilde{v} = 3055$, 3000, 2933, 2907, 2836, 2043, 1884, 1607, 1582, 1510, 1463, 1443, 1418, 1303, 1258, 1177, 1141, 1110, 1035, 1015, 909, 891, 864, 835, 787, 771, 752, 736, 700, 648, 628, 610, 578, 537 cm⁻¹. MS (%): m/z $= 450 (100) [M]^+, 121 (43), 265 (37), 451 (34), 329 (33), 252 (25),$ 165 (25), 239 (23). HRMS (EI): calcd. for C₃₁H₂₇O₂F 450.1994; found 450.1995.

3i: A colorless oil (13 mg, 25%). ¹H NMR (CDCl₃, 300 MHz, TMS): $\delta = 2.99$ (dd, J = 3.0, 16.8 Hz, 1 H, CH₂), 3.65 (d, J = 16.8 Hz, 1 H, CH₂), 3.67 (s, 3 H, OCH₃), 3.75 (s, 3 H, OCH₃), 4.61 (s, 1 H, CH), 6.21 (s, 1 H, =CH), 6.53 (d, J = 8.7 Hz, 2 H, Ar), 6.67–6.70 (m, 4 H, Ar), 6.74 (d, J = 8.7 Hz, 2 H, Ar), 6.92 (d, J = 8.7 Hz, 2 H, Ar), 7.17–7.54 (m, 6 H, Ar) ppm. ¹³C NMR (CDCl₃, 75 MHz, TMS): $\delta = 40.0$, 55.1, 55.2, 61.1, 61.3, 113.4, 113.8, 123.2, 126.0, 127.2, 127.7, 127.8, 128.6, 128.9, 129.6, 130.7, 131.9, 134.7, 142.8, 143.7, 146.0, 158.1, 158.9 ppm. IR (CH₂Cl₂): $\tilde{v} = 3052$, 3000, 2954, 2932, 2835, 1887, 1607, 1583, 1573, 1510, 1492, 1465, 1441, 1399, 1303, 1256, 1177, 1111, 1093, 1035, 1013, 863, 837, 815, 786, 753, 738, 681, 626, 607, 539 cm⁻¹. MS (%): m/z = 500 (100) [M]⁺, 121 (62), 125 (13), 165 (26), 227 (20), 239 (24), 252 (31), 265 (86), 501 (36), 502 (69), 503 (23). HRMS (EI): calcd. for C₃₁H₂₆O₂Cl₂ 500.1310; found 500.1303.

3j: A colorless oil (28 mg, 60%). ¹H NMR (CDCl₃, 300 MHz, TMS): $\delta = 1.27$ (t, J = 6.9 Hz, 3 H, CH₃), 1.36 (t, J = 6.9 Hz, 3 H, CH₃), 2.69 (dd, J = 2.7, 16.9 Hz, 1 H, CH₂), 3.76 (d, J = 16.9 Hz, 1 H, CH₂), 3.82 (q, J = 6.9 Hz, 2 H, OCH₂), 3.96 (q, J = 6.9 Hz, 2 H, OCH₂), 4.77 (s, 1 H, CH), 6.17 (s, 1 H, =CH), 6.49 (d, J = 8.4 Hz, 2 H, Ar), 6.76 (d, J = 8.4 Hz, 2 H, Ar), 6.84 (d, J = 8.4 Hz, 2 H, Ar), 6.93-7.03 (m, 5 H, Ar), 7.11-7.22 (m, 5 H, Ar), 7.35 (d, J = 8.4 Hz, 2 H, Ar) ppm. ¹³C NMR (CDCl₃, 75 MHz, TMS): δ = 14.7, 14.8, 43.8, 59.9, 61.6, 63.0, 63.3, 113.7, 114.2, 122.0, 125.3, 125.5, 127.2, 127.3, 127.5, 128.1, 128.9, 130.0, 131.1, 145.4, 145.6, 151.9, 157.1, 158.2 ppm. IR (CH₂Cl₂): $\tilde{v} = 3054$, 2979, 2927, 2903, 2848, 1943, 1882, 1607, 1580, 1509, 1493, 1477, 1443, 1392, 1302, 1244, 1177, 1116, 1090, 1047, 1002, 923, 863, 828, 803, 778, 756, 744, 699, 669, 648, 619, 548, 531 cm⁻¹. MS (%): m/z = 460 (100) [M]⁺, 107 (15), 165 (15), 293 (20), 325 (25), 461 (35). HRMS (EI): calcd. for C₃₃H₃₂O₂ 460.2402; found 460.2410.

3k: A colorless oil (34 mg, 60%). ¹H NMR (CDCl₃, 300 MHz, TMS): $\delta = 0.92-0.94$ (m, 6 H, CH₃), 1.33-1.54 (m, 12 H, CH₂), 1.67-1.80 (m, 4 H, CH₂), 2.74 (dd, J = 3.0, 16.2 Hz, 1 H, CH₂),

3.80 (d, J = 16.2 Hz, 1 H, CH₂), 3.81 (t, J = 6.6 Hz, 2 H, OCH₂), 3.93 (t, J = 6.6 Hz, 2 H, OCH₂), 4.82 (s, 1 H, CH), 6.22 (s, 1 H, =CH), 6.54 (d, J = 8.4 Hz, 2 H, Ar), 6.81 (d, J = 8.4 Hz, 2 H, Ar), 6.88 (d, J = 8.4 Hz, 2 H, Ar), 6.98–7.08 (m, 5 H, Ar), 7.16–7.26 (m, 5 H, Ar), 7.40 (d, J = 8.4 Hz, 2 H, Ar) ppm. ¹³C NMR (CDCl₃, 75 MHz, TMS): $\delta = 14.0$, 22.6, 25.7, 29.2, 31.5, 31.6, 43.8, 59.9, 61.6, 67.6, 67.8, 113.8, 114.2, 121.9, 125.3, 125.5, 127.2, 127.3, 127.5, 127.9, 128.1, 128.9, 129.9, 131.0, 145.4, 145.7, 151.9, 157.2, 158.4 ppm. IR (CH₂Cl₂): $\tilde{v} = 3055$, 2957, 2855, 1939, 1881, 1607, 1580, 1509, 1493, 1469, 1444, 1390, 1302, 1243, 1176, 1112, 1033, 937, 907, 888, 864, 831, 780, 756, 744, 699, 670, 649, 620, 556, 531 cm⁻¹. MS (%): m/z = 572 (100) [M]⁺, 43 (60), 91 (11), 167 (10), 381 (13), 573 (48), 574 (11). HRMS (EI): calcd. for C₄₁H₄₈O₂ 572.3654; found 572.3666.

31: A colorless oil (27 mg, 55%). ¹H NMR (CDCl₃, 300 MHz, TMS): δ = 2.69 (dd, J = 3.0, 16.5 Hz, 1 H, CH₂), 3.75 (d, J = 16.5 Hz, 1 H, CH₂), 4.31 (d, J = 4.8 Hz, 2 H, OCH₂), 4.46 (d, J = 4.8 Hz, 2 H, OCH₂), 4.77 (s, 1 H, CH), 5.18 (dd, J = 1.5, 10.5 Hz, 1 H, =CH₂), 5.24 (dd, J = 1.5, 10.5 Hz, 1 H, =CH₂), 5.27 (dd, J = 1.5, 17.1 Hz, 1 H, =CH₂), 5.34 (dd, *J* = 1.5, 17.1 Hz, 1 H, =CH₂), 5.87-6.05 (m, 2 H, =CH), 6.18 (s, 1 H, =CH), 6.51 (d, J = 8.4 Hz, 2 H, Ar), 6.79 (d, J = 8.4 Hz, 2 H, Ar), 6.84 (d, J = 8.4 Hz, 2 H, Ar), 6.91–7.03 (m, 5 H, Ar), 7.10–7.22 (m, 5 H, Ar), 7.34 (d, J =8.4 Hz, 2 H, Ar) ppm. $^{13}\mathrm{C}$ NMR (CDCl₃, 75 MHz, TMS): δ = 43.7, 59.9, 61.6, 68.4, 68.6, 113.9, 114.5, 117.3, 117.6, 122.2, 125.3, 125.5, 127.2, 127.3, 127.5, 128.1, 128.3, 128.8, 129.9, 131.3, 133.2, 133.3, 145.3, 145.5, 151.8, 156.7, 157.8 ppm. IR (CH₂Cl₂): $\tilde{v} =$ 3055, 2957, 2855, 1939, 1881, 1607, 1580, 1509, 1493, 1469, 1444, 1390, 1302, 1243, 1176, 1112, 1033, 937, 907, 888, 864, 831, 780, 756, 744, 699, 670, 649, 620, 556, 531 cm⁻¹. MS (%): m/z = 484(100) [M]⁺, 41 (98), 91 (31), 115 (21), 147 (11), 165 (21), 167 (42), 191 (36), 309 (14), 485 (39). HRMS (EI): calcd. for C₃₅H₃₂O₂ 484.2402; found 484.2407.

3m: A colorless oil (22 mg, 45%). $^1\mathrm{H}$ NMR (CDCl₃, 300 MHz, TMS): $\delta = 2.70$ (dd, J = 3.0, 16.2 Hz, 1 H, CH₂), 3.37 (s, 3 H, OCH₃), 3.45 (s, 3 H, OCH₃), 3.77 (d, *J* = 16.2 Hz, 1 H, CH₂), 4.78 (s, 1 H, CH), 5.00 (s, 2 H, OCH₂), 5.14 (s, 2 H, OCH₂), 6.22 (s, 1 H, =CH), 6.63 (d, J = 8.4 Hz, 2 H, Ar), 6.85 (d, J = 8.4 Hz, 2 H, Ar), 6.92–7.01 (m, 7 H, Ar), 7.1–7.20 (m, 5 H, Ar), 7.38 (d, J = 8.4 Hz, 2 H, Ar) ppm. ¹³C NMR (CDCl₃, 75 MHz, TMS): δ = 43.8, 55.9, 56.0, 59.9, 61.6, 94.3, 94.4, 115.5, 116.0, 122.7, 125.3, 125.5, 127.2, 127.3, 127.5, 128.1, 128.8, 129.3, 130.0, 132.4, 145.2, 145.4, 151.8, 155.5, 156.4 ppm. IR (CH₂Cl₂): $\tilde{v} = 3054$, 2997, 2928, 2848, 2824, 2786, 1994, 1947, 1888, 1727, 1606, 1578, 1510, 1443, 1406, 1312, 1240, 1200, 1150, 1111, 1077, 1000, 921, 888, 864, 808, 789, 772, 757, 744, 699, 669, 656, 631, 619, 594, 555, 531 cm⁻¹. MS (%): $m/z = 492 (100) [M]^+$, 167 (86), 169 (63), 91 (53), 115 (42), 101 (40), 191 (40), 69 (38). HRMS (EI): calcd. for C₃₃H₃₂O₄ 492.2301; found 492.2309.

3n: A colorless oil (27 mg, 57%). ¹H NMR (CDCl₃, 300 MHz, TMS): $\delta = 2.42$ (t, J = 2.4 Hz, 1 H, CH), 2.47 (t, J = 2.4 Hz, 1 H, CH), 2.70 (dd, J = 3.0, 16.2 Hz, 1 H, CH₂), 3.75 (d, J = 16.2 Hz, 1 H, CH₂), 4.47 (d, J = 2.4 Hz, 2 H, OCH₂), 4.62 (d, J = 2.4 Hz, 2 H, OCH₂), 4.78 (s, 1 H, CH), 6.21 (s, 1 H, =CH), 6.56 (d, J = 8.4 Hz, 2 H, Ar), 6.86 (d, J = 8.4 Hz, 4 H, Ar), 6.91–7.03 (m, 5 H, Ar), 7.11–7.22 (m, 5 H, Ar), 7.37 (d, J = 8.4 Hz, 2 H, Ar) ppm. ¹³C NMR (CDCl₃, 75 MHz, TMS): $\delta = 43.7$, 55.6, 59.8, 61.6, 75.2, 75.6, 78.5, 78.6, 114.1, 114.6, 122.7, 125.3, 125.5, 127.2, 127.3, 127.5, 128.0, 128.8, 128.9, 129.9, 132.0, 145.1, 145.2, 151.7, 155.8, 156.7 ppm. IR (CH₂Cl₂): $\tilde{v} = 3290$, 3053, 2924, 2850, 2121, 1878, 1605, 1509, 1493, 1444, 1375, 1304, 1266, 1219, 1178, 1114, 1030, 925, 864, 832, 783, 757, 744, 700, 644, 553, 530 cm⁻¹. MS (%): *m/z*



= 480 (76) [M]⁺, 191 (100), 167 (95), 91 (89), 115 (89), 165 (89), 178 (53), 192 (43). HRMS (EI): calcd. for $C_{35}H_{28}O_2$ 480.2089; found 480.2093.

30: A colorless oil (22 mg, 52%). ¹H NMR (CDCl₃, 300 MHz, TMS): δ = 1.46–1.82 (m, 8 H, CH₂), 2.06 (t, *J* = 7.2 Hz, 2 H, CH₂), 2.27–2.34 (m, 1 H, CH), 2.82 (t, *J* = 7.2 Hz, 2 H, CH₂), 3.72 (s, 3 H, OCH₃), 3.86 (s, 3 H, OCH₃), 6.64 (d, *J* = 8.7 Hz, 2 H, Ar), 6.89 (d, *J* = 8.7 Hz, 2 H, Ar), 7.01 (d, *J* = 8.7 Hz, 2 H, Ar), 7.14–7.30 (m, 5 H, Ar) ppm. ¹³C NMR (CDCl₃, 75 MHz, TMS): δ = 30.9, 33.0, 33.3, 35.0, 43.6, 52.7, 55.0, 55.1, 113.0, 113.6, 125.8, 126.7, 128.2, 128.8, 130.1, 130.7, 131.1, 135.3, 145.5, 147.7, 157.8, 158.2 ppm. IR (CH₂Cl₂): \tilde{v} = 3058, 3027, 2999, 2926, 2850, 2063, 1884, 1605, 1570, 1513, 1441, 1338, 1284, 1248, 1178, 1106, 1053, 1035, 895, 830, 802, 788, 757, 700, 626, 586, 570, 540 cm⁻¹. MS (%): *m/z* = 424 (71) [M]⁺, 91 (16), 121 (27), 171 (48), 292 (21), 305 (100), 306 (25), 425 (25). HRMS (EI): calcd. for C₃₀H₃₂O₂ 424.2402; found 424.2403.

3p: A white solid (32 mg, 65%), m.p. 188–190 °C. ¹H NMR $(CDCl_3, 400 \text{ MHz}, TMS): \delta = 1.52-1.57 \text{ (m, 2 H, CH}_2), 1.64-1.71$ (m, 4 H, CH₂), 1.81–1.85 (m, 2 H, CH₂), 2.07 (t, J = 7.2 Hz, 2 H, CH₂), 2.31–2.39 (m, 1 H, CH), 2.83 (t, J = 7.2 Hz, 2 H, CH₂), 3.71 (s, 3 H, OCH₃), 3.83 (s, 3 H, OCH₃), 6.64–6.66 (m, 2 H, Ar), 6.89– 6.91 (m, 2 H, Ar), 6.98-7.03 (m, 4 H, Ar), 7.24-7.33 (m, 3 H, Ar), 7.39-7.43 (m, 2 H, Ar), 7.49-7.55 (m, 2 H, Ar), 7.56-7.58 (m, 2 H, Ar) ppm. ¹³C NMR (CDCl₃, 100 MHz, TMS): δ = 30.9, 33.1, 33.3, 35.0, 43.2, 52.8, 55.06, 55.12, 113.1, 113.7, 126.93, 126.97, 127.01, 127.2, 128.7, 128.8, 130.1, 130.8, 131.1, 135.4, 138.8, 141.1, 145.5, 146.8, 157.8, 158.3 ppm. IR (CH₂Cl₂): $\tilde{v} = 3026$, 2924, 2848, 1882, 1605, 1571, 1510, 1485, 1461, 1439, 1409, 1287, 1240, 1173, 1106, 1054, 1028, 908, 827, 788, 765, 736, 698, 586, 562, 545, 511 cm⁻¹. MS (%): m/z = 500 (55) [M]⁺, 305 (100), 171 (47), 292 (31), 306 (24), 501 (23), 121 (22), 180 (10). $C_{36}H_{36}O_2$ (500.2715): calcd. C 86.36, H 7.25; found C 86.68, H 7.23. HRMS (EI): calcd. for C₃₀H₃₆O₂ 500.2715; found 500.2717.

3q: A white solid (30 mg, 67%), m.p. 163–165 °C. ¹H NMR $(CDCl_3, 400 \text{ MHz}, TMS)$: $\delta = 1.47-1.55 \text{ (m, 2 H, CH_2)}, 1.63-1.69$ (m, 5 H, CH₂), 2.07 (t, J = 7.2 Hz, 2 H, CH₂), 2.21 (s, 3 H, CH₃), 2.27 (s, 3 H, CH₃), 2.43–2.50 (m, 1 H, CH), 2.82 (t, J = 7.2 Hz, 2 H, CH₂), 3.72 (s, 3 H, OCH₃), 3.84 (s, 3 H, OCH₃), 6.64–6.66 (m, 2 H, Ar), 6.89-6.93 (m, 3 H, Ar), 6.96-7.02 (m, 5 H, Ar), 7.14 (d, J = 8.0 Hz, 1 H, Ar) ppm. $^{13}\mathrm{C}$ NMR (CDCl₃, 100 MHz, TMS): δ = 19.3, 20.8, 30.1, 33.1, 33.3, 35.2, 38.7, 52.9, 55.06, 55.10, 113.0, 113.7, 125.2, 126.7, 128.8, 130.1, 130.8, 131.0, 131.1, 134.9, 135.1, 135.4, 142.5, 145.6, 157.8, 158.3 ppm. IR (CH₂Cl₂): \tilde{v} = 3032, 2999, 2926, 2849, 1605, 1570, 1442, 1283, 1246, 1178, 1105, 1053, 1035, 909, 829, 788, 733, 591, 568, 539 cm⁻¹. MS (%): m/z = 452 (65) [M]⁺, 305 (100), 132 (46), 171 (46), 119 (31), 292 (27), 121 (27), 453 (25). $C_{32}H_{36}O_2$ (452.2715): calcd. C 84.91, H 8.02; found C 85.37, H 7.83. HRMS (EI): calcd. for C₃₂H₃₆O₂ 452.2715; found 452.2714.

3r: A colorless oil (28 mg, 61%). ¹H NMR (CDCl₃, 400 MHz, TMS): $\delta = 1.11-1.26$ (m, 3 H, CH₂), 1.55–1.67 (m, 3 H, CH₂), 1.74–1.77 (m, 2 H, CH₂), 2.04–2.08 (m, 2 H, CH₂), 2.66–2.74 (m, 1 H, CH), 2.81 (t, J = 7.2 Hz, 1 H, CH₂), 3.70 (s, 3 H, OCH₃), 3.81 (s, 3 H, OCH₃), 6.62–6.64 (m, 2 H, Ar), 6.85–6.90 (m, 2 H, Ar), 6.93–6.96 (m, 4 H, Ar), 7.06–7.10 (m, 2 H, Ar), 7.21–7.24 (m, 2 H, Ar) ppm. ¹³C NMR (CDCl₃, 100 MHz, TMS): $\delta = 23.3$, 33.1, 33.40, 33.43, 33.9, 40.4, 42.9, 53.9, 55.0, 55.1, 113.0, 113.7, 128.2, 128.3, 128.8, 129.8, 130.0, 130.5, 131.1, 131.3, 135.6, 145.2, 146.0, 157.9, 158.3 ppm. IR (CH₂Cl₂): $\tilde{v} = 3032, 2997, 2927, 2849, 1882, 1605, 1511, 1493, 1463, 1442, 1283, 1247, 1178, 1091, 1035, 1013, 908, 828, 732, 572, 530 cm⁻¹. MS (%): <math>m/z = 458$ (100) [M]⁺, 415

(75), 460 (36), 459 (34), 291 (34), 121 (30), 171 (29), 417 (28). HRMS (EI): calcd. for $C_{30}H_{31}O_2CI$: 458.2013; found 458.2014.

4aa and 4ac: A colorless oil (6 mg, 14%); **4aa:** ¹H NMR (CDCl₃, 300 MHz, TMS): $\delta = 2.72-2.83$ (m, 4 H, CH₂), 3.67 (s, 3 H, OCH₃), 3.76 (s, 3 H, OCH₃), 6.50 (d, J = 8.7 Hz, 2 H, Ar), 6.67 (d, J = 8.7 Hz, 2 H, Ar), 7.12–7.26 (m, 8 H, Ar), 7.32–7.35 (m, 4 H, Ar) ppm. **4ac:** ¹H NMR (CDCl₃, 300 MHz, TMS): $\delta = 3.55$ (d, J = 6.9 Hz, 2 H, CH₂), 3.80 (s, 3 H, OCH₃), 3.83 (s, 3 H, OCH₃), 6.04 (t, J = 6.9 Hz, 1 H, =CH), 6.49–7.35 (m, 19 H, Ar) ppm. IR (CH₂Cl₂): $\tilde{v} = 3043$, 2996, 2955, 2836, 2359, 2342, 1788, 1686, 1664, 1602, 1572, 1506, 1463, 1442, 1409, 1289, 1248, 1178, 1166, 1109, 1032, 1014, 830, 795, 766, 616, 576, 540 cm⁻¹. MS (%): m/z = 432 (100) [M]⁺, 121 (29), 239 (20), 265 (34), 311 (38), 433 (36). HRMS (EI): calcd. for C₃₁H₂₈O₂ 432.2091; found 432.2089.

4ca, 4cb and 4cc: A colorless oil (9 mg, 19%); 4ca: ¹H NMR $(CDCl_3, 300 \text{ MHz}, TMS): \delta = 2.66-2.70 \text{ (m, 2 H, CH}_2), 2.76-2.82$ (m, 2 H, CH₂), 3.69 (s, 3 H, OCH₃), 3.76 (s, 3 H, OCH₃), 6.53 (d, J = 8.7 Hz, 2 H, Ar), 6.64 (d, J = 8.7 Hz, 2 H, Ar), 6.73 (d, J =8.7 Hz, 2 H, Ar), 6.90 (d, J = 8.7 Hz, 2 H, Ar), 6.93 (d, J = 8.7 Hz, 2 H, Ar), 7.12 (d, J = 8.7 Hz, 2 H, Ar), 7.23 (d, J = 8.7 Hz, 4 H, Ar) ppm. **4cb**: ¹H NMR (CDCl₃, 300 MHz, TMS): δ = 3.19 (d, J = 7.2 Hz, 2 H, CH₂), 3.72 (s, 3 H, OCH₃), 3.80 (s, 3 H, OCH₃), 6.10 (t, J = 7.2 Hz, 1 H, =CH), 6.33 (s, 1 H, =CH), 6.53–7.23 (m, 16 H, Ar) ppm. 4cc: ¹H NMR (CDCl₃, 300 MHz, TMS): δ = 3.52 $(d, J = 6.9 \text{ Hz}, 2 \text{ H}, \text{CH}_2), 3.79 (s, 3 \text{ H}, \text{OCH}_3), 3.82 (s, 3 \text{ H}, \text{OCH}_3),$ 5.95 (t, J = 6.9 Hz, 1 H, =CH), 6.53–7.23 (m, 17 H, Ar) ppm. IR (CH_2Cl_2) : $\tilde{v} = 3041, 2999, 2955, 2836, 2359, 2342, 1782, 1683, 1662,$ 1602, 1572, 1506, 1463, 1442, 1409, 1289, 1248, 1178, 1161, 1106, 1032, 1014, 830, 795, 766, 616, 576, 540 cm⁻¹. MS (%): m/z = 468(97) [M]⁺, 109 (20), 121 (34), 265 (16), 361 (21), 373 (100), 374 (28), 469 (35). HRMS (EI): calcd. for C₃₁H₂₆O₂F₂: 468.1893; found 468,1901.

4da, 4db and 4dc: A colorless oil (5 mg, 10%); 4da: ¹H NMR $(CDCl_3, 300 \text{ MHz}, TMS): \delta = 2.65-2.69 \text{ (m, 2 H, CH}_2), 2.77-2.81$ (m, 2 H, CH₂), 3.69 (s, 3 H, OCH₃), 3.76 (s, 3 H, OCH₃), 6.54 (d, J = 8.7 Hz, 2 H, Ar), 6.65 (d, J = 8.7 Hz, 2 H, Ar), 6.73 (d, J =8.7 Hz, 2 H, 2 H, Ar), 7.12 (d, J = 8.7 Hz, 2 H, Ar), 7.21 (s, 8 H, Ar) ppm. 4db: ¹H NMR (CDCl₃, 300 MHz, TMS): δ = 3.19 (d, J = 7.5 Hz, 2 H, CH₂), 3.77 (s, 3 H, OCH₃), 3.81 (s, 3 H, OCH₃), 6.15 (t, J = 7.5 Hz, 1 H, =CH), 6.32 (s, 1 H, =CH) 6.54–7.21 (m, 16 H, Ar) ppm. 4dc: ¹H NMR (CDCl₃, 300 MHz, TMS): δ = 3.52 (d, J = 6.3 Hz, 2 H, CH₂), 3.73 (s, 3 H, OCH₃), 3.83 (s, 3 H, OCH₃), 6.00 (t, J = 6.3 Hz, 1 H, =CH) 6.54–7.21 (m, 17 H, Ar) ppm. IR (CH_2Cl_2) : $\tilde{v} = 3032, 3000, 2955, 2933, 2835, 2360, 2342, 1683, 1604,$ 1575, 1508, 1463, 1441, 1419, 1398, 1287, 1248, 1176, 1091, 1033, 1013, 830, 756, 738, 519 cm⁻¹. MS (%): $m/z = 500 (100) [M]^+$, 121 (62), 135 (17), 165 (21), 239 (36), 252 (21), 253 (22), 265 (54), 389 (49), 501 (38), 502 (73) 503 (25). HRMS (EI): calcd. for C₃₁H₂₆O₂Cl₂ 500.1309; found 500.1310.

4ea and 4ec: A colorless oil (16 mg, 32%); **4ea**: ¹H NMR (CDCl₃, 300 MHz, TMS): δ = 2.65–2.69 (m, 2 H, CH₂), 2.74–2.79 (m, 2 H, CH₂), 3.68 (s, 3 H, OCH₃), 3.76 (s, 3 H, OCH₃), 3.78 (s, 6 H, 2 OCH₃), 6.52 (d, *J* = 8.7 Hz, 2 H, Ar), 6.65–6.78 (m, 8 H, Ar), 7.12 (d, *J* = 8.7 Hz, 2 H, Ar), 7.22 (d, *J* = 8.7 Hz, 2 H, Ar), 7.25 (d, *J* = 8.7 Hz, 2 H, Ar) ppm. **4ec**: ¹H NMR (CDCl₃, 300 MHz, TMS): δ = 3.54 (d, *J* = 6.6 Hz, 2 H, CH₂), 3.78 (s, 3 H, OCH₃), 3.79 (s, 3 H, OCH₃), 3.82 (s, 3 H, OCH₃), 3.83 (s, 3 H, OCH₃), 5.89 (t, *J* = 6.6 Hz, 1 H, =CH), 652–7.25 (m, 17 H, Ar) ppm. IR (CH₂Cl₂): \tilde{v} = 3033, 2997, 2953, 2928, 2835, 2359, 2342, 1605, 1575, 1507, 1463, 1441, 1289, 1248, 1178, 1112, 1036, 829, 802 cm⁻¹. MS (%): *m/z* = 492 (83) [M]⁺, 121 (27), 227 (11), 253 (12), 371 (14), 384 (11), 385

(100), 386 (29), 493 (30). HRMS (EI): calcd. for $C_{33}H_{32}O_4{:}$ 492.2300; found 492.2301.

4fa, 4fb and 4fc: A colorless oil (18 mg, 31%); 4fa: ¹H NMR $(CDCl_3, 300 \text{ MHz}, TMS)$: $\delta = 2.64-2.68 \text{ (m, 2 H, CH}_2), 2.77-2.81$ (m, 2 H, CH₂), 3.70 (s, 3 H, OCH₃), 3.76 (s, 3 H, OCH₃), 6.54 (d, J = 8.7 Hz, 2 H, Ar), 6.64 (d, J = 8.7 Hz, 2 H, Ar), 6.72 (d, J =8.7 Hz, 2 H, Ar), 7.11 (d, J = 8.7 Hz, 2 H, Ar), 7.15 (d, J = 8.7 Hz, 4 H, Ar), 7.35 (d, J = 8.7 Hz, 4 H, Ar) ppm. 4fb: ¹H NMR (CDCl₃, 300 MHz, TMS): δ = 3.18 (d, J = 6.6 Hz, 2 H, CH₂), 3.73 (s, 3 H, OCH₃), 3.78 (s, 3 H, OCH₃), 6.15 (t, J = 6.6 Hz, 1 H, =CH), 6.32 (s, 1 H, =CH), 6.54–7.35 (m, 16 H, Ar) ppm. 4fc: ¹H NMR (CDCl₃, 300 MHz, TMS): δ = 3.51 (d, J = 6.9 Hz, 2 H, CH₂), 3.82 (s, 3 H, OCH₃), 3.83 (s, 3 H, OCH₃), 6.01 (t, J = 6.9 Hz, 1 H, =CH), 6.54-7.35 (m, 17 H, Ar) ppm. IR (CH₂Cl₂): \tilde{v} = 2953, 2927, 2835, 2360, 1653, 1604, 1570, 1558, 1507, 1487, 1458, 1439, 1289, 1248, 1177, 1076, 1034, 1008, 829, 796 cm⁻¹. MS (%): m/z = 588 (48) [M]⁺, 121 (47), 265 (22), 433 (81), 434 (25), 435 (77), 436 (22), 589 (18), 590 (100), 591 (34), 592 (54), 593 (17). HRMS (EI): calcd. for C₃₁H₂₆O₂Br₂ 588.0302; found 588.0300.

4ha, 4hb and 4hc: A colorless oil (10 mg, 22%). ¹H NMR (CDCl₃, 300 MHz, TMS): $\delta = 2.68-2.73$ (m, 2 H, CH₂), 2.76–2.81 (m, 2 H, CH₂), 3.68 (s, 3 H, OCH₃), 3.76 (s, 3 H, OCH₃), 6.52 (d, J = 8.7 Hz, 2 H, Ar), 6.66 (d, J = 8.7 Hz, 2 H, Ar), 6.72 (d, J = 8.7 Hz, 2 H, Ar), 6.90 (t, J = 8.7 Hz, 2 H, Ar), 7.13 (d, J = 8.7 Hz, 2 H, Ar), 7.24–7.28 (m, 5 H, Ar) ppm. IR (CH₂Cl₂): $\tilde{v} = 3055$, 2999, 2954, 2836, 2538, 2358, 2048, 1888, 1683, 1662, 1602, 1574, 1506, 1463, 1443, 1289, 1248, 1177, 1161, 1033, 831, 812, 797, 765, 702, 576, 557, 539 cm⁻¹. MS (%): m/z = 450 (100) [M]⁺, 121 (34), 343 (18), 355 (34), 373 (63), 374 (19), 451 (37). HRMS (EI): calcd. for C₃₁H₂₇O₂F: 450.1991; found 450.1995.

4ka and 4kb: A colorless oil (5 mg, 8%); **4ka**: ¹H NMR (CDCl₃, 300 MHz, TMS): $\delta = 0.83-0.90$ (m, 6 H, CH₃), 1.27–1.45 (m, 12 H, CH₂), 1.66–1.77 (m, 4 H, CH₂), 2.72–2.80 (m, 4 H, 2 CH₂), 3.79 (t, J = 6.6 Hz, 2 H, OCH₂), 3.90 (t, J = 6.6 Hz, 2 H, OCH₂), 6.49 (d, J = 8.7 Hz, 2 H, Ar), 6.65 (d, J = 8.7 Hz, 2 H, Ar), 6.71 (d, J = 8.7 Hz, 2 H, Ar), 7.10–7.23 (m, 8 H, Ar), 7.33 (d, J = 8.7 Hz, 4 H, Ar) ppm. **4kb**: ¹H NMR (CDCl₃, 300 MHz, TMS): $\delta = 3.22$ (d, J = 7.5 Hz, 2 H, CH₂), 6.19 (t, J = 7.5 Hz, 1 H, =CH), 6.35 (s, 1 H, =CH), 6.49–7.33 (m, 18 H, Ar) ppm. IR (CH₂Cl₂): $\tilde{v} = 3057$, 2930, 2858, 2357, 2054, 1653, 1604, 1575, 1506, 1469, 1446, 1285, 1245, 1177, 1031, 830, 760, 701 cm⁻¹. MS (%): m/z = 572 (100) [M]⁺, 43 (73), 107 (18), 495 (70), 496 (28), 573 (46), 574 (11). HRMS (EI): calcd. for C₄₁H₄₈O₂ 572.3652; found 572.3654.

4ab: A white solid (1 mg, 1%), m.p. 90–92 °C. ¹H NMR (CDCl₃, 300 MHz, TMS): δ = 3.22 (d, J = 7.8 Hz, 2 H, CH₂), 3.71 (s, 3 H, OCH₃), 3.78 (s, 3 H, OCH₃), 6.18 (t, J = 7.8 Hz, 1 H, =CH), 6.55 (s, 1 H, =CH), 6.64 (d, J = 8.4 Hz, 2 H, Ar), 6.81 (d, J = 8.4 Hz, 2 H, Ar), 6.89 (d, J = 8.4 Hz, 2 H, Ar), 7.03 (d, J = 8.4 Hz, 2 H, Ar), 7.12–7.37 (m, 10 H, Ar) ppm. ¹³C NMR (CDCl₃, 75 MHz, TMS): δ = 40.6, 55.06, 55.12, 113.2, 113.8, 125.8, 126.7, 126.96, 127.01, 127.3, 128.0, 128.1, 129.6, 129.8, 130.09, 130.14, 133.6, 139.6, 139.8, 142.6, 142.9, 157.8, 158.4 ppm. IR (CH₂Cl₂): $\tilde{\nu}$ = 3028, 2926, 2853, 2834, 1608, 1510, 1461, 1442, 1287, 1247, 1178, 1074, 1035, 900, 830, 805, 766, 745, 701, 600 cm⁻¹. MS (%): m/z = 432 (100) [M]⁺, 121 (48), 433 (37), 311 (35), 265 (24), 165 (19), 310 (18), 91 (17). HRMS (EI): calcd. for C₃₁H₂₈O₂: 432.2091; found 432.2089.

4aB: A white solid (25 mg, 98%), m.p. 94–96 °C. ¹H NMR (CDCl₃, 300 MHz, TMS): $\delta = 3.73$ (s, 3 H, OCH₃), 3.76 (s, 3 H, OCH₃), 4.06 (s, 2 H, CH₂), 6.72 (d, J = 11.4 Hz, 1 H, =CH), 6.74 (d, J = 8.7 Hz, 2 H, Ar), 6.81 (d, J = 8.7 Hz, 2 H, Ar), 7.09 (d, J = 11.4 Hz, 1 H, =CH), 7.17 (d, J = 8.7 Hz, 2 H, Ar), 7.23–7.44 (m, 12 H, Ar)

ppm. ¹³C NMR (CDCl₃, 100 MHz, TMS): δ = 34.9, 55.19, 55.21, 113.7, 113.9, 124.6, 125.2, 127.27, 127.32, 127.4, 127.7, 128.2, 129.0, 130.7, 131.8, 134.6, 139.5, 139.9, 142.8, 143.3, 157.9, 158.8 ppm. IR (CH₂Cl₂): \tilde{v} = 3028, 2932, 2834, 1608, 1510, 1442, 1284, 1247, 1178, 1111, 1074, 1036, 898, 829, 804, 766, 746, 701, 600 cm⁻¹. MS (%): *m/z* = 432 (100) [M]⁺, 121 (48), 433 (37), 311 (35), 265 (24), 165 (19), 310 (18), 91 (17). HRMS (EI): calcd. for C₃₁H₂₈O₂ 432.2091; found 432.2089.

5a: A white solid (98 mg, 68%), m.p. 244-246 °C. ¹H NMR $(CDCl_3, 300 \text{ MHz}, \text{TMS}): \delta = 2.39 \text{ (dd, } J = 10.2, 16.2 \text{ Hz}, 1 \text{ H},$ CH₂), 2.56 (dd, *J* = 7.8, 16.2 Hz, 1 H, CH₂), 3.21 (d, *J* = 10.5 Hz, 1 H, CH), 3.63 (s, 3 H, OCH₃), 3.67 (s, 3 H, OCH₃), 3.72 (s, 9 H, 3 OCH₃), 3.87 (s, 3 H, OCH₃), 4.23-4.32 (m, 1 H, CH), 6.41 (d, J = 8.7 Hz, 2 H, Ar), 6.44 (d, J = 8.7 Hz, 2 H, Ar), 6.52 (d, J = 8.7 Hz, 2 H, Ar), 6.57 (d, J = 8.7 Hz, 2 H, Ar), 6.65 (d, J = 8.7 Hz, 2 H, Ar), 6.68 (d, J = 8.7 Hz, 2 H, Ar), 6.84 (d, J = 8.7 Hz, 2 H, Ar), 6.91 (d, J = 8.7 Hz, 2 H, Ar), 6.97 (d, J = 8.7 Hz, 2 H, Ar), 7.00 (d, J = 8.7 Hz, 2 H, Ar), 7.12 (d, J = 8.7 Hz, 2 H, Ar), 7.17 (d, *J* = 8.7 Hz, 2 H, Ar) ppm. ¹³C NMR (CDCl₃, 75 MHz, TMS): $\delta = 41.6, 52.1, 53.0, 54.8, 55.0, 55.1, 55.2, 68.1, 112.0, 112.5, 112.7,$ 112.8, 113.1, 113.6, 128.7, 129.5, 129.6, 129.8, 130.1, 130.7, 131.1, 131.3, 132.1, 135.9, 136.6, 136.8, 138.3, 143.9, 157.1, 157.2, 157.4, 157.6, 158.0, 158.2 ppm. IR (CH₂Cl₂): $\tilde{v} = 3033$, 3000, 2933, 2905, 2834, 2532, 2054, 1890, 1606, 1581, 1514, 1461, 1441, 1419, 1336, 1290, 1254, 1177, 1115, 1035, 831, 788, 773, 738, 704, 684, 653, 642, 632, 617, 599, 584, 566, 543, 517 cm⁻¹. MS (%): m/z = 718 (2) [M]⁺, 227 (100), 478 (37), 357 (29), 121 (26), 240 (18), 228 (18), 491 (16), 238 (15). C₄₈H₄₆O₆ (718.89): calcd. C 80.20, H 6.45; found C 80.25, H 6.61. HRMS (EI): calcd. for C₄₈H₄₆O₆ 718.3294; found 718.3306.

5b: A white solid (77 mg, 50%), m.p. 197–199 °C. $^1\mathrm{H}$ NMR $(CDCl_3, 300 \text{ MHz}, TMS)$: $\delta = 1.26-1.36 \text{ (m, 12 H, 4 CH}_3), 2.38$ (dd, *J* = 10.5, 15.9 Hz, 1 H, CH₂), 2.56 (dd, *J* = 7.5, 15.9 Hz, 1 H, CH₂), 3.19 (d, J = 10.5 Hz, 1 H, CH), 3.70 (s, 3 H, OCH₃), 3.78-3.98 (m, 11 H, CH₂ and OCH₃), 4.22-4.32 (m, 1 H, CH), 6.40 (d, J = 9.3 Hz, 2 H, Ar), 6.44 (d, J = 9.3 Hz, 2 H, Ar), 6.50–6.73 (m, 8 H, Ar), 6.81–7.00 (m, 8 H, Ar), 7.12 (d, J = 8.7 Hz, 2 H, Ar), 7.18 (d, J = 8.7 Hz, 2 H, Ar) ppm. ¹³C NMR (CDCl₃, 75 MHz, TMS): $\delta = 14.76, 14.84, 41.6, 52.2, 53.1, 55.1, 55.2, 62.9, 63.1, 63.2,$ 68.1, 112.0, 112.5, 113.2, 113.4, 113.7, 113.9, 114.1, 128.7, 129.5, 129.9, 130.0, 130.7, 131.1, 131.3, 132.1, 135.8, 136.6, 136.8, 138.2, 143.9, 156.6, 156.8, 157.0, 157.1, 157.5, 158.0 ppm. IR (CH₂Cl₂): $\tilde{v} = 3035, 2979, 2932, 2904, 2835, 1606, 1581, 1509, 1478, 1463,$ 1442, 1392, 1298, 1251, 1178, 1116, 1045, 923, 828, 808, 739, 611, 549, 518 cm⁻¹. MS (%): m/z = 774 (18) [M]⁺, 506 (100), 255 (75), 519 (47), 227 (35), 268 (20), 371 (18). C₅₂H₅₄O₆ (775.00): calcd. C 80.59, H 7.02; found C 80.57, H 7.28. HRMS (EI): calcd. for C₅₂H₅₄O₆ 774.3929; found 774.3920.

5c: A colorless oil (102 mg, 50%). ¹H NMR (CDCl₃, 300 MHz, TMS): $\delta = 0.86-0.90$ (m, 12 H, CH₃), 1.27–1.38 (m, 24 H, CH₂), 1.57–1.75 (m, 8 H, CH₂), 2.38 (dd, J = 7.8, 15.9 Hz, 1 H, CH₂), 2.57 (dd, J = 7.5, 15.9 Hz, 1 H, CH₂), 3.20 (d, J = 10.8 Hz, 1 H, CH), 3.71 (s, 3 H, OCH₃), 3.74–3.98 (m, 11 H, CH₂ and CH₃), 4.20–4.32 (m, 1 H, CH), 6.40 (d, J = 9.3 Hz, 2 H, Ar), 6.43 (d, J = 9.3 Hz, 2 H, Ar), 6.50–6.72 (m, 8 H, Ar), 6.77–7.00 (m, 8 H, Ar), 7.12 (d, J = 9.3 Hz, 2 H, Ar), 7.19 (d, J = 9.3 Hz, 2 H, Ar) ppm. ¹³C NMR (CDCl₃, 75 MHz, TMS): $\delta = 14.0$, 22.6, 25.65, 25.70, 29.2, 29.3, 31.5, 41.6, 52.1, 53.1, 55.1, 55.2, 67.5, 67.7, 68.1, 112.0, 112.5, 113.3, 113.4, 113.7, 113.8, 113.9, 114.1, 128.7, 129.4, 129.5, 129.8, 129.9, 130.7, 131.1, 131.4, 132.0, 132.1, 135.8, 136.7, 136.8, 138.2, 143.8, 156.8, 157.0, 157.1, 157.2, 157.7, 158.0 ppm. IR (CH₂Cl₂): $\tilde{v} = 3035$, 2931, 2859, 1606, 1581, 1509, 1468, 1390, 1285,



1244, 1178, 1114, 1036, 938, 830, 787, 728, 614, 552 cm⁻¹. ESI-MS (%): $m/z = 1021 \text{ [M + Na]^+}$. HRMS (ESI): calcd. for C₆₈H₈₆O₆ 1021.6324; found 1021.6317.

5d: A white solid (90 mg, 53%), m.p. 182-184 °C. ¹H NMR $(CDCl_3, 300 \text{ MHz}, TMS): \delta = 2.39 \text{ (dd, } J = 10.8, 16.2 \text{ Hz}, 1 \text{ H},$ CH₂), 2.56 (dd, J = 7.2, 16.2 Hz, 1 H, CH₂), 3.20 (d, J = 10.5 Hz, 1 H, CH), 3.70 (s, 3 H, OCH₃), 3.85 (s, 3 H, OCH₃), 4.22–4.51 (m, 9 H, CH₂ and CH), 5.17-5.38 (m, 8 H, =CH₂), 5.88-6.06 (m, 4 H, =CH), 6.43 (s, 3 H, Ar), 6.54 (d, J = 8.4 Hz, 2 H, Ar), 6.56 (d, J = 8.4 Hz, 2 H, Ar), 6.65 (d, J = 8.4 Hz, 2 H, Ar), 6.70 (d, J = 8.4 Hz, 2 H, Ar), 6.82–7.01 (m, 9 H, Ar), 7.12 (d, J = 8.4 Hz, 2 H, Ar), 7.17 (d, J = 8.4 Hz, 2 H, Ar) ppm. ¹³C NMR (CDCl₃, 75 MHz, TMS): δ = 41.5, 52.1, 53.0, 55.0, 55.2, 68.1, 68.3, 68.5, 68.7, 112.0, 112.5, 113.5, 113.6, 113.9, 114.3, 117.4, 117.5, 117.6, 128.7, 129.5, 129.7, 129.9, 130.2, 130.7, 131.0, 131.2, 132.1, 133.1, 133.2, 133.3, 133.4, 136.0, 136.5, 136.7, 138.4, 143.9, 156.3, 156.5, 156.7, 157.1, 157.2, 158.0 ppm. IR (CH₂Cl₂): \tilde{v} = 3034, 2995, 2932, 2835, 2533, 2051, 1884, 1648, 1606, 1581, 1508, 1461, 1442, 1424, 1361, 1294, 1247, 1179, 1115, 1032, 997, 926, 830, 787, 740, 623, 568, 546 cm⁻¹. C₅₆H₅₄O₆ (823.04): calcd. C 81.72, H 6.61; found C 81.47, H 6.66. ESI-MS (%): $m/z = 822 [M + Na]^+$. HRMS (ESI): calcd. for $C_{55}H_{54}O_6Na^+$ 845.3801; found 845.3813.

5e: A white solid (95 mg, 57%), m.p. 188–190 °C. ¹H NMR $(CDCl_3, 400 \text{ MHz}, \text{TMS}): \delta = 2.39 \text{ (dd, } J = 10.4, 16.0 \text{ Hz}, 1 \text{ H},$ CH₂), 2.43–2.49 (m, 4 H, CH), 2.56 (dd, J = 7.6, 16.0 Hz, 1 H, CH₂), 3.22 (d, J = 10.4 Hz, 1 H, CH), 3.71 (s, 3 H, OCH₃), 3.87 (s, 3 H, OCH₃), 4.24–4.31 (m, 1 H, CH), 4.49 (d, *J* = 2.8 Hz, 2 H, CH₂), 4.54 (d, J = 2.8 Hz, 2 H, CH₂), 4.59 (d J = 2.8 Hz, 2 H, CH₂), 4.60 (d, J = 2.8 Hz, 2 H, CH₂), 6.43–6.50 (m, 4 H, Ar), 6.56 (d, J = 8.8 Hz, 2 H, Ar), 6.59 (d, J = 8.8 Hz, 2 H, Ar), 6.72 (d, J= 8.8 Hz, 2 H, Ar), 6.76 (d, J = 8.8 Hz, 2 H, Ar), 6.83 (d, J =8.8 Hz, 2 H, Ar), 6.92 (d, J = 8.8 Hz, 2 H, Ar), 6.98 (d, J = 8.8 Hz, 2 H, Ar), 7.01 (d, J = 8.8 Hz, 2 H, Ar), 7.09 (d, J = 8.8 Hz, 2 H, Ar), 7.15 (d, J = 8.8 Hz, 2 H, Ar) ppm. ¹³C NMR (CDCl₃, 100 MHz, TMS): δ = 41.6, 52.1, 53.1, 55.1, 55.2, 55.6, 55.7, 55.8, 58.4, 68.2, 75.2, 75.30, 75.33, 75.5, 78.5, 78.60, 78.67, 78.72, 112.1, 112.6, 113.8, 113.9, 114.2, 114.6, 128.8, 129.5, 129.9, 130.4, 130.7, 130.9, 131.06, 131.11, 132.1, 136.3, 136.6, 136.8, 139.0, 144.1, 155.4, 155.6, 155.9, 156.2, 157.2, 158.1 ppm. IR (CH₂Cl₂): $\tilde{v} =$ 3291, 3036, 2999, 2931, 2864, 2836, 2121, 2048, 1888, 1605, 1584, 1510, 1455, 1418, 1373, 1216, 1178, 1115, 1075, 1028, 925, 832, 739, 680, 643, 586, 543 cm⁻¹. 3(C₅₆H₄₆O₆)·CH₂Cl₂ (2526.9417): calcd. C 80.24, H 5.58; found C 80.14, H 5.70. ESI-MS (%): m/z = 837 [M + Na]⁺. HRMS (ESI): calcd. for $C_{56}H_{46}O_6Na^+$ 837.3168; found 837.3187.

Supporting Information (see also the footnote on the first page of this article): The spectroscopic data (¹H and ¹³C spectroscopic data), HRMS of the compounds shown in Tables 1–5 and Schemes 1–5, the X-ray crystal structures of **3a**, **4aB**, **5a** and **5b** along with the detailed description of experimental procedures are included in the Supporting Information.

Acknowledgments

We thank the Shanghai Municipal Committee of Science and Technology (06XD14005 and 08dj1400100-2), National Basic Research Program of China (grant no. 973-2009CB825300) and the National Natural Science Foundation of China (20872162, 20672127, 20872162, 20821002 and 20732008) for financial support.

- For recent reviews, see: a) I. Nakamura, Y. Yamamoto, *Adv. Synth. Catal.* 2002, *344*, 111–129; b) A. Brandi, S. Cicchi, F. M. Cordero, A. Goti, *Chem. Rev.* 2003, *103*, 1213–1270; c) E. Nakamura, S. Yamago, *Acc. Chem. Res.* 2002, *35*, 867–877; d) For the synthesis of MCPs, see: A. Brandi, A. Goti, *Chem. Rev.* 1998, *98*, 589–636.
- [2] Selected recent articles about transition-metal-catalyzed reactions of MCPs: a) D. H. Camacho, I. Nakamura, S. Saito, Y. Yamamoto, J. Org. Chem. 2001, 66, 270–275; b) M. Lautens, C. Meyer, A. Lorenz, J. Am. Chem. Soc. 1996, 118, 10676–10677; c) M. Lautens, Y. Ren, J. Am. Chem. Soc. 1996, 118, 9597–9605; d) S. Saito, M. Masuda, S. Komagawa, J. Am. Chem. Soc. 2004, 126, 10540–10541; e) M. Shi, B.-Y. Wang, J.-W. Huang, J. Org. Chem. 2005, 70, 5606–5610; f) I. Nakamura, H. Itagaki, Y. Yamamoto, J. Org. Chem. 1998, 63, 6458–6459; g) N. Tsukada, A. Hibuya, I. Nakamura, Y. Yamamoto, J. Matsuda, Y. Ito, J. Am. Chem. Soc. 2000, 122, 11015–11016; i) M. E. Scott, Y. Bethuel, M. Lautens, J. Am. Chem. Soc. 2007, 129, 1482–1483.
- [3] Selected recent articles about Lewis-acid-mediated reactions of MCPs: a) M. Shi, B. Xu, Org. Lett. 2002, 4, 2145–2148; b) B. Xu, M. Shi, Org. Lett. 2003, 5, 1415–1418; c) X. Huang, H.-W. Zhou, Org. Lett. 2002, 4, 4419–4422; d) J.-W. Huang, M. Shi, Tetrahedron Lett. 2003, 44, 9343–9347; e) ¹.-F. Yao, M. Shi, Org. Lett. 2007, 9, 5187–5190 and references cited herein.
- [4] a) D. H. Camacho, I. Nakamura, S. Saito, Y. Yamamoto, Angew. Chem. Int. Ed. 1999, 38, 3365–3367; b) I. Nakamura, B. H. Oh, S. Saito, Y. Yamamoto, Angew. Chem. Int. Ed. 2001, 40, 1298–1300; c) B. H. Oh, I. Nakamura, S. Saito, Y. Yamamoto, Tetrahedron Lett. 2001, 42, 6203–6205; d) B. H. Oh, I. Nakamura, S. Saito, Y. Yamamoto, Heterocycles 2003, 61, 247–257; e) S. Braese, A. de Meijere, Angew. Chem. Int. Ed. Engl. 1995, 34, 2545–2547; f) I. Nakamura, T. Nemoto, Y. Yamamoto, A. de Meijere, Angew. Chem. Int. Ed. 5176–5179.
- [5] a) M. Shi, B. Xu, J.-W. Huang, Org. Lett. 2004, 6, 1175–1178;
 b) M. Shi, L.-X. Shao, B. Xu, Org. Lett. 2003, 5, 579–582; c)
 L.-X. Shao, B. Xu, J.-W. Huang, M. Shi, Chem. Eur. J. 2006, 12, 510–517; d) J.-W. Huang, M. Shi, Synlett 2004, 2343–2346;
 e) L. Patient, M. B. Berry, J. D. Kilburn, Tetrahedron Lett. 2003, 44, 1015–1017.
- [6] a) S.-M. Ma, J.-L. Zhang, J. Am. Chem. Soc. 2003, 125, 12386–12387; b) M. Lautens, W. Han, J. Am. Chem. Soc. 2002, 124, 6312–6316; c) M. Lautens, W. Han, J. H.-C. Liu, J. Am. Chem. Soc. 2003, 125, 4028–4029; d) M. E. Scott, W. Han, M. Lautens, Org. Lett. 2004, 6, 3309–3312; e) M. E. Scott, M. Lautens, Org. Lett. 2005, 7, 3045–3047. Ring enlargement of MCPs; f) M. Shi, L. P. Liu, J. Tang, J. Am. Chem. Soc. 2006, 128, 7430–7431; g) A. Furstner, C. Aissa, J. Am. Chem. Soc. 2006, 128, 6306–6307.
- [7] a) L. P. Liu, M. Shi, J. Org. Chem. 2004, 69, 2805–2808; b)
 L. F. Yao, M. Shi, Org. Lett. 2007, 9, 5187–5189; c) M. Shi,
 M. Jiang, L.-P. Liu, Org. Biomol. Chem. 2007, 5, 438–440.
- [8] Compound 4ab was determined to be (E)-configured based on the following two reasons: 1) no interaction between H^a and H^b was detected from the NOE spectroscopy and 2) the interactions between H^a and the aromatic ring and between H^b and the aromatic ring could be recognized by NOE spectroscopy.



- [9] CCDC-675136 (for 3a), -712669 (for 4aB), -684691 (for 5a), -720536 (for 5b) contain the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.
- [10] a) A. W. Schrecker, L. Y. Greenberg, L. L. Hartwell, J. Am. Chem. Soc. 1952, 74, 5669–5671; b) E. L. Jackson, L. Pastiut,

J. Am. Chem. Soc. **1928**, 50, 2249–2260; c) D. J. Cram, J. Am. Chem. Soc. **1949**, 71, 3863–3870; d) D. J. Cram, J. Am. Chem. Soc. **1952**, 74, 2129–2137.

Received: May 18, 2009 Published Online: September 1, 2009