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Quinoline-Fused Lactones via Tandem Oxidation Cyclization: Metal-Free sp³ C–H Functionalization

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ABSTRACT: A unique lactonization of 2-methyl-3-acyl-4-phenylquinolines using PhIO as the oxidant and selectfluor as an additive is reported. The reaction occurs under ambient conditions through tandem oxidation and cyclization of sp^3 C–H bonds under metal-free conditions. The heterocycle-fused lactones are obtained in moderate to good yield.

INTRODUCTION

Aryl- and heteroaryl-fused lactones are useful scaffolds that display a range of biological properties.¹ Generally, the synthesis of these molecules is carried out in multiple steps using metal, ligand, or acid. Scheme 1A(i-viii) enlists some of the classical prefunctionalized starting materials that enable access to benzolactones. The methodologies involve (i) metalmediated oxidative lactonization of benzene-1,2-dimethanol,² (ii) Tishchenko reaction of phthalaldehyde with actinide coordination complexes,³ (iii) Zn/ligand-mediated lactonization of 2-formylbenzoates,⁴ and (iv) lactonization of 2-(halomethyl)benzoic acid.⁵ Alternatively, benzolactone formation has also been achieved via metal-catalyzed and directing group-assisted C-H activation/C-O bond formation in (v) *o*-toluic acid^{6a,b} and aliphatic carboxylic acid,^{6c} (vi) (Z)-2-(3-hydroxyprop-1-en-1-yl)benzoic acid,⁷ (vii) γ -(2methoxycarbonylphenyl)propargylic alcohol,⁸ and (viii) benzyl alcohol and CO₂.⁹ Pyridine-fused lactones (Scheme 1B) have been synthesized from pyridine-2,3-dimethanol or pyridine-3,4-dimethanol in the presence of metal and ligand (i),¹⁰ while quinoline-fused lactones have been prepared from 2-aminobenzophenone in multiple steps using metal and/or acid at high temperature (ii).¹¹ In general, most of these methods suffer from limitations of prefunctionalized substrates, harsh reagents, multiple steps, high temperature, and toxicity. Therefore, seeking efficient, atom economical, and mild strategies for synthesizing heteroaryl-fused lactones is highly desirable. Direct sp³ C-H functionalization has emerged as a powerful tool for construction of C-C, C-N, and C-O bonds.¹² Because of the inertness of the C(sp³)-H bond, its late-stage functionalization is challenging, though attractive. Recent approaches for C(sp³)-H functionalization describe the use of metal/oxidant via ionic pathway/free-radical

process¹³ or oxidative strategy using hypervalent iodine reagents.¹⁴ As shown in (Scheme 1B(ii)), quinoline lactones serve as key synthetic intermediates for aza-podophyllotoxin derivatives. 1,4-Dihydroquinoline lactones (I), derived by reduction of 9-phenylfuro[3,4-*b*]quinolin-1(3*H*)-one, are known for their anti-leishmanial and anti-leukemic properties. Another biologically important molecule, 3-fluoromethylquino-line-2-carboxamide ([¹⁸F]-AB5186) (II), known as a positron emission tomography imaging agent can be derived from 9-phenylfuro[3,4-*b*]quinolin-1(3*H*)-one.¹⁵

In view of the immense biological significance of heteroarylfused lactones and limited synthetic procedures, we delved into hypervalent iodine-assisted protocols to access them. In this work, we report an unprecedented synthesis of quinoline and quinoxaline lactones via oxidative cyclization of sp³ C–H bonds using PhIO (iodosobenzene) and selectfluor under room temperature (RT) conditions (Scheme 1C).

RESULTS AND DISCUSSION

The study initiated with the reaction of 1-(2-methyl-6-nitro-4phenylquinolin-3-yl)ethan-1-one (1a) using PhIO and selectfluor (1:1) at RT. After 24 h, a cyclic product, 7-nitro-9phenylfuro[3,4-*b*]quinolin-1(3*H*)-one (2a) was isolated in 31% yield along with the formation of 7-nitro-5-phenyl-1*H*pyrano[3,4-*b*]quinolin-4(3*H*)-one (2a') in 17% yield (Table 1,



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Article

Scheme 1. Synthetic Routes for Benzolactone and Heteroaryl-Fused Lactones



Table 1. Optimization Table^a

	O_2N $Hypervalent iodine O_2N$ $Ph O O_2N$ $Ph O O_2N$ $Hypervalent iodine O_2N$ $Hypervalent$			
	1a	2a	2a'	
s. no	hypervalent iodine	additive	solvent	yield ^b (%)
1	PhIO (1)	selectfluor (1)	CH ₃ CN	31, 17 ^c
2	PhIO (1)	selectfluor (2)	CH ₃ CN	40, 13 [°]
3	PhIO (2)	selectfluor (1)	CH ₃ CN	58, 7 ^c
4	PhIO (3)	selectfluor (2)	CH ₃ CN	86
5	PhIO (3)	selectfluor (2)	CH ₃ CN	69 ^d , 56 ^e
6	DPIB (3)	selectfluor (2)	CH ₃ CN	67
7	PIDA (3)	selectfluor (2)	CH ₃ CN	61
8	PIFA (3)	selectfluor (2)	CH ₃ CN	53
9	HTIB (3)	selectfluor (2)	CH ₃ CN	49
10	PhIO (3)	PTSA/BF ₃ ·OEt/NFSI (2)	CH ₃ CN	51/53/8
11	PhIO(3)	selectfluor (2)	solvent ^{f,g,h}	0

°C. ^fDCE. ^gDioxane. ^hTHF.

entry 1). To improve the yield, optimization studies were carried out (Table 1, entries 2-4). On changing the stoichiometry of PhIO and selectfluor to 1:2 (Table 1, entry 2), the yield of 2a increased slightly (40%) and 2a' was formed in 13% yield (Table 1, entry 2). With a 2:1 stoichiometry, 2a increased to 58% and 2a' got reduced to 7% (Table 1, entry 3). Changing the ratio to 3:2 helped in driving the reaction to completion, and 2a was isolated in 86% yield (Table 1, entry

4). Raising the temperature to 50 and 80 $^{\circ}$ C reduced the yield to 69 and 56%, respectively, because of the formation of unknown polar products (Table 1, entry 5). Use of other hypervalent iodine reagents such as dipivaloyloxyiodobenzene (DPIB), phenyliodine(III) diacetate (PIDA), [bis-(trifluoroacetoxy)iodo]benzene (PIFA), and hydroxy-(tosyloxy)iodobenzene (HTIB) did not help, and lower yields were obtained in all cases (Table 1, entries 6–9).

Article

Scheme 2. Substrate Scope^a



"Reaction conditions: 1 (1.0 equiv), PhIO (3.0 equiv), selectfluor (2.0 equiv), CH₃CN, RT, 24 h. Yields are isolated yields. ^bReaction carried on a gram scale with 1 g of 1a.

Scheme 3. Control Experiments for Mechanistic Investigation



On substituting selectfluor with N-fluorobenzenesulfonimide, the yield of **2a** dropped considerably (8%). With acid additives such as p-toluenesulfonic acid (PTSA) and $BF_3 \cdot OEt_2$, **2a** was obtained in 51 and 53% yields, respectively (Table 1, entry 10). No reaction took place in dichloroethene (DCE), dioxane, and tetrahydrofuran (THF) (Table 1, entry 11), and acetonitrile was found to be the best solvent for the reaction.

With the optimized reaction conditions in hand, the generality of the reaction was examined. Quinoline derivatives

with electron-withdrawing substituents such as NO₂, F, Cl, and Br at 5, 6, and 8 positions afforded the corresponding lactones in 61–86% yields (Scheme 2, entries 2a-2d). However, unsubstituted and 7-methyl-substituted quinolines gave lower yield (Scheme 2, entries 2e and 2f), suggesting higher reactivity of methyl sp³ C–H under the influence of electron-withdrawing groups. Next, the effect of substituents on 4-phenyl ring was examined. F, Cl, Br, and CF₃ groups at the para position showed good compatibility and afforded the

Scheme 4. Proposed Mechanism



corresponding products in moderate yields (Scheme 2, entries 2g-2j). Lower yield was obtained with electron-donating *p*-methoxy and *p*-benzyloxy substituted derivatives as a substantial amount of starting material was left unreacted (Scheme 2, entries 2k-2l). Next, disubstituted derivates with halo-/nitrosubstituents on both the phenyl rings were explored, and moderate-to-good product yields were obtained (Scheme 2, entries 2m-2s). Interestingly, with 1-(3-methyl-quinoxalin-2-yl)ethanone, the desired lactone furo[3,4-*b*]-quinoxalin-1(3*H*)-one (2t) was obtained in 39% yield with considerable amount of unreacted 1t.

Notably, it was found that the desired reaction did not take place with 1-(2-methylquinolin-3-yl)ethan-1-one (1u) and 1-(2,4-dimethylquinolin-3-yl)ethan-1-one (1v) (Scheme 2), and a mixture of polar products was formed in either case suggesting the importance of electron-withdrawing effect of 4-phenyl group in the reaction. The results indicated that the electron-withdrawing effect played a dominant role in activating the sp³-C-H bond and enabling lactonization. Reaction was carried out on a gram scale, and 2a was isolated in 67% yield (670 mg) under optimized reaction conditions (Scheme 2).

To gain insight into the reaction mechanism, some control experiments were carried out. The role of PhIO and selectfluor was studied by carrying out the reaction of 1a in the presence of PhIO alone (Scheme 3(i)) and then in the presence of selectfluor alone (Scheme 3(ii)). In both the cases, 2a was not formed. Interestingly, with selectfluor alone, fluorination of 2methyl group took place, and 1-(2-(fluoromethyl)-6-nitro-4phenylquinolin-3-yl)ethanone (3a) was obtained in 30% yield (Scheme 3(ii)). Few other substituted quinolines were also subjected to the reaction with selectfluor. However, small amounts of fluorinated products were obtained in all cases (Scheme 3(ii), 3a-3d).¹⁶ The involvement of methyl free radical in the reaction was ascertained by carrying out the reaction in the presence of 2,2,6,6-tetramethylpiperidine-Noxyl (TEMPO) (2 equiv). It was found that the desired lactone 2a did not form under these conditions, and a TEMPO adduct 2a" was isolated in 69% yield (Scheme 3(iii)).

Based on the experimental results, a reaction mechanism has been proposed as shown in Scheme 4. The reaction is believed to initiate via in situ generation of fluorine radical and radical cation intermediate [A] by the reaction of la with selectfluor.^{17a} [A] reacts with fluorine radical and forms intermediate [B], which subsequently loses the fluorine radical to generate a methylene radical intermediate [C] at the C-2 position. The formation of $[\mathbf{C}]$ is supported by the formation of 2a" (TEMPO adduct) and 3a (Scheme 3(ii)). [C] undergoes intramolecular cyclization by radical coupling with carbonyl oxygen and yields intermediate [E] (m/z 305.0936,Figure S1, Supporting Information) via [D]. [E] undergoes ring opening reaction with protonated PhIO, and the cationic site is attacked by the amine nucleophile derived from selectfluor^{17b} to form [F]. It is noteworthy to mention that selectfluor plays a dual role. First, it activates the substrate (1a-[A]), and second the diazoniabicyclo [2.2.2] octane moeity acts as a nucleophile and facilitates the ring-opening step ([E] -[F]).^{17c} Further, [F] undergoes reductive elimination of hypervalent iodine followed by hydroxide attack resulting in the formation of [G]. On reaction with protonated PhIO, [G]forms [H], while in it's absence, 2a' is formed as a side product (Table 1). [H] undergoes reductive elimination of hypervalent iodine and loss of formaldehyde to give [I], which eventually yields 2a after elimination of ammonia.

CONCLUSIONS

In conclusion, we report an unprecedented, metal-free sp³ C– H functionalization of methyl group in 2-methyl-3-acyl-4phenylquinolines. The reaction occurs under ambient conditions *via* oxidative cyclization using PhIO and selectfluor. The protocol offers metal- and ligand-free conditions, smooth scalability, and good functional group tolerance and provides quinoline-fused lactones in moderate to good yields.

EXPERIMENTAL SECTION

General Information. All reagents and solvents were of pure analytical grade. Thin-layer chromatography (TLC) was performed on 60 F254 silica gel, precoated on aluminum plates, and revealed with a UV lamp ($\lambda_{max} = 254$ nm). The products were purified by

column chromatography on silica gel 230–400 mesh. ¹H and ¹³C NMR spectra were recorded on 300 MHz (¹H 300 MHz, ¹³C 75 MHz), 400 MHz (¹H 400 MHz, ¹³C 100 MHz, ¹⁹F 376 MHz,) and 500 MHz spectrometers (¹H 500 MHz, ¹³C 125 MHz) using CDCl₃ as the solvent with tetramethylsilane (TMS) as the internal standard at RT. Chemical shifts are in δ (ppm) relative to TMS. The coupling constants (*J*) are in Hz. High-resolution mass spectra (HRMS) were recorded on a mass spectrometer using electrospray ionization time-of-flight (ESI-TOF) reflectron experiments.

General Procedure for Synthesis of Compounds (1a-1s). A mixture of 2-aminoaryl ketones (0.50 mmol, 1 equiv), β -ketoesters/ketones (1.0 mmol, 2 equiv), and 4-toluenesulfonic acid (0.10 mmol, 0.2 equiv) in EtOH (3 mL) was added into a Schlenk flask (25 mL), and the mixture was stirred at 80 °C using oil bath and stirred until the reaction was complete. The solvent was evaporated under reduced pressure, and the residue was purified by column chromatography on silica gel using ethyl acetate-hexane.^{18a} Isolated yields: 1a^{18c} (97%), 1b^{18c} (93%), 1c (92)%, 1d (89)%, 1e^{18a} (99%), 1f (87%), 1g^{18c} (77%), 1h^{18a} (81%), 1i^{18c} (80%), 1j (81%), 1k (85%), 11^{18c} (86%), 1m (71%), 1n^{18a} (75%), 1o^{18a} (61%), 1p^{18a} (72%), 1q (69%), 1r (70%), 1s (71%).

General Procedure for Synthesis of Compound 1t. To a roundbottom flask containing water (15 mL), acetyl acetone (185 mg, 1 mmol) was added followed by N-bromo succinimide (393 mg, 1.2 mmol) and stirred for 20 min at 70 °C using oil bath. To this reaction mixture, 1,2-phenylene diamine (200 mg, 1.0 mmol) was added, and the contents were stirred until completion of the reaction as indicated by TLC.^{18b} The reaction mixture was then cooled to RT and poured onto ice and extracted with ethyl acetate. The organic extract was washed successively with brine, dried over Na₂SO₄, and concentrated under reduced pressure. The product was purified by column chromatography on silica gel using ethyl acetate-hexane.

General Procedure for Synthesis of Lactone-Fused Quinoline and Quinoxaline (2a-2t). To a 25 mL round-bottom flask was added 1a (60 mg, 1 equiv), PhIO (128 mg, 3 equiv), selectfluor (138 mg, 2 equiv), and CH₃CN (4 mL). The reaction mixture was stirred for 24 h at RT, and the reaction was monitored through TLC. After reaction completion, the contents were diluted with water, extracted with ethyl acetate, and concentrated at reduced pressure. The product was purified by column chromatography on silica gel using ethyl acetate-hexane.

General Procedure for Synthesis of Fluorinated Compounds 3a-3d. To a 25 mL round-bottom flask was added 1a (60 mg, 1 equiv), selectfluor (138 mg, 2 equiv), and CH₃CN (3 mL). The reaction mixture was stirred for 24 h at RT, and the reaction was monitored through TLC. After reaction completion, the contents were diluted with water, extracted with ethyl acetate, and concentrated at reduced pressure. The product was purified by column chromatography on silica gel using ethyl acetate-hexane.

General Procedure for Preparation of TEMPO Adduct 2a". To a 25 mL round-bottom flask was added 1a (60 mg, 1 equiv), PhIO (128 mg, 3 equiv), selectfluor (138 mg, 2 equiv), TEMPO (61.1 mg, 2 equiv), and CH₃CN (3 mL). The reaction mixture was stirred for 24 h at RT, and the reaction was monitored through TLC. After reaction completion, the contents were diluted with water, extracted with ethyl acetate, and concentrated at reduced pressure. The product was purified by column chromatography on silica gel using ethyl acetate-hexane.

Physical Properties and Characterization Data of the Synthesized Compounds. *1-(5-Fluoro-2-methyl-4-phenylquino-lin-3-yl)ethan-1-one* (*1c*). White solid, yield 92% (128.34 mg), hexane/EtOAc (92:8); ¹H NMR (400 MHz, CDCl₃): δ 7.89 (d, *J* = 8.48 Hz, 1H), 7.64–7.58 (m, 1H), 7.43–7.42 (m, 3H), 7.32–7.31 (m, 2H), 7.08–7.03 (m, 1H), 2.66 (s, 3H), 1.95 (s, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃): δ 205.1, 158.6 (d, ¹*J*_{C-F} = 257.90 Hz), 154.1, 148.7, 141.0, 137.0 (d, ⁴*J*_{C-F} = 3.54 Hz), 136.5, 129.7 (d, ³*J*_{C-F} = 9.58 Hz), 129.0 (d, ⁴*J*_{C-F} = 9.15 Hz), 118.8 (d, ²*J*_{C-F} = 21.65 Hz), 31.8, 23.7; HRMS (ESI-TOF) *m/z*: [M + H]⁺ calcd for C₁₈H₁₅FNO⁺, 280.1132; found, 280.1080.

Article

1-(8-Bromo-6-chloro-2-methyl-4-phenylquinolin-3-yl)ethan-1one (1d). White solid, yield 89% (165.5 mg), hexane/EtOAc (92:7); ¹H NMR (500 MHz, CDCl₃): δ 8.02 (d, J = 2.25 Hz, 1H), 7.54–7.53 (m, 4H), 7.32–7.30 (m, 2H), 2.73 (s, 3H), 1.99 (s, 3H); ¹³C{¹H} NMR (125 MHz, CDCl₃): δ 204.7, 155.0, 143.5, 143.1, 136.2, 134.1, 133.9, 132.0, 129.9, 129.4, 129.0, 126.7, 125.6, 124.8, 31.7, 24.1; HRMS (ESI-TOF) m/z: [M + H]⁺ calcd for C₁₈H₁₄ BrClNO⁺, 373.9942; found, 373.9872.

1-(2,7-Dimethyl-4-phenylquinolin-3-yl)ethan-1-one (**1f**). White solid, yield 87% (119.6 mg), hexane/EtOAc (93:7); ¹H NMR (500 MHz, CDCl₃): δ 7.86 (s, 1H), 7.51–7.49 (m, 4H), 7.35–7.33 (m, 2H), 7.283–7.280 (m, 1H), 2.67 (s, 3H), 2.55 (s, 3H), 1.98 (s, 3H); ¹³C{¹H} NMR (125 MHz, CDCl₃): δ 205.9, 153.3, 147.7, 143.8, 140.6, 135.4, 134.1, 130.0, 128.8, 128.7, 128.6, 127.9, 125.8, 123.0, 32.0, 23.8, 21.7; HRMS (ESI-TOF) *m*/*z*: [M + H]⁺ calcd for C₁₉H₁₈NO⁺, 276.1383; found, 276.1392.

1-(2-Methyl-4-(4-(trifluoromethyl)phenyl)quinolin-3-yl)ethan-1one (1j). White solid, yield 81% (133.2 mg), hexane/EtOAc (92:8); ¹H NMR (500 MHz, CDCl₃): δ 8.09 (d, *J* = 8.30 Hz, 1H), 7.79 (d, *J* = 8.10 Hz, 2H), 7.76–7.73 (m, 1H), 7.51–7.46 (m, 4H), 2.71 (s, 3H), 2.06 (s, 3H); ¹³C{¹H} NMR (125 MHz, CDCl₃): δ 205.1, 153.4, 147.5, 142.1, 139.0, 134.9, 130.5, 130.3, 129.1, 126.9, 125.7, 125.6 (2C), 124.5, 123.8 (q, ¹*J*_{C-F} = 270.62 Hz), 32.2, 23.8; HRMS (ESI-TOF) *m/z*: [M + H]⁺; calcd for C₁₉H₁₅F₃NO⁺, 330.1100; found, 330.1036.

1-(4-(4-(Benzyloxy)phenyl)-2-methylquinolin-3-yl)ethan-1-one (1k). White solid, yield 85% (155.9 mg), hexane/EtOAc (93:7); ¹H NMR (400 MHz, CDCl₃): δ 8.07 (d, J = 8.48 Hz, 1H), 7.70–7.67 (m, 2H), 7.47–7.44 (m, 2H), 7.42–7.38 (m, 3H), 7.35–7.33 (m, 1H), 7.29–7.26 (m, 2H), 7.12–7.10 (m, 2H) 5.12 (s, 2H), 2.69 (s, 3H), 2.00 (m, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃): δ 205.9, 159.3, 153.3, 147.5, 143.7, 136.5, 134.9, 131.4, 130.0, 128.8, 128.6, 128.1, 127.5, 127.4, 126.4, 126.1, 125.3, 115.1, 70.1, 31.9, 23.8; HRMS (ESI-TOF) m/z: [M + H]⁺ calcd for C₂₅H₂₂NO₂⁺, 368.1645; found, 368.1647.

1-(4-(3-Chloro-4-fluorophenyl)-2-methylquinolin-3-yl)ethan-1one (1m). White solid, yield 71% (111.1 mg), hexane/EtOAc (93:7); ¹H NMR (500 MHz, CDCl₃): δ 8.10 (d, *J* = 8.40 Hz, 1H), 7.77–7.74 (m, 1H), 7.55–7.54 (m, 1H), 7.51–7.48 (m, 1H), 7.46–7.45 (m, 1H), 7.33–7.30 (m, 1H), 7.28–7.25 (m, 1H), 2.71 (s, 3H), 2.13 (s, 3H); ¹³C{¹H} NMR (125 MHz, CDCl₃): δ 205.0, 158.4 (d, ¹*J*_{C-F} = 250.48 Hz), 153.4, 147.5, 141.2, 135.0, 132.18, 132.12, 130.3, 130.0 (d, ³*J*_{C-F} = 7.22 Hz), 129.1, 127.0, 125.5, 124.7, 121.7 (d, ²*J*_{C-F} = 18.18 Hz), 117.1 (d, ²*J*_{C-F} = 21.05 Hz), 32.2, 23.8; HRMS (ESI-TOF) *m*/*z*: [M + H]⁺ calcd for C₁₈H₁₄ClFNO⁺, 314.0742; found, 314.0680.

1-(5-Fluoro-4-(4-fluorophenyl)-2-methylquinolin-3-yl)ethan-1one (1q). White solid, yield 69% (104.6 mg), hexane/EtOAc (92:8); ¹H NMR (500 MHz, CDCl₃): δ 7.89 (d, *J* = 8.45 Hz, 1H), 7.67–7.63 (m, 1H), 7.31–7.28 (m, 2H), 7.15–7.12 (m, 2H), 7.10–7.08 (m, 1H), 2.65 (s, 3H), 1.99 (s, 3H); ¹³C{¹H} NMR (125 MHz, CDCl₃): δ 205.1, 162.8 (d, ¹*J*_{C-F} = 247.05 Hz), 158.6 (d, ¹*J*_{C-F} = 257.52 Hz), 154.1, 148.8, 140.0, 136.8, 132.9, 130.9 (d, ⁴*J*_{C-F} = 3.60 Hz), 130.8 (d, ⁴*J*_{C-F} = 3.55 Hz), 129.9 (d, ³*J*_{C-F} = 9.57 Hz), 125.3 (d, ⁴*J*_{C-F} = 4.37 Hz), 115.1 (d, ²*J*_{C-F} = 26.60 Hz), 112.0 (d, ²*J*_{C-F} = 21.71 Hz), 32.0, 23.7; HRMS (ESI-TOF) *m*/*z*: [M + H]⁺ calcd for C₁₈H₁₄F₂NO⁺, 298.1038; found, 298.1042.

1-(4-(4-Bromophenyl)-5-fluoro-2-methylquinolin-3-yl)ethan-1one (1r). White solid, yield 70% (124.9 mg), hexane/EtOAc (93:7); ¹H NMR (500 MHz, CDCl₃): δ 7.88 (d, *J* = 8.50 Hz, 1H), 7.64–7.60 (m, 1H), 7.55 (d, *J* = 8.20 Hz, 2H), 7.18 (d, *J* = 8.10 Hz, 2H), 7.09– 7.05 (m, 1H), 2.63 (s, 3H), 2.00 (s, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃): δ 204.9, 158.5 (d, ¹*J*_{C-F} = 257.50 Hz), 154.1, 148.7, 139.6, 136.5, 135.9, 131.1, 130.6 (d, ⁴*J*_{C-F} = 3.55 Hz), 129.9 (d, ³*J*_{C-F} = 9.55 Hz), 125.3 (d, ⁴*J*_{C-F} = 4.18 Hz), 122.9, 115.1 (d, ³*J*_{C-F} = 9.13 Hz), 112.0 (d, ²*J*_{C-F} = 21.71 Hz), 32.1, 23.6; HRMS (ESI-TOF) *m/z*: [M + H]⁺ calcd for C₁₈H₁₄BrFNO⁺, 358.0237; found, 358.0238.

1-(4-(4-Chlorophenyl)-2-methyl-6-nitroquinolin-3-yl)ethan-1one (1s). White solid, yield 71% (120.7 mg), hexane/EtOAc (93:7); ¹H NMR (500 MHz, CDCl₃): δ 8.51–8.48 (m, 2H), 8.20 (d, J = 9.05

Hz, 1H), 7.57 (d, *J* = 8.35 Hz, 2H), 7.31 (d, *J* = 8.35 Hz, 2H), 2.73 (s, 3H), 2.07 (s, 3H); $^{13}C{}^{1}H$ NMR (125 MHz, CDCl₃): δ 204.2, 157.8, 149.5, 145.7, 144.1, 136.4, 136.3, 131.9, 131.2, 130.9, 129.6, 124.1, 123.7, 122.7, 31.9, 24.2; Anal. Calcd for C₁₈H₁₃ClN₂ O₃: C, 63.45; H, 3.85; N, 8.22. Found: C, 61.61; H, 3.52; N, 7.62%.

7-Nitro-9-phenylfuro[*3*,*4-b*]*quinolin-1(3H)-one* (*2a*). White solid, yield 86% (51.5 mg), hexane/EtOAc (83:17); ¹H NMR (300 MHz, CDCl₃): δ 8.86 (d, *J* = 2.52 Hz, 1H), 8.64 (dd, *J*₁ = 9.21 Hz, *J*₂ = 2.40 Hz, 1H), 8.35 (d, *J* = 9.30 Hz, 1H), 7.68–7.64 (m, 3H), 7.50–7.47 (m, 2H), 5.51 (s, 2H); ¹³C{¹H} NMR (100 MHz, CDCl₃): δ 167.1, 166.7, 153.6, 153.1, 146.1, 131.3, 130.5, 130.0, 129.9, 128.7, 126.3, 125.7, 124.9, 115.1, 69.6; HRMS (ESI-TOF) *m*/*z*: [M + H]⁺ calcd for C₁₇H₁₁N₂O₄⁺, 307.0713; found, 307.0714.

7-*Chloro-9-phenylfuro*[3,4-*b*]*quinolin-1*(3*H*)-one (**2b**).^{11e} White solid, yield 61% (36.6 mg), hexane/EtOAc (92:8); ¹H NMR (500 MHz, CDCl₃): δ 8.14 (d, *J* = 8.95 Hz, 1H), 7.86 (d, *J* = 2.25 Hz, 1H), 7.83 (dd, *J*₁ = 8.95 Hz, *J*₂ = 2.35 Hz, 1H), 7.60–7.59 (m, 3H), 7.44–7.42 (m, 2H), 5.44 (s, 2H); ¹³C{¹H} NMR (125 MHz, CDCl₃): δ 167.4, 163.8, 150.5, 149.6, 133.6, 133.4, 130.9, 130.9, 129.8, 129.7, 128.4, 127.8, 126.6, 114.2, 69.5; HRMS (ESI-TOF) *m*/*z*: [M + Na]⁺ calcd for C₁₇H₁₀ClNNaO₂⁺, 318.0292; found, 318.0303.

8-*Fluoro-9-phenylfuro*[3,4-*b*]*quinolin-1(3H)-one* (2*c*). White solid, yield 69% (41.3 mg), hexane/EtOAc (91:9); ¹H NMR (500 MHz, CDCl₃): δ 8.06 (d, *J* = 8.50 Hz, 1H), 7.88–7.83 (m, 1H), 7.53–7.52 (m, 3H), 7.39–7.38 (m, 2H), 7.27–7.28 (m, 1H), 5.43 (s, 2H); ¹³C{¹H} NMR (125 MHz, CDCl₃): δ 167.0, 164.1, 160.07 (d, ¹*J*_{C-F} = 260.37 Hz), 152.2, 149.7, 134.0 (d, ⁴*J*_{C-F} = 4.56 Hz), 132.7 (d, ³*J*_{C-F} = 9.67 Hz), 128.9, 127.9 (d, ⁴*J*_{C-F} = 3.38 Hz), 127.6, 125.7 (d, ⁴*J*_{C-F} = 4.55 Hz), 117.8 (d, ³*J*_{C-F} = 9.27 Hz), 115.0, 112.9 (d, ²*J*_{C-F} = 21.67 Hz), 69.1; HRMS (ESI-TOF) *m*/*z*: [M + Na]⁺ calcd for C₁₇H₁₀FNNaO₂⁺, 302.0587; found, 302.0594.

5-Bromo-7-chloro-9-phenylfuro[3,4-b]quinolin-1(3H)-one (2d). White solid, yield 65% (39 mg), hexane/EtOAc (92:8); ¹H NMR (500 MHz, CDCl₃): δ 8.20 (d, J = 2.25 Hz, 1H), 7.83 (d, J = 2.25 Hz, 1H), 7.61 (d, J = 1.85 Hz, 2H), 7.60 (d, J = 2.35 Hz, 1H), 7.42 (d, J = 1.85 Hz, 1H), 7.41–7.40 (m, 1H), 5.52 (s, 2H); ¹³C{¹H} NMR (125 MHz, CDCl₃): δ 166.8, 164.5, 151.2, 146.9, 136.4, 133.2, 130.6, 130.0, 129.6, 128.8, 128.5, 126.4, 125.7, 115.0, 69.7; HRMS (ESI-TOF) m/z: [M + H]⁺ calcd for C₁₇H₁₀BrClNO₂⁺, 373.9577; found, 373.9588.

9-Phenylfuro[3,4-b]quinolin-1(3H)-one (2e)¹¹. White solid, yield 53% (31.7 mg), hexane/EtOAc (93:7); ¹H NMR (400 MHz, CDCl₃): δ 8.21 (d, J = 8.24 Hz, 1H), 7.93–7.89 (m, 2H), 7.61–7.57 (m, 4H), 7.47–7.45 (m, 2H), 5.46 (s, 2H); ¹³C{¹H} NMR (125 MHz, CDCl₃): δ 167.8, 163.6, 151.4, 151.3, 132.5, 131.6, 129.8, 129.5, 129.3, 128.2, 128.0, 127.4, 127.0, 113.5, 69.5; HRMS (ESI-TOF) *m*/*z*: [M + Na]⁺ calcd for C₁₇H₁₁NNaO₂⁺, 284.0681; found, 284.0680.

6-Methyl-9-phenylfuro[3,4-b]quinolin-1(3H)-one (**2f**). White solid, yield 51% (30.6 mg), hexane/EtOAc (93:7); ¹H NMR (400 MHz, CDCl₃): δ 7.97 (s, 1H), 7.79 (d, J = 8.68 Hz, 1H), 7.59–7.56 (m, 3H), 7.45–7.44 (m, 2H), 7.42–7.40 (m, 1H) 5.43 (s, 2H), 2.62 (s, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃): δ 168.1, 163.8, 151.6, 151.2, 143.7, 131.8, 129.8, 129.7, 129.4, 128.3, 128.2, 127.7, 125.1, 112.7, 69.8, 22.0 HRMS (ESI-TOF) m/z: [M + H]⁺ calcd for C₁₈H₁₄NO₂⁺, 276.1019; found, 276.1011.

9-(4-fluorophenyl)furo[3,4-b]quinolin-1(3H)-one (**2g**).^{11e} White solid, yield 61% (36.6 mg), hexane/EtOAc (92:8); ¹H NMR (400 MHz, CDCl₃): δ 8.22 (d, J = 8.40 Hz, 1H), 7.94–7.89 (m, 2H), 7.63–7.59 (m, 1H), 7.48–7.44 (m, 2H), 7.30–7.29 (m, 2H), 5.46 (s, 2H); ¹³C{¹H} NMR (125 MHz, CDCl₃): δ 167.8, 163.5 (d, ¹J_{C-F} = 248.26 Hz), 163.6, 151.3, 150.3, 132.6, 131.8 (d, ³J_{C-F} = 8.45 Hz), 129.5, 127.64 (d, ²J_{C-F} = 20.37 Hz) (2C), 127.01, 115.55 (d, ²J_{C-F} = 21.76 Hz) (2C), 113.57, 69.5; HRMS (ESI-TOF) *m/z*: [M + H]⁺ calcd for C₁₇H₁₁FNO₂⁺, 280.0768; found, 280.0770.

9-(4-Chlorophenyl)furo[3,4-b]quinolin-1(3H)-one (2h). White solid, yield 59% (35.4 mg), hexane/EtOAc (92:8); ¹H NMR (500 MHz, CDCl₃): δ 8.21 (d, J = 8.45 Hz, 1H), 7.93–7.87 (m, 2H), 7.62–7.59 (m, 2H), 7.56 (d, J = 8.45 Hz, 1H), 7.40 (d, J = 8.40 Hz, 2H), 5.45 (s, 2H); ¹³C{¹H} NMR (125 MHz, CDCl₃): δ 167.8,

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163.5, 151.3, 150.0, 135.9, 132.7, 131.2, 129.9, 129.5, 128.6, 127.7, 127.6, 126.7, 113.4, 69.6; HRMS (ESI-TOF) m/z: [M + Na]⁺ calcd for C₁₇H₁₀ClNNaO₂⁺, 318.0292; found, 318.0275.

9-(4-Bromophenyl)furo[3,4-b]quinolin-1(3H)-one (2i). White solid, yield 57% (34.2 mg), hexane/EtOAc (92:8); ¹H NMR (500 MHz, CDCl₃): δ 8.22 (d, J = 8.52 Hz, 1H), 7.94–7.87 (m, 2H), 7.73–7.71 (d, J = 6.4 Hz, 2H), 7.63–7.59 (d, J = 6.2 Hz, 1H), 7.36–7.33 (m, 2H), 5.46 (s, 2H); ¹³C{¹H} NMR (125 MHz, CDCl₃): δ 167.8, 163.5, 151.3, 149.9, 132.7, 131.6, 131.4, 130.4, 129.5, 127.6 (2C), 126.7, 124.1, 113.4, 69.6; HRMS (ESI-TOF) m/z: [M + H]⁺ calcd for C₁₇H₁₁BrNO₂⁺, 339.9967; found, 339.9961.

9-(4-(*Trifluoromethyl*)phenyl)furo[3,4-b]quinolin-1(3H)-one (**2***j*). White solid, yield 60% (36.0 mg), hexane/EtOAc (92:8); ¹H NMR (500 MHz, CDCl₃): δ 8.24 (d, *J* = 8.45 Hz, 1H), 7.95–7.91 (m, 1H), 7.84 (d, *J* = 8.0 Hz, 2H), 7.81 (d, *J* = 8.5 Hz, 1H), 7.63 (d, *J* = 8.05 Hz, 1H), 7.59 (d, *J* = 7.95 Hz, 2H), 5.47 (s, 2H); ¹³C{¹H} NMR (125 MHz, CDCl₃): δ 167.7, 163.4, 151.3, 149.4, 135.3, 132.8, 131.5 (d, ²*J*_{C-F} = 32.5 Hz), 130.2, 129.6, 127.9, 127.4, 126.5, 125.3 (d, ⁴*J*_{C-F} = 3.76 Hz), 123.94 (d, ¹*J*_{C-F} = 270.62 Hz), 113.5, 69.7; HRMS (ESITOF) *m*/*z*: [M + Na]⁺ calcd for C₁₈H₁₀F₃NNaO₂⁺, 352.0555; found, 352.0548.

9-(4-(Benzyloxy)phenyl)furo[3,4-b]quinolin-1(3H)-one (2k). White solid, yield 52% (31.2 mg), hexane/EtOAc (93:7); ¹H NMR (500 MHz, CDCl₃): δ 8.19 (d, J = 7.40 Hz, 1H), 8.00 (d, J = 7.20 Hz, 1H), 7.91–7.89 (m, 1H), 7.61–7.57 (m, 1H), 7.51–7.49 (m, 2H), 7.44–7.43 (m, 4H), 7.38–7.35 (m, 1H) 7.19–7.17 (m, 2H), 5.44 (s, 2H), 5.17 (s, 2H); ¹³C{¹H} NMR (125 MHz, CDCl₃): δ 168.1, 163.7, 160.0, 151.5, 151.3, 136.6, 132.4, 131.7, 129.3, 128.7, 128.6, 128.19, 128.17, 127.6, 127.2, 123.8, 114.6, 113.3, 70.2, 69.4; HRMS (ESI-TOF) *m*/*z*: [M + Na]⁺ calcd for C₂₄H₁₇NNaO₃⁺, 390.1106; found, 390.1081.

9-(4-Methoxyphenyl)furo[3,4-b]quinolin-1(3H)-one (2I). White solid, yield 50% (30.0 mg), hexane/EtOAc (91:9); ¹H NMR (500 MHz, CDCl₃): δ 8.18 (d, *J* = 8.40 Hz, 1H), 8.00 (d, *J* = 8.15 Hz, 1H), 7.90–7.87 (m, 1H), 7.58 (t, *J* = 7.25 Hz, 1H), 7.43 (d, *J* = 8.65 Hz, 2H), 7.10 (d, *J* = 8.65 Hz, 2H), 5.43 (s, 2H), 3.92 (s, 3H); ¹³C{¹H} NMR (125 MHz, CDCl₃): δ 168.2, 163.7, 160.7, 151.6, 151.3, 132.5, 131.6, 129.3, 128.2, 127.29, 127.24, 123.4, 113.7, 113.3, 69.4, 55.3; HRMS (ESI-TOF) *m*/*z*: [M + H]⁺ calcd for C₁₈H₁₄NO₃⁺, 292.0968; found, 292.0975.

9-(3-Chloro-4-fluorophenyl)furo[3,4-b]quinolin-1(3H)-one (2m). White solid, yield 70% (70.0/42 mg), hexane/EtOAc (92:8); ¹H NMR (500 MHz, CDCl₃): δ 8.22 (d, J = 8.50 Hz, 1H), 7.95–7.91 (m, 1H), 7.87 (d, J = 8.60 Hz, 1H), 7.66–7.62 (m, 1H), 7.51 (d, J = 6.80 Hz, 1H), 7.36 (d, J = 6.80 Hz, 2H), 5.46 (s, 2H); ¹³C{¹H} NMR (125 MHz, CDCl₃): δ 167.7, 163.5, 158.8 (d, ¹J_{C-F} = 250.8 Hz), 151.3, 148.5, 132.8, 132.0, 129.9 (d, ³J_{C-F} = 7.5 Hz), 129.6, 128.5 (d, ⁴J_{C-F} = 4.2 Hz), 127.9, 127.3, 126.6, 121.4 (d, ²J_{C-F} = 17.10 Hz), 116.7 (d, ²J_{C-F} = 21.50 Hz), 113.6, 69.6; HRMS (ESI-TOF) *m*/*z*: [M + H]⁺ calcd for C₁₇H₁₀CIFNO₂⁺, 314.0378; found, 314.0370.

7-Chloro-9-(2-chlorophenyl)furo[*3,4-b*]*quino*[*in-1(3H*)-*one* (*2n*). White solid, yield 70% (42.0 mg), hexane/EtOAc (92:8); ¹H NMR (500 MHz, CDCl₃): δ 8.17 (d, J = 9.05 Hz, 1H), 7.83 (dd, J_1 = 9.0 Hz, J_2 = 2.3 Hz, 1H), 7.63–7.61 (m, 2H), 7.55 (td, J_1 = 7.60 Hz, J_2 = 1.70 Hz, 1H), 7.48 (td, J_1 = 7.45 Hz, J_2 = 1.30 Hz, 1H), 7.28 (dd, J_1 = 7.60 Hz, J_2 = 1.76 Hz, I_2 = 1.75 Hz, 1H), 5.46 (s, 2H); ¹³C{¹H} NMR (125 MHz, CDCl₃): δ 167.0, 163.6, 149.7, 146.9, 133.9, 133.6, 132.9, 131.1, 131.0, 130.6, 130.5, 130.0, 127.5, 126.9, 126.0, 115.3, 69.8; HRMS (ESI-TOF) *m/z*: [M + Na]⁺ calcd for C₁₇H₉Cl₂NNaO₂⁺, 351.9902; found, 351.9903.

9-(2-Chlorophenyl)-7-nitrofuro[3,4-b]quinolin-1(3H)-one (2o). White solid, yield 74% (74.0/44.3 mg), hexane/EtOAc (90:10); ¹H NMR (500 MHz, CDCl₃): δ 8.17 (d, *J* = 9.0 Hz, 1H), 7.83 (dd, *J*₁ = 9.05 Hz, *J*₂ = 2.40 Hz, 1H), 7.64–7.61 (m, 2H), 7.57–7.53 (m, 1H), 7.50–7.47 (m, 1H), 7.29 (dd, *J*₁ = 7.55 Hz, *J*₂ = 1.65 Hz, 1H), 5.48 (s, 2H); ¹³C{¹H} NMR (125 MHz, CDCl₃): δ 167.0, 163.6, 149.7, 146.9, 133.9, 133.6, 132.9, 131.1, 131.0, 130.6, 130.5, 130.0, 127.4, 126.9, 126.0, 115.3, 69.8; HRMS (ESI-TOF) *m*/*z*: [M + H]⁺ calcd for C₁₇H₁₀ClN₂O₄⁺, 341.0323; found, 341.0328.

7-Chloro-9-(2-fluorophenyl)furo[3,4-b]quinolin-1(3H)-one (**2p**). White solid, yield 71% (42.6 mg), hexane/EtOAc (93:7); ¹H NMR (500 MHz, CDCl₃): δ 8.17 (d, J = 9.0 Hz, 1H), 7.85 (dd, J_1 = 9.05 Hz, J_2 = 2.35 Hz, 1H), 7.79 (s, 1H), 7.63–7.59 (m, 1H), 7.39–7.36 (m, 2H), 7.34–7.31 (m, 1H), 5.47 (s, 2H); ¹³C{¹H} NMR (125 MHz, CDCl₃): δ 167.0, 163.5, 159.60 (d, ¹ J_{C-F} = 247.50 Hz), 149.7, 143.8, 133.9, 133.5, 132.07 (d, ³ J_{C-F} = 8.15 Hz), 131.24 (d, ⁴ J_{C-F} = 2.47 Hz), 131.0, 127.6, 126.0, 124.27 (d, ⁴ J_{C-F} = 3.60 Hz), 118.98 (d, ² J_{C-F} = 15.9 Hz), 116.19 (, ² J_{C-F} = 21.2 Hz), 115.3, 69.6; HRMS (ESI-TOF) *m*/*z*: [M + H]⁺ calcd for C₁₇H₁₀ClFNO₂⁺, 314.0378; found, 314.0370.

8-Fluoro-9-(4-fluorophenyl)furo[3,4-b]quinolin-1(3H)-one (2q). White solid, yield 78% (46.8 mg), hexane/EtOAc (92:8); ¹H NMR (500 MHz, CDCl₃): δ 8.05–8.03 (m, 1H), 7.87–7.83 (m, 1H), 7.37–7.34 (m, 2H), 7.26–7.23 (m, 1H), 7.22–7.20 (m, 1H), 7.19–7.17 (m, 1H), 5.42 (s, 2H); ¹³C{¹H} NMR (100 MHz, CDCl₃): δ 167.1, 164.0, 163.6 (d, ¹ J_{C-F} = 246.81 Hz), 159.9 (d, ¹ J_{C-F} = 261.03 Hz), 152.3, 148.7, 132.8 (d, ³ J_{C-F} = 9.81 Hz), 130.0 (d, ⁴ J_{C-F} = 3.6 Hz), 129.9 (d, ⁴ J_{C-F} = 3.53 Hz), 125.8 (d, ⁴ J_{C-F} = 4.35 Hz), 114.9 (2C) (d, ¹ J_{C-F} = 28.7 Hz), 113.0 (2C) (d, ¹ J_{C-F} = 21.7 Hz), 69.2; HRMS (ESI-TOF) *m*/*z*: [M + Na]⁺ calcd for C₁₇H₉F₂NNaO₂⁺, 320.0493; found, 320.0476.

9-(4-Bromophenyl)-8-fluorofuro[3,4-b]quinolin-1(3H)-one (2r). White solid, yield 75% (45 mg), hexane/EtOAc (92:8); ¹H NMR (500 MHz, CDCl₃): δ 8.05 (d, *J* = 8.55 Hz, 1H), 7.89–7.84 (m, 1H), 7.65–7.63 (m, 2H), 7.27–7.26 (m, 1H), 7.25–7.23 (s, 2H), 5.44 (s, 2H); ¹³C{¹H} NMR (100 MHz, CDCl₃): δ 167.0, 164.5, 159.6 (d, ¹*J*_{C-F} = 277.65 Hz), 152.3, 148.3, 132.9, 132.88, 132.81, 131.0, 129.6 (d, ⁴*J*_{C-F} = 3.53 Hz), 125.9 (d, ⁴*J*_{C-F} = 4.31 Hz), 123.4, 115.6, 113.1 (d, ⁴*J*_{C-F} = 4.31 Hz), 69.2; HRMS (ESI-TOF) *m/z*: [M + Na]⁺ calcd for C₁₇H₉BrFNNaO₂⁺, 379.9692; found, 379.9693.

9-(4-Chlorophenyl)-7-nitrofuro[3,4-b]quinolin-1(3H)-one (2s). White solid, yield 75% (45 mg), hexane/EtOAc (90:10); ¹H NMR (500 MHz, CDCl₃): δ 8.83 (d, J = 2.55 Hz, 1H), 8.66 (dd, $J_1 = 9.25$ Hz, $J_2 = 2.55$ Hz, 1H), 8.37 (d, J = 9.25 Hz, 1H), 7.63 (d, J = 7.90, 2H), 7.44 (d, J = 8.35, 2H), 5.51 (s, 2H); ¹³C{¹H} NMR (125 MHz, CDCl₃): δ 167.1, 166.6, 153.1, 152.2, 146.3, 137.1, 131.5, 131.3, 129.2, 128.3, 126.1, 125.8, 124.4, 115.2, 69.6; HRMS (ESI-TOF) *m*/*z*: [M + H]⁺ calcd for C₁₇H₁₀ClN₂O₄⁺, 341.0323; found, 341.0324.

Furo[3,4-*b*]*quinoxalin*-1(3*H*)-one (2*t*). White solid, yield 39% (23.4 mg), hexane/EtOAc (92:8); ¹H NMR (500 MHz, CDCl₃): δ 8.41 (dd, J_1 = 8.45 Hz, J_2 = 1.7 Hz, 1H), 8.24 (dd, J_1 = 8.42 Hz, J_2 = 1.45 Hz, 1H), 8.04–8.00 (m, 1H), 7.98–7.95 (m, 1H), 5.61 (s, 2H); ¹³C{¹H} NMR (125 MHz, CDCl₃): δ 166.3, 156.5, 144.2 (2C), 138.4, 133.6, 131.5, 131.3, 129.3, 69.0; HRMS (ESI-TOF) *m/z*: [M + H]⁺ calcd for C₁₀H₇N₂O₂⁺, 187.0502; found, 187.0491.

7-Nitro-5-phenyl-1H-pyrano[3,4-*b*]quinolin-4(3*H*)-one (2*a*'). White solid, hexane/EtOAc (85:15), ¹H NMR (400 MHz, CDCl₃): δ 8.56 (dd, *J*₁ = 9.20 Hz, *J*₂ = 2.44 Hz, 1H), 8.51 (d, *J* = 2.44 Hz, 1H), 8.23 (d, *J* = 9.20 Hz, 1H), 7.60–7.58 (m, 3H), 7.25–7.23 (m, 2H), 5.16 (s, 2H), 4.41 (s, 2H); ¹³C{¹H} NMR (125 MHz, CDCl₃): δ 192.7, 162.6, 153.5, 150.6, 146.0, 134.5, 130.8, 129.1, 128.7, 128.2, 127.0, 125.4, 125.0, 122.1, 74.3, 71.2; HRMS (ESI-TOF) *m/z*: [M + H]⁺ calcd for C₁₈H₁₃N₂O₄⁺, 321.0869; found, 321.0880.

1-(2-(Fluoromethyl)-6-nitro-4-phenylquinolin-3-yl)ethan-1-one (**3a**). White solid, yield 30% (19.05 mg), hexane/EtOAc (94:6); ¹H NMR (400 MHz, CDCl₃): δ 8.62 (d, *J* = 2.44 Hz, 1H), 8.53 (dd, *J*₁ = 9.16 Hz, *J*₂ = 2.52 Hz, 1H), 8.28 (d, *J* = 9.20 Hz, 1H), 7.62–7.60 (m, 3H), 7.38–7.36 (m, 2H), 5.74 (d, ¹*J*_{H–F} = 46.44 Hz, 2H), 1.99 (s, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃): δ 203.7, 155.6 (d, ²*J*_{C–F} = 18.6 Hz), 148.9, 146.8, 146.4, 134.9, 133.2, 131.6, 130.1, 129.8, 129.4, 125.6, 123.9, 123.1, 84.7 (d, ¹*J*_{C–F} = 168.41 Hz), 31.5; ¹⁹F NMR (376 MHz, CDCl₃): δ –115.0; HRMS (ESI-TOF) *m*/*z*: [M + H]⁺ calcd for C₁₈H₁₄FN₂O₃⁺, 325.0982; found, 325.0990.

1-(2-(Fluoromethyl)-4-phenylquinolin-3-yl)ethan-1-one (**3b**). White solid, yield 16% (10.2 mg), hexane/EtOAc (95:5); ¹H NMR (500 MHz, CDCl₃): δ 8.15 (d, *J* = 8.35 Hz, 1H), 7.79–7.76 (m, 1H), 7.69–7.68 (d, *J* = 8.35 Hz, 1H), 7.54–7.51 (m, 4H), 7.38–7.36 (m, 2H), 5.72 (d, ¹*J*_{H-F} = 46.70 Hz, 2H), 1.98 (s, 3H); ¹³C{¹H} NMR (125 MHz, CDCl₃): δ 204.9, 151.65 (d, ²*J*_{C-F} = 18.43 Hz), 146.9,

145.0, 134.8, 133.6, 130.5, 129.9 (2C), 129.6, 129.1, 128.8, 127.8, 126.2, 85.06 (d, ${}^{1}J_{C-F}$ = 166.55 Hz), 31.8; ${}^{19}F$ NMR (376 MHz, CDCl₃): δ –113.9; HRMS (ESI-TOF) *m/z*: [M + H]⁺ calcd for C₁₈H₁₅FNO⁺, 280.1132; found: 280.1125.

1-($\dot{4}$ -(4-Bromophenyl)-2-(fluoromethyl)quinolin-3-yl)ethan-1one (**3***c*). White solid, yield 20% (12.6 mg), hexane/EtOAc (95:5); ¹H NMR (500 MHz, CDCl₃): δ 8.16 (d, J = 8.45 Hz, 1H), 7.80–7.77 (m, 1H), 7.69 (d, J = 3.3 Hz, 2H), 7.63 (dd, $J_1 = 7.63$ Hz, $J_2 = 1.45$ Hz, 1H), 7.56–7.53 (m, 1H), 7.26–7.25 (m, 2H), 5.72 (d, ¹ $J_{H-F} = 46.65$ Hz, 2H), 2.05 (s, 3H); ¹³C{¹H} NMR (125 MHz, CDCl₃): δ 204.5, 151.6 (d, ² $J_{C-F} = 18.65$ Hz), 146.9, 143.5, 133.6, 132.1 (2C), 131.5, 130.6 (2C), 129.7, 128.0, 125.9, 125.8, 123.7, 85.0 (d, ¹ $J_{C-F} = 166.8$ Hz), 32.0; ¹⁹F NMR (376 MHz, CDCl₃): δ –113.5; HRMS (ESI-TOF) *m*/*z*: [M + H]⁺ calcd for C₁₈H₁₄BrFNO⁺, 358.0237; found, 358.0247.

1-(4-(4-Chlorophenyl)-2-(fluoromethyl)-6-nitroquinolin-3-yl)ethan-1-one (**3d**). White solid, yield 28% (17.6 mg), hexane/EtOAc (95:5); ¹H NMR (500 MHz, CDCl₃): δ 8.56–8.53 (m, 1H), 8.52 (d, J = 2.28 Hz, 1H), 8.29 (d, J = 9.04 Hz, 1H), 7.61–7.59 (m, 2H), 7.33–7.32 (m, 2H), 5.73 (d, ¹ $J_{H-F} = 46.44$ Hz, 2H), 2.05 (s, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃): δ 202.2, 154.58 (d, ² $J_{C-F} = 18.88$ Hz), 147.8, 145.5, 144.3, 135.5, 134.0, 130.7, 130.5, 130.1, 128.7, 124.3, 123.0, 121.6, 83.64 (d, ¹ $J_{C-F} = 168.53$ Hz), 30.7; ¹⁹F NMR (376 MHz, CDCl₃): δ –114.5. HRMS (ESI-TOF) m/z: [M + H]⁺ calcd for C₁₈H₁₃ClFN₂O₃⁺, 359.0593; found, 359.0600.

1-(6-Chloro-4-phenyl-2-(((2,2,6,6-tetramethylpiperidin-1-yl)oxy)methyl)quinolin-3-yl)ethan-1-one (**2a**"). White solid, yield 69% (62.37 mg), hexane/EtOAc (98:2); ¹H NMR (500 MHz, CDCl₃): δ 8.59 (d, *J* = 2.40 Hz, 1H), 8.47 (d, *J* = 2.45 Hz, 1H), 8.26 (d, *J* = 9.20 Hz, 1H), 7.60–7.58 (m, 3H), 7.39–7.38 (m, 2H), 5.23 (s, 2H), 2.05 (s, 3H), 1.53–1.51 (m, 3H), 1.48–1.46 (m, 3H), 1.22 (s, 6H), 1.05 (s, 6H); ¹³C{¹H} NMR (125 MHz, CDCl₃): 203.3, 157.4, 149.2, 146.0 (2C), 135.6, 133.7, 131.6, 129.9, 129.8, 129.2, 125.3, 123.3, 123.0, 80.3, 60.1, 39.9, 32.8, 32.1, 20.6, 17.0; HRMS (ESI-TOF) *m*/*z*: [M + H]⁺ calcd for C₂₇H₃₂N₃O₄⁺, 462.2387; found, 462.2309.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/acs.joc.0c02238.

Copies of ¹H NMR, ¹³C NMR, and HRMS for all synthesized compounds and mechanistic investigation (PDF)

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Notes

The authors declare no competing financial interest.

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REFERENCES

(1) Foley, P.; Eghbali, N.; Anastas, P. T. Silver-Catalyzed One-Pot Synthesis of Arylnaphthalene Lactone Natural Products. *J. Nat. Prod.* **2010**, *73*, 811–813.

(2) Paudel, K.; Pandey, B.; Xu, S.; Taylor, D. K.; Tyer, D. L.; Torres, C. L.; Gallagher, S.; Kong, L.; Ding, K. Cobalt-Catalyzed Acceptorless Dehydrogenative Coupling of Primary Alcohols to Esters. *Org. Lett.* **2018**, *20*, 4478–4481.

(3) (a) Karmel, I. S. R.; Fridman, N.; Tamm, M.; Eisen, M. S. Mixed Imidazolin-2-iminato-Cp* Thorium(IV) Complexes: Synthesis and Reactivity Toward Oxygen-Containing Substrates. *Organometallics* **2015**, *34*, 2933–2942. (b) Karmel, I. S. R.; Fridman, N.; Tamm, M.; Eisen, M. S. Mono(imidazolin-2-iminato) Actinide Complexes: Synthesis and Application in the Catalytic Dimerization of Aldehydes. *J. Am. Chem. Soc.* **2014**, *136*, 17180–17192.

(4) Carlos, A. M. M.; Stieler, R.; Lüdtke, D. S. Catalytic Asymmetric Synthesis of 3-Aryl Phthalides Enabled by Arylation–Lactonization of 2-Formylbenzoates. *Org. Biomol. Chem.* **2019**, *17*, 283–289.

(5) Yudin, A. K. Catalyzed Carbon Heteroatom Bond Formation; Wiley-VCH: Weinheim, 2011; pp 35-68.

(6) (a) Lee, J. M.; Chang, S. Pt-Catalyzed Sp³ C–H Bond Activation of O-Alkyl Substituted Aromatic Carboxylic Acid Derivatives for the Formation of Aryl Lactones. *Tetrahedron Lett.* **2006**, 47, 1375–1379. (b) Novák, P.; Correa, A.; Gallardo-Donaire, J.; Martin, R. Synergistic Palladium-Catalyzed C(Sp³)-H Activation/C(Sp³)-O Bond Formation: A Direct, Step-Economical Route to Benzolactones. *Angew. Chem., Int. Ed.* **2011**, 50, 12236–12239. (c) Das, J.; Dolui, P.; Ali, W.; Biswas, J. P.; Chandrashekar, H. B.; Prakash, G.; Maiti, D. A Direct Route to Six and Seven Membered Lactones via γ -C(Sp3)–H Activation: A Simple Protocol to Build Molecular Complexity. *Chem. Sci.* **2020**, *11*, 9697–9702.

(7) Liu, J.; Miotto, R. J.; Segard, J.; Erb, A. M.; Aponick, A. Catalytic Dehydrative Lactonization of Allylic Alcohols. *Org. Lett.* **2018**, *20*, 3034–3038.

(8) Arcadi, A.; Fabrizi, G.; Goggiamani, A.; Marinelli, F. Pd- and Rh-Catalyzed Hydroarylation of γ -(2-Methoxycarbonylphenyl) Propargylic Alcohols: Approaches to 4- or 5-Substituted Seven-Membered Benzolactones and 3,3-Disubstituted Phthalides. *J. Org. Chem.* **2015**, *80*, 6986–6995.

(9) Song, L.; Zhu, L.; Zhang, Z.; Ye, J.-H.; Yan, S.-S.; Han, J.-L.; Yin, Z.-B.; Lan, Y.; Yu, D.-G. Catalytic Lactonization of Unactivated Aryl C–H Bonds with CO₂: Experimental and Computational Investigation. *Org. Lett.* **2018**, *20*, 3776–3779.

(10) Xie, X.; Stahl, S. S. Efficient and Selective Cu/Nitroxyl-Catalyzed Methods for Aerobic Oxidative Lactonization of Diols. *J. Am. Chem. Soc.* **2015**, *137*, 3767–3770.

(11) (a) Blair, A.; Zmuda, F.; Malviya, G.; Tavares, A. A. S.; Tamagnan, G. D.; Chalmers, A. J.; Dewar, D.; Pimlott, S. L.; Sutherland, A. A Novel ¹⁸F-Labelled High Affinity Agent for PET Imaging of the Translocator Protein. Chem. Sci. 2015, 6, 4772-4777. (b) Tufail, F.; Saquib, M.; Singh, S.; Tiwari, J.; Singh, M.; Singh, J.; Singh, J. Bioorganopromoted Green Friedländer Synthesis: A Versatile New Malic Acid Promoted Solvent Free Approach to Multisubstituted Quinolines. New J. Chem. 2017, 41, 1618-1624. (c) Soleimani, E.; Namivandi, M. N.; Sepahvand, H. ZnCl2 Supported on Fe₃O₄@SiO₂ Core-Shell Nanocatalyst for the Synthesis of Quinolines via Friedländer Synthesis under Solvent-Free Condition. Appl. Organomet. Chem. 2017, 31, No. e3566. (d) Blair, A.; Stevenson, L.; Dewar, D.; Pimlott, S. L.; Sutherland, A. Structure-Activity Relationships of Novel Iodinated Quinoline-2-Carboxamides for Targeting the Translocator Protein. MedChem-Comm 2013, 4, 1461-1466. (e) Schmidt, D. G.; Seemuth, P. D.; Zimmer, H. Substituted. Gamma. -Butyrolactones. Part 31. 2,4(3H,5H)-Furandione: Heteroannulations with Aromatic o-Amino Carbonyl Compounds and Condensations with Some Vic-Polyones. J. Org. Chem. **1983**, 48, 1914–1916.

(12) (a) Cheng, T.; Yin, W.; Zhang, Y.; Zhang, Y.; Huang, Y. Palladium Catalyzed Acetoxylation of Benzylic C-H Bonds Using a Bidentate Picolinamide Directing Group. Org. Biomol. Chem. 2014, 12, 1405–1411. (b) Zhang, G.; Xie, X.; Zhu, J.; Li, S.; Ding, C.; Ding, P. Pd(II)-Catalyzed C(Sp3)–H Arylation of Amino Acid Derivatives with Click-Triazoles as Removable Directing Groups. Org. Biomol. Chem. 2015, 13, 5444–5449. (c) Pati, T. K.; Debnath, S.; Kundu, M.; Khamrai, U.; Maiti, D. K. 3-Amino-1-Methyl-1H-Pyridin-2-One-Directed PdII Catalysis: C(Sp³)–H Activated Diverse Arylation Reaction. Org. Lett. 2018, 20, 4062–4066.

(13) (a) Clark, J. R.; Feng, K.; Sookezian, A.; White, M. C. Manganese-Catalysed Benzylic $C(Sp^3)$ -H Amination for Late-Stage Functionalization. Nat. Chem. 2018, 10, 583-591. (b) Bai, Q.; Tang, J.; Wang, H. Functionalization of Sulfonamide-Containing Peptides through Late-Stage Palladium-Catalyzed C(Sp³)-H Arylation. Org. Lett. 2019, 21, 5858-5861. (c) Kang, T.; Kim, Y.; Lee, D.; Wang, Z.; Chang, S. Iridium-Catalyzed Intermolecular Amidation of Sp³ C-H Bonds: Late-Stage Functionalization of an Unactivated Methyl Group. J. Am. Chem. Soc. 2014, 136, 4141-4144. (d) White, M. C.; Zhao, J. Aliphatic C-H Oxidations for Late-Stage Functionalization. J. Am. Chem. Soc. 2018, 140, 13988-14009. (e) Liu, D.; Liu, C.; Li, H.; Lei, A. Direct Functionalization of Tetrahydrofuran and 1,4-Dioxane: Nickel-Catalyzed Oxidative C(Sp³)-H Arylation. Angew. Chem., Int. Ed. 2013, 52, 4453-4456. (f) Liu, D.; Liu, C.; Li, H.; Lei, A. Copper-Catalysed Oxidative C-H/C-H Coupling between Olefins and Simple Ethers. Chem. Commun. 2014, 50, 3623-3626.

(14) (a) Zhang, N.; Cheng, R.; Zhang-Negrerie, D.; Du, Y.; Zhao, K. Hypervalent Iodine-Mediated Oxygenation of N,N-Diaryl Tertiary Amines: Intramolecular Functionalization of Sp³ C-H Bonds Adjacent to Nitrogen. J. Org. Chem. 2014, 79, 10581-10587. (b) Li, G.-X.; Hu, X.; He, G.; Chen, G. Photoredox-Mediated Remote C(Sp³)-H Heteroarylation of Free Alcohols. Chem. Sci. 2019, 10, 688-693. (c) Shu, X.-Z.; Xia, X.-F.; Yang, Y.-F.; Ji, K.-G.; Liu, X.-Y.; Liang, Y.-M. Selective Functionalization of Sp³ C-H Bonds Adjacent to Nitrogen Using (Diacetoxyiodo)Benzene (DIB). J. Org. Chem. 2009, 74, 7464-7469. (d) Budhwan, R.; Yadav, S.; Murarka, S. Late Stage Functionalization of Heterocycles Using Hypervalent Iodine(III) Reagents. Org. Biomol. Chem. 2019, 17, 6326-6341. (e) Džambaski, Z.; Bondžić, B. P. Dehydrogenative $C(Sp^3)$ -H Bond Functionalization of Tetrahydroisoquinolines Mediated by Organic Oxidants under Mild Conditions. Org. Biomol. Chem. 2019, 17, 6420-6425. (f) Muramatsu, W.; Nakano, K.; Li, C.-J. Direct Sp³ C-H Bond Arylation, Alkylation, and Amidation of Tetrahydroisoquinolines Mediated by Hypervalent Iodine(III) under Mild Conditions. Org. Biomol. Chem. 2014, 12, 2189-2192.

(15) (a) Labruère, R.; Helissey, P.; Desbène-Finck, S.; Giorgi-Renault, S. Design and Effective Synthesis of the First 4-Aza-2,3-Didehydropodophyllotoxin Rigid Aminologue: A N-Methyl-4-[(3,4,5-Trimethoxyphenyl)Amino)]-1,2 Dihydroquinoline-Lactone. J. Org. Chem. 2008, 73, 3642-3645. (b) da Rocha Pissurno, A. P.; Santos, F. A.; Candido, A. C. B. B.; Magalhães, L. G.; da Silva de Laurentiz, R. In Vitro Leishmanicidal Activity of Lactone 1,4-Dihydroquinoline Derivatives against Leishmania (Leishmania) Amazonensis. Med. Chem. Res. 2018, 27, 2224-2229. (c) Frackenpohl, J.; Adelt, I.; Antonicek, H.; Arnold, C.; Behrmann, P.; Blaha, N.; Böhmer, J.; Gutbrod, O.; Hanke, R.; Hohmann, S.; van Houtdreve, M.; Lösel, P.; Malsam, O.; Melchers, M.; Neufert, V.; Peschel, E.; Reckmann, U.; Schenke, T.; Thiesen, H.-P.; Velten, R.; Vogelsang, K.; Weiss, H.-C. Insecticidal Heterolignans-Tubuline Polymerization Inhibitors with Activity against Chewing Pests. Bioorg. Med. Chem. 2009, 17, 4160-4184. (d) Jeedimalla, N.; Flint, M.; Smith, L.; Haces, A.; Minond, D.; Roche, S. P. Multicomponent Assembly of 4-Aza-Podophyllotoxins: A Fast Entry to Highly Selective and Potent Anti-Leukemic Agents. Eur. J. Med. Chem. 2015, 106, 167-179.

(16) Pang, X.; Xiang, L.; Ma, J.; Yang, X.; Yan, R. Halogenations of Substituted 2-Alkylquinoline with Iodine and Halide Exchange with

AgF₂. RSC Adv. **2016**, *6*, 111713–111717. (b) Danahy, K. E.; Cooper, J. C.; Van Humbeck, J. F. Benzylic Fluorination of Aza-Heterocycles Induced by Single-Electron Transfer to Selectfluor. Angew. Chem., Int. Ed. **2018**, *57*, 5134–5138.

(17) (a) Xie, L.-Y.; Qu, J.; Peng, S.; Liu, K.-J.; Wang, Z.; Ding, M.-H.; Wang, Y.; Cao, Z.; He, W.-M. Selectfluor-Mediated Regioselective Nucleophilic Functionalization of N-Heterocycles under Metal- and Base-Free Conditions. *Green Chem.* **2018**, *20*, 760–764. (b) Pitts, C. R.; Ling, B.; Snyder, J. A.; Bragg, A. E.; Lectka, T. Aminofluorination of Cyclopropanes: A Multifold Approach through a Common, Catalytically Generated Intermediate. J. Am. Chem. Soc. **2016**, *138*, 6598–6609. (c) Cao, Z.; Zhu, Q.; Lin, Y.-W.; He, W.-M. The Concept of Dual Roles Design in Clean Organic Preparation. Chin. Chem. Lett. **2019**, *30*, 2132–2138.

(18) (a) Zhou, X.-Y.; Chen, X.; Wang, L.-G. Highly Efficient Brønsted Acid and Lewis Acid Catalysis Systems for the Friedländer Quinoline Synthesis. *Synth. Commun.* **2018**, *48*, 830–837. (b) Anil Kumar, B. S. P.; Madhav, B.; Harsha Vardhan Reddy, K.; Nageswar, Y. V. D. Quinoxaline Synthesis in Novel Tandem One-Pot Protocol. *Tetrahedron Lett.* **2011**, *52*, 2862–2865. (c) Wang, C.-C.; Chan, C.-K.; Lai, C.-Y. Environmentally Friendly Nafion-Mediated Friedländer Quinoline Synthesis of Substituted Quinolinyl Chalcones. *Synthesis* **2020**, *52*, 1779–1794.