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# Synthesis, vasorelaxant activity and 2D-QSAR study of some novel pyridazine derivatives

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# Synthesis, vasorelaxant activity and 2D-QSAR study of some novel pyridazine derivatives

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#### Abstract

Novel 3,6-disubstituted pyridazines were synthesized by facile method and screened for their vasorelaxant properties utilizing isolated thoracic rat aortic rings. Compounds **8a** and **11a** exerted potent vasorelaxant activity (IC<sub>50</sub> = 198 and 177  $\mu$ M, respectively) relative to doxazosin mesylate (used reference standard, IC<sub>50</sub> = 226  $\mu$ M), that, they may represent promising hits for treatment of cardiovascular disorders. The observed activity was validated by a statistically significant QSAR model (N=32, n=6, R<sup>2</sup>=0.811782, R<sup>2</sup><sub>cvOO</sub> = 0.7153, R<sup>2</sup><sub>cvMO</sub> = 0.7209, F=17.9708, s<sup>2</sup>=9.65226x10<sup>-8</sup>) that was obtained employing CODESSA-Pro software.

Keywords: pyridazines, vasodilator, QSAR

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#### 1. Introduction

Cardiovascular diseases (CVDs), in particular coronary heart disease and stroke, are the leading cause of mortality in the developed countries. According to the World Health Organization (WHO), 17 million people die every year by CVDs, accounting for almost one third of deaths worldwide per year [1]. Among cardiovascular disorders, hypertension is known as a silent killer as it is the most common risk factor that can cause coronary disease, myocardial infarction, stroke and sudden death. In addition, it is the major contributor to cardiac failure and renal insufficiency [2,3]. Moreover, hypertension is not only responsible for high morbidity and mortality, but it impacts negatively the quality of life of a huge number of people across the world. Therefore, prevention and treatment of hypertension is an important public health challenge. Over the past decades, great efforts have been made to discover many antihypertensive drugs that act through different mechanisms such as: diuretics [4], angiotensin-converting enzyme inhibitors [5,6], angiotensin II receptor blockers [6,7], calcium channel blockers [8], centrally sympathetic  $\alpha$ 2-adrenoreceptors stimulants [9], and drugs that prevent the action of peripheral sympathetic activity as  $\beta$ -adrenergic [10,11] and  $\alpha$ -adrenergic blocking agents [12]. It is noteworthy that the reduction in blood pressure achieved with most of the aforementioned classes of drugs is directly or indirectly related to the relaxation of vascular smooth muscles; which makes vasorelaxation one of important strategies in controlling hypertension [13]. Several agents have been developed; however they are all associated with side effects such as fatigue, mood change, sleep disturbances, etc [14]. Therefore, there is a continuous need to explore and develop new vasorelaxant agents with minimal side effects.

The pyridazine nucleus represents a versatile scaffold to develop new pharmacologically active compounds. This heterocyclic system has a wide range of biological activities and can also be used to link other pharmacophoric groups [15–18]. For instance, the 6-(aryl or heteroaryl)pyridazinone derivatives represent an important core with a wide pharmacological profile that includes interesting activities on cardiovascular system, such as cardiotonic effects [19,20], antihypertensive activity [21] and platelet aggregation inhibition [22]. Zardaverine **I**, levosimendan **II** and motapizone **III** are some characteristic drugs bearing this pharmacophoric moiety [23–25] (Fig. 1).

Most of the described pyridazinone derivatives, showing activity on cardiovascular system, possess aryl residues at position 6 of the pyridazine ring [23,26,27]. Interestingly, 6-hydrazino-3-phenylpyridazine IV which exhibited strong hypotensive properties, is the structural analog of the well known vasodilator hydralazine V that possesses a phenyl ring attached to the pyridazine nucleous instead of being fused to it as in hydralazine [28] (Figure 1). Encouraged by these findings; this work focused on design and synthesis of some phenylpyridazines substituted with a heterocyclic ring, namely, pyrrolidine, imidazole or pyrazole, (general structure A, Figure 2). These ring systems can be considered as pharmacophoric moieties in many vasorelaxant agents [9,19,21,25,29-31]. Moreover, 6-hydrazino-3-phenylpyridazine IV was used for further structural modification through derivatization of the hydrazine moiety into a semicarbazide group (general structure B), or an aryl hydrazone (general structure C, Figure 2) in order to study their effect on the modulation of the activity. Additionally, quantitative structureactivity relationship (QSAR) studies were considered in the present work to validate the obtained vasorelaxant activity and detect the most important structural parameters controlling it.

Figure 1

Figure 2

#### 2- Results and discussion

#### 2.1. Chemistry

The targeted pyridazines were synthesized according to the general procedures outlined in Schemes 1 and 2. Thus, the 6-aryl-3-pyridazinones **3a-d** were synthesized from the reaction of the appropriate acetophenone **1a-d** and glyoxilic acid **2** followed by cyclisation with hydrazine hydrate [32]. Chlorination of **3a-d** with phosphorous oxychloride afforded 3-chloro-6-arylpyridazines **4a-d** which were further treated with hydrazine hydrate in absolute ethanol to obtain the 3-hydrazino derivatives **5a-d** [33]. The target compounds **6a-d** and **7a-c** were obtained from the precursor chloropyridazines **4a-d** through a nucleophilic substitution reaction with imidazole and pyrrolidine, respectively (Scheme 1). The structures of the obtained compounds **6a-d** were confirmed by <sup>1</sup>H NMR which revealed three broad singlet signals at 7.30-7.32, 7.81-7.83 and 8.49-

8.53 ppm corresponding to the imidazole protons. Compound 6b revealed triplet corresponding to the imidazole proton at C2 at 7.83 ppm due to long range coupling with protons at C4, C5 of the imidazole ring. However, the pyridazine protons of 6a-d appeared as two doublets at 7.59-7.68 and 7.97-8.24 ppm, while the phenyl ring protons of **6a** revealed two multiplets at 7.53-7.59 and 8.10-8.12 ppm. In case of the *p*-substituted derivatives **6b**, **6d**, two doublets appeared at the range 7.07-7.56 and 8.07 ppm. On the other hand, the o-substituted derivative 6c showed doublet at 7.08 ppm corresponding to the proton at C3 that coupled with H of C4, two doublet triplets at 7.18 and 7.49 ppm corresponding to protons at C4 and C5 which coupled with protons at C3, C5, C6 and C4, C6, C3, respectively, and one doublet of doublet at 8.04 ppm due to the proton at C6 that coupled with protons at C5 and C4. Moreover, the appearance of multiplet signals in the ranges 2.06-2.12 and a broad singlet at 3.63 ppm corresponding to the pyrrolidine protons attested the obtained compounds 7a,b. Regarding compound 7c, the pyrrolidine protons appeared as a multiplet at 1.98-2.01 ppm and a triplet at 3.56 ppm. Meanwhile, cyclocondensation of the hydrazine derivatives 5a-d with acetyl acetone in absolute ethanol afforded 3-(3,5-dimethyl-1*H*-pyrazol-1-yl)-6-arylpyridazines **8a-d**. <sup>1</sup>H NMR spectra of 8c,d confirmed the obtained structure through the presence of three singlet signals at 2.34, 2.81-2.82 and 3.89-3.90 ppm corresponding to two methyl and one methoxy groups, respectively, in addition to a singlet attributed to the pyrazole proton at 6.08-6.09 ppm. Similarly, the reaction of **5a-d** with either ethoxymethylene malononitrile or ethyl ethoxymethylene cyanoacetate in absolute ethanol gave the 5-aminopyrazole derivatives 9a-d and 10a-d, respectively. The IR spectra of 9b-d and 10 b-d showed the NH<sub>2</sub> stretching vibration as two bands at 3391-3285 and 3453-3312 cm<sup>-1</sup>, in addition to the CN band at 2230-2210 cm<sup>-1</sup> in case of compounds **9b-d**. Moreover, <sup>1</sup>H NMR spectra of **10b-d** revealed the triplet-quartet pattern of the C<sub>2</sub>H<sub>5</sub> protons at 1.37-1.39 and 4.31-4.34 ppm along with two singlet signals at 7.60-7.62 and 7.82-7.84 ppm corresponding to NH<sub>2</sub> and pyrazole proton, respectively. The former signal for NH<sub>2</sub> was disappeared upon deuteration with D<sub>2</sub>O. On the other hand, nucleophilic addition of hydrazines 5a-d to the appropriate isocyanate in methylene chloride in the presence of triethylamine afforded the semicarbazides 11a-c, whose structures were confirmed by the appearance of three NH bands at 3329-3300, 3250-3223 and 3138-3134 cm<sup>-1</sup> in the IR spectra. In addition, <sup>1</sup>H NMR showed three singlet signals corresponding to three NH groups that were exchanged with  $D_2O$ . Additionally, the condensation of the hydrazine derivatives **5a-d** with different aromatic aldehydes in absolute ethanol in the presence of catalytic amount of glacial acetic acid revealed the corresponding phenyl hydrazones **12a-h**. These derivatives were characterized by their spectral and elemental data, where <sup>1</sup>H NMR spectra revealed the presence of the characteristic singlet signal of methylidene proton at 8.09-8.43 ppm in addition to a singlet signal of NH at 7.84-7.95 or 11.40-11.93 ppm that was disappeared upon treatment with D<sub>2</sub>O.

# Scheme 1

Scheme 2

#### 2.2. In-vitro vasorelaxant activity

Vasorelaxant properties of the synthesized pyridazines were investigated in vitro using isolated thoracic aortic rings of rats precontracted with norepinephrine hydrochloride standard according to the reported procedure [34-40] and compared with doxazosin mesylate as a reference standard. The observed data (Table 1) revealed that all pyridazine derivatives exhibited remarkable vasodilating activity with IC<sub>50</sub> values ranging from 177–323  $\mu$ M compared to the reference standard doxazosin mesylate (IC<sub>50</sub> = 226  $\mu$ M). It is noteworthy that thirteen compounds (5a, 6c, 8a, 8b, 10c, 10d, 11a, 11c, 12a, 12c, 12d, 12e, 12h) showed higher activity than doxazosin mesylate (Table 1). Structure activity relationship (SAR) based on the observed vasorelaxant activity indicated that the impact of the electronic nature of the substituent X viz H, 4-Cl, 2-OCH<sub>3</sub>, 4-OCH<sub>3</sub> upon the activity is not absolute. However, substitution with different R resulted in either positive or negative change in the observed activity. Thus, considering the hydrazine derivatives 5a-d as the prototypes of the target compounds, it could be concluded that best results were obtained when the hydrazine group was structurally extended to the semicarbazides 11a-c and hydrazones 12a-h. Moreover, replacement of the hydrazine moiety with rigid azacyclic groups was sometimes in favor of activity as in some imidazolyl **6b,c**, 3,5-dimethylpyrazolyl 8a,b and 5-amino-4ethoxycarbonylpyrazolyl derivatives 10b-d. However, with other ring system such as the pyrrolidinyl **7a**–**c** and 5-amino-4-cyanopyrazolyl **9a**–**d**, the activity was diminished.

#### 2.3. 2D- QSAR modeling

#### 2.3.1. Dataset

QSAR is capable of generating a relationship between the chemical structure of an organic compound and its aximu-chemical properties. The vasorelaxant active pyridazines **5a-d**, **6a-d**, **7a,c**, **8a–d**, **9a–d**, **10a–c**, **11a–c** and **12a–h** were used as a training set in the present QSAR study (homogeneous dataset of common heterocyclic scaffold). The QSAR study was undertaken using comprehensive descriptors for structural and statistical analysis (CODESSA-Pro) software.

#### 2.3.2. Methodology

Geometry of the training set compounds was optimized using the molecular mechanics force field ( $MM^+$ ) followed by the semi-empirical AM1 method implemented in the HyperChem 8.0 package. The structures were fully optimized without fixing any parameters, thus bringing all geometric variables to their equilibrium values. The energy minimization protocol employed the Polak–Ribiere conjugated gradient algorithm. Convergence to a local minimum was achieved when the energy gradient was  $\leq 0.01$  kcal mol<sup>-1</sup>. The RHF (Restricted Hartree-Fock) method was used in spin pairing for the two semi-empirical tools [41–50]. The resulting output files were exported to CODESSA-Pro that includes MOPAC capability for the final geometry optimization.

CODESSA-Pro calculated 725 molecular descriptors including constitutional, topological, geometrical, charge-related, semi-empirical, thermodynamic, molecular-type, atomic-type and bond-type descriptors for the exported 32 bio-active pyridazine-based compounds which were used as training set in the present study. Different mathematical transformations of the experimentally observed vasodilation property/activity (IC<sub>50</sub>,  $\mu$ M) of the training set compounds were utilized for the QSAR modeling determination including property (IC<sub>50</sub>,  $\mu$ M), 1/property, log(property) and 1/log(property) values in searching for the best QSAR model.

#### 2.3.3. QSAR modeling

The best multi-linear regression (BMLR) technique was utilized which is a stepwise search for the best n-parameter regression equations (where n stands for the number of descriptors used), based on the highest  $R^2$  (squared correlation coefficient),  $R^2$ cvOO (squared cross-validation "leave one-out, LOO" coefficient),  $R^2$ cvMO (squared cross-validation "leave many-out, LMO" coefficient), F (Fisher statistical significance criteria) values, and  $s^2$  (standard deviation). The QSAR models up to 6 descriptor model describing the bio-activity of the vasorelaxant pyridazine-based active agents were generated (obeying the thumb rule of 5:1, which is the ratio between the data points and the number of QSAR descriptor). The observed and predicted values of the training set compounds **5a–d**, **6a–d**, **7a,c**, **8a–d**, **9a–d**, **10a–c**, **11a–c** and **12a–h** according to the multilinear QSAR models are presented in Table 1.

The statistical characteristics of the BMLR-QSAR model attained are presented in Table 2. The established QSAR model is statistically significant. The descriptors are sorted in the descending order of the respective values of the Student's *t*-criterion, which is a widely accepted measure of statistical significance of individual parameters in multiple linear regressions. Figure 3 shows the QSAR multi-linear model plot of correlations representing the observed *vs*. predicted  $1/\text{IC}_{50}$  ( $\mu$ M) values for the vasodilation pyridazine-based active agents. The scattered plots are uniformly distributed, covering ranges, observed 0.0031-0.0056, predicted 0.0030-0.0056 log (IC<sub>50</sub>,  $\mu$ M) units.

Figure 3

#### 2.3.4. Molecular descriptors

Molecular descriptors are the aximu-chemical parameters used to correlate the chemical structure and the bio-property value expressed as  $1/IC_{50}$ ,  $\mu$ M. The descriptors were obtained based on the BMLR method. The first descriptor controlling the attained BMLR-QSAR model based on its level of significance (*t*-criterion = 6.4616) is FHACA fractional HACA (HACA/TMSA) (MOPAC PC) which is a charge-related descriptor.. Fractional hydrogen bonding acceptor ability of the molecule *FHACA1* is determined by equation (1) [51].

Where, HASA1 is the hydrogen bonding acceptor ability, TMSA is the total molecular surface area. The second most important descriptor controlling the BMLR-QSAR model (t = 4.6771) is shadow plane YZ, which is a geometrical descriptor reflecting molecular size. Most geometrical descriptors are calculated directly from the (x,y,z) coordinates and other quantities derived from the coordinates, *e.g.* interatomic distances or distances from a specified origin (*e.g.* the molecule bary-center) [52,53]. The third and fourth descriptors of QSAR model based on their *t*-value (4.1118, -4.9294, respectively) are HA dependent HDSA-1 (Zefirov PC) (all) and HA dependent HDSA-1/TMSA (MOPAC PC) (all), which are a charge-related descriptors. The hydrogen bonding acceptor ability of the molecule HASA1 is determined by equation (2) [51].

Where,  $s_A$  stands for solvent-accessible surface area of H-bonding acceptor atoms. The fifth descriptor of the attained QSAR model (t = -5.7082) is structural information content (order 2), which is a topological descriptor determined by equations (3, 4) [51].

$${}^{k}SIC = {}^{k}IC / \log_{2} n \dots (3)$$
$${}^{k}IC = -\sum_{i=1}^{k} \frac{n_{i}}{n} \log_{2} \frac{n_{i}}{n} \dots (4)$$

Where  $n_i$  stands for number of atoms in the *i*<sup>th</sup> class, *n* is the total number of atoms in the molecule, k is the number of atomic layers in the coordination sphere around a given atom that are accounted for. The last descriptor controlling the BMLR-QSAR is 8aximum nucleophilic reaction index for atom C, which is a semi-empirical descriptor. Molecular descriptor values controlling the attained BMLR-QSAR model are presented in Table 3.

# 2.3.5. Validation of the BMLR-QSAR model

## 2.3.5.1. Internal validation

Internal validation is applied by the CODESSA-Pro technique employing both leave one out (LOO), which involves developing a number of models with one example omitted at a time, and leave many out (LMO), which involves developing a number of models with many data points omitted at a time (up to 20% of the total data points). The observed

correlations by the internal validation technique are  $R^2 \text{cvOO} = 0.715$ ,  $R^2 \text{cvMO} = 0.721$ , respectively which are significantly correlated with the squared correlation coefficient of the attained QSAR model ( $R^2 = 0.812$ ). Standard deviation of the regression ( $s^2 = 9.652 \times 10^{-8}$ ) is also a measurable value for the attained model together with the Fisher test value (F = 17.971) that reflects the ratio of the variance explained by the model and the variance due to their errors. A high value of the *F*-test relative to the s2 value is also validation of the model.

The predicted vasodilation properties due to the attained QSAR model of most of the synthesized pyridazines are compatible with their experimentally observed values preserving their relative potencies (Table 1). The predicted IC<sub>50</sub> value of compound **11a**, which is considered the most potent analogue among all the synthesized pyridazines ( $IC_{50}$ ) = 179  $\mu$ M) is compatible to its experimentally observed value (IC<sub>50</sub> = 177  $\mu$ M) with minor error (error "difference between the observed and predicted values" = -2). The same words for the second most potent pyridazine analogue 8a (IC<sub>50</sub> = 198, 210  $\mu$ M for the observed and predicted values, respectively, error = -12). Compounds 5a, 6c, 8b, 10c, 11c, 12a, 12c and 12e which exhibit experimental potency higher than that of doxazosin mesylate (standard reference used), reveal predicted properties correlated with their biodata (IC<sub>50</sub> = 206, 205, 213, 225, 207, 209, 217, 205; 214, 213, 216, 232, 198, 215, 219, 201 for the observed and predicted values of 5a, 6c, 8b, 10c, 11c, 12a, 12c and 12e  $\mu$ M, respectively). Although high error values due to the observed and predicted vasodilation properties of compounds 12d and 12h (which are considered high potent analogue) are slightly high, their bio-properties are still preserved (IC<sub>50</sub> = 200, 221; 214, 244  $\mu$ M for the observed and predicted values of 12d and 12h, respectively; error = -14, -23 for compounds 12d and 12h, respectively). Additionally, correlated predicted bio-data were also revealed due to the QSAR modeling by the promising vasodilation active pyridazines (IC<sub>50</sub> = 230-323, 226-331  $\mu$ M for the observed and predicted values) of observed potencies lower than that of doxazosin mesylate.

#### 2.3.5.2. External validation

Compounds **7b** and **10d** were used as an external test set not only for validating the attained QSAR model but also for examining its predicative ability. The test set

analogues experimentally exhibit higher and lower vasodilation potencies relative to the used standard reference (doxazosin mesylate). The variation in potency can indicate the predication capabilities of the attained BMLR-QSAR model. Table 4 reveals the observed and predicted IC<sub>50</sub> values of the test set compounds. From the observed data, it has been noticed that compound **10d** which is considered a high potent agent (IC<sub>50</sub> = 216  $\mu$ M) reveals a predicted IC<sub>50</sub> = 214  $\mu$ M with minimum error value = 2. Compounds **7b** that is considered a low potent vasorelaxant active analogue (IC<sub>50</sub> = 300  $\mu$ M) afforded predicted IC<sub>50</sub> = 283  $\mu$ M (error value = 17).

All the above internal and external validation observations give good sign for the predictive capability of the attained BMLR-QSAR model and support the previous statement concerning its predictive power due to the statistical values.

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Table 1. Observed and predicted vasorelaxant activity (IC50,  $\mu$ M) of the training set compounds 5a-d, 6a-d, 7a,c, 8a-d, 9a-d, 10a-c,11a-c and 12a-h and doxazosin mesylate.

				HN-R	R. C.		
		X N-N		N	x N-N	IH	
		5-9, 7a-c, 10a-d	11а-с		12a-h		
Entry	Compd.	R		X	Observed IC <sub>50</sub> , µM	Predicted IC <sub>50</sub> , µM	Error
1	5a	NHNH <sub>2</sub>		Н	206	214	-8
2	5b	NHNH <sub>2</sub>		4-C1	275	257	18
3	5c	NHNH <sub>2</sub>		2-OCH <sub>3</sub>	279	259	20
4	5d	NHNH <sub>2</sub>	R	4-OCH <sub>3</sub>	235	252	-17
5	6a	1-imidazolyl		Н	273	256	17
6	6b	1-imidazolyl		4-C1	255	257	-2
7	6с	1-imidazolyl		2-OCH <sub>3</sub>	205	213	-8
8	6d	1-imidazolyl		4-OCH <sub>3</sub>	259	269	-10

9	7a	1-pyrrolidinyl	Н	235	253	-18
10	7c	1-pyrrolidinyl	2-OCH <sub>3</sub>	284	253	31
11	8a	3,5-dimethyl-1 <i>H</i> -Pyrazol-1-yl	Н	198	210	-12
12	8b	3,5-dimethyl-1 <i>H</i> -Pyrazol-1-yl	4-C1	213	216	-3
13	8c	3,5-dimethyl-1 <i>H</i> -Pyrazol-1-yl	2-OCH <sub>3</sub>	299	316	-17
14	8d	3,5-dimethyl-1 <i>H</i> -Pyrazol-1-yl	4-OCH <sub>3</sub>	323	315	8
15	9a	5-amino-4-cyano-1 <i>H</i> -pyrazol-1-yl	H	287	291	-4
16	9b	5-amino-4-cyano-1 <i>H</i> -pyrazol-1-yl	4-C1	310	276	34
17	9c	5-amino-4-cyano-1 <i>H</i> -pyrazol-1-yl	2-OCH <sub>3</sub>	316	331	-15
18	9d	5-amino-4-cyano-1 <i>H</i> -pyrazol-1-yl	4-OCH <sub>3</sub>	260	235	25
19	10a	5-amino-4-ethoxycarbonyl-1 <i>H</i> -pyrazol-1-yl	Н	254	248	6
20	10b	5-amino-4-ethoxycarbonyl-1 <i>H</i> -pyrazol-1-yl	4-C1	230	268	-38
21	10c	5-amino-4-ethoxycarbonyl-1 <i>H</i> -pyrazol-1-yl	2-OCH <sub>3</sub>	225	232	-7
22	11a	C <sub>6</sub> H <sub>11</sub>	4-Cl	177	179	-2

23	11b	$C_{6}H_{11}$	2-OCH <sub>3</sub>	266	244	22
24	11c	$C_6H_5$	4-Cl	207	198	9
25	12a	4-CF <sub>3</sub>	4-Cl	209	215	-6
26	12b	4-CF <sub>3</sub>	2-OCH <sub>3</sub>	248	226	22
27	12c	4-OCH <sub>3</sub>	4-Cl	217	219	-2
28	12d	4-OCH <sub>3</sub>	2-OCH <sub>3</sub>	200	214	-14
29	12e	2,4- (OCH <sub>3</sub> ) <sub>2</sub>	4-Cl	205	201	4
30	12f	2,4- (OCH <sub>3</sub> ) <sub>2</sub>	2-OCH <sub>3</sub>	259	230	29
31	12g	3,4,5-(OCH <sub>3</sub> ) <sub>3</sub>	4-C1	255	262	-7
32	12h	3,4,5-(OCH <sub>3</sub> ) <sub>3</sub>	2-OCH <sub>3</sub>	221	244	-23
33	Doxazosin mesvlate			226		

N = 32	N = 32, n = 6, $R^2$ = 0.812, $R^2$ cvOO = 0.715, $R^2$ cvMO = 0.721, F = 17.971, $s^2$ = 9.652 x 10 <sup>-8</sup>								
Entry	ID	Coefficient	S	t	Descriptor				
1	0	0.0096	0.0009	10.1478	Intercept				
2	$D_1$	0.0417	0.0065	6.4616	FHACA Fractional HACA (HACA/TMSA) (MOPAC PC)				
3	$D_2$	5.4842 x 10 <sup>-5</sup>	1.1726 x 10 <sup>-5</sup>	4.6771	Shadow plane YZ				
4	$D_3$	3.3666 x 10 <sup>-5</sup>	8.1877 x 10 <sup>-6</sup>	4.1118	HA dependent HDSA-1 (Zefirov PC) (all)				
5	$D_4$	-0.0245	0.0050	-4.9294	HA dependent HDSA-1/TMSA (MOPAC PC) (all)				
6	$D_5$	-0.0002	3.5491 x 10 <sup>-5</sup>	-5.7082	Structural information content (order 2)				
7	$D_6$	-0.0662	0.0105	-6.2774	Max. nucleoph. React. Index for atom C				
1/IC <sub>50</sub>	$\frac{1}{10} = 0.0096 + (0.0417 \text{ x } D_1) + [(5.4842 \text{ x } 10^{-5}) \text{ x } D_2] + [(3.3666 \text{ x } 10^{-5}) \text{ x } D_3] - (0.0245 \text{ x } D_4) - (0.0002 \text{ x } D_5) - (0.0662 \text{ x } D_6)$								

Table 2. Descriptor of the BMLR-QSAR model for the vasodilatory active pyridazines.

CERTE

Entry	Compd	Descriptors*							
Elluy	Compa.	$D_l$	$D_2$	$D_3$	$D_4$	$D_5$	$D_6$		
1	5a	0.01996	20.8	101.1514	0.22673	19.1222	0.01396		
2	5b	0.01899	20.42	117.851	0.27135	20.51033	0.01251		
3	5c	0	32.48	134.0734	0.25586	24.09871	0.01404		
4	5d	0.02653	20.86	127.3935	0.25483	23.68268	0.0177		
5	6a	0.01231	21.58	92.5631	0.20305	22.71714	0.01462		
6	6b	0.01147	21.5	111.6483	0.22665	23.63504	0.01243		
7	6c	0.01273	33.52	127.8707	0.21346	27.61804	0.01192		
8	6d	0.01789	21.4	126.9164	0.22294	27.21434	0.0172		
9	7a	0	31.84	62.98108	0.14203	23.92709	0.01841		
10	7c	0	38.8	101.1514	0.16894	28.89063	0.01856		
11	<b>8</b> a	0	35.94	83.97477	0.14995	25.34215	0.01317		
12	8b	0	31.84	114.5111	0.18407	26.60386	0.01089		
13	8c	0	36.62	97.33439	0.1944	30.27769	0.01304		
14	8d	0	30.14	115.4653	0.19416	29.89377	0.01807		
15	9a	0.02014	27.86	109.7398	0.20324	25.44222	0.03215		
16	9b	0.02249	28.08	109.7398	0.19697	26.73928	0.02933		
17	9c	0.02006	35.28	164.6096	0.27947	30.31351	0.02928		
18	9d	0.02829	28.34	165.5639	0.27831	29.92039	0.01221		
19	10a	0.03798	33.08	132.1648	0.21973	32.45125	0.02267		
20	10b	0.03055	32.76	162.224	0.25276	33.66402	0.02149		
21	10c	0.03443	39.2	176.0607	0.25226	37.26924	0.01691		
22	11a	0.03008	54.34	107.3541	0.16984	33.74725	0.01374		
23	11b	0.02384	38.48	134.5505	0.18985	37.48443	0.01397		
24	11c	0.02546	45.86	91.60884	0.14897	29.75213	0.0236		
25	12a	0.00534	33.04	74.90931	0.11377	29.853	0.01069		
26	12b	0.0079	42	153.6357	0.22227	33.79116	0.01112		
27	12c	0.01256	35.1	127.8707	0.19712	30.91076	0.01098		

**Table 3.** Molecular descriptor values of the BMLR-QSAR model for the vasodilatory active pyridazines.

28	12d	0.01251	41.26	129.3021	0.19786	32.49193	0.01013
29	12e	0.01348	43.2	167.4724	0.22933	34.16959	0.01059
30	12f	0.018	40.9	151.25	0.22414	34.81357	0.01263
31	12g	0.01261	41.94	171.7666	0.24847	35.55921	0.01719
32	12h	0.01479	46.84	176.0607	0.25802	36.45079	0.01442

\*  $D_1$  = FHACA Fractional HACA (HACA/TMSA) (MOPAC PC),  $D_2$  = Shadow plane YZ,  $D_3$  = HA dependent HDSA-1 (Zefirov PC) (all),  $D_4$  = HA dependent HDSA-1/TMSA (MOPAC PC) (all),  $D_5$  = Structural information content (order 2),  $D_6$  = Max. nucleoph. React. Index for atom C.

Entry	Comnd	Observed	Predicted	Error	Descriptors*							
	Compu.	$IC_{50}, \mu M$	$IC_{50}, \mu M$	EIIOI	$D_I$	$D_2$	$D_3$	$D_4$	$D_5$	$D_6$		
1	7b	300	283	17	0	25.32	72.52367	0.15211	25.2	0.01664		
2	10d	216	214	2	0.03495	32.34	178.4464	0.24968	36.89834	0.00912		

 Table 4. Observed, predicated and molecular descriptor values of external test set compounds 7b and 10d according to the BMLR 

 QSAR model.

\*  $D_1$  = FHACA Fractional HACA (HACA/TMSA) (MOPAC PC),  $D_2$  = Shadow plane YZ,  $D_3$  = HA dependent HDSA-1 (Zefirov PC) (all),  $D_4$  = HA dependent HDSA-1/TMSA (MOPAC PC) (all),  $D_5$  = Structural information content (order 2),  $D_6$  = Max. nucleoph. React. Index for atom C.

#### **3-** Conclusion

In summary, different pyridazines substituted with different aryl/heterocyclic moieties at position 3 and 6 were synthesized by facile methods. All the targeted compounds were screened for their in vitro vasorelaxant activity using isolated thoracic aortic rings of rats precontracted with norepinephrine hydrochloride standard reported procedure and compared with doxazosin mesylate, which was used as a reference standard. All synthesized pyridazine derivatives exhibited remarkable vasodilating activity with IC<sub>50</sub> values ranges from 177-323 µM compared to the reference standard doxazosin mesylate (IC<sub>50</sub> = 226  $\mu$ M). Compounds **8a** and **11a** exerted the highest activity with IC<sub>50</sub> values 198 and 177  $\mu$ M, respectively, that, they may represent promising leads for future optimization. 2D-QSAR study was undertaken utilizing CODESSA-Pro software in order to validate the vasorelaxant observed bio-data and determine the most important parameters controlling bio-activity. Statistically significant robust QSAR model describing the pyridazines bio-properties was obtained. External validation technique utilizing high and promising potent synthesized agents, support the predictive power of the attained QSAR model in addition to the internal validation parameters. Homogeneity of the training set analogues (the same chemical scaffold) may be the main factor for the success of the QSAR model.

#### 4. Experimental

#### 4.1. Chemistry

Melting points were recorded on Stuart SMP10 digital melting point apparatus. IR spectra (KBr disc) were recorded on a Shimadzu FT-IR 8400S infrared spectrophotometer. NMR spectra were recorded on a Bruker Ascend 400/ R (<sup>1</sup>H: 400, <sup>13</sup>C: 100 MHz) spectrometer. Some <sup>13</sup>C NMR spectra were recorded on a Varian Mercury VX 300 spectrometer (75 MHz). Mass spectra were recorded on a Shimadzu GCMS-QP 1000 EX (EI, 70 eV) spectrometer. Elemental analyses were carried out at the Regional center for Mycology and Biotechnology, Al-Azhar University, Egypt. Compounds **3a–d**, **4a–d** and **5a–d** were prepared according to the reported procedures [32,33].

#### 4.1.1. General procedure for preparation of 6a–d and 7a–c

An equimolar amount of the appropriate secondary amine (2 mmol) and sodium hydride (60% in mineral oil) in dry dimethylformamide (5 ml) was stirred at R.T. for 1 h. The appropriate chloropyridazine derivative 4a-d was added and the mixture was heated at 100°C for 6 h. The reaction mixture was cooled and poured on ice water. The obtained precipitate was filtered off and washed with water and crystallized from ethanol.

## 4.1.1.1. 3-(1H-Imidazol-1-yl)-6-phenylpyridazine 6a

Yield 78% (0.43 g), mp 155-157°C. IR:  $v_{max}$ / cm<sup>-1</sup> 3059, 1589, 1562, 1481. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  7.31 (br s, 1H, imidazole proton), 7.53-7.59 (m, 3H, arom. protons), 7.68 (d, 1H, *J*= 9.20 Hz, pyridazine proton), 7.83 (br s, 1H, imidazole proton), 8.04 (d, 1H, *J*= 9.20 Hz, pyridazine proton), 8.10-8.12 (m, 2H, arom. protons), 8.53 (br s, 1H, imidazole proton). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 100 MHz):  $\delta$  116.1, 117.4, 126.6, 126.9, 129.2, 130.5, 131.4, 134.9, 135.1, 150.8, 158.5 (arom. carbons). Anal.Calcd. for C<sub>13</sub>H<sub>10</sub>N<sub>4</sub> (222.25): C, 70.26; H, 4.54; N, 25.21. Found: C, 70.49; H, 4.61; N, 25.42.

# 4.1.1.2. 3-(4-Chlorophenyl)-6-(1H-imidazol-1-yl)pyridazine 6b

Yield 88% (0.50 g), mp 206-208° C. IR:  $v_{max}$ / cm<sup>-1</sup> 3042, 1593, 1557, 1489, 1441. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  7.32 (br s, 1H, imidazole proton), 7.56 (d, 2H, *J*= 8.64 Hz, arom. protons), 7.67 (d, 1H, *J*= 9.20 Hz, pyridazine proton), 7.83 (t, 1H, *J*=1.24 Hz, imidazole proton), 8.02 (d, 1H, *J*= 9.20 Hz, pyridazine proton), 8.07 (d, 2H, *J*= 8.64 Hz, arom. protons), 8.51 (br s, 1H, imidazole proton). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 75 MHz):  $\delta$  115.9, 117.3, 125.8, 126.2, 128.0, 129.4, 131.5, 134.7, 138.4, 158.9, 159.0 (arom. carbons). MS: m/z (%) 256 (M<sup>+</sup>, 100), 258 (M<sup>+</sup>+2, 34). Anal.Calcd. for C<sub>13</sub>H<sub>9</sub>ClN<sub>4</sub> (256.69): C, 60.83; H, 3.53; N, 21.83. Found: C, 60.98; H, 3.50; N, 21.98.

# 4.1.1.3. 3-(1*H*-Imidazol-1-yl)-6-(2-methoxyphenyl)pyridazine 6c

Yield 80% (0.46 g), mp 159-161°C. IR:  $v_{max}$ / cm<sup>-1</sup> 3050, 2955, 2924, 2853, 1601, 1560, 1491, 1481, 1443. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  3.92 (s, 3H, OCH<sub>3</sub>), 7.08 (d, 1H, *J*= 8.28 Hz, arom. proton), 7.18 (dt, 1H, *J*= 0.74, 7.52, 11.24 Hz, arom.proton), 7.32 (br s, 1H, imidazole proton), 7.49 (dt, 1H, *J*=1.36, 7.86, 10.69 Hz, arom. proton), 7.59 (d, 1H,

J= 9.20 Hz, pyridazine proton), 7.83 (br s, 1H, imidazole proton), 8.04 (dd, 1H, J= 1.68, 7.68 Hz, arom. proton), 8.24 (d, 1H, J= 9.20 Hz, pyridazine proton), 8.49 (br s, 1H, imidazole proton). MS: m/z (%) 252 (M<sup>+</sup>, 34), 253 (M<sup>+</sup>+1, 7), 185 (100). Anal.Calcd. for C<sub>14</sub>H<sub>12</sub>N<sub>4</sub>O (252.27): C, 66.65; H, 4.79; N, 22.21. Found: C, 66.98; H, 4.84; N, 22.37.

## 4.1.1.4. 3-(1H-Imidazol-1-yl)-6-(4-methoxyphenyl)pyridazine 6d

Yield 60% (0.27 g), mp 178-180°C. IR:  $v_{max}$ / cm<sup>-1</sup> 3049-3009, 2965-2837, 1607, 1582, 1508, 1487. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  3.91 (s, 3H, OCH<sub>3</sub>), 7.07 (d, 2H, *J*= 8.84 Hz, arom. protons), 7.30 (br s, 1H, imidazole proton), 7.62 (d, 1H, *J*= 9.20 Hz, pyridazine proton), 7.81 (br s, 1H, imidazole proton), 7.97 (d, 1H, *J*= 9.24 Hz, pyridazine proton), 8.07 (d, 2H, *J*= 8.84 Hz, arom. proton), 8.49 (br s, 1H, imidazole proton). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 100 MHz):  $\delta$  55.46 (OCH<sub>3</sub>), 114.6, 116.0, 117.4, 125.9, 127.6, 128.3, 131.4, 134.8, 150.4, 158.1, 161.65 (arom. carbons). Anal.Calcd. for C<sub>14</sub>H<sub>12</sub>N<sub>4</sub>O (252.27): C, 66.65; H, 4.79; N, 22.21. Found: C, 66.81; H, 4.85; N, 22.43.

## 4.1.1.5. 6-(Pyrrolidin-1-yl)phenylpyridazine 7a

Yield 65 % (0.39 g), mp 135-137°C. IR:  $v_{max}$ / cm<sup>-1</sup> 3044, 2924, 2851, 1605, 1543, 1462. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  2.07-2.11 (m, 4H, pyrrolidine protons), 3.65 (br s, 4H, pyrrolidine protons), 6.80 (d, 1H, *J*= 9.44 Hz, pyridazine proton), 7.38-7.42 (m, 1H, arom. proton), 7.45-7.49 (m, 2H, arom. protons), 7.69 (d, 1H, *J*= 9.48 Hz, pyridazine proton), 8.00-8.06 (m, 2H, arom. protons). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 100 MHz):  $\delta$  25.4, 46.9 (pyrrolidine carbons), 119.7, 125.7, 126.3, 127.4, 128.3, 128.7, 129.1, 131.6, 141.8, 149.5 (arom. carbons). MS: m/z (%) 259 (M<sup>+</sup>, 69), 261 (M<sup>+</sup>+2, 23), 70 (100). Anal.Calcd. for C<sub>14</sub>H<sub>15</sub>N<sub>3</sub> (225.30): C, 74.64; H, 6.71; N, 18.65. Found: C, 74.81; H, 6.79; N, 18.84.

#### 4.1.1.6. 3-(4-Chlorophenyl)-6-(pyrrolidin-1-yl)pyridazine 7b

Yield 66% (0.39 g), mp 213-215°C. IR:  $v_{max}$ / cm<sup>-1</sup> 3042, 2957, 2855, 1605, 1549, 1487, 1458. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  2.06-2.12 (m, 4H, pyrrolidine protons), 3.63 (br s, 4H, pyrrolidine protons), 6.70 (d, 1H, J= 9.48 Hz, pyridazine proton), 7.43 (d, 2H, J= 8.52 Hz, arom. protons), 7.58 (d, 1H, J= 9.44 Hz, pyridazine proton), 7.94 (d, 2H, J= 8.56 Hz, arom. protons). MS: m/z (%) 259 (M<sup>+</sup>, 69), 261 (M<sup>+</sup>+2, 23), 70 (100).

Anal.Calcd. for C<sub>14</sub>H<sub>14</sub>ClN<sub>3</sub> (259.73): C, 64.74; H, 5.43; N, 16.18. Found: C, 64.85; H, 5.52; N, 16.26.

4.1.1.7. 3-(2-Methoxyphenyl)-6-(pyrrolidin-1-yl)pyridazine 7c

Yield 57% (0.33 g), mp 130-132°C. IR:  $v_{max}$ / cm<sup>-1</sup> 3055, 2972, 2849, 1603, 1541, 1493, 1470, 1456. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  1.98-2.01 (m, 4H, pyrrolidine protons), 3.56 (t, 4H, *J*= 6.70 Hz, pyrrolidine protons), 3.77 (s, 3H, OCH<sub>3</sub>), 6.64 (d, 1H, *J*= 9.44 Hz, pyridazine proton), 6.91 (d, 1H, *J*= 8.08 Hz, arom. proton), 6.97-7.02 (m, 1H, arom.proton), 7.29 (t, 1H, *J*= 7.20 Hz, arom. proton), 7.73 (d, 1H, *J*= 9.44 Hz, pyridazine proton), 7.85 (dd, 1H, *J*= 1.24, 7.60 Hz, arom. proton). Anal.Calcd. for C<sub>15</sub>H<sub>17</sub>N<sub>3</sub>O (255.31): C, 70.56; H, 6.71; N, 16.46. Found: C, 70.81; H, 6.78; N, 16.63.

4.1.2. General procedure for the preparation of 8a-d

A solution of the hydrazide derivative 5a-d (2.15 mmol) and acetylacetone (2.30 mmol) in absolute ethanol (5 ml) was heated under reflux for 2 h. The crystals separated on cooling were filtered off, washed and recrystallized from ethanol.

4.1.2.1. 3-(3,5-Dimethyl-1*H*-pyrazol-1-yl)-6-phenylpyridazine **8a** Yield 41% (0.22 g), mp 109-111°C (reported 106-107°C) [54].

4.1.2.2. 3-(4-Chlorophenyl)6-(3,5-dimethyl-1*H*-pyrazol-1-yl)pyridazine **8b** Yield 60% (0.36 g), mp 177-179°C (reported 174-175°C) [55].

4.1.2.3. 3-(3,5-Dimethyl-1*H*-pyrazol-1-yl)-6-(2-methoxyphenyl)pyridazine **8c** Yield 64% (0.39 g), mp 114-116°C. IR:  $v_{max}$ / cm<sup>-1</sup> 3005, 2978, 2924, 2839, 1605, 1575, 1543, 1435. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  2.34 (s, 3H, CH<sub>3</sub>), 2.81 (s, 3H, CH<sub>3</sub>), 3.89 (s, 3H, OCH<sub>3</sub>), 6.08 (s, 1H, pyrazole proton), 7.05 (d, 1H, *J*= 8.28 Hz, arom. proton), 7.15 (t, 1H, *J*= 7.20 Hz, arom. proton), 7.45-7.49 (m, 1H, arom. proton), 8.01 (dd, 1H, *J*= 1.68, 7.64 Hz, arom. proton), 8.13-8.18 (m, 2H, pyridazine protons. Anal.Calcd. for C<sub>16</sub>H<sub>16</sub>N<sub>4</sub>O (280.33): C, 68.55; H, 5.75; N, 19.99. Found: C, 68.78; H, 5.83; N, 20.17.

# 4.1.2.4. 3-(3,5-Dimethyl-1*H*-pyrazol-1-yl)-6-(4-methoxyphenyl)pyridazine **8d**

Yield 65% (0.39 g), mp 148-150°C. IR:  $v_{max}$ / cm<sup>-1</sup> 3003, 2986, 2970, 2924, 2837, 1609, 1571, 1543, 1516, 1468. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  2.34 (s, 3H, CH<sub>3</sub>), 2.82 (s, 3H, CH<sub>3</sub>), 3.90 (s, 3H, OCH<sub>3</sub>), 6.09 (s, 1H, pyrazole proton), 7.06 (d, 2H, *J*= 8.84 Hz, arom. proton), 7.92 (d, 1H, *J*= 9.28 Hz, pyridazine proton), 8.08 (d, 2H, *J*= 8.84 Hz, arom. protons), 8.18 (d, 1H, *J*= 9.28 Hz, pyridazine proton). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 100 MHz):  $\delta$  13.7 (CH<sub>3</sub>), 15.0 (CH<sub>3</sub>), 55.4 (OCH<sub>3</sub>), 110.0, 114.4, 120.5, 125.5, 128.2, 128.3, 142.4, 151.2, 155.4, 156.6, 161.3 (arom. carbons). Anal.Calcd. for C<sub>16</sub>H<sub>16</sub>N<sub>4</sub>O (280.33): C, 68.55; H, 5.75; N, 19.99. Found: C, 68.82; H, 5.81; N, 20.23.

#### 4.1.3. General procedure for the preparation of **9a–d** and **10a–d**

An equimolar mixture of the hydrazine derivative **5a-d** and ethoxymethylene malononitrile or ethyl ethoxymethylene cyanoacetate (2.30 mmol) in absolute ethanol (5ml) was refluxed for 2 h. The formed precipitate was filtered off on hot and washed with ethanol to give **9a-d** or **10a-d**, respectively in pure form.

4.1.3.1. 5-Amino-1-(6-phenylpyridazin-3-yl)-1*H*-pyrazole-4-carbonitrile **9a** Yield 84% (0.51 g), mp 254-255°C (reported 251-252°C) [56].

4.1.3.2. 5-Amino-1-[6-(4-chlorophenyl)pyridazin-3-yl]-1*H*-pyrazole-4-carbonitrile **9b** Yield 75% (0.51 g), mp 294-296°C. IR:  $v_{max}$ / cm<sup>-1</sup> 3397, 3294, 3078, 3063, 2230, 1622, 1595, 1560, 1551, 1524, 1493. <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>, 400 MHz):  $\delta$  7.65 (d, 2H, *J*= 8.60 Hz, arom. protons), 8.04 (s, 1H, pyrazole proton), 8.19-8.23 (m, 3H, 2 arom. protons + pyridazine proton), 8.26 (s, 2H, NH<sub>2</sub> exchanged with D<sub>2</sub>O), 8.49 (d, 1H, *J*= 9.40 Hz, pyridazine proton). <sup>13</sup>C NMR (DMSO-*d*<sub>6</sub>, 100 MHz):  $\delta$  74.1 (*C*-CN), 114.6 (CN), 120.0, 128.4, 129.0, 130.0, 134.4, 135.7, 144.4, 153.6, 156.1, 156.8 (arom. carbons). Anal.Calcd. for C<sub>14</sub>H<sub>9</sub>ClN<sub>6</sub> (296.72): C, 56.67; H, 3.06; N, 28.32. Found: C, 56.84; H, 3.04; N, 28.49. 4.1.3.3. 5-Amino-1-[6-(2-methoxyphenyl)pyridazin-3-yl]-1*H*-pyrazole-4-carbonitrile **9c** Yield 69% (0.47 g), mp 266-268°C. IR:  $v_{max}$ / cm<sup>-1</sup> 3387, 3285, 3096, 3082, 3069, 3007, 2980, 2947, 2918, 2839, 2218, 1636, 1603, 1584, 1549, 1524, 1495. <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>, 400 MHz):  $\delta$  3.86 (s, 3H, OCH<sub>3</sub>), 7.14 (t, 1H, *J*= 7.46 Hz, arom. proton), 7.23 (d, 1H, *J*= 8.32 Hz, arom. proton), 7.53 (t, 1H, *J*= 7.44 Hz, arom. proton), 7.79 (d, 1H, *J*= 7.24 Hz, arom. proton), 8.02 (s, 1H, pyrazole proton), 8.16 (d, 1H, *J*= 9.32 Hz, pyridazine proton), 8.25 (s, 2H, NH<sub>2</sub> exchanged with D<sub>2</sub>O), 8.28 (d, 1H, *J*= 9.40 Hz, pyridazine proton). Anal.Calcd. for C<sub>15</sub>H<sub>12</sub>N<sub>6</sub>O (292.30): C, 61.64; H, 4.14; N, 28.75. Found: C, 61.87; H, 4.19; N, 28.93.

4.1.3.4. 5-Amino-1-[6-(4-methoxyphenyl)pyridazin-3-yl]-1*H*-pyrazole-4-carbonitrile **9d** Yield 97% (0.65 g), mp 294-295°C. IR:  $v_{max}$ / cm<sup>-1</sup> 3397, 3296, 3078, 3038, 3005, 2970, 2918, 2843, 2210, 1622, 1605, 1580, 1551, 1512, 1449. <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>, 400 MHz):  $\delta$  3.85 (s, 3H, OCH<sub>3</sub>), 7.13 (d, 2H, *J*= 8.80 Hz, arom. protons), 8.02 (s, 1H, pyrazole proton), 8.13-8.17 (m, 3H, 2 arom. protons + pyridazine proton), 8.23 (s, 2H, NH<sub>2</sub> exchanged with D<sub>2</sub>O), 8.42 (d, 1H, *J*= 9.40 Hz, pyridazine proton). Anal.Calcd. for C<sub>15</sub>H<sub>12</sub>N<sub>6</sub>O (292.30): C, 61.64; H, 4.14; N, 28.75. Found: C, 61.83; H, 4.21; N, 28.97.

4.1.3.5. Ethyl 5-amino-1-(6-phenylpyridazin-3-yl)-1*H*-pyrazole-4-carboxylate **10a** Yield 79% (0.56 g), mp 188-190°C (reported 190-191°C) [57].

4.1.3.6. Ethyl 5-amino-1-[6-(4-chlorophenyl)pyridazin-3-yl]-1*H*-pyrazole-4-carboxylate10b

Yield 68% (0.54 g), mp 209-211°C. IR:  $v_{max}$ / cm<sup>-1</sup> 3443, 3333, 3084, 3073, 2986, 2941, 2916, 2903, 1674, 1607, 1593, 1541, 1493. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  1.39 (t, 3H, *J*= 7.12 Hz, CH<sub>3</sub>), 4.34 (q, 2H, *J*= 7.12 Hz, CH<sub>2</sub>), 7.53 (d, 2H, *J*= 8.48 Hz, arom. protons), 7.61 (s, 2H, NH<sub>2</sub> exchanged with D<sub>2</sub>O), 7.84 (s, 1H, pyrazole proton), 7.98-8.02 (m, 3H, 2 arom. protons + pyridazine proton), 8.26 (d, 1H, *J*= 9.32 Hz, pyridazine proton). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 75 MHz):  $\delta$  14.4 (CH<sub>3</sub>), 59.6 (OCH<sub>2</sub>), 119.0, 125.8, 126.4, 127.9, 129.3, 133.8, 136.5, 138.4, 142.7, 152.0, 155.8 (arom. carbons), 164.0 (C=O). MS

(m/z, %): 343 (M<sup>+</sup>, 100), 344 [M<sup>+</sup>+1], 297 (50). Anal.Calcd. for C<sub>16</sub>H<sub>14</sub>ClN<sub>5</sub>O<sub>2</sub> (343.77): C, 55.90; H, 4.11; N, 20.37. Found: C, 56.07; H, 4.15; N, 20.49.

4.1.3.7. Ethyl 5-amino-1-[6-(2-methoxyphenyl)pyridazin-3-yl]-1*H*-pyrazole-4carboxylate **10c** 

Yield 64% (0.50 g), mp 137-139°C. IR:  $v_{max}$ / cm<sup>-1</sup> 3424, 3312, 3080, 2990, 2959, 2918, 2839, 1688, 1618, 1601, 1545, 1493. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  1.37 (t, 3H, *J*= 7.12 Hz, CH<sub>3</sub>), 3.87 (s, 3H, OCH<sub>3</sub>), 4.31 (q, 2H, *J*= 7.10 Hz, CH<sub>2</sub>), 7.03 (d, 1H, *J*= 8.32 Hz, arom. proton), 7.12 (t, 1H, *J*= 7.48 Hz, arom. proton), 7.41 (t, 1H, *J*= 8.48 Hz, arom. proton), 7.62 (s, 2H, NH<sub>2</sub> exchanged with D<sub>2</sub>O), 7.82 (s, 1H, pyrazole proton), 7.90 (dd, 1H, *J*= 1.24, 7.60 Hz, arom. proton), 8.14 (s, 2H, pyridazine protons). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 100 MHz):  $\delta$  14.6 (CH<sub>3</sub>), 55.6 (OCH<sub>3</sub>), 59.7 (CH<sub>2</sub>), 95.5, 111.4, 117.5, 121.3, 125.0, 130.5, 130.8, 131.3, 142.5, 152.1, 156.2, 156.5, 157.1 (arom. carbons), 164.2 (C=O). Anal.Calcd. for C<sub>17</sub>H<sub>17</sub>N<sub>5</sub>O<sub>3</sub> (339.36): C, 60.17; H, 5.05; N, 20.64. Found: C, 60.34; H, 5.08; N, 20.87.

4.1.3.8. Ethyl 5-amino-1-[6-(4-methoxyphenyl)pyridazin-3-yl]-1*H*-pyrazole-4carboxylate **10d** 

Yield 79% (0.62 g), mp 198-200°C. IR:  $v_{max}$ / cm<sup>-1</sup> 3453, 3331, 3082, 3046, 2988, 2965, 2922, 2901, 2833, 1676, 1612, 1580, 1545, 1516. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  1.39 (t, 3H, *J*= 7.12 Hz, CH<sub>3</sub>), 3.90 (s, 3H, OCH<sub>3</sub>), 4.33 (q, 2H, *J*= 7.12 Hz, CH<sub>2</sub>), 7.06 (d, 2H, *J*= 8.84 Hz, arom. protons), 7.60 (br s, 2H, NH<sub>2</sub> exchanged with D<sub>2</sub>O), 7.84 (s, 1H, pyrazole proton), 7.96 (d, 1H, *J*= 9.36 Hz, pyridazine proton), 8.02 (d, 2H, *J*= 8.84 Hz, arom. protons), 8.20 (d, 1H, *J*= 9.36 Hz, pyridazine proton). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 75 MHz):  $\delta$  14.4 (CH<sub>3</sub>), 55.3 (OCH<sub>3</sub>), 59.6 (OCH<sub>2</sub>), 114.5, 118.9, 126.1, 127.8, 128.1, 142.4, 151.9, 156.2, 156.5, 161.4 (arom. carbons), 164.1 (C=O). Anal.Calcd. for C<sub>17</sub>H<sub>17</sub>N<sub>5</sub>O<sub>3</sub> (339.36): C, 60.17; H, 5.05; N, 20.64. Found: C, 60.31; H, 5.12; N, 20.91.

#### 4.1.4. General procedure for the preparation of 11a-c

To a mixture of the hydrazine derivative **5b,c** (1.65 mmol) and triethylamine (0.25 ml) in dry methylene chloride (5 ml), the appropriate isocyanate (1.65 mmol) was added

dropwise. The reaction was stirred at room temperature for 6 h, the obtained precipitate was filtered off, washed and crystallized from ethanol.

# 4.1.4.1. 2-[6-(4-Chlorophenyl)pyridazin-3-yl]-N-cyclohexylhydrazine-1-carboxamide11a

Yield 43% (0.24 g), mp 211-213°C. IR:  $v_{max}$ / cm<sup>-1</sup> 3300, 3242, 3134, 3051, 3034, 3015, 2930, 2853, 1645, 1614, 1601, 1558, 1464. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  1.01-1.09 (m, 2H, cyclohexyl protons), 1.23-1.33 (m, 2H, cyclohexyl protons), 1.51-1.64 (m, 4H, cyclohexyl protons), 1.85 (br d, 2H, *J*= 9.4 Hz, cyclohexyl protons), 3.56-3.66 (m, 1H, cyclohexyl proton), 5.59 (s, 1H, NH exchanged with D<sub>2</sub>O), 5.61 (s, 1H, NH exchanged with D<sub>2</sub>O), 6.77 (s, 1H, NH exchanged with D<sub>2</sub>O), 7.13 (d, 1H, *J*= 9.24 Hz, pyridazine proton), 7.42 (d, 2H, *J*= 8.68 Hz, arom. protons), 7.66 (d, 1H, *J*= 9.24 Hz, pyridazine proton), 7.81 (d, 2H, *J*= 8.28 Hz, arom. protons). Anal.Calcd. for C<sub>17</sub>H<sub>20</sub>ClN<sub>5</sub>O (345.83): C, 59.04; H, 5.83; N, 20.25. Found: C, 59.25; H, 5.91; N, 20.41.

# 4.1.4.2. *N*-Cyclohexyl-2-[6-(2-methoxyphenyl)pyridazin-3-yl]hydrazine-1-carboxamide11b

Yield 74% (0.42 g), mp 197-198°C. IR:  $v_{max}$ / cm<sup>-1</sup> 3306, 3223, 3073, 2999, 2928, 2853, 1659, 1599, 1580, 1541, 1493, 1460. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  1.08 (td, 2H, *J*= 2.87, 11.89, 14.80 Hz, cyclohexyl protons), 1.21-1.31 (m, 2H, cyclohexyl protons), 1.50-1.56 (m, 2H, cyclohexyl protons), 1.61-1.65 (m, 2H, cyclohexyl protons), 1.85 (dd, 2H, *J*= 2.92, 12.16 Hz, cyclohexyl protons), 3.53-3.61 (m, 1H, cyclohexyl proton), 3.78 (s, 3H, OCH<sub>3</sub>), 3.81 (s, 1H, NH exchanged with D<sub>2</sub>O), 3.82 (s, 1H, NH exchanged with D<sub>2</sub>O), 5.95 (s, 1H, NH exchanged with D<sub>2</sub>O), 6.94 (d, 1H, *J*= 8.32 Hz, arom. proton), 7.02 (td, 1H, *J*= 0.83, 7.53, 8.33 Hz), 7.20 (d, 1H, *J*= 9.24 Hz, pyridazine proton), 7.36 (td, 1H, *J*= 1.74, 7.86, 9.12), 7.70 (dd, 1H, *J*= 1.64, 7.60 Hz, arom.proton), 7.88 (d, 1H, *J*= 9.32 Hz, pyridazine proton. <sup>13</sup>C NMR (CDCl<sub>3</sub>, 75 MHz):  $\delta$  24.8, 25.4, 33.5, 48.7 (cyclohexyl carbons), 55.4 (OCH<sub>3</sub>), 111.2, 112.0, 121.1, 125.7, 130.5, 130.7, 138.5, 156.8, 160.0 (arom. carbons), 158.1 (C=O). Anal.Calcd. for C<sub>18</sub>H<sub>23</sub>N<sub>5</sub>O<sub>2</sub> (341.42): C, 63.32; H, 6.79; N, 20.51. Found: C, 63.48; H, 6.85; N, 20.67.

4.1.4.3. 2-[6-(4-Chlorophenyl)pyridazin-3-yl]-*N*-phenylhydrazine-1-carboxamide **11c** Yield 64% (0.35 g), mp 237-239°C. IR:  $v_{max}$ / cm<sup>-1</sup> 3329, 3250, 3138, 3061, 3007, 2984, 1659, 1601, 1597, 1557, 1448. <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>, 400 MHz):  $\delta$  6.95 (t, 1H, *J*=7.34Hz, arom. proton), 7.08 (d, 1H, *J*= 9.36 Hz, pyridazine proton), 7.25 (t, 2H, *J*= 7.94 Hz, arom. protons), 7.51 (d, 2H, *J*= 7.92 Hz, arom. protons), 7.56 (d, 2H, *J*= 8.64 Hz, arom. protons), 8.02 (d, 1H, *J*= 9.36 Hz, pyridazine proton), 8.05 (d, 2H, *J*= 8.68 Hz, arom. protons), 8.34 (s, 1H, NH exchanged with D<sub>2</sub>O), 8.91 (s, 1H, NH exchanged with D<sub>2</sub>O), 8.96 (s, 1H, NH exchanged with D<sub>2</sub>O). <sup>13</sup>C NMR (DMSO-*d*<sub>6</sub>, 75 MHz):  $\delta$  113.2, 118.6, 121.8, 125.6, 127.4, 128.5, 128.8, 135.5, 139.6, 151.1, 160.5 (arom. carbons), 156.1 (C=O). Anal.Calcd. for C<sub>17</sub>H<sub>14</sub>ClN<sub>5</sub>O (339.78): C, 60.09; H, 4.15; N, 20.61. Found: C, 60.18; H, 4.19; N, 20.89.

4.1.5. General procedure for the preparation of 12a-h

To an equimolar mixture of the hydrazine derivative **5b,c** and the appropriate aldehyde in absolute ethanol (25ml) was added glacial acetic acid (0.50 ml). The reaction was heated under reflux for 3 h. The obtained precipitate was filtered off and washed with ethanol to give **12a–h** in a pure form.

4.1.5.1. 3-(4-Chlorophenyl)-6-{2-[4-(trifluoromethyl)benzylidene]hydrazinyl}pyridazine12a

Yield 87% (0.59 g), mp > 300°C. IR:  $v_{max}$ / cm<sup>-1</sup> 3120, 3050, 1611, 1593, 1549, 1489. <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>, 400 MHz):  $\delta$  7.59 (d, 2H, *J*= 7.04 Hz, arom. protons), 7.72-7.79 (m, 3H, 2 arom. protons + pyridazine proton), 7.95 (d, 2H, *J*= 7.24 Hz, arom. protons), 8.11-8.16 (m, 3H, 2 arom. protons + pyridazine proton), 8.23 (s, 1H, N=CH), 11.93 (s, 1H, NH exchanged with D<sub>2</sub>O). Anal.Calcd. for C<sub>18</sub>H<sub>12</sub>ClF<sub>3</sub>N<sub>4</sub>O (376.76): C, 57.38; H, 3.21; N, 14.87. Found: C, 57.49; H, 3.23; N, 15.01.

4.1.5.2.3-(4-Methoxyphenyl)-6-{2-[4-(trifluoromethyl)benzylidene]hydrazinyl}pyridazine 12b

Yield 77% (0.53 g), mp 262-264°C. IR:  $v_{max}$ / cm<sup>-1</sup> 3125, 3060, 2940, 2839, 1611, 1589, 1545, 1493. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  3.83 (s, 3H, OCH<sub>3</sub>), 6.98 (d, 1H, arom.

proton), 7.05 (t, 1H, J= 7.50 Hz, arom. protons), 7.42 (t, 1H, J= 7.74 Hz, arom. protons), 7.60 (d, 3H, J= 7.84 Hz, arom. protons+ pyridazine proton), 7.70 (d, 1H, J= 6.64 Hz, arom. proton), 7.77 (d, 2H, J= 7.92 Hz, arom. protons), 7.95 (s, 1H, NH, exchanged with D<sub>2</sub>O), 8.09 (d, 1H, J= 9.12 Hz, pyridazine proton), 8.39 (s, 1H, N=CH). Anal.Calcd. for C<sub>19</sub>H<sub>15</sub>F<sub>3</sub>N<sub>4</sub>O (372.34): C, 61.29; H, 4.06; N, 15.05. Found: C, 61.43; H, 4.12; N, 15.23.

4.1.5.3. 3-(4-Chlorophenyl)-6-[2-(4-methoxybenzylidene)hydrazinyl]pyridazine **12c** Yield 82% (0.63 g), mp 272-273°C. IR:  $v_{max}$ / cm<sup>-1</sup> 3125, 3060, 2951, 2916, 2835, 1618, 1603, 1551, 1508, 1489. <sup>1</sup>H NMR (DMSO- $d_6$ , 400 MHz):  $\delta$  3.81 (s, 3H, OCH<sub>3</sub>), 7.00 (d, 2H, *J*= 7.36 Hz, arom. proton), 7.56-7.68 (m, 5H, arom. protons), 8.09 (d, 4H, *J*= 8.28 Hz, arom. protons + N=CH), 11.53 (s, 1H, NH, exchanged with D<sub>2</sub>O). Anal.Calcd. for C<sub>18</sub>H<sub>15</sub>ClN<sub>4</sub>O (338.79): C, 63.81; H, 4.46; N, 16.54. Found: C, 63.96; H, 4.53; N, 16.80.

4.1.5.4. 3-[2-(4-Methoxybenzylidene)hydrazinyl] 6-(4-methoxyphenyl)pyridazine **12d** Yield 80% (0.62 g), mp 239-241°C. IR:  $v_{max}$ / cm<sup>-1</sup> 3120, 3005, 2930, 2906, 2831, 1611, 1593, 1541, 1508, 1493. <sup>1</sup>H NMR (DMSO- $d_6$ , 400 MHz):  $\delta$  3.81 (s, 3H, OCH<sub>3</sub>), 3.84 (s, 3H, OCH<sub>3</sub>), 7.00 (d, 2H, J= 8.64 Hz, arom. protons), 7.09 (t, 1H, J= 7.38 Hz, arom. proton), 7.17 (d, 1H, J= 8.24 Hz, arom. proton), 7.43 (t, 1H, J= 7.16 Hz, arom. proton), 7.57 (d, 1H, J= 9.32 Hz, pyridazine proton), 7.65 (d, 2H, J= 8.64 Hz, arom. protons), 7.71 (d, 1H, J= 6.44 Hz, arom. proton), 7.86 (d, 1H, J= 9.36 Hz, pyridazine proton), 8.11 (s, 1H, N=CH), 11.40 (s, 1H, NH, exchanged with D<sub>2</sub>O). <sup>13</sup>C NMR (DMSO- $d_6$ , 75 MHz):  $\delta$  55.2 (OCH<sub>3</sub>), 55.6 (OCH<sub>3</sub>), 111.0, 111.8, 114.2, 120.5, 126.2, 128.2, 129.7, 141.0, 151.5, 156.5, 158.0, 160.0 (arom. carbons). Anal.Calcd. for C<sub>19</sub>H<sub>18</sub>N<sub>4</sub>O<sub>2</sub> (334.38): C, 68.25; H, 5.43; N, 16.76. Found: C, 68.47; H, 5.49; N, 16.94.

4.1.5.5. 3-(4-Chlorophenyl)-6-[2-(2,4-dimethoxybenzylidene)hydrazinyl]pyridazine **12e** Yield 79% (0.53 g), mp 225-226°C. IR:  $v_{max}$ / cm<sup>-1</sup> 3150, 3003, 2968, 2940, 2928, 2835, 1609, 1568, 1539, 1504, 1489. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  3.77 (s, 3H, OCH<sub>3</sub>), 3.79 (s, 3H, OCH<sub>3</sub>), 6.39 (d, 1H, *J*= 1.96 Hz, arom. proton), 6.48 (dd, 1H, *J*= 1.86, 8.66 Hz, arom. proton), 7.39 (d, 2H, *J*= 8.48 Hz, arom. protons), 7.70-7.87 (m, 6H, 2 pyridazine protons + 3 arom. protons + NH exchanged with  $D_2O$ ), 8.43 (s, 1H, N=CH). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 100 MHz):  $\delta$  55.5 (OCH<sub>3</sub>), 55.6 (OCH<sub>3</sub>), 98.1, 105.8, 116.1, 126.5, 127.1, 127.4, 129.1, 135.2, 140.3, 145.0, 153.5, 156.5, 157.0 (arom. carbons). Anal.Calcd. for  $C_{19}H_{17}CIN_4O_2$  (368.82): C, 61.87; H, 4.65; N, 15.19. Found: C, 62.13; H, 4.69; N, 15.35.

4.1.5.6. 3-[2-(2,4-Dimethoxybenzylidene)hydrazinyl]-6-(2-methoxyphenyl)pyridazine12f

Yield 72% (0.48 g), mp 214-216°C. IR:  $v_{max}$ / cm<sup>-1</sup> 3200, 3013, 3003, 2955, 2918, 2897, 2833, 1607, 1566, 1539, 1503, 1491. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  3.73 (s, 3H, OCH<sub>3</sub>), 3.78 (s, 3H, OCH<sub>3</sub>), 3.80 (s, 3H, OCH<sub>3</sub>), 6.38 (d, 1H, *J*= 2.20 Hz, arom. proton), 6.47 (dd, 1H, *J*= 2.16, 8.64 Hz, arom. proton), 6.94 (d, 1H, *J*= 8.28 Hz, arom. proton), 7.02 (t, 1H, *J*= 7.46 Hz, arom. proton), 7.31-7.35 (m, 1H, arom. proton), 7.59 (d, 1H, *J*= 9.40 Hz, pyridazine proton), 7.81-7.84 (m, 4H, pyridazine proton + 2 arom. protons + NH exchanged with D<sub>2</sub>O), 8.36 (s, 1H, N=CH). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 75 MHz):  $\delta$  55.3 (OCH<sub>3</sub>), 55.4 (OCH<sub>3</sub>), 55.5 (OCH<sub>3</sub>), 98.1, 105.4, 111.2, 112.1, 117.0, 121.0, 125.8, 126.8, 129.7, 130.5, 130.6, 138.2, 151.8, 156.9, 158.4, 158.7, 161.6 (arom. carbons). Anal.Calcd. for C<sub>20</sub>H<sub>20</sub>N<sub>4</sub>O<sub>3</sub> (364.41): C, 65.92; H, 5.53; N, 15.38. Found: C, 66.11; H, 5.64; N, 15.52.

# 4.1.5.7. 3-(4-Chlorophenyl)-6-[2-(3,4,5-trimethoxybenzylidene)hydrazinyl]pyridazine12g

Yield 83% (0.60 g), mp 294-296°C. IR:  $v_{max}$ / cm<sup>-1</sup> 3219, 3086, 2997, 2924, 2851, 2833, 1614, 1599, 1539, 1506, 1491. <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400 MHz):  $\delta$  3.70 (s, 3H, OCH<sub>3</sub>), 3.86 (s, 6H, 2OCH<sub>3</sub>), 7.03 (s, 2H, arom. protons), 7.57 (d, 2H, *J*=8.16 Hz, arom. protons), 7.70 (d, 1H, *J*= 9.36 Hz, pyridazine proton), 8.08-8.10 (m, 4H, pyridazine proton + 2 arom. protons + N=CH), 11.70 (s, 1H, NH exchanged with D<sub>2</sub>O). Anal.Calcd. for C<sub>20</sub>H<sub>19</sub>ClN<sub>4</sub>O<sub>3</sub> (398.85): C, 60.23; H, 4.80; N, 14.05. Found: C, 60.47; H, 4.87; N, 14.13.

4.1.5.8. 3-(2-Methoxyphenyl)-6-[2-(3,4,5-trimethoxybenzylidene)hydrazinyl]pyridazine12h

Yield 75% (0.55 g), mp 252-254°C. IR:  $v_{max}$ / cm<sup>-1</sup> 3200, 3086, 2997, 2955, 2926, 2837, 1601, 1589, 1576, 1506, 1491. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  3.81 (s, 6H, 2OCH<sub>3</sub>), 3.82

(s, 6H, 2OCH<sub>3</sub>), 6.88 (s, 2H, arom. protons), 6.97 (d, 1H, J= 8.32 Hz, arom. proton), 7.02 (t, 1H, J= 7.48 Hz, arom. proton), 7.36 (t, 1H, J= 7.14 Hz, arom. proton), 7.68 (d, 1H, J= 9.40 Hz, pyridazine proton), 7.84 (d, 2H, arom. proton + NH exchanged with D<sub>2</sub>O), 7.91 (d, 1H, J= 9.40 Hz, pyridazine proton), 8.38 (s, 1H, N=CH). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 75 MHz):  $\delta$  55.5 (OCH<sub>3</sub>), 56.0 (OCH<sub>3</sub>), 60.8(OCH<sub>3</sub>), 94.6, 103.5, 111.4, 112.2, 120.8, 130.0, 130.3, 130.8, 130.9, 143.2, 144.2, 153.4, 157.5 (arom. carbons). Anal.Calcd. for C<sub>21</sub>H<sub>22</sub>N<sub>4</sub>O<sub>4</sub> (394.42): C, 63.95; H, 5.62; N, 14.20. Found: C, 64.12; H, 5.69; N, 14.37.

#### 4.2. Vasorelaxant activity

The vasorelaxant activity screening procedure was carried out according to the standard reported techniques [34–40] by testing the effects of the synthesized compounds **5a–d**, **6a–d**, **7a–c**, **8a–d**, **9a–d**, **10a–d**, **11a–c**, **12a–h** and the reference standard, doxazosin mesylate, on isolated thoracic aortic rings of male Wister rats (250-350 g). After light ether anesthesia, the rats were sacrificed by cervical dislocation.

The aorta was immediately excised, freed of extraneous tissues and prepared for isometric tension recording. Aorta was cut into (3-5 mm width) rings and each ring was placed in a vertical chamber "10 ml jacketed automatic multi-chamber organ bath system (Model no. ML870B6/C, Panlab, Spain)" filled with Krebs solution composed of (in mM): NaCl, 118.0; KCl, 4.7; NaHCO<sub>3</sub>, 25.0; CaCl<sub>2</sub>, 1.8; NaH<sub>2</sub>PO<sub>4</sub>, 1.2; MgSO<sub>4</sub>, 1.2; glucose, 11.0 and oxygenated with carbogen gas (95% O<sub>2</sub>/5% CO<sub>2</sub>) at 37+0.5°C. Each aortic ring was mounted between two stainless steel hooks passed through its lumen. The lower hook was fixed between two plates, while the upper one was attached to a force displacement transducer (Model no. MLT0201, Panlab, Spain) connected to an amplifier (PowerLab, AD Instruments Pty. Ltd.) which is connected to a computer. The Chart for windows (v 3.4) software was used to record and elaborate data.

Preparations were stabilized under 2 g resting tension during 2 h and then the contracture response to norepinephrine hydrochloride  $(10^{-6} \text{ M})$  was measured before and after exposure to increasing concentrations of the tested synthesized compounds.

The tested compounds were dissolved in dimethylsulfoxide (DMSO) as stock solution (10 ml of 0.01 M). Control experiments were performed in the presence of DMSO alone, at the same concentrations as those used with the derivatives tested, which demonstrated

that the solvent did not affect the contractile response of isolated aorta. The observed vasodilatation activity screening data were reported (Table 1) and the potency ( $IC_{50}$ , concentration necessary for 50% reduction of maximal norepinephrine hydrochloride induced contracture) was determined by the best-fit line technique.

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#### **Figure captions**

Figure 1. Representative pyridazine based compounds with vasorelaxant activity.

Figure 2. General structures of the designed target compounds.

**Figure 3.** BMLR-QSAR plot of correlation representing the observed *vs.* predicted  $1/IC_{50}$  ( $\mu$ M).

#### **Table captions:**

**Table 1.** Observed and predicted vasorelaxant activity (IC<sub>50</sub>,  $\mu$ M) of the training set compounds **5a–d**, **6a–d**, **7a,c**, **8a–d**, **9a–d**, **10a–c**, **11a–c** and **12a–h** and doxazosin mesylate.

Table 2. Descriptor of the BMLR-QSAR model for the vasodilatory active pyridazines.

**Table 3.** Molecular descriptor values of the BMLR-QSAR model for the vasodilatory active pyridazines.

 Table 4. Observed, predicated and molecular descriptor values of external test set compounds 7b and 10d according to the BMLR-QSAR model.



Figure 1. Representative pyridazine based compounds with vasorelaxant activity.



Figure 2. General structures of the designed target compounds.



Figure 3. BMLR-QSAR plot of correlation representing the observed vs. predicted  $1/IC_{50}$ 

(µM).



# Highlights

- Some phenylpyridazines were prepared by facile methods.
- They were screened for their vasorelaxant activity.
- Compounds 8a and 11a revealed promising activity compared to doxazosin mesylate.
- A reliable 2D-QSAR model was obtained to validate the obtained results.

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