

# Pd(0)–Cu(I)-catalyzed cross-coupling of alkynylsilanes with triarylantimony(v) diacetates

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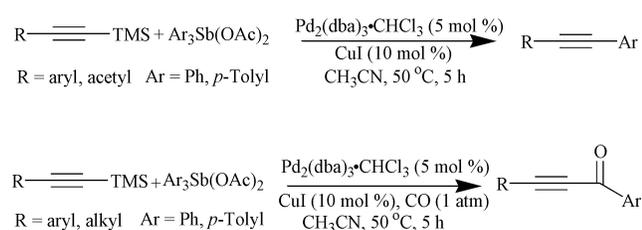
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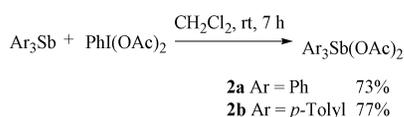
Treatment of alkynylsilanes with triarylantimony diacetates in the presence of Pd<sub>2</sub>(dba)<sub>3</sub>·CHCl<sub>3</sub> (5 mol%) and CuI (10 mol%) in CH<sub>3</sub>CN at 50 °C for 5 h afforded aryl-substituted alkynes in good yield. Alternatively, direct carbonylative coupling of triarylantimony diacetates with alkynylsilanes was accomplished under atmospheric pressure of carbon monoxide.

The palladium-catalyzed cross-coupling reaction of terminal alkynes with aryl or vinyl halides in the presence of a Pd(0)–Cu(I) co-catalyst system, known as Sonogashira and Hagihara coupling,<sup>1</sup> has been widely utilized as a practical synthetic method for carbon–carbon bond formation in organic synthesis. In synthesis, for the introduction of appropriate functionalities, a terminal alkyne moiety is often protected with a trimethylsilyl group. Thus the direct coupling of alkynylsilanes should be more accessible. There have only been a few reports of such direct coupling reactions of alkynylsilanes with aryl halides; in the presence of fluoride ion by Hiyama *et al.*<sup>2</sup> or under basic conditions by Rossi<sup>3</sup> and Huang<sup>4</sup> and their co-workers. Recently Hiyama *et al.*<sup>5</sup> reported a copper(I)-salt mediated direct Pd(0)-cross coupling reaction of alkynylsilanes with triflates in the presence of CuCl in DMF. Nagasaka *et al.*<sup>6</sup> also reported the direct silver-promoted cross-coupling of alkynylsilanes with aryl iodides to form aryl-substituted alkynes. Alternatively, in order to synthesize alkynyl ketones CuCl-catalyzed reaction of alkynylsilanes and acyl halides by one-pot desilylation–coupling was reported by Hosomi *et al.*<sup>7</sup> We have investigated the use of triarylantimony(v) diacetates<sup>8</sup> in palladium-catalyzed cross-coupling reactions; here we wish to report the direct palladium-catalyzed cross-coupling and carbonylative cross-coupling of alkynylsilanes with triarylantimony diacetates to afford the aryl-substituted alkynes and alkynyl ketones (Scheme 1).<sup>9</sup>



Scheme 1

Triarylantimony(v) diacetates **2a**<sup>10</sup> and **2b**<sup>11</sup> were prepared by reaction of triarylantimony(III) with PhI(OAc)<sub>2</sub> by stirring in CH<sub>2</sub>Cl<sub>2</sub> at room temperature for 7 h (Scheme 2).

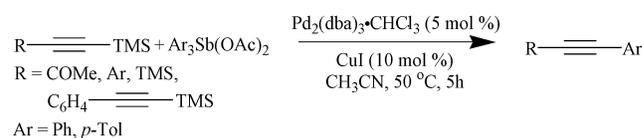


Scheme 2

## Results and discussion

### 1 Pd(0)–Cu(I)-catalyzed cross-coupling of alkynylsilanes with organoantimony(v) compounds

The results of the palladium–copper catalyzed cross-coupling of alkynylsilanes with triarylantimony diacetates are summarized in Scheme 3 and Table 1.



Scheme 3

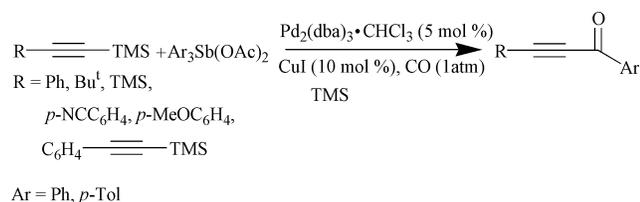
The (phenylethynyl)trimethylsilane (**1a**) reacted with triphenylantimony diacetate (**2a**) in the presence of Pd<sub>2</sub>(dba)<sub>3</sub>·CHCl<sub>3</sub> (5 mol%) and CuI (10 mol%) in CH<sub>3</sub>CN at 50 °C for 5 h to afford 1,2-diphenylacetylene (**3a**) in 81% yield (entry 1 in Table 1). No homocoupling reaction of alkynylsilanes<sup>12</sup> was observed using Ar<sub>3</sub>Sb(OAc)<sub>2</sub>. The addition of CuI is critical and improved the yield. Of the catalysts (Pd<sub>2</sub>(dba)<sub>3</sub>·CHCl<sub>3</sub>, PdCl<sub>2</sub>, (π-allyl)<sub>2</sub>PdCl<sub>2</sub>, Pd<sub>2</sub>(dba)<sub>3</sub>, PdCl<sub>2</sub>(PhCN)<sub>2</sub>) tested, Pd<sub>2</sub>(dba)<sub>3</sub>·CHCl<sub>3</sub> was the best choice. Among the solvents (DMF, CH<sub>3</sub>CN, toluene and DME) tested, CH<sub>3</sub>CN was the most suitable at 50 °C. Under the same conditions the reaction of **1a** with tri-*p*-tolylantimony diacetate (**2b**) gave 1-phenyl-2-*p*-tolylacetylene (**3b**<sup>13a</sup>) in 80% yield (entry 2). For the aryl-substituted silanes **1b** and **1c**, triphenylantimony diacetate (**2a**) was successfully coupled to give **3c**<sup>12c</sup> and **3d**<sup>13b</sup> under the same conditions in 73 and 70% yields, respectively (entries 3 and 4). The acetyl-substituted ethynylsilane **1d** was treated with **2a** and **2b** to provide **3e** and **3f** in 85 and 82% yields, respectively (entries 5 and 6). When the 1,2-bis(trimethylsilyl)ethyne (**1e**) was reacted with **2a** and **2b**, the disubstituted acetylene **3a** and **3g**<sup>13c</sup> were afforded in 80 and 77% yields, respectively (entries 7 and 8). Finally for the 1,4-bis(trimethylsilyl)ethynylbenzene (**1f**), disubstituted product **3h**<sup>13d</sup> was obtained using Pd<sub>2</sub>(dba)<sub>3</sub>·CHCl<sub>3</sub> as a catalyst (entry 9). However, when PdCl<sub>2</sub> was employed as a catalyst, the mono-substituted product **3i**<sup>13e</sup> was afforded in 42% yield (entry 10).

### 2 Pd(0)–Cu(I)-catalyzed carbonylative cross-coupling of alkynylsilanes with organoantimony(v) compounds

This coupling was extended to carbonylative cross-coupling of silanes with antimony(v) compounds. The results are summarized in Scheme 4 and Table 2.

**Table 1** Pd(0)–Cu(I)-catalyzed cross-coupling of alkynylsilanes with triarylantimony diacetates

Entry	Silanes	Triarylantimony diacetates	Product	Yield (%)
1	Ph—C≡C—TMS <b>1a</b>	Ph <sub>3</sub> Sb(OAc) <sub>2</sub> <b>2a</b>	Ph—C≡C—Ph <b>3a</b>	81
2	<b>1a</b>	( <i>p</i> -Tol) <sub>3</sub> Sb(OAc) <sub>2</sub> <b>2b</b>	Ph—C≡C—C <sub>6</sub> H <sub>4</sub> — <b>3b</b>	80
3	MeO—C <sub>6</sub> H <sub>4</sub> —C≡C—TMS <b>1b</b>	<b>2a</b>	MeO—C <sub>6</sub> H <sub>4</sub> —C≡C—Ph <b>3c</b>	73
4	NC—C <sub>6</sub> H <sub>4</sub> —C≡C—TMS <b>1c</b>	<b>2a</b>	NC—C <sub>6</sub> H <sub>4</sub> —C≡C—Ph <b>3d</b>	70
5	H <sub>3</sub> C—C(=O)—C≡C—TMS <b>1d</b>	<b>2a</b>	H <sub>3</sub> C—C(=O)—C≡C—Ph <b>3e</b>	85
6	<b>1d</b>	<b>2b</b>	H <sub>3</sub> C—C(=O)—C≡C—C <sub>6</sub> H <sub>4</sub> — <b>3f</b>	82
7	TMS—C≡C—TMS <b>1e</b>	<b>2a</b>	Ph—C≡C—Ph <b>3a</b>	77
8	<b>1e</b>	<b>2b</b>	—C <sub>6</sub> H <sub>4</sub> —C≡C—C <sub>6</sub> H <sub>4</sub> — <b>3g</b>	80
9	TMS—C≡C—C <sub>6</sub> H <sub>4</sub> —C≡C—TMS <b>1f</b>	<b>2a</b>	Ph—C≡C—C <sub>6</sub> H <sub>4</sub> —C≡C—Ph <b>3h</b>	75
10	<b>1f</b>	<b>2a</b>	TMS—C≡C—C <sub>6</sub> H <sub>4</sub> —C≡C—Ph <b>3i</b>	42

**Scheme 4**

(3,3-Dimethylbut-1-ynyl)trimethylsilane (**1g**) was treated with triphenylantimony diacetate (**2a**) in the presence of Pd<sub>2</sub>(dba)<sub>3</sub>·CHCl<sub>3</sub> (5 mol%) and CuI (10 mol%) under atmospheric pressure of carbon monoxide in CH<sub>3</sub>CN to give ynone **4e**<sup>13f</sup> in 80% yield (entry 6 in Table 2). Under the same conditions phenylethyne (**1a**) reacted with **2a** to afford benzoyl substituted alkyne **4a**<sup>13g</sup> in 62% yield (entry 1). For the substituted arylethyne silanes **1b** and **1c**, the reactions with triphenylantimony diacetate (**2a**) provided the carbonylated coupled products **4b**<sup>13h</sup> and **4c** in 65 and 68% yields, respectively (entries 2 and 3). The bis-silyl substituted **1e** and **1f** were readily coupled with **2a** to afford the mono carbonylated silanes **4a** and **4d** in 66 and 58% yields, respectively (entries 4 and 5). This method was also applied to *p*-tolyl substituted antimony diacetate **2b**. The alkynylsilanes **1g** and **1b** were subjected to carbonylative cross-coupling to afford ynones **4f** and **4g** in 77 and 81% yields, respectively (entries 7 and 8). To the best of our knowledge this is the first direct Pd(0)-catalyzed carbonylative cross-coupling of alkynylsilanes without deprotection of a silyl group.

In considering the plausible mechanism for carbonylative cross-coupling, the palladium complex **A** is generated by oxidative addition of the Ar–Sb bond of triarylantimony(v) diacetate onto Pd(0) followed by carbonylation. Alternatively, the alkynyl group is transferred from alkynylsilane to copper to form alkynylcopper **B**. The alkynyl group in **B** migrates from copper to palladium to furnish intermediate **C** which is subjected to reductive elimination to give the direct coupled product, regenerating Pd(0) and Cu(I) as a catalyst (Scheme 5).<sup>14</sup>

In conclusion triarylantimony(v) diacetates were prepared conveniently and cross-coupling and carbonylative cross-coupling of triarylantimony(v) derivatives with alkynylsilanes were achieved in the presence of Pd<sub>2</sub>(dba)<sub>3</sub>·CHCl<sub>3</sub> (5 mol%) and CuI (10 mol%), at 50 °C at atmospheric pressure of carbon monoxide for the carbonylative cross-coupling.

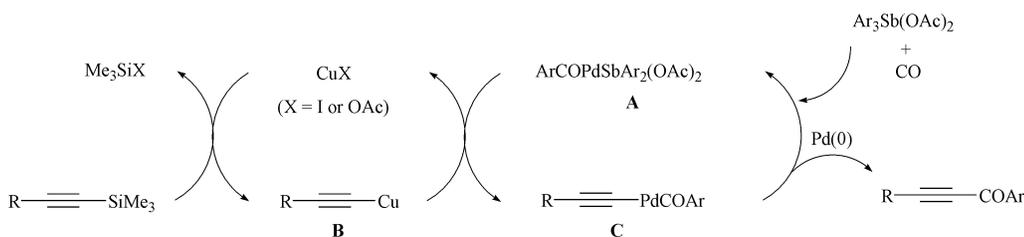
## Experimental

### Typical procedure for the preparation of triarylantimony(v) diacetate

A mixture of triarylantimony (2.0 mmol) and (diacetoxyiodo)-benzene diacetate (2.2 mmol) in dichloromethane (20 mL) was stirred at room temperature for 7 h. The solvent was concentrated under reduced pressure to a small volume. A mixture of diethyl ether–pentane was added and the solution was kept overnight at –15 °C. The solid was filtered and recrystallized from a mixture of dichloromethane and pentane.

**Table 2** Pd(0)–Cu(I)-catalyzed carbonylative cross-coupling of alkynylsilanes with triarylantimony diacetates

Entry	Silanes	Triarylantimony diacetates	Product	Yield (%)
1	Ph—C≡C—TMS <b>1a</b>	Ph <sub>3</sub> Sb(OAc) <sub>2</sub> <b>2a</b>	Ph—C≡C—C(=O)Ph <b>4a</b>	62
2	MeO—C <sub>6</sub> H <sub>4</sub> —C≡C—TMS <b>1b</b>	<b>2a</b>	MeO—C <sub>6</sub> H <sub>4</sub> —C≡C—C(=O)Ph <b>4b</b>	65
3	NC—C <sub>6</sub> H <sub>4</sub> —C≡C—TMS <b>1c</b>	<b>2a</b>	NC—C <sub>6</sub> H <sub>4</sub> —C≡C—C(=O)Ph <b>4c</b>	68
4	TMS—C≡C—TMS <b>1e</b>	<b>2a</b>	Ph—C≡C—C(=O)Ph <b>4a</b>	66
5	TMS—C≡C—C <sub>6</sub> H <sub>4</sub> —C≡C—TMS <b>1f</b>	<b>2a</b>	TMS—C≡C—C <sub>6</sub> H <sub>4</sub> —C≡C—C(=O)Ph <b>4d</b>	58
6	Bu <sup>t</sup> —C≡C—TMS <b>1g</b>	<b>2a</b>	Bu <sup>t</sup> —C≡C—C(=O)Ph <b>4e</b>	80
7	<b>1g</b>	( <i>p</i> -Tol) <sub>3</sub> Sb(OAc) <sub>2</sub> <b>2b</b>	Bu <sup>t</sup> —C≡C—C(=O)—C <sub>6</sub> H <sub>4</sub> —CH <sub>3</sub> <b>4f</b>	77
8	<b>1b</b>	<b>2b</b>	MeO—C <sub>6</sub> H <sub>4</sub> —C≡C—C(=O)—C <sub>6</sub> H <sub>4</sub> —CH <sub>3</sub> <b>4g</b>	80

**Scheme 5**

Triphenylantimony(v) diacetate (**2a**): mp 210–212 °C (lit.<sup>15</sup> 208–209 °C);  $\delta_{\text{H}}$  1.83 (s, 6H), 7.48 (m, 9H), 7.99 (m, 6H).

Tri(*p*-tolyl)antimony(v) diacetate (**2b**): mp 157–159 °C;  $\delta_{\text{H}}$  1.82 (s, 6H), 2.39 (s, 9H), 7.27 (d, 6H,  $J = 8$  Hz), 7.86 (d, 6H,  $J = 8$  Hz).

#### Typical procedure for the cross-coupling of alkynylsilanes with organoantimony(v) compounds

**Diphenylacetylene (3a).** To a mixture of triphenylantimony diacetate (**2a**) (412 mg, 1.00 mmol), Pd<sub>2</sub>(dba)<sub>3</sub>·CHCl<sub>3</sub> (112 mg, 5 mol%) and CuI (19 mg, 10 mol%) was added (phenylethynyl)trimethylsilane (**1a**) (169 mg, 1.00 mmol) under N<sub>2</sub> charged at 50 °C in CH<sub>3</sub>CN (20 mL). The reaction mixture was stirred at 50 °C for 5 h, extracted with ether (20 mL × 3), and washed with water (20 mL × 3). The organic layer was dried over anhydrous MgSO<sub>4</sub> and evaporated *in vacuo*. The crude product was separated by SiO<sub>2</sub> column chromatography (hexanes,  $R_{\text{f}} = 0.48$ ) to afford the coupled product diphenylacetylene (**3a**) (144 mg, 81%);  $\delta_{\text{H}}$  (500 MHz; CDCl<sub>3</sub>) 7.26 (m, 2H), 7.36 (m, 4H), 7.54 (m, 4H);  $\nu_{\text{max}}$  (KBr)/cm<sup>-1</sup> 3063, 1600, 1499, 1070, 755;  $m/z$  178 (100), 176, 152, 89, 88, 76.

Compounds **3b–3i** were prepared following the above procedures using the appropriate starting material.

**1-Phenylethynyl-4-methylbenzene (3b).** Hexanes,  $R_{\text{f}} = 0.40$ ;  $\delta_{\text{H}}$  (500 MHz; CDCl<sub>3</sub>) 2.37 (s, 3 H), 7.17 (d, 1 H,  $J = 7.8$  Hz), 7.34 (m, 3 H), 7.42 (m, 2 H), 7.53 (m, 2 H);  $\nu_{\text{max}}$  (KBr)/cm<sup>-1</sup> 3084, 2964, 1612, 1480, 1372, 695;  $m/z$  192 (100), 191, 189, 165, 115.

**1-Methoxy-4-(phenylethynyl)benzene (3c).** EtOAc–hexanes 1 : 10,  $R_{\text{f}} = 0.52$ ;  $\delta_{\text{H}}$  (500 MHz; CDCl<sub>3</sub>) 3.83 (s, 3 H), 6.89 (dd, 1 H,  $J = 4.2, 2.0$  Hz), 7.33 (m, 3 H), 7.51 (m, 4 H);  $\nu_{\text{max}}$  (KBr)/cm<sup>-1</sup> 3080, 2958, 1617, 1405;  $m/z$  208, 207 (100), 193 (49), 165 (52).

**4-Phenylethynylbenzonitrile (3d).** EtOAc–hexanes 1 : 30,  $R_{\text{f}} = 0.33$ ;  $\delta_{\text{H}}$  (500 MHz; CDCl<sub>3</sub>) 7.37 (m, 3 H), 7.50 (m, 2 H), 7.62 (m, 4 H);  $m/z$  204 (11), 103 (100), 156 (10), 88 (11).

**4-Phenylbut-3-yn-2-one (3e).** EtOAc–hexanes 1 : 10,  $R_{\text{f}} = 0.48$ ;  $\delta_{\text{H}}$  (500 MHz; CDCl<sub>3</sub>) 2.45 (s, 3 H), 7.38 (m, 2 H), 7.45 (m, 1 H), 7.56 (m, 2 H);  $\nu_{\text{max}}$  (KBr)/cm<sup>-1</sup> 3054, 2865, 1680, 1456.

**4-(4-Methylbenzene)but-3-yn-2-one (3f).** EtOAc–hexanes 1 : 10,  $R_{\text{f}} = 0.47$ ;  $\delta_{\text{H}}$  (500 MHz; CDCl<sub>3</sub>) 2.38 (s, 3 H), 2.45 (s, 3 H), 7.19 (d, 2 H,  $J = 9.9$  Hz), 7.47 (dd, 2 H,  $J = 9.9, 2.1$  Hz);

$\delta_{\text{C}}$  (125 MHz;  $\text{CDCl}_3$ ) 200.9, 157.9, 150.5, 137.6, 137.2, 130.0, 128.7, 30.1, 21.9;  $\nu_{\text{max}}$  (KBr)/ $\text{cm}^{-1}$  3054, 2865, 1680, 1456;  $m/z$  158 (27), 143 (100), 115 (13), 89 (9); HRMS for  $\text{C}_{11}\text{H}_{10}\text{O}$  calcd: 158.0732, found: 158.0728.

**Bis(4-methylphenyl)acetylene (3g).** Hexanes,  $R_f = 0.38$ ;  $\delta_{\text{H}}$  (500 MHz;  $\text{CDCl}_3$ ) 2.36 (s, 3 H), 7.24 (d, 2 H,  $J = 8.0$  Hz), 7.50 (d, 2 H,  $J = 8.0$  Hz);  $\nu_{\text{max}}$  (KBr)/ $\text{cm}^{-1}$  3080, 2964, 1609, 1480, 1377;  $m/z$  207 (10), 106 (100), 189 (15), 102 (10), 101 (11), 89 (14).

**1,4-Bis(phenylethynyl)benzene (3h).** Hexanes,  $R_f = 0.27$ ;  $\delta_{\text{H}}$  (500 MHz;  $\text{CDCl}_3$ ) 7.45 (m, 3 H), 7.58 (m, 4 H);  $m/z$  278 (100), 139 (28), 126 (9).

**Trimethyl[4-(phenylethynyl)phenylethynyl]silane (3i).** Hexanes,  $R_f = 0.29$ ;  $\delta_{\text{H}}$  (500 MHz;  $\text{CDCl}_3$ ) 0.12 (s, 9 H), 7.32 (m, 3 H), 7.49 (m, 4 H), 7.52 (m, 2 H);  $m/z$  274 (13), 273 (55), 260 (22), 259 (100), 130 (17).

#### Typical procedure for the carbonylative cross-coupling of alkynylsilanes with organoantimony(v) compounds

**3-(4-Methoxyphenyl)-1-phenylprop-2-yn-1-one (4b).** To a mixture of triphenylantimony diacetate (**2a**) (412 mg, 1.00 mmol),  $\text{Pd}_2(\text{dba})_3 \cdot \text{CHCl}_3$  (112 mg, 5 mol%) and CuI (19 mg, 10 mol%) was added (4-methoxyphenylethynyl)trimethylsilane (**1b**) (168 mg, 1.00 mmol) under atmospheric pressure of CO at 50 °C in  $\text{CH}_3\text{CN}$  (20 mL). The reaction mixture was stirred at 50 °C for 5 h, extracted with ether (20 mL  $\times$  3), and washed with water (20 mL  $\times$  3). The organic layer was dried over anhydrous  $\text{MgSO}_4$  and evaporated *in vacuo*. The crude product was separated by  $\text{SiO}_2$  column chromatography (EtOAc–hexanes 1 : 10,  $R_f = 0.19$ ) to afford the coupled product 3-(4-methoxyphenyl)-1-phenylprop-2-yn-1-one (**4b**) (118 mg, 65%);  $\delta_{\text{H}}$  (500 MHz;  $\text{CDCl}_3$ ) 3.89 (s, 3 H), 6.98 (m, 2 H), 7.41 (m, 2 H), 7.45 (m, 1 H), 7.66 (m, 2 H), 8.19 (m, 2 H);  $\nu_{\text{max}}$  (KBr)/ $\text{cm}^{-1}$  3055, 2200, 1632, 1264;  $m/z$  237, 236, 208, 193, 165, 159 (100), 144.

Compounds **4a**, **c**–**g** were prepared following the above procedures using the appropriate starting material.

**1,3-Diphenylprop-2-yn-1-one (4a).** EtOAc–hexanes 1 : 10,  $R_f = 0.34$ ;  $\delta_{\text{H}}$  (500 MHz;  $\text{CDCl}_3$ ) 7.41 (m, 2 H), 7.48 (m, 1 H), 7.52 (m, 1 H), 7.62 (m, 1 H), 7.68 (m, 1 H), 8.22 (m, 2 H);  $\nu_{\text{max}}$  (KBr)/ $\text{cm}^{-1}$  3055, 2200, 1641;  $m/z$  206 (95), 178 (100), 129 (94).

**4-(3-Oxo-3-phenylprop-1-ynyl)benzointrile (4c).** EtOAc–hexanes 1 : 7,  $R_f = 0.34$ ;  $\delta_{\text{H}}$  (500 MHz;  $\text{CDCl}_3$ ) 7.36–7.82 (m, 7 H), 8.19 (m, 2 H);  $\delta_{\text{C}}$  (125 MHz;  $\text{CDCl}_3$ ) 174.9, 134.0, 132.1, 130.8, 130.6, 129.9, 127.2, 126.4, 122.5, 115.4, 111.6, 87.1;  $m/z$  235 (15), 234 (99), 233 (100), 105 (14), 191 (12), 128 (29), 103 (82), 91 (23), 77 (33);  $\nu_{\text{max}}$  (KBr)/ $\text{cm}^{-1}$  3056, 2987, 2305, 2205, 1644; HRMS for  $\text{C}_{16}\text{H}_9\text{NO}$  calcd: 231.0684, found: 231.0681.

**1-Phenyl-3-(4-trimethylsilanylethynylphenyl)prop-2-yn-1-one (4d).** EtOAc–hexanes 1 : 10,  $R_f = 0.43$ ;  $\delta_{\text{H}}$  (500 MHz;  $\text{CDCl}_3$ ) 0.22 (s, 9 H), 7.58 (m, 2 H), 7.62 (m, 2 H), 7.74 (m, 1 H), 7.80 (m, 2 H), 8.18 (m, 2 H);  $\delta_{\text{C}}$  (125 MHz;  $\text{CDCl}_3$ ) 178.5, 137.5, 134.9, 133.5, 132.8, 130.3, 129.5, 126.4, 120.6, 104.7, 98.9, 92.9, 88.9, 0.5;  $\nu_{\text{max}}$  (KBr)/ $\text{cm}^{-1}$  3055, 2987, 2199, 1641, 1423, 1265, 744; HRMS for  $\text{C}_{20}\text{H}_{18}\text{OSi}$  calcd: 302.1127, found: 302.1123.

**4,4-Dimethyl-1-phenylpent-2-yn-1-one (4e).** EtOAc–hexanes 1 : 10,  $R_f = 0.41$ ;  $\delta_{\text{H}}$  (500 MHz;  $\text{CDCl}_3$ ) 1.34 (s, 9 H), 7.61 (m, 2 H), 7.73 (m, 1 H), 8.05 (m, 2 H);  $m/z$  186 (10), 143 (34), 128 (23), 105 (100), 77 (19).

**4,4-Dimethyl-1-(p-tolyl)pent-2-yn-1-one (4f).** EtOAc–hexanes 1 : 10,  $R_f = 0.47$ ;  $\delta_{\text{H}}$  (500 MHz;  $\text{CDCl}_3$ ) 1.56 (s, 9 H), 2.42 (s, 3

H), 7.27 (m, 2 H), 7.69 (m, 2 H), 8.00 (m, 2 H);  $\delta_{\text{C}}$  (125 MHz;  $\text{CDCl}_3$ ) 178.8, 145.5, 135.5, 130.3, 129.9, 104.1, 78.9, 30.9, 28.7, 22.5;  $\nu_{\text{max}}$  (KBr)/ $\text{cm}^{-1}$  3055, 2888, 1706, 1450; HRMS for  $\text{C}_{14}\text{H}_{16}\text{O}$  calcd: 200.1201, found: 200.1203.

**3-(4-Methoxyphenyl)-1-(p-tolyl)prop-2-yn-1-one (4g).** EtOAc–hexanes 1 : 10,  $R_f = 0.16$ ;  $\delta_{\text{H}}$  (500 MHz;  $\text{CDCl}_3$ ) 2.45 (s, 3 H), 3.89 (s, 3 H), 6.98 (m, 2 H), 7.41 (m, 2 H), 7.45 (m, 1 H), 7.66 (m, 2 H), 8.19 (m, 2 H);  $\delta_{\text{C}}$  (125 MHz;  $\text{CDCl}_3$ ) 178.5, 162.3, 145.7, 135.5, 130.3, 130.0, 129.7, 115.1, 112.7, 94.5, 87.6, 56.1, 22.5;  $\nu_{\text{max}}$  (KBr)/ $\text{cm}^{-1}$  3055, 2963, 2200, 1632, 1264; HRMS for  $\text{C}_{17}\text{H}_{14}\text{O}_2$  calcd: 250.0994, found: 250.0988.

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