

One-Pot Synthesis of B-Ring ortho-Hydroxylated Sappanin-type Homoisoflavanoids

Galyna P Mrug, Natalia V Myshko, Svitlana P. Bondarenko, Vitaliy M Sviripa, and Mykhaylo S. Frasinyuk

J. Org. Chem., Just Accepted Manuscript • DOI: 10.1021/acs.joc.9b00814 • Publication Date (Web): 15 May 2019

Downloaded from <http://pubs.acs.org> on May 15, 2019

Just Accepted

"Just Accepted" manuscripts have been peer-reviewed and accepted for publication. They are posted online prior to technical editing, formatting for publication and author proofing. The American Chemical Society provides "Just Accepted" as a service to the research community to expedite the dissemination of scientific material as soon as possible after acceptance. "Just Accepted" manuscripts appear in full in PDF format accompanied by an HTML abstract. "Just Accepted" manuscripts have been fully peer reviewed, but should not be considered the official version of record. They are citable by the Digital Object Identifier (DOI®). "Just Accepted" is an optional service offered to authors. Therefore, the "Just Accepted" Web site may not include all articles that will be published in the journal. After a manuscript is technically edited and formatted, it will be removed from the "Just Accepted" Web site and published as an ASAP article. Note that technical editing may introduce minor changes to the manuscript text and/or graphics which could affect content, and all legal disclaimers and ethical guidelines that apply to the journal pertain. ACS cannot be held responsible for errors or consequences arising from the use of information contained in these "Just Accepted" manuscripts.



ACS Publications

is published by the American Chemical Society, 1155 Sixteenth Street N.W., Washington, DC 20036

Published by American Chemical Society. Copyright © American Chemical Society. However, no copyright claim is made to original U.S. Government works, or works produced by employees of any Commonwealth realm Crown government in the course of their duties.

1
2
3
4
5
6
7
8
9

One-Pot Synthesis of B-Ring *ortho*-Hydroxylated Sappanin-type Homoisoflavanoids

Galyna P. Mrug,[†] Nataliia V. Myshko,[‡] Svitlana P. Bondarenko,[§] Vitaliy M. Sviripa,^{*†,⊥} and Mykhaylo S. Frasinyuk ^{*†}.

[†] Department of Chemistry of Bioactive Nitrogen-Containing Heterocyclic Bases, V. P. Kukhar Institute of Bioorganic Chemistry and Petrochemistry, National Academy of Science of Ukraine, Kyiv 02094, Ukraine

[‡] Institute of High Technologies, Taras Shevchenko Kyiv National University, Kyiv 03022, Ukraine

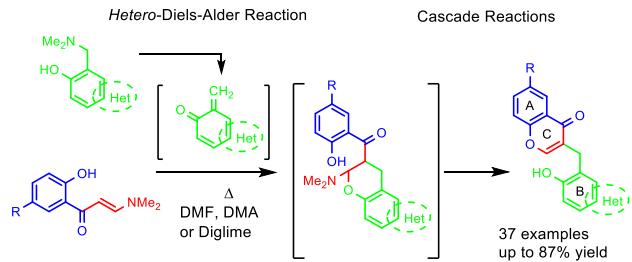
[§] Department of Food Chemistry, National University of Food Technologies, Kyiv 01601, Ukraine

^{||}Center for Pharmaceutical Research and Innovation, College of Pharmacy, University of Kentucky, Lexington, KY 40536-0509, USA

[⊥] Department of Pharmaceutical Sciences, College of Pharmacy, University of Kentucky, Lexington, KY 40536-0509, USA

Supporting Information Placeholder

ABSTRACT: A reliable method for the synthesis of B-ring hydroxylated homoisoflavanoids and 3-hetarylmethyl chromones has been developed. The method involves an initial *oxa*-Diels–Alder reaction of *ortho*-(*N,N*-dimethylaminomethyl)phenols with (2*E*)-3-(*N,N*-dimethylamino)-1-(2-hydroxyphenyl)prop-2-en-1-ones and the subsequent cascade of reactions. This synthetic strategy avoids conventional multistep protocols and does not require the protection of hydroxyl groups, thus allowing the facile synthesis of a library of various aromatic and heterocyclic analogs of naturally occurring homoisoflavanoids.



INTRODUCTION

The homoisoflavanoids are a small but diverse family of natural compounds that derive biosynthetically from 2'-methoxychalcones and usually classified into five main structural subtypes including sappanin-, scillasillin-, brazilin-, caesalpin-, and protosappanin-types.¹ Among them, the sappanin-type homoisoflavanoids (Fig. 1), that contain chromanone or chromone ring with 3-benzyl substituents, are the most diverse and well-studied ones. They are often used in biosynthetic pathways as precursors for the synthesis of other types.² Like other flavonoid counterparts, naturally occurring homoisoflavanoids typically bear hydroxyl, methoxy, and methylenedioxy groups in rings A and B³ and possess a broad spectrum of biological activities, ranging from anti-oxidant, anti-microbial, anti-diabetic and immunomodulatory effects, to specific effects as protein kinase inhibitors. Therefore, the development of innovative and efficient approaches to the natural/semisynthetic homoisoflavanoids continues to be an active synthetic direction for the discovery of novel analogs with unique bioactivities.

Methodology in the literature for the synthesis of homoisoflavanoids follows patterns seen in the synthesis of the corresponding isoflavones from 2-hydroxydeoxybenzoines by the ring-closure reactions and typically involves an applica-

tion of the corresponding 1-(2-hydroxyphenyl)-3-phenylpropan-1-ones in Vilsmayer–Haak reaction,⁴ Claisen condensation,⁵ *ortho*-ether condensation⁶ reactions, and in cyclization with anhydrides⁷. Alternatively, 9-hydroxyderivatives have been synthesized by Baylis–Hillman reaction of chromones with aromatic aldehydes.⁸ These approaches include multiple steps and require the introduction of B-ring fragment at the initial stage of the synthesis 1-(2-hydroxyphenyl)-3-(aryl/hetaryl)propan-1-ones, and therefore, they are often inconvenient for the synthesis of B-ring hydroxylated homoisoflavanoids and their hydroxylated heteroaryl analogs.

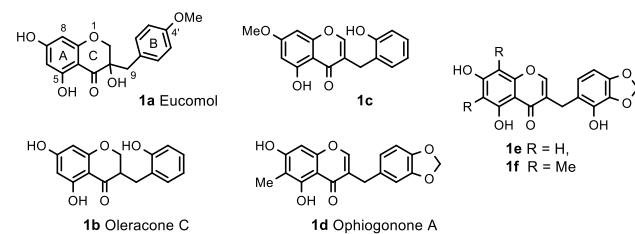


Figure 1. Chemical structures of some sappanin-type natural homoisoflavanoids

On the other hand, 2'-hydroxy derivatives of (*E*)-1-phenyl-3-(*N,N*-dimethylamino)prop-2-en-1-ones **2** (commonly

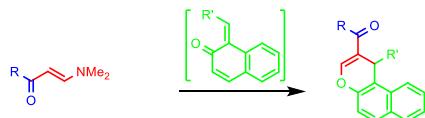
named as enaminones) have proven to be efficient substrates for constructing chromone frameworks. Thus, treatment of enaminones with acids is a common pathway to the formation of 2,3-unsubstituted naphthopyrones and chromones.⁹ Oxidative reactions of enaminones with halogens or *tert*-butyl hypochlorite¹⁰ are simple methods for the synthesis of 3-halogenochromones¹¹ which in turn are useful compounds for the synthesis of isoflavones by Suzuki coupling reactions. A related oxidative reaction for the synthesis of 3-thiocyanatochromones from enaminones with KSCN in the presence of K₂S₂O₈ was also reported.¹² At last, alkylation/acylation of the enamine fragment and subsequent cyclization reactions are effective methods for the preparation of alkylchromones¹³ and 3-acylchromones.¹⁴ As example, the alkylation of 2-hydroxylated enaminones with benzyl bromides led to the formation of homoisoflavones^{13,15} (Scheme 1a) and recently reported interaction of β -naphthol Mannich bases with enaminoketones provided access to the synthesis of 3-acyl-4*H*-chromenes (Scheme 1b).¹⁶ The latter transformation was achieved by inverse-electron-demand hetero-Diels–Alder reaction of enaminones with *ortho*-quinone methides (*o*-QMs), which have found to be useful in the syntheses of oxygen-containing heterocycles.^{17,18}

Scheme 1. Application of enaminones in the synthesis of chromones

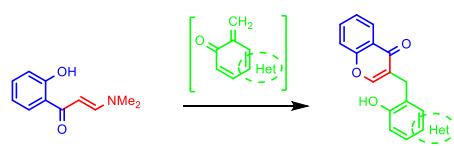
a) known: CH-alkylation and cascade reactions



b) known: Diels-Alder addition and elimination



c) in this work: Diels-Alder addition and cascade reactions

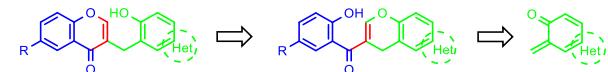


Taking into account this information and our previous report,¹⁹ we anticipated that utilizing of *ortho*-hydroxyphenyl enaminones in Diels–Alder addition with thermally generated *o*-QMs from Mannich bases would provide adducts similar to the ones shown in Scheme 1b. However, these adducts due to the presence of *ortho*-hydroxyl could undergo the subsequent intramolecular rearrangement involving nucleophilic attack of the hydroxyl group at position 2 of 4*H*-chromene ring that resulted in the formation of the 3-(2-hydroxybenzyl)-4*H*-chromen-4-one skeleton (Scheme 1c). The retrosynthetic pathway of such homoisoflavanoids synthesis is presented in Scheme 2.

Herein, in continuation of our research program aimed at the development of novel flavonoids for the search of pharmaceutical agents, we report a one-pot approach for the synthesis of hydroxylated homoisoflavanoids. This approach involves initial inverse-electron-demand Diels–Alder reaction of thermally generated from phenolic Mannich bases *ortho*-quinone methides with (2*E*)-3-(*N,N*-dimethylamino)-

1-(2-hydroxyphenyl)prop-2-en-1-ones and the subsequent cascading reactions.

Scheme 2. Retrosynthetic Pathway



RESULTS AND DISCUSSION

To optimize the reaction conditions, we investigated the reaction between (*E*)-3-(dimethylamino)-1-(2-hydroxy-5-methylphenyl)prop-2-en-1-one (**2a**) and 1-((dimethylamino)methyl)naphthalen-2-ol (**3d**). A screen of various solvents revealed that DMF was the best choice for this transformation to afford the desired product **4da** in 44% yield. The reaction could also proceed in other solvents, such as diglyme and dimethylacetamide (DMA), but the yields of **4da** were lower (data not shown). All attempts to improve the yield by extending the reaction time were unsuccessful.

Table 1. Substrate Scope of Enaminoketones 2 and Mannich Bases 3^a

2a-2d	3a-3j	4
		Solvent temp, time
4aa, R = Me, 29%	4ba, R = Me, 59%	4ca, R = Me, 24%
4ab, R = F, 21%	4bb, R = F, 45%	4cb, R = F, 24%
4ac, R = Cl, 35%	4bc, R = Cl, 51%	4cc, R = Cl, 42%
4ad, R = Br, 19%	4bd, R = Br, 36%	4cd, R = Br, 53%
4da, R = Me, 44%	4ea, R = Me, 12%	4fa, R = Me, 39%
4db, R = F, 34%	4eb, R = F, 35%	4fb, R = F, 35%
4dc, R = Cl, 43%	4ec, R = Cl, 21%	4fc, R = Cl, 44%
4dd, R = Br, 63%	4ed, R = Br, 52%	4fd, R = Br, 20%
4ga, R = Me, 71%	4hb, R = F, 51%	4ja, R = Me, 47%
4gb, R = F, 67%	4hc, R = Cl, 87%	4jb, R = F, 43%
4gc, R = Cl, 66%		4jc, R = Cl, 54%
4gd, R = Br, 67%		4jd, R = Br, 39%
Enami-	Mannich	Solvent
none	base	Temp, °C
2a-2d	3a	Diglyme
2a-2d	3b	160
2a-2d	3c	160
2a-2d	3d	154
2a-2d	3e	154
2a-2d	3f	154
2a-2d	3g	154
2a-2d	3h	154
2a-2d	3i	154
2a-2d	3i	180
		Time, h
		20
		20
		20
		16
		16
		16
		16
		16
		16
		16
		20
4ia, R = Me, 37%		4ja, R = Me, 47%
4ib, R = F, 43%		4jb, R = F, 43%
4ic, R = Cl, 23%		4jc, R = Cl, 54%
4ja, R = Me, 47%		4jd, R = Br, 39%

^a Isolated yield

To define the scope of this approach to homoisoflavanoids **4**, we studied the reactions of methyl- and halogen-substituted (*E*)-1-(2-hydroxyphenyl)-3-(*N,N*-dimethylamino)prop-2-en-1-ones **2a-2d** with *ortho*-*N,N*-dimethylaminomethyl derivatives of various carbocyclic/heterocyclic phenols **3a-3j** (Table 1).

In all cases, the *oxa*-Diels-Alder addition of the ortho-quinone methides, generated from **3a-j**, with the enaminones **2a-d** in diglyme, DMF or DMA at temperatures more than 150°C led to the formation of the homoisoflavanoids **4** in yields ranging from moderate to high (Table 1). Thus, the reaction of enaminones **2a-2d** with phenolic Mannich bases **3a-3c** provided better yields of the corresponding products **4** by refluxing in diglyme, whereas the reaction of the same enaminones **2a-2d** with bases **3d-3i** gave higher yields of **4** by refluxing in DMF or DMA (in case of the reaction with compound **3j**).

The substituents on the aryl ring of the enaminoketones **2** had little influence on the reaction outcome. The reaction mainly depended on nature of Mannich bases. For example, the reaction of **2a-2d** with the Mannich bases that contained condensed aromatic rings such as **3d**, **3g**, **3h**, and **3j** furnished higher yields of corresponding **4** compared to their reaction with Mannich base **3a** (Table 1). Such results could be rationalized by their propensity to the charges delocalization effects that contributed to the stabilization of *o*-QM intermediates.

The molecular structure of products **4** was unambiguously confirmed by 2D NMR techniques. Thus, the presence of HMBC correlations between H-2, H-5, H-7 and C-8a in compound **4ga** confirmed the formation of chromone ring (Fig. 2, see Supporting Information for spectra and detailed information).

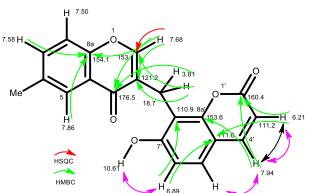


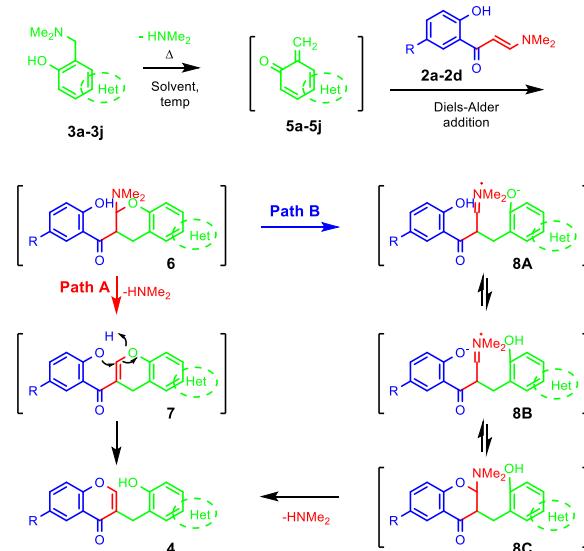
Figure 2. Principal 2D NMR correlations for 4ga

In our opinion, the initial steps of such transformation proceeded in the same manner as it was described for the aforementioned reaction of non-hydroxylated enaminoketones with phenol or naphtol Mannich bases.¹⁶ Obviously, that initial thermal generation of *o*-QMs **5** from carbocyclic or heterocyclic phenolic Mannich bases **3** and their following inverse electron-demand *oxa*-Diels-Alder addition to the hydroxylated enaminoketones **2** furnished hemi-aminal adducts **6** (Scheme 3).

Such unstable Diels–Alder adducts **6** underwent further deamination and afforded 4*H*-benzopyrane derivatives **7**, which underwent a base-catalyzed intramolecular nucleophilic attack of the hydroxylic group at position 2 of 4*H*-chromene ring and provided the desired products **4**. (Scheme 3, Path A) We also can consider an alternate rearrangement of adducts **6** to intermediates **8A-8C** and their further deamination to derivatives **4** (Scheme 3, Path B). The preferable formation of compounds **4** from intermediates **7** can be explained by the thermodynamic control. The formation of 10 π electron aromatic system for **4** is more energetically favorable than the 6 π +2 π electron non-conjugated system for compounds **7**. It should be also noted, that in all cases we only isolated compounds **4** as the final products and did not detect the formation of compounds **6-8**.

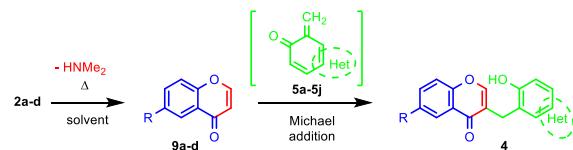
Taking into account Michael addition of *o*-QMs to 2-aminochromones reported earlier,²⁰ we could not ignore the possible Michael addition reaction of *o*-QMs **5** with chromones **9** as a plausible pathway in the formation of homoisoflavanoids **4** (Scheme 4).

Scheme 3. Proposed Reaction Mechanism



It is important to notice that during the conditions optimizing studies of the reaction of *ortho*-hydroxyphenyl enaminones **2** with Mannich bases we indeed isolated traces of chromones **9a-d** in those cases when the reaction was not completed at low temperatures. In order to determine a contribution of such mechanism to the synthesis of **4**, we performed additional studies and carried out the reactions of enaminone **2a** with **3a** or **3d** (Table 2) and Michael addition reactions of chromone **9a** with **3a** or **3d** (Table 3) at different conditions during 4 h. According to our observations, this time was usually insufficient for full conversion of enaminones **2** or chromones **9** to homoisoflavanoids **4**, therefore, we expected to isolate intermediates of the reaction of enaminone **2a** with Mannich bases **3a** or **3d**.

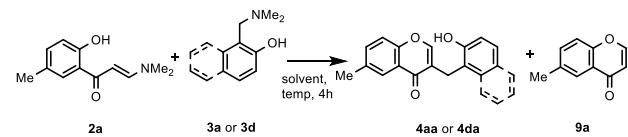
Scheme 4. Alternate Reaction Mechanism via Michael Addition of *o*-QMs **5** to Chromones **9**



Thus, the reaction of enaminone **2a** with 2-(*N,N*-dimethylaminomethyl)phenol (**3a**) at temperatures less than 120°C, led to the increased amounts of **9a** relative to the target homoisoflavanoids **4aa** (Table 2, entries 1–3). However, the direct addition of **5a** generated from **3a** to chromone **9a** at the same conditions did not occur (Table 3, entries 1–3). Moreover, the reaction of enaminone **2a** with **3a** at temperatures 154–160 °C gave much higher yields of **4aa** (Table 2, entries 4,5) compared to the reaction of chromone **9a** with **3a** that resulted in very low yield of **4aa** (Table 3, entries 4,5). The similar results were observed for the reactions of enaminone **2a** or chromone **9a** with 1-(*N,N*-dimethylaminomethyl)-2-naphthol (**3d**) that forms more

stable *o*-QM intermediate **5d**. In all cases, the direct addition of *o*-QM **5d** generated from **3d** to chromone **9a** provided much lower yields of the desired homoisoflavanoid **4da** (Table 3, entries 6–10) than the corresponding reactions of enaminone **2a** with **3d** (Table 2, entries 6–10).

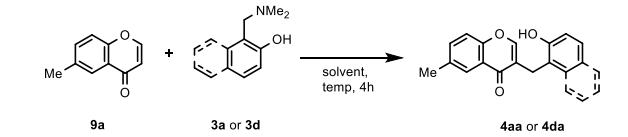
Table 2. Screening Conditions for the Reaction of Enaminone **2a with Mannich Bases **3a**, **3d**^a**



entry	solvent	temp (°C)	Mannich base	4aa or 4da yield (%) ^b	9a yield (%) ^b
1	MeOCH ₂ CH ₂ OMe	80	3a	-	32
2	1,4-Dioxane	100	3a	5.6	78.6
3	EtOCH ₂ CH ₂ OEt	121	3a	10.1	86.1
4	DMF	154	3a	75.9	24.1
5	Diglyme	160	3a	43.6	56.4
6	MeOCH ₂ CH ₂ OMe	80	3d	81.5	-
7	1,4-Dioxane	100	3d	87.5	1.9
8	EtOCH ₂ CH ₂ OEt	121	3d	72.5	-
9	DMF	154	3d	90.8	-
10	Diglyme	160	3d	95	-

^a All reactions were carried out using **2a** (1 mmol), **3a** or **3d** (1 mmol), and corresponding solvent (5mL) for 4 h at the indicated temperatures. ^b Yields were determined by the LC-MS.

Table 3. Screening Conditions for Michael Addition of **9a with Mannich Bases **3a**, **3d**^a**



entry	solvent	temp (°C)	Mannich base	product	yield (%) ^b
1	MeOCH ₂ CH ₂ OMe	80	3a	4aa	-
2	1,4-Dioxane	100	3a	4aa	-
3	EtOCH ₂ CH ₂ OEt	121	3a	4aa	-
4	DMF	154	3a	4aa	7.6
5	Diglyme	160	3a	4aa	2.8
6	MeOCH ₂ CH ₂ OMe	80	3d	4da	6.4
7	1,4-Dioxane	100	3d	4da	14.2
8	EtOCH ₂ CH ₂ OEt	121	3d	4da	39.7
9	DMF	154	3d	4da	70.0
10	Diglyme	160	3d	4da	30.9

^a All reactions were carried out using **9a** (1 mmol), either **3a** or **3d** (1 mmol), and corresponding solvent (5 mL) for 4 h at the indicated temperatures. ^b Yields were determined by the LC-MS.

Such observations suggested that 6-methyl-4H-chromen-4-one (**9a**) was not a dominant player in the reactions of enamones **2** with *o*-QMs **5a** and **5d** generated from the corre-

sponding Mannich bases **3a** and **3d**. Otherwise, the yields of **4aa** or **4ad** obtained in the reactions of enaminone **2a** or chromone **9a** with **3a** and **3d** under the same reaction conditions would be comparable. These results indicate that, although possible, the alternative conversion of enaminone **2a** to chromone **9a** and its further Michael addition to *o*-QMs (path **2a** → **9a** → **4aa**, Scheme 4) cannot be the primary synthetic route to homoisoflavanoid **4aa**. With respect to the aforementioned studies, we conclude that the most appropriate pathway is based on *oxa*-Diels-Alder addition of thermally generated *o*-QMs **5** to enaminones **2** with formation of hemiaminals **6**, followed by their deamination to 4H-chromene derivatives **7**, and their subsequent intramolecular base-catalyzed rearrangement to the target compounds **4**. Despite the moderate yields of the synthesized homoisoflavanoids, the simplicity, accessibility, and available variety of starting reagents make this method an attractive choice for the synthesis of various aromatic and heterocyclic analogs of naturally occurring homoisoflavanoids that would be difficult or impossible to synthesize by other methods.

In conclusion, we have developed an efficient one-pot method for the synthesis of B-ring hydroxylated homoisoflavanoids that does not require the protection of the phenolic groups. The reaction provides a novel preparative route for the synthesis of sappanin-type flavonoids and their heteroaryl analogs in moderate yields. Further practical scope and potential application of these promising homoisoflavanoids are currently under investigation in our laboratory.

EXPERIMENTAL SECTION

Chemicals were purchased from Sigma Aldrich or Fisher Scientific or were synthesized according to literature procedures. Solvents were used from commercial vendors without further purification unless otherwise noted. ¹H and ¹³C{¹H} NMR spectra were recorded on a Bruker 500 (500 MHz / 125 MHz) or Bruker 400 (400 MHz / 100 MHz) spectrometers in CDCl₃ or DMSO-d₆. ¹⁹F NMR spectra were recorded on a Bruker 400 (376 MHz) relative to CFCl₃. Melting points were determined in open capillarity tubes using a Buchi B-535 apparatus and were uncorrected. Mass spectra were obtained with an Agilent 1100 spectrometer under chemical ionization conditions. Elemental analysis was performed on a varioMICROcube automated CHNS-analyzer. Column chromatography was performed using Macherey-Nagel Silica 60, 0.04–0.063 mm silica gel.

The enaminones **2a**–**2d**²¹ and Mannich adducts **3a**,²² **3d**,²³ **3f**,²⁴ **3h**,²⁵ **3i**,²⁶ and **3j**²⁷ were prepared as previously described.

6-[Dimethylamino]methyl-1,3-benzodioxol-5-ol (3b). A solution of sesamol (966 mg, 7 mmol), paraformaldehyde (210 mg, 7 mmol), and dimethylamine (0.840 mL, 40% aqueous solution, 7.4 mmol) in ethanol (10 mL) was stirred at room temperature for 24 h. The solvent was evaporated *in vacuo* and residue was purified by recrystallization from isopropanol-hexanes mixture (1:5). Yield 1.110 g (82%). Beige solid, mp 87–89 °C. ¹H NMR (400 MHz, CDCl₃) δ: 2.26 (s, 6H), 3.47 (s, 2H), 5.81 (s, 2H), 6.37 (s, 1H), 6.40 (s, 1H), 10.81 ppm (s, 1H). ¹³C{¹H} NMR (125 MHz, CDCl₃) δ: 44.2, 62.6, 98.4, 100.7, 107.8, 113.0, 139.9, 147.5, 153.1 ppm. MS (APCI) m/z: 196.0 [M+H]⁺. Anal. Calcd (%) for C₁₀H₁₃NO₃: C, 61.53; H, 6.71; N, 7.18. Found (%): C, 61.38; H, 6.89; N, 6.92.

5-[(Dimethylamino)methyl]-1,3-benzodioxol-4-ol (3c) was synthesized by the similar procedure described for **3b** from 4-hydroxy-1,3-benzodioxol. Yield 675 mg (49%). Beige solid, ^1H NMR (400 MHz, CDCl_3) δ : 2.31 (s, 6H), 3.61 (s, 2H), 5.93 (s, 2H), 6.30 (d, $J = 7.8$ Hz, 1H), 6.43 (d, $J = 7.8$ Hz, 1H), 10.55 ppm (s, 1H). $^{13}\text{C}\{\text{H}\}$ NMR (100 MHz, CDCl_3) δ : 44.4, 62.8, 99.5, 101.3, 117.6, 120.7, 134.2, 142.5, 148.5 ppm. MS (APCI) m/z : 196.0 [M+H] $^+$. Anal. Calcd (%) for $\text{C}_{10}\text{H}_{13}\text{NO}_3$: C, 61.53; H, 6.71; N, 7.18. Found (%): C, 61.77; H, 6.99; N, 7.24.

7-[(Dimethylamino)methyl]-6-hydroxy-2-(1-methylethylidene)-1-benzofuran-3(2H)-one (3e). To a stirred suspension of 6-hydroxy-2-(1-methylethylidene)-1-benzofuran-3(2H)-one (950 mg, 5 mmol) in 10 mL of isopropanol were added 0.816 mL (6 mmol, 1.2 eq) of bis(N,N-dimethylamino)methane at 70°C. After stirring at 80°C for 2 h the mixture was cooled, triturated with hexane and formed precipitate collected by filtration to afford **3e** that was recrystallized from isopropanol-hexane. Yield 888 mg (72%). Yellow solid, mp 153 - 155 °C. ^1H NMR (400 MHz, CDCl_3) δ : 2.05 (s, 3H), 2.32 (s, 3H), 2.41 (s, 6H), 3.81 (s, 2H), 6.56 (d, $J = 8.5$ Hz, 1H), 7.52 (d, $J = 8.5$ Hz, 1H), 10.65 ppm (s, 1H). $^{13}\text{C}\{\text{H}\}$ NMR (100 MHz, CDCl_3) δ : 17.2, 20.0, 44.6, 54.8, 104.0, 112.7, 115.5, 124.6, 129.2, 145.7, 163.9, 166.9, 182.5 ppm. MS (APCI) m/z : 248.0 [M+H] $^+$. Anal. Calcd (%) for $\text{C}_{14}\text{H}_{17}\text{NO}_3$: C, 68.00; H, 6.93; N, 5.66. Found (%): C, 68.21; H, 6.78; N, 5.49.

8-Dimethylaminomethyl-7-hydroxy-chromen-2-one (3g) was synthesized similar to compound **3e**. Yield 613 mg (56%). Beige solid, mp 117 - 119 °C. ^1H NMR (400 MHz, CDCl_3) δ : 2.40 (s, 6H), 4.01 (s, 2H), 6.17 (d, $J=9.4$ Hz, 1H), 6.75 (d, $J=8.5$ Hz, 1H), 7.27 (d, $J=8.5$ Hz, 1H), 7.61 (d, $J=9.4$ Hz, 1H), 12.37 ppm (s, 1H). $^{13}\text{C}\{\text{H}\}$ NMR (100 MHz, CDCl_3) δ : 44.5, 55.0, 108.2, 111.1, 111.3, 113.8, 128.0, 144.3, 153.0, 161.3, 163.3 ppm. MS (APCI) m/z : 220.0 [M+H] $^+$. Anal. Calcd (%) for $\text{C}_{14}\text{H}_{17}\text{NO}_3$: C, 65.74; H, 5.98; N, 6.39. Found (%): C, 65.93; H, 7.06; N, 5.24.

General Procedure for the synthesis of Homoisoflavonoids 4. A mixture of enaminone **2a-d** (2 mmol) and corresponding Mannich base **3a-j** (2.4 mmol) in appropriate solvent (10 mL) was stirred at reflux for the indicated period of time (Tab.1). The reaction mixture was cooled, solvent was evaporated, and formed residue was washed with water and purified by recrystallization from EtOH or column chromatography.

3-(2-Hydroxybenzyl)-6-methyl-4H-chromen-4-one

(4aa). Purified by column chromatography using 1:100 methanol-dichloromethane. Yield 154 mg (29%). Off-white solid, mp 175 - 177 °C. ^1H NMR (400 MHz, $\text{DMSO}-d_6$) δ : 2.38 (s, 3H), 3.64 (s, 2H), 6.70 (t, $J = 7.4$ Hz, 1H), 6.82 (d, $J = 8.0$ Hz, 1H), 7.02 (t, $J = 7.7$ Hz, 1H), 7.05 - 7.14 (m, 1H), 7.44 (d, $J = 8.6$ Hz, 1H), 7.53 (dd, $J = 8.6, 2.2$ Hz, 1H), 7.82 (d, $J = 2.2$ Hz, 1H), 8.04 (s, 1H), 9.48 ppm (s, 1H). $^{13}\text{C}\{\text{H}\}$ NMR (125 MHz, $\text{DMSO}-d_6$) δ : 20.4, 25.4, 115.1, 118.1, 118.9, 122.4, 122.8, 124.2, 125.1, 127.4, 130.0, 134.7, 135.0, 153.9, 154.2, 155.1, 176.5 ppm. MS (APCI) m/z : 267.2 [M+H] $^+$. Anal. Calcd (%) for $\text{C}_{17}\text{H}_{14}\text{O}_3$: C, 76.68; H, 5.30. Found (%): C, 76.84; H, 5.11.

6-Fluoro-3-(2-hydroxybenzyl)-4H-chromen-4-one

(4ab). Purified by column chromatography using 1:100 methanol-dichloromethane. Yield 114 mg (21%). Off-white solid, mp 194 - 196 °C. ^1H NMR (400 MHz, $\text{DMSO}-d_6$) δ :

3.63 (s, 2H), 6.69 (t, $J = 7.4$ Hz, 1H), 6.82 (d, $J = 7.9$ Hz, 1H), 7.01 (t, $J = 7.8$ Hz, 1H), 7.08 (d, $J = 7.5$ Hz, 1H), 7.51 - 7.73 (m, 3H), 8.09 (s, 1H), 9.51 ppm (s, 1H). $^{13}\text{C}\{\text{H}\}$ NMR (125 MHz, $\text{DMSO}-d_6$) δ : 25.3, 109.4 (d, $J_{C-F} = 23.5$ Hz), 115.1, 118.9, 121.1 (d, $J_{C-F} = 8.3$ Hz), 122.0, 122.0 (d, $J_{C-F} = 25.5$ Hz), 124.2 (d, $J_{C-F} = 7.1$ Hz), 124.8, 127.4, 130.1, 152.3, 154.4, 155.2, 158.8 (d, $J_{C-F} = 244.1$ Hz), 175.8 ppm (d, $J = 2.2$ Hz). ^{19}F NMR (376 MHz, $\text{DMSO}-d_6$) δ : -116.1 ppm. MS (APCI) m/z : 271.0 [M+H] $^+$. Anal. Calcd (%) for $\text{C}_{16}\text{H}_{11}\text{FO}_3$: C, 71.11; H, 4.10. Found (%): C, 70.93; H, 4.35.

6-Chloro-3-(2-hydroxybenzyl)-4H-chromen-4-one

(4ac). Purified by column chromatography using 1:100 methanol-dichloromethane. Yield 201 mg (35%). Off-white solid, mp 197 - 199 °C. ^1H NMR (400 MHz, $\text{DMSO}-d_6$) δ : 3.64 (s, 2H), 6.69 (t, $J = 7.4$ Hz, 1H), 6.82 (d, $J = 8.0$ Hz, 1H), 7.02 (t, $J = 7.8$ Hz, 1H), 7.08 (d, $J = 7.5$ Hz, 1H), 7.60 (d, $J = 9.0$ Hz, 1H), 7.74 (d, $J = 8.5$ Hz, 1H), 7.93 (s, 1H), 8.08 (s, 1H), 9.48 ppm (s, 1H). $^{13}\text{C}\{\text{H}\}$ NMR (125 MHz, $\text{DMSO}-d_6$) δ : 25.3, 115.1, 118.9, 120.8, 122.7, 124.0, 124.1, 124.6, 127.4, 129.7, 130.0, 133.8, 154.4, 154.4, 155.2, 175.3 ppm. MS (APCI) m/z : 287.0 [M+H] $^+$. Anal. Calcd (%) for $\text{C}_{16}\text{H}_{11}\text{ClO}_3$: C, 67.03; H, 3.87. Found (%): C, 67.31; H, 4.12.

6-Bromo-3-(2-hydroxybenzyl)-4H-chromen-4-one

(4ad). Purified by column chromatography using 1:100 methanol-dichloromethane. 126 mg (19%). Off-white solid, mp 171 - 173 °C. ^1H NMR (400 MHz, $\text{DMSO}-d_6$) δ : 3.63 (s, 2H), 6.69 (t, $J = 7.4$ Hz, 1H), 6.80 (d, $J = 8.0$ Hz, 1H), 7.02 (t, $J = 7.7$ Hz, 1H), 7.07 (d, $J = 7.6$ Hz, 1H), 7.58 (d, $J = 8.6$ Hz, 1H), 7.85 - 7.94 (m, 1H), 8.04 - 8.16 (m, 2H), 9.48 ppm (s, 1H). $^{13}\text{C}\{\text{H}\}$ NMR (125 MHz, $\text{DMSO}-d_6$) δ : 25.3, 115.0, 117.6, 118.9, 121.1, 122.8, 124.6, 124.6, 127.1, 127.4, 130.0, 136.5, 154.4, 154.8, 155.1, 175.2 ppm. MS (APCI) m/z : 331.0 [M+H] $^+$. Anal. Calcd (%) for $\text{C}_{16}\text{H}_{11}\text{BrO}_3$: C, 58.03; H, 3.35. Found (%): C, 57.88; H, 3.42.

3-[(6-Hydroxy-1,3-benzodioxol-5-yl)methyl]-6-methyl-4H-chromen-4-one (4ba). Purified by column chromatography using 1:100 methanol-dichloromethane. Yield 366 mg (59%). Pale yellow solid, mp 239 - 241 °C. ^1H NMR (400 MHz, $\text{DMSO}-d_6$) δ : 2.38 (s, 3H), 3.54 (s, 2H), 5.85 (s, 2H), 6.45 (s, 1H), 6.71 (s, 1H), 7.45 (d, $J = 8.6$ Hz, 1H), 7.54 (dd, $J = 8.6, 2.2$ Hz, 1H), 7.82 (d, $J = 2.2$ Hz, 1H), 8.04 (s, 1H), 9.23 ppm (s, 1H). $^{13}\text{C}\{\text{H}\}$ NMR (125 MHz, $\text{DMSO}-d_6$) δ : 20.4, 25.4, 97.8, 100.5, 109.6, 116.5, 118.1, 122.8, 124.2, 134.7, 135.0, 139.6, 145.9, 149.5, 153.9, 154.2, 176.7 ppm. MS (APCI) m/z : 311.0 [M+H] $^+$. Anal. Calcd (%) for $\text{C}_{18}\text{H}_{14}\text{O}_5$: C, 69.67; H, 4.55. Found (%): C, 69.49; H, 4.69.

6-Fluoro-3-[(6-hydroxy-1,3-benzodioxol-5-yl)methyl]-4H-chromen-4-one (4bb). Purified by column chromatography using 1:100 methanol-dichloromethane. Yield 283 mg (45%). Pale yellow solid, mp 229 - 231 °C. ^1H NMR (400 MHz, $\text{DMSO}-d_6$) δ : 3.54 (s, 2H), 5.84 (s, 2H), 6.44 (d, $J = 1.9$ Hz, 1H), 6.70 (d, $J = 1.9$ Hz, 1H), 7.46 - 7.76 (m, 3H), 8.07 (s, 1H), 9.22 ppm (s, 1H). $^{13}\text{C}\{\text{H}\}$ NMR (100 MHz, $\text{DMSO}-d_6$) δ : 25.4, 97.7, 100.5, 109.4 (d, $J_{C-F} = 23.5$ Hz), 109.7, 116.2, 121.1 (d, $J_{C-F} = 8.3$ Hz), 122.1 (d, $J_{C-F} = 25.5$ Hz), 122.3, 124.2 (d, $J_{C-F} = 7.1$ Hz), 139.6, 146.0, 149.6, 152.4, 154.3, 158.8 (d, $J_{C-F} = 244.1$ Hz), 176.0 ppm (d, $J = 2.3$ Hz). ^{19}F NMR (376 MHz, $\text{DMSO}-d_6$) δ : -116.1 ppm. MS (APCI) m/z : 315.0 [M+H] $^+$. Anal. Calcd (%) for $\text{C}_{17}\text{H}_{11}\text{FO}_5$: C, 64.97; H, 3.53. Found (%): C, 65.11; H, 3.27.

6-Chloro-3-[(6-hydroxy-1,3-benzodioxol-5-yl)methyl]-4H-chromen-4-one (4bc). Purified by column chromatography using 1:100 methanol-dichloromethane. Yield 337 mg (51%). Pale yellow solid, mp 208 - 210 °C. ¹H NMR (400 MHz, DMSO-*d*₆) δ: 3.55 (s, 2H), 5.84 (s, 2H), 6.44 (s, 1H), 6.70 (s, 1H), 7.67 (d, *J* = 9.0 Hz, 1H), 7.81 (d, *J* = 9.0 Hz, 1H), 7.98 (s, 1H), 8.09 (s, 1H), 9.15 ppm (s, 1H). ¹³C{¹H} NMR (100 MHz, DMSO-*d*₆) δ: 25.3, 97.7, 100.5, 109.7, 116.1, 120.7, 123.0, 123.9, 124.1, 129.7, 133.7, 139.6, 146.0, 149.6, 154.2, 154.4, 175.5 ppm. MS (APCI) *m/z*: 331.0 [M+H]⁺. Anal. Calcd (%) for C₁₇H₁₁ClO₅: C, 61.74; H, 3.35. Found (%): C, 61.97; H, 3.48.

6-Bromo-3-[(6-hydroxy-1,3-benzodioxol-5-yl)methyl]-4H-chromen-4-one (4bd). Purified by column chromatography using 1:100 methanol-dichloromethane. Yield 270 mg (36%). Pale yellow solid, mp 224 - 226 °C. ¹H NMR (400 MHz, DMSO-*d*₆) δ: 3.55 (s, 2H), 5.84 (s, 2H), 6.44 (s, 1H), 6.70 (s, 1H), 7.58 (d, *J* = 8.9 Hz, 1H), 77.84 - 7.94 (m, 1H), 8.03 - 8.17 (m, 2H), 9.14 ppm (s, 1H). ¹³C{¹H} NMR (125 MHz, DMSO-*d*₆) δ: 25.3, 97.7, 100.5, 109.7, 116.1, 117.6, 121.1, 123.0, 124.6, 127.1, 136.5, 139.5, 145.9, 149.6, 154.3, 154.8, 175.4 ppm. MS (APCI) *m/z*: 375.0 [M+H]⁺. Anal. Calcd (%) for C₁₇H₁₁BrO₅: C, 54.42; H, 2.96. Found (%): C, 54.18; H, 3.17.

3-[(4-Hydroxy-1,3-benzodioxol-5-yl)methyl]-6-methyl-4H-chromen-4-one (4ca). Purified by column chromatography using 1:100 methanol-dichloromethane. Yield 149 mg (24%). Pale yellow solid, mp 198 - 200 °C. ¹H NMR (400 MHz, DMSO-*d*₆) δ: 2.39 (s, 3H), 3.54 (s, 2H), 5.84 (s, 2H), 6.45 (s, 1H), 6.71 (s, 1H), 7.46 (d, *J* = 8.6 Hz, 1H), 7.55 (dd, *J* = 8.6, *J* = 2.2 Hz, 1H), 7.82 (d, *J* = 2.2 Hz, 1H), 8.04 (s, 1H), 9.22 ppm (s, 1H). ¹³C{¹H} NMR (125 MHz, DMSO-*d*₆) δ: 20.5, 25.1, 100.1, 100.6, 118.1, 121.5, 122.3, 122.7, 122.9, 124.2, 134.8, 134.9, 135.2, 138.8, 146.9, 154.1, 154.2, 176.7 ppm. MS (APCI) *m/z*: 311.0 [M+H]⁺. Anal. Calcd (%) for C₁₈H₁₄O₅: C, 69.67; H, 4.55. Found (%): C, 69.88; H, 4.78.

6-Fluoro-3-[(4-hydroxy-1,3-benzodioxol-5-yl)methyl]-4H-chromen-4-one (4cb). Purified by column chromatography using 1:100 methanol-dichloromethane. Yield 151 mg (24%). Pale yellow solid, mp 236 - 237 °C. ¹H NMR (400 MHz, DMSO-*d*₆) δ: 3.61 (s, 2H), 5.93 (s, 2H), 6.36 (d, *J*=8.0, 1H), 6.62 (d, *J*=8.0, 1H), 7.57 - 7.78 (m, 3H), 8.12 (s, 1H), 9.53 (s, 1H). ¹³C{¹H} NMR (125 MHz, DMSO-*d*₆) δ = 25.0, 100.0, 100.6, 109.5 (d, *J*_{C-F} = 23.5 Hz), 121.2, 121.2 (d, *J*_{C-F} = 8.5 Hz), 122.3 (d, *J*_{C-F} = 25.6 Hz), 122.4, 124.1 (d, *J*_{C-F} = 7.1 Hz), 134.7, 138.8, 146.9, 152.4, 154.5, 158.9 (d, *J*_{C-F} = 244.0 Hz), 176.0 ppm. ¹⁹F NMR (376 MHz, DMSO-*d*₆) δ: -116.0 ppm. MS (APCI) *m/z*: 315.0 [M+H]⁺. Anal. Calcd (%) for C₁₇H₁₁FO₅: C, 64.97; H, 3.53. Found (%): C, 64.86; H, 3.78.

6-Chloro-3-[(4-hydroxy-1,3-benzodioxol-5-yl)methyl]-4H-chromen-4-one (4cc). Purified by column chromatography using 1:100 methanol-dichloromethane. Yield 278 mg (42%). Pale yellow solid, mp 230 - 232 °C. ¹H NMR (400 MHz, DMSO-*d*₆) δ: 3.60 (s, 2H), 5.94 (s, 2H), 6.36 (d, *J* = 8.0 Hz, 1H), 6.62 (d, *J* = 8.0 Hz, 1H), 7.68 (d, *J* = 8.9 Hz, 1H), 7.81 (d, *J* = 8.9 Hz, 1H), 7.97 (s, 1H), 8.11 (s, 1H), 9.53 ppm (s, 1H). ¹³C{¹H} NMR (100 MHz, DMSO-*d*₆) δ: 25.0, 100.0, 100.6, 120.9, 121.1, 122.4, 123.1, 124.0, 124.1, 129.7, 133.9, 134.7, 138.8, 146.9, 154.4, 175.5 ppm. MS (APCI) *m/z*: 331.0 [M+H]⁺. Anal. Calcd (%) for C₁₇H₁₁ClO₅: C, 61.74; H, 3.35. Found (%): C, 61.52; H, 3.09.

6-Bromo-3-[(4-hydroxy-1,3-benzodioxol-5-yl)methyl]-4H-chromen-4-one (4cd). Purified by column chromatography using 1:100 methanol-dichloromethane. Yield 398 mg (53%). Pale yellow solid, mp 208 - 210 °C. ¹H NMR (400 MHz, DMSO-*d*₆) δ: 3.59 (s, 2H), 5.94 (s, 2H), 6.35 (d, *J* = 8.0 Hz, 1H), 6.62 (d, *J* = 8.0 Hz, 1H), 7.55 (d, *J* = 8.9 Hz, 1H), 7.87 (dd, *J* = 8.9, 2.6 Hz, 1H), 8.02 - 8.13 (m, 2H), 9.55 ppm (s, 1H). ¹³C{¹H} NMR (125 MHz, DMSO-*d*₆) δ: 25.1, 100.0, 100.6, 117.7, 121.0, 121.1, 122.4, 123.2, 124.5, 127.1, 134.7, 136.6, 138.8, 146.9, 154.4, 154.8, 175.4 ppm. MS (APCI) *m/z*: 375.0 [M+H]⁺. Anal. Calcd (%) for C₁₇H₁₁BrO₅: C, 54.42; H, 2.96. Found (%): C, 54.70; H, 3.22.

3-[(2-Hydroxy-1-naphthyl)methyl]-6-methyl-4H-chromen-4-one (4da). Yield 278 mg (44%). Pale yellow solid, mp 193 - 195 °C. ¹H NMR (400 MHz, DMSO-*d*₆) δ: 2.38 (s, 3H), 4.12 (s, 2H), 7.16 - 7.30 (m, 2H), 7.31 - 7.43 (m, 2H), 7.45 - 7.56 (m, 2H), 7.72 (d, *J* = 8.8 Hz, 1H), 7.77 (d, *J* = 8.1 Hz, 1H), 7.84 (d, *J* = 8.5 Hz, 1H), 7.90 (s, 1H), 9.93 ppm (s, 1H). ¹³C{¹H} NMR (125 MHz, DMSO-*d*₆) δ: 20.1, 20.4, 115.5, 118.0, 118.3, 122.4, 122.5, 122.7, 124.3, 126.4, 128.3, 128.3, 128.4, 133.1, 134.8, 135.0, 153.0, 153.4, 154.0, 176.7 ppm. MS (APCI) *m/z*: 317.3 [M+H]⁺. Anal. Calcd (%) for C₂₁H₁₆O₃: C, 79.73; H, 5.10. Found (%): C, 79.45; H, 5.23.

6-Fluoro-3-[(2-hydroxy-1-naphthyl)methyl]-4H-chromen-4-one (4db). Yield 218 mg (34%). Pale yellow solid, mp 202 - 204 °C. ¹H NMR (400 MHz, DMSO-*d*₆) δ: 4.12 (s, 2H), 7.20 - 7.29 (m, 2H), 7.34 - 7.43 (m, 1H), 7.55 (s, 1H), 7.56 - 7.68 (m, 2H), 7.72 (d, *J* = 8.9 Hz, 1H), 7.75 - 7.81 (m, 1H), 7.83 (d, *J* = 8.6 Hz, 1H), 9.91 ppm (s, 1H). ¹³C{¹H} NMR (100 MHz, DMSO-*d*₆) δ: 20.0, 109.5 (d, *J*_{C-F} = 23.5 Hz), 115.2, 118.3, 121.1 (d, *J*_{C-F} = 8.4 Hz), 122.1, 122.3, 122.5, 122.7, 123.9 (d, *J*_{C-F} = 7.2 Hz), 126.5, 128.3, 128.3, 128.4, 133.1, 152.2, 153.1, 153.8, 158.8 (d, *J*_{C-F} = 244.1 Hz), 176.1 ppm. ¹⁹F NMR (376 MHz, DMSO-*d*₆) δ: -115.9 ppm. MS (APCI) *m/z*: 321.0 [M+H]⁺. Anal. Calcd (%) for C₂₀H₁₃FO₃: C, 74.99; H, 4.09. Found (%): C, 75.26; H, 4.30.

6-Chloro-3-[(2-hydroxy-1-naphthyl)methyl]-4H-chromen-4-one (4dc). Yield 290 mg (43%). Pale yellow solid, mp 211 - 213 °C. ¹H NMR (400 MHz, DMSO-*d*₆) δ: 4.11 (s, 2H), 7.17 - 7.29 (m, 2H), 7.38 (t, *J* = 7.7 Hz, 1H), 7.44 - 7.59 (m, 2H), 7.66 - 7.85 (m, 4H), 8.02 (s, 1H), 9.90 ppm (s, 1H). ¹³C{¹H} NMR (125 MHz, DMSO-*d*₆) δ: 20.0, 115.0, 118.2, 120.7, 122.4, 122.7, 123.0, 123.9, 124.0, 126.5, 128.2, 128.4, 129.7, 133.1, 133.8, 153.1, 153.8, 154.3, 175.6 ppm. MS (APCI) *m/z*: 337.0 [M+H]⁺. Anal. Calcd (%) for C₂₀H₁₃ClO₃: C, 71.33; H, 3.89. Found (%): C, 71.09; H, 4.07.

6-Bromo-3-[(2-hydroxy-1-naphthyl)methyl]-4H-chromen-4-one (4dd). Yield 480 mg (63%). Pale yellow solid, mp 210 - 212 °C. ¹H NMR (400 MHz, DMSO-*d*₆) δ: 4.12 (s, 2H), 7.16 - 7.31 (m, 2H), 7.38 (t, *J* = 7.8 Hz, 1H), 7.44 (d, *J* = 8.9 Hz, 1H), 7.52 (s, 1H), 7.64 - 7.90 (m, 4H), 8.15 (s, 1H), 9.89 ppm (s, 1H). ¹³C{¹H} NMR (125 MHz, DMSO-*d*₆) δ: 20.1, 115.1, 117.6, 118.3, 120.7, 122.4, 122.7, 123.1, 124.2, 126.4, 127.1, 128.2, 128.3, 133.1, 136.4, 153.1, 153.7, 154.5, 175.4 ppm. MS (APCI) *m/z*: 381.0 [M+H]⁺. Anal. Calcd (%) for C₂₀H₁₃BrO₃: C, 63.01; H, 3.44. Found (%): C, 79.48; H, 4.82.

3-[(6-Hydroxy-2-(1-methylethylidene)-3-oxo-2,3-dihydro-1-benzofuran-7-yl)methyl]-6-methyl-4H-

chromen-4-one (4ea). Yield 167 mg (23%). Off-white solid, mp 204 - 206 °C. ¹H NMR (500 MHz, DMSO-*d*₆) δ: 1.87 (s, 3H), 2.19 (s, 3H), 2.37 (s, 3H), 3.76 (s, 2H), 6.73 (d, *J* = 8.5 Hz, 1H), 7.33 - 7.46 (m, 2H), 7.46 - 7.56 (m, 1H), 7.81 (s, 1H), 7.94 (s, 1H), 10.87 ppm (s, 1H). ¹³C{¹H} NMR (125 MHz, DMSO-*d*₆) δ: 16.37, 18.66, 19.36, 20.41, 107.89, 111.79, 115.08, 117.96, 121.28, 122.57, 123.19, 124.18, 128.74, 134.73, 134.98, 144.74, 153.48, 154.13, 163.20, 164.38, 176.36, 181.47 ppm. MS (APCI) *m/z*: 363.1 [M+H]⁺. Anal. Calcd (%) for C₂₂H₁₈O₅: C, 72.92; H, 5.01. Found (%): C, 73.20; H, 5.30.

6-Fluoro-3-[(5-hydroxy-2-(1-methylethylidene)-3-oxo-2,3-dihydro-1-benzofuran-4-yl)methyl]-4H-chromen-4-one (4eb). Yield 256 mg (35%). Off-white solid, mp 239 - 241 °C. ¹H NMR (400 MHz, DMSO-*d*₆) δ: 1.87 (s, 3H), 2.19 (s, 3H), 3.76 (s, 2H), 6.72 (d, *J* = 8.4 Hz, 1H), 7.40 (d, *J* = 8.4 Hz, 1H), 7.54 - 7.80 (m, 3H), 8.04 (s, 1H), 10.92 ppm (s, 1H). ¹³C{¹H} NMR (125 MHz, DMSO-*d*₆) δ: 16.4, 18.6, 19.3, 107.6, 109.4 (d, *J*_{C-F} = 23.6 Hz), 111.7, 115.1, 120.9, 121.0 (d, *J*_{C-F} = 8.7 Hz), 122.0 (d, *J*_{C-F} = 25.5 Hz), 123.2, 124.0 (d, *J*_{C-F} = 7.1 Hz), 128.6, 144.8, 152.3, 153.9, 158.8 (d, *J*_{C-F} = 244.2 Hz), 163.2, 164.4, 175.7 (d, *J*_{C-F} = 2.2 Hz), 181.4 ppm. ¹⁹F NMR (376 MHz, DMSO-*d*₆) δ: -116.0 ppm. MS (APCI) *m/z*: 367.2 [M+H]⁺. Anal. Calcd (%) for C₂₁H₁₅FO₅: C, 68.85; H, 4.13. Found (%): C, 71.36; H, 4.27.

6-Chloro-3-[(6-hydroxy-2-(1-methylethylidene)-3-oxo-2,3-dihydro-1-benzofuran-7-yl)methyl]-4H-chromen-4-one (4ec). Yield 161 mg (21%). Off-white solid, mp 232 - 234 °C. ¹H NMR (500 MHz, DMSO-*d*₆) δ: δ 1.87 (s, 3H), 2.19 (s, 3H), 3.75 (s, 2H), 6.72 (d, *J* = 8.4 Hz, 1H), 7.40 (d, *J* = 8.4 Hz, 1H), 7.62 (d, *J* = 9.0 Hz, 1H), 7.76 (dd, *J* = 9.0, 2.9 Hz, 1H), 7.94 (s, 1H), 8.03 (s, 1H), 10.92 ppm (s, 1H). ¹³C{¹H} NMR (125 MHz, DMSO-*d*₆) δ: 16.4, 18.6, 19.4, 107.5, 111.7, 115.0, 120.8, 121.6, 123.3, 123.9, 128.8, 129.7, 133.8, 144.7, 154.0, 154.4, 163.2, 164.3, 175.2, 181.4 ppm. MS (APCI) *m/z*: 383.1 [M+H]⁺. Anal. Calcd (%) for C₂₁H₁₅ClO₅: C, 65.89; H, 3.95. Found (%): C, 66.04; H, 4.20.

6-Bromo-3-[(5-hydroxy-2-(1-methylethylidene)-3-oxo-2,3-dihydro-1-benzofuran-4-yl)methyl]-4H-chromen-4-one (4ed). Yield 444 mg (52%). Off-white solid, mp 278 - 280 °C. ¹H NMR (400 MHz, DMSO-*d*₆) δ: 1.90 (s, 3H), 2.22 (s, 3H), 3.80 (s, 2H), 6.73 (d, *J* = 8.4 Hz, 1H), 7.40 (d, *J* = 8.4 Hz, 1H), 7.56 (d, *J* = 8.8 Hz, 1H), 7.88 (dd, *J* = 8.8, *J* = 2.5 Hz, 1H), 8.03 (s, 1H), 8.13 (s, 1H), 10.66 ppm (s, 1H). ¹³C{¹H} NMR (125 MHz, DMSO-*d*₆) δ: 16.4, 18.7, 19.4, 107.5, 111.7, 115.0, 117.7, 121.0, 121.7, 123.3, 124.4, 127.1, 128.8, 136.6, 144.7, 154.0, 154.8, 163.2, 164.4, 175.1, 181.5 ppm. MS (APCI) *m/z*: 427.0 [M+H]⁺. Anal. Calcd (%) for C₂₁H₁₅BrO₅: C, 59.04; H, 3.54. Found (%): C, 58.86; H, 3.69.

Ethyl 5-Hydroxy-4-[(6-methyl-4-oxo-4H-chromen-3-yl)methyl]-2-phenyl-1-benzofuran-3-carboxylate (4fa). Yield 355 mg (39%). Off-white solid, mp 150 - 152 °C. ¹H NMR (400 MHz, DMSO-*d*₆) δ: 0.92 (t, *J* = 7.1 Hz, 3H), 2.40 (s, 3H), 3.80 - 4.10 (m, 4H), 7.03 (d, *J* = 8.9 Hz, 1H), 7.32 (s, 1H), 7.37 - 7.58 (m, 6H), 7.56 - 7.72 (m, 2H), 7.89 (s, 1H), 9.56 (s, 1H) ppm. ¹³C{¹H} NMR (100 MHz, DMSO-*d*₆) δ: 13.3, 20.4, 21.9, 61.3, 110.4, 110.7, 114.4, 114.5, 118.0, 122.3, 122.5, 124.2, 126.1, 126.9, 128.8, 129.0, 129.8, 134.7, 135.0, 147.6, 152.2, 152.5, 154.1, 154.8, 165.2, 176.4 ppm.

MS (APCI) *m/z*: 455.2 [M+H]⁺. Anal. Calcd (%) for C₂₈H₂₂O₆: C, 74.00; H, 4.88. Found (%): C, 74.27; H, 5.12.

Ethyl 4-[(6-Fluoro-4-oxo-4H-chromen-3-yl)methyl]-5-hydroxy-2-phenyl-1-benzofuran-3-carboxylate (4fb). Yield 321 mg (35%). Off-white solid, mp 177 - 178 °C. ¹H NMR (400 MHz, DMSO-*d*₆) δ: 0.93 (t, *J* = 7.1 Hz, 3H), 3.93 - 4.07 (m, 4H), 7.04 (d, *J* = 8.8 Hz, 1H), 7.38 - 7.54 (m, 5H), 7.60 - 7.68 (m, 4H), 7.76 (d, *J* = 8.3 Hz, 1H), 9.58 ppm (s, 1H). ¹³C{¹H} NMR (100 MHz, DMSO-*d*₆) δ: 13.3, 21.8, 61.4, 109.4 (d, *J*_{C-F} = 23.4 Hz), 110.5, 110.7, 114.2, 114.5, 121.0 (d, *J*_{C-F} = 8.3 Hz), 121.9, 122.2 (d, *J*_{C-F} = 25.4 Hz), 123.8 (d, *J*_{C-F} = 7.2 Hz), 126.1, 127.0, 128.8, 129.0, 129.8, 147.7, 152.3, 152.3, 153.2, 155.0, 158.8 (d, *J*_{C-F} = 244.2 Hz), 165.2, 175.8 ppm (d, *J*_{C-F} = 2.2 Hz). ¹⁹F NMR (376 MHz, DMSO-*d*₆) δ: -116.0 ppm. MS (APCI) *m/z*: 459.0 [M+H]⁺. Anal. Calcd (%) for C₂₇H₁₉FO₆: C, 70.74; H, 4.18. Found (%): C, 70.97; H, 4.32.

Ethyl 4-[(6-chloro-4-oxo-4H-chromen-3-yl)methyl]-5-hydroxy-2-phenyl-1-benzofuran-3-carboxylate (4fc). Yield 418 mg (44%). Off-white solid, mp 179 - 181 °C. ¹H NMR (400 MHz, DMSO-*d*₆) δ: 0.94 (t, *J* = 7.1 Hz, 3H), 3.80 - 4.23 (m, 4H), 7.03 (d, *J* = 8.9 Hz, 1H), 7.40 - 7.53 (m, 5H), 7.61 (d, *J* = 9.0 Hz, 1H), 7.64 - 7.69 (m, 2H), 7.78 (dd, *J* = 8.9, *J* = 2.7 Hz, 1H), 7.93 - 8.06 (m, 1H), 9.57 ppm (s, 1H). ¹³C{¹H} NMR (125 MHz, DMSO-*d*₆) δ: 13.3, 21.8, 61.3, 110.5, 110.7, 114.1, 114.4, 120.8, 122.6, 123.8, 123.9, 126.1, 127.0, 128.8, 129.0, 129.7, 129.8, 133.9, 147.7, 152.2, 153.2, 154.3, 155.0, 165.2, 175.4 ppm. MS (APCI) *m/z*: 475.0 [M+H]⁺. Anal. Calcd (%) for C₂₇H₁₉ClO₆: C, 68.29; H, 4.03. Found (%): C, 68.37; H, 4.29.

Ethyl 4-[(6-bromo-4-oxo-4H-chromen-3-yl)methyl]-5-hydroxy-2-phenyl-1-benzofuran-3-carboxylate (4fd). Yield 208 mg (20%). Off-white solid, mp 182 - 184 °C. ¹H NMR (500 MHz, DMSO-*d*₆) δ: 0.95 (t, *J* = 7.1 Hz, 3H), 3.95 - 4.05 (m, 4H), 7.02 (d, *J* = 8.9 Hz, 1H), 7.44 (s, 1H), 7.46 - 7.54 (m, 4H), 7.58 (d, *J* = 9.0 Hz, 1H), 7.63 - 7.73 (m, 2H), 7.93 (dd, *J* = 9.0, 2.5 Hz, 1H), 8.17 (d, *J* = 2.5 Hz, 1H), 9.53 ppm (s, 1H). ¹³C{¹H} NMR (125 MHz, DMSO-*d*₆) δ: 13.3, 21.8, 61.3, 110.5, 110.7, 114.0, 114.4, 120.8, 122.6, 123.8, 123.9, 126.1, 127.0, 128.8, 129.0, 129.7, 129.8, 133.9, 147.7, 152.2, 153.2, 154.3, 155.0, 165.2, 175.2 ppm. MS (APCI) *m/z*: 519.0 [M+H]⁺. Anal. Calcd (%) for C₂₇H₁₉BrO₆: C, 62.44; H, 3.69. Found (%): C, 73.74; H, 4.97.

7-Hydroxy-8-[(6-methyl-4-oxo-4H-chromen-3-yl)methyl]-2H-chromen-2-one (4ga). Yield 475 mg (71%). Off-white solid, mp 252 - 254 °C. ¹H NMR (400 MHz, DMSO-*d*₆) δ: 2.42 (s, 3H, CH₃), 3.81 (s, 2H, CH₂), 6.21 (d, *J* = 9.4 Hz, 1H, H-3'), 6.89 (d, *J* = 8.5 Hz, 1H, H-6'), 7.45 - 7.50 (m, 2H, H-8, 5'), 7.58 (dd, *J* = 8.5, *J* = 2.2 Hz, 1H, H-7), 7.68 (s, 1H, H-2), 7.86 (d, *J* = 2.2 Hz, 1H, H-5), 7.94 (d, *J* = 9.4 Hz, 1H, H-4'), 10.63 ppm (s, 1H, OH-7'). ¹³C{¹H} NMR (100 MHz, DMSO-*d*₆) δ: 18.7 (CH₂), 20.5 (CH₃), 110.9 (C-8'), 111.2 (CH-3'), 111.6 (C-4a'), 112.6 (CH-6'), 118.1 (CH-8), 121.2 (C-3), 122.5 (C-4a), 124.2 (CH-5), 127.8 (CH-5'), 134.9 (C-6), 135.2 (CH-7), 144.9 (CH-4'), 153.1 (CH-2), 153.6 (C-8a'), 154.1 (C-8a), 159.2 (C-7'), 160.4 (C-2'), 176.5 ppm (CO-4). MS (APCI) *m/z*: 335.1 [M+H]⁺. Anal. Calcd (%) for C₂₀H₁₄O₅: C, 71.85; H, 4.22. Found (%): C, 71.97; H, 4.03.

8-[(6-Fluoro-4-oxo-4H-chromen-3-yl)methyl]-7-hydroxy-2H-chromen-2-one (4gb). Yield 453 mg (67%).

Off-white solid, mp 254 - 256 °C. ¹H NMR (400 MHz, DMSO-*d*₆) δ: 3.82 (s, 2H), 6.20 (d, *J* = 9.4 Hz, 1H), 6.88 (d, *J* = 8.4 Hz, 1H), 7.47 (d, *J* = 8.4 Hz, 1H), 7.60 - 7.75 (m, 3H), 7.77 (s, 1H), 7.93 (d, *J* = 9.4 Hz, 1H), 10.62 ppm (s, 1H). ¹³C{¹H} NMR (100 MHz, DMSO-*d*₆) δ: 18.5, 109.2 (d, *J*_{C-F} = 23.4 Hz), 110.5, 111.0, 111.4, 112.3, 120.6, 120.9 (d, *J*_{C-F} = 8.2 Hz), 122.0 (d, *J*_{C-F} = 25.7 Hz), 123.7 (d, *J*_{C-F} = 7.3 Hz), 127.6, 144.6, 152.1, 153.4, 153.5, 158.6 (d, *J*_{C-F} = 244.1 Hz), 159.1, 160.2, 175.7 ppm. ¹⁹F NMR (376 MHz, DMSO-*d*₆) δ = -116.0 ppm. MS (APCI) *m/z*: 339.0 [M+H]⁺. Anal. Calcd (%) for C₁₉H₁₁FO₅: C, 67.46; H, 3.28. Found (%): C, 67.73; H, 3.53.

8-[(6-Chloro-4-oxo-4*H*-chromen-3-yl)methyl]-7-hydroxy-2*H*-chromen-2-one (4gc). Yield 468 mg (66%). Off-white solid, mp 268 - 270 °C. ¹H NMR (400 MHz, DMSO-*d*₆) δ: 3.84 (s, 2H), 6.19 (d, *J* = 9.4 Hz, 1H), 6.89 (d, *J* = 8.4 Hz, 1H), 7.47 (d, *J* = 8.4 Hz, 1H), 7.65 (d, *J* = 9.0 Hz, 1H), 7.75 - 7.86 (m, 2H), 7.93 (d, *J* = 9.4 Hz, 1H), 8.01 (d, *J* = 2.7 Hz, 1H), 10.52 ppm (s, 1H). ¹³C{¹H} NMR (100 MHz, DMSO-*d*₆) δ: 18.7, 110.6, 111.2, 111.6, 112.5, 120.9, 121.6, 123.9, 124.0, 127.9, 129.8, 133.9, 144.9, 153.6, 153.7, 154.4, 159.3, 160.4, 175.4 ppm. MS (APCI) *m/z*: 354.9 [M+H]⁺. Anal. Calcd (%) for C₁₉H₁₁ClO₅: C, 64.33; H, 3.13. Found (%): C, 64.08; H, 3.02.

8-[(6-Bromo-4-oxo-4*H*-chromen-3-yl)methyl]-7-hydroxy-2*H*-chromen-2-one (4gd). Yield 615 mg (77%). Off-white solid, mp 314 - 315 °C. ¹H NMR (400 MHz, DMSO-*d*₆) δ: 3.82 (s, 2H), 6.20 (d, *J* = 9.4 Hz, 1H), 6.89 (d, *J* = 8.5 Hz, 1H), 7.47 (d, *J* = 8.5 Hz, 1H), 7.58 (d, *J* = 8.9 Hz, 1H), 7.77 (s, 1H), 7.88 - 8.00 (m, 2H), 8.14 (d, *J* = 2.5 Hz, 1H), 10.60 ppm (s, 1H). ¹³C{¹H} NMR (125 MHz, DMSO-*d*₆) δ: 18.7, 110.6, 111.2, 111.6, 112.5, 117.7, 121.0, 121.6, 124.2, 127.1, 127.9, 136.6, 144.8, 153.6, 153.6, 154.7, 159.3, 160.4, 175.35 ppm. MS (APCI) *m/z*: 310.2 [M+H]⁺. Anal. Calcd (%) for C₁₉H₁₁BrO₅: C, 57.17; H, 2.78. Found (%): C, 57.43; H, 3.03.

8-[(6-Fluoro-4-oxo-4*H*-chromen-3-yl)methyl]-7-hydroxy-3-(4-methoxyphenyl)-4*H*-chromen-4-one (4hb). Yield 453 mg (51%). Off-white solid, mp 118 - 120 °C. ¹H NMR (400 MHz, DMSO-*d*₆) δ: 3.76 (s, 3H), 3.90 (s, 2H), 6.95 (d, *J* = 8.4 Hz, 2H), 7.05 (d, *J* = 8.7 Hz, 1H), 7.49 (d, *J* = 8.4 Hz, 2H), 7.59 - 7.77 (m, 3H), 7.83 (s, 1H), 7.93 (d, *J* = 8.7 Hz, 1H), 8.35 (s, 1H), 10.84 ppm (s, 1H). ¹³C{¹H} NMR (100 MHz, DMSO-*d*₆) δ: 18.8, 55.1, 109.5 (d, *J*_{C-F} = 23.5 Hz), 110.9, 113.6, 114.5, 117.0, 120.9, 121.1 (d, *J*_{C-F} = 8.2 Hz), 122.3 (d, *J*_{C-F} = 25.2 Hz), 122.8, 123.9 (d, *J*_{C-F} = 7.1 Hz), 124.3, 125.3, 130.0, 152.3, 153.2, 153.8, 155.6, 158.9 (d, *J*_{C-F} = 244.2 Hz), 158.9, 160.4, 175.0, 175.9 ppm. ¹⁹F NMR (376 MHz, DMSO-*d*₆) δ: -116.0 ppm. MS (APCI) *m/z*: 445.0 [M+H]⁺. Anal. Calcd (%) for C₂₆H₁₇FO₆: C, 70.27; H, 3.86. Found (%): C, 70.54; H, 4.11.

8-[(6-Chloro-4-oxo-4*H*-chromen-3-yl)methyl]-7-hydroxy-3-(4-methoxyphenyl)-4*H*-chromen-4-one (4hc). Yield 802 mg (87%). Off-white solid, mp 250 - 252 °C. ¹H NMR (400 MHz, DMSO-*d*₆) δ: 3.77 (s, 3H), 3.91 (s, 2H), 6.97 (d, *J* = 8.7 Hz, 2H), 7.05 (d, *J* = 8.7 Hz, 1H), 7.50 (d, *J* = 8.7 Hz, 2H), 7.66 (d, *J* = 9.0 Hz, 1H), 7.81 (dd, *J* = 9.0, *J* = 2.7 Hz, 1H), 7.84 (s, 1H), 7.94 (d, *J* = 8.7 Hz, 1H), 8.01 (d, *J* = 2.7 Hz, 1H), 8.37 (s, 1H), 10.84 ppm (s, 1H). ¹³C{¹H} NMR (125 MHz, DMSO-*d*₆) δ: 18.8, 55.1, 110.8, 113.5, 114.4, 116.9, 120.8, 121.6, 122.8, 123.8, 123.9, 124.2, 125.3, 129.7, 129.9, 133.9, 153.1, 153.7, 154.4, 155.6, 158.9, 160.3,

175.0, 175.4 ppm. MS (APCI) *m/z*: 461.0 [M+H]⁺. Anal. Calcd (%) for C₂₆H₁₇ClO₆: C, 67.76; H, 3.72. Found (%): C, 67.53; H, 3.99.

Ethyl 5-hydroxy-1,2-dimethyl-4-[(6-methyl-4-oxo-4*H*-chromen-3-yl)methyl]-1*H*-indole-3-carboxylate (4ia). Yield 300 mg (37%). Off-white solid, mp 157 - 159 °C. ¹H NMR (400 MHz, DMSO-*d*₆) δ: 1.01 (t, *J* = 7.1 Hz, 3H), 2.40 (s, 3H), 2.50 (s, 3H), 3.64 (s, 3H), 3.95 (q, *J* = 7.1 Hz, 2H), 4.15 (s, 2H), 6.87 (d, *J* = 8.7 Hz, 1H), 7.06 (s, 1H), 7.26 (d, *J* = 8.7 Hz, 1H), 7.38 (d, *J* = 8.6 Hz, 1H), 7.52 (d, *J* = 8.6 Hz, 1H), 7.89 (s, 1H), 8.96 ppm (s, 1H). ¹³C{¹H} NMR (100 MHz, DMSO-*d*₆) δ: 11.9, 14.0, 20.5, 22.7, 29.7, 59.2, 104.8, 109.0, 111.6, 113.1, 117.9, 122.6, 123.4, 124.2, 125.2, 131.3, 134.5, 134.8, 143.0, 150.4, 152.1, 154.0, 165.6, 176.8 ppm. MS (APCI) *m/z*: 406.2 [M+H]⁺. Anal. Calcd (%) for C₂₄H₂₃NO₅: C, 71.10; H, 5.72; N, 3.45. Found (%): C, 71.34; H, 5.99; N, 3.73.

Ethyl 4-[(6-fluoro-4-oxo-4*H*-chromen-3-yl)methyl]-5-hydroxy-1,2-dimethyl-1*H*-indole-3-carboxylate (4ib). Yield 352 mg (43%). Off-white solid, mp 179 - 180 °C. ¹H NMR (400 MHz, DMSO-*d*₆) δ: 1.01 (t, *J* = 7.1 Hz, 3H), 2.52 (s, 3H), 3.65 (s, 3H), 3.95 (q, *J* = 7.1 Hz, 2H), 4.14 (s, 2H), 6.86 (d, *J* = 8.7 Hz, 1H), 7.13 (s, 1H), 7.28 (d, *J* = 8.7 Hz, 1H), 7.61 - 7.69 (m, 2H), 7.80 - 7.72 (m, 1H), 8.96 ppm (s, 1H). ¹³C{¹H} NMR (125 MHz, DMSO-*d*₆) δ: 11.9, 13.9, 22.6, 29.7, 59.2, 104.7, 109.0, 109.4 (d, *J*_{C-F} = 23.6 Hz), 111.5, 112.8, 121.0 (d, *J*_{C-F} = 7.6 Hz), 122.0 (d, *J*_{C-F} = 25.6 Hz), 123.1, 123.9 (d, *J*_{C-F} = 6.8 Hz), 125.2, 131.3, 143.0, 150.4, 152.2, 152.5, 158.7 (d, *J*_{C-F} = 243.4 Hz), 165.5, 176.1 ppm. ¹⁹F NMR (376 MHz, DMSO-*d*₆) δ: -116.3 ppm. MS (APCI) *m/z*: 410.0 [M+H]⁺. Anal. Calcd (%) for C₂₃H₂₀FNO₅: C, 67.48; H, 4.92; N, 3.42. Found (%): C, 67.71; H, 4.66; N, 3.20.

Ethyl 4-[(6-chloro-4-oxo-4*H*-chromen-3-yl)methyl]-5-hydroxy-1,2-dimethyl-1*H*-indole-3-carboxylate (4ic). Yield 196 mg (23%). Off-white solid, mp 185 - 187 °C. ¹H NMR (400 MHz, DMSO-*d*₆) δ: 1.01 (t, *J* = 7.1 Hz, 3H), 2.50 (s, 3H), 3.63 (s, 3H), 3.94 (q, *J* = 7.1 Hz, 2H), 4.13 (s, 2H), 6.85 (d, *J* = 8.7 Hz, 1H), 7.11 (s, 1H), 7.26 (d, *J* = 8.7 Hz, 1H), 7.57 (d, *J* = 8.9 Hz, 1H), 7.76 (dd, *J* = 8.9, *J* = 2.7 Hz, 1H), 8.02 (d, *J* = 2.7 Hz, 1H), 8.95 ppm (s, 1H). ¹³C{¹H} NMR (125 MHz, DMSO-*d*₆) δ: 11.9, 13.9, 22.6, 29.7, 59.2, 104.7, 109.0, 111.5, 112.7, 120.7, 123.8, 123.9, 125.2, 129.5, 131.3, 133.7, 143.1, 150.4, 152.5, 154.3, 165.5, 175.6 ppm. MS (APCI) *m/z*: 426.1 [M+H]⁺. Anal. Calcd (%) for C₂₃H₂₀ClNO₅: C, 64.87; H, 4.73; N, 3.29. Found (%): C, 65.03; H, 4.68; N, 3.40.

3-[(8-Hydroxyquinolin-7-yl)methyl]-6-methyl-4*H*-chromen-4-one (4ja). Purified by column chromatography using 1:100 methanol-dichloromethane. Yield 298 mg (47%). Pale yellow solid, mp 156 - 158 °C. ¹H NMR (400 MHz, CDCl₃) δ: 2.37 (s, 3H), 4.01 (s, 2H), 7.19 - 7.27 (m, 2H), 7.28 - 7.40 (m, 2H), 7.56 (d, *J* = 8.4 Hz, 1H), 7.86 (s, 1H), 7.97 (d, *J* = 2.1 Hz, 1H), 8.04 (dd, *J* = 8.4, 1.6 Hz, 1H), 8.70 ppm (dd, *J* = 4.2, 1.6 Hz, 1H). ¹³C{¹H} NMR (101 MHz, CDCl₃) δ: 21.0, 26.0, 117.5, 117.8, 120.9, 121.3, 123.1, 123.6, 125.2, 127.4, 130.4, 134.6, 134.8, 136.1, 138.3, 148.0, 149.8, 153.7, 154.8, 177.8 ppm. MS (APCI) *m/z*: 318.2 [M+H]⁺. Anal. Calcd (%) for C₂₀H₁₅NO₃: C, 75.70; H, 4.76; N, 4.41. Found (%): C, 75.90; H, 4.97; N, 4.64.

6-Fluoro-3-[(8-hydroxyquinolin-7-yl)methyl]-4*H*-chromen-4-one (4jb). Purified by column chromatography

using 1:100 methanol-dichloromethane. Yield 276 mg (43%). Pale yellow solid, mp 161 - 163 °C. ¹H NMR (400 MHz, DMSO-*d*₆) δ: 3.89 (s, 2H), 7.33 (d, *J* = 8.4 Hz, 1H), 7.43 (d, *J* = 8.4 Hz, 1H), 7.50 (dd, *J* = 8.3, 4.2 Hz, 1H), 7.62 - 7.81 (m, 3H), 8.20 (s, 1H), 8.28 (d, *J* = 8.3 Hz, 1H), 8.83 (d, *J* = 4.2 Hz, 1H), 9.87 ppm (s, 1H). ¹³C{¹H} NMR (101 MHz, CDCl₃) δ: 26.03, 110.66 (d, *J*_{C-F} = 23.6 Hz), 117.62, 120.22 (d, *J*_{C-F} = 8.1 Hz), 120.47, 121.42, 121.67 (d, *J*_{C-F} = 25.6 Hz), 122.75, 125.10 (d, *J*_{C-F} = 7.3 Hz), 127.50, 130.45, 136.12, 138.23, 148.03, 149.85, 152.76, 153.98, 159.41 (d, *J*_{C-F} = 246.4 Hz), 177.03 ppm. ¹⁹F NMR (376 MHz, CDCl₃) δ: -116.1 ppm. MS (APCI) *m/z*: 322.0 [M+H]⁺. Anal. Calcd (%) for C₁₉H₁₂FNO₃: C, 71.03; H, 3.76; N, 4.36. Found (%): C, 71.31; H, 3.94; N, 4.46.

6-Chloro-3-[(8-hydroxyquinolin-7-yl)methyl]-4H-chromen-4-one (4jc). Purified by column chromatography using 1:100 methanol-dichloromethane. Yield 365 mg (54%). Pale yellow solid, mp 180 - 182 °C. ¹H NMR (400 MHz, CDCl₃) δ: 4.01 (s, 2H), 7.28 (d, *J* = 8.5 Hz, 1H), 7.32 (d, *J* = 8.5 Hz, 1H), 7.37 (dd, *J* = 8.3, 4.3 Hz, 1H), 7.52 (dd, *J* = 8.5, 2.6 Hz, 1H), 7.57 (d, *J* = 8.5 Hz, 1H), 7.90 (s, 1H), 8.09 (d, *J* = 8.3 Hz, 1H), 8.16 (d, *J* = 2.6 Hz, 1H), 8.73 ppm (d, *J* = 4.3 Hz, 1H). ¹³C{¹H} NMR (101 MHz, CDCl₃) δ: 26.0, 117.7, 119.9, 120.3, 121.5, 123.5, 125.0, 125.4, 127.5, 130.5, 130.9, 133.7, 136.1, 138.3, 148.1, 149.9, 153.9, 154.9, 176.6 ppm. MS (APCI) *m/z*: 338.0 [M+H]⁺. Anal. Calcd (%) for C₁₉H₁₂ClNO₃: C, 67.57; H, 3.58; N, 4.15. Found (%): C, 67.39; H, 3.72; N, 4.23.

6-Bromo-3-[(8-hydroxyquinolin-7-yl)methyl]-4H-chromen-4-one (4jd). Purified by column chromatography using 1:100 methanol-dichloromethane. Yield 298 mg (39%). Pale yellow solid, mp 198 - 200 °C. ¹H NMR (400 MHz, DMSO-*d*₆) δ: 3.89 (s, 2H), 7.33 (d, *J* = 8.4 Hz, 1H), 7.43 (d, *J* = 8.4 Hz, 1H), 7.51 (dd, *J* = 8.3, 4.2 Hz, 1H), 7.63 (d, *J* = 8.9 Hz, 1H), 7.93 (dd, *J* = 9.0, 2.6 Hz, 1H), 8.11 (d, *J* = 2.5 Hz, 1H), 8.21 (s, 1H), 8.29 (dd, *J* = 8.3, 1.7 Hz, 1H), 8.83 (dd, *J* = 4.2, 1.7 Hz, 1H), 9.87 ppm (s, 1H). ¹³C{¹H} NMR (126 MHz, DMSO-*d*₆) δ: 25.5, 117.0, 117.6, 120.8, 121.1, 121.3, 122.6, 124.6, 127.1, 127.3, 129.3, 136.0, 136.6, 138.0, 148.1, 150.3, 154.6, 154.8, 175.2 ppm. MS (APCI) *m/z*: 382.0 [M+H]⁺, 384.0 [M+H]⁺. Anal. Calcd (%) for C₁₉H₁₂BrNO₃: C, 59.71; H, 3.16; N, 3.66. Found (%): C, 59.87; H, 3.42; N, 3.78.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website.

Copies of ¹H, ¹³C{¹H} spectra for all new synthesized compounds and 2D for compound 4ga (PDF).

AUTHOR INFORMATION

Corresponding Authors

* VMS E-mail: vitaliy.sviripa@uky.edu

* MFS E-mail: mykhaylo.frasinyuk@ukr.net

Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENT

VMS was supported by grant #IRG 16-182-28 from the American Cancer Society (ACS). We thank the College of Pharmacy NMR Center (University of Kentucky) for NMR support.

REFERENCES

- (1) (a) Castelli, M. V.; López, S. N. In *Studies in Natural Products Chemistry*; Atta-ur-Rahman, Ed.; Elsevier: 2017; Vol. 54, p 315-354. (b) Lin, L.-G.; Liu, Q.-Y.; Ye, Y. Naturally Occurring Homoisoflavanoids and Their Pharmacological Activities. *Planta Med.* **2014**, *80*, 1053-1066.
- (2) Dewick, P. M. Biosynthesis of the 3-benzylchroman-4-one eucomin in Eucomis bicolor. *Phytochemistry* **1975**, *14*, 983-988.
- (3) (a) Chang, J.-M.; Shen, C.-C.; Huang, Y.-L.; Chien, M.-Y.; Ou, J.-C.; Shieh, B.-J.; Chen, C.-C. Five new homoisoflavonoids from the tuber of *Ophiopogon japonicus*. *J. Nat. Prod.* **2002**, *65*, 1731-1733. (b) González, A. G.; León, F.; Sánchez-Pinto, L.; Padrón, J. I.; Bermejo, J. Phenolic Compounds of Dragon's Blood from *Dracaena draco*. *J. Nat. Prod.* **2000**, *63*, 1297-1299. (c) Hoang Anh, N. T.; Van Sung, T.; Porzel, A.; Franke, K.; Wessjohann, L. A. Homoisoflavonoids from *Ophiopogon japonicus* Ker-Gawler. *Phytochemistry* **2003**, *62*, 1153-1158. (d) Qi, J.; Xu, D.; Zhou, Y.-F.; Qin, M.-J.; Yu, B.-Y. New features on the fragmentation patterns of homoisoflavonoids in *Ophiopogon japonicus* by high-performance liquid chromatography/diode-array detection/electrospray ionization with multi-stage tandem mass spectrometry. *Rapid Commun. Mass Spectrom.* **2010**, *24*, 2193-2206.
- (4) (a) Gan, L.-S.; Zeng, L.-W.; Li, X.-R.; Zhou, C.-X.; Li, J. New homoisoflavonoid analogues protect cells by regulating autophagy. *Bioorg. Med. Chem. Lett.* **2017**, *27*, 1441-1445. (b) Siddaiyah, V.; Rao, C. V.; Venkateswarlu, S.; Subbaraju, G. V. A concise synthesis of polyhydroxydihydrochalcones and homoisoflavonoids. *Tetrahedron* **2006**, *62*, 841-846. (c) Thanigaimalai, P.; Le Hoang, T. A.; Lee, K.-C.; Sharma, V. K.; Bang, S.-C.; Yun, J. H.; Roh, E.; Kim, Y.; Jung, S.-H. Synthesis and evaluation of novel chromone analogs for their inhibitory activity against interleukin-5. *Eur. J. Med. Chem.* **2010**, *45*, 2531-2536. (d) Basha, G. M.; Yadav, S. K.; Srinuvasarao, R.; Prasanthi, S.; Ramu, T.; Mangarao, N.; Siddaiyah, V. A mild and efficient protocol to synthesize chromones, isoflavones, and homoisoflavones using the complex 2,4,6-trichloro-1,3,5-triazine/dimethylformamide. *Can. J. Chem.* **2013**, *91*, 763-768. (e) Lee, B.; Basavarajappa, H. D.; Sulaiman, R. S.; Fei, X.; Seo, S.-Y.; Corson, T. W. The first synthesis of the antiangiogenic homoisoflavanone, cremastranone. *Org. Biomol. Chem.* **2014**, *12*, 7673-7677. (f) Ismail, K. A.; Abd El Azem, T. Synthesis and biological evaluation of some novel 4H-benzopyran-4-one derivatives as nonsteroidal antiestrogens. *Eur. J. Med. Chem.* **2001**, *36*, 243-253.
- (5) (a) Tada, A.; Saitoh, T.; Shoji, J. Studies on the Constituents of Ophiopogonis Tubers. VII. Synthetic Studies of Homoisoflavonoids. (3). *Chem. Pharm. Bull.* **1980**, *28*, 2487-2493. (b) Davis, F. A.; Chen, B. C. Enantioselective synthesis of (+)-O-trimethylsappanone B and (+)-O-trimethylbrazilin. *J. Org. Chem.* **1993**, *58*, 1751-1753. (c) Kirkiacharian, S.; Tong, H. G.; Bastide, J.; Bastide, P.; Grenie, M. M. Synthese et activité angioprotectrice, anti-allergique et antihistaminique de benzyl-3 chromones (homo-isoflavones). *Eur. J. Med. Chem.* **1989**, *24*, 541-546. (d) Davis, F. A.; Chen, B.-C. A highly enantioselective synthesis of (R)- and (S)-5,7-dimethyleucomol. *Tetrahedron Lett.* **1990**, *31*, 6823-6826.
- (6) (a) Poisson, T.; Gembus, V.; Dalla, V.; Oudeyer, S.; Levacher, V. Organocatalyzed Enantioselective Protonation of Silyl Enol Ethers: Scope, Limitations, and Application to the Preparation of Enantioenriched Homoisoflavones. *J. Org. Chem.* **2010**, *75*, 7704-7716. (b) Kirkiacharian, B. S.; Gomis, M. New Convenient Synthesis of Homoisoflavanones and (±)-Di-O-methyldihydroeucomin. *Synth. Commun.* **2005**, *35*, 563-569. (c)

Zhao, W.; Sun, J.; Xiang, H.; Zeng, Y. Y.; Li, X. B.; Xiao, H.; Chen, D. Y.; Ma, R. L. Synthesis and biological evaluation of new flavonoid fatty acid esters with anti-adipogenic and enhancing glucose consumption activities. *Bioorg. Med. Chem.* **2011**, *19*, 3192-3203.

(7) Wang, L.; Li, Z.-W.; Zhang, W.; Xu, R.; Gao, F.; Liu, Y.-F.; Li, Y.-J. Synthesis, Crystal Structure, and Biological Evaluation of a Series of Phloretin Derivatives. *Molecules* **2014**, *19*, 16447-16457.

(8) (a) Luo, S.; Mi, X.; Xu, H.; Wang, P. G.; Cheng, J.-P. Efficient Baylis–Hillman Reactions of Cyclic Enones in Methanol As Catalyzed by Methoxide Anion. *J. Org. Chem.* **2004**, *69*, 8413-8422. (b) Basavaiah, D.; Jannapu Reddy, R.; Srivardhana Rao, J. Applications of Baylis–Hillman adducts: a simple, convenient, and one-pot synthesis of 3-benzoylquinolines. *Tetrahedron Lett.* **2006**, *47*, 73-77. (c) Bharate, S. B.; Kumar, V.; Jain, S. K.; Mintoo, M. J.; Guru, S. K.; Nuthakki, V. K.; Sharma, M.; Bharate, S. S.; Gandhi, S. G.; Mondhe, D. M.; Bhushan, S.; Vishwakarma, R. A. Discovery and Preclinical Development of IIIM-290, an Orally Active Potent Cyclin-Dependent Kinase Inhibitor. *J. Med. Chem.* **2018**, *61*, 1664-1687. (d) Basavaiah, D.; Jagannathan Rao, A. 1-Benzopyran-4(4H)-ones as novel activated alkenes in the Baylis–Hillman reaction: a simple and facile synthesis of indolizine-fused-chromones. *Tetrahedron Lett.* **2003**, *44*, 4365-4368.

(9) (a) Badawy, D. S.; Kandeel; Awad, N. M.; Abdel-Rahman, A.-R. H. Synthesis of some new naphthopyran, pyrazole, pyridine, and thienobenzochromene derivatives using 1-(1-hydroxy-2-naphthyl) ethanone as a versatile starting material. *Phosphorus, Sulfur Silicon Relat. Elem.* **2008**, *184*, 179-196. (b) Gammill, R. B.; Day, C. E.; Schurr, P. E. Khellin analogs. 1. General topological requirements for lipid-altering activity in furochromones. *J. Med. Chem.* **1983**, *26*, 1672-1674. (c) Boggu, P. R.; Venkateswararao, E.; Manickam, M.; Kim, Y.; Jung, S.-H. Discovery of novel 3-(hydroxalkoxy)-2-alkylchromen-4-one analogs as interleukin-5 inhibitors. *Eur. J. Med. Chem.* **2017**, *139*, 290-304.

(10) Bolós, J.; Gubert, S.; Anglada, L.; Planas, J. M.; Burgarolas, C.; Castelló, J. M.; Sacristán, A.; Ortiz, J. A. 7-[3-(1-Piperidinyl)propoxy]chromenes as Potential Atypical Antipsychotics. *J. Med. Chem.* **1996**, *39*, 2962-2970.

(11) (a) Biegasiewicz, K. F.; Denis, J. D. S.; Carroll, V. M.; Priefer, R. An efficient synthesis of daidzein, dimethyl daidzein, and isoformonetin. *Tetrahedron Lett.* **2010**, *51*, 4408-4410. (b) Denis, J. D. S.; Gordon Iv, J. S.; Carroll, V. M.; Priefer, R. Novel synthesis of the isoflavone genistein. *Synthesis* **2010**, 1590-1592. (c) Selepe, M. A.; Drewes, S. E.; Van Heerden, F. R. Total synthesis of the pyranoisoflavone kraussianone 1 and related isoflavones. *J. Nat. Prod.* **2010**, *73*, 1680-1685. (d) Li, G.; Zhang, Z.-T.; Dai, L.-Y.; Du, Y.-L.; Xue, D. Synthesis of Novel Disubstituted Pyrazolo[1,5-al]pyrimidines, Imidazo[1,2-a]pyrimidines, and Pyrimido[1,2-al]benzimidazoles Containing Thioether and Aryl Moieties. *Helv. Chim. Acta* **2012**, *95*, 989-997. (e) Dejon, L.; Mohammed, H.; Du, P.; Jacob, C.; Speicher, A. Synthesis of chromenoindole derivatives from *Robinia pseudoacacia*. *MedChemComm* **2013**, *4*, 1580-1583. (f) Biegasiewicz, K. F.; Gordon, J. S.; Rodriguez, D. A.; Priefer, R. Development of a general approach to the synthesis of a library of isoflavonoid derivatives. *Tetrahedron Lett.* **2014**, *55*, 5210-5212.

(12) Zhang, X.-Z.; Ge, D.-L.; Chen, S.-Y.; Yu, X.-Q. A catalyst-free approach to 3-thiocyanato-4H-chromen-4-ones. *RSC Adv.* **2016**, *6*, 66320-66323.

(13) (a) Panja, S. K.; Maiti, S.; Bandyopadhyay, C. Synthesis of 3-allylchromones, homoisoflavones and bischromones from (*E*)-1-(2-hydroxyphenyl)-3-(*N,N*-dimethylamino)prop-2-en-1-one. *J. Chem. Res.* **2010**, *34*, 555-558. (b) Xiang, H.; Zhao, Q.; Tang, Z.; Xiao, J.; Xia, P.; Wang, C.; Yang, C.; Chen, X.; Yang, H. Visible-light-driven, radical-triggered tandem cyclization of *o*-hydroxyaryl enaminones: Facile access to 3-CF₂/CF₃-containing chromones. *Org. Lett.* **2017**, *19*, 146-149. (c) Joussot, J.; Schoenfelder, A.; Larquetoux, L.; Nicolas, M.; Suffert, J.; Blond, G. Synthesis of 3-Substituted Chromones and Quinolones from Enaminones. *Synthesis* **2016**, *48*, 3364-3372.

(14) (a) Iaroshenko, V. O.; Mkrtchyan, S.; Ghazaryan, G.; Hakobyan, A.; Maalik, A.; Supe, L.; Villinger, A.; Tolmachev, A.; Ostrovsky, D.; Sosnovskikh, V. Y.; Ghochikyan, T. V.; Langer, P. 3-(Dichloroacetyl)chromone; A new building block for the synthesis of formylated purine isosteres: Design and Synthesis of Fused α -(formyl)pyridines. *Synthesis* **2010**, 469-479. (b) Iaroshenko, V. O.; Bunescu, A.; Spannenberg, A.; Supe, L.; Milyutina, M.; Langer, P. 3-Methoxalylchromones - versatile reagents for the regioselective synthesis of functionalized 2,4'-dihydroxybenzophenones, potential UV-filters. *Org. Biomol. Chem.* **2011**, *9*, 7554-7558.

(15) Lin, Y.-F.; Fong, C.; Peng, W.-L.; Tang, K.-C.; Liang, Y.-E.; Li, W.-T. Synthesis of 3-(2-Olefinbenzyl)-4H-chromen-4-one through Cyclobenzylation and Catalytic C–H Bond Functionalization Using Palladium(II). *J. Org. Chem.* **2017**, *82*, 10855-10865.

(16) Lukashenko, A. V.; Osyanin, V. A.; Osipov, D. V.; Klimochkin, Y. N. Reaction of Push–Pull Enaminoketones and *in Situ* Generated *ortho*-Quinone Methides: Synthesis of 3-Acyl-4H-chromenes and 2-Acyl-1H-benzof[*l*]chromenes as Precursors for Hydroxybenzylated Heterocycles. *J. Org. Chem.* **2017**, *82*, 1517-1528.

(17) For examples of construction of oxygen-containing heterocycles see: Zhang, T.; Ma, C.; Zhou, J. Y.; Mei, G. J.; Shi, F. Application of Homophthalic Anhydrides as 2C Building Blocks in Catalytic Asymmetric Cyclizations of *ortho*-Quinone Methides: Diastereo and Enantioselective Construction of Dihydrocoumarin Frameworks. *Adv. Synth. Catal.* **2017**, *360*, 1128-1137 and references cited herein.

(18) For an overview of reactions of the o-QMs generated from Mannich bases, see: Barta, P.; Fülop, F.; Szatmári, I. Mannich base-connected syntheses mediated by *ortho*-quinone methides. *Beilstein J. Org. Chem.* **2018**, *14*, 560-575 and references cited herein.

(19) Frasinyuk, M. S.; Mrug, G. P.; Bondarenko, S. P.; Khilya, V. P.; Sviripa, V. M.; Syrotchuk, O. A.; Zhang, W.; Cai, X.; Fiandalo, M. V.; Mohler, J. L.; Liu, C.; Watt, D. S. Antineoplastic Isoflavonoids Derived from Intermediate *ortho*-Quinone Methides Generated from Mannich Bases. *ChemMedChem* **2016**, *11*, 600-611.

(20) (a) Mazzei, M.; Nieddu, E.; Folli, C.; Caci, E.; Galletta, L. V. J. 2-(Dialkylamino)-4H-1-benzopyran-4-one derivatives modify chloride conductance in CFTR expressing cells. *Farmaco* **2003**, *58*, 961-970. (b) Mazzei, M.; Dondero, R.; Sottofattori, E.; Melloni, E.; Minafra, R. Inhibition of neutrophil O₂⁻ production by unsymmetrical methylene derivatives of benzopyrans: their use as potential antiinflammatory agents. *Eur. J. Med. Chem.* **2001**, *36*, 851-861.

(21) Mutai, P.; Pavadai, E.; Wiid, I.; Ngwane, A.; Baker, B.; Chibale, K. Synthesis, antimycobacterial evaluation and pharmacophore modeling of analogues of the natural product formononetin. *Bioorg. Med. Chem. Lett.* **2015**, *25*, 2510-2513.

(22) Epstein, J.; Michel, H. O.; Rosenblatt, D. H.; Plapinger, R. E.; Stephani, R. A.; Cook, E. Reactions of Isopropyl Methylphosphonofluoride with Substituted Phenols. II. *J. Am. Chem. Soc.* **1964**, *86*, 4959-4963.

(23) Monti, S. A.; Castillo, G. D. Reverse Mannich reaction of some 5-hydroxyindoles. *J. Org. Chem.* **1970**, *35*, 3764-3767.

(24) Grinev, A. N.; Venevtseva, N. K. Mannich bases in series of derivatives of 5-hydroxybenzofuran. *Russ. J. Gen. Chem.* **1963**, *33*, 806-810.

(25) Frasinyuk, M.; Mrug, G. P.; Bondarenko, S. P.; Sviripa, V. M.; Zhang, W.; Cai, X.; Fiandalo, M.; Mohler, J. L.; Liu, C.; Watt, D. Application of Mannich Bases to the Synthesis of Hydroxymethylated Isoflavonoids As Potential Antineoplastic Agents. *Org. Biomol. Chem.* **2015**, *13*, 11292-11301.

(26) Böhme, T. M.; Augelli-Szafran, C. E.; Hallak, H.; Pugsley, T.; Serpa, K.; Schwarz, R. D. Synthesis and Pharmacology of Benzoxazines as Highly Selective Antagonists at M₄ Muscarinic Receptors. *J. Med. Chem.* **2002**, *45*, 3094-3102.

(27) Phillips, J. P.; Fernando, Q. Chelating Properties of 8-Quinolinol Mannich Bases. *J. Am. Chem. Soc.* **1953**, *75*, 3768-3769.