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An efficient asymmetric synthesis of Fmoc-L-cyclopentylglycine via diastereoselective alkylation of glycine enolate equivalent

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Abstract—Stereoselective alkylation of the enolate derived from benzyl (2R,3S)-(-)-6-oxo-2,3-diphenyl-4-morpholinecarboxylate (1) with cyclopentyl iodide afforded anti- α -monosubstituted product, benzyl (2R,3S,5S)-(-)-6-oxo-2,3-diphenyl-5-cyclopentyl-4-morpholinecarboxylate (3) in 60% yield. Catalytic hydrogenolysis over PdCl₂ cleaved the auxiliary ring system to give L-cyclopentylglycine (4) in 84% yield. Subsequent protection of the α -amino function with Fmoc-OSu gave Fmoc-L-cyclopentylglycine (5) in high yield. © 2003 Elsevier Science Ltd. All rights reserved.

1. Introduction

Nature has afforded us 22 naturally occurring coded amino acids commonly found in proteins isolated from eukaryotic and prokaryotic sources. A host of enzymes are present in nature to extend this repertoire much further by utilizing post-translational modification.¹ A variety of synthetic means have also been established to produce a plethora of non-proteinogenic amino acids as tools to investigate enzymatic mechanisms, extend biological half-life, establish a specific conformational determinant or increase potency of therapeutically interesting peptides.² Of these, stereoselective homologation of readily available chiral auxiliaries, such as cyclic glycine enolate equivalents derived from bis-lactim ethers,³ imidazolidinones,⁴ and oxazinones⁵ are parglycylsultam⁶ useful. Besides, pseudoephedrine glycinate hydrate⁷ are also useful αamino acid templates.

Cyclopentylglycine (Cpg) is a competitive inhibitor of isoleucine uptake in *E. coli*⁸ and also has been used in designing angiotensin II antagonists. It has been synthesized via S_N2 displacement of bromoglycinate with an organometallic reagent followed by epimerization. Syntheses of racemic 2-cyclopentenylglycine, 11 cyclopentylglycine, 8 and 2-cyclopentadieneylglycine have also been reported. In this publication, we wish to report a short and efficient asymmetric synthesis of

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Fmoc-L-cyclopentylglycine (5) by using benzyl (2R,3S)-(-)-6-oxo-2,3-diphenyl-4-morpholinecarboxylate (1) as a template. The reasons behind choosing this chiral auxiliary were: (1) commercial availability, (2) excellent optical purity of the final product, (3) high reactivity towards unactivated electrophiles and (4) scalability.

2. Results and discussion

As shown in Scheme 1, chiral auxiliary 1 was alkylated with cyclopentyl iodide in the presence of lithium bis(trimethylsilyl)amide base. Enolate generation at -78° C followed by quenching with alkylating agent at the same temperature did not result in any reaction. Optimum conditions utilized dissolving 1 and cyclopentyl iodide in THF/HMPA (10:1) by heating to $\sim 35^{\circ}$ C, generating enolate at -78° C and allowing the reaction mixture to warm to room temperature over a period of 2 h. Under these conditions, alkylated product 3 was obtained in 60% yield.[†]

Variations in experimental conditions, such as longer reaction time, increasing the amount of base (>1.5

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[†] **Compound 3**: ESI-MS: 456 ($C_{29}H_{30}NO_4$) (M+H)⁺. ¹H NMR (600 MHz, CDCl₃): δ 7.26–7.07 (13H, m), 6.52 (2H, d, J=6.8 Hz), 6.00 (1H, d, J=3.0 Hz), 5.13 (1H, d, J=3.0 Hz), 4.91 (1H, d, J=12.0 Hz), 4.85 (1H, d, J=12.0 Hz), 3.74 (1H, m), 2.44 (1H, m), 1.90–1.80 (4H, m), 1.65–1.55 (4H, m); mp 216–217°C; [α]_D²⁴ –44.88° (c 0.5, CH₂Cl₂). Anal. (recrystallized from EtOAc/hexanes) calcd for C₂₉H₂₉NO₄: C, 76.48; H, 6.37; N, 3.07. Found: C, 76.58, H, 6.29; N, 3.28.

Scheme 1.

equiv.) or cyclopentyl iodide (>5 equiv.) and substituting the solvating agent HMPA with DMPU either did not result in any improvements or lowered the yield of the alkylation product 3. Since cyclopentyl iodide is an unactivated electrophile, raising the reaction temperature to 40°C was expected to improve the yield of alkylation product 3. However, reaction at 40°C for 1 h resulted in the formation of 10–15% of side product 6 (Fig. 1), in addition to alkylation product 3.

It is important to note that no dialkylation product was formed. Furthermore, only *trans*-alkylated product (3) formed and no *cis*-diastereoisomer (i.e. 2R,3S,5R) was detected by HPLC. The high diastereoselectivity of the enolate alkylation can be explained by considering the expected boat conformation of 2 that disposes the phenyl ring at C-3 in a pseudoaxial orientation, creating steric shielding of the same face at C-5 position from electrophilic attack as shown in Scheme 1.

No purification by column chromatography was necessary. The *trans*-alkylated product **3** was easily crystallized out as a white solid from EtOAc/hexane in 60% of isolated yield (>99% de by HPLC) after standard workup (cf. *syn*-alkylated product is usually an oil). Additional evidence of stereoselectivity was apparent from ¹H NMR as the methine protons (CH) at C-2 and C-3 appeared at δ 6.00 and 5.13 ppm, respectively (0.87 ppm apart, characteristic of *anti*-alkylation product; cf. 0.6–0.7 ppm apart for the *syn*-alkylated product). ¹⁰

Cleavage of the auxiliary ring system 3 was performed using $\rm H_2$ and $\rm PdCl_2$ as a catalyst in THF/MeOH (2:1) solvent mixture at 60 psi for 48–60 h. After removing the catalyst (pyrophoric), the solvent was removed and the residue was triturated with ether to afford Cpg 4 in almost quantitative yield. The contaminating byproduct, 1,2-diphenylethane was easily removed by extracting the dilute aqueous HCl solution of 4 with EtOAc. Removal of aqueous solvent followed by crys-

tallization of the syrup from MeOH/EtOAc afforded 4 in 84% yield.[‡]

Protection of the α -amino function of Cpg 4 was accomplished by treating it with Fmoc-OSu in the presence of Na₂CO₃ overnight in dioxane/H₂O (1.5:1) solvent mixture. Fmoc-Cpg-OH (5) was obtained in quantitative yield after crystallization from EtOAc/hexane.

Thus, an efficient asymmetric synthesis of Fmoc-Cpg-OH (5) from a commercially available chiral auxiliary was successfully accomplished on multigram scale. Synthesis was easy to perform, as no chromatographic purification was required at any step. Furthermore, excellent optical purity (>99%) and high yield (50% overall) was obtained.

Figure 1.

^{*} Compound 4: ESI-MS: 144 ($C_7H_{13}NO_2$) (M+H)⁺. ¹H NMR (600 MHz, DMSO- d^6): δ 4.74 (1H, m), 2.48 (1H, m), 1.78–1.69 (2H, m), 1.53–1.42 (4H, m), 1.39–1.25 (2H, m); mp 245–247°C (dec.); [α] $_D^{24}$ +12.34° (c 0.5, MeOH); (lit. $_0^{10}$, +11.6°, c 0.49, 1N HCl)). Anal. (recrystallized from MeOH/EtOAc) calcd for $C_7H_{13}NO_2$: C, 58.74; H, 9.09; N, 9.79. Found: C, 58.70, H, 8.85; N, 9.99.

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