

Liebigs Ann. Chem., 726, 25 (1969).

- (28) J. G. Cannon, J. C. Kim, and M. A. Aleem, *J. Med. Chem.*, **15**, 348 (1972).
 (29) J. H. Burn, "Practical Pharmacology," Blackwell Scientific Pub-

lications, Oxford, 1952, pp 35-37.

- (30) S. Z. Langer, *J. Pharmacol. Exp. Ther.*, **154**, 14 (1966).
 (31) E. E. Smitsman, S. El-Antably, L. W. Hedrich, E. J. Walaszek, and L-F. Tseng, *J. Med. Chem.*, **16**, 109 (1973).

Antiparasitic 5-Nitrothiazoles and 5-Nitro-4-thiazolines. 2[†]

Peter J. Islip,* Michael D. Closier, and John E. Weale

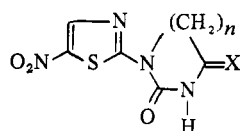
Chemistry Department, Research and Development Division, Parke, Davis and Co., Pontypool, United Kingdom.

Received February 12, 1973

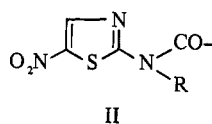
The synthesis of a wide variety of 3-[*N*-(5-nitro-2-thiazolyl)acylamino]propionamides IV and [1-(2-substituted ethyl)-1-(5-nitro-2-thiazolyl)-3-substituted]ureas V as potential antibacterial and antiparasitic agents is described. Treatment of 2-bromo-5-nitrothiazole with 3-aminopropionitrile afforded 3-[(5-nitro-2-thiazolyl)amino]propionitrile (1), the key intermediate. Some of the compounds prepared showed potent schistosomicidal, trichomonocidal, and/or antibacterial activities.

Many 2-amino-5-nitrothiazole derivatives have been shown to possess antiamebic,^{2,3} antihistomonal,⁴ and antitrichomonal^{3,5} properties, and some have antischistosomal³ activity. Thus, 1-(5-nitro-2-thiazolyl)-2-imidazolidinone (niridazole) (Ia) has been found to be effective in the treatment of human schistosomiasis and amebiasis and to give good results when used against dracunculiasis and strongyloidiasis. Recently the nitrothiazolylhydantoin Ib and -hydrouracil Ic have also been shown to possess antiparasitic activities.^{6,7}

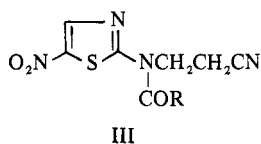
Whereas 2-(alkyl- and arylamino)-5-nitrothiazoles are largely devoid of antischistosome activity,⁸ many antiparasitic nitrothiazoles, including niridazole (Ia), 2-acetamido-5-nitrothiazole (aminithiazole), and 1-ethyl-3-(5-nitro-2-thiazolyl)urea (nithiazide), contain partial structure II in which R is H, CH₂, etc.



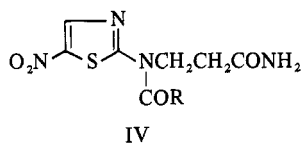
Ia, X = H₂; n = 1
 b, X = O; n = 1
 c, X = O; n = 2



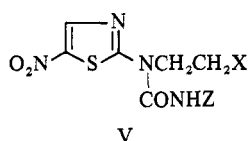
As part of a program to prepare novel chemotherapeutic agents, the synthesis of some 3-[*N*-(5-nitro-2-thiazolyl)acylamino]propionamides IV and [1-(2-substituted ethyl)-1-(5-nitro-2-thiazolyl)-3-substituted]ureas V was undertaken. Thiazoles IV contain partial structure II, in which R is



III



IV



V

CH₂CH₂CONH₂, and may be regarded as open-chain analogs of 1-(5-nitro-2-thiazolyl)hydrouracil (Ic). In particular, the preparation of 3-[*N*-(5-nitro-2-thiazolyl)acetamido]propionamide (24) was examined since this compound could be re-

garded as a possible metabolite of the known^{6,7} active antischistosomal agent Ic.

Chemistry. The compounds described in the present work were derived from 3-[(5-nitro-2-thiazolyl)amino]propionitrile (1), which was prepared by treatment of 2-bromo-5-nitrothiazole with 3-aminopropionitrile in THF. Compounds 5-58 are listed in Tables I-III, and details of the synthesis of these and nitrothiazoles 1-4 and 59-61 are given in the Experimental Section.

	Z	R		Z	R
1	CN	H	59	CO ₂ H	COMe
2	CONH ₂	H	60	CO ₂ Et	COMe
3	CO ₂ Et	H	61	CONH(CH ₂) ₃ NMe ₂	COMe
4	CO ₂ H	H			

Biological Activity. The compounds described in this paper were tested against a Puerto Rican strain of *Schistosoma mansoni* in mice by Dr. Paul E. Thompson and co-workers of Parke, Davis and Co., Ann Arbor, Mich.[‡] As in previous work, drugs were administered in a powdered diet for 14 days. Table IV lists the more active nitrothiazole derivatives, and it can be seen that schistosomicidal activity is present in a limited number of widely varying structural types. It was found that while 3-[*N*-(5-nitro-2-thiazolyl)acetamido]propionamide (24) did possess moderate antischistosome activity, the butyryl derivative 27 appeared to be the most potent propionamide IV. This latter compound effected an 85% kill of worms in mice when administered at ca. 305 mg/kg per day. While the formyl congener 23 possessed slight but significant schistosomicidal properties, somewhat surprisingly the chloroacetyl and propionyl analogs 25 and 26 were inactive in the mouse primary screen. Lengthening the carbon chain of the acyl group (RCO) in IV appeared to reduce efficacy (28 and 29 and higher homologs), as did the use of cycloalkyl R moieties 35-37. Alkyl substitution at the amide nitrogen apparently had little effect, since *N,N*-diethylamide 60 (corresponding to primary amide 24) was also an active schistosomicide. However, the basic *N*-(dimethylamino)propylamide 61 had no activity in *S. mansoni* infected mice. Other propionamides IV, 3-[(5-nitro-2-thiazolyl)amino]propionamide (2), all nitriles III, and 3-[(5-nitro-2-thiazolyl)amino]propionitrile (1) were inactive in the mouse primary screen.

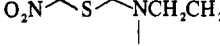
In the nitrothiazolylurea series V, surprisingly in view of

[†] For part 1 of this series, see ref 1.

* Author to whom correspondence should be addressed at Chemical Research Laboratories, Wellcome Research Laboratories, Beckenham, Kent, England.

[‡] For a description of test methods, see ref 9.

Table III. [1-(2-X-Ethyl)-1-(5-nitro-2-thiazolyl)-3-Z]ureas

<div style="text-align: center;">  </div>									
Compd	X	Z	Method	Reaction time, hr	Starting compd	% yield	Mp, °C	Recrystn solvent	Formula
43	CN	H	D		1	37	187-189 dec	EtOH	C ₇ H ₇ N ₃ O ₃ S
44	CN	COCH ₃	E	2	1	60	158-159 dec	EtOH	C ₉ H ₉ N ₃ O ₄ S
45	CN	COCH ₂ Br	E	2.5	1	29	153-155 dec	EtOH	C ₉ H ₈ BrN ₃ O ₄ S
46	CN	COC ₂ H ₅	E	3.5	1	45	148-150 dec	EtOH	C ₁₀ H ₁₁ N ₃ O ₄ S
47	CN	COCH ₂ CH ₂ Br	E	16	1	38	157-158 dec	EtOH	C ₁₀ H ₁₀ BrN ₃ O ₄ S
48	CN	Et	E	4.5 ^b	1	39	170-172	<i>i</i> -PrOH	C ₉ H ₁₁ N ₃ O ₃ S
49	CONH ₂	H	C ^a	0.75	43	63	201-202 dec	AcOH	C ₇ H ₉ N ₃ O ₄ S
50	CONH ₂	COCH ₃	E	4	2	43	196-198 dec	Aqueous DMF	C ₉ H ₁₁ N ₃ O ₅ S
51	CONH ₂	COCH ₂ Cl	E	4	2	63	173-175 dec	EtOAc-petrol ^c	C ₉ H ₁₀ ClN ₃ O ₅ S
52	CONH ₂	COCHCl ₂	E	4	2	69	165-166 dec	EtOAc-petrol ^c	C ₉ H ₉ Cl ₂ N ₃ O ₅ S ^d
53	CONH ₂	COC ₂ H ₅	E	5	2	41	180-182 dec	EtOAc-petrol ^c	C ₁₀ H ₁₁ ClN ₃ O ₅ S
54	CONH ₂	COCH ₂ CH ₂ Br	E	4	2	37	173-174 dec	EtOAc-petrol ^c	C ₁₀ H ₁₂ BrN ₃ O ₅ S
55	CONH ₂	Et	C ^a	0.75	48	27	185-186	<i>i</i> -PrOH	C ₉ H ₁₁ ClN ₃ O ₄ S
56	CO ₂ C ₂ H ₅	COCH ₂ CH ₂ Br	E	2	3	81	129-131	EtOH	C ₁₂ H ₁₅ BrN ₄ O ₆ S
57	CO ₂ H	COC ₂ H ₅	E	2	4	31	146-147 dec	EtOH	C ₁₀ H ₁₂ N ₄ O ₆ S · EtOH
58	CO ₂ H	COCH ₂ CH ₂ Br	E	3	4	37	148-149 dec	EtOH	C ₁₀ H ₁₁ BrN ₄ O ₆ S · EtOH

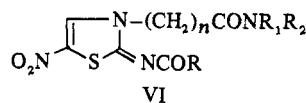
^aReaction mixture not diluted with H₂O. ^bReaction effected in refluxing toluene. ^cPetroleum ether, bp 60–80°. ^dC: calcd, 29.2; found, 30.0.

Table IV. Effects of Nitrothiazoles 1-61 against *S. mansoni* in Mice

Compd	Drug		% reduction in live schistosomes ^c
	Route × days ^a	mg/kg per day	
23	D × 14	347	20
24	D × 7 then D × 7	118, then 238 ^b	59
27	D × 14	305	85
28	D × 14	305	5
29	D × 14	426	39
35	D × 7 then D × 7	160, then 238	15
37	D × 14	360	25
43	D × 14	308	95
44	D × 7 then D × 7	167, then 198	33
60	D × 14	339	49
Niridazole	D × 14	249	99

^aD represents drug-diet. ^bRepresents a dose reduction based on percentage in diet. ^cGroups of six animals were used in the diet studies. The worm burden of the controls averaged 15 per mouse.

amides **24** and **27** did possess marked antischistosome properties, the degree of activity is somewhat less than that shown by the 5-nitro-4-thiazoline-3-acetamides (**VI**, $n = 1$) described earlier.¹ In the latter series of compounds, however, it was shown that analogs bearing the propionamide side chain (*i.e.*, **VI**, $n = 2$) were all inactive in the mouse primary screen. A more detailed comparison of the structure-



activity relationships existing in the present series of compounds, other nitrothiazoles, and the 5-nitro-4-thiazolines described earlier¹ is given in the following paper.¹⁰ However, it may not be inappropriate at this stage to point out that the discovery of schistosomicidal properties in propionamides IV is unexpected in view of the structure-specificity generally associated with analogs of niridazole (Ia).¹¹⁻¹³

Compounds 1-61 were also evaluated against *Trichomonas vaginalis* *in vitro* and in mice.⁸ In the *in vitro* screen, many of the nitrothiazoles were trichomonocidal at concentrations

of 1.56–6.25 $\mu\text{g/ml}$. When tested further against intraperitoneal *T. vaginalis* infections in mice, compounds **7**, **11**, **26**, and **30** were found to cure the infections when administered by gavage at 100 mg/kg twice daily for 3 days; compound **35** was effective at 50 mg/kg twice daily for 3 days; and thiazoles **24**, **27**, **31**, and **47** were curative at 25 mg/kg twice daily when administered similarly.

Nitrothiazoles **1-61** were also tested *in vitro* against a variety of bacteria according to procedures described previously.[#] The most active of these compounds (**24**, **51**, and **52**) had minimum inhibitory concentrations of 0.16, 0.31, and 0.31 µg/ml against *Streptococcus pyogenes* C203; 10, 5, and 5 µg/ml against *Salmonella typhimurium* V-31; and 20, 20, and 20 µg/ml against *Escherichia coli* Vogel, respectively. When tested against *S. pyogenes* infections in mice by oral (po) and subcutaneous (sc) administration, *N*-(2-carbamoyl-ethyl)-*N*-(5-nitro-2-thiazolyl)butyramide (**27**) had ED₅₀ 8.2 (sc) and 9.6 mg/kg (po), and nitrothiazoles **24** and **52** had ED₅₀ <25 mg/kg (sc and po).

Experimental Section**

The physical properties of most of the compounds prepared are collected in Tables I-III, and the experimental details below relate to these tables.

3-[(5-Nitro-2-thiazolyl)amino]propionitrile (1). 2-Bromo-5-nitrothiazole (121 g) was added in portions to a solution of 3-aminopropionitrile (86 g) in THF (1350 ml) at room temperature. The mixture was stirred 3.5 hr at room temperature and then the filtered solution was evaporated *in vacuo*. The residue was stirred with H₂O, and solid was collected and washed with H₂O, cold *i*-PrOH, and then Et₂O to give the product (103.5 g), mp 155–157° dec after recrystallization from 50% EtOH. *Anal.* (C₆H₆N₄O₂S) C, H, N.

3-[(5-Nitro-2-thiazolyl)amino]propionamide (2). A mixture of nitrile **1** (10.0 g) and concentrated HCl (200 ml) was stirred until solid all dissolved (*ca.* 0.5 hr). Dilution with H₂O (100 ml) and neutralization (NaHCO₃) precipitated the amide **2** (5.14 g), mp 190–191° (from EtOAc). *Anal.* (C₆H₈N₄O₃S) C, H, N.

***N*-(5-Nitro-2-thiazolyl)- β -alanine Ethyl Ester (3).** Nitrile 1 (12.0 g) was stirred 1 hr at room temperature with concentrated HCl

For the general *in vitro* and *in vivo* test procedures, see ref 15.

**Melting points are corrected and were determined in capillary tubes. Analytical results were obtained for C, H, and N for all compounds and unless otherwise stated were within $\pm 0.4\%$ of the theoretical values.

§ For a description of test methods, see ref 14.

(240 ml), and solution was evaporated *in vacuo* at 100°. Recrystallization of the residue from EtOH afforded the ester **3** (5.3 g), mp 135–137°. An analytical sample (recrystallized three times from EtOH) had mp 141–143°. *Anal.* (C₈H₁₁N₃O₄S) C, H, N.

N-(5-Nitro-2-thiazolyl)-β-alanine (4). A mixture of the nitrile **1** (5.0 g) and concentrated HCl (100 ml) was heated 1 hr at 100° and then cooled. Addition of H₂O and isolation with EtOAc furnished the acid **4** (3.2 g), mp 162–163° [from EtOAc–petroleum ether (bp 60–80°)]. *Anal.* (C₈H₇N₃O₄S) C, H, N.

2-[N-(2-Cyanoethyl)acetamido]-5-nitrothiazole (5). Method A. A mixture of nitrile **1** (5.0 g), Ac₂O (10 ml), and AcOH (10 ml) was heated at 100° (reaction time given) and cooled, and the solution was then poured slowly onto crushed ice. The oil that separated rapidly crystallized, and the solid was collected, washed well with H₂O, and recrystallized from EtOH to afford **5** (56%), mp 182–184°.

N-(2-Cyanoethyl)-3-methyl-N-(5-nitro-2-thiazolyl)butyramide (11). Method B. Isovaleryl chloride (10.88 g, 20% excess) was added dropwise to a solution of nitrile **1** (14.85 g) in pyridine (75 ml) at 10°. The mixture was stirred at room temperature (time given) and then poured into H₂O. The aqueous layer was decanted from the oil, which was then washed with H₂O and triturated with *i*-PrOH. Several crops of material were obtained, and these were combined and recrystallized (twice) from *i*-PrOH (charcoal) to give **11** (48%), mp 89–90.5°.

N-(2-Carbamoylethyl)-N-(5-nitro-2-thiazolyl)propionamide (26). Method C. A suspension of *N*-(2-cyanoethyl)-*N*-(5-nitro-2-thiazolyl)-propionamide (**7**) (8.8 g) in concentrated HCl (88 ml) was stirred at room temperature (time given) and then poured into H₂O. Solid was collected, washed thoroughly with H₂O, dried, and recrystallized (charcoal) from *i*-PrOH. Amide **26** (42%) had mp 158–160°.

1-(2-Cyanoethyl)-1-(5-nitro-2-thiazolyl)urea (43). Method D. A solution of nitrile **1** (27.0 g) in THF (800 ml) was added over 0.75 hr to a stirred solution of COCl₂ in toluene (12.5% w/v, 900 ml), and the mixture was then stirred 3 hr at room temperature and 0.5 hr at 40°. Excess COCl₂ was removed with a stream of N₂, and the reaction mixture was left overnight at room temperature. The cooled (0°) vigorously stirred mixture was then saturated with NH₃ and evaporated *in vacuo*. The residue was stirred with H₂O (ca. 800 ml), and the solid was collected, dried, and recrystallized from EtOH (charcoal) to give urea **43** (37%), mp 187–189° dec.

3-Acetyl-1-(2-cyanoethyl)-1-(5-nitro-2-thiazolyl)urea (44). Method E. Acetyl isocyanate (3.34 g, 20% excess) in THF (10 ml) was added dropwise to a solution of nitrile **1** (5.9 g) in THF (150 ml), and the mixture was then stirred at room temperature (reaction time given). A small amount of insoluble material was filtered off, and the filtrate was evaporated *in vacuo* to furnish urea **44** (60%), mp 158–159° dec (from EtOH).

N-Acetyl-N-(5-nitro-2-thiazolyl)-β-alanine (59). A mixture of acid **4** (13.29 g), Ac₂O (40 ml), and AcOH (40 ml) was heated 2 hr at 100° and then poured into ice–H₂O to give the *N*-acetyl derivative **59** of acid **4** (59%), mp 171–173° (from EtOAc). *Anal.* (C₈H₇N₃O₅S) C, H, N.

***N,N*-Diethyl-3-[N-(5-nitro-2-thiazolyl)acetamido]propionamide (60).** Ethyl chloroformate (3.55 ml) was added to a solution of

acid **59** (9.6 g) and NEt₃ (5.5 ml) in CHCl₃ (300 ml) at 0°, and the mixture was stirred 0.5 hr at 0°. Redistilled NHET₃ (11.0 ml) was added dropwise at 0°, and the mixture was stirred 5 min at 0° and 1 hr at room temperature. The organic layer was washed with H₂O (5 × 100 ml), dried (MgSO₄), and evaporated to provide the *N,N*-diethyl analog **60** of primary amide **24** (33%), mp 122–124° (from *i*-PrOH). *Anal.* (C₁₂H₁₈N₄O₄S) C, H, N.

***N*-[3-(Dimethylamino)propyl]-3-[N-(5-nitro-2-thiazolyl)acetamido]propionamide (61).** Treatment of the mixed anhydride of acid **59** (from 7.1 g of acid) with 3-(dimethylamino)propylamine (8.3 g) at 0° (*cf.* preparation of **60**) afforded the *N*-(dimethylamino)-propyl analog **61** of **24** (39%), mp 128–130° (from *i*-PrOH). *Anal.* (C₁₃H₂₁N₅O₄S) C, H, N.

Acknowledgments. The authors are indebted to Dr. R. E. Bowman for advice and encouragement, to Miss E. M. Tanner for spectroscopic measurements, to Mr. F. H. Oliver for microanalyses, to Mr. J. P. Guyett for some technical assistance, and to Drs. M. W. Fisher and P. E. Thompson and their associates of the Departments of Microbiology and Experimental Therapeutics, respectively, Parke, Davis and Co., Inc., Detroit and Ann Arbor, for the biological results.

References

- (1) P. J. Islip, M. D. Closier, M. C. Neville, L. M. Werbel, and D. B. Capps, *J. Med. Chem.*, **15**, 951 (1972).
- (2) E. F. Elslager in "Medicinal Chemistry," A. Burger, Ed., 3rd ed, Wiley-Interscience, New York, N. Y., 1969.
- (3) L. M. Werbel, *Top. Med. Chem.*, **3**, 125 (1970).
- (4) L. P. Joyner, S. F. M. Davies, and S. B. Kendall, *Exp. Chemother.*, **1**, 333 (1963).
- (5) R. J. Schnitzer, *ibid.*, **1**, 289 (1963).
- (6) Ciba Ltd., Netherlands Patent Application 6,505,226 (1965).
- (7) P. J. Islip, British Patent 1,070,675 (1967).
- (8) L. M. Werbel, E. F. Elslager, A. A. Phillips, D. F. Worth, P. J. Islip, and M. C. Neville, *J. Med. Chem.*, **12**, 521 (1969).
- (9) P. E. Thompson, J. E. Meisenhelder, and H. Najarian, *Amer. J. Trop. Med. Hyg.*, **11**, 31 (1962).
- (10) P. J. Islip, M. D. Closier, and M. C. Neville, *J. Med. Chem.*, **16**, 1030 (1973).
- (11) P. Schmidt and M. Wilhelm, *Angew. Chem., Int. Ed. Engl.*, **5**, 857 (1966).
- (12) C. H. Robinson, E. Bueding, and J. Fisher, *Mol. Pharmacol.*, **6**, 604 (1970).
- (13) L. M. Werbel and J. R. Battaglia, *J. Med. Chem.*, **14**, 10 (1971).
- (14) P. E. Thompson, A. Bayles, S. F. Herbst, B. Olszewski, and J. E. Meisenhelder, *Antibiot. Chemoth.*, **9**, 618 (1959).
- (15) M. W. Fisher, M. C. Manning, L. A. Gagliardi, M. R. Gaetz, and A. R. Erlandson, *Antibiot. Annu.*, 1959–1960, 293 (1960); M. W. Fisher, *Proc. Soc. Exp. Biol. Med.*, **85**, 538 (1954).

Antiparasitic 5-Nitrothiazoles and 5-Nitro-4-thiazolines. 3¹

Peter J. Islip,* Michael D. Closier, and Martin C. Neville

Chemistry Department, Research and Development Division, Parke, Davis and Company, Pontypool, United Kingdom.

Received February 12, 1973

Alkylation of the sodium salt of 2-formamido-5-nitrothiazole in *N,N*-dimethylformamide with a variety of alkylating agents is shown to give exclusively exocyclic *N*-alkylated products IV. Removal of the *N*-formyl group was readily achieved with hydrazine hydrate or 1 equiv of sodium hydroxide, and the resulting aminothiazoles V were treated with several acid chlorides and isocyanates to give (acylamino)-thiazoles VI. Some of the nitrothiazoles IV, V, and VI exhibited moderate activity against *Schistosoma mansoni*, *Trichomonas vaginalis*, and a range of gram-positive and gram-negative bacteria.

The potent antischistosome activity of various 2-(acylimino)-5-nitro-4-thiazoline-3-acetamides I against *Schistosoma mansoni* infections in mice has been described re-

cently.² Thiazolines I, in which R is alkyl, aryl, or alkoxy, etc., were prepared (together with varying amounts of the exocyclic *N*-alkylated thiazoles III) by alkylation of the sodium salt of the appropriate thiazolylamide II in DMF.

In the present work, it was found that alkylation of the sodium salt of 2-formamido-5-nitrothiazole under similar

*Author to whom correspondence should be addressed at Chemical Research Laboratory, Wellcome Research Laboratories, Beckenham, Kent, England.